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In chemistry, an ionic compound is a chemical compound composed of ions held together by electrostatic forces termed ionic bonding. The compound is neutral overall, but consists of positively charged ions called cations and negatively charged ions called anions. These can be simple ions such as the sodium (Na^+) and chloride (Cl^-) in sodium chloride, or polyatomic species such as the ammonium ...

Ionic compound - Wikipedia

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For a solution, Raoult's law relates the relative concentrations of the components in solution with their relative vapor pressures above the solution.

Definition of Raoult's Law | Chegg.com

Nickel(II) chloride (or just nickel chloride), is the chemical compound NiCl_2 . The anhydrous salt is yellow, but the more familiar hydrate $\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$ is green. Nickel(II) chloride, in various forms, is the most important source of nickel for chemical synthesis. The nickel chlorides are deliquescent, absorbing moisture from the air to form a solution. ...

Nickel(II) chloride - Wikipedia

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pH Buffers in the Blood - chemistry.wustl.edu

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