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Thermodynamics and Its Applications

Based on the authors' graduate courses at MIT, this text and reference provides a unified understanding of both the critical concepts of chemical thermodynamics and their applications. KEY TOPICS: Part I of this book provides the theoretical basis of classical thermodynamics, including the 1st and 2nd laws, the Fundamental Equation, Legendre transformations, and general equilibrium criteria.

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The third law of thermodynamics states that "The entropy of a perfect crystal at absolute zero temperature is exactly equal to zero." At absolute zero that is zero Kelvin, the system is said to possess minimum energy. This statement of the third law of thermodynamics holds true only when the perfect crystal has minimum energy state.

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