# AY23-1 Chemical Engineering Course and Program Brief

LTC Matthew Armstrong

LTC Sam Cowart

Dr. Simuck Yuk

MAJ Caspar Yi

**MAJ Patrick Bowers** 

Dr. Enoch Nagelli

Dr. Andy Biaglow

MAJ Jeff Chin

LTC John Belanger

CPT(P) Galen Mandes

# Agenda

- Course Briefings
  - -CH365
  - -CH485
  - -CH363
  - -CH350
  - -CH459
- Summer training for CH459 Transition
- Signature Writing Event (SWE) in CH459
- Bioengineering: Curriculum Committee Memos for CH300 and CH350
  - For cadets to sign up for courses AY24 and beyond: memos
  - CH350 Update going into AY23
- ABET and Advisory Board Update (07APR22)
  - Dinner, Hotel, and Schedule
- Majors Update from Class of 2025
- Recruiting/Mentorship plan for inbound faculty

### CH365: Chemical Engineering Thermodynamics

Course Director: Dr. Biaglow Course Supervisor: Dr. Nagelli

Credit Hours: 3.0 (BS=0, ET=3.0, MA=0) Prerequisites: CH363, CH364, MA366, MC312

> Co-requisite: None Lessons: 40 @ 55 min Special Requirements: None

This course covers the body of thermodynamic knowledge necessary for understanding modern chemical process simulation. Students learn the theory behind the thermodynamic methods used in the software. The course includes calculus- and numerical-based thermodynamics approaches for determining the properties of substances, solutions, and multiphase mixtures. Topics include equations of state, pure component properties, transport properties, properties of mixtures, fugacity, excess properties, activity coefficients, and phase equilibria. The problems in the course emphasize engineering applications. Topics covered in class are related to real systems through the use of chemical process simulators.

### Course Assessment – Items from Section III

#### **Sustain:**

Writing project - resume exercise involved intense use of instructor time but positive changes were observed; many teachable moments

Capstone project – calculation of properties and comparison with CAD

Addition of writing to capstone

#### Improve:

CH365

More emphasis on Ch 11, specifically excess properties More practice on calculating fugacity and activity. Use of writing rubric to assess capstone.

### Topics – by Chapter

Chemical Engineering Thermodynamics, Smith, van Ness, Abbott, and Swihart, 9th Edition (2021)

- Introduction (Ch. 1)
- First Law (Ch. 2)
- Equations of State (Ch. 3)
- Heat (Ch. 4)
- Entropy and Second Law (Ch. 5)
- Fluid Properties (Ch. 6)
- Equilibrium (Ch. 10)

\*Individual Points :

Solution Thermodynamics (Ch. 11)

### Assessment – Graded Events

1 *Term End Exam @ 500 pts:	500	22.42%
1 *Capstone Design Project @ 300 pts	300	13.45%
3 *Capstone IPRs @ 30 pts each:	90	4.04%
3 *WPRs @ 200 pts each:	600	26.91%
64 *Problems @ 10 pts each:	640	28.70%
1 *Writing Assignment @ 100 pts:	100	4.48%
Total:	2230	100.00%

2230

100%

### CH485: Heat and Mass Transfer

Course Director: LTC Sam Cowart Course Supervisor: Dr. Biaglow

Credit Hours: 3.5 (BS=0, ET=3.5, MA=0)
Prerequisites: MA364 and MC312
Co-requisite: None
Lessons: 30 @ 75 min, 7 @ 120 min
Special Requirements: None

This course includes the study of the mechanisms of energy and mass transport, with special emphasis on applications in engineering systems. Coverage includes Fourier's Law of Heat Conduction, and Fick's Law of Diffusion, the development of shell energy and species balances, and the use of these equations to solve for temperature and concentration profiles in chemical engineering systems. An important emphasis in the course is the use of transport equations to understand species diffusion, convection, and chemical reaction in equipment design.

CH485

### Course Assessment – Items from Section III

### Sustain:

Problem solving days prior to WPR/ICPS Lesson structure & content\*

### **Improve:**

- Continue to revise problem sets Too much reliance on old solutions.
- \*Incorporate radiation heat transfer material (external source).
- Update Lab 6 Arnold Cell
  - Hands-on in the ChemE Lab, with conditions allowing experimental data to be taken in ~30 minutes.
  - Eliminates reliance on pre-recorded video of acetone diffusion.

### **Topics**

Mass and Heat Transfer, T.W. Fraser Russell, A.S. Robinson, and N.J. Wagner, Cambridge University Press, Copyright 2008

• Introduction (1 Lesson)

\*Individual Points :

- Macroscopic mass, energy, and species balances
  - Chemical reactors (2 lessons)
  - Heat exchangers (4 lessons)
  - Mass contactors (2 lessons)
- Microscopic mass, energy, and species balances
  - Conduction and diffusion (9 lessons)
  - Convective heat and mass transfer (7 lessons)
  - Transport Analogies (4 lessons)

### Assessment – Graded Events

8 *Problem Sets @ 50 pts each:	400	18%
2 *In-Class Prob. Sets @ 100 pts each:	200	9%
3 *WPRs @ 200 pts each:	600	27%
1 *Writing Assignment @ 200 pts:	225	10%
7 Labs @ 40 pts each:	280	13%
1 *Term End Exam @ 500 pts:	500	23%
Total:	2205	

87%

1925

### CH363: Separations Processes

Course Director: Dr. Enoch Nagelli Course Supervisor: Dr. Biaglow

Credit Hours: 3.5 (BS=0, ET=3.5, MA=0)
Prerequisites: CH362
Co-requisite: None
Lessons: 30 @ 75 min, 7 @ 120 min

Special Requirements: None

This course covers methods for the physical separation of chemicals. Topics include dew point and bubble point calculations, adiabatic flash, distillation, chromatography, liquid-liquid and gas-liquid absorption/stripping. Students are taught the significance of staging of unit operations. Heavy emphasis is placed on theory of operations, numerical methods of solution, and simulation.

### Course Assessment – Items from Section III

#### Sustain:

- Technical Writing: Written Research Paper
- Capstone project Ties all key concepts together. Assesses communication outcome.

ChemCAD use in concert with theory/ every day CHEMCAD

### Improve:

**CH363** 

- Need extra distillation lesson in block 3; topic needs more treatment and fits better in block 3
- Improve cadets reliance on the text as a resource
- More instructor problems for HW/ less book problems

### Topics – by Chapter

Separation Process Principles, 4<sup>th</sup> Ed., by J.D. Seader, E.J. Henley and D.K. Roper

- Introduction to Separations/DOF Analysis (Ch. 1)
- Vapor-liquid, gas-liquid, solid-liquid, flash(Ch. 4)
- Cascading configurations (Ch. 5)
- Designing trayed towers and packed columns (Ch. 6)
- Optimizing towers and columns (Ch. 7)
- Liquid-liquid extraction (Ch. 8)
- Multi-component distillation (Ch. 9)
- Capstone Project

Assessment –	Grade	d Events	
8 Problem Sets @ 40 pts each:	320	13.3%	
4 *In-Class Prob. Sets @ 75 pts e	each:	300	12.5%
3 *WPRs @ 200 pts each:		600	25.0%
1 *Research Paper @ 100 pts:	100	4.2%	
7 *Labs @ 40 pts each:		280	11.6%
1 *Term End Exam @ 500 pts:	500	20.8%	
1 *Capstone		300	12.5%
Total:		2400	
*Individual Points :		1820	75.8%

Consideral Francisco

### CH350: Introduction to Bioengineering

Course Director: Dr. Simuck Yuk Course Supervisor: COL Burpo

Credit Hours: 3.0 (BS=0, ET=3.0, MA=0)
Prerequisites:
Co-requisite: None
Lessons: 30 @ 75 min

Special Requirements: None

The purpose of this course is to provide the introductory knowledge for understanding the biotechnology/bioprocessing engineering. Topic includes enzyme kinetics, molecular biology, cell growth, bioreactors, and bioprocesses. The bioprocess control and its application to different biological systems are covered in the classroom instruction. An important emphasis is made on the use of kinetics and process controls on the biological systems for engineering application.

