

# Competishun

52/6, Opposite Metro Mas Hospital, Shipra Path, Mansarovar

**Date:** 07/01/2024

**Time:** 3 hours

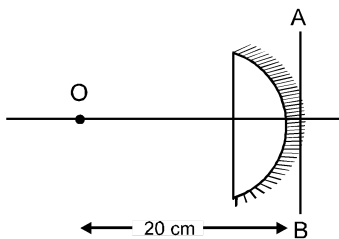
**Max. Marks: 300**

## MFST-11 (23-24) & UT-2\_MT-3

# Physics

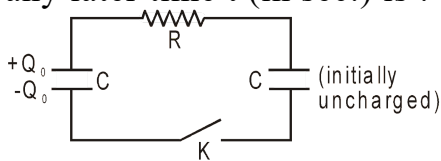
### Single Choice Question

- Q1** A point object is placed at a distance of 20 cm from a thin plane convex lens of focal length 15 cm ( $n = 1.5$ ). Now the curved surface is silvered. The image will be formed at :

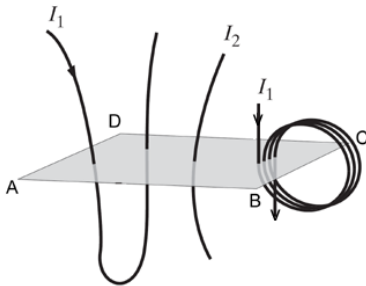


- a) 60 cm left of AB                      b) 30 cm left of AB  
c)  $\frac{20}{7}$  cm left of AB                      d) 60 cm right of AB
- Q2** The following data are given for a crown glass prism ;  
refractive index for blue light  $n_b = 1.521$   
refractive index for red light  $n_r = 1.510$   
refractive index for yellow light  $n_y = 1.550$   
Dispersive power of a parallel glass slab made of the same material is :  
a) 0.01                      b) 0.02                      c) 0.03                      d) 0
- Q3** Two coaxial dipoles of dipole moments  $P_1$  &  $P_2$  are separated by a distance ‘ r ’. The magnitude of electric force on  $P_1$  due to  $P_2$  is: ( $k = 1/4 \pi \epsilon_0$ )  
a)  $\frac{2 k P_1 P_2}{r^3}$                       b)  $\frac{2 k P_1 P_2}{r^2}$                       c)  $\frac{6 k P_1 P_2}{r^4}$                       d) zero
- Q4** Two racing cars of masses  $m_1$  and  $m_2$  are moving in circles of radii  $r$  and  $2r$  respectively and their angular speeds are equal. The ratio of the time taken by cars to complete one revolution is :  
a)  $m_1 : m_2$                       b)  $1 : 2$                       c)  $1 : 1$                       d)  $m_1 : 2m_2$

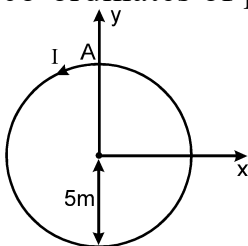
- Q5** The shown circuit comprises of two identical capacitors of capacitance  $C$  Farad and resistor of resistance  $R \Omega$ . The key  $K$  is initially open. At time  $t = 0$  the charge on left capacitor is  $Q_0$  Coloumbs and the right capacitor is uncharged as shown. The key  $K$  is closed at time  $t = 0$ . Then the magnitude of current in amperes through the resistor at any later time  $t$  (in sec.) is :



- a)  $\frac{Q_0}{RC} e^{-\frac{t}{RC}}$       b)  $\frac{Q_0}{RC} e^{-\frac{2t}{RC}}$       c)  $\frac{Q_0}{2RC} e^{-\frac{2t}{RC}}$       d)  $\frac{Q_0}{2RC} e^{-\frac{t}{RC}}$
- Q6** Three distinct current carrying wires intersect a finite rectangular plane ABCD. The current in left wire and the loop is  $I_1$ . The direction of current in left most wire and right most loop is downwards as shown in figure. The current  $I_2$  through middle wire is adjusted so that the path integral of the total magnetic field along the perimeter of the rectangle is zero, that is,  $\oint_{ABCD} \vec{B} \cdot d\vec{\ell} = 0$ . Then the current  $I_2$  is -

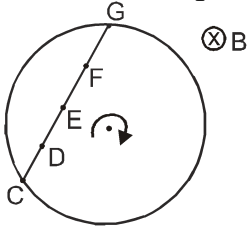


- a)  $2 I_1$  and upwards      b)  $2 I_1$  and downwards  
c)  $4 I_1$  and upwards      d)  $3 I_1$  and downwards
- Q7** A ring of radius 5 m is lying in the x-y plane and is carrying current of 1 A in anti-clockwise sense. If a uniform magnetic field  $\vec{B} = 3\hat{i} + 4\hat{j}$  is switched on, then the co-ordinates of point about which the loop will lift up is:

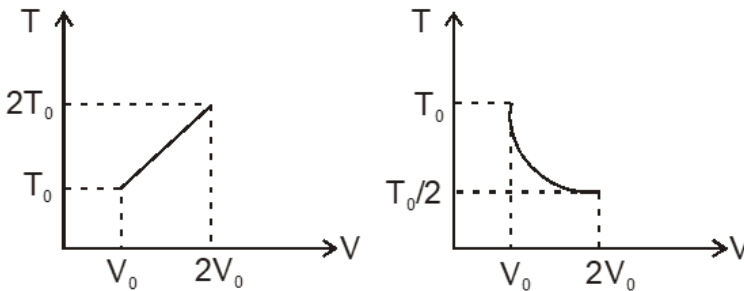


- a) (3, 4)      b) (4, 3)      c) (3, 0)      d) (0, 3)

- Q8** A conducting disc rotates about its axis (passing through centre and perpendicular to the plane of disc) in a uniform magnetic field  $B$  and magnetic field is perpendicular to plane of disc. There are certain points marked on a chord joining two points  $C$  and  $G$  on the periphery of the disc. Point  $E$  is the midpoint of the chord. Points  $D$  and  $F$  are the midpoints of segments  $CE$  and  $EG$  respectively. The emf induced is maximum between the points: (Do not consider the emf due to centrifugal forces)

