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In a hypothetical lab of this nature a student found that 10.0 grams of KCIOx contained 6.10 g of KCI. 1) What is the mass of oxygen that was left? 2) How many moles of KCI were present? 3) how many moles of O were present? 4) what is the smallest whole-number ratio of moles of KCI to moles of O? 5) what is the formula of this hypothetical compound of KCIOx?

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Labs play a significant role in AP Chemistry, and knowledge of experimental procedures is critical for the final exam. If you're an AP Chemistry student (or are considering becoming one), this guide will help you understand what to expect from the lab components of the course.

How to Ace Your AP Chemistry Labs - PrepScholar

Department of Chemistry GENERAL CHEMISTRY 101 LABORATORY MANUAL An Inquiry Approach through an Environmental Awareness The following laboratories have been compiled and adapted by Alan Khuu, M.S. & Armando Rivera, Ph.D.

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