### Course Assessment – Items from Section III

#### **Sustain:**

Continue to use of DeLisa (3<sup>rd</sup> Ed.)
Continue to introduce problem demos

### CH350 Improve:

Increase number of in-class problem demos.

Improve problem sets and WRPs.

Increase number of coding-related problems.

### Topics – by Chapter

Bioprocess Engineering Basic Concepts, 3rd Ed., by Michael L. Shulter, Fikret Kargi, Matthew DeLisa, Prentice Hall.

Quantitative Fundamentals of Molecular and Cellular Bioengineering, by K. Dane Wittrup, Brice Tidor, Benjamin J. Hackel, and Casim A. Sarkar, The MIT Press.

- Introduction
- Enzyme Kinetics
- Central Dogma to Molecular Biology
- Cell Growth
- Bioreactor Selection
- Bioprocess Consideration

### Assessment – Graded Events

1 *Capstone Presentation	100	9%
1 *Capstone Paper	200	18%
2 *WPRs	400	36%
5 Problem Sets(50pts/ea.)	250	23%
1 *Discussion	150	14%
Total:	1100	100.00%

### CH459: Chemical Engineering Lab

Course Director: LTC Belanger

Course Supervisor: LTC Cowart/Dr. Nagelli

Credit Hours: 3.5 (BS=0, ET=3.5, MA=0) Prerequisites: CH362/CH363/CH364

Co-requisite: None

Lessons: 40 @ 120 min, 7 @ 120 min Special Requirements: None

This course provides laboratory experience in selected chemical engineering unit operations, such as gas absorption, evaporation, distillation, liquid-liquid extraction, cooling tower, heat exchanger, and chemical reactors. Process control and process safety are emphasized in laboratory and classroom instruction. Written and oral reports required.

### Course Assessment – Items from Section III

#### **Sustain:**

Technical Writing: Written Lab Reports/ Posters/ Executive Sum.

**SACHE Modules** 

In Progress Reviews (IPRs)

Use of CHEMCAD to model all lab data

Videos of each experiment for cadets to develop procedure

#### Improve:

CH459

Cadet peer evaluations at assigned role during lab rotation; make sure cadets assess the cadets performance at role

More practice with setting up flow charts by hand and labeling streams with process variables with units

- 1. Unit Operations of Chemical Engineering, 7th Edition, by Warren L. McCabe, Julian C. Smith and Peter Harriott; McGraw-Hill. 2005.
- 2. Plant Design and Economics for Chemical Engineers 5<sup>th</sup> ed., Peters, Max S. and Klaus D. Timmerhause, McGraw-Hill, New York 2003, ISBN-10: 0071240446

1. Batch and CSTR EC: Short Ht Exchgnr

- 2. Cooling Tower
- 3. Hydrogen Fuel Cell
- 4. Single/Double Effect Evaporator
- 5. Distillation
- Carbon Dioxide Absorber

### Assessment – Graded Events

6 Lab HWs@ 20 pts each	120*	4.1%
6 Lab Exec. Sum/ Poster/ Report	600	20.5%
6 IPRs @25 pts ea	150	5.1%
6 Lab Execution (aka. lab rubric)	450	15.4%
2 *WPRs @ 550 pts each:	1100*	37.7%
1 *Term End Exam @ 500 pts:	500*	17.1%
Total:	2920	
*Individual Points :	1720	58.9%

# FDW for CH459: Summer Training

- FDW Mentors: Dr. Enoch Nagelli (CH459) and LTC Sam Cowart (CH101 and CH459)
- FDW Mentee: LTC Belanger
- Summer of 2022: 27JUN-01JULY (CH101) and 01JULY-29JULY (CH459)
  - LTC Belanger to attend beginning portion of CH101 FDW
  - Schedule daily will include experiments and calculations overview/discussion
- Round Robin 1 and 2 Experiments, Procedure and Execution
- Process controls analysis, P&ID, mass & energy balance calculations, and other key calculations
- Process simulation modeling via ChemCAD

### SWE in CH459: EXSUM

- EXSUM (Individual submission) with established Rubric for Assessment in Course
  - Heat Exchanger Lab

- Cadet teams assigned a lab hour to conduct experiment and collect data
  - Pre-safety startup review and procedure
  - Laplace block controls analysis
  - P&ID
  - Calculations and ChemCAD process simulation model
  - Error Analysis

- ChemE Faculty as Mentors for each cadet in course
  - Scheduled IPRs

# Bioengineering

- Curriculum Committee Memos for CH300 and CH350
  - Allows class of 2024 and 2025 to enroll into courses once approved
  - Future offerings in Redbook
  - Memorandum for permanently placing CH300 and CH350 in the Redbook was written and program review complete. Currently, pending approval from COL Burpo and COL James.

# CH350 Update

- DeLisa's textbook generally gives a good overview of how major engineering concepts (mostly from chemical engineering) can be modified and readjusted for various biological systems.
  - 1) Some key concepts introduced in the book can better be understood for cadets who have taken CH363 and CH364.
  - 2) Units introduced in the book are sometimes misleading.
- The introduction of additional coding-related problems during the class can enhance cadet's understanding on the quantitative analysis of biological pathways and system designs.

# ABET Advisory Board (07-08APR22)

- In-person meeting with the Teams option for those who cannot travel
- Combination of representatives from our constituencies:
  - Industry
  - Academia
  - Army
- Review data; take survey; interact with Chemical Engineering Cows and Firsties
- Give curriculum advice
- Inform program decisions
- Helps write the program educational objectives and mission statement (these are ABET requirements)

# **Advisory Board**

					Attending	In-persion
No	Name	Title	School - Company	Email	?	or remote?
		EleCent Team	LLNL (Lawrence Livermore			
1	Lucy Hair	Leader	National Laboratory)	hair1@llnl.gov	Yes	In-Person
2	COL (Ret) Paul Dietrich	Chemical Officer	Chemical Officer/Industry	paul@the-dietrichs.com	Yes	In-Person
3	Kisondra Tanev	Vice President	Bank of America	kisondra@gmail.com	Yes	In-Person
4	Donald Glaser	President	Simulation Solutions	dglaser@simulation- solutions.com	Yes	In-Person
•	Boriara Graser	rresident		mgarvey@simulation-	100	111 1 013011
5	Matt Garvey	Engineer	Simulation Solutions	solutions.com	Yes	In-Person
	,			matthew.liberatore@Utole		
6	Matthew Liberatore	Professor	University of Toledo	do.edu	Yes	In-Person
7	COL Aaron Hill	Academy Professor	USMA; CME	aaron.hill@westpoint.edu	Yes	In-Person
/	COLABIOITIIII	Automation	OSIVIA, CIVIL	aaron.niii@westpoint.edd	163	III-reison
8	Kevin Shipe	Engineer	NALCO Champion	kevin.a.shipe@gmail.com	Yes	Remote
9	Patrick Underhill	Professor	RPI	underhill@rpi.edu	Yes	Remote
				gautham.krishnamoorthy@		
10	Gautham Krishnamoorthy	Professor	UND	und.com	Yes	Remote
11	Kelly Schutz	Assistant Professor	Lehigh University	kes513@lehigh.edu	Yes	Remote
	Lynn Walker	Professor	Carnegie Mellon University	lwalker@anderw.cmu.edu		
	,	COL(RET)/				
	Greg Ritch	<b>Engineering Officer</b>	MIL/Industry	garitch@cvzoom.net		
		Industry, Chem E				
	Mike Deforest	(Old Grad '07)	SMK Packaging	mike@smkpackaging.com		