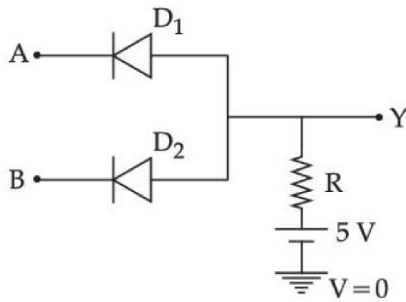


- a) C and D                      b) C and E                      c) C and G                      d) D and F
- Q9** A monoatomic ideal gas undergoes a process given by  $2dU + 3dW = 0$ , then the process is:  
a) isobaric                      b) adiabatic                      c) isothermal                      d) none of these
- Q10** For two thermodynamic process temperature and volume diagram are given. In first process, it is a straight line having initial and final coordinates as  $(V_0, T_0)$  and  $(2V_0, 2T_0)$ , where as in second process it is a rectangular hyperbola having initial and final coordinates  $(V_0, T_0)$  and  $(2V_0, T_0/2)$ . Then ratio of work done in the two processes must be

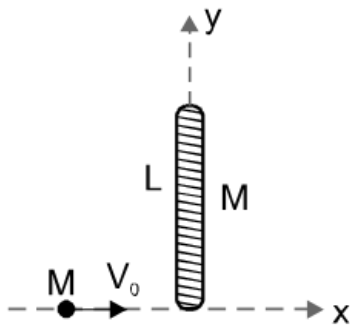


- a) 1 : 2                      b) 2 : 1                      c) 1 : 1                      d) None of these
- Q11** A photon is incident upon a hydrogen atom ejects an electron with a kinetic energy 10.7 eV. If the ejected electron was in the first excited state. Energy of photon incident nearly will be :  
a) 14.5 eV                      b) 20.9 eV                      c) 24.3 eV                      d) 14.1 eV

- Q12** In the circuit, the logical value of  $A = 1$  or  $B = 1$  when potential at A or B is 5V and the logical value of  $A = 0$  or  $B = 0$  when potential at A or B is 0 V. The truth table of the given circuit will be :

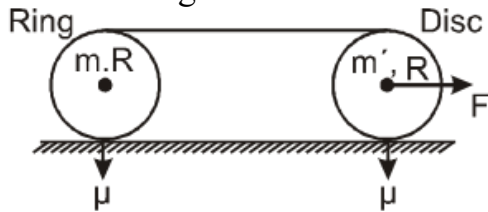


- | <p><b>a)</b></p> <table border="1" style="margin-left: 20px;"> <thead> <tr> <th>A</th> <th>B</th> <th>Y</th> </tr> </thead> <tbody> <tr><td>0</td><td>0</td><td>0</td></tr> <tr><td>1</td><td>0</td><td>0</td></tr> <tr><td>0</td><td>1</td><td>0</td></tr> <tr><td>1</td><td>1</td><td>1</td></tr> </tbody> </table> | A | B | Y | 0 | 0 | 0 | 1 | 0 | 0 | 0 | 1 | 0 | 1 | 1 | 1 | <p><b>b)</b></p> <table border="1" style="margin-left: 20px;"> <thead> <tr> <th>A</th> <th>B</th> <th>Y</th> </tr> </thead> <tbody> <tr><td>0</td><td>0</td><td>0</td></tr> <tr><td>1</td><td>0</td><td>1</td></tr> <tr><td>0</td><td>1</td><td>1</td></tr> <tr><td>1</td><td>1</td><td>1</td></tr> </tbody> </table> | A | B | Y | 0 | 0 | 0 | 1 | 0 | 1 | 0 | 1 | 1 | 1 | 1 | 1 | <p><b>c)</b></p> <table border="1" style="margin-left: 20px;"> <thead> <tr> <th>A</th> <th>B</th> <th>Y</th> </tr> </thead> <tbody> <tr><td>0</td><td>0</td><td>0</td></tr> <tr><td>1</td><td>0</td><td>0</td></tr> <tr><td>0</td><td>1</td><td>0</td></tr> <tr><td>1</td><td>1</td><td>0</td></tr> </tbody> </table> | A | B | Y | 0 | 0 | 0 | 1 | 0 | 0 | 0 | 1 | 0 | 1 | 1 | 0 | <p><b>d)</b></p> <table border="1" style="margin-left: 20px;"> <thead> <tr> <th>A</th> <th>B</th> <th>Y</th> </tr> </thead> <tbody> <tr><td>0</td><td>0</td><td>1</td></tr> <tr><td>1</td><td>0</td><td>1</td></tr> <tr><td>0</td><td>1</td><td>1</td></tr> <tr><td>1</td><td>1</td><td>0</td></tr> </tbody> </table> | A | B | Y | 0 | 0 | 1 | 1 | 0 | 1 | 0 | 1 | 1 | 1 | 1 | 0 |
|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|---|
| A   | B | Y |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 0   | 0 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 1   | 0 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 0   | 1 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 1   | 1 | 1 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| A   | B | Y |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 0   | 0 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 1   | 0 | 1 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 0   | 1 | 1 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 1   | 1 | 1 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| A   | B | Y |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 0   | 0 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 1   | 0 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 0   | 1 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 1   | 1 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| A   | B | Y |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 0   | 0 | 1 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 1   | 0 | 1 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 0   | 1 | 1 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
| 1   | 1 | 0 |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |   |
- Q13** A heavy nucleus having mass number 200 gets disintegrated into two small fragments of mass number 80 and 120. If binding energy per nucleon for parent atom is 6.5 MeV and for daughter nuclei is 7 MeV and 8 MeV respectively, then the energy released in the decay will be :
- a)** 200 MeV      **b)** - 220 MeV      **c)** 220 MeV      **d)** 180 MeV
- Q14** A uniform rod of length  $L$  and mass  $M$  rests on horizontal frictionless table. A particle of equal mass  $M$  is moving along the 'x' axis at a speed  $V_0$ . At time  $t = 0$  the particle strikes one end of the rod and sticks to it. Consider (particle + rod) as combined system. The position  $\vec{R}_{cm}$  of centre of mass of system as a function of time is :

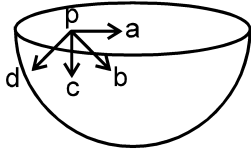


- |  |  |  |  |
|--|--|--|--|
| <p><b>a)</b> <math>\frac{V_0}{2}t \hat{i} + \frac{L}{4} \hat{j}</math></p> | <p><b>b)</b> <math>\frac{V_0}{2}t \hat{i} + \frac{L}{2} \hat{j}</math></p> | <p><b>c)</b> <math>V_0t \hat{i} + \frac{L}{2} \hat{j}</math></p> | <p><b>d)</b> <math>V_0t \hat{i} + \frac{L}{4} \hat{j}</math></p> |
|--|--|--|--|

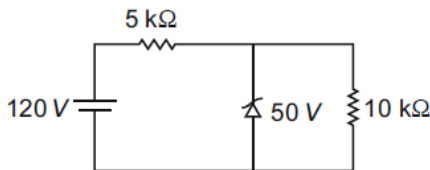
- Q15** A disc and a ring having different masses and same radius are connected with an ideal string as shown in the figure. Both are placed on rough surface of coefficient of friction  $\mu$ . A force  $F$  is applied on the centre of disc horizontally. Assume that initially both the bodies were at rest and they would be rolling without slipping when they start moving. Then frictional force acting on the ring is.



