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3-1 review and reinforcement. STUDY. PLAY. Matter is neither created or destroyed in chemical reactions. Lavoisier. A given compound always has the same relative numbers and kind of atoms. Dalton. All atoms of a given element are identical, but they differ from those of any other element.

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15-1 and 15-2 Review and Reinforcement Learn with flashcards, games, and more — for free. Search. Create. ... 13 terms. powersm15 PLUS. Chemistry: Chapter 15 Concentration and Nature of Solutions. 15-1 and 15-2 Review and Reinforcement. STUDY. PLAY. What is the concentration of a solution expressed as the number of moles of solute dissolved ...

Chemistry: Chapter 15 Concentration and Nature of Solutions

11—1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. consdruption of matter coefficients subscripts mass

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10—2 Review and Reinforcement Mole Conversions Answer each of the following questions in the space provided. I. How would you calculate the number of moles resent in a given mass of a substance? 2. How would you calculate the number of particles present in a given number of moles of a substance? 23 10 3.

10-2 Review and Reinforcements

2 Answer.Key 2-1 Explore, page 5 The original source of energy for this activity would be the stored chemical energy in the dry cell battery. The chemical energy in the ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed 6. natural gas 7. kinetic . 8. f . 9. c 10. b 11. a . 12. e .

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24-1 Review and Reinforcement (continued) 13. There are 3.29 g of iodine-126 remaining in a sample originally containing 26.3 g of iodine-126. The half-life of iodine-126 is 13 days. How old is the sample? 14. A sample that originally contained 2.5 g of rubidium-87 now contains 1.25 g. The half-life of rubidium-87 is 6 x 1010 years.

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