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### 19 ACIDS AND BASES REVIEWSHEET ANSWERS

ACIDS AND BASES REVIEW SHEET. Identify each acid, base, conjugate acid, and conjugate base. 1.  $\text{HNO}_2 + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{NO}_2^-$

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accept a positive hydrogen ion. Thus, an acid-base reaction in the reverse direction can occur. The acid  $\text{H}_3\text{O}^+$  can react with the base  $\text{X}^-$  to form water and  $\text{HX}$  and the following equilibrium is established. 598 Chapter 19 Acids and Bases  $\text{HX(aq)} + \text{H}_2\text{O(l)} \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{X}^-(\text{aq})$  Acid Base Conjugate acid Conjugate base

### Chapter 19: Acids and Bases - cchs.ccusd.org

Acids, Bases, and Salts 587 19.1 FOCUS Objectives 19.1.1 Define the properties of acids and bases. 19.1.2 Compare and contrast acids and bases as defined by the theories of Arrhenius, Brønsted-Lowry, and Lewis. Guide for Reading Build Vocabulary Word Parts Explain that the prefixes mono-, di-, and tri- have Greek origins,

### 19.1 Acid-Base Theories 19

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Acids, Bases, and Solutions ANSWER KEY Acids, Bases, and Solutions Describing Acids and Bases Review and Reinforce 1. Sour 2. Bitter 3. Corrosive to magnesium, zinc, and iron; eats them away and produces bubbles of hydrogen gas 4. Doesn't react with metals 5. Produces carbon dioxide 6. Doesn't react with carbonates 7. Red 8. Blue 9.

### **Acids, Bases, and Solutions ANSWER KEY - Lab35**

Title: PowerPoint Presentation Author: Debbie Munson Created Date: 5/12/2014 8:17:21 AM

### **19.1 Acid-Base Theories> - Quia**

Lewis base. 6. Acids have a tart or sour taste and cause indicators to change color. Acids react with compounds containing hydroxide ions to produce a salt and water. 7. Aqueous solutions of bases taste bitter and feel slippery. They react with acids to produce a salt and water. Bases cause indicators to change colors. -10M 10.  $\text{pH} = -\log[\text{H}^+]$

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Acids and Bases Cheat Sheet. Acids and Bases. Matters can be classified in many ways. Acids and bases are another way of classification of matters. Most of the reactions taking place in water solutions are in acid or base mediums. Definitions of acids and bases are given by Arrhenius and Bronsted -Lowery.

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According to this Odel, in an acid-base reaction, each acid has a (17) , and each base has a (18) Study Guide for Content Mastery Chemistry: Matter and Change Chapter 19 . Name 19 Date Class Section Strengths of Acids and Bases In your textbook, read about strengths of acids.

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