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the law that states that for a fixed amount of gas at a constant pressure, the volume of the gas increases as its temperature increases, and the volume of the gas decreases as its temperature decreases. plasma. the state of matter that does not have a definite shape or volume and whose particles have broken apart. change of state.

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Chapter 3 States of Matter Section 3.3 Phase Changes (pages 84–91) This section explains what happens when a substance changes from one state of matter to another and describes six phase changes. Reading Strategy (page 84) Summarizing As you read, complete the description of energy flow during phase changes in the diagram below. For more ...

Chapter 3 States of Matter Section 3.3 Phase Changes

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