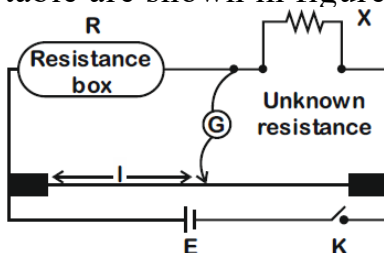
- a)  $\frac{2F}{3}$                       b)  $\frac{F}{3}$                       c) zero                      d) None of these
- Q16** Figure show a hemispherical shell having uniform mass density. The direction of gravitational field intensity at point P will be along:



- a) a                      b) b                      c) c                      d) d
- Q17** For the circuit shown below, the current through the Zener diode is:



- a) Zero                      b) 9 mA                      c) 14 mA                      d) 5 mA
- Q18** In a meter bridge experiment, the circuit diagram and the corresponding observation table are shown in figure.

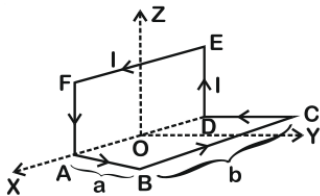


Sl. No	$R(\Omega)$	$l(\text{cm})$
1.	1000	60
2.	100	13
3.	10	1.5
4.	1	1.0

Which of the readings is inconsistent ?

- a) 3                      b) 2                      c) 1                      d) 4

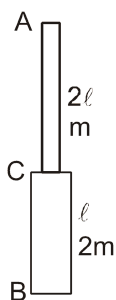
- Q19** A capillary tube made of glass of radius 0.15 mm is dipped vertically in a beaker filled with methylene iodide (surface tension =  $0.05 \text{ Nm}^{-1}$ , density =  $667 \text{ kg m}^{-3}$ ) which rises to height  $h$  in the tube. It is observed that the two tangents drawn from liquid-glass interfaces (from opp. sides of the capillary) make an angle of  $60^\circ$  with one another. Then  $h$  is close to ( $g = 10 \text{ ms}^{-2}$ )
- a) 0.049 m                      b) 0.087 m                      c) 0.137 m                      d) 0.172 m
- Q20** A wire carrying current  $I$  is bent in the shape ABCDEFA as shown, where rectangle ABCDA and ADEFA are perpendicular to each other. If the sides of the rectangles are of lengths  $a$  and  $b$ , then the magnitude and direction of magnetic moment of the loop ABCDEFA is



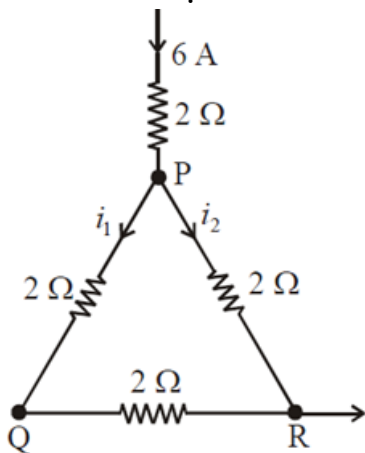
- a)  $abl$ , along  $\left(\frac{\hat{j}}{\sqrt{5}} + \frac{2\hat{k}}{\sqrt{5}}\right)$                       b)  $\sqrt{2}abl$ , along  $\left(\frac{\hat{j}}{\sqrt{2}} + \frac{\hat{k}}{\sqrt{2}}\right)$
- c)  $abl$ , along  $\left(\frac{\hat{j}}{\sqrt{2}} + \frac{\hat{k}}{\sqrt{2}}\right)$                       d)  $\sqrt{2}abl$ , along  $\left(\frac{\hat{j}}{\sqrt{5}} + \frac{2\hat{k}}{\sqrt{5}}\right)$

## Numerical

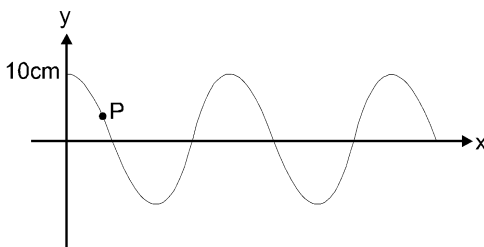
- Q21** A composite rod is made of two rods: one of mass  $m$ , length  $2\ell$  and another of mass  $2m$  and length  $\ell$  as shown. When the rod is suspended from A and allowed small oscillation in vertical plane, time period is  $T$ . Find the distance (in centimeter) of the point of suspension on other side from the junction so that time period is still the same. Take  $\ell = 60 \text{ cm}$ .



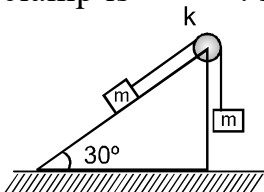
- Q22** A current of 6 A enters one corner P of an equilateral triangle PQR having 3 wires of resistance  $2\ \Omega$  each and leaves by the corner R. The currents  $i_1$  in ampere is



- Q23** A ball with a speed of 9 m/s collides with another identical ball at rest. After the collision, the direction of each ball makes an angle of  $30^\circ$  with the original direction. The ratio of velocities of the balls after collision is  $x : y$ , where  $x$  is \_\_\_\_\_.
- Q24** A transverse sinusoidal wave moves along a string in the positive  $x$ -direction at a speed of 10 cm/s. The wavelength of the wave is 0.5 m and its amplitude is 10 cm. At a particular time  $t$ , the snap-shot of the wave is shown in figure. The speed of point P is  $x\sqrt{2}\pi$  cm/s when its displacement is  $\frac{10}{\sqrt{2}}$  cm, then the value of  $x$  is :



- Q25** A position dependent force  $F = \frac{1}{2}|x - 4|$  is acting along  $x$ -axis, where  $x$  is in meter and  $F$  is in Newton. Find the work done (in joule) by the force in moving a particle from origin to  $x = 8$  m rectilinearly along  $x$ -axis.
- Q26** Two blocks of equal mass ' $m$ ' are connected by massless string and pulley as shown in figure. Now the blocks are left to move. Then the force exerted by pulley on the clamp is  $\frac{3\sqrt{3}Mg}{k}$ . Find the value  $k$  ?



