
11th i-CoMSE Workshop: Mesoscale Particle-Based Modeling

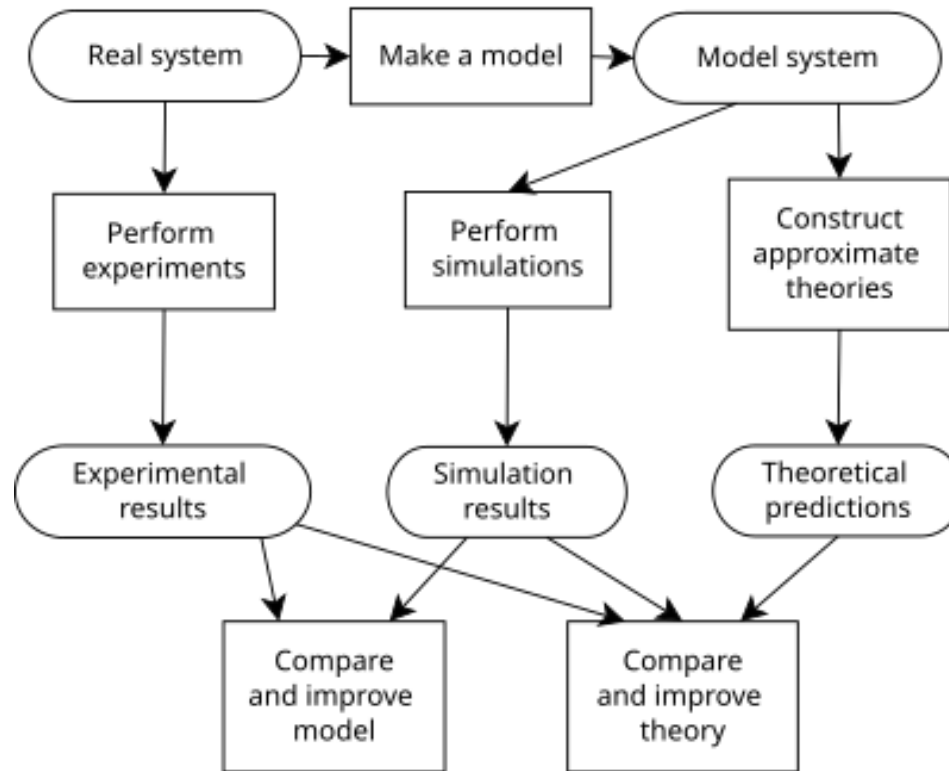
**Mississippi State University
July 21–25, 2025**

Session 2: Top-down coarse graining



Top-down coarse graining

- **Objective:** use macroscopic/mesoscale information to parameterize a simplistic model or build simplistic model to match theory

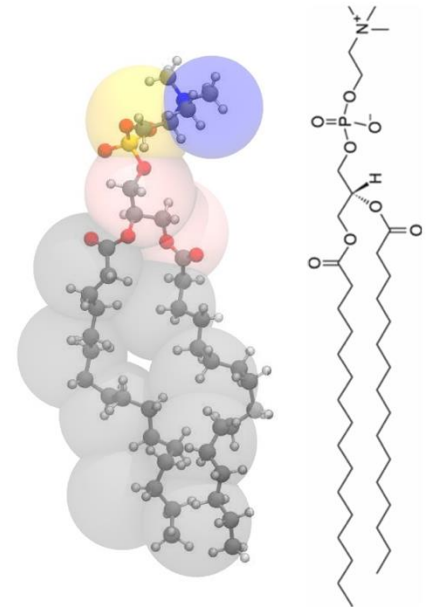
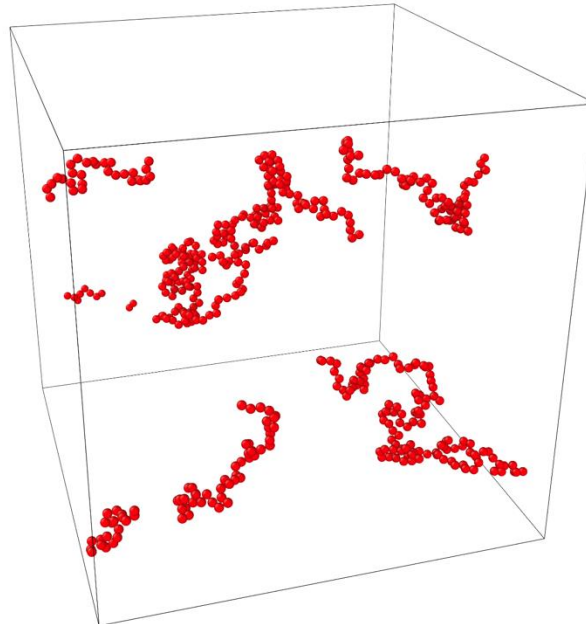
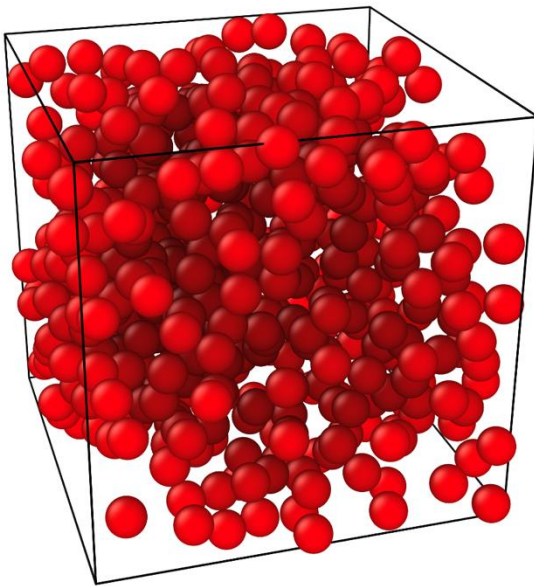


