

## ***16 3 Review And Reinforcement Answer Key***

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16—3 Review and Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperature affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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**16 3 REVIEW AND REINFORCEMENT ANSWERS**

3—3 Review and Reinforcement Modern Atomic Theory Complete the following table. 1. 2. 3. If the statement is true, write "true." If it is false, change the underlined word or words to make the statement true. Write your answer on the line provided. ... I 16. 3 protons and 2 electrons 17 12 protons and 10 electrons

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3-2 Review and Reinforcement . Discovering Atomic Structure . On the line at the left, write the letter of the answer that best completes each statement. 1. The scientist who suggested that the structure of the atom was somehow related to electricity was a. Benjamin Franklin. b. Democritus. c. Michael Faraday. d. John Dalton. 2.

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16 1 Review And Reinforcement Answers.pdf Free Download Here Study Guide and Reinforcement - Answer Key - NovakWiki - home ... The Process of Chemical Reactions Review Skills 16.1 Collision Theory: ... According to Table 16.1, the KP for this reaction is 2.2 ... Key Ideas Answers

**16 1 Review And Reinforcement Answers**

7-2 Review and Reinforcement. Covalent Bonding. ... Answer the following questions as directed. 21. Explain why the molecules SF<sub>4</sub> is an exception to the octet rule.

**2 3 Review And Reinforcement Answer Key**

10—2 Review and Reinforcement Mole Conversions Answer each of the following questions in the space provided. 1. How would you calculate the number of moles present in a given mass of a substance? 2. How would you calculate the number of particles present in a given number of moles of a substance? 23 10 3.

**10-2 Review and Reinforcements**

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Chapter 16 Review Answers ... Section 16.3 . 22. All organic molecules contain the element carbon. 23. Carbon has four available bonding positions. It can form bonds with other . carbon atoms in chains or rings, or with atoms of other elements. 24. The compound is likely a protein because lipids and carbohydrates do not

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3—1 Review and Reinforcement Early Models of the Atom On the line at the left, write the letter of the chemist who proposed each of the ideas listed below. Each letter may be used once, more than once, or not at all. a. Democritus ... 16. 3 protons and 2 electrons 17. 12 protons and 10 electrons

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3-4 Review and Reinforcement . Changes in the Nucleus . Complete each of the sentences below. 1. Nuclear reactions change the composition of an atom's \_ 2. The attractive force that overcomes the electric repulsion between protons is the force. 3. Almost all the atoms you encounter have nuclei. 4. All nuclei with atomic numbers greater than 83 ...

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II—I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. ... 16. Calcium carbonate ( $\text{CaCO}_3$ ) combines with HCl to produce calcium c oride ( $\text{CaCl}_2$ ), water, and carbon dioxide gas. Write the balanced equation for this reaction. How many moles of HCl

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3. 4. What is the molecular shape of carbon tetrachloride ( $\text{CCl}_4$ )? What are the bond angles in this molecule? h 100115 A molecule of hydrogen cyanide contains 1- hydrogen atom, -1 carbon atom, and -1 nitrogen atom. The carbon is bonded to the hydrogen atom and to the nitrogen atom. Predict the bond angle in this molecule. )

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