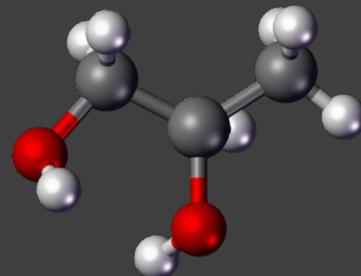
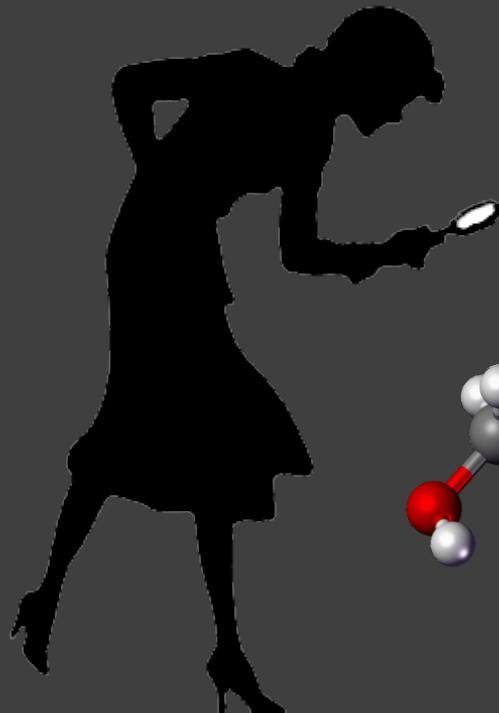




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Fun fact: the Sun is made of mostly hydrogen and helium. If you could put it in a very big pool, it would float!



Spectral Detective Work: Finding Molecules in Space

Alyssa Bulatek (she/her)
Astronomy



Outline

- Where do elements come from?
- How do molecules form from atoms?
- Can molecules that form on Earth also exist in space?
- How do we detect molecules in space?
- What does an astronomer's job look like?

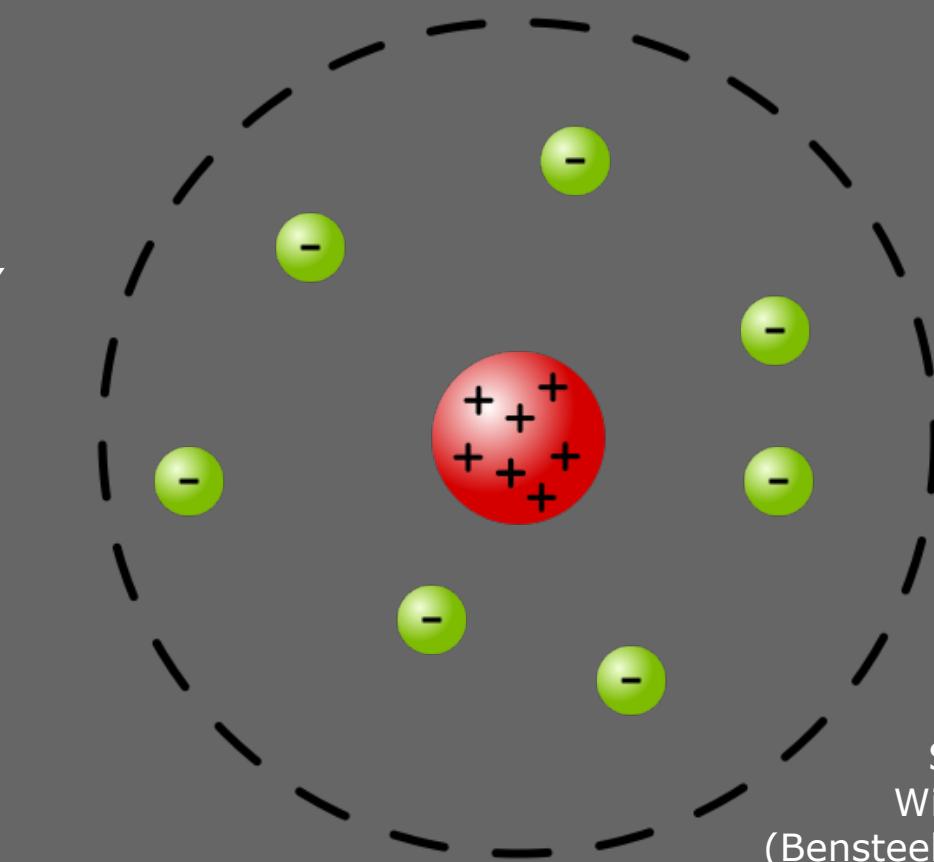
Takeaway: elements on the periodic table are relevant to everyday life, and some of the same elements we encounter daily can also be found in space!



