

Applied Physics

Abdul Rafel Problem 1-1 CSC-205-104 Consider a copper penny that Coulains both positive and Negative charges, each of magnifule 1.83 x 10°C. Sup. pose that these charges could be concentrated into two Seperate bundles, held I am apart which attractive force would ait on each bundle?

Data: Copper penny Positive Charge, $94 = 1.83 \times 10^6 \text{ C}$ Copper penny hegative Charge, $92 = -1.83 \times 10^6 \text{ C}$ Separation b/w Charges, t = 100 m

Required: F = ??

 $F = \frac{1}{4\pi \epsilon_0} \left(\frac{9192}{+^2} \right) = K \frac{9192}{+^2}$

Using Colombis law formula, colombis Constant 1/428 = 8.99 × 109 N-m²/c²

Politing values in Cormula, $F = \frac{18.99 \times 10^9 \, \text{N} \cdot \text{m}^2/c^2}{(1.83 \times 10^6 \, \text{c})(-1.83 \times 10^6 \, \text{c})}$ (100 m)²

F=-3.01 x 1018 N

Result: The attractive force that acts on each bundle is $F = -3.01 \times 10^{18} \, \text{N}$



Rollen 1-2

The average distance + between the electron and the Proton an atom is 7.4 x 10" m. what is the magnitude of the average electrostatic for that acts between those two Particles?

Oata - Charge on elettron, ve = 1.60×10" C Charge on Booton, vp = 1.60×10" C Oistance, r = 7.4×10" m

Required 1- Fa??

Formula: F= 1 (9192) = K. 9192
+2

F- 1 (9192) = K. 9192
+2

Solution:Continue (orstant 1/4 No = 8.99 x 10° N. n. /c²
Putting values in formula,

 $F = (8.99 \times 10^{9})(1.60 \times 10^{-19})(1.60 \times 10^{-19})$ $(7.4 \times 10^{-11} \text{ m})^{2}$ $(7.4 \times 10^{-11} \text{ m})^{2}$

F= 4.20×10-8N

Kesulli-The Electrostatic Force acts 6/w to Particles.

F= 4.20×10-8N



Roblem 1-3

The nucleus of an atom has a radius of about 6x10 is and contains 26 protons what repulsive electrostatic force at blu two protons in Such a revelous that are Separated by a distance of one hadins?

Dala: Separation 1=6×10"5 M

Chang on proton, PP= 1.60×10-19 C

Requiered:- Repulsive Force = 3?

Formula: F= 1 (9492) = K 9192 - 418 (+2) +2

Solution .-Colombis Constant 1/4780 = 8.99×10 N.m²/c²
Putting Values in formulla,

