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End point the point in a titration at which the indicator changes color. Endothermic refers to a reaction in which energy (as heat) flows into the system. Energy the capacity to do work or to cause the flow of heat. Enthalpy at constant pressure, the change in enthalpy equals the energy flow as heat.

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Steve Zumdahl is the author of market-leading textbooks in introductory chemistry, general chemistry, honors-level chemistry, and high school chemistry. Recently retired from his long-time

position as Director of Undergraduate Programs at the University of Illinois, he has received numerous awards for his contributions to chemical education.

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a comprehensive system of units understood by everyone in the world. volume. amount of 3-d space occupied by a substance. liter. L. milliliter. mL. 1 L = 1000 mL. how many liters in a milliliter. mass. quantity of matter present in an object. gram. fundamental SI unit of mass. significant figures.

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(Fig. 1.2). One of the main challenges of chemistry is to understand the connection between the macroscopic world that we experience and the microscopic world of atoms and molecules. To truly understand chemistry, you must learn to think on the atomic level. We will spend much time in this text helping you learn to do that.

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