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ST AT 14. small compared to the total volume of the gas. Air will rush into a sealed container when the container is opened. Gas flows from a region of lower pressure to a region of higher pressure. Adding air to an object will cause the object to inflate.

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THE BEHAVIOR OF GASES SECTION 14.1 PROPERTIES OF GASES (pages 413-417) This section uses kinetic theory to explain the properties of gases. This section also explains how gas pressure is affected by the amount of gas, its volume, and its temperature. Compressibility (pages 413-414) 1. Look at Figure 14.1 on page 413.

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SECTION 14.1 PROPERTIES OF GASES(pages 413-417)

Upon closer study, they began observing consistent properties that defined gases. The single distinction that initially baffled scientists -- that of gas particles having more space to move freely than particles of solids or liquids -- informs each of the properties that all gases have in common.

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