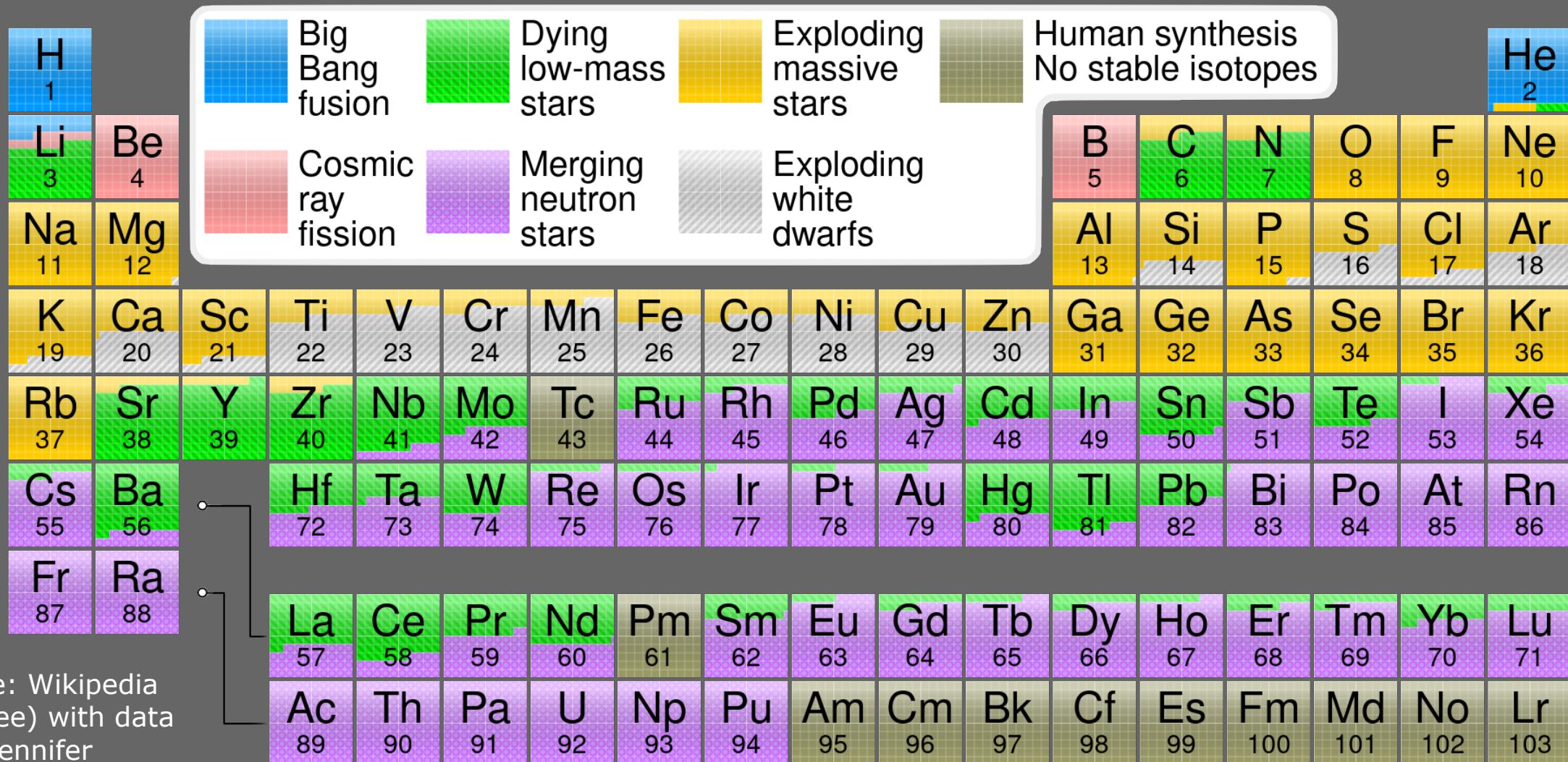
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What is the anatomy of an atom?

