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Endothermic vs. Exothermic vel 5-1: Endothermic vs. Exothermic Invarious chemical processes such as reactions and the dissolving of salts, heat is either absorbed or given off. We call these events either an endothermic (heat in) or exothermic (heat out) process. It is usual to detect these heat events by measuring the temperature change associated with the process ...

vel 5-1: Endothermic vs. Exothermic - Reeths-Puffer

In your lab notebook decide what data you will need to collect in order to answer the research question. Develop your procedures and decide how you will collect your data. Perform the virtual experiment and analyze your results. Develop a scientific argument (claim, evidence, reasoning) that answers the research question.

Virtual Lab Spectroscopy - Mr. Palermo's Flipped Chemistry ...

Analysis. When measuring volume of air in the flask at the first temperature, a volume of 250 mL was recorded, known as V1. The temperature of the air in the flask in boiling water was recorded as 99°C, known as T2.

Charles Law: Volume & Temperature Lab Answers ...

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