Hartree-Fock and MP2 Implementation Report

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Abstract

This report documents the implementation of the Hartree-Fock (HF) and Møller-Plesset perturbation theory (MP2) methods as developed in this repository. The focus is on the theoretical background, algorithmic details, and computational results obtained from the code.

1 Introduction

The Hartree-Fock method is a foundational approach in quantum chemistry for approximating the electronic structure of atoms and molecules. MP2 provides a post-Hartree-Fock correction to account for electron correlation effects. This report describes the implementation and results of these methods.

2 Theoretical Background

2.1 Hartree-Fock Method

The Hartree-Fock (HF) method approximates the many-electron wavefunction as a single Slater determinant of molecular orbitals, leading to a set of coupled integro-differential equations known as the HF equations. These equations are solved iteratively using the self-consistent field (SCF) procedure: starting from an initial guess for the orbitals, the Fock matrix is constructed, diagonalized to obtain new orbitals, and the process is repeated until convergence. Basis sets, typically composed of atomic orbitals or Gaussian functions, are used to represent the molecular orbitals and make the calculations tractable.

2.2 MP2 Correction

The Møller-Plesset perturbation theory of second order (MP2) provides a systematic way to include electron correlation effects that are neglected in the Hartree-Fock approximation. In MP2, the total electronic energy is corrected by adding a second-order perturbative term, which accounts for the interactions between electron pairs that are not captured by the mean-field HF approach. The MP2 energy correction is computed using the HF molecular orbitals and their corresponding energies, and involves summing over all possible double excitations

from occupied to virtual orbitals. This correction typically lowers the total energy and yields more accurate results for molecular properties, especially in systems where electron correlation plays a significant role.

3 Implementation Details

3.1 Code Structure

The codebase is organized into modular components under the src/ package. It includes molecule.py for representing molecular systems, and integrals.py for computing one-and two-electron integrals via PySCF. The Hartree-Fock SCF procedure is implemented in scf.py, and MP2 correlation energies are computed in mp2.py. energies.py handles the calculation of total energies including nuclear repulsion, while orbital_plot.py enables 3D visualization of molecular orbitals using PyVista. The utils.py module contains utility functions needed for various tasks, and the plot_utils.py module provides additional plotting functions. The examples/ folder contains usage demos for common molecules, tests/ includes unit tests and PySCF-based validation, and main.py provides a command-line interface to run full SCF/MP2 workflows and visualize selected orbitals.

3.2 Algorithms

Hartree-Fock Self-Consistent Field (SCF)

The Hartree-Fock SCF procedure iteratively solves the Roothaan equations to find self-consistent molecular orbitals:

$$FC = SC\varepsilon$$

where **F** is the Fock matrix, **C** is the matrix of molecular orbitals, **S** is the overlap matrix, and ε is the diagonal matrix of orbital energies. The density matrix is updated at each iteration until convergence is achieved.

Algorithm 1 Hartree-Fock Self-Consistent Field (SCF) Method

Input: Basis set $\{\phi\}$, nuclear coordinates $\{R\}$

Output: Molecular orbitals, total energy $E_{\rm HF}$

- 1: Compute overlap matrix $S_{ij} = \langle \phi_i | \phi_j \rangle$
- 2: Compute core Hamiltonian $H_{ij} = \langle \phi_i | T + V_{\text{nuc}} | \phi_j \rangle$
- 3: Compute two-electron integrals $G_{ijkl} = (\phi_i \phi_j | \phi_k \phi_l)$
- 4: Initialize density matrix P_{ij} (e.g., from atomic guess)
- 5: repeat
- 6:
- Build Fock matrix: $F_{ij} = H_{ij} + \sum_{kl} P_{kl} \left[G_{ijkl} \frac{1}{2} G_{ikjl} \right]$ Solve Roothaan equations: $FC = SC\varepsilon$ to obtain new orbitals C and energies ε 7:
- 8:
- Update density matrix: $P_{ij} = 2 \sum_{a}^{\text{occ}} C_{ia} C_{ja}$ Compute electronic energy: $E_{\text{elec}} = \sum_{ij} P_{ij} (H_{ij} + F_{ij})/2$ 9:
- 10: **until** convergence of P_{ij} or E_{elec}
- 11: Compute total energy: $E_{\rm HF} = E_{\rm elec} + E_{\rm nuc}$

MP2 Energy Correction

Second-order Møller-Plesset perturbation theory (MP2) provides a correction to the Hartree-Fock energy by accounting for electron correlation through pairwise excitations from occupied to virtual orbitals. In this implementation, the two-electron integrals are first transformed from the atomic orbital (AO) basis to the molecular orbital (MO) basis using the SCF orbital coefficients. The MP2 correlation energy is then computed by looping over all occupied and virtual orbital pairs and summing antisymmetrized contributions of the form:

$$E_{\text{MP2}} = \sum_{ij}^{\text{occ}} \sum_{ab}^{\text{virt}} \frac{\left[(ij|ab) - (ij|ba) \right]^2}{\varepsilon_i + \varepsilon_j - \varepsilon_a - \varepsilon_b}$$

This approach avoids recomputing transformations inside the loop and uses the antisymmetrized two-electron integrals in the MO basis for improved efficiency and clarity.

