

BIRLA INSTITUTE OF TECHNOLOGY AND SCIENCE, PILANI

K. K. BIRLA GOA CAMPUS

FIRST SEMESTER 2020-2021

Course Handout

Date: 19/08/2020

Course No. : **CHE F414**
Course Title : **TRANSPORT PHENOMENA**
Instructor-in-charge : **Dr. Asima Shaukat**

1. Course Description:

Analogy among momentum, heat and mass transport phenomena; Shell balance approach for all three transport phenomena, Boundary layer concepts, velocity, temperature and concentration distribution in laminar flow; Velocity distributions in turbulent flow.

2. Scope and Objective of the Course:

Transport phenomena is a subject of importance in the field of engineering. All the three transport mechanisms namely momentum, heat and mass transfer frequently occur in chemical processes either individually or simultaneously. The aim of this course is to understand the analogy between the transport mechanisms and how the same methods of solution can be applied to the problems involving different transport processes.

3. Books:

Text Book

1. Bird, Stewart and Lightfoot, 'Transport Phenomena', John Wiley & Sons, 2002, 2nd edition

Reference Books

1. Fox and McDonald, 'Introduction to fluid dynamics,' John Wiley & Sons, 2008, 7th edition
2. Holman, J.P., 'Heat transfer', McGraw Hill, 1997, 8th edition

4. Course Plan:

Lecture No.	Learning Objectives	Topics to be covered	Text Book Chap./Sec
1	Introduction	Scope and objectives of course	Ch. 0
2	Molecular momentum transport	Newton's law of viscosity	1.1-1.2
3	Convective momentum transport	Convective momentum transport	1.7
4	Momentum balances	Shell momentum balances, boundary conditions	2.1
5 - 7	Velocity distributions in	Examples	2.2-2.5

	laminar flow		
8	Problems	Problems for Ch. 2	
9	Equations of change for isothermal systems	Equations of continuity, motion and mechanical energy	3.1-3.3
10	Equations of change in terms of substantial derivative	Partial, total and substantial derivative	3.5
11 - 12	Applications of equations of change	Examples	3.6
13	Problems	Problems for Ch. 3	
14	Molecular energy transport	Fourier's law of heat conduction	9.1
15	Convective energy transport	Convective transport of energy, work associated with molecular motions	9.7-9.8
16-18	Energy balances	Shell energy balances, examples	10.1-10.7
19-20	Convection	Forced and free convection	10.8-10.9
21	Problems	Problems for Ch. 10	
22	Equations of change for non-isothermal systems	Various forms of energy equations	11.1-11.3
23-24	Applications of equations of change	Examples	11.4
25	Problems	Problems for Ch. 11	
26	Molecular mass transport	Fick's law of binary diffusion	17.1
27	Convective mass transport	Mass and molar transport by convection, summary of mass and molar fluxes	17.7-17.8
28	Mass balances	Shell mass balance, boundary conditions	18.1
29 - 30	Concentration distributions in solids and laminar flow	Examples	18.2-18.7
31	Problems	Problems for Ch. 18	
32-33	Equations of change for multicomponent systems	Equations of continuity, summary of multicomponent equations of change	19.1-19.4
34	Problems	Problems for Ch. 19	
35	Velocity distributions with more than one independent variables	Time-dependent flow of Newtonian fluids	4.1
36-37	Stream function and velocity potential	Solving flow problems using stream functions and velocity potential	4.2-4.3
38	Flow near solid surfaces	Boundary layer theory	4.4
39	Problems	Problems for Ch. 4	
40-41	Temperature distributions with more than one independent variable	Unsteady state heat conduction in solids, Steady potential flow	12.1-12.3 (R2 - 3.5, 4.6)
42	Turbulent flow	Time-smoothed equations of change and velocity profiles	5.1-5.3

4. Evaluation Scheme:

EC No.	Evaluation Component	Weightage (%)	Date & Time	Duration
1	Test 1	15	September 14 (During scheduled class hour)	30 min
2	Test 2	15	October 16 (During scheduled class hour)	30 min
3	Test 3	15	November 16 (During scheduled class hour)	30 min
4	Quizzes/assignments	20	To be announced	-
5	Comprehensive	35	05/12/20 (AN)	120 min

Notices:

Notices, if any, concerning the course will be uploaded on the Moodle

Make-up:

Make-up will be granted for genuine cases only. Prior permission of IC is compulsory. No make-up for Quizzes will be given.

Consultation Hour: Monday, 5:00 to 6:00 pm.

**Instructor-in-charge
(CHE F414)**