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Section 2–3 Carbon Compounds This section explains how the element carbon able to form millions of carbon, or organic, compounds. It describes the four groups of organic compounds found in living things. The Chemistry of Carbon How many valence electrons does each carbon atom have? Each carbon atom has four electrons.

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2.3 Carbon Compounds A. The Chemistry of Carbon Organic any substance that contains carbon except CO and CO.2 Carbon is very important to living things 1. Carbon has 4 Valence electrons Makes it very versatile Will easily bond with H, O, N, P, S..all important for living things

Carbon Compounds Ch 2 - Northern Highlands

Answer: carbon compounds in which all the four valencies of carbon atom are satisfied by forming single covalent bonds are known saturated carbon compounds. For example, ethane: In unsaturated carbon compounds, at least one or more double covalent bond or triple bonds satisfy the four valencies of a carbon atom. For example, ethylene or acetylene:

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Nomenclature and Formula Writing Worksheet Answers

Macromolecules Worksheet. Compounds can be organic or inorganic. Organic - compounds that contain both carbon and hydrogen atoms. Inorganic - compounds that DO NOT contain both carbon and hydrogen. There are . four. classes of organic compounds that are central to life on earth. ... nucleic acids 23. DNA and RNA belong to this category of ...

Macromolecules Worksheet - Manatee School for the Arts

The Chemistry of Carbon Organic chemistry is the study of compounds with bonds between carbon atoms. Carbon atoms have four valence electrons, allowing them to form strong covalent bonds with many other elements, including hydrogen, oxygen, phosphorus, sulfur, and nitrogen. Living organisms are made up of molecules made of carbon and these

The Chemistry of Life - Weebly

Section 2–3 Carbon Compounds (pages 44–48) Key Concept • What are the functions of each group of organic compounds? The Chemistry of Carbon ... SC06_GRSW_CH02 5/23/06 3:13 PM Page 19. 11. What are polysaccharides? They are large macromolecules formed from monosaccharides. 12.

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Section 2-3 Carbon Compounds (pages 44-48) Key Concept • What are the functions of each group

of organic compounds? The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? 2. What gives carbon the ability to form chains that are almost unlimited in length? Macromolecules (page 45) 3.

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