- Different models of the atom have been suggested
 - *Rutherford-Bohr: tight **nucleus**, broad "**electron cloud**"*
- The number of protons that an atom has determines which element it is
- Cool interactive periodic table:
<https://ptable.com/>



Source:
Wikipedia
(Bensteele1995)



Source: Wikipedia
(Cmglee) with data
from Jennifer
Johnson (OSU)

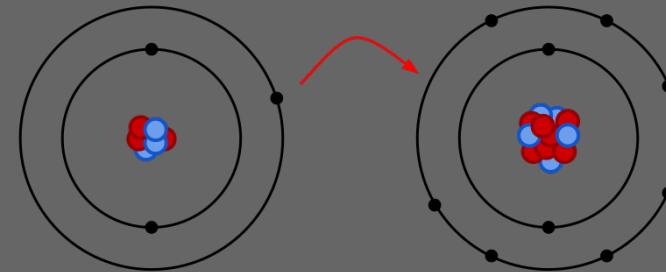
Where do the elements come from?



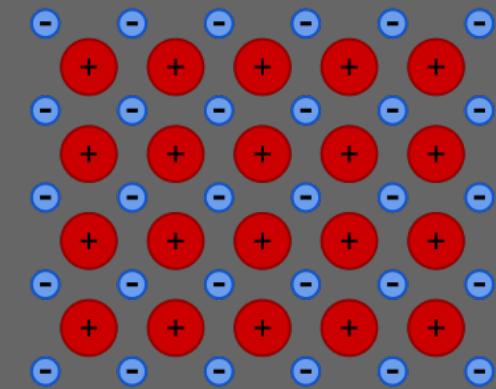
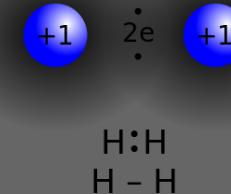
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How do elements combine to make molecules?

- When atoms get close to each other, the **electrons** orbiting their nuclei interact
- Three main types of bonds:
 - Ionic*
 - Covalent*
 - Metallic*
- Bonds between molecules exist, though they are weaker than atomic bonds



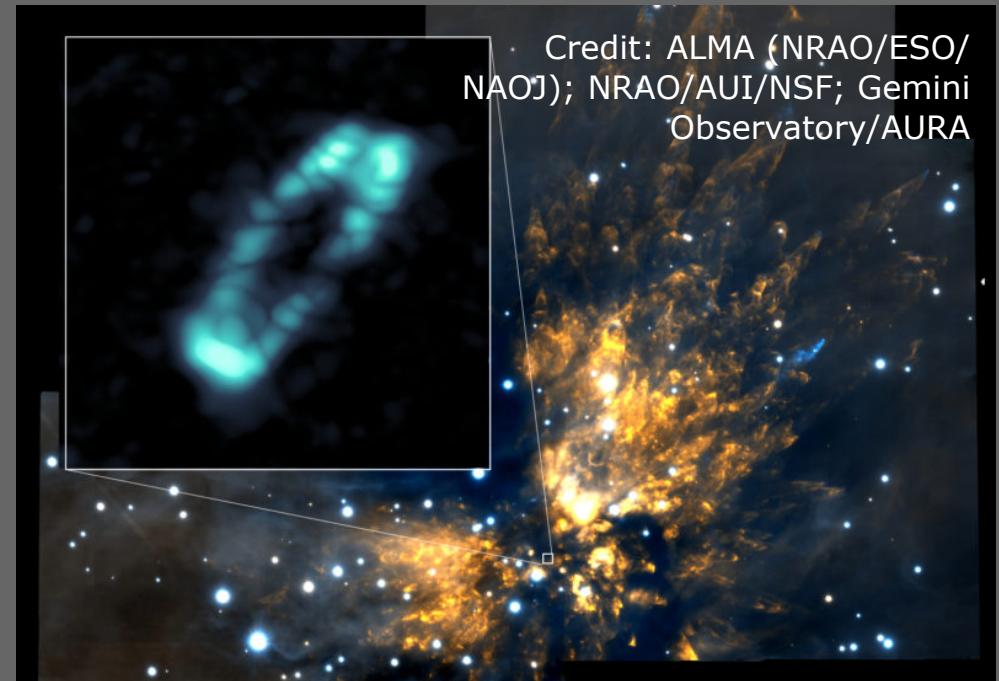
Source: Wikipedia (Jacek FH)



Where are molecules found?

