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(a)  $150\text{cm}^3$  of sodium trioxosulphur sulphate(vi) solution in a corked flask or bottle, labelled "An". These should all be the same containing 6.0g of  $\text{Na}_2\text{S}_2\text{O}_3$  per  $\text{dm}^3$  of solution. (b)  $150\text{cm}^3$  of iodine in potassium iodide solution in a corked flask or bottle labelled "Bn".

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(VII)solution in a corked flask or bottled labelled 'A'n. These should all be the same containing 7.7g of  $\text{KMnO}_4$  per  $\text{dm}^3$  of solution. (b)  $150\text{cm}^3$  of iron(II)chloride solution in a corked flask or bottle labelled 'Nn'. These should all be the same containing 19.1g of  $\text{FeCl}_2$  per  $\text{dm}^3$  of solution. NB-  $\text{FeCl}_2$  solution should be freshly prepared.

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