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WebAssign is built to withstand the rigors of teaching calculus to today's students. Our patent-pending grading engine works like a real professor and recognizes all algebraically equivalent answers to even the most complex problems.

Calculus - WebAssign

I think the correct answer from the choices listed above is the first option. If a pure substance consists of hydrogen and oxygen and a mixture consists of hydrogen and oxygen, then both the pure substance and the mixture will be identical.

If a pure substance consists of hydrogen and oxygen and a ...

Find an answer to your question A base is added to water. What happens when the position of equilibrium lies to the left? A. The base is completely dissociatin...

A base is added to water. What happens when the position ...

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EXAMPLE 2 - Predicting the Effect of Disruptions on Equilibrium: Ammonia gas, which is used to make fertilizers and explosives, is made from the reaction of nitrogen gas and hydrogen gas. The forward reaction is exothermic. Consider a system in which the gases are compressed to a volume that is small enough to yield a total pressure of about 300 atm.

Changing Volumes and Equilibrium - Mark Bishop

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