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Unfortunately, the Supplementary Application submission deadline is firm, and no extensions will be given if this deadline is missed. The questions are posted in mid-September, to provide you with several months to prepare your answers.

FAQs | McMaster BHSC

Taking chemistry further: If you imagine a horizontal plane through the nucleus, with one lobe of the orbital above the plane and the other beneath it, there is a zero probability of finding the electron on that plane. So how does the electron get from one lobe to the other if it can never pass through the plane of the nucleus? At this introductory level you just have to accept that it does!

ATOMIC ORBITALS - chemguide

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