- Q27** On an inclined plane of inclination  $30^\circ$ , a ball is thrown at an angle of  $60^\circ$  with the horizontal from the foot of the incline with a velocity of  $10\sqrt{3}\text{ ms}^{-1}$ . If  $g = 10\text{ ms}^{-2}$ , then the time in which ball will hit the inclined plane is -

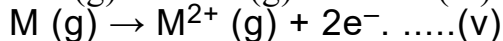
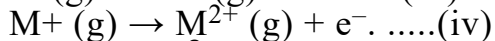
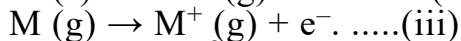
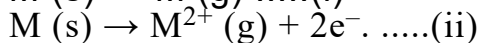
- Q28** A uniform chain of 6 m length is placed on a table such that a part of its length is hanging over the edge of the table. The system is at rest. The co-efficient of static friction between the chain and the surface of the table is 0.5, the maximum length of the chain hanging from the table is \_\_\_\_\_ m.
- Q29** The displacement current of  $4.425 \mu\text{A}$  is developed in the space between the plates of parallel plate capacitor when voltage is changing at a rate of  $10^6 \text{ Vs}^{-1}$ . The area of each plate of the capacitor is  $40 \text{ cm}^2$ . The distance between each plate of the capacitor is  $x \times 10^{-3} \text{ m}$ . The value of x is,  
(Permittivity of free space,  $\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2 \text{ N}^{-1} \text{ m}^{-2}$ )
- Q30** Two identical conducting spheres with negligible volume have  $2.1 \text{ nC}$  and  $-0.1 \text{ nC}$  charges, respectively. They are brought into contact and then separated by a distance of  $0.5 \text{ m}$ . The electrostatic force acting between the spheres is \_\_\_\_\_  $\times 10^{-9} \text{ N}$ .  
[ Given :  $4\pi\epsilon_0 = \frac{1}{9 \times 10^9} \text{ SI unit}$  ]



## Chemistry

### Single Choice Question

**Q31** Consider the following changes :



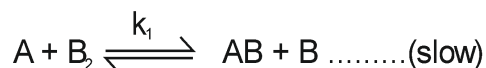
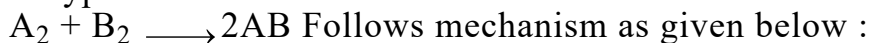
The second ionization energy of M could be calculated from the energy values associated with :

- a)  $1 + 3 + 4$                       b)  $2 - 1 + 3$                       c)  $1 + 5$                       d)  $5 - 3$ .

**Q32** In which of the following molecules number of lone pairs and bond pairs on central atom are not equal ?

- a)  $H_2O$                       b)  $I_3^{-}$                       c)  $O_2F_2$                       d)  $SCl_2$

**Q33** A hypothetical reaction :



The order of overall reaction is :

- a) 2.5                      b) 1                      c)  $3/2$                       d) Zero

**Q34** In which of the following reactions, the equilibrium remains unaffected on addition of small amount of argon at constant volume.

- a)  $H_2 + I_2 \rightleftharpoons 2HI(g)$   
 b)  $PCl_5(g) \rightleftharpoons PCl_3(g) + Cl_2(g)$   
 c)  $N_2 + 3H_2 \rightleftharpoons 2NH_3(g)$   
 d) The equilibrium will be remain unaffected in all the three cases.

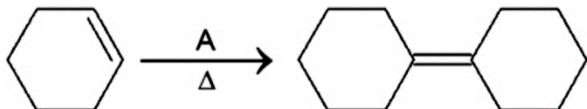
**Q35** Which of the followings is/are correct :

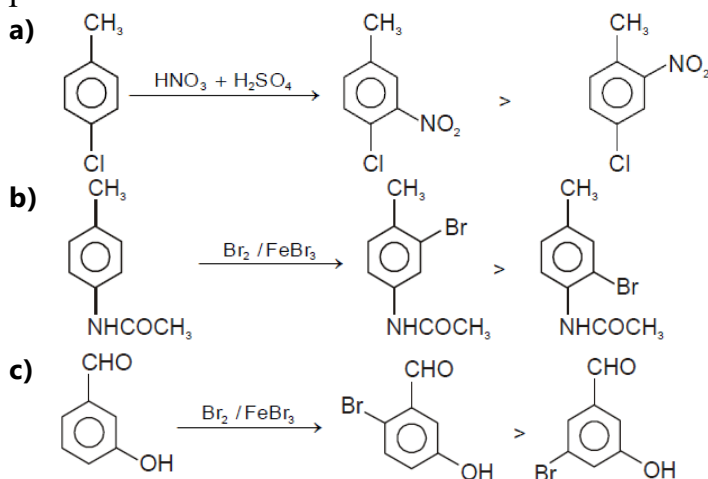
- (i)  $\alpha$ -D-glucose and  $\beta$ -D-glucose are anomers of each other.  
 (ii) D-Galactose is C-4 epimer of D-glucose.  
 (iii) Sucrose is a reducing sugar.  
 (iv) Cellulose is natural polymer of  $\beta$ -D-glucopyranose

- a) i, ii, iii, iv                      b) i, ii, iii                      c) i, ii, iv                      d) ii, iii, iv

**Q36** Which of the following will have three stereoisomeric forms?

- (i)  $[Cr(NO_3)_3(NH_3)_3]$  (ii)  $K_3[Co(C_2O_4)_3]$  (iii)  $K_3[CoCl_2(C_2O_4)_2]$  (iv)  $[CoBrCl(en)_2]$   
 a) (iii) and (iv)                      b) (i), (iii) and (iv)                      c) (iv) only                      d) All four

