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ThermoChemistry (Energy) WebQuest Name _____ Period ____ Part 5- Some more Specific Heat Problems ... Part 6 - Mixed Review Answer the following questions. Show all equations used and calculations done. 1. Use the thermochemical equations shown below to determine the enthalpy for the reaction: ...

ThermoChemistry (Energy) WebQuest

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ThermoChemistry (Energy) WebQuest

ThermoChemistry (Energy) WebQuest Name _____ Period _____ 5 Part 6 Some more Specific Heat Problems

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Calculate the answer and check it. 1. Problem #1. 2. Problem #2. 3. Problem #3. Exothermic. Endothermic. Problem : Given : Show work here. Δ H = Title: ThermoChemistry (Energy) WebQuest Author: CCSD Last modified by: CCSD Created Date: 2/8/2007 4:45:00 AM Company: Cobb County School District Other titles: ThermoChemistry (Energy) WebQuest ...

ThermoChemistry (Energy) WebQuest - Quia

Thermochemistry Answers - Worksheet Number One We will ignore any heats losses to the walls of the container and losses to the air. These is a typical position to take since, in a real experiment, both would have to be accounted for, making for much more complexity.

Thermochemistry Answers - Worksheet Number One

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ThermoChemistry-Webquest | Heat (504 views)

THERMOCHEMISTRY WORKSHEET KEY 1. The following describes the reaction that takes place when a typical fat, glyceryl trioleate, is metabolized by the body: 2C 57H 104O 6(s) + 160O 2(g) \rightarrow 114CO 2(g) + 104H 2O(l) DH° = -6.70 x 104 kJ a. How much heat energy must be expelled by the body to rid the body of one pound of fat?

THERMOCHEMISTRY WORKSHEET KEY - Faculty

Enlow, Jessica (from bhes to HR) Mcnutt, Megan (hrms-location-TBD) Miles, Margaret ... Predicting Reactions WS 1 front answer key back page answer key ... Thermochemistry and Nuclear Chemistry Tutorials. Thermochemistry WS #1 and #2 Answer Key scanned.

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Unit 13: Thermochemistry. STUDY. PLAY. Thermochemistry. The study of energy changes that occur during chemical reactions and changes in state. Absolute Zero. The coldest temperature, 0 Kelvin, that can be reached. It is the hypothetical temperature at which all molecular motion stops.

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