# 5 3 Review Reinforcement Periodic Trends Answers

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element's atoms is called the atomic number.Name Date Class 3—3 Review and Reinforcement 5 3 review reinforcement periodic trends answers.pdf FREE PDF DOWNLOAD Review of Periodic Trends - ScienceGeek.net Homepage ... Study Guide and Reinforcement 3 ANSWER KEY 7. opposes the

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5—3 Review and Reinforcement Periodic Trends Use the periodic table and your knowledge of periodic trends to answer the following questions, ... Review and Reinforcement (continued) Class Write the charge that each of the f0i/owtng atoms will acquire when It has a complete set of valence electrons, 23. O

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3—3 Review and Reinforcement (continued) Use the periodic table to determine how manyprotons,
neutrons, and electrons arepresent in each of the following atoms. Write your answers in the spaces
provided. Atom Protons Neutrons Electrons 12 iodine 125 13. potassium-39 / / 14. iron-56
3 Write the chemical symbol for each of the ions

## Name Date Class 3—3 Review and Reinforcement - Weebly

5-1 Review and Reinforcement (continued) Use the skills you developed in Section . 5-1 . to answer the following questions. 9. Find the atomic masses for lithium, potassium, and sodium on your periodic table and write them below. 10. Calculate the average atomic mass of lithium and potassium. Compare this calculated value

## Class 5-1 Review and Reinforcement - MRS. Nairn

2 Answer.Key 2-1 Explore, page 5 The original source of energy for this activity ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed ... periodic table 14. compound 15. carbon dioxide 16. pure, properties 17. element 18. compound

## 2 Answer - Weebly

3-1 review and reinforcement. STUDY. PLAY. Matter is neither created or destroyed in chemical reactions. Lavoisier. A given compound always has the same relative numbers and kind of atoms. Dalton. All atoms of a given element are identical, but they differ from those of any other element.

## 3-1 review and reinforcement Flashcards | Quizlet

\_\_\_\_ Chemistry Chapter ... 9-3 Review and Reinforcement (Continued) Answer each of the following questions in the space provided. 16-2 Review and Reinforcement pages 15-16

## **Review And Reinforcement Chemistry Answer**

CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in

372EEADF93A20E2BA6352A17FCD25225

the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an electron from a neutral ...

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