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16—3 Review.nd Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperatuÈe affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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18. Gas molecules are said to be in	19. The ϵ	collisions betwee	n gas partio	cles are	20	. The
temperature of a gas is a measure of	of the average	of the gas .	particles. 2	21. Gas	particles	exert
on one another 12-1 Review	w and Reinforce	ement				

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10—2 Review and Reinforcement Mole Conversions ... 18. In an experiment, 12.1 g of dry ice (solid C02) were converted into gaseous C02 at STP. How many moles were in the sam le? What was f gaseous C02? 12,1 Coz K C ICE. Ch'3'ter 10 19 . Title: 10-2 Review and Reinforcements ...

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