- A quick quiz! What "common" substances do these formulas represent?
- All of those molecules have also been detected in space!

H₂O
CO
NaCl
N₂O
NH₃
CH₃CH₂OH
(CH₃)₂CO



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H_2O

CO

NaCl

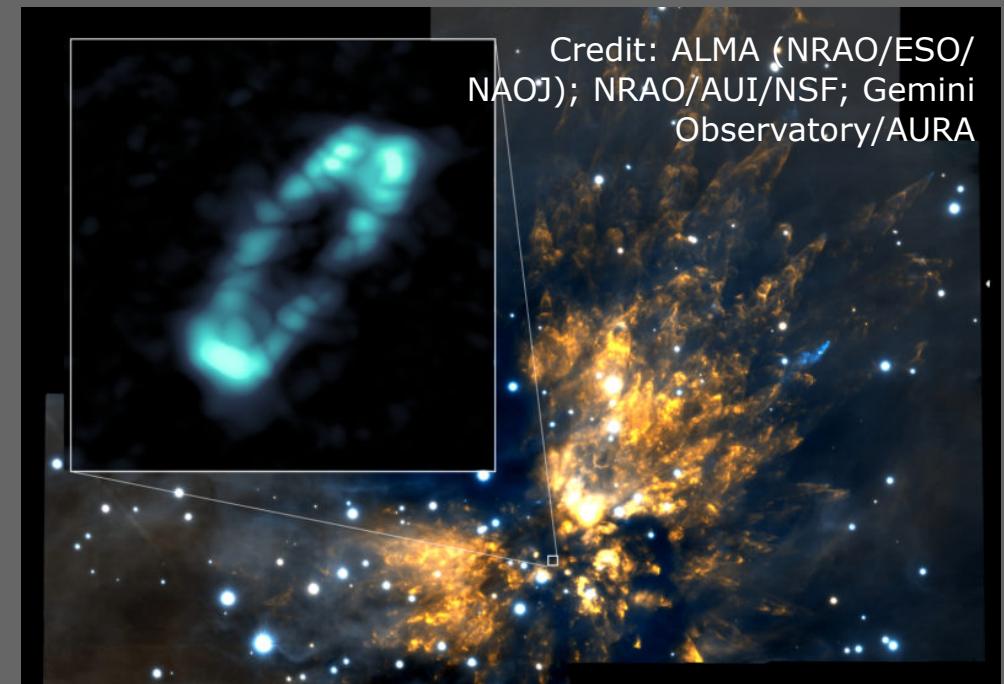
N_2O

NH₃

$\text{CH}_3\text{CH}_2\text{OH}$

$(\text{CH}_3)_2\text{CO}$

water



Where are molecules found?

