

Tutorial 12

P101 $I = neAv_d$

Using this data:

Atomic weight of Cu = 63.5, density = 8.94 g/cc.

8940 kg occupies 1 m^3 , each contributes 1 electron.

$$n = \frac{8940}{63.5} \times \frac{N_A}{1 \text{ m}^3} \times 10^{23} = 8.475 \times 10^{28} \text{ m}^{-3}$$

$$\frac{I}{neA} = v_d = \frac{10}{8.475 \times 10^{28} \times 1.6 \times 10^{-19} \times 10^{-6}} = 7.375 \times 10^{-4} \text{ m/s}$$

P102 Relaxation time $\tau = \frac{\sigma m}{ne^2} = \frac{m}{\rho ne^2}$

$$= \frac{9.1 \times 10^{-31}}{1.5 \times 10^{-8} \times 8.5 \times 10^{28} \times (1.6 \times 10^{-19})^2} = 2.78 \times 10^{-14} \text{ sec.}$$

P103

$$\text{Resistivity } \rho = \frac{5.8 \times 10^{-3}}{1} \times 3.3 \times 10^{-6} = 1.914 \times 10^{-8} \text{ ohm m}$$

$$\therefore \text{conductivity } \sigma = 5.22 \times 10^7 (\text{ohm-m})^{-1}$$

$$\text{Current density } J = \frac{25}{3.3 \times 10^{-6}} = 7.58 \times 10^6 \text{ A/m}^2$$

Now $J = nev_D$ we have to find n to find drift velocity v_D .

63.5 kg of Cu has $10^3 N_A$ atoms ($N_A \rightarrow$ Avogadro's number)
 8940 kg of Cu will have $\frac{8940(N_A \times 10^3)}{63.5} = 8.475 \times 10^{28}$

As 8940 kg occupies 1 m^3 volume, assuming each atom contributes one free electron, we have

$$n = 8.475 \times 10^{28} \text{ m}^{-3}$$

$$\text{Thus, } v_D = \frac{7.58 \times 10^6}{8.475 \times 10^{28} \times 1.6 \times 10^{-19}} = 5.6 \times 10^{-4} \text{ m/s}$$

Note: drift velocity is very small in comparison to even classical rms speed. To calculate mean free path in Drude model, one has to calculate rms speed.

$$\frac{1}{2} m u_{\text{rms}}^2 = \frac{3}{2} kT \quad kT \approx 0.025 \text{ eV at room temperature}$$

$$\therefore u_{\text{rms}} = \sqrt{\frac{3 \times 0.025 \times 1.6 \times 10^{-19}}{9.1 \times 10^{-31}}} \approx 1.15 \times 10^5 \text{ m/s}$$

$$\text{Relaxation time } \tau = \frac{\sigma m}{ne^2} = \frac{5.22 \times 10^7 \times 9.1 \times 10^{-31}}{8.475 \times 10^{28} \times (1.6 \times 10^{-19})^2} \approx 2.19 \times 10^{-14} \text{ s}$$

Mean free path in Drude model

$$l = u_{rms} \times \tau = 1.15 \times 10^5 \times 2.19 \times 10^{-14} \approx 2.52 \text{ nm}$$

Mean free path in Sommerfield model

$$l = v_F \times \tau$$

$$E_F \text{ for Cu} = 7.06 \text{ eV} \Rightarrow v_F = 1.58 \times 10^6 \quad (E_F = \frac{1}{2} m v_F^2)$$

$$\therefore l = 34.7 \text{ nm}$$

P104 $\sigma = \frac{n e^2 \tau}{m^*}$, where m^* is the effective mass

for Na

$$n = \frac{6.02 \times 10^{26} \times 970}{23} \approx 2.54 \times 10^{28}$$

$$\tau = \frac{\sigma m^*}{n e^2} = \frac{2.17 \times 10^7 \times 1.2 \times 9.1 \times 10^{-31}}{2.54 \times 10^{28} \times (1.6 \times 10^{-19})^2} \approx 3.64 \times 10^{-14} \text{ s}$$

$$l = 1.15 \times 10^5 \times 3.64 \times 10^{-14} \approx 4.19 \text{ nm (in Drude model)}$$

(for sommerfield, $E_F = \frac{1}{2} m^* v_F^2$ & calculate l using v_F)

$$v_D = \frac{J}{n e} = \frac{\sigma E}{n e} = \frac{2.17 \times 10^7 \times 100}{2.54 \times 10^{28} \times 1.6 \times 10^{-19}} \approx 0.53 \text{ m/s}$$

P105

$$\text{Let } E, v \text{ vary as } e^{-i\omega t} \quad \left| \begin{array}{l} \therefore v(t) = v e^{-i\omega t} \\ \therefore E(t) = E e^{-i\omega t} \\ \therefore \frac{dv(t)}{dt} = -i\omega v e^{-i\omega t} \end{array} \right.$$

$$m \left(\frac{dv}{dt} + \frac{v}{\tau} \right) = -eE$$

$$m \left(-i\omega v e^{-i\omega t} + \frac{v e^{-i\omega t}}{\tau} \right) = -eE e^{-i\omega t}$$

$$m \left(-i\omega v + \frac{v}{\tau} \right) = -eE \Rightarrow v = \frac{-eE/m}{1/\tau - i\omega} = \frac{-eE\tau}{m} \frac{1+i\omega\tau}{1+(\omega\tau)^2}$$

$$\text{Electric current density} = J = n(-e)v = \sigma E$$

$$\frac{ne^2E\tau}{m} \left[\frac{1+i\omega\tau}{1+(\omega\tau)^2} \right] = \sigma E \Rightarrow \sigma = \frac{ne^2\tau}{m} \left[\frac{1+i\omega\tau}{1+(\omega\tau)^2} \right]$$

P106.

