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answer. 157.5 g 106.5 g 51 g Yes. Mass of reactants equals mass of products. Section 3.2 Assessment page 79 10. Classify each example as a physical change or a chemical change. a. crushing an aluminum can physical b. recycling used aluminum cans to make new aluminum cans physical c. aluminum combining with oxygen to form aluminum oxide chemical 11.

#### Matter—Properties and ChangesMatter—Properties and Changes

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Covalent BondingCovalent Bonding Solutions Manual Chemistry: Matter and Change  $\bullet$  Chapter 8 121 Section 8.1 The Covalent Bond pages 240–247 Practice Problems page 244 Draw the Lewis structure for each molecule. 1. PH 3 H HH — H H P respectively, for single, double, and triple P — - 2. H 2 S H H H — H S S — 3. HCl H Cl H — Cl 4 ...

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# **Chapter 1: Introduction to Chemistry**

These sub outcomes are used in the Assessment Schedule Template for Achievement Standard 3.1. Rather than have a whole series of check boxes corresponding to each of these sub outcomes it is better to make an overall decision on each of the criteria as indicated by the single check box at each level of achievement for each of the criteria.

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