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[OUTLINE]: Parameterization of Non-Bonded Classical Mechanics Potentials for Neat Organic Liquids using a Multi-fidelity Bayesian Inference Approach

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I. Preliminaries

Definitions

- V: Volume
- U: Total energy (including potential and kinetic, excluding external energy such as due to gravity, etc)
- S: Entropy
- N: Number of particles
- T: Temperature
- P: Pressure
- k_B: Boltzmann constant
- β: $(k_B T)^{-1}$
- M: Molar mass
- ρ: Density (M/V)
- H: Enthalpy
- G: Gibbs Free Energy (free enthalpy)
- A: Helmholtz Free Energy
- u: reduced energy
- f: reduced free energy

II. Introduction

- MD as a critical research tool
  - Force fields that are transferable and quantitatively accurate are necessary for molecular simulation to be useful. [1–3]
- Transferability and inaccuracy issues
  - Transferability of MD force fields, and particularly sets of force field parameters, is a current limitation in the molecular simulation field.[4–7]
  - Inaccurate and poorly parameterized force fields have been shown to grossly misrepresent molecular systems. [8–10]
  - It has been shown that depending on the choice of force field, the same experiments for the same or similar systems can produce quantitatively different results, making the choice of force field far more important than it should be. [8, 9, 11–13]

- Parameterization efforts
  - Early
    * Until very recently, force fields have been parameterized manually, guided by the intuition of expert computational chemists.[14–25]
    * Despite attempts at improvement, many of the functional forms and parameters of popular force fields remain mostly unchanged due to the lack of clear, systematic methods for updating them.[26, 27]
    * Force fields like AMBER *parm94* showed intuitive departure by shrinking parameter space with clever atom typing defined by expert computational chemists.[14]
  - Second Gen
    * The parameterization of GAFF used a semi-automated genetic algorithm approach to select parameters.[28]
    * The parameterization of the rigid Tip4p-Ew model utilized a unique gradient assisted method. [29]
    * An incredible amount of work over a long period of time has still been necessary to get biomolecular force fields somewhat correct.[30]
  - Current efforts
    * A few notable attempts, such as GAAMP and ForceBalance, have been made in recent years towards the development of more automated and systematic force field parameterization methods.[31–34]
    * Each made important contributions to automated force field parameterization through clever use of objective function optimization, exploiting a variety of fitting data and allowing exploration of functional forms.
- Bayesian parameterization
  - Previous uses
    * Bayesian inference provides a robust statistical framework for force field parameterization. It has been shown that Bayesian approaches can be applied to a wide variety of data driven sciences. [35–42]
    * Bayesian inference methods have also been applied for uncertainty quantification in MD as well as limited parameterization problems on simple Lennard-Jones systems. [43–45]
  - Surrogate models/metamodels
    * Metamodeling has been critical in accelerating sampling driven processes which involve expensive calculations [46]
    * Some previous work has utilized efficient metamodels to accelerate Bayesian infernence driven parameterization of LJ models with mixed results. [45]
    * COFFE papers [47]
  - What our ideas for parameterization are/paper overall thesis
    * **Through systematically testing different multi-fidelity likelihood estimation workflows, we have found an optimal process which maximizes computational efficiency while yielding a force field consistent with the experimental data it was trained on.**
    * How can we combine different techniques, to find reasonable force fields in medium dimensionality in a computationally efficient manner

III. Methods

- Simulation protocol
- What parameters?
  - Non-bonded for cyclohexane and ethanol
  - I'm going to add a few more molecules, but try to keep the number of parameters capped at 10 (chain alkanes, cyclic alcohols, etc.)
  - 10
  - Specific SMIRKS:
    * $[\#8X2H1{+}0:1],[\#6X4:1],[\#1:1]{-}[\#6X4],[\#1:1]{-}[\#8],[\#1:1]{-}[\#6X4]{-}[\#7,\#8,\#9,\#16,\#17,\#35]$
- Property calculation For this paper we will be optimizing our parameters on two thermophysical properties; molar volume, $\hat{V}$, and heat of vaporization, ΔH_{vap}. This section details the methods we will use to calculate each.
  - Molar Volume System volume, V, can easily be calculated as:

$$V = x/timesy/timesz \tag{1}$$

    where x, y and z are the edge lengths of the simulation. This can be converted to a molar volume by dividing by the number of moles in the periodic box. We can write molar volume as:

$$\hat{V} = \frac{V}{N_{mol}} = \frac{V \times N_{Av}}{N_{part}} \tag{2}$$

    Where N_{mol} are the number of moles per box, N_{part} are the number of particles per box and N_{Av} is Avogadro's number.

1. Heat of Vaporization

The definition of the enthalpy of vaporization is:

$$\Delta H_{vap} = H_{gas} - H_{liq} = E_{gas} - E_{liq} + P(V_{gas} - V_{liq}) \tag{3}$$

The uncertainty in this calculation can be computed by bootstrapping or analytical estimation using MBAR. We will compare both results in order to determine whether the cheaper analytical estimate is accurate enough to be used.

