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2) $4.75\text{g} \times 1\text{ kg} = 0.00475\text{ kg}$. 1000g. 3) $0.000278\text{ mole} = 0.0582\text{ m}$. 0.00475 kg . The molality of a solution of chlorine and water is 0.0362m . This solution contains 3500g of water. How much chlorine, in grams, was used to prepare this solution? ... 15-2 Review and Reinforcement ...

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Chemistry: Chapter 15 Concentration and Nature of Solutions

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2 Answer.Key 2-1 Explore, page 5 ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed 6. natural gas ... 15. chromatography 16. electrolysis 17. distillation 18. A magnet could be used to separate them

2 Answer - Weebly

'16—1A Review and Reinforcement, pages 9-10 6. Answers Will Vary. When the rates Of forward and reverse reactions are equal, equilibrium is established. For the example of ... 15. The forward and reverse reactions are both occurring, but with no net change in concentration. -16—2 Review Reinforcement, pages 15-16

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11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. 1. 2. 5. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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15. Why is water called the universal solvent? ____ ____ 16. How could you determine experimentally whether a given substance is an electrolyte? ... Title: 13-1 Review and Reinforcement Author: NPSD Last modified by: NPSD Created Date: 9/27/2005 1:14:00 PM Company: NPSD

13-1 Review and Reinforcement - North Penn School District

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5-2 Review and Reinforcement . Reading the Periodic Table . On the line at the left, write the letter of the appropriate location of each group of elements on the periodic table below. Some letters will be used more than once. 1. carbon family 8. f-block elements 2. alkaline earth metals 9. noble gases 3. inner transition metals 10. p-block ...

5-2 Review and Reinforcement - MRS. Nairn - HOME

10—2 Review and Reinforcement (continued) 9. 0.0621 mol SS = (Q \$ gK2S 10. 65.8gMg= l. atoms Mg Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NOZ gas.

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