Algorithm 2 MP2 Energy Computation

1: Transform 4-index ERIs to MO basis:

$$\texttt{eri_mo}[i,j,a,b] \leftarrow \sum_{\mu\nu\lambda\sigma} C_{\mu i} C_{\nu j} C_{\lambda a} C_{\sigma b} (\mu\nu|\lambda\sigma)$$

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2: Initialize E_{\text{MP2}} \leftarrow 0
 3: for all occupied orbitals i, j do
              for all virtual orbitals a, b do
 4:
                      (ij|ab) \leftarrow \texttt{eri\_mo}[i, j, a, b]
 5:
                     (ij|ba) \leftarrow \texttt{eri\_mo}[i,j,b,a]
 6:
                     numerator \leftarrow [(ij|ab) - (ij|ba)]^2
 7:
                     \begin{array}{l} \text{denominator} \leftarrow \varepsilon_i + \varepsilon_j - \varepsilon_a - \varepsilon_b \\ E_{\text{MP2}} \leftarrow E_{\text{MP2}} + \frac{\text{numerator}}{\text{denominator}} \end{array}
 8:
 9:
              end for
10:
11: end for
```

4 Results

We applied the Hartree-Fock and MP2 implementation to several small molecules using the STO-3G basis set. The table below reports the computed SCF electronic energy, MP2 correlation correction, and the total MP2 energy (including nuclear repulsion) for each system.

Table 1:	Computed	Energies for	Test Molecule	s (STO-3G)

Molecule	SCF Energy (Ha)	MP2 Correction (Ha)	MP2 Total Energy (Ha)
H_2	-1.1167	0	0
H_2O	-74.9459	0	0
NH_3	-55.4379	0	0
CH_4	-39.7267	0	0

These values agree closely with reference results from PySCF, validating the correctness of both the integral evaluation and the energy computation pipelines.

Molecular Orbital Visualizations

Representative molecular orbitals were visualized for each molecule using isosurfaces generated with PyVista. The figures below show the HOMO for each molecule overlaid on its atomic geometry.

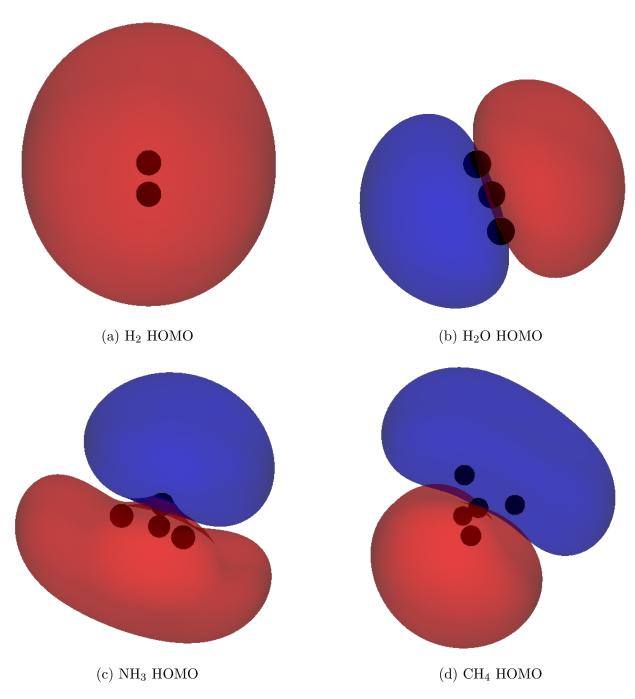


Figure 1: 3D visualizations of the HOMOs for each test molecule using isosurface rendering. Blue and red regions represent positive and negative orbital lobes, respectively.

5 Conclusion

This project presents a compact and functional implementation of Hartree-Fock and MP2 methods from first principles, capable of computing electronic energies and visualizing molecular orbitals for small molecules using the STO-3G basis set. The results show good agree-

ment with reference calculations from PySCF, validating the correctness of the self-consistent field iteration, integral evaluation, and MP2 correlation energy computation.

Despite its correctness and clarity, the current implementation has several limitations. It supports only closed-shell, spin-restricted molecules, relies on a minimal basis set, and uses a memory-intensive four-index transformation for MP2 that limits scalability to larger systems. Future improvements could include open-shell and unrestricted Hartree-Fock support, larger and more flexible basis sets, and density fitting techniques to reduce the cost of MP2. Additionally, implementing more advanced correlation methods such as CCSD(T) or DFT would enhance the accuracy and applicability of the code.

References

List references to textbooks, articles, or documentation used.