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5 3 Review And Reinforcement Chemistry Answers

3—3 Review and Reinforcement Modern Atomic Theory Complete the following table. 1. 2. 3. ... 5. In an individual, neutral atom, the number of protons always equals the number of electrons. T 6. When an atom loses or gains one or more electrons, it is called an. iZ 7. The average mass of an element's atoms is called the atomic number.

Name Date Class 3—3 Review and Reinforcement

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16—3 Review.nd Reinforcement, pages 20-2J 5. endothermic 6. true 7. increases 8. true 9. The Concentrations of reactants and products, pressure, and temperatuËe affect the equilibrium 10. Le Chatelier's principle states that if a change is imposed on a system at equilibrium, the equilibrium position will shift in the

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12-5 Review and Reinforcement - Course Hero

10—2 Review and Reinforcement Mole Conversions Answer each of the following questions in the space provided. 1. How would you calculate the number of moles resent in a given mass of a substance? 2. How would you calculate the number of particles present in a given number of moles of a substance? 23 10 3.

10-2 Review and Reinforcements

5 3 review and reinforcement continued answers

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6—2 Review and Reinforcement (continued) Identify each of the following elements as a meta/ (M), nonmetal (NM), or semimetal (SM). 19. sodium —SM— silicon 21. neon 22. 23. nitrogen Write the family names that have been given to each of the following groups. 24. Group IA alkan- 25. Group 2A 26. Group 7 A 27. Group 8A hvohlt

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II—I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. ... 5 1420 16. Calcium carbonate (CaCO_3) combines with HCl to produce calcium chloride (CaCl_2), water, and carbon dioxide gas. Write the balanced equation for this reaction. How many moles of HCl

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4-3 Review and Reinforcement. Another Look at the Atom. Complete the following sentences. Every element has a uniquely characteristic atomic ____ spectrum. De Broglie referred to the wavelike behavior of particles as _____. Heisenberg's uncertainty principle states that the position and the _____ of a moving object cannot simultaneously be ...

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REVIEW and REINFORCEMENT Chemical Families Section 6-3.... KEY CONCEPTS. - ~ Elements within the same family of the periodic table have similar properties because they have the same number of valence electrons. M Vocabulary Skills: Applying Definitions On the periodic table below, label each of the following: acidoid series alkaline earth ...

REVIEW and REINFORCEMENT Section 6-3 - npsd.k12.nj.us

10-1 Review and Reinforcement (continued) Find the formula mass of each of the following compounds. Show all your work. 11. CaCO_3 12. MgSO_4 13. NaOH 14. KCl 15. BaF_2 o JC 04) 137B Find the molar mass of each of the following compounds. Show all your work. ... 3/7/2012 7:42:08 PM ...

Name Date Class 10-1 Review and Reinforcement (continued ...

2-5 Review and Reinforcement (continued) Answer each of the following questions in the space provided. 18. How could you separate a mixture of iron filings and aluminum filings? What property of these metals would allow such a separation? 19. Could distillation be used to separate air into oxygen, nitrogen, carbon dioxide, argon, and so forth ...

2-5 Review and Reinforcement - Stone Science

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