

Student Solutions Manual For Thermodynamics Statistical

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Student Solutions Manual For Thermodynamics

Theoretical background. The acid dissociation constant for an acid is a direct consequence of the underlying thermodynamics of the dissociation reaction; the pK_a value is directly proportional to the standard Gibbs free energy change for the reaction. The value of the pK_a changes with temperature and can be understood qualitatively based on Le Châtelier's principle: when the reaction is ...

Acid dissociation constant - Wikipedia

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