

rTPCflowmeter_v3

June 29, 2020

```
[1]: import numpy as np
      from scipy import interpolate

[2]: import matplotlib.pyplot as plt
      %matplotlib inline
      plt.rcParams["figure.figsize"] = (15,8)
      x=np.linspace(5.0,150.0,300)
```

1 Executive Summary

The total flow prior to mixing is lower than the values extracted from the tube with mixed gas. Consequently, the amount of carbon dioxide appears to be lower than reported. The inconsistency between the two flow measurements, i.e., “before” and “after” the mixing, could be due to the definitive and known calibration of the rotameters for a single gas, while it is reported by the manufacturer that the calibration for a gas mixture is subjected to variations.

2 Calibration

2.1 Setup for atmosphere Argon and Stainless Steel ball

[https://www.mathesongas.com/pdfs/flowchart/602%20\(E300\)/ARGON%20602\(E300\)%20SS%200%20PSIG.pdf](https://www.mathesongas.com/pdfs/flowchart/602%20(E300)/ARGON%20602(E300)%20SS%200%20PSIG.pdf)

$\phi_{\text{Ar}}(z)$

```
[3]: ro,cal=np.loadtxt('ARGON_602(E300)_SS_0_PSIG.dat',unpack=True) # sccm
      arflow=interpolate.interp1d(ro, cal)
```

2.2 Setup for atmosphere Carbon Dioxide and Stainless Steel ball

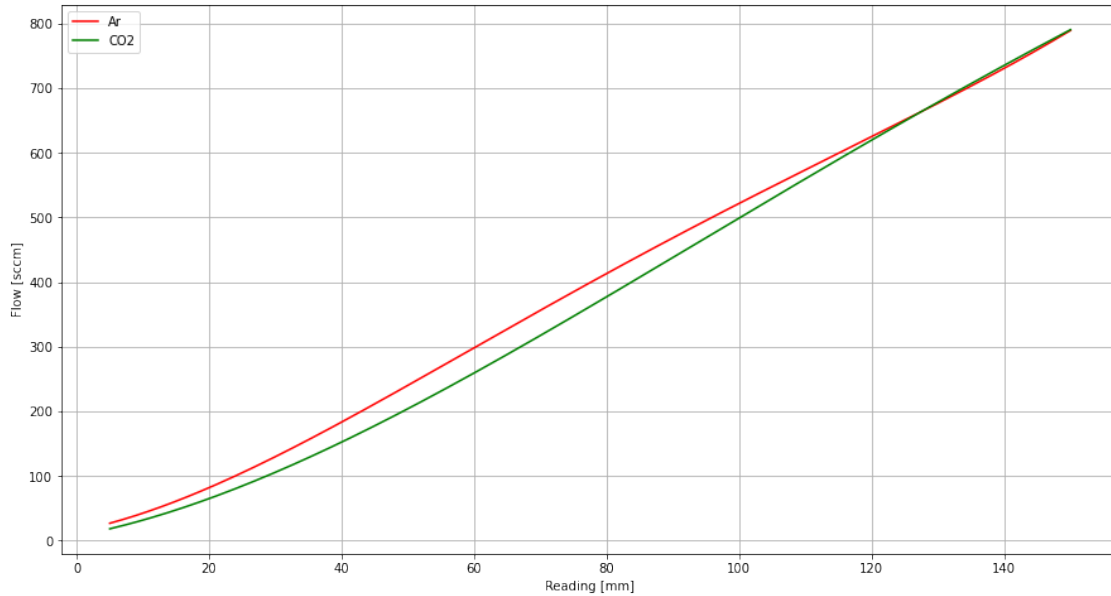
[https://www.mathesongas.com/pdfs/flowchart/602%20\(E300\)/CARBON%20DIOXIDE%20602\(E300\)%20SS%200%20PSIG.pdf](https://www.mathesongas.com/pdfs/flowchart/602%20(E300)/CARBON%20DIOXIDE%20602(E300)%20SS%200%20PSIG.pdf)

$\phi_{\text{CO}_2}(z)$

```
[4]: ro,cal=np.loadtxt('CARBON_DIOXIDE_602(E300)_SS_0_PSIG.dat',unpack=True) # sccm
      co2flow=interpolate.interp1d(ro, cal)
```