Top-down coarse graining

- **Objective:** use macroscopic/mesoscale information to parameterize a simplistic model or build simplistic model to match theory
- Macroscopic/mesoscale information:
 - Experimental measurements
 - Thermodynamic quantities (densities, concentrations, liquid–liquid partitioning, ..., critical temperatures, interfacial tensions, radius of gyration, system free energies, ..)
- Use coarse-grained “beads” to express the general character of molecular structure, de-emphasizing specific chemical features
- Remove solvent

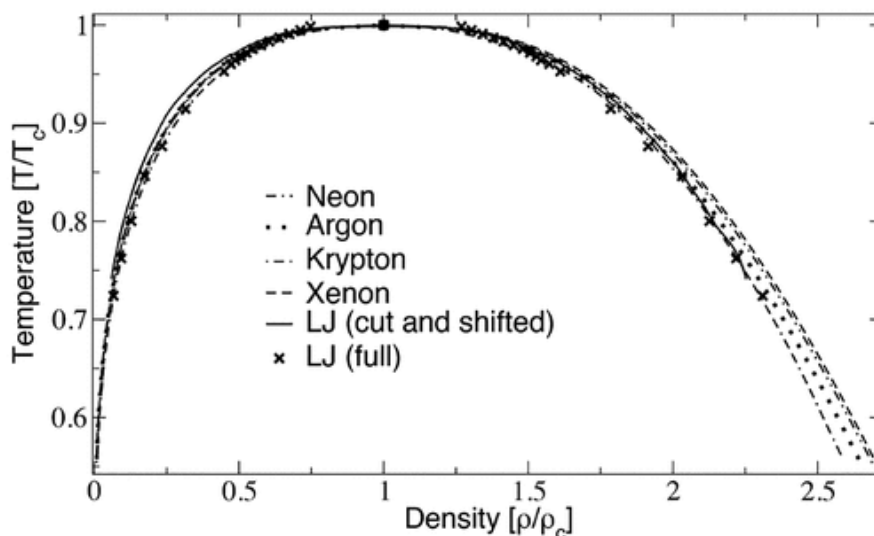
Top-down coarse graining

- Examples of top-down models:
 - Martini force field for lipids and biomolecules (non-bonded interactions based on experimental partitioning free energies, four-to-one mapping)
 - Kremer-Grest and Bead-Spring models for polymers
 - LJ for fluids, ...