# Proposed Agenda for Advisory Board Meeting

Date	Time	Event	Location	OIC	Task
	NLT1700	Pick up 15 PAX van	Motorpool	MAJ Yi/ MAJ Chin	
7APR	1530	Dr. Matthew Liberatore Chem E. Seminar (abstract to follow)	BH465 (Chair-side)	CPT(P) Bowers/ CPT(P) Mandes	
	1830 - UTC	Dinner	Bear Mountin Inn Fort Montgomery, NY	LTC Armstrong	make reservation
	0730	Pick up	Holiday Inn Express 1106 Route 9W Fort Montgomery, NY (845) 446 - 4277	MAJ Chin	Drive van
	0745 - 0800	Arrival and lite breakfast	Bartlett Hall 465 (table side)	LTC Armstrong Dr. Nagelli	Breakfast
	0800 - 0830	Session1: Introductory remarks and ABET orientation	BH465	COL Burpo LTC Armstrong	
	0830 - 0920	Session2: Program Assessment Student Outcomes Assessment Discussion of Program Objectives	BH465		
8APR	0920 - 0935	Board Surveys	BH465	LTC Armstrong	Survey Parts 1 & 2
SAPK	0935 - 0950	Session3: CH367 Update	BH465 (table side)	COL James	
	1000 - 1050	Career Panel (Cows)	BH465 (chair side)	LTC Armstrong	
	1050 - 1115	Cadet Discussions (Fisties and Cows)	BH465 (chair side)		
	1130 - 1300	Lunch/Firstie Mock Interview Round Robin	BH465	Dr. Nagelli	
	1300 - 1330	Board backbrief on cadet interactions	s BH465	LTC Armstrong	
	10.10 1.100	2	211465	Dr. Nagelli	
	1340 - 1430	Session 4: Future Challenges	BH465	LTC Armstrong	
	1430 - 1455	Admin and Unit Ops Lab Tour	SBBH	Dr. Nagelli	
	1500 - 1600	Wrap up	BH465		

# Majors Update Class of 2025

- Currently 36 cadets declared ChemE
  - 2 sections of major ChemE course
  - CH459 3 sections a possibility similar to AY20

# Recruiting Plan for Faculty in ChemE

- Communication and outreach by current ChemE faculty
  - Research that leverages the department's current research threads for execution when inbound
  - Thesis committee members (non-voting) for senior faculty (Dr. Biaglow, LTC Cowart, Dr. Yuk, and Dr. Nagelli)
  - Collaboration between faculty researchers and PI at respective universities

	Future Faculty Member	ACS Start	USMA Arrival	School	Cost Category	Research Focus
A WINDOWS	LTC John Belanger (Sponsor LTC Cowart)	AUG 2019	JUN 2022	Stanford	High	Complex Materials
Discourse of the second of the	CPT Sam Lowell (Sponsor MAJ Bowers)	N/A	JUN 2022	NPS	N/A	NPS Thesis: Reduction Expansion Synthesis for Magnetic Alloy Powders
* * * * * * * * * * * * * * * * * * *	CPT Louis Tobergte (Sponsor CPT(P) Mandes)	AUG 2022	JUN 2023	Carnegie Mellon	High	Thesis completed at Leeds: Water
TO COLUMN TO STANK	CPT Liz Golonski (Sponsor MAJ Bowers)	AUG 2022	JUN 2024	Colorado School of Mines	Low	Catalytic membrane reactors
	CPT Nijel Rogers (Sponsor CPT(P) Mandes)	AUG 2022	JUN 2024	Princeton University	High	Soft matter
	CPT Christopher Stewart (Sponsor CPT Lowell)	AUG 2023	JUN 2025	TBD	TBD	TBD

### LTC John Belanger

- Sponsored by LTC Cowart
- CH459 mentor is Dr. Nagelli
- Programmed for CH459, and CH400 (eventually)
- AIChE OIC?
- Will conduct CH459 FDW Summer 2022
- Currently researching complex material/ tree sap/ kino/ rheology

### **CPT Sam Lowell**

- Sponsored by MAJ Bowers
  - Sub: Dr. Yuk and CPT(P) Mandes
- Completed M.S. degree directly after USMA graduation at NPS (MechE)

# Questions

# Backup Slides

AY22		AY23		AY24	
AY22-1 (Fall)	AY22-2 (Spring)	AY23-1 (Fall)	AY23-2 (Spring)	AY24-1 (Fall)	AY24-2 (Spring)
CH363 (Armstrong)	CH362 (Cowart)	CH363 (Nagelli)	CH362 (Cowart)	CH363 (Nagelli)	CH362 (Cowart)
CH459 (Nagelli)	CH364 (Yuk)	CH459 (Belanger)	CH364 (Yuk)	CH459 (Belanger)	CH364 (Yuk)
CH485 (Cowart)	CH402 (Biaglow)	CH485 (Cowart)	CH402 (Biaglow)	CH485 (Cowart)	CH402 (Biaglow)
CH365 (Biaglow)	CH400 (Nagelli)	CH365 (Biaglow)	CH400 (Nagelli)	CH365 (Biaglow)	CH400 (Nagelli)
CH350 (Yuk)	CH367 (James)	CH350 (Yuk)	CH367 (James)	CH350 (Yuk)	CH367 (James)
	CH300 (Burpo)		CH300 (Burpo)		CH300 (Burpo)
	CH450 (Burpo)		CH450 (Burpo)		CH450 (Burpo)
Yi (2x GC sections)	Mandes (2x GC sections)	Mandes (GC)	Mandes (GC)	Mandes (GC)	Mandes (GC)
Bowers (GC 3x sections)	Bowers (GC 3x sections)	Yi (GC)	Belanger (GC)	Bowers (GC)	Bowers (GC)
Mandes (GC 3x sections)	Yi (2x GC sections)	Chin (GC+S1)	Chin (GC+S1)	Lowell (GC)	Lowell (GC)
Chin (2x sections+S1)	Chin (2x sections+S1)	Bowers (GC)	Bowers (GC)		Belanger (GC)
		Lowell(GC)	Yi (2x GC sections)		
			Lowell(GC)		

