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2 titration with a known amount of a stable acid, which in this case is the monoprotic acid potassium hydrogen phthalate (abbreviated KHP). The reaction of KHP and NaOH is

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[4055] – 204-2- iv) v) b) What is configuration and conformation? Explain with suitable examples. Write in brief about Atropisomerism. 5 2. Answer the following (any 5): 15a) What is mutarotation?

[4055] - 204 - Savitribai Phule Pune University

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ION EXCHANGE REACTIONS OF CLAYS

You will use Beer's law. $A = \epsilon mCl$ The basic idea here is to use a graph plotting Absorbance vs. Concentration of known solutions. Once you have that you can compare the absorbance value of an unknown sample to figure out its concentration. You will be applying Beer's law to calculate the concentration. The equation for Beer's law is: $A = \epsilon mCl$ (A = absorbance, $\epsilon m = molar$ extinction coefficient ...

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Answer: Correct option is . Explanation: Exothermic reactions are the reactions in which energy is released to the surroundings. For these reactions, the energy term should be present on the product side and in the options given the energy term is present in the reaction,

Which reaction is exothermic? CH4 + 202 -- > CO2 + 2H2O ...

Fill in the gap sheet to explain why concentration. surface area, temperature and the presence of a catalyst affect rate of reaction.

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