Toy Models: Generic Top-down models

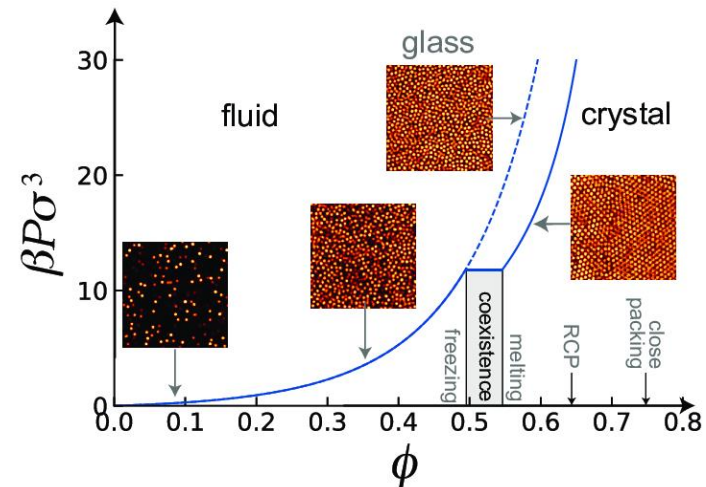
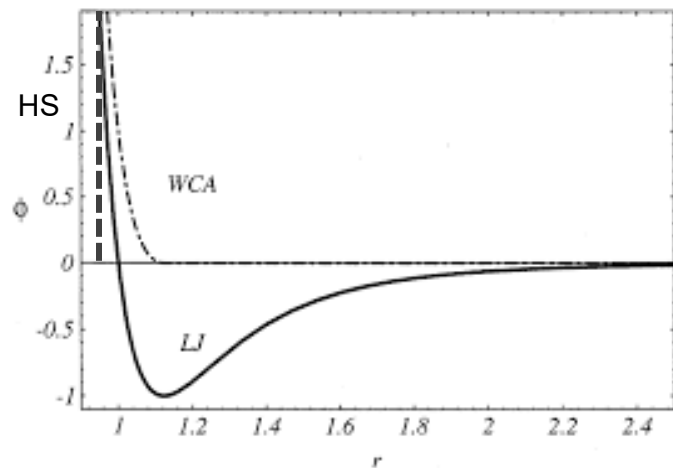
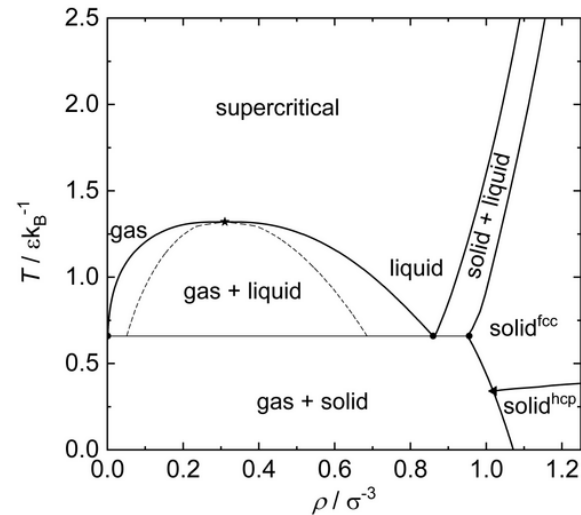
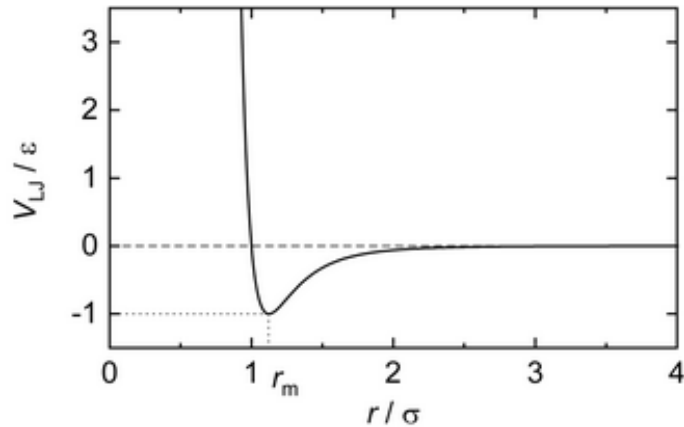
- Phenomenological with relatively low resolution with few/no chemical detail
- Do not describe any particular system
- Used for investigation of generic consequences of basic physical principles (systematic variation of parameters)
- Simple potentials with relatively few parameters



100 Years of the Lennard-Jones Potential
Peter Schwerdtfeger and David J. Wales
Journal of Chemical Theory and
Computation 2024 20 (9), 3379-3405
DOI: 10.1021/acs.jctc.4c00135

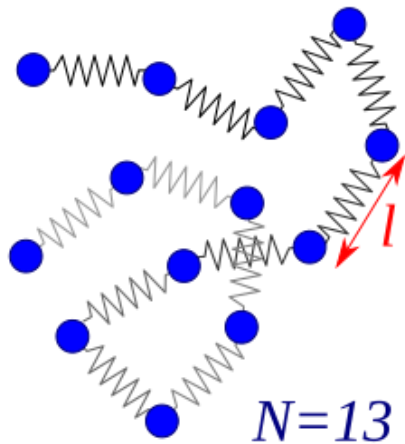
Toy Models: Generic Top-down models

Simple Fluids (Lennard Jones, Hard Spheres/WCA,...)