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H_2O

water

CO

carbon monoxide

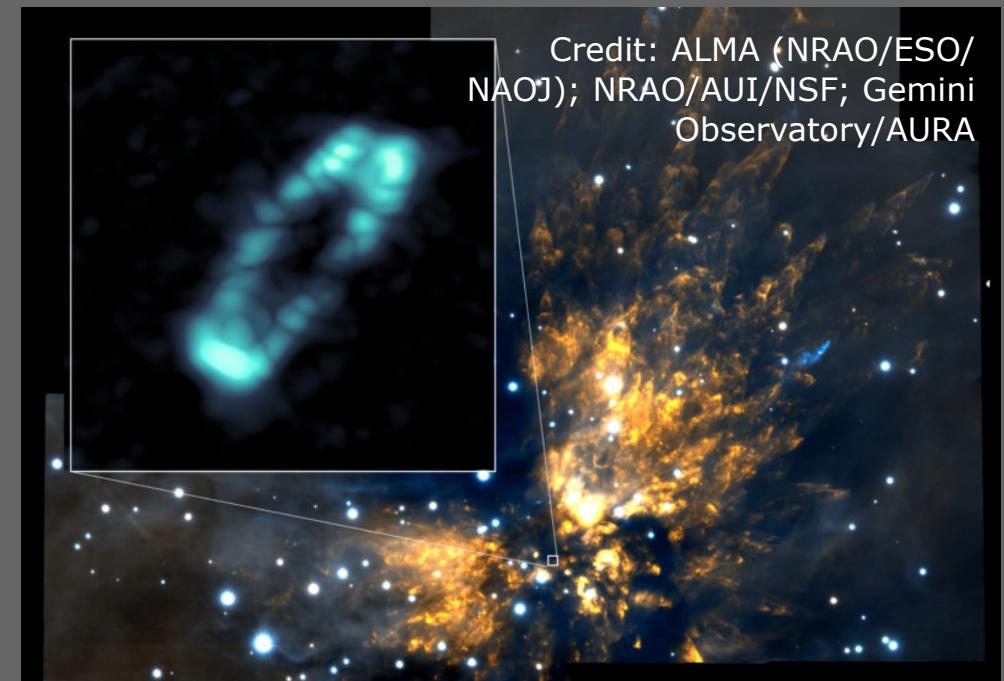
NaCl

N_2O

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$\text{CH}_3\text{CH}_2\text{OH}$

