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1. heterogeneous 2. pure 3. heterogeneous 4. pure 5. homogeneous 6. heterogeneous 7. homogeneous 8. heterogeneous 9. homogeneous 10. pure 11. filtration 12. filtration 13. crystallization 14. distillation 15. chromatography 16. electrolysis 17. distillation 18. A magnet could be used to separate them because iron is attracted to a magnet but aluminum is not.

2 Answer - urbanbhs.weebly.com

16—3 Review.nd Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperatuÈe affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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18. Gas molecules are said to be in	19. The co	ollisions betwee	n gas particle	s are	e 20	. The
temperature of a gas is a measure of	of the average _	$_{__}$ of the gas .	particles. 21	. Gas	particles	exert
on one another 12-1 Review	w and Reinforce	ment				

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Chemistry Chapter ... 9-3 Review and Reinforcement (Continued) Answer each of the following questions in the space provided. 16-2 Review and Reinforcement pages 15-16 Download Chemistry 9 1 Review And Reinforcement Answers PDF chemistry 9 1 review and reinforcement answers such as: alternative investments investor profile questionnaire,

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11—1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. consdrvption of matter coefficients subscripts mass

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10—2 Review and Reinforcement Mole Conversions ... 18. In an experiment, 12.1 g of dry ice (solid C02) were converted into gaseous C02 at STP. How many moles were in the sam le? What was f gaseous C02? 12,1 Coz K C ICE. Ch'3'ter 10 19 . Title: 10-2 Review and Reinforcements ...

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