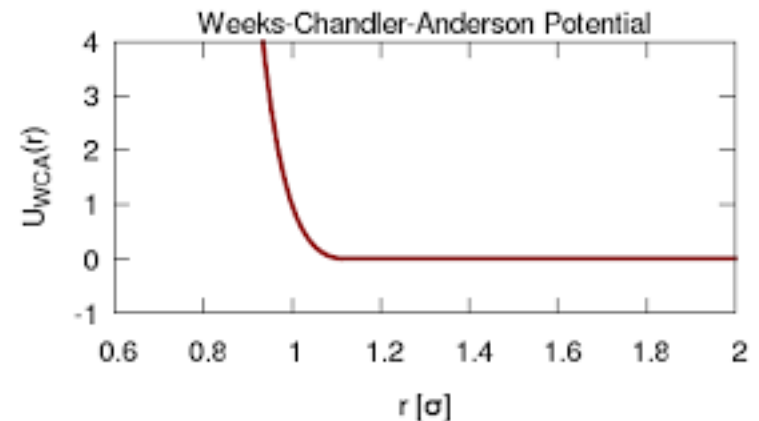


Toy Models: Generic Top-down models

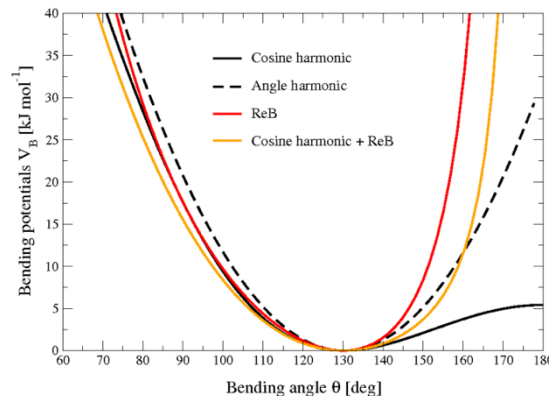
Bead-Spring models (Kremer-Grest,...)



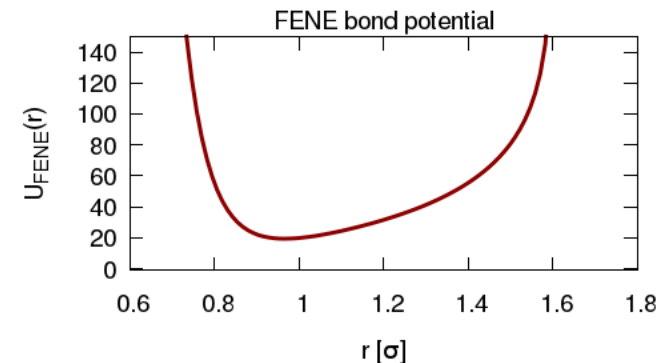
Non-bonded pair interactions:



Sometimes also: angles, dihedrals



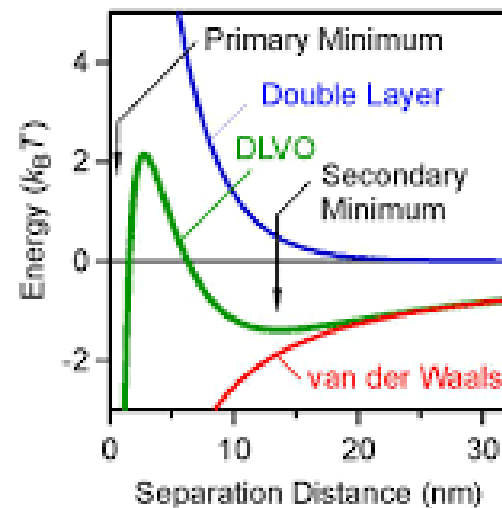
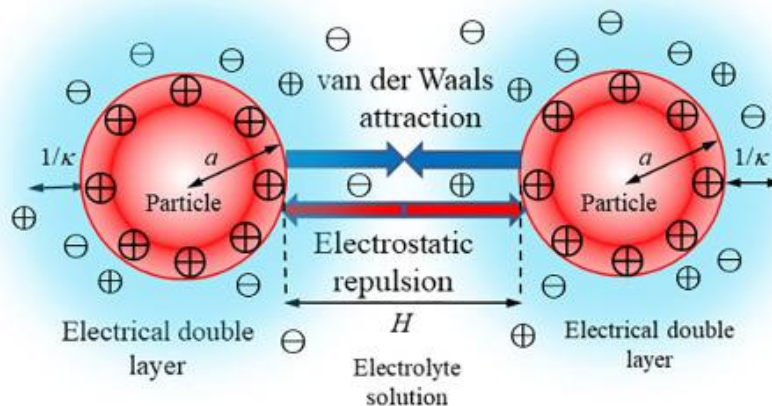
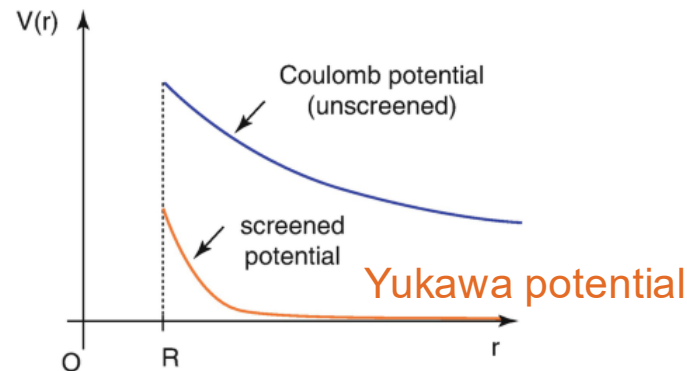
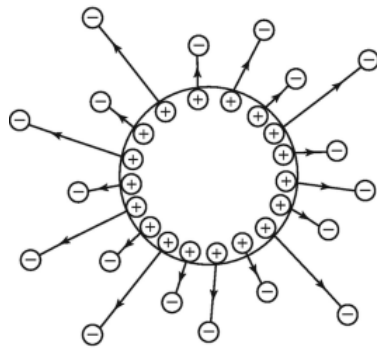
Bonded interactions:



$$E = -0.5KR_0^2 \ln \left[1 - \left(\frac{r}{R_0} \right)^2 \right] + 4\epsilon \left[\left(\frac{\sigma}{r} \right)^{12} - \left(\frac{\sigma}{r} \right)^6 \right] + \epsilon$$

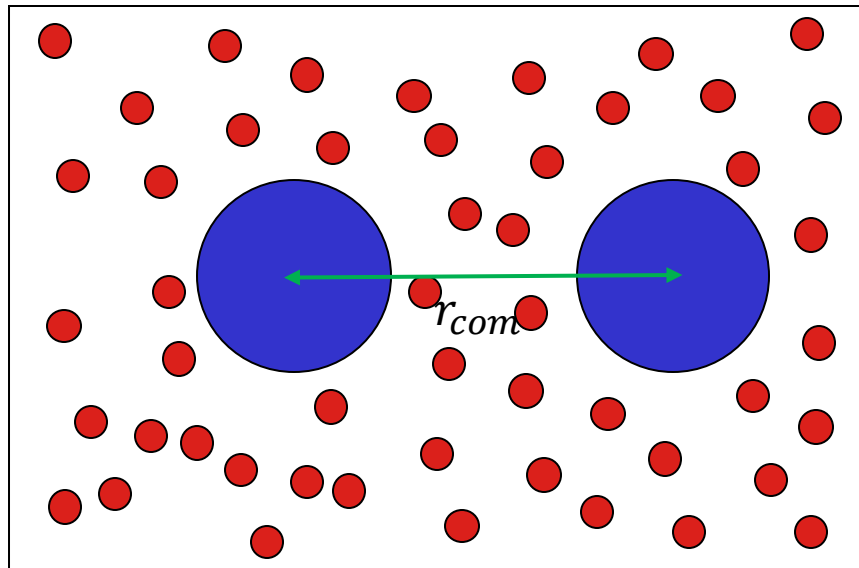
Other Interactions

Screened electrostatics, Van der Waals, DLVO theory



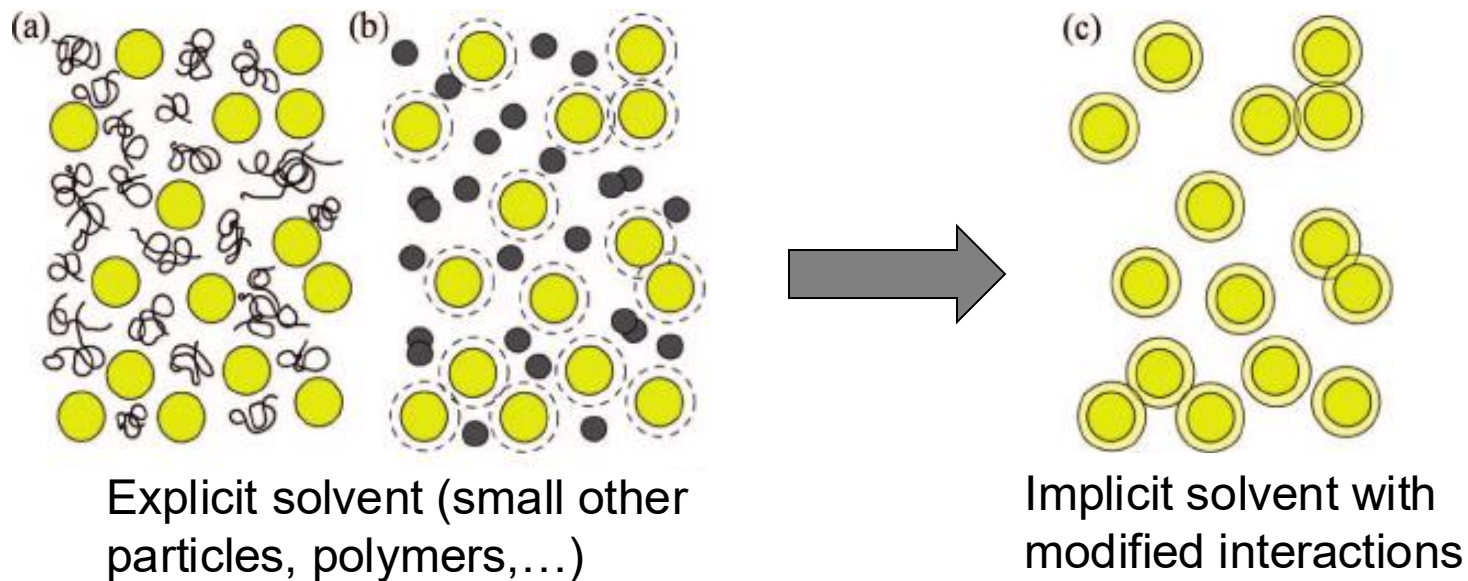
Example: NP in Solvent

- CG nanoparticles (NP) or colloids in a bath of N solvent (or depletant) particles
- Typical experimental system: each 10 nm diameter nanoparticle is surrounded by $> 10^6$ solvent molecules under dilute conditions (0.1 volume fraction)



