

111 Describing Chemical Reactions Section Review Answer Key

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CHAPTER 111 DESCRIBING CHEMICAL REACTIONS

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SECTION 111 DESCRIBING CHEMICAL REACTIONS ANSWER KEY

reactions Section 11.1 Describing Chemical Reactions 323 withChemASAP Chemical Equations Word equations adequately describe chemical reactions, but they are cumbersome. It's easier to use the formulas for the reactants and products to write chemical equations.

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In a chemical reaction, one or more reactants change into one or more products. Figure 11.1a shows the ingredients for making leavened bread— flour, salt, yeast, and water. Chemical reactions take place when the ingredients are mixed together and heated in the oven. Figure 11.1b shows the product—a loaf of bread.

11.1 Describing Chemical Reactions 11

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a representation of a chemical reaction; the formulas of the reactants (on the left) are connected by an arrow with the formulas of the products (on the right). Write the formulas of the reactants to the left of the yields sign (arrow) and the formulas of the products to the right. first write the skeleton equation.

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Chapter 11 - Chemical Reactions - 11.1 Describing Chemical Reactions - 11.1 Lesson Check - Page 354: 7 Answer Write the chemical formulas of the reactants on the left of the yield sign and the chemical formulas of the products on the right of the yield sign.

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The production of a new substance, a gas, is evidence of a chemical change. Two new substances are produced in this reaction, oxygen gas and liquid water. You could describe this reaction by saying, "Hydrogen peroxide decomposes to form water and oxygen gas." You could also write a word equation.

11.1 Describing Chemical Reactions 11

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