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KEY - CP - Mixed Mole Problems

7. How many moles of nitrogen gas will occupy a volume of 347 mL at 6680 torr and 270 C? (C 300k) 8. What volume .11 454 grams (1 lb) of hydrogen occupy at 1.05 atm and 250 C? (I K) qqq 9. Find the number of grams of CO₂ that exert a pressure of 785 torrs at the volume 0 32.5 L and a temperature of 320 C. 10. An elemental gas has a mass of 10.3 g.

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Moles and Mass Directions: Determine the number of moles in each of the quantities below. 1) 52 g of NaCl ANSWER: 0.89 mole 2) 145 g of H₂ SO₄ ANSWER: 1.48 mole 3) 110. g of KMnO₄ ANSWER: 0.70 mole 4) 74 g of KCl ... ANSWER: 570 g Chemistry IF8766 Page 50 Instructional Fair, Inc. Title: Name:

Moles and Mass - Ms. Agostine's Chemistry Page

How many moles of oxygen are necessary to react completely with four moles of propane (C₃H₈)? 5. + 3KNO₃ + How many moles of potassium nitrate are produced when two moles of potassium phosphate react with moles of aluminum nitrate? enstructional Fair, Inc. Chemistry IF8766 62

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The Mole and Volume Directions ... What volume will the following quantities of gasses occupy at STP? 1) 5.00 mole of H₂ ANSWER: 112. L 2) 2.50 moles of O₂ ANSWER: 56.0 L 3) 0.850 moles of N₂ ... ANSWER: 87.5 L 8) 56.0g of N₂ ANSWER: 44.8 L 9) 70. g of CO₂ ANSWER: 36. L 10) 20. g of NH₃ ANSWER: 26 L Chemistry IF8766 Page 51 Instructional ...

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KEY Molarity: • a ____ description of solution concentration. ... Problems: Show all work and circle your final answer. 1. To make a 4.00 M solution, how many moles of solute will be needed if 12.0 liters of solution are required? 4.00 M = moles of solute 12.0 L moles of solute = 48.0 mol 2. How many moles of sucrose are dissolved in 250 mL ...

Molarity: Molarity = 1. 2. - Central Bucks School District

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