

## *11 2 Practice Problems Continued Answers*

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### 11 2 Practice Problems Continued

!\\jame ----- Date \_ Class \_ 11-2 Practice Problems (continued) 13. Find the mass of sodium required to 20. What volumes of H<sub>2</sub>S gas and oxygen are produce 5.68 L of hydrogen gas at STP necessary to produce 14.2 L of sulfur from the reaction described by the dioxide gas?

### Name 11-2 Practice Problems - Harrison High ... - FlipHTML5

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0.001934 moles NaI x 2 moles NaCl = 0.001934 mole NaCl. 2 mole NaI. 0.001934 mole NaCl x 58.45 g = 0.11 g NaCl. 1 mole NaCl. Determine the mass of carbon dioxide, CO<sub>2</sub>, produced when 0.85 g of butane, C<sub>4</sub>H<sub>10</sub>, reacts with oxygen according to the following equation: (Hint to balance you will have to multiply by 2) ... 11-2 Practice Problems ...

### 11-2 Practice Problems - Mr. Fischer.com

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2. Identify the limiting reactant when 4.68 g of Fe reacts with 2.88 g of sulfur to produce FeS. 3. ... 11-3 Practice Problems ... Fraction Review Sheet - Quia

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View 11-3-Problems-2 from MATH 102 at Miami University. 11-3 Practice Problems 1. Identify the limiting reactant when 1.22 g of O<sub>2</sub> reacts with 1.05 g of H<sub>2</sub> to produce water. 2. Identify the limiting

### 11-3-Problems-2 - 11-3 Practice Problems 1 Identify the ...

AE Stoichiometry Answer Keys. AE An Introduction to Stoichiometry Answers. AE 11-1 Practice Problems Answers. AE 11-2 Practice Problems Answers. AE Extra Stoichiometry Problems Answers. AE Stoich Practice Sheet Answers. AE 11-3 Practice Problems Answers. AE Percent Yield and Limiting Reactant 2.0 Answers.

### AE Stoichiometry Answer Keys - Baumritter - Google Sites

Jefferson Township High School Science Department. 11-3 Practice Problems Answer key. February 1st, 2013 | Author: kkula

### 11-3 Practice Problems Answer key

Chapter 11 Section 11.2 (continued) CONCEPTUAL PROBLEM 11.5 Answers 15. 2HI → H<sub>2</sub> + I<sub>2</sub> 16. HBr Use Visuals Conceptual Problem 11.5 Have students study the photograph in the problem. Explain that the explosive properties of dynamite are due to the rapid production of large amounts of gases. A related reaction is the decomposition of ...

### 11.2 Types of Chemical Reactions - bleiker.weebly.com

Practice Problems: Stoichiometry (Answer Key) Balance the following chemical reactions: a. 2 CO + O<sub>2</sub> → 2 CO<sub>2</sub> b. 2 KNO<sub>3</sub> → 2 KNO<sub>2</sub> + O<sub>2</sub> c. 2 O<sub>3</sub> → 3 O<sub>2</sub> d. NH<sub>4</sub>NO<sub>3</sub> → N<sub>2</sub>O + 2 H<sub>2</sub>O e. 4 CH<sub>3</sub>NH<sub>2</sub> + 9 O<sub>2</sub> → 4 CO<sub>2</sub> + 10 H<sub>2</sub>O + 2 N<sub>2</sub> f. Cr(OH)<sub>3</sub> + 3 HClO<sub>4</sub> → Cr(ClO<sub>4</sub>)<sub>3</sub> + 3 H<sub>2</sub>O Write the balanced chemical equations of each reaction:

### Practice Problems: Stoichiometry (Answer Key)

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Chapter 11.3 Practice Problems EXPECTED SKILLS: Know how to compute the dot product of two vectors. Be able to use the dot product to find the angle between two vectors; and, in particular, be able to determine if two vectors are orthogonal. Know how to compute the direction cosines of a vector.

### Chapter 11.3 Practice Problems - Drexel University

Practice Problems (continued) II. Analysis of a sample of maple syrup reveals that the concentration of  $\text{OH}^-$  ions is  $5.0 \times 10^{-4} \text{ M}$ . What is the pH of this syrup? Is it acidic, neutral, or basic? 70 12. In a sample of bananas and water, it is found that  $[\text{H}^+] = 2.51 \times 10^{-5} \text{ M}$ . What is the corresponding pH value, and what is the

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