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1) Water is exposed to infrared radiation of wavelength  $2.9 \times 10^{-4} \text{ cm}$  . Assume that all the radiation

is absorbed and converted to heat. How many photons will be required to raise the temperature of 2.5g of water by 2.4K ? 2) Suppose that in an alternate universe, the possible values of  $l$  were the integer values from 0 to  $n$  (instead of 0 to  $n-1$ ).

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answer.  $q = mcT = (110.0 \text{ g})(4.18 \text{ J/(g}\cdot\text{C)})(35.6^\circ\text{C} - 15.0^\circ\text{C})$  ... Question (d) Explain why  $\text{I}$  is a solid at room temperature whereas  $\text{Br}$  is a liquid. Your explanation should clearly reference the types and relative strengths of the intermolecular forces present in each substance. ... AP Chemistry Scoring Guidelines, 2016

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