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Reg. Number:

**NATIONAL INSTITUTE OF TECHNOLOGY CALICUT  
DEPARTMENT OF CHEMISTRY**

**B. Tech. Semester -I, Monsoon Semester -Test II, October 2013**

**CY1001 Chemistry**

**Time: 1 hr**

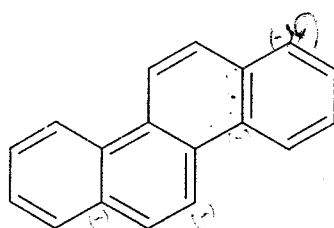
**Maximum Marks: 20**

**Answer all questions**

1. Which of the following molecules are microwave active? Justify your answer.

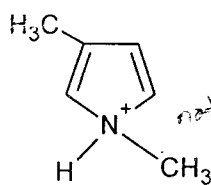
(a) HD (b) *cis*-CH<sub>3</sub>CH=CHCH<sub>3</sub> (c) *meta*-dichlorobenzene (d) CH<sub>2</sub>Cl<sub>2</sub> (2 Marks)

2. Suggest the most stable resonance structure of the following molecule.

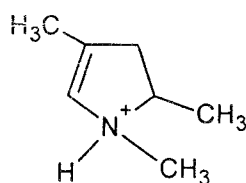


(2 Marks)

3. Arrange the following compounds in the increasing order of their pK<sub>a</sub> values. Explain.



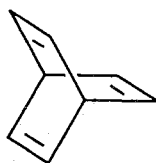
(i)



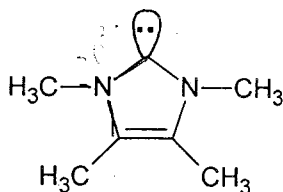
(ii)

(2 Marks)

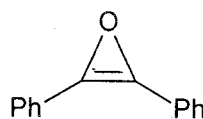
4. Classify the following compounds as aromatic, anti-aromatic and non-aromatic with proper justification.



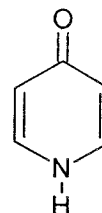
(i)



(ii)



(iii)

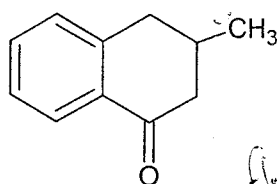


(iv)

(2 Marks)

5. Sketch the first four lines of the rotation spectrum of the mixture of <sup>12</sup>C<sup>16</sup>O and <sup>13</sup>C<sup>16</sup>O where the inter-nuclear distance is 112.8 pm. Calculate the number of revolutions per second which <sup>12</sup>C<sup>16</sup>O and <sup>13</sup>C<sup>16</sup>O undergo at J = 5 state. Comment on your answer. (4 Marks)

6. Suggest a possible synthetic route with appropriate mechanism for the preparation of the following compound from benzene.

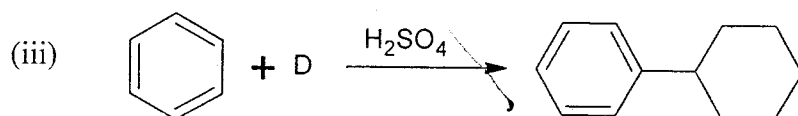
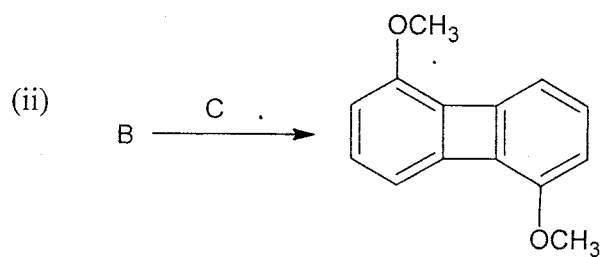
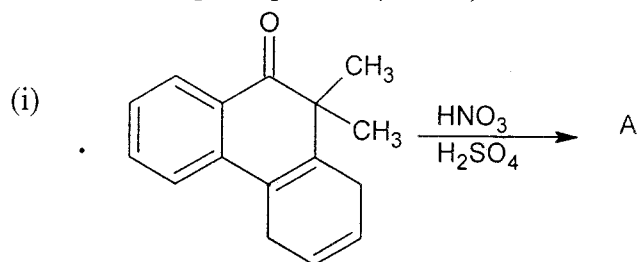


$J=0$        $E=0$   
 $J=1$        $E=2B$   
 $J=2$        $E=6B$   
 $J=3$        $E=12B$   
 $J=4$        $E=20B$

(4 Marks)

P. T. O.

7. Identify the missing compounds (A to D) in the following reactions.



(4 Marks)