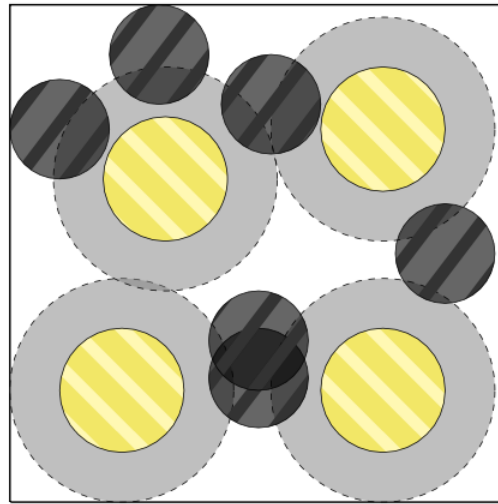
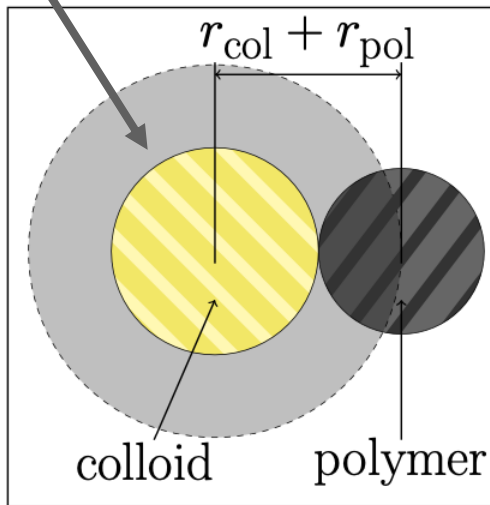
Solvent coarse graining

Asakura–Oosawa theory

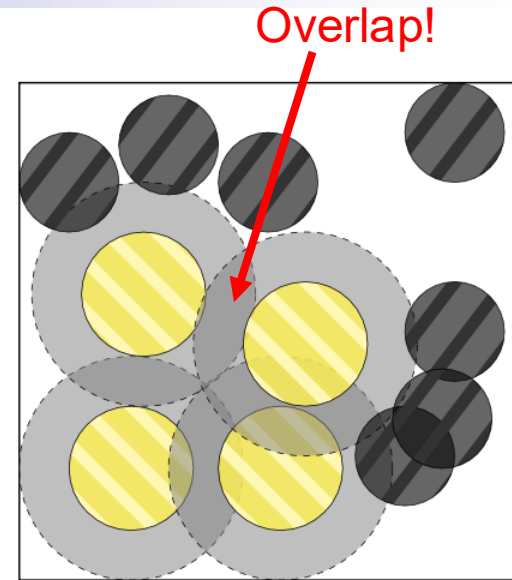


Asakura-Oosawa Theory

Depletion "zone"



Dispersed colloids



Aggregated colloids

Small depletants (solvent, polymers,..) can cause effective attraction between NP/Colloids due to free volume and entropy arguments.

Asakura, Sho; Oosawa, F. (1 January 1954). "On Interaction between Two Bodies Immersed in a Solution of Macromolecules". *The Journal of Chemical Physics*. 22 (7): 1255. doi:10.1063/1.1740347.

