15 4 Practice Problems Answers Chemistry

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1/4

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2/4

15 4 Practice Problems Answers

Name Date Class 15—4 Practice Problems 1. 2. 3. 4. 5. What is the boiling point elevation when 11.4 g of ammonia (NH3) is dissolved in 200. g of water?

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Practice Problems: Solutions (Answer Key) a. 1.00 L of 0.125 M K 2 SO 4 21.8 g K 2 SO 4 b. 375 mL of 0.015 M NaF 0.24 g NaF c. 500 mL of 0.350 M C 6 H 12 O 6 31.5 g C 6 H 12 O 6 Calculate the molarity of each of the following solutions: a. 12.4 g KCl in 289.2 mL solution 0.576 M KCl b. 16.4 g CaCl 2 in 0.614 L solution 0.241 M CaCl 2 c.

Practice Problems: Solutions (Answer Key)

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15-4 Practice Problems 1) What is the boiling point elevation when 11.4 g of ammonia (NH 3) is dissolved in 200g of water? K b for water is 0.52 C o /m. 2) How many grams of benzoic acid (C 7 H 6 O 2) must be dissolved in 78.1g of ethanol to raise the boiling point by 4.00 C o? K b for ethanol is 1.2C o /m.

15-4prac - 15-4 Practice Problems 1 What is the boiling ...

238 Unit Practice Problem Answers Additional Practice Problem Answers 14 1) tenths 2) hundredths 3) thousandths 1) tenths 2) hundredths 3) thousandths 4) ten thousandths 5) hundred thousandths 4) ten thousandths 5) hundred thousandths 6) millionths 7) four-tenths 8) forty-three hundredths 6) millionths 7) six-tenths 8) sixty-three hundredths 9) four ...

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Name _ 15-4 Practice Problems (continued) 11. How many grams of silver would have to be dissolved in 1120 g of ethanol to lower the freezing point by 0.25 CO?

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4/4