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Chemistry of Life packet 1 ANSWER KEY (1) - The Chemistry... It also explains how atoms of the same element can have a different number of neutrons and describes the two main types of chemical bonds. Atoms (page 35) 1. The basic unit of matter is called a (an) atom . It is the center of an atom, made up of protons and 2. Describe the nucleus of an atom. neutrons. Complete the table about subatomic particles.

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Therefore, students' answers should mention a function involving recovery from fever, or reduction of body temperature after exercise. 1. nucleus 2. element 3. compound 4. ionic, covalent 5. van der Waals forces 6. A sodium atom is a neutral particle that contains equal numbers of protons and electrons.

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Chapter 2: Introduction to the Chemistry of Life. The elements carbon, hydrogen, nitrogen, oxygen, sulfur, and phosphorus are the key building blocks of the chemicals found in living things. They form the carbohydrates, nucleic acids, proteins, and lipids (all of which will be defined later in this chapter) that are the fundamental molecular...

Chapter 2: Introduction to the Chemistry of Life ...

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Chemistry of Life - Macromolecules and Enzymes

The Chemistry Of Life ANSWER KEY Section 2-1 The Nature Of Matter (pages 35-39) This Section Identifies The Three Particles That Make Up Atoms. It Also Explains How Atoms Of The Same Element Can Have A Different Number Of Neutrons And Describes The Two Main Types Of Chemical Bonds. Atoms (page 35) 1.

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