Asakura, Sho; Oosawa, F. (1958). "Interaction between Particles Suspended in Solutions of Macromolecules". *Journal of Polymer Science*. 33 (126): 183–192. doi:10.1002/pol.1958.1203312618.

Asakura–Oosawa Theory

Assumptions:

- low concentrations of macromolecules
- homogeneous, uniform density
- Pairwise interactions

Force between two particles:

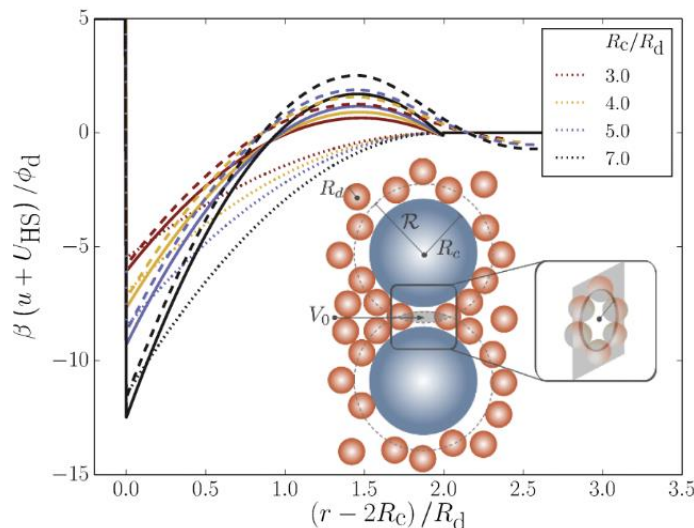
$$F = \Pi \Delta S_{\text{exclusion}}$$

↑
osmotic pressure

↖
area of the circular cross section
of the over-lapping region

→ Integrate → Potential Energy U

Solution is exact when the depletants are small (size ratio $q = \sigma_p/\sigma_c < 0.1547$)



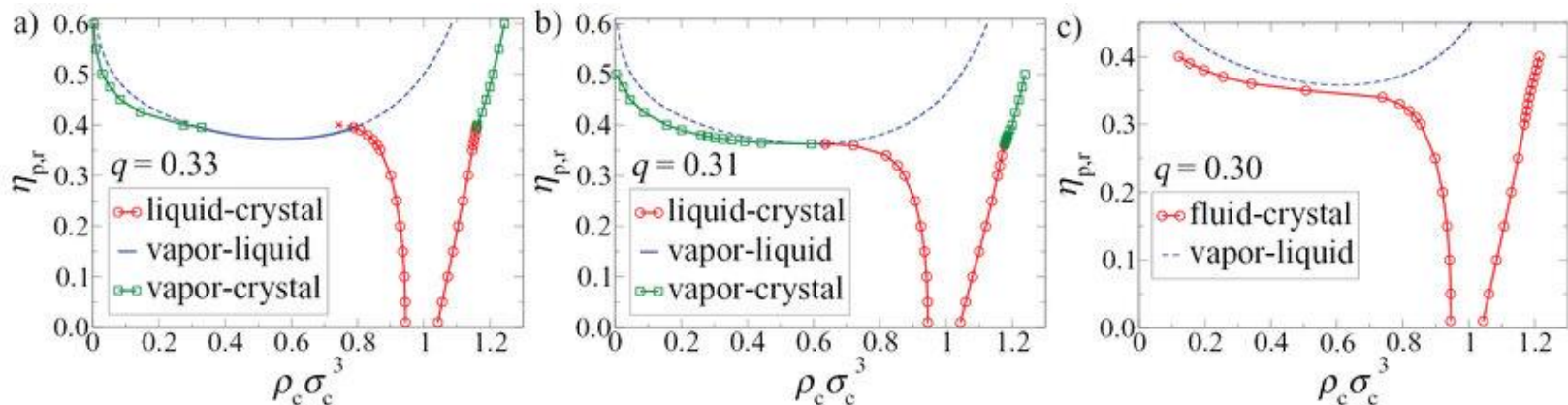
$$\beta\phi_{AO}(r) = \begin{cases} \infty & (r < \sigma_c) \\ -\frac{\pi}{6}\sigma_p^3\rho_p\left(1 + \frac{1}{q}\right)^3\left(1 - \frac{3r}{2\sigma_c(1+q)} + \frac{r^3}{2\sigma_c^3(1+q)^3}\right) & (\sigma_c \leq r \leq \sigma_c + \sigma_p) \\ 0 & (\text{otherwise}) \end{cases}$$

σ_c = colloid size , σ_p = polymer/depletant size
polymer fugacity $z_p = \rho_p$, number density ideal polymers"

