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iv Chemistry: Matter and Change Study Guide for Content Mastery This Study Guide for Content Masteryfor Chemistry: Matter and Change will help you learn more easily from your textbook. Each textbook chapter has six study guide pages of questions and exercises for you to complete as you read the text.

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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 2 11 In your textbook, read about significant figures. Use each of the terms below just once to complete the statements. 7. The digits that are reported in an answer are called . 8. The numeral 9.66 has three significant figures, two known figures and one figure.

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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 10 55 Chemical ReactionsChemical Reactions Section 10.1 Reactions and Equations In your textbook, read about evidence of chemical reactions. For each statement, write yes if evidence of a chemical reaction is present. Write no if there is no evidence of a chemical reaction. 1.

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Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 21 123 In your textbook, read about calculating cell electrochemical potential. Use the table of standard reduction potentials below to answer the following questions. 20. Suppose you have two voltaic cells whose half-cells are represented by the following

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CHAPTER STUDY GUIDE FOR CONTENT MASTERY

Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 15 89 Section 15.3 Colligative Properties of Solutions In your textbook, read about electrolytes and colligative properties, vapor pressure low-ering, boiling point elevation, and freezing point depression. Use the table to answer the following questions. 1.

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Study Guide for Content Mastery Chapter 2Earth Science: Geology, the Environment, and the Universe7 SECTION 2.1 Latitude and Longitude In your textbook, read about latitude and longitude. Match the definition in Column A with the term in Column B. Column A Column B 1. Science of mapmaking 2.

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Chapter 21 Study Guide For Content Mastery Answers ...

STUDY GUIDE FOR CONTENT MASTERY Section 6.3 Periodic Trends In your textbook, read about atomic radius and ionic radius. Circle the letter of the choice that best completes the statement or answers the question. 1. Atomic radii cannot be measured directly because the electron cloud surrounding the

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STUDY GUIDE FOR CONTENT MASTERY Name CHAPTER 10.1 continued Class STUDY GUIDE FOR CONTENT MASTERY Chemical Reactions Section 10.1 Reactions and Equations In your textbook, read about evidence of chemical reactions. For each statement, write yes if evidence of a chemical reaction is present. Write no if there is no evidence or a chemical reaction.

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Universe 5 CHAPTER 1 STUDY GUIDE FOR CONTENT MASTERY Time (s) 10 20 30 40 50 60 70 80 Mass of product (g) 1.5 3.2 4.3 6.0 7.7 9.2 10.4 12.1 3. On the grid below, plot the mass of product versus time. Connect the data points with a line. 4.

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