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How many phosphorus atoms does it take to equal 30.973 grams? 130.973M 15 1449's I 0"" gletom) x (3.02m x10" moms 5. Compare your answers to questions 2 and 4. 4--e - . . u A l -. "I - J hey are both :tpproxnnatcly the some. lhcrlurc. notices uncut some" atoms ol carbon to equal the atomic muss in grams and it takes ("03XI l)" :tiums ol ...

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All the elements in the universe heavier than helium â€¦. www.answers.com > â€¦ > Elements and Compounds > Noble Gases Yes, a mole of hydrogen weighs 1 g, a mole of helium weighs 4 g. Still lighter than air, as a mole of air weighs about 30 g.

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(Base the answer to this question on the number of moles of propane that actually get combusted—which is your answer to part a.) 12 moles. For every mole of propane that combusts 3 moles of CO<sub>2</sub> are produced, so the number of moles of CO<sub>2</sub> that can be produced when 4 moles of propane combusts = 4(3) = 12.

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