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3—3 Review and Reinforcement. Modern Atomic Theory. Complete the following table. 1. 2. 3. If the statement is true, write "true." If it is false, change the underlined word or words to make the statement true. Write your answer on the line provided.

Name Date Class 3—3 Review and Reinforcement

11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. 1. 2. 5. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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10-2 Review and Reinforcement (continued) 9. 0.0621 mol SS = (Q \$ gK2S 10. 65.8gMg= I. atoms Mg Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NOZ gas.

10-2 Review and Reinforcements

16—3 Review.nd Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperatuÈe affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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10-1 Review and Reinforcement (continued) Find the formula mass of each of the following compôunds. Show all your work. 11. caC03 12. MgS04 13. NaOH 14. KCI 15. BaF2 o JC 04) 137B Find the molar mass of each of the following compounds. Show all your work. 16. nitric acid (HN03) (D 17. ammonia (NH3) 18. calcium sulfate (CaS04) 19. potassium ...

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