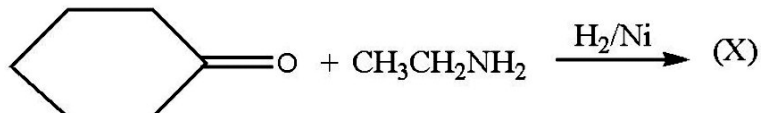
- Q37** Sulphide ions reacts with sodium nitropruside giving a coloured solution. In the reaction, the oxidation state of iron :
- a) Changes from +2 to +4                      b) Changes from +3 to +2  
c) Changes from +2 to +3                      d) Does not change
- Q38** Three faradays of electricity are passed through molten  $\text{Al}_2\text{O}_3$ , aqueous solution of  $\text{CuSO}_4$  and molten  $\text{NaCl}$  taken in different electrolytic cells. The amount of Al, Cu and Na deposited at the cathodes will be in the ratio of
- a) 1 mole : 2 mole : 3 mole                      b) 3 mole : 2 mole : 1mole  
c) 1 mole : 1.5 mole : 3 mole                      d) 1.5 mole : 2 mole : 3 mole
- Q39** 
- A can be
- a) Conc.  $\text{H}_2\text{SO}_4$                       b) Alc. KOH                      c)  $\text{Et}_3\text{N}$                       d)  $t\text{-BuOK}$
- Q40** A 0.001 molal solution of a complex  $[\text{MA}_8]$  in water has the freezing point of  $-0.0054^\circ\text{C}$ . Assuming 100% ionization of the complex salt and  $K_f$  for  $\text{H}_2\text{O} = 1.86 \text{ km}^{-1}$ , write the correct representation for the complex
- a)  $[\text{MA}_8]$                       b)  $[\text{MA}_7]\text{A}$                       c)  $[\text{MA}_6]\text{A}_2$                       d)  $[\text{MA}_5]\text{A}_3$
- Q41** The thermal stability of the hydrides of group 15 follows the order:
- a)  $\text{NH}_3 < \text{PH}_3 < \text{AsH}_3 < \text{SbH}_3 < \text{BiH}_3$                       b)  $\text{NH}_3 > \text{PH}_3 > \text{AsH}_3 > \text{SbH}_3 > \text{BiH}_3$   
c)  $\text{PH}_3 > \text{NH}_3 > \text{AsH}_3 > \text{SbH}_3 < \text{BiH}_3$                       d)  $\text{AsH}_3 < \text{PH}_3 > \text{SbH}_3 > \text{BiH}_3 > \text{NH}_3$
- Q42** For the first-order reaction  $T_{\text{av}}$  (average life),  $T_{50}$  and  $T_{75}$  in the increasing order are :
- a)  $T_{50} < T_{\text{av}} < T_{75}$                       b)  $T_{50} < T_{75} < T_{\text{av}}$                       c)  $T_{\text{av}} < T_{50} < T_{75}$                       d)  $T_{\text{av}} = T_{50} < T_{75}$
- Q43** How many ml water should be added to 100ml HCl solution ( $d = 1.5 \text{ g/ml}$ ) 80% by wt. to make it a solution of 40% by wt. of density = 1 g/ml.
- a) 100 ml                      b) 300 ml                      c) 200 ml                      d) none of these
- Q44** Select the reaction in which the correct orientation have been mentioned in the major products.



d) All of these are correct

- Q45** Aryl halides are less reactive towards nucleophilic substitution reactions as compared to alkyl halides due to
- The formation of less stable carbocation
  - Shorter carbon halogen bond than usual
  - $sp^2$ -hybridized carbon attached to the halogen
  - All of the above

- Q46** What is the product (X) in the given reaction?



- 
- 
- 
- 

- Q47**  $\text{CH}_3 - \text{CH} = \text{O} + \text{HCN} \xrightarrow{\text{Dil. NaOH}} \xrightarrow{\text{H}_3\text{O}^+}$   
Products are :

- (50% d + 50% l)
- (Only one isomer)
- (Only one isomer)
- (pair of enantiomers)

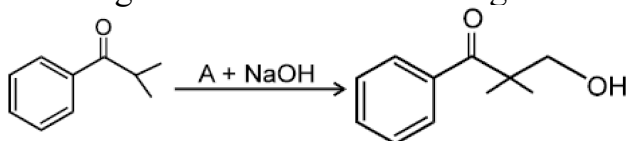
- Q48**  $\text{H}_3\text{C}-\text{CH}_2-\text{C}(=\text{O})-\text{CH}_3 + \text{Ph}_3\text{P}=\text{C}(\text{CH}_3)_2 \xrightarrow{\Delta} \text{X}$  Product X will be :

- 
- 
- Both (1) and (2)
-

**Q49** For the conversion of benzene into m-chlorobenzoic acid, the correct sequence of reagents is :

- a) (i)  $\text{C}_2\text{H}_5\text{Cl}$ ,  $\text{AlCl}_3$  (ii)  $\text{Cl}_2$ , Fe  
 (iii)  $\text{KMnO}_4$ ,  $\text{H}^+$ ,  $\Delta$
- b) (i)  $\text{CO}$ ,  $\text{HCl}$ ,  $\text{AlCl}_3$  (ii)  $\text{Cl}_2$ , Fe  
 (iii)  $\text{KMnO}_4$ ,  $\text{H}^+$ ,  $\Delta$
- c) (i)  $\text{Cl}_2$ ,  $\text{AlCl}_3$  (ii)  $\text{C}_2\text{H}_5\text{Cl}$ ,  $\text{AlCl}_3$   
 (iii)  $\text{KMnO}_4$ ,  $\text{H}^+$ ,  $\Delta$
- d) (i)  $\text{HCOOH}$ ,  $\text{AlCl}_3$  (ii)  $\text{Cl}_2$ , Fe

**Q50** The reagent 'A' for the following reaction would be



- a)  $\text{CH}_3 - \text{C}(=\text{O}) - \text{OC}_2\text{H}_5$
- b)  $\text{CH}_3\text{CHO}$
- c)  $\text{HCHO}$
- d)  $\text{CH}_2 = \text{CH}_2$