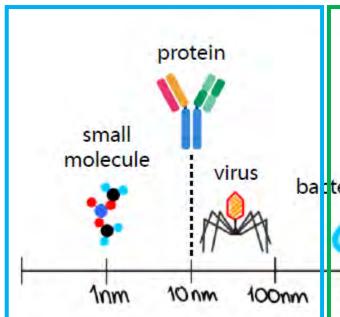
AY25			AY26	AY27		
AY25-1 (Fall)	AY25-2 (Spring)	AY25-1 (Fall)	AY25-2 (Spring)	AY25-1 (Fall)	AY25-2 (Spring)	
CH363 (Nagelli)	CH362 (Cowart)	CH363 (Nagelli)	CH362 (Cowart)	CH363 (Nagelli)	CH362 (Cowart)	
CH459 (Belanger)	CH364 (Yuk)	CH459 (Belanger?)	CH364 (Yuk)	CH459 (Belanger?)	CH364 (Yuk)	
CH485 (Cowart)	CH402 (Biaglow)	CH485 (Cowart)	CH402 (Biaglow)	CH485 (Cowart)	CH402 (Biaglow)	
CH365 (Biaglow)	CH400 (Belanger)	CH365 (Biaglow)	CH400 (Nagelli)	CH365 (Biaglow)	CH400 (Nagelli)	
CH350 (Yuk)	CH367 (James)	CH350 (Yuk)	CH367 (James)	CH350 (Yuk)	CH367 (James)	
	CH300 (Burpo)		CH300 (Burpo)		CH300 (Burpo)	
	CH450 (Burpo)		CH450 (Burpo)		CH300 (Burpo)	
Rogers (GC)	Rogers (GC)	Rogers (GC)	Rogers (GC)	Rogers (GC)	Rogers (GC)	
Lowell(GC)	Lowell(GC)	Golonski(GC)	Golonski(GC)	Golonski(GC)	Golonski(GC)	
Golonski(GC)	Golonski(GC)					

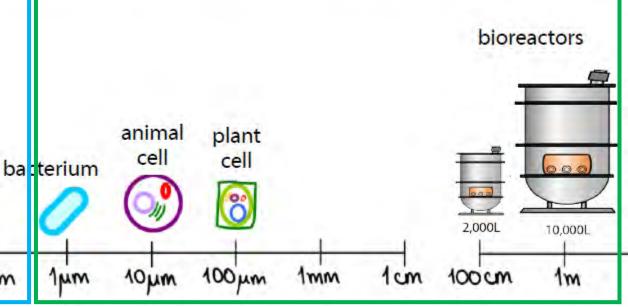
# Bioengineering

- Curriculum Memoranda completed and approved by Curriculum Committee for CH300 & CH350
- Bioengineering track
- 3-course engineering sequence
- Bioengineering minor (2-3 years)
- Bioengineering major (5 − 15+ years); ABET accredited ~2030

# CH350 Course Layout

Molecular - Cellular - Bioreactor/Purification Systems - Societal







Block 1 - Biomolecular Considerations 12 lectures (plus 1 WPR) LSN 1-13

Block 2 - Cellular Considerations 8 lectures (plus 1 WPR) LSN 14 - 22

Block 3 Downstream
Considerations 6
lectures (Plus 2
lessons of Group
Presentations)
LSN 23-30

### CH350 Detailed Overview

Day	2-Day	LSN	Subject	Reading	Instructor	Event	Point
			Block I: Biomolecul	ar Considerations			
Th	21-Jan	1	Introduction to Bioprocess Engineering	Ch 1 (Schuler et al)	Dr. Yuk		
M	25-Jan	2	Introduction to Cells and Biomolecules Part 1	Ch 2.1 (Schuler et al)	Dr. Yuk		
Th	28-Jan	3	Introduction to Cells and Biomolecules Part 2	Ch 2.2-2.3 (Schuler et al)	Dr. Yuk	Problem Set 1 Due	50/50
М	1-Feb	4	Enzyme kinetics and immobilized enzyme systems Part 1	Ch 3.1-3.2.3.1 (Schuler et al)	Dr. Yuk		
ŧ	5-Feb	5	Enzyme kinetics and immobilized enzyme systems Part 2	Ch 3.2.3.2-3.2.6 (Schuler et al) Skim: MIT Press Chap 3.3	Dr. Yuk	Discussion Grade 1	30/80
Ťu	9-Feb	6	Enzyme kinetics and immobilized enzyme systems Part 3	Ch 3.3 (Schuler et al)	Dr. Yuk		
Ti.	11-Feb	7	Non-covalent binding interactions; Binding equilibria and kinetics	MIT Press Chap 2.1-2.4	Dr. Yuk	Problem Set 2 Due	50/130
To	16-Feb	8	Central dogma of molecular biology	Ch 4 (Schuler et al)	Dr. Yuk	2110100 0117 000	2 11 10
Th	18-Feb	9	Genetic/metabolic regulation	Ch 5 (Schuler et al)	Dr. Yuk		
M	Z2-Feb	10	Recombinant DNA technology / Heterologous protein expression	Ch 8 (Schuler et al)	Dr. Yuk	Problem Set 3 Due	50/180
Ŧ	36-Feb	11	Protein engineering	Li, C. et al (Article)	Dr. Yuk		
In	2-Mnr	12	Metabolic engineering	Varma, A et al (Article)	Dr. Yuk	Discussion Grade 2	30/21
Πb	+Mar	13	WPRI		Dr. Yuk	WPR 1	200/41
			Block II: Cellular				
Γu	9-Mar	14	General concepts of cell growth - Part 1	Ch 6.1 (Schuler et al)	Dr. Yuk		
8	12-Mar	15	General concepts of cell growth - Part 2	Ch 6.2-6.4 (Schuler et al)	Dr. Yuk		
Tu	16-34ar	16	General concepts of cell growth - Part 3	Ch 7 (Schuler et al)	Dr. Yuk	Problem Set 4 Due	50/460
Th	18-54nr	17	Operational Considerations for Bioreactors: Suspension and Immobolized Cultures	Ch 9.1-9.3 (Schuler et al)	Dr. Yuk	Discussion Grade 3	30/490
M	22-Mar	18	Bioreactor selection, scale-up and process control Part 1	Ch 9.4-10.1 (Schuler et al)	Dr. Yuk		545
Ξ.	26-Mar	19	Bioreactor selection, scale-up and process control Part 2	Ch 10.2-10.4 (Schniler et al)	Dr. Yuk	Problem Set 5 Due	50/540
W	31-Mar	20	Upstream operational challenges and other biomanufacturing considerations		rofessor DeLisa		
3	3-Apt	21	Guest Lecture - Erin Johnson (MERK) Up Stream Processing		Erin Johnson	The second secon	****
<u> </u>	9-Apr	22	WPR II Block III: Downstre	im Considerations	Dr. Yuk	WPR 2	200/74
Γū	13-Apr	23	Product recovery and separation - Part 1	Ch 11.1-11.3 (Schuler et al)	Dr. Yuk		
Th	15-Apr	24	Product recovery and separation Part 2	Ch 11.4-11.7 (Schuler et al)	Dr. Yuk		
M	19-Apr	25	Bio Process Considerations using Animal Cells	Ch 12 (Schuler et al)	Dr. Yuk	Discussion Grade 4	30/77
Τb	22-Apr	26	Bio Process Considerations using Plant Cells	Ch 13 (Schuler et al)	Dr. Yuk	Distantion Of the 4	30///
50	26-Apr	27	Medical Applications of Bioprocess Engineering	Ch 15 (Schuler et al)	Dr. Yuk		
Ŧ	30-Apr	28	Traditional Industrial Bioprocesses	Appendix A (Schuler et al)	Dr. Yuk	Discussion Grade 5	30/80
Tir	4-May	29	Capstone Presentations	- primari (remer ci m)	Dr. Yuk	Capstone Paper	200/100
λī	10-May	30	Cadet Presentations		Dr. Yuk	Capstone Presentation	100/110
			TEE Week of 12-15 DEC 2021			TOTAL POINTS	11
					51. 101		PS's WPR's DG's

### CH350 Detailed Overview

### • Textbook:

- Bioprocess Engineering Basic Concepts, 3rd Ed., by Michael L. Shulter, Fikret Kargi, Matthew DeLisa, Prentice Hall.
- 2. Quantitative Fundamentals of Molecular and Cellular Bioengineering, by K. Dane Wittrup, Brice Tidor, Benjamin J. Hackel, and Casim A. Sarkar, The MIT Press.

