

## *Acids And Bases Chapter 19 Answers*

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accept a positive hydrogen ion. Thus, an acid-base reaction in the reverse direction can occur. The acid  $\text{H}_3\text{O}^+$  can react with the base  $\text{X}^-$  to form water and  $\text{HX}$  and the following equilibrium is established. 598 Chapter 19 Acids and Bases  $\text{HX(aq)} + \text{H}_2\text{O(l)} \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{X}^-(\text{aq})$  Acid Base Conjugate acid Conjugate base

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Chapter 19 "Acids, Bases, and Salts ... Figure 19.22, page 615. Neutralization. The solution of known concentration is called the standard solution. added by using a buret. Continue adding until the indicator . changes color.

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