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Page 2 of 3. Modeling Chemistry 2 U2 ws 3 v2.1 5. Sam's bike tire contains 15 units of air particles and has a volume of 160mL. Under these conditions the pressure reads 13 psi. The tire develops a leak. Now it contains 10 ... Displaying Unit 2 Worksheet 3 PVTn Problems.doc.

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Unit 2 Worksheet 3 - PVTn Problems On each of the problems below, start with the given P, V, T, or n; then make a decision as to how a change in P, V, T, or n will affect the starting quantity, and then multiply

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And by problem #3, they are throwing the STP in there. I would have liked to have built the students' confidence up with some more straight forward problems before I start switching units or worrying about STP. If I have such low-performing students next year, I think I will add several easier problems at the beginning.

Unit 2 Reflections: Worksheet 3 PVTn Problems - Blogger

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Date Pd Unit 2 Worksheet 3 PVTn Problems - Buckeye Valley

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pvt n problems

Unit 2, Worksheet 4— Pressure Conversions and Manometer Problems 1.00 atm = 760 torr = 760 mm Hg = 14.7 psi Show all work including units and the appropriate number of significant figures.

1. On a sunny day the atmospheric pressure is 14.9 psi. Convert this pressure to atm, mmHg and torr. 2.

Unit 2, Worksheet 1— Observations of the Phases of Matter

Modeling Chemistry 1 U2 ws 3 v2.1 Name Date Pd Unit 2 Worksheet 3 - PVTn Problems On each of the problems below, start with the given P, V, T, or n; then make a decision as to how a change in P, V, T, or n will affect the starting quantity, and then multiply by the appropriate factor.

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