### Major Blocks:

- 1. Biomolecular considerations: enzyme kinetics and molecular biology
- 2. Cellular considerations: cell growths and process considerations of bioreactor development
- Downstream considerations: bioprocess development in different cell cultures

### CH350: Introduction to Bioengineering

Course Director: Dr. Simuck Yuk Course Supervisor: COL Burpo

> Credit Hours: 3.0 (BS=0, ET=3.0, MA=0) Prerequisites: Co-requisite: None Lessons: 30 @ 75 min

> > Special Requirements: None

The purpose of this course is to provide the introductory knowledge for understanding the biotechnology/bioprocessing engineering. Topic includes enzyme kinetics, molecular biology, cell growth, bioreactors, and bioprocesses. The bioprocess control and its application to different biological systems are covered in the classroom instruction. An important emphasis is made on the use of kinetics and process controls on the biological systems for engineering application.

### Course Assessment – Items from Section III

#### **Sustain:**

Continue to use of DeLisa (3<sup>rd</sup> Ed.) Continue to introduce problem demos

#### Improve: CH350

Increase number of in-class problem demos.

Improve problem sets and WRPs.

Increase number of coding-related problems.

### Topics – by Chapter

Bioprocess Engineering Basic Concepts, 3rd Ed., by Michael L. Shulter, Fikret Kargi, Matthew DeLisa, Prentice Hall.

Quantitative Fundamentals of Molecular and Cellular Bioengineering, by K. Dane Wittrup, Brice Tidor, Benjamin J. Hackel, and Casim A. Sarkar, The MIT Press.

- Introduction
- **Enzyme Kinetics**
- Central Dogma to Molecular Biology
- Cell Growth
- **Bioreactor Selection**
- **Bioprocess Consideration**

### Assessment – Graded Events

1 *Capstone Presentation	100	9%
1 *Capstone Paper	200	18%
2 *WPRs	400	36%
5 Problem Sets(50pts/ea.)	250	23%
1 *Discussion	150	14%
Total:	1100	100.00%

### CH300: Introduction to Biomedical Engineering

Course Director: COL Burpo Course OIC: MAJ Jeffrey Chin

Credit Hours: 3.0 (BS=0, ET=3.0, MA=0) Prerequisites: CH102, MA205 Co-requisite: None Lessons: 30 @ 75 min Special Requirements: None

This course provides a basis for understanding the application of engineering principles to problems in medicine and biology. It provides preparation for future graduate work in medical school, biomedical engineering, and chemical engineering. Specifically, the objectives of the course are: (1) to provide an introduction to the field and how it relates to other fields of engineering and science, (2) the develop the ability to apply mathematics, science, and engineering to solve problems, (3) to develop an understanding of the impact of engineering solutions on the medical field and society as a whole, and (4) to understand current topics within the field.

#### COA 1

Block I: Molecular and cellular principles

- Biomolecular principles (Ch. 2)
- Biomolecular principles: Nucleic acids (Ch. 3)
- Biomolecular principles: proteins (Ch. 4)
- Cellular principles (Ch. 5)

Block II: Physiological principles

- Communication systems in the body (Ch. 6)
- Engineering balances: respiration and digestion (Ch. 7)
- Circulation (Ch. 8)
- Removal of molecules from the body (Ch. 9)

### Assessment – Graded Events

6 *HWs@ 50 pts each	300	21.4%
2 *WPRs @ 200 pts each:	400	28.6%
1 *Capstone	200	14.3%
1 *TEE	500	35.7%
Total:	1400	

\*Individual Points : 1400 100%

TEXT: Biomedical Engineering, 2<sup>nd</sup> Edition, by W. Mark Saltzman; Cambridge University Press, 2015.

### COA 2

Block I: Molecular and cellular principles

- Biomolecular principles (Ch. 2)
- Biomolecular principles: Nucleic acids (Ch. 3)
- Biomolecular principles: proteins (Ch. 4)
- Cellular principles (Ch. 5)

Block II: Physiological principles

- Communication systems in the body (Ch. 6)
- Engineering balances: respiration and digestion (Ch. 7)
- Circulation (Ch. 8)
- Removal of molecules from the body (Ch. 9)

Block III: Biomedical Engineering

- Biomechanics (Ch. 10)
- Bioinstrumentation (Ch. 11)
- Bioimaging (Ch. 12)
- Biomolecular Engineering I: Biotechnology (Ch. 13)
- Biomolecular Engineering II: Engineering of Immunity (Ch. 14)

# CH450: Bioengineering Modeling and Analysis Course Director: COL John Burpo

Credit Hours: 3.0 (BS=0, ET=3.0, MA=0)
Prerequisites: CH102, MA205
Co-requisite: None
Lessons: 30 @ 75 min
Special Requirements: None

This course provides a broad understanding of bioengineering disciplines to include biomechanics, biomaterials, tissue engineering, biocatalysis, biochemical engineering, and biosensors. Fundamental concepts of molecular kinetics, thermodynamics, and mass transport are applied in problem sets in each bioengineering sub-discipline and capstone design project providing students the opportunity for modeling, analysis, and design from the biomolecular to physiological length scale and across multiple time scales. Modeling software such as MATLAB and Mathematica is extensively used.

### Topics – by Chapter

TEXT: Introduction to Biomedical Engineering, 3<sup>rd</sup> Edition, by John Enderle and Joseph Bronzino; Academic Press, 2012.

- Part I: Biomechanics (Ch. 1 and 4)
- Part II: Biomaterials (Ch. 5)
- Part III: Tissue Engineering (Ch. 6)
- Part IV: Biomedical Enzyme Kinetics (Ch. 7 and 8)
- Part V: Biochemical Engineering (Handouts)
- Part VI: Biosensors (Ch. 10)

### Course Assessment – Items from Section III

#### **Sustain:**

- Best looking Dept Head/ CD/ Prof at the academy
- Strong instructor personal experience in mathematics, engineering, and chemistry

Each lesson considers interdisciplinary science and engineering topics –science topics are taught in context of engineering applications

Improve:

### Assessment – Graded Events

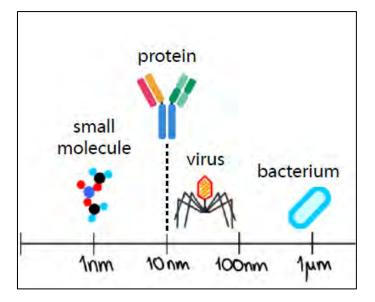
5 *Problem Sets@ 25 pts each	500	47.2%
6 *Quizes @ 200 pts each:	180	17.0%
1 *Paper	150	14.1%
1 *Presentation	50	4.7%
6 *Discussion	180	17.0%
Total:	1060	

\*Individual Points: 1060 100%

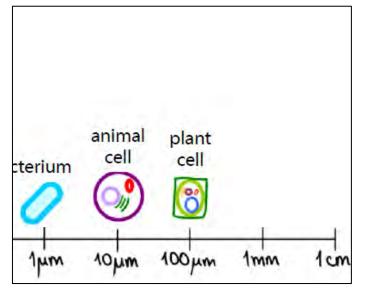
CH450

### CH350 Updates

### **Molecular Scale**



### **Cellular Scale**



- Finished with the block 1/2 and on-going with the block 3 of CH350.
- Cadets were introduced to both <u>quantitative</u> and <u>qualitative</u> perspectives of bioprocess engineering:
  - 1) Qualitative:
    - a. Cell types (animal vs. plant)
    - b. Central dogma of molecular biology
    - c. Metabolic pathways and regulations

### 2) Quantitative:

- a. Enzyme kinetics
- b. Mass transfer at immobilized enzymes
- c. Material balances on bioreactors
- Mathematica and excel were utilized to develop governing models for enzyme kinetic description and reactor designs.
- The block 1 was ended with WPR #1.
- The block 2 was ended with guest lectures and WPR #2.
- Curriculum memo to add CH350 permanently to REDBOOK.
- Assessment data to reinforce the memo.
- Dr. Biaglow/Dr. Yuk spearheading this effort JAN22.

