

## *Answers To Electron Configuration Orbital Diagram And*

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### Answers To Electron Configuration Orbital

26 For starters, you know that your element is located in period 4 of the Periodic Table because all the transition metals that have 3d configurations are located in period 4 of the Periodic Table. This means that you can write the noble-gas shorthand configuration for this unknown element, let's say "X", like this "X"^(3+): ["Ar"] 3d^5 Now, the trick here is to realize that when a transition ...

### An unknown element "X" has electron configuration ["Ar"]3d ...

Sodium, like all the group 1 alkali metals, has one valence electron. Valence electrons are the outermost electrons, and are the ones involved in bonding. Sodium has 11 electrons: its atomic number is 11, so it has 11 protons; atoms are neutral, so this means sodium also has 11 electrons. Electrons are arranged in "shells" or energy levels.

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