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14.1 Structure of the Atom Atoms are made of three tiny subatomic particles: protons, neutrons, and electrons. The protons and neutrons are grouped together in the nucleus, which is at the center of the atom. The chart below compares electrons,

14.1 Structure of the Atom - North Penn School District

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Chapter 14.1 The Structure of The Atom. Vocabulary. Electric Charge. Elementary Charge. Electron. Nucleus. Proton. Neutron. Atomic Number. Isotopes. Mass number. Stable. Radioactive. Electric Charge. It was once believed that atoms were the smallest particles of matter. We now know that matter consists of three basic particles.

Chapter 14.1 The Structure of The Atom

So we know that all matter is made up of atoms, but what is an atom made out of? Chemists describe the structure of atoms using models. This section will cover the Bohr model, photoelectric effect, absorption and emission spectra, quantum numbers, and electron configurations.

Electronic structure of atoms | Chemistry | Science | Khan ...

Structure of a beryllium atom: four protons, four neutrons and four electrons. Credit: general-fmv Shutterstock Atoms are the basic units of matter and the defining structure of elements.

What is an Atom? | Parts of an Atom - Live Science

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STRUCTURE OF THE ATOM (CH_20)

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STRUCTURE OF THE ATOM. There are 92 elements that occur naturally. The elements hydrogen, carbon, nitrogen and oxygen are the elements that make up most living organisms. Some other elements found in living organisms are: magnesium, calcium, phosphorus, sodium, potassium.

Structure of the Atom (grades 6-8) - New York University

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