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Predict the product(s) along with the states, indicate the type of reaction, and balance the following chemical reactions. a. A solution of lead (II) nitrate is mixed with a solution of sodium iodide. b. Solid zinc sulfide reacts with oxygen in the air. c. Liquid butane (C4H10 (I)) is used as a fuel to ignite a lighter.

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Write the word equations for each of the following chemical reactions: 1) When dissolved beryllium chloride reacts with dissolved silver nitrate in water, aqueous beryllium nitrate and silver chloride powder are made. 2) When isopropanol (C3H8O) burns in oxygen, carbon dioxide, water, and heat are produced. ... Balancing Equations Practice ...

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