### Numerical

- Q51** In Duma's method of estimation of nitrogen, 0.1840 g of an organic compound gave 30 mL of nitrogen collected at 287 K and 758 mm of Hg pressure. The percentage composition of nitrogen in the compound is \_\_\_\_\_. (Round off to the Nearest Integer). [Given : Aqueous tension at 287 K = 14 mm of Hg]
- Q52** The number of orbitals with  $n = 5$ ,  $m_l = +2$  is \_\_\_\_\_. (Round off to the Nearest Integer).
- Q53** Sulphurous acid ( $\text{H}_2\text{SO}_3$ ) has  $K_{a1} = 1.7 \times 10^{-2}$  and  $K_{a2} = 6.4 \times 10^{-8}$ . The pH of 0.588 M  $\text{H}_2\text{SO}_3$  is \_\_\_\_\_. (Round off to the Nearest Integer)
- Q54** When 35 mL of 0.15 M lead nitrate solution is mixed with 20 mL of 0.12 M chromic sulphate solution, \_\_\_\_\_  $\times 10^{-5}$  moles of lead sulphate precipitate out. If your answer is X then what will be the value of  $\frac{X}{25}$ . (Round off to the Nearest Integer).
- Q55** For the reaction  $\text{A(g)} \rightleftharpoons \text{B(g)}$  at 495 K,  $\Delta_r G^\circ = -9.478 \text{ kJ mol}^{-1}$ . If we start the reaction in a closed container at 495 K with 22 millimoles of A, the amount of B in the equilibrium mixture is \_\_\_\_\_ millimoles. (Round off to the Nearest Integer). [R =  $8.314 \text{ J mol}^{-1} \text{ K}^{-1}$ ;  $\ln 10 = 2.303$ ]

**Q56** The standard enthalpies of formation of  $\text{Al}_2\text{O}_3$  and  $\text{CaO}$  are  $-1675 \text{ kJ mol}^{-1}$  and  $-635 \text{ kJ mol}^{-1}$  respectively.

For the reaction

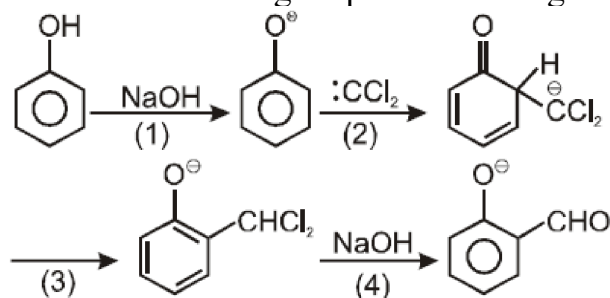
$3\text{CaO} + 2\text{Al} \rightarrow 3\text{Ca} + \text{Al}_2\text{O}_3$  the standard reaction enthalpy  $\Delta_r H^0 = \underline{\hspace{2cm}}$  kJ.

If your answer is X then what will be the value of  $\frac{X}{10}$ .

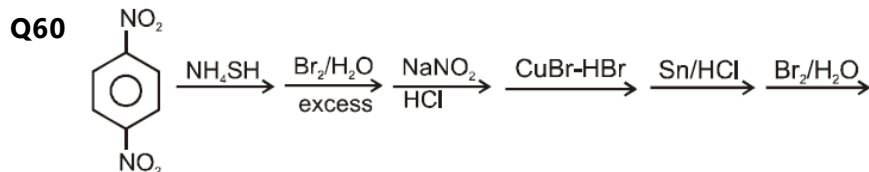
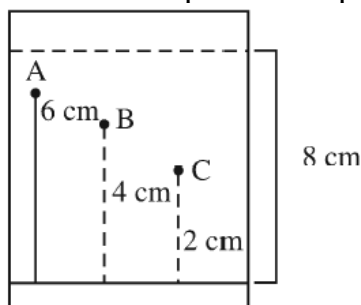
(Round off to the Nearest Integer).

**Q57** The number of moles of potassium dichromate ( $\text{K}_2\text{Cr}_2\text{O}_7$ ) required for the complete reaction of 36 mole of cuprous sulphide ( $\text{Cu}_2\text{S}$ ) with its acidified solution producing  $\text{Cu}^{2+}$  ions and  $\text{SO}_2(\text{g})$  is :

**Q58** The rate determining step of following reaction is:



**Q59** Three organic compounds A, B and C were allowed to run in thin layer chromatography using hexane and gave the following result (see figure). The  $R_f$  value of the least polar compound is  $\underline{\hspace{2cm}} \times 10^{-2}$



major end product.

Find the total number of halogen atoms present in the major end product :

## Mathematics

### Single Choice Question

- Q61** Let  $f(x) = e^x - x$  and  $g(x) = x^2 - x$ ,  $\forall x \in \mathbb{R}$ . Then the set of all  $x \in \mathbb{R}$ , where the function  $h(x) = (f \circ g)(x)$  is increasing, is :
- a)  $\left[-1, \frac{-1}{2}\right] \cup \left[\frac{1}{2}, \infty\right)$       b)  $[0, \infty)$       c)  $\left[0, \frac{1}{2}\right] \cup [1, \infty)$       d)  $\left[\frac{-1}{2}, 0\right] \cup [1, \infty)$
- Q62** If  $f(x) = \sqrt{\frac{1 + \sin^{-1} x}{1 - \tan^{-1} x}}$ ; then  $f'(0)$  is equal to :
- a) 4      b) 3      c) 2      d) 1
- Q63** From a pack of 52 well shuffled cards, cards are drawn one by one without replacement. If 4<sup>th</sup> drawn card is found to be ace, then what is the probability, that there are no more aces left in the pack is :-
- a)  $\frac{1}{48C_3 + 3 \cdot 49C_2 + 1}$       b)  $\frac{1}{48C_3 + 49C_2 + 1}$       c)  $\frac{1}{3 \cdot 48C_3 + 49C_2 + 1}$       d)  $\frac{1}{52C_4 + 1}$
- Q64** Let  $A = \begin{bmatrix} 2 & b & 1 \\ b & b^2 + 1 & b \\ 1 & b & 2 \end{bmatrix}$  where  $b > 0$ . Then find minimum value of  $\frac{|A|}{b}$  is
- a)  $-\sqrt{3}$       b)  $\sqrt{3}$       c)  $2\sqrt{3}$       d)  $-2\sqrt{3}$
- Q65** Let  $\vec{a} = 3\hat{i} + 2\hat{j} + x\hat{k}$  and  $\vec{b} = \hat{i} - \hat{j} + \hat{k}$ , for some real  $x$ . Then  $|\vec{a} \times \vec{b}| = r$  is possible if :
- a)  $3\sqrt{\frac{3}{2}} < r < 5\sqrt{\frac{3}{2}}$       b)  $\sqrt{\frac{3}{2}} < r \leq 3\sqrt{\frac{3}{2}}$       c)  $0 < r \leq \sqrt{\frac{3}{2}}$       d)  $r \geq 5\sqrt{\frac{3}{2}}$
- Q66** if  $a^2 + b^2 + c^2 = 0$  and matrix  $A = \begin{vmatrix} b^2 + c^2 & ab & ac \\ ab & c^2 + a^2 & bc \\ ac & bc & a^2 + b^2 \end{vmatrix}$  and if  $|\text{adj}(\text{adj} A)| = 32\lambda$
- $a^8 b^8 c^8$ , ( $a, b, c \neq 0$ ), then  $\lambda =$
- a) 8      b) 16      c) 32      d) 4
- Q67** If  $2a + 2b + 3c = \frac{1}{5}$  and  $a, b, c \in \mathbb{R}^+$ , then maximum value of term independent of  $x$  in the expansion of  $(abx^{1/2} + cx^{-1/3})^{25}$  is -
- a)  ${}^{25}C_{10}$       b)  ${}^{25}C_{10}(35)^{25}$       c)  ${}^{25}C_{15}\left(\frac{1}{35}\right)^{35}$       d) None of these

