

## *Supplemental Problems Stoichiometry Answer*

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### **Supplemental Problems Stoichiometry Answer**

Chemistry: Supplemental Stoichiometry Problems Directions: Solve each of the following problems. Assume excess of any reactant that isn't mentioned, unless otherwise specified. Show your work, including proper units, to earn full credit. ... ANSWERS: 2. 18.0 mol O 2 3B. 13.8 g NaCl 3C. NaNO 3 3D. 0.118 mol Cu(NO 3) 2 3E. 4.2 g CuCl 2.

### **Name: Hour: Date: Chemistry: Supplemental Stoichiometry ...**

Chemistry: Supplemental Stoichiometry Problems. Directions: Solve each of the following problems. Assume excess of any reactant that isn't mentioned, unless otherwise . specified. Show your work, including proper units, to earn full credit. 1. Given the balanced equation, show what the following molar ratios (i.e., the conversion factors ...

### **Stoichiometry Worksheet - teachnlearnchem.com**

Chemistry Supplemental Problems Stoichiometry Answer Ap chemistry darrell feebeck, ap chemistry chemistry & chemical reactivity 6th ed kotz, treichel and weaver thomson brookes cole, 2006 / isbn: 978 0 534 99766 3

### **Chemistry Supplemental Problems Stoichiometry Answer PDF ...**

Supplemental Stoichiometry Problems 1. (a) Iron (III) oxide can be reduced with carbon monoxide according to the following equation.  $\text{Fe}_2\text{O}_3(\text{s}) + 3 \text{CO}(\text{g}) \rightarrow 2 \text{Fe}(\text{s}) + 3 \text{CO}_2(\text{g})$  A 16.2 L sample of CO (g) at 1.50 atm and 200. °C is combined with 15.39g of Fe  $\text{Fe}_2\text{O}_3(\text{s})$ .

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Stoichiometry Worksheet #1 Answers 1. Given the following equation:  $2 \text{C}_4\text{H}_{10} + 13 \text{O}_2 \rightarrow 8 \text{CO}_2 + 10 \text{H}_2\text{O}$ , show what the following molar ratios should be. a.  $\text{C}_4\text{H}_{10} / \text{O}_2$  b.  $\text{O}_2 / \text{CO}_2$  c.  $\text{O}_2 / \text{H}_2\text{O}$  d.  $\text{C}_4\text{H}_{10} / \text{CO}_2$  e.  $\text{C}_4\text{H}_{10} / \text{H}_2\text{O}$  2. Given the following equation:  $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$  a. How many moles of  $\text{O}_2$  can be produced by ...

### Stoichiometry Worksheet #1 Answers

Answer keys for homework assignments are listed below. You should use answer keys as a tool, not to plagiarize. For you to be successful in this class you will need to do your own work and ask questions when you need clarification. Do not depend on answer keys to do your homework.

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