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Physical Chemistry Laboratory I CHEM 445 Experiment 2 Partial Molar Volume (Revised, 08/14/06) ... in 10.00 mL H2O is not identical to a solution of 1.000 g{X} in 10.00 mL ... The standard treatment of partial molar quantities considers the differential change in that property with the addition of more (mol) of material. ...

Physical Chemistry Laboratory - University of Delaware

10.1 Measuring Matter MAIN Idea Chemists use the mole to count atoms, molecules, ions, and formula units. 10.2 Mass and the Mole MAIN Idea A mole always contains the same number of particles; however, moles of different substances have different masses. 10.3 Moles of Compounds MAIN Idea The molar mass of a compound can be calculated from

Chapter 10: The Mole - Middlesex County Vocational and ...

MOLAR AND PARTIAL MOLAR VOLUMES IN AQUEOUS SOLUTIONS (10/21/03) QUANTITATIVE TECHNIQUES Use of an Analytical Balance ... PRE LABORATORY EXERCISE Use a spreadsheet to do the calculations in the next 5 parts. Include a sample of each calculation in your pre lab report. 1. Find the literature values of densities of solute/water solutions for ...

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