Department of Chemistry and Life Science, Chemical Engineering Program 05APR22

### CH300 Updates

- Majority of block 1 of CH350 will be moved into the block 1 and 2 of CH300:
  - a) Molecular description of cells (animal vs. plant)
  - b) Central dogma of molecular biology (DNA, RNA, protein, etc)
  - c) Genetic engineering
  - d) Enzyme functionality and inhibitions
  - e) Metabolic pathways (respiration and digestion)
- Block 3 will be focused on biomedical engineering application.
- Suggesting that concepts related to enzyme kinetics (especially quantitative perspective) and metabolic pathways to be reintroduced in CH350.
- Curriculum memo to add CH300 permanently to REDBOOK.
- Assessment data to reinforce the memo.
- Dr. Biaglow/Dr. Yuk spearheading this effort JUN22.

### Bio-engineering Track/Sequence/minor

COA 1	COA 2	COA 3	
CH362-CH364-CH450	CH300-CH350-CH450	CH362-CH350-CH450	
Will require minimum of 2 additional faculty (if new sections added)	Will require additional faculty, minimum of 2. Ongoing AP searches. Faculty allocation will be	Will require additional faculty, minimum of 1. Ongoing AP searches. Faculty allocation will be	
Does not give Chem E cadets new electives.	the same in either COA, so use new faculty to develop new courses.	the same in either COA, so use new faculty to develop new courses.	
Existing course material/objectives are tied to ABET outcomes. These courses are highly optimized for chemical engineering content	Chem E cadets currently have no electives in DCLS. Continuing area of concern. Chem E cadets get 3x elective courses.	Chem E cadets currently have no electives in DCLS. Continuing area of concern. Chem E cadets get 2x elective courses.	
leading to high performance in these areas. Will require extensive re-configuring of <b>two</b> key program courses for a different audience without basis in assessment; max. 2 sections	Courses should be designed to meet student needs. Memo constructed in AY18 was reviewed by other programs and is ready to drop now. Even if proposed courses are not	Courses should be designed to meet student needs. Memo constructed in AY18 was reviewed by other programs and is ready to drop now.  Even if proposed courses are not optimal or we	
Less resource encumbered	optimal or we do not agree on content, get the	do not agree on content, get the courses in	
Course material/objectives tied to ABET courses in Redbook.  outcomes 1-7,8  Better for the department. Opportunity for	courses in Redbook.	Redbook.	
	Better for the department. Opportunity for		
Cannot dilute or alter content	interdisciplinary collaboration and team building.	interdisciplinary collaboration and team building.	



### **COA2 - Bioengineering Venn Diagram**

### Bioengineering Mod. & Design

**Biosensors** 

**Biomechanics** 

Biomaterials

Tissue engineering

Biochemical enzyme kinetics

Biochemical engineering

Legend
CH450
CH350
CH300

Upstream (biomolecular)
Bioreactors (cellular)

Downstream (purification)

