

24005

B.Tech. 2nd Semester F-Scheme Examination,

May-2018

ENGINEERING CHEMISTRY

Paper-CH-101-F

(Common for All Branches)

Time allowed : 3 hours] [Maximum marks : 100

Note : Attempt five questions in all, selecting at least one question from each section. Q. No. 1 is compulsory. All questions carry equal marks.

1. (a) What do you mean by congruent melting point ?
(b) What is reduced phase rule ?
(c) What is induced catalysis ?
(d) Distinguish between hard water and soft water.
(e) Define coagulation.
(f) What is meant by electrochemical corrosion ?
(g) What do you understand by viscosity index of a lubricant ?
(h) Define Iodine value.
(i) What do you understand by homopolymer and copolymer ?
(j) Define Lambert's law. 2×10=20

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[P.T.O.]

Section-A

2. (a) Draw and explain the phase diagram of Lead-Silver system in detail. 10
- (b) Write short note on cooling curves. 10
3. (a) Give a brief account of enzyme catalysis and explain its mechanism. 10
- (b) Give a brief account of the theories given to explain the mechanism of homogeneous and heterogeneous catalysis. 10

Section-B

4. (a) What is hardness of water ? Describe the estimation of hardness of water by any one method. 10
- (b) Discuss the boiler corrosion, in brief. 10
5. (a) Discuss the zeolite process for the removal of hardness of water. 10
- (b) What is meant by Desalination ? Describe the process by electrodialysis method. 10

Section-C

6. (a) Define corrosion. Explain dry corrosion in detail. 10
- (b) Write a note on stress corrosion. 10
7. (a) Define lubricants. How are they classified ? 10
- (b) Describe the following properties of lubricants :
 (i) Cloud point and Pour point 5
 (ii) Flash point and Pour point 5

Section-D

8. (a) Give the preparation, properties and uses of PF and UF resins. 10
- (b) What are silicones ? Discuss their important properties and uses. 10
9. (a) Describe the principle technique and applications of thermogravimetric analysis. 10
- (b) Write a note on flame photometry. 10