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10—2 Review and Reinforcement (continued) 9.  $0.0621 \text{ mol SS} = (Q \text{ } \$ \text{ gK2S } 10. 65.8\text{gMg} = \text{l. atoms Mg}$  Solve each of the following problems as directed. Show all your work. 11. Calculate the number Of atoms in 0.40 mol of sulfur- 12. Calculate the number of atoms in 2.30 mol of silver. G.oux 13. You have a tank of NO<sub>2</sub> gas.

**10-2 Review and Reinforcements - Front Page**

12—2 Review and Reinforcement (continued) 10. 20.0 g of potassium react with water to produce potassium hydroxide and hydrogen gas. How many liters would the hydrogen gas occupy at STP? 573 L u 301, 1 2 k 11. If 30.2 g of aluminum react with HCl to produce alurninum chloride and hydrogen gas, how

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View Notes - 12-5 Review and Reinforcement from CHEMISTRY 1A at Evergreen Valley College. 12-5 Review and Reinforcement Name \_ How Gases Work Date \_Period \_ Molar mass CFCs ultraviolet

**12-5 Review and Reinforcement - Course Hero**

2 Answer.Key 2-1 Explore, page 5 ... 2-1 Review and Reinforcement, pages 8-9 . 1. energy 2. radiant 3. chemical potential 4. joule 5. created, destroyed ... 12. filtration 13. crystallization 14. distillation 15. chromatography 16. electrolysis 17. distillation

**2 Answer - Weebly**

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\_\_\_\_\_ 12. The volume of a gas is equal to the volume of its container. Indicate the relationship between the pairs of gas variables below by writing "increase," "decrease," or "not be affected" in

the space provided. Assume that all other variables remain constant. ... 12-1 Review and Reinforcement ...

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Indicate whether each of the€ 12 4 Review And Reinforcement Answer Key : free ebook 4-1 Review and Reinforce - Monmouth Telecom User Pages ... 2-4 Review and Reinforcement. ... 12. Elements are abbreviated with \_\_\_\_\_, which ... Answer each of the following questions. 1 / 2.

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### Lesson Plan Overview for Writing & Grammar 12, 2nd ed.

4—2 Review and Reinforcement (continued) 12. 13. Could you use a highly intense beam of red light to eject electrons from the surface of a metal? Explain your answer. Give two examples of photoelectric cells and explain how they work. On the line at the left, write the letter of the scientist who made the contribution to the quantum

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