F= (8.99×10 NM/C2) (1.60×10-19c) (2.60×10'9c) (6x10-15m)2

F= 6.39N

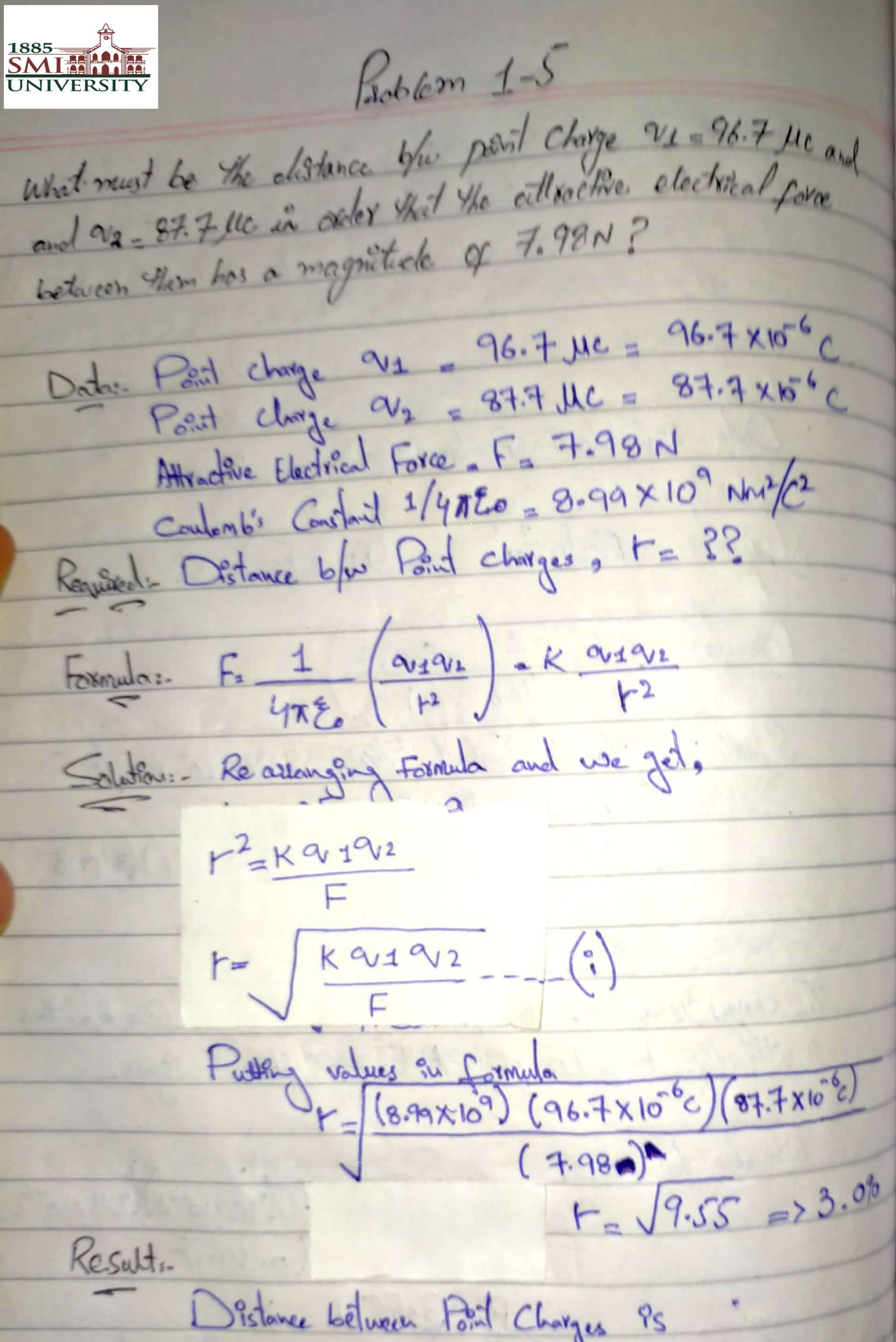
The Repulsive Force b/w Protons

F= 6.39N/

12066m 1-4 Figure. Shows three Charged particles, held in placed by forces not Shown what electrostatic force Dwing to the other charges, all on my? (Find the value of F12 4 F13)? Take or = 4.2 MC, 92= 8.4 MC, 93= 3.4 MC, 12= 19 cm, 113= 19 cm Data available en Problem question. Required: Electrostatic Foce (F124 F13) = 3? Formula: F = 1 (219/2)
476 (+2) Schutienie Colombis Constant 1/4 1/2 = 8.99 x 109 NM/c2

Putting value in Formula,

F12 = (8.99 x 10° Nm/c²) (4.2 x 10° c) (8.4 x 10° c) (0.19m)2 F-12 = 8.78N The Charges VI and V2 have opposite Signs So that the force b/w them is attractive. Herce FIZ points to the right shown in figure. We also have F13. F13 = (8.99×10° NAI/c)(4.2×10°C)(3.4×10°C)
(0.19m) Results Electrostatic force is away from other charges. F12 = 8.78N F13 = 3.55N



Distance bétween Point Charges Ps

r= 3.090m

Fa 3.000



Roblem 1-6

A Point charge of +7.12 × 106 C is 13.4 cm distant from a Second point charge of -2.48 × 10-6 C. Calculate the magneture of the force on each charge.

Data: Pont Charge $a_1 = +7.12 \times 10^6 C$ Point Charge $a_2 = -2.48 \times 10^6 C$ Separation, r= 13.4 cm = 13.8 x10 m

Colombis Contant 1/4780 = 8.99 x109 NN1/c2

Requiradi F= ??

Solutions. Putting values in formulas

F= (899×10° NM/c²) (+7.12×10°6) (-2.48×10°2)

F= -8.840N

Resulti- Mangritude of the force of each charge is

F=-8.840N/

F=-8.840N



12066m 1-7 A Point charge of +4.00× 106 C = 13.0 cm distant from a Second & Charge of 6.50× 106 C Calculate the magnitude of the force on each charge.

Data. Point charge $91 = 44.00 \times 10^{-6} \text{ C}$ Point charge $91 = 6.50 \times 10^{-6} \text{ C}$ Seprention, r= 13.0 = 13.0 × 10 m Coulomb's Constant 1/4720 = 8.99×109 m/c2

Requisedin

F= 83

F= 1 (9192) = Ka 919/2 4720 (+2) = Ka 919/2

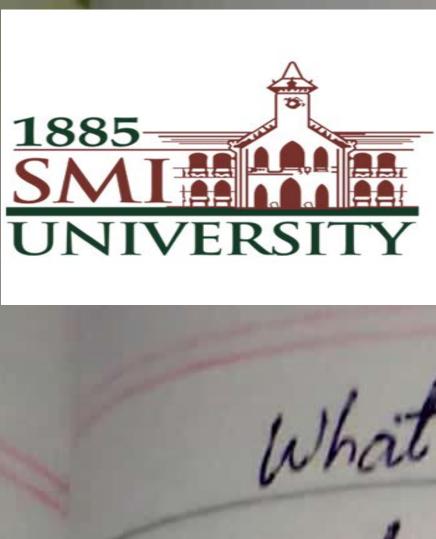
Solution: Futting Values in formular,

F=(8.99×109Nm²(c) (44.00×10°C) (6.50×10°C) $(13.0 \times 10^{-2} \text{ m})^2$

F= 13.83 N

Result: Magnetude of the force of each charge is

F= 13.83N



1006km 1-8

what must be the distance blu point charge $\alpha_1 = 23.0 \mu e$ and $\alpha_2 = 25.0 \mu c$ for the electrostatic force blu them to have a magnitude of 7.50 N?

Data: Point charge $91 = 23.0 \,\mu C = 23.0 \times 10^6 \, c$ Point Charge $92 = 25. \,\mu C = 25.0 \times 10^6 \, c$ Eleiborable force $F = 7.50 \, N$ Coulombs Constant 1/4220 = $K = 8.99 \times 10^9 \, Nm^2/c^2$ Required: Distance be a Point Charges of $F = 3.99 \times 10^9 \, Nm^2/c^2$

F= = (919/2) = K919/2 47/60 (+2) = +2 tomula:-

F2 K9192 Kensing ong

= (8.99×109) (23.0×10-6) (25.0×166)

Result: Distance b/w Point Charge is

r= 0.830 m