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ii) 2 hydrogen, 1 sulfur, 4 oxygen iii) 12 carbon, 22 hydrogen, 11 oxygen. iv) 1 carbon, 1 hydrogen, 3 chlorine . 3 Nitrogen has a diatomic molecule with two atoms of nitrogen.

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4 Physical chemistry Charges, chemical formulae and equations Question Answers and guidance  
Marks 1 a)  $2\text{CuCO}_3(\text{s}) + \text{C}(\text{s}) \rightarrow 2\text{Cu}(\text{s}) + 3\text{CO}_2(\text{g})$  1 mark for correct formulae; 1 mark for correct  
balancing 2 b)  $63.5 + 12 + 16 + 16 + 16 = 123.5$  1 1 c) i) Solid 1 ii)  $3 \times 2 = 1.5 \text{ mol}$  1 iii)  $12 + 16 + 16 = 44$   $44 \times 1.5 = 66 \text{ g}$  1 1 1 Total 9 Question ...

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