- Q68** If  $\alpha, \beta$  are the roots of equation  $x^2 - 2x + 5 = 0$ , then equation, whose roots are :  $\alpha^3 + \alpha^2 - \alpha + 22, \beta^3 + 4\beta^2 - 7\beta + 35$  is :
- a)  $x^2 - 57x + 770 = 0$                       b)  $x^2 - 12x + 35 = 0$   
 c)  $x^2 - 2x + 5 = 0$                               d)  $x^2 - 11x + 25 = 0$
- Q69** If the equation  $4y^3 - 8a^2yx^2 - 3ay^2x + 8x^3 = 0$  represent three straight lines, two of them are perpendicular then sum of all possible values of a is equal to :
- a)  $\frac{3}{8}$                               b)  $-\frac{3}{4}$                               c)  $\frac{1}{4}$                               d)  $-2$
- Q70** Let  $f(-1, \infty) \rightarrow \mathbb{R}$  be defined by  $f(0) = 1$  and  $f(x) = \frac{1}{x} \log_e(1+x), x \neq 0$ . Then the function f
- a) decreases in  $(-1, 0)$  and increases in  $(0, \infty)$   
 b) increases in  $(-1, \infty)$   
 c) increases in  $(-1, 0)$  and decreases in  $(0, \infty)$   
 d) decreases in  $(-1, \infty)$
- Q71** Let  $f(x) = 3x^{10} - 7x^8 + 5x^6 - 21x^3 + 3x^2 - 7$
- $$265 \left( \lim_{h \rightarrow 0} \frac{h^4 + 3h^2}{(f(1-h) - f(1)) \sin 5h} \right) =$$
- a) 1                              b) 2                              c) 3                              d) -3
- Q72** Given that for  $a, b, c, d \in \mathbb{R}$ , if  $a \sec(200^\circ) - c \tan(200^\circ) = d$  and  $b \sec(200^\circ) + d \tan(200^\circ) = c$ , then find the value of  $\left( \frac{a^2 + b^2 + c^2 + d^2}{bd - ac} \right) \sin 20^\circ$ .
- a) 2                              b) 1                              c) 0                              d) None of these
- Q73** Let  $f(x) = \min. (x+1, \sqrt{1-x})$  for all  $x \leq 1$ . Then the area bounded by  $y = f(x)$  and the x-axis is :-
- a)  $\frac{7}{3}$  sq. units                      b)  $\frac{1}{6}$  sq. units                      c)  $\frac{11}{6}$  sq. units                      d)  $\frac{7}{6}$  sq. units
- Q74** Evaluate  $\int \frac{\left( \sqrt[3]{x + \sqrt{2-x^2}} \right) \left( \sqrt[6]{1-x\sqrt{2-x^2}} \right) dx}{\sqrt[3]{1-x^2}}; x \in (0,1)$  :
- a)  $\frac{1}{2^6}x + C$                       b)  $\frac{1}{2^{12}}x + C$                       c)  $\frac{1}{2^3}x + C$                       d) None of these
- Q75** The general solution of differential equation,  $\sin 2x \left( \frac{dy}{dx} - \sqrt{\tan x} \right) - y = 0$  is
- a)  $y\sqrt{\cot x} = \tan x + C$                       b)  $y\sqrt{\cot x} = x + C$                       c)  $y\sqrt{\tan x} = \cot x + C$                       d)  $y\sqrt{\tan x} = x + C$

- Q76** The set of values of  $k$  for which the circle  $C : 4x^2 + 4y^2 - 12x + 8y + k = 0$  lies inside the fourth quadrant and the point  $\left(1, -\frac{1}{3}\right)$  lies on or inside the circle  $C$  is :
- a) An empty                      b)  $\left(6, \frac{95}{9}\right]$                       c)  $\left[\frac{80}{9}, 10\right)$                       d)  $\left(9, \frac{92}{9}\right]$
- Q77** The vertex  $C$  of a triangle  $ABC$  is  $(4, -1)$ . The equation of altitude  $AD$  and Median  $AE$  are  $2x - 3y + 12 = 0$  and  $2x + 3y = 0$  respectively then slope of side  $AB$  is :
- a)  $-\frac{3}{7}$                       b)  $-\frac{3}{2}$                       c)  $-\frac{9}{11}$                       d) None of these
- Q78** The foci of a hyperbola lie at the vertices of the ellipse  $\frac{x^2}{100} + \frac{y^2}{64} = 1$  and its directrices pass through the foci of the ellipse. The equation of the hyperbola must be :-
- a)  $\frac{x^2}{100} - \frac{y^2}{64} = 1$                       b)  $\frac{x^2}{40} - \frac{y^2}{60} = 1$                       c)  $\frac{x^2}{60} - \frac{y^2}{40} = 1$                       d) None of these
- Q79** The adjacent side vectors  $\vec{OA}$  and  $\vec{OB}$  of a rectangle  $OACB$  are  $\vec{a}$  and  $\vec{b}$  respectively, where  $O$  is the origin. If  $16|\vec{a} \times \vec{b}| = 3(|\vec{a}| + |\vec{b}|)^2$  and  $\theta$  be the acute angle between the diagonals  $OC$  and  $AB$  then the value of  $\tan(\theta/2)$  is :
- a)  $\frac{1}{\sqrt{2}}$                       b)  $\frac{1}{2}$                       c)  $\frac{1}{\sqrt{3}}$                       d)  $\frac{1}{3}$
- Q80** A batsman can score 0, 1, 2, 3, 4 or 6 runs from a ball. The number of different sequences in which he can score exactly 30 runs in an over of six balls :
- a) 4                      b) 72                      c) 56                      d) 71

