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CBSE Class 12 Chemistry Quick Revision Notes Chapter 2 Solutions. The difference in boiling points of solution T_b and pure solvent T° is called elevation in boiling point. Solutions: Solutions are the homogeneous mixtures of two or more than two components. Binary solution: A solution having two components is called a binary solution.

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Class 12 Chemistry Chapter 2 Solutions Notes- PDF DOWNLOAD. 2.3.1 Solubility of a Solid in a Liquid 2.3.2 Solubility of a Gas in a Liquid 2.4 Vapour Pressure of Liquid Solutions 2.4.1 Vapour Pressure of Liquid-Liquid Solutions 2.4.2 Raoult's Law as a special case of Henry's Law 2.4.3 Vapour Pressure of Solutions of Solids in Liquids 2.5 Ideal and Nonideal Solutions.

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Solutions – CBSE Notes for Class 12 Chemistry. A solution may be classified as solid, liquid or a gaseous solution. Solubility is defined as the amount of solute in a saturated solution per 100g of a solvent. The effect of pressure on the solubility of a gas in a liquid is governed by Henry's Law.

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CBSE Class 12 Chemistry Notes: Solutions – Basics. Saturated solution is a solution which remains in contact with excess of solute. The amount of solute dissolved per 100 g of solvent in a saturated solution at a specific temperature represents the solubility of the solute. For exothermic substances such as KOH, CaO, Ca(OH)_2 , M_2CO_3 ,...

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A solution consists of a solvent and a solute. It is defined as a homogeneous mixture of two or more than two substances. Download notes for Solutions Class 12 Chemistry PDF available on BYJU'S.

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