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ChemActivity 32 Molarity 185 Figure 1: Cations and anions in a solid crystal and in water. ions in a crystal Hno ions in aqueous solution Critical Thinking Questions 1. ionic compound, NaCl or NapS04, does Fioure 1 represent? Explain your reasonina clearly. 2. a) \That is the identity of the cation in Fioure 1?

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ChemActivity 32 Molarity Information Some molecules do not dissociate into ions when dissolved in water. Sugars (glucose, sucrose, dextrose, etc.) and alcohols are examples. These compounds do not dissociate into ions upon dissolution and they do not increase the conductivity of water.

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ChemActivity 32 Molarity Which is more concentrated with respect to sodium ions, 50.0 g of NaCl in

500.0 mL of solution or 59.0 g ofNa2S04 in 500.0 ml. of solution? Which is more concentrated with respect to sodium ions, 0.50 M NaCl or 0.30 M Na2S04? Which C TQ was easier to answer, 8 or 9? Why?

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