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For a solution, Raoult's law relates the relative concentrations of the components in solution with their relative vapor pressures above the solution.

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In order for NaCl to be soluble, the Na + and Cl-ions must break free from the crystal-lattice structure of the solid. When the ions are in solution, they are surrounded by water molecules, and the ions are said to be solvated, or dissolved in an aqueous solution, denoted (aq). Hence, the process of dissolving a NaCl crystal can be described by the following chemical equation (Equation 1):

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In more detail, human thermodynamics is the study of the energy and entropy aspects of the work cycles involved in human life, namely those existent between heat, spontaneity, irreversibility and the laws defining therein. In short, human thermodynamics is the study of heat and its relation to the motion and changes in the equilibriums of human bodies.

Human Thermodynamics - the science of energy transformations

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