Fermi energy for any element:

$$E_F = \frac{\hbar^2}{2m} \left(3\pi^2 \frac{N}{V} \right)^{2/3} \approx 5.84 \times 10^{-38} \left(\frac{N}{V} \right)^{2/3} \text{ Joules}$$

$$\frac{N}{V} = \frac{N_A \times \text{density in g/cc} \times 10^6}{\text{Atomic weight in g.}} \quad \left| \quad \approx 3.646 \times 10^{-19} \left(\frac{N}{V} \right)^{2/3} \text{ eV} \right.$$

Example for Na.

$$\frac{N}{V} = \frac{6.02 \times 10^{26} \times 0.971 \times 10^6}{22.99} = 2.543 \times 10^{28} \cdot \text{Substituting}$$

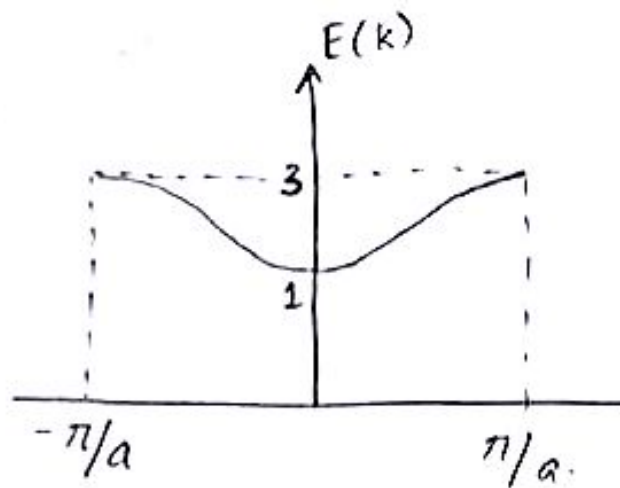
$$E_F \approx 3.16 \text{ eV}$$

	Li	Na	K	Rb	Cs.
E_F (in eV)	4.72	3.16	2.14	1.82	1.53

(approx).

P108.

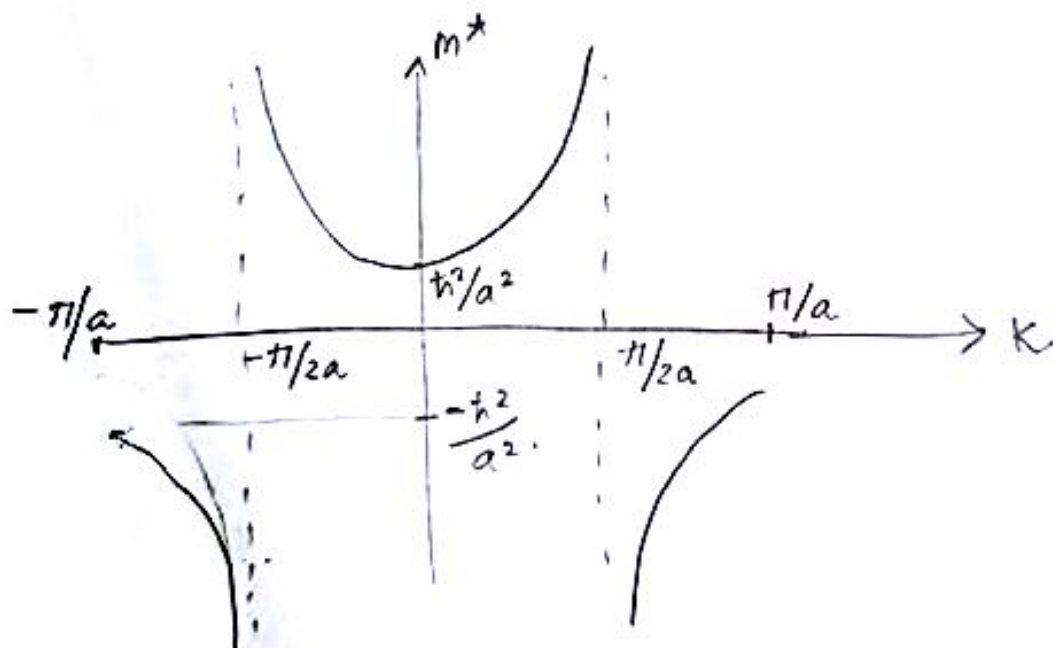
(a)



(b)
$$v = \frac{p}{m} = \frac{\hbar k}{m} = \frac{1}{\hbar} \frac{dE}{dk} = \frac{1}{\hbar} a \sin ka$$

$$m^* = \frac{\hbar^2}{\frac{d^2 E}{dk^2}} = \frac{\hbar^2}{a^2 \cos(ka)}$$

(c)



(P107 → same as P99, Tutorial 11).