2.3 Plots

```
[5]: plt.plot(x, arflow(x), 'r', label='Ar')
plt.plot(x, co2flow(x), 'g', label='CO2')
plt.xlabel('Reading [mm]')
plt.ylabel('Flow [sccm]')
plt.grid()
plt.legend(loc='upper left')
plt.show()
```



3 Flow of a the gas mixture

Mixture fraction $f_{Ar} = 1 - f_{CO_2}$

```
[24]: f_co2=0.3
```

3.1 Determination of the return fraction

Take the reading from the middle of the ball z_{in}

<https://photos.app.goo.gl/wgFVmY5JfMx3TixcA>

and z_{out}

<https://photos.app.goo.gl/bVF7iXANj8YrsXhC7>

```
[25]: zin=101
      zout=66
```

Determine input Ar flow as $f_{\text{Ar}}\phi_{\text{Ar}}(z_{\text{in}})$

```
[26]: inar=arflow(zin)*(1.0-f_co2)
      print(f'Ar inflow {inar:.1f} sccm')
```

Ar inflow 369.0 sccm

Determine input CO₂ flow as $f_{\text{CO}_2}\phi_{\text{CO}_2}(z_{\text{in}})$

```
[27]: inco2=co2flow(zin)*f_co2
      print(f'CO2 inflow {inco2:.1f} sccm')
```

CO₂ inflow 151.6 sccm

Determine input flow as $\phi_{\text{in}} = f_{\text{CO}_2}\phi_{\text{CO}_2}(z_{\text{in}}) + f_{\text{Ar}}\phi_{\text{Ar}}(z_{\text{in}})$

```
[28]: inFlow=inar+inco2
      print(f'Gas Mixture inflow {inFlow:.1f} sccm')
```

Gas Mixture inflow 520.6 sccm

Determine output flow ϕ_{out} as above by replacing $z_{\text{in}} \rightarrow z_{\text{out}}$

```
[29]: outFlow=arflow(zout)*(1.0-f_co2)+co2flow(zout)*f_co2
      print(f'Gas Mixture outflow {outFlow:.1f} sccm')
```

Gas Mixture outflow 321.6 sccm

Determine return fraction as $\frac{\phi_{\text{out}}}{\phi_{\text{in}}}$

```
[30]: print(f'Return fraction {outFlow/inFlow*100.0:.1f}%')
```

Return fraction 61.8%

3.2 Determination of the flow prior to mixing

Carbon Dioxide flow $\phi_{\text{CO}_2}(z_1)$ for reading z_1

<https://photos.app.goo.gl/5inXHJiXBmY5LRRJ6>

```
[13]: z1=41 # reading
      inCO2Flow=co2flow(z1)
      print(f'CO2 inflow {inCO2Flow:.1f} sccm')
```

CO₂ inflow 157.5 sccm

Argon flow $\phi_{\text{Ar}}(z_2)$ for reading z_2

<https://photos.app.goo.gl/p2os5w12aqfPPWdU8>

```
[14]: z2=89
      inArFlow=arflow(z2)
      print(f'Ar inflow {inArFlow:.1f} sccm')
```

Ar inflow 463.1 sccm

Total flow prior to mixing $\phi_{\text{sum}} = \phi_{\text{CO}_2}(z_1) + \phi_{\text{Ar}}(z_2)$

```
[15]: totFlow=inCO2Flow+inArFlow  
      print(f'Total input Flow {totFlow:.1f} sccm')
```

Total input Flow 620.7 sccm

Calculate CO₂ fraction $f_{\text{CO}_2} = \frac{\phi_{\text{CO}_2}(z_1)}{\phi_{\text{sum}}}$

```
[16]: co2frac=inCO2Flow/totFlow  
      print(f'CO2 fraction {co2frac*100:.1f}%')
```

CO2 fraction 25.4%

4 Conclusion

The weighted average of the calibrated flow for the two components of the gas mixture, where the weights are given by the volume fraction of each component, seems to yield resonable values for the input flow. The calculated value of the return is in accordance with what was previously seen with Argon only.

However, if one tries to determine the input flow from the rotameters prior to mixing, the sum flow is about 20% higher. The largest difference between the two methods appears in the determination of the Ar flow. It follows that the fraction of carbon dioxide is lower than expected.