

Sweet 16 Compound Tournament Answers

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Sweet 16 Compound Tournament Answers

Sweet 16 Chemistry Compound Tournament With spring just around the corner, your students' thoughts will soon be turning to sunshine, prom, and the NCAA basketball tournament. This clever activity combines the ever-popular March Madness basketball pool with a review of chemical formulas, the structure and charges of ions, and molar mass cal-

Sweet 16 Chemistry Compound Tournament SCIENTIFIC

tournament. This clever activity combines the ever-popular basketball office pool with a review of the formulas of ionic compounds and their solubility. The result is the Sweet 16 Chemistry Ion Tournament. This activity will spark interest in your chemistry class because students will have fun playing the "Tournament."

Sweet 16 Chemistry Ion Tournament

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Sweet 16 Chemistry Compound Tournament - Flinn Scientific

Sweet 16 Chemistry Ion Tournament First Round Second Round Semifinals Finals Winner NiCl_2 $\text{Ni(II)}_2 + \text{Fe(II)}$ S Zn CO_3 Na CrO_4 Mg PO_4 Ag(I) SO_4 Ba NO_3 Al OH Rules 1. First Round →Add Charges 2. Second Round →Write Formula 3. Semis →Soluble Compound (compound that will dissolve in water) 4. Finals →Soluble Product of Chemical Reaction 5.

Name Class/Period Sweet 16 Chemistry Ion Tournament

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