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Use (s) for a solid, (l) for a liquid, (g) for a gas, and (aq) for a substance in aqueous solution (a substance dissolved in water). (s) In many chemical reactions, a catalyst is added to the reaction mixture. A catalyst is a substance that speeds up the reaction but is not used up in the reaction.

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The production of a new substance, a gas, is evidence of a chemical change. Two new substances are produced in this reaction, oxygen gas and liquid water. You could describe this reaction by saying, "Hydrogen peroxide decomposes to form water and oxygen gas." You could also write a word equation.

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SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329) This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical ... 114 Guided Reading and Study Workbook CHAPTER 11, Chemical Reactions(continued) 9.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)

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