CH365 CHEMICAL ENGINEERING THERMODYNAMICS

Lesson 4: Internal Energy, Energy Balances, & State Functions

Read: Sections 2.1-2.5, pp. 24-34

Problems: 2.1, 2.4

Objectives:

- 1. Describe Joule's experiments.
- 2. Describe the relationship between internal energy and heat and work.
- 3. State the first law of thermodynamics in word and equation form.
- 4. Use concepts of thermodynamic state and state functions to calculate heat, work, and internal energy associated with changes of state (see Examples 2.3, and 2.4).

Definitions:

Internal energy, 1st Law of Thermodynamics, system, surroundings, extensive property, intensive property, thermodynamic state, state function.

Cadet Notes: