

# Temperature conversion

Read this carefully and practice with the questions at the bottom of the page

Degrees Celsius                       $^{\circ}\text{C} = \frac{5}{9}(^{\circ}\text{F} - 32)$

Degrees Fahrenheit                   $^{\circ}\text{F} = \frac{9}{5}(^{\circ}\text{C}) + 32 = 1.8 (^{\circ}\text{C}) + 32$

Temperature differences:

The Boiling Point Rise is the difference between the boiling point of the pure solvent and the boiling point of the solution.

Example: Temperature of pure solvent is  $\theta_1$  ; Temperature of solution is  $\theta_2$

$$\theta_{1\text{F}} = 212^{\circ}\text{F}; \quad \theta_{2\text{F}} = 221^{\circ}\text{F}$$

Convert to  $^{\circ}\text{C}$ :  $\theta_{1\text{C}} = \frac{5}{9}(212-32) = 100 ^{\circ}\text{C}$  ;  $\theta_{2\text{C}} = \frac{5}{9}(221-32) = 105 ^{\circ}\text{C}$ .

$$\text{BPR } ^{\circ}\text{F} = (\theta_{2\text{F}} - \theta_{1\text{F}}) = 221 - 212 = 9 ^{\circ}\text{F}$$

$$\text{BPR } ^{\circ}\text{C} = (\theta_{2\text{C}} - \theta_{1\text{C}}) = 105 - 100 = 5 ^{\circ}\text{C}$$

Check: BPR, Fahrenheit

$$\theta_{2\text{F}} - \theta_{1\text{F}} = 221 - 212 = 9 ^{\circ}\text{F}$$

BPR, Celsius

$$\theta_{2\text{C}} - \theta_{1\text{C}} = \frac{5}{9}(221-32) - \frac{5}{9}(212 - 32) = \frac{5}{9}(221) - \frac{5}{9}(32) - \frac{5}{9}(212) + \frac{5}{9}(32) = \frac{5}{9}(221-212) = 5 ^{\circ}\text{C}$$

$$\text{BPR } ^{\circ}\text{C} = \frac{5}{9} (\text{BPR } ^{\circ}\text{F})$$

$$\text{BPR } ^{\circ}\text{F} = \frac{9}{5} (\text{BPR } ^{\circ}\text{C}) = 1.8 (\text{BPR } ^{\circ}\text{C})$$

Question 1. Convert 239  $^{\circ}\text{F}$  to  $^{\circ}\text{C}$ .

(Answer: 115  $^{\circ}\text{C}$ )

Question 2. Convert 120  $^{\circ}\text{C}$  to  $^{\circ}\text{F}$

(Answer: 248  $^{\circ}\text{F}$ )

Question 3. BPR is 13  $^{\circ}\text{F}$ ; calculate BPR in  $^{\circ}\text{C}$ .

(Answer: 7.2  $^{\circ}\text{C}$ )

Question 4. BPR is 6  $^{\circ}\text{C}$ ; calculate BPR in  $^{\circ}\text{F}$ .

(Answer: 10.8  $^{\circ}\text{F}$ )