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Chemical Changes in the Blood During Exercise. In previous tutorials ("Hemoglobin and the Heme Group: Metal Complexes in the Blood for Oxygen Transport", "Iron Use and Storage in the Body: Ferritin and Molecular Representations", "Maintaining the Body's Chemistry: Dialysis in the Kidneys") you learned about the daily maintenance required in the blood for normal everyday activities such as ...

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Natural Acidity of Rainwater. Pure water has a pH of 7.0 (neutral); however, natural, unpolluted rainwater actually has a pH of about 5.6 (acidic).[Recall from Experiment 1 that pH is a measure of the hydrogen ion (H +) concentration.]The acidity of rainwater comes from the natural presence of three substances (CO 2, NO, and SO 2) found in the troposphere (the lowest layer of the atmosphere).

#### **Acid Rain - Department of Chemistry**

The term Activation Energy was introduced in 1889 by Svante Arrhenius, a Swedish scientist.. It is defined as the least possible amount of energy (minimum) which is required to start a reaction or the amount of energy available in a chemical system for a reaction to take place.

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