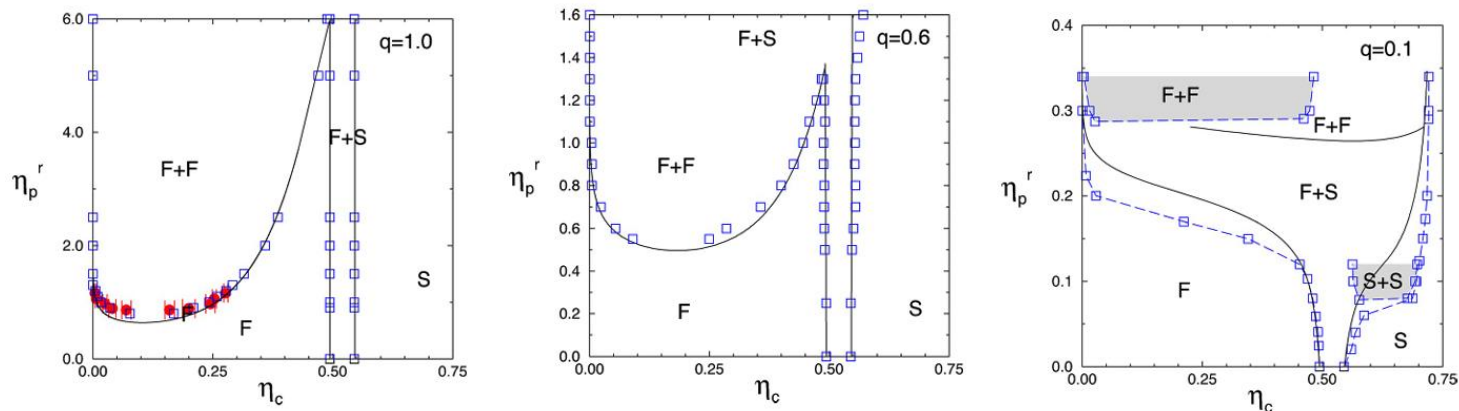
Polymer reservoir packing fraction $\eta_p^r = \frac{\pi\sigma_p^3 z_p}{6}$

Asakura–Oosawa Theory

Free Volume Theory, measure density functionals, simulations:



Mortazavifar, M., & Oettel, M. (2016). A fundamental measure density functional for fluid and crystal phases of the Asakura–Oosawa model. *Journal of Physics: Condensed Matter*, 28(24), 244018.

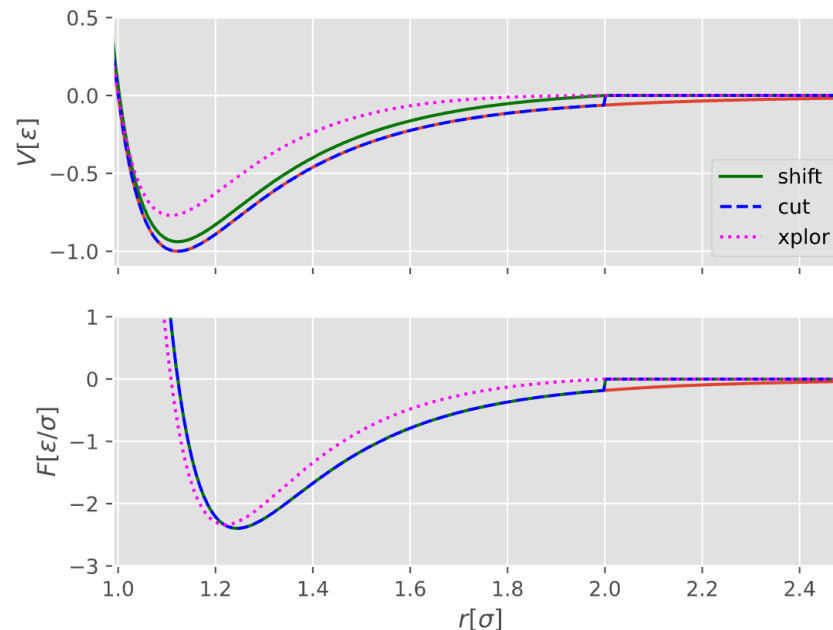


Dijkstra, M., van Roij, R., Roth, R., & Fortini, A. (2006). Effect of many-body interactions on the bulk and interfacial phase behavior of a model colloid-polymer mixture. *Physical Review E—Statistical, Nonlinear, and Soft Matter Physics*, 73(4), 041404.

Exercise

Objective:

- Plot potentials and forces
- Set up simple HOOMD simulation with two particles
- Run simple AO model simulation



Questions / Common Pitfalls

- How **transferable** and how **representative** is a model?
- Ignoring reduced units
- How many “ad hoc” or other empirical fixes does a model need to describe a particular system?
- Matching experiments is difficult!
- Not comparing with existing models, theory, scaling predictions
- What are the **limitations** of the model?