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Chapters 13-16 Resources - pgasd.com

Date Name TEACHING TRANSPARENCY WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in

bryant.dearbornschools.org

Teaching Transparency . Name Date Class TEACHING TRANSPARENCY The Electromagnetic Spectrum 1. What kinds of waves have the longest wavelength? ... Estimate your answer to the nearest power of ten. KS X 104 Chemistry: Matter and Change Teaching Transparency Worksheet ... 1/13/2016 3:25:16 PM ...

Name Date Class TEACHING TRANSPARENCY The Electromagnetic ...

TEACHING TRANSPARENCY WORKSHEET The Activity Series Use with Chapter 9, Section 9.2 1. For each of the following pairs of elements, underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

TEACHING TRANSPARENCY WORKSHEET The Activity Series

Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 5 9 1. What is the shape of an s orbital? 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a sublevel? 4. How many electrons can each ...

TEACHING TRANSPARENCY MASTER 16 Atomic Orbitals Use with ...

Teaching Transparency 18-1 Teaching Transparency 18-2 Connecting Math to Physics FAST FILEChapters 16-20 Resources, Chapter 18 Transparency 18-3 Master, p. 93 Study Guide, pp. 75-80 Reinforcement, p. 85 Section 18-2 Quiz, p. 82 Mini Lab Worksheet, p. 69 Teaching Transparency 18-3 Connecting Math to Physics FAST FILEChapters 16-20 ...

Section/Objectives Standards Lab and Demo Planning

MATH SKILLS TRANSPARENCY MASTER 16 Math Skills Transparency Masters Chemistry: Matter and Change • Chapter 11 31 number of moles of known substance C H C H C H O O O CO CO H O H O H O number of moles of unknown substance O CO H O C H CO H O C H O C H O CO use mole ratio 5 mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol ...

MATH SKILLS TRANSPARENCY MASTER 16 - trenholm science

TEACHING TRANSPARENCY 18 Chemistry: Matte	ter and Change Teaching Transparency Worksheet 2	
Use with Chapter 6, Section 6.1 The Periodic Table 1. How many elements are listed in the periodic table? 2. What is the atomic number of selenium? 3. What is the symbol for palladium?		
TEACHING TRANSPARENCY		
TEACHING TRANSPARENCY 22 Name Da	Date Class TEACHING TRANSPARENCY 22	
Chemistry: Matter and Change Teaching Transp	sparency Worksheet 2 Use with Chapter 7, Formation	

of Ions Section 7.1 1. ... Chemistry: Matter and Change 3 Teacher Guide and Answers Teaching Transparency 22 ...

Name Date Class TEACHING TRANSPARENCY

TEACHING TRANSPARENCY MASTER ... 16 3 3 10 2 8 10 18 3 3 10 2 10 10 20 3 3 10 2 12 10 22 3 3 10 2 14 Wavelengths (I) in meters Visible light 3 TV, FM AM Microwaves X rays 3 3 10 2 Energy increases The Electromagnetic Spectrum Use with Chapter 5, Section 5.1 15. Name Date Class ... answer to the nearest power of ten.

TEACHING TRANSPARENCY MASTER 15 The Electromagnetic ...

Section Focus Transparency 20 on the overhead projector. Have students answer the accompanying questions using Section Focus Transparency Master 20. 2 Teach Concept Development Help students understand why the periodic table has become such a valuable tool for chemists. Reinforce the point that in Lavoisier's time, the late 1790s, there were only

Chapter 6: The Periodic Table and Periodic Law - irion-isd.org

FAST FILEChapters 16–20 Resources Transparency 16-1 Master, p. 21 Study Guide, pp. 10–11 Section 16-1 Quiz, p. 15 Teaching Transparency 16-1 Connecting Math to Physics This CD-ROM is an editable Microsoft® PowerPoint® presentation that includes: Section presentations Interactive graphics Image bank All transparencies Audio reinforcement

Section/Objectives Standards Lab and Demo Planning

. 16 Study Guide Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 9 7 Examine the parts of the chemical equation at the top of the transparency. Use this information to answer the following questions about Equation 1 and Equation 2. 1. Write Equation 1 as a sentence. 2

Chapters 9-12 Resources - pgasd.com

Math Skills Transparency 25 and Master Teaching Transparency 49 and Master Study Guide for Content Mastery, pp. 94–95 TCR Multimedia Resources Chemistry Interactive CD-ROM, Section 16.4 Experiment and Video MindJogger Videoquizzes, Ch. 16 Guided Reading Audio Program, Section 16.4 Cosmic Chemistry Videodisc, Disc 1, Side 2

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