**Bioprocess Engineering** 

**Bioimaging/instrumentation** 

**Biomechanics** 

Biotechnology

**Physiological Principles** 

Biomolecular/Cellular

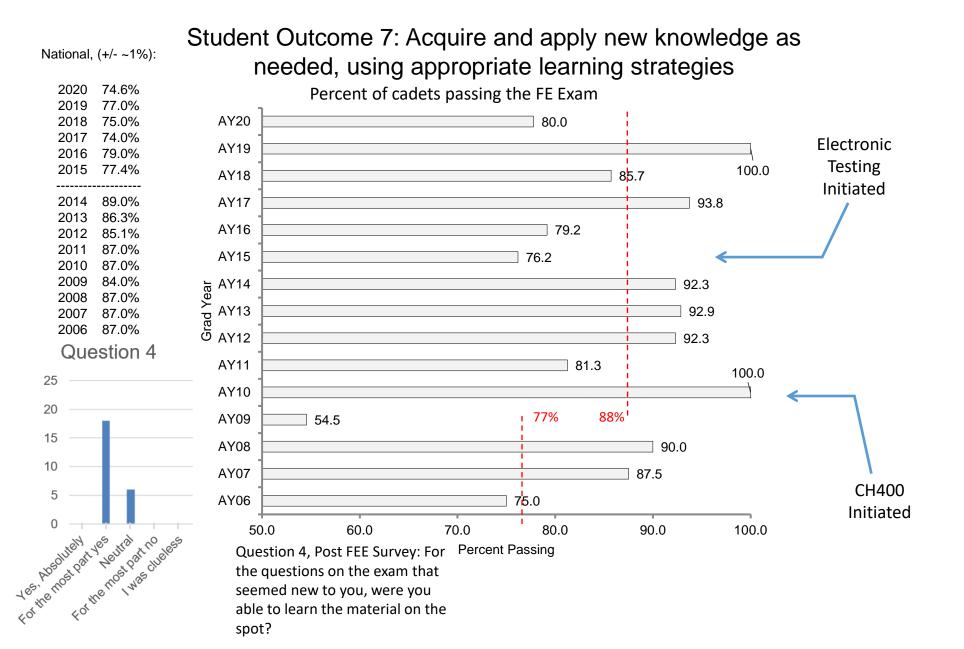
**Biomedical Engineering** 

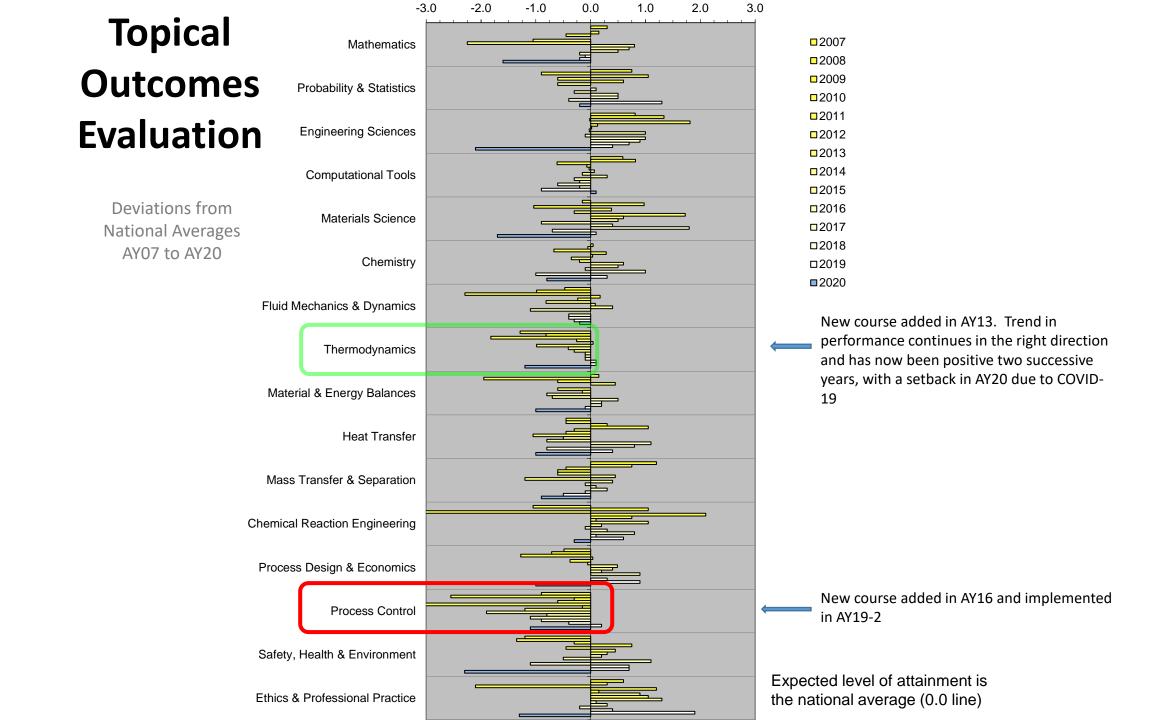
### Bio-engineering Track

- Select Bioengineering AP...ongoing Fall 2020
- Select Bioengineering T10
- QA/QC 3.0 ET credit for CH450
- Stand up new courses...CH350 & CH300
- Get to curriculum committee... Dr. Biaglow is the OIC for this effort
- Get courses in Redbook
- Get Bioengineering sequence approved
- ABET-compatibility (minor point)

# FEE Program Data

### As of 15APR Class of 2021: 75% passing FEE first time







# Program Mission



The mission of the chemical engineering program is to prepare commissioned leaders of character who are proficient in applying chemical and engineering principles to solve problems in a complex operational environment.

12/22/2023 40



# Engineering Technology Accreditation Commission



Accredited 1 October 2012 to present

Next Record Year: AY20

Next ABET Visit: Fall 2020 (AY21-1)

## Data used to support:

## **Assessment**

## **Program Assessment Cycle**

#### **Performance Indicators:**

1. FE Exam Results **Decision making** 2. Coursework Embedded Indicators **Evaluate** 3. Surveys (Cadet, Faculty, Adv. Board) 4. Course Grades Assess **Program-level** Collect **Program Loop**  $\Delta t$ **Course-level** Evaluate Time Course Assessments (Sect. III) **Embedded Indicators** Assess Adjust CH459 CH485 Collect  $\Delta t$ **Course Loops Evaluate** Etc. Assess Adjust 42

Collect

12/22/2023

## \*Student Outcomes

Identical to ABET 1-7 plus one additional outcome (8)

## On completion of the chemical engineering program, our graduates will be able to:

- 1. Identify, formulate, and solve complex engineering problems by applying principles of engineering, science, and mathematics.
- 2. Apply engineering design to produce solutions that meet specified needs with consideration of public health, safety, and welfare, as well as global, cultural, social, environmental, and economic factors.
- 3. Communicate effectively with a range of audiences.
- 4. Recognize ethical and professional responsibilities in engineering situations and make informed judgments, which must consider the impact of engineering solutions in global, economic, environmental, and societal contexts.
- 5. Function effectively on a team whose members together provide leadership, create a collaborative and inclusive environment, establish goals, plan tasks, and meet objectives.
- 6. Develop and conduct appropriate experimentation, analyze and interpret data, and use engineering judgment to draw conclusions.
- 7. Acquire and apply new knowledge as needed, using appropriate learning strategies.
- 8. Understand the chemical engineering curriculum, including chemistry, material and energy balances, safety and environmental factors, thermodynamics of physical and chemical equilibria, heat, mass, and momentum transfer, chemical reaction engineering, continuous and staged separation processes, process dynamics and control, modern experimental and computing techniques, and process design.

# Performance Indicators

- Coursework embedded indicators (example to be shown)
- Cadet end-of-semester surveys
- FE Exam results
- Cadet grades in program courses
- Advisory Board student outcome surveys
- Faculty student outcome surveys

## **Course Assessment Cycle**

#### Table 4-1. Outline of the course assessment process

#### **Description Assessment Recommendations**

SECTION I. COURSE DESCRIPTION: This section summarizes the course, exactly as it was taught in the most recently completed semester.	SECTION II. COURSE ASSESSMENT - This section provides data and analysis to answer the following questions:	SECTION III. RECOMMENDED CHANGES - All proposed changes to the course, in each of the specified areas. Recommendations should be based			
		on assessments from Section II.			
Redbook Description - List the current Redbook description.	Redbook Description - Does the Redbook description match what is taught in the course?	Redbook Description- For changes, include a cut and paste Redbook entry and use "track changes" when submitting recommendation.			
2. Enrollment - This AY and next AY (projected)	Enrollment - How does the student population compare from one year to the other? Assess effect of population on course.				
3. Course Content - Abbreviated list of subjects or lesson blocks covered in the course (not the syllabus).	Course Content - Is the course content appropriate?	3. Course Content - Recommended changes to course content.			
4. Course Objectives - List course objectives here.	Course Objectives - Were the course objectives achieved? Do the course objectives cover the body of knowledge appropriately? Do the course objectives lend themselves to assessment?	e body of objectives.			
	4a. Coverage - Indicate coverage of objectives by graded events.	4a. Coverage – Recommended changes to coverage of objectives by graded events.			
	4b. Performance - Indicate performance on course objectives.	<ol> <li>Performance - Recommendations to address shortcomings in performance on course objectives.</li> </ol>			
5. Survey Questions - List web-based and any other survey questions administered to cadets (If used).	5. Survey Questions - Are the survey questions appropriate?	<ol><li>Survey Questions - Recommended changes to survey questions.</li></ol>			
Examples include course questions, program questions, and USMA web-based survey questions.	Sa. Survey Results – Include analysis of Course-End Feedback or other surveys to include significant trends, suggestions or input that you believe should be incorporated into the course in the future.	Sa. Survey Results – Recommendations to address any shortcomings identified by survey results (if necessary).			
	5b. Survey Freeform Comments - (If used.) Results of any free-form comments from cadets about the course – summarize the most prevalent positive and negative comments.	5b. Survey Freeform Comments - Recommendations to address shortcomings identified from free-form comments, if necessary.			
6. Course GPA - List course GPA here. Include numbers from the last six terms.	6. Course QPA – Discuss any discernible trends or abrupt changes in course GPA over past several terms.	<ol><li>Course QPA – Recommendations to address any perceived problems.</li></ol>			
7. TEE Grade - List course TEE grade here from the last six terms.	7. TEE Grade – Discuss any discernible trends or abrupt changes in TEE grade over past several terms.	7. TEE Grade – Recommendations to address concerns with TEE grades.			
8. Course Processes	8. Course Processes	8. Course Process			
8a. Textbook - Title, author, and edition	8a. Textbook - Is the current textbook appropriate?	8a. Textbook - Recommended changes to textbook.			
8b. Lessons and Labs - List of lessons and labs in the course (syllabus).      8c. Summary of Graded Requirements - Number, type,	8b. Lessons and labs - Are the number of lessons and labs appropriate?     8c. Summary of Graded Requirements - Are the graded	8b. List of lessons and labs - Recommended changes to the number of lessons and labs.     8c. Summary of Graded Requirements - Recommended			
and weight of drill problems, Problem Sets, Special Problems, EDP's, Lab Reports, Writs, WPR's, TEE, and Instructor Grade (as applicable).	requirements appropriate?	changes to the graded requirements.			
8d. Areas of Special Emphasis - Any special topics not included in the Redbook description or program embedded indicators go here.	8d. Areas of Special Emphasis - Are the areas of special emphasis appropriate?	8d. Areas of Special Emphasis - Recommended changes to the areas of special emphasis.			
9. Contribution to Student Outcomes - List student outcomes here.	9. Contribution to the Student Outcomes - Does the course contribute to the student outcomes? How?	9. Contribution to Student Outcomes			
	9a. Coverage - Indicate coverage of objectives by graded events.  9b. Performance - Indicate performance on course objectives.	Qa. Coverage - Recommendations to address shortcomings in coverage of outcomes.     Performance - Recommendations to address problems in performance on student outcomes.			
10. Resources and Laboratories 10a. Laboratories - List laboratories lab projects used in the course.	10. Resources and Laboratories 10a. Laboratories - Was equipment available for desired experiments? Was equipment working?	10. Resources and Laboratories 10a. Laboratories - Recommendations to address any shortcomings in equipment.			
10b. Computer Labs - List computer labs used in the course.	10b. Computer Labs - Were adequate computing facilities available for the course?	10b. Computer Labs - Recommendations to improve computing facilities.			
10c. Physical Models & Demos - List physical models and demos used in the course.	10c. Physical Models & Demos - Were physical models and demos adequate? In good working order?	<ol> <li>Physical Models &amp; Demos - Recommendations for new demos or models, or to improve condition of existing models and demos.</li> </ol>			
10d. Technician Support - List technician support used in this course (wet lab or IT).	10d. Technician Support - Was technician support adequate?	10d. Technician Support - Recommendations to improve technician support.			
10e. Supplies - List any wet lab or computer supplies used in this course.	10e. Supplies - Were supplies adequate?	10e. Supplies - Recommend additional supplies for this course.			
10f. Additional Facilities - List any additional facilities used.	10f. Additional Facilities - Were the additional facilities adequate?	10f. Additional Facilities - Recommendations to address perceived shortcomings in additional facilities?			
10g. Unfunded Requests - List any unfunded requests	10g. Unfunded Requests - If provided, were the items	10g. Unfunded Requests - Recommendations for any			



