

## *Worksheet 3 Periodic Trends Answer Key*

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### **Worksheet 3 Periodic Trends Answer**

Worksheet 3-3 Periodic Trends Name \_\_\_\_\_ Period \_\_\_\_ 1. Discuss the importance of Mendeleev's periodic law. e 2. Identify each element as a metal, metalloids, or nonmetal. a) fluorine b) germanium c) zinc d) phosphorous e) lithium 3. Give two examples of elements for each category. a) noble gases b) halogens c) alkali metals d) alkaline earth metals 4.

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Periodic(Trends(Worksheet(((Answer Key 1. Circle)the)element)withthe)largest)atomic(radius)andput)a)square)aroundthe)element)withthe)smallest) atomic(radius:))) Cu) K) Ni) Br))) Largest - K Smallest - Br) Explainwhy)you)madethesechoices.) Atomic radius decreases as you go left to right across a period. Potassium is in the far

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Answers for Comparing Atomic Sizes Here are answers for the questions above. a. Li, C, F All are in the same period and thus have the same number of energy levels. Therefore, the important factor is the nuclear charge. Li is the largest because it has the smallest nuclear charge and pulls the ... Microsoft Word - Periodic Trends Worksheet Author:

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Name \_\_\_\_\_ Worksheet: Periodic Trends Period \_\_\_\_ 1. Which statement best describes Group 2 elements as they are considered in order from top to bottom of the Periodic Table? (A) The number of principal energy levels increases, and the number of valence electrons increases.

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and cesium. Explain why there is a spike on the graph at element numbers 3, 11, 19, 37, and 55. 8. Bromine ( $z = 35$ ) forms a diatomic molecule. Sketch a picture of this molecule and determine the distance in picometers between the two bromine nuclei. Periodic Trends - Atomic Radius Name \_\_\_\_ Chem Worksheet 6-3 eriod 3 3 1 19 37 55

**Periodic Trends - Atomic Radius Name Chem Worksheet 6-3**

ANSWER KEY Trends on the Periodic Table Part 1: Identifying chemical properties/characteristics of the elements For each of the following elements, draw a caricature or cartoon above the description to show its "chemical personality" based on the description provided. (Note: See the Neon example for ideas.) a "magnetic" personality.

**ANSWER KEY Trends on the Periodic Table - rocklin.k12.ca.us**

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