An alternate, but similar, method for calculating the enthlapy of vaporization is recommended by Horn et al [29].

$$\Delta H_{vap} = -\frac{E_{liq,potential}}{N} + RT - PV_{liq} + C \tag{4}$$

In the above equation C is a correction factor for vibrational energies, polarizability, non-ideality of the gas and pressure. It can be calculated as follows.

$$\begin{aligned} C_{vib} = C_{vib,intra} + C_{vib,inter} & \\ &= (E_{vib,QM,gas,intra} - E_{vib,QM,liq,intra}) \\ &\quad + (E_{vib,QM,liq,inter} - E_{vib,CM,liq,inter}) \end{aligned} \tag{5}$$

The QM and CM subscripts stand for quantum and classical mechanics, resectively.

$$C_{pol} = \frac{N}{2} \frac{\left(d_{gas} - d_{liq}\right)^2}{\alpha_{p,gas}} \tag{6}$$

Where d_i is the dipole moment of a molecule in phase i and $\alpha_{p,gas}$ is the mean polarizability of a molecule in the gas phase.

$$C_{ni} = P_{vap} \left(B - T \frac{dB}{dT} \right) \tag{7}$$

Where B is the second virial coefficient.

$$C_x = \int_{P_{ext}}^{P_{vap}} \left[V\left(P_{ext}\right) \left[1 - \left(P - P_{ext}\right) \kappa_T \right] - TV\alpha \right] dP \tag{8}$$

Where P_{ext} is the external pressure and $V\left(P_{ext}\right)$ is the volume at P_{ext}.
This is frequently done as a single simulation calculation by assuming the average intramolecular energy remains constant during the phase change, which is rigorously correct for something like a rigid water molecule (intramolecular energies are zero), but less true for something with structural rearrangement between gas and liquid phases.

- Methods for metamodeling
  - MBAR
  - Surrogate models
    - GP models
      - Formalism for estimating some quantity Z at unknown location x_0 ($Z\left(x_0\right)$) from N pairs of observed values $w_i\left(x_0\right)$ and $Z\left(x_i\right)$ where $i = 1, ..., N$
      .

$$\hat{Z}\left(x_0\right) = \sum_{i=1}^{N} w_i\left(x_0\right) \times Z\left(x_i\right) \tag{9}$$

      - We find our weight matrix, **W**, by minimizing **W** subject to the following system of equations:
      .

$$\underset{W}{\text{minimize}} \qquad W^T \cdot \mathrm{Var}_{x_i} \cdot W - \mathrm{Cov}_{x_i x_0}^T \cdot W - W^T \cdot \mathrm{Cov}_{x_i x_0} + \mathrm{Var}_{x_0} \tag{10}$$

$$\text{subject to} \qquad \mathbf{1}^T \cdot W = 1 \tag{11}$$

      - where the literals

$$\left\{ \mathrm{Var}_{x_i}, \mathrm{Var}_{x_0}, \mathrm{Cov}_{x_i x_0} \right\} \tag{12}$$

        stand for

$$\left\{ \mathrm{Var}\left(\begin{bmatrix} Z(x_1) & \cdots & Z(x_N) \end{bmatrix}^T \right), \mathrm{Var}(Z(x_0)), \mathrm{Cov}\left(\begin{bmatrix} Z(x_1) & \cdots & Z(x_N) \end{bmatrix}^T, Z(x_0) \right) \right\} \tag{13}$$

      - The weights summarize important procedures of the inference process:
        - They reflect the structural closeness of samples to the estimation location, x_0

        - They have a desegregating effect, to avoid bias caused by sample clustering
    - **Hypothesis: With a multi-fidelity hierarchical observable calculation scheme, we can quickly approach the true forward model produced by MD simulation**
    - Explanation of potential multi-fidelity posterior sampling algorithms:
        * 3 Levels of property calculation
            - High fidelity: Full MD simulation at a single point in parameter space
            - Medium fidelity: Use MBAR to estimate properties over a conservative range of parameter space in order to create a hypervolume of data over which we can construct a model
            - Low fidelity: Use data from medium fidelity calculations in order to fit a regression model over a hypervolume of parameter space
                - For right now, most plausible technique is GP regression, but could brainstorm some others

IV. Experiments + Results and Analyses

- **Hypothesis: Using a multi-fidelity likelihood calculation scheme described in the previous section will provide not only a substantial speed up over a traditional inference approach with purely simulation used in the likelihood estimate, but will also result in a final force field with accuracy rivaling that of the expensive approach.**
- Experiments for testing sampling workflow
- Results from 2D experiments with $[\#6X4:1]$
    - Stability of solution (multiple starting points both in phase and out of phase)
    - Simulation of properties using final distribution of parameters
- Same results from 4D experiments with $[\#1:1]-[\#6X4]$ added
- Scale up to all 5 SMIRKS
- **Hypothesis: Different sampling algorithms will affect the speed of convergence to our final force field as well final distribution of parameters sampled.**
- Ideas for comparing force fields resultant from different sampling methods
    - KL divergence
    - Speed of convergence
    - Stability of solution (from different starting points do we get same answer?)
    - Simulation of properties, using final parameters, that were not in training set
        * 3-fold verification
            - Different properties not in training set
            - Extrapolation to thermodynamic state outside of training set (T, P)
            - Other molecules outside of training set that have the same SMIRKS types

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