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Blue-black SnO can be produced by heating the tin(II) oxide hydrate, $\text{SnO} \cdot x\text{H}_2\text{O}$ ($x < 1$) precipitated when a tin(II) salt is reacted with an alkali hydroxide such as NaOH. Metastable, red SnO can be prepared by gentle heating of the precipitate produced by the action of aqueous ammonia on a tin(II) salt.

Tin(II) oxide - Wikipedia

Cadmium sulfide is the inorganic compound with the formula CdS. Cadmium sulfide is a yellow solid. It occurs in nature with two different crystal structures as the rare minerals greenockite and hawleyite, but is more prevalent as an impurity substituent in the similarly structured zinc ores sphalerite and wurtzite, which are the major economic sources of cadmium.

Cadmium sulfide - Wikipedia

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