Published guidance in CD Handbook

 Required for all chemical engineering courses used for assessment (not optional)

Similar system for courses outside department

Sustain and improve from last AY

## Example Schedule for Chemical Engineering Classes of 20xx and Beyond

Fall Term	Course	Credit	Spring	Course		Credit
4th CLASS		Hours	Term			Hours
MA103	Math. Modeling & Intro. Calculus	4.5	<b>MA104</b>	Calculus I		4.5
CH101	<b>General Chemistry I</b>	4.0	CH102	<b>General Chemistry II</b>		4.0
EN101	Composition	3.0	EN102	Literature		3.0
HI107	Western Civilization	3.0	HI108	Regional Studies in World History		3.0
IT105	Introduction to Computing & IT	3.0	PL100	General Psychology		3.0
PE11x	Combatives / Boxing / Movement	0.5	<b>MS100</b>	Introduction to Warfighting		1.5
			PE150	Fundamentals/Personal Fitness		1.5
3rd CLASS		<b>Total</b> 18.0			Total	20.5
<b>MA205</b>	Calculus II	4.0	CH362	Mass and Energy Balances		<i>3.5</i>
PH205	Physics I	4.0	MA364	Applied Engineering Math		<i>3.0</i>
Lx203	Foreign Language	4.0	PH206	Physics II		4.0
SS201	Economics	3.0	Lx204	Foreign Language		4.0
PY201	Philosophy	3.0	<b>SS202</b>	American Politics		3.0
MS200	<b>Fundamentals: Army Operations</b>	1.5	EV203	Physical Geography		3.0
	· -		PE 2xx	Lifetime Physical Activity		0.5
2nd CLASS		Total 19.5			Total	21.0
CH363	Separation Processes	<i>3.5</i>	CH364	Chemical Reaction Engineering		<i>3.5</i>
<i>EE301</i>	Fundamentals of Electrical Engineering	<i>3.5</i>	CH367	Introduction to Automatic Process Control		<i>3.0</i>
CH383	Organic Chemistry 1	<i>3.5</i>	MC312	Thermal-Fluid Systems 2		<i>3.0</i>
MC311	Thermal-Fluid Systems 1	3.5	Elective	Engineering Elective 1		<i>3.0</i>
PL300	Military Leadership	3.0	<b>SS307</b>	International Relations		3.0
<b>MA206</b>	Probability and Statistics	3.0	<b>MS300</b>	Platoon Operations		1.5
PE32x	Survival Swimming	0.5	PE360	Combat Applications		1.5
1st CLASS		<b>Total 20.5</b>			Total	18.5
CH459	Chemical Engineering Laboratory	<i>3.5</i>	CH402	Chemical Engineering Process Design 2		<i>3.5</i>
CH365	Chemical Engineering Thermodynamics	3.0	CH400	Chemical Engineering Prof. Practice		<i>3.0</i>
CH485	Heat & Mass Transfer	<i>3.5</i>	Elective	Engineering Elective 3		<i>3.0</i>
CH401	Chemical Engineering Process Design 1	3.5	HI302	History of the Military Art		3.0
Elective	Engineering Elective 2	<i>3.0</i>	LW403	Constitutional & Military Law		3.0
PE450	Army Fitness Development	1.5	<b>MX400</b>	Officership		3.0
	<u> </u>	<b>Total 18.0</b>		-	Total	18.5

Required Courses * (for classes 2020 and beyond)			
MA364	Engineering Mathematics		
CH362	Mass & Energy Balances		
CH363	Separation Processes		
CH364	Chemical Reaction Engineering		
CH367	Introduction to Automatic Process Control (XE472 2019 and previous)		
CH485	Heat & Mass Transfer		
CH459	Chemical Engineering Laboratory		
CH402	Chemical Engineering Process Design		
CH400	Professional Practice		
MC311	Thermal-Fluid Systems I		
MC312	Thermal-Fluid Systems II		
EE301	Fundamentals of Electrical Engineering		
MC300	Fundamentals of Engineering Mechanics & Design (Statics & Dynamics)		
CH365	Chemical Engineering Thermodynamics		
CH383	Organic Chemistry 1		

## **Chemical Engineering Student Outcomes**

On completion of the chemical engineering program, our graduates demonstrate an ability to

- 1. Identify, formulate, and solve complex engineering problems by applying principles of engineering, science, and mathematics.
- 2. Apply engineering design to produce solutions that meet specified needs with consideration of public health, safety, and welfare, as well as global, cultural, social, environmental, and economic factors.
- 3. Communicate effectively with a range of audiences.
- Recognize ethical and professional responsibilities in engineering situations and make informed judgments, which must consider the impact of engineering solutions in global, economic, environmental, and societal contexts.
- Function effectively on a team whose members together provide leadership, create a collaborative and inclusive environment, establish goals, plan tasks, and meet objectives.
- 6. Develop and conduct appropriate experimentation, analyze and interpret data, and use engineering judgment to draw conclusions.
- 7. Acquire and apply new knowledge as needed, using appropriate learning strategies.
- 8. Understand the chemical engineering curriculum, including:
  - · Chemistry,
  - · Material and energy balances,
  - · Safety and environmental factors,
  - · Thermodynamics of physical and chemical equilibria,
  - · Heat, mass, and momentum transfer,
  - Chemical reaction engineering,
  - Continuous and staged separation processes
  - · Process dynamics and control,
  - · Modern experimental and computing techniques, and
  - · Process design.

# **Chemical Engineering Program Objectives**

During a career as commissioned officers in the United States Army and beyond, program graduates:

- Contribute to the solution of infrastructure or operational problems in a complex operational environment.
- Succeed in graduate school or other advanced study programs.
- Advance their careers through clear and precise technical communication.
- Demonstrate effective leadership and chemical engineering expertise.

# Chemical Engineering Program Population

