Major Test (PHL654, Experimental techniques)

Semester -I (2008-09)

Maximum Marks: 50 Duration: 2 Hr

Attempt all questions.

- 1. (a) Explain the basic principle of spectroscopic ellipsometry?
 - (b) What information you can get from ellipsometric data?
 - (c) How does parallel and perpendicular components of reflection varies with incidence angle for Si (n=3.5, k=0), glass (n=1.5) Show the proper graphs.

(4+3+4)

- 2. (a) Why do you need thermal analysis?
 - (b) Which information can you get from TGA.
 - (b) Compare DSC and DTA techniques.

(2x3)

3. Answer the following (any FIVE)

- a. Which vibrational mode in a molecule is IR active? Give two examples of IR active bonds.
- b. Name any two materials which are used for making windows and cells in IR spectroscopy instrument and why?
- c. Which transitions are responsible for absorption in the 200-700 nm range?
- d. What are the various fundamental modes of vibration in CH₂ group?
- e. How can you separate isotopes using IR spectroscopy?
- f. How does the stretching frequency vary as we change the bonds from
 - (i) C-H, C-D, C-C and (ii) C-C, C=C, C=C ⁹ Give reasons.

(3x5)

4. Answer the following (any SIX)

- (i) How can you prepare sample for TEM studies in the case of (i) powder, (ii) thin films?
- (j) Why do we need differential pumping in electron or ion gun sources?
- (k) What are the factors which influence resolution in SEM and how?
- (l) Compare secondary and backscattered electrons. (Give three points)
- (m) Why do you get broad core level peaks in XPS?
- (n) Name two sources which are used in Ultra violet photoelectron spectroscopy. Why?
- (o) X rays has high penetration depth still XPS is a surface sensitive technique, why?

(3x6)