## Numerical

- Q81** If  $f(x) = \int \frac{5x^8 + 7x^6}{(x^2 + 1 + 2x^7)^2} dx$ , ( $x \geq 0$ ),  $f(0) = 0$  and  $f(1) = \frac{1}{K}$ , then the value of  $K$  is
- Q82** Let  $A = \{2, 3, 4\}$  and  $B = \{8, 9, 12\}$ . Then the number of elements in the relation  $R = \{((a_1, b_1), (a_2, b_2)) \in (A \times B, A \times B) : a_1 \text{ divides } b_2 \text{ and } a_2 \text{ divides } b_1\}$  is :
- Q83** Let the image of the point  $P(1, 2, 3)$  in the line  $L : \frac{x-6}{3} = \frac{y-1}{2} = \frac{z-2}{3}$  be  $Q$ . Let  $R(\alpha, \beta, \gamma)$  be a point that divides internally the line segment  $PQ$  in the ratio 1 : 3. Then the value of  $22(\alpha + \beta + \gamma)$  is equal to
- Q84** Let  $f : \mathbb{R} \rightarrow \mathbb{R}$  satisfy the equation  $f(x + y) = f(x) \cdot f(y)$  for all  $x, y \in \mathbb{R}$  and  $f(x) \neq 0$  for any  $x \in \mathbb{R}$ . If the function  $f$  is differentiable at  $x = 0$  and  $f'(0) = 3$ , then  $\lim_{h \rightarrow 0} \frac{1}{h}(f(h) - 1)$  is equal to \_\_\_\_\_.
- Q85** The parabola  $y = 4 - x^2$  has vertex  $P$ . It intersects  $x$ -axis at  $A$  and  $B$ . If the parabola is translated from its initial position to a new position by moving its vertex along the line  $y = x + 4$ , so that it intersects  $x$ -axis at  $B$  and  $C$ , then abscissa of  $C$  will be :



**Q86** The numbers  $\frac{1}{3}, \frac{1}{3} \log_x y, \frac{1}{3} \log_y z, \frac{1}{7} \log_z x$  are in H.P. If  $y = x^r$  and  $z = x^s$ , then  $4(r + s) =$

**Q87** If  $\sum_{n=0}^{\infty} 2 \cot^{-1} \left( \frac{n^2 + n + 4}{2} \right) = k\pi$ , then find the value of  $k$ .

**Q88** Let  $f(x) = \int_0^x e^{x-y} f'(y) dy - (x^2 - x + 1)e^x$ . Find the number of roots of the equation  $f(x) = 0$ .

**Q89** Complex number  $z_1$  and  $z_2$  satisfy  $z + \bar{z} = 2|z-1|$  and  $\arg(z_1 - z_2) = \frac{\pi}{4}$ . Then the value of  $\operatorname{Im}(z_1 + z_2)$  is: (where  $z_1$  &  $z_2$  are non zero complex number)

**Q90** If the mean and variance of the frequency distribution.

$x_i$	2	4	6	8	10	12	14	16
$f_i$	4	4	$\alpha$	15	8	$\beta$	4	5

are 9 and 15.08 respectively, then the value of  $\alpha^2 + \beta^2 - \alpha\beta$  is \_\_\_\_\_.

# Rankers Academy JEE

## Answer Key

Que.	1	2	3	4	5	6	7	8	9	10
<b>Ans.</b>	<b>C</b>	<b>B</b>	<b>C</b>	<b>C</b>	<b>B</b>	<b>C</b>	<b>A</b>	<b>B</b>	<b>D</b>	<b>B</b>
Que.	11	12	13	14	15	16	17	18	19	20
<b>Ans.</b>	<b>D</b>	<b>A</b>	<b>C</b>	<b>A</b>	<b>C</b>	<b>C</b>	<b>B</b>	<b>D</b>	<b>B</b>	<b>B</b>
Que.	21	22	23	24	25	26	27	28	29	30
<b>Ans.</b>	<b>20</b>	<b>2</b>	<b>1</b>	<b>2</b>	<b>8</b>	<b>4</b>	<b>2</b>	<b>2</b>	<b>8</b>	<b>36</b>
Que.	31	32	33	34	35	36	37	38	39	40
<b>Ans.</b>	<b>D</b>	<b>B</b>	<b>C</b>	<b>D</b>	<b>C</b>	<b>A</b>	<b>D</b>	<b>C</b>	<b>A</b>	<b>C</b>
Que.	41	42	43	44	45	46	47	48	49	50
<b>Ans.</b>	<b>B</b>	<b>A</b>	<b>C</b>	<b>C</b>	<b>D</b>	<b>D</b>	<b>D</b>	<b>C</b>	<b>B</b>	<b>C</b>
Que.	51	52	53	54	55	56	57	58	59	60
<b>Ans.</b>	<b>19</b>	<b>3</b>	<b>1</b>	<b>21</b>	<b>20</b>	<b>23</b>	<b>48</b>	<b>2</b>	<b>75</b>	<b>5</b>
Que.	61	62	63	64	65	66	67	68	69	70
<b>Ans.</b>	<b>C</b>	<b>D</b>	<b>A</b>	<b>C</b>	<b>D</b>	<b>A</b>	<b>C</b>	<b>B</b>	<b>B</b>	<b>D</b>
Que.	71	72	73	74	75	76	77	78	79	80
<b>Ans.</b>	<b>C</b>	<b>A</b>	<b>D</b>	<b>A</b>	<b>B</b>	<b>D</b>	<b>C</b>	<b>C</b>	<b>D</b>	<b>D</b>
Que.	81	82	83	84	85	86	87	88	89	90
<b>Ans.</b>	<b>4</b>	<b>36</b>	<b>125</b>	<b>3</b>	<b>25</b>	<b>6</b>	<b>1</b>	<b>1</b>	<b>2</b>	<b>25</b>