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A Absolute electrode potential In electrochemistry, according to an IUPAC definition, is the electrode potential of a metal measured with respect to a universal reference system (without any additional metal-solution interface). Absolute pressure Is zero-referenced against a perfect vacuum, using an absolute scale, so it is equal to gauge pressure plus atmospheric pressure.

Glossary of engineering - Wikipedia

Fluorescence quenching rate constants, k_q , ranging from 10^6 to $2 \times 10^{10} \text{ M}^{-1} \text{ sec}^{-1}$, of more than 60 typical electron donor-acceptor systems have been measured in de-oxygenated acetonitrile and are shown to be correlated with the free enthalpy change, ΔG° , involved in the actual electron transfer process . in the encounter complex and varying between + 5 and –60 kcal/mole.

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