

## *Amadeo Creations Stoichiometry Answers*

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2 a. How many moles of  $\text{O}_2$  can be produced by ...

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370 Chapter 11 • Stoichiometry EXAMPLE Problem 11.1 Interpreting Chemical Equations The combustion of propane (  $C_3H_8$  ) provides energy for heating homes, cooking food, and soldering metal parts. Interpret the equation for the combustion of propane in terms of representative particles, moles, and mass.

### Chapter 11: Stoichiometry

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Practice Problems: Stoichiometry. Balance the following chemical reactions: Hint a.  $CO + O_2 \rightarrow CO_2$  b.  $KNO_3 \rightarrow KNO_2 + O_2$  c.  $O_3 \rightarrow O_2$  d.  $NH_4NO_3 \rightarrow N_2O + H_2O$  e.  $CH_3NH_2 + O_2 \rightarrow CO_2 + H_2O + N_2$  Hint f.  $Cr(OH)_3 + HClO_4 \rightarrow Cr(ClO_4)_3 + H_2O$  Write the balanced chemical equations of each reaction:

### Practice Problems: Stoichiometry

Counting units are used for certain kinds of objects depending on their size. For example, eggs are counted in twelves, called dozens, and shoes are counted in pairs. Paper is counted in reams and pencils are counted in the gross. Chemists also need a unit for counting very small particles like ...

### Stoichiometry : Counting Particles & Avogadro's Number Quiz

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