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CO

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NaCl

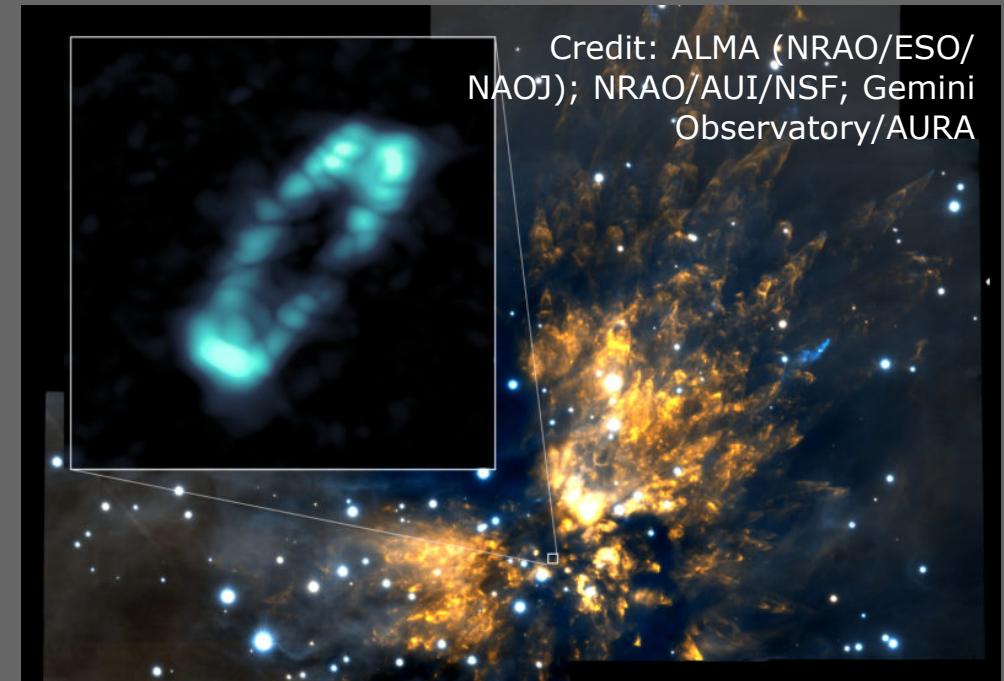
salt

N_2O

NH_3

$\text{CH}_3\text{CH}_2\text{OH}$

$(\text{CH}_3)_2\text{CO}$



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H_2O

water

CO

carbon monoxide

NaCl

salt

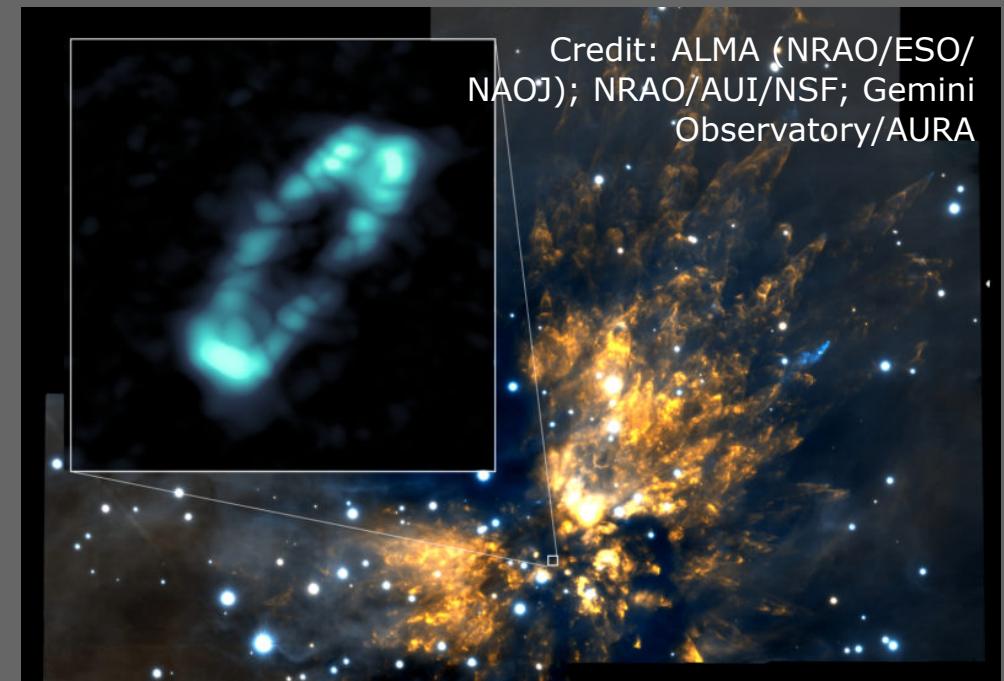
N_2O

laughing gas

NH_3

$\text{CH}_3\text{CH}_2\text{OH}$

$(\text{CH}_3)_2\text{CO}$



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H_2O

water

CO

carbon monoxide

NaCl

salt

N_2O

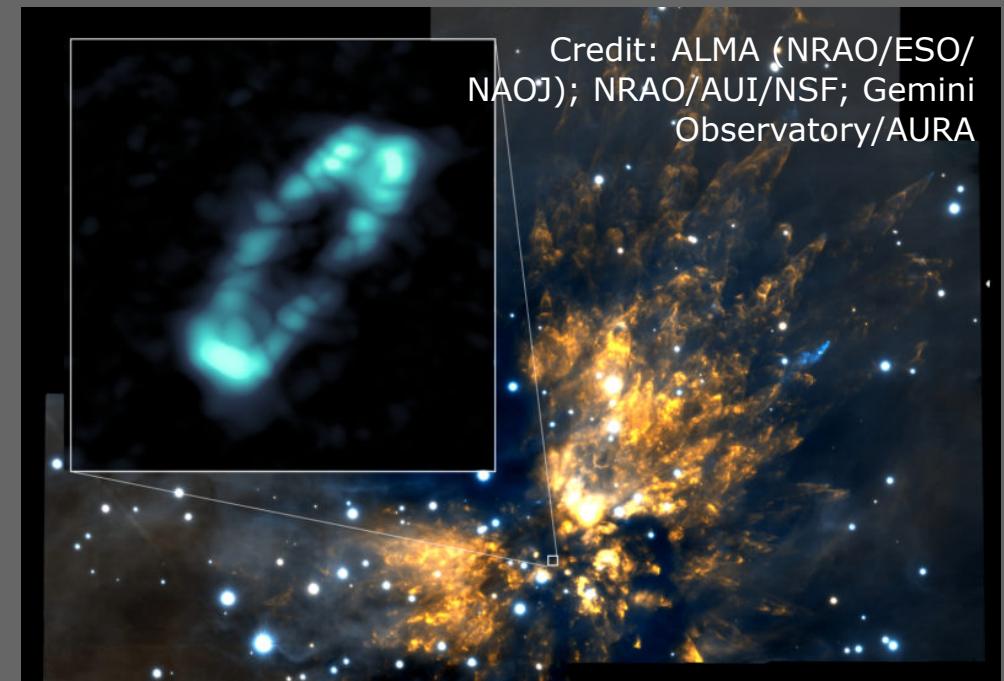
laughing gas

NH_3

ammonia

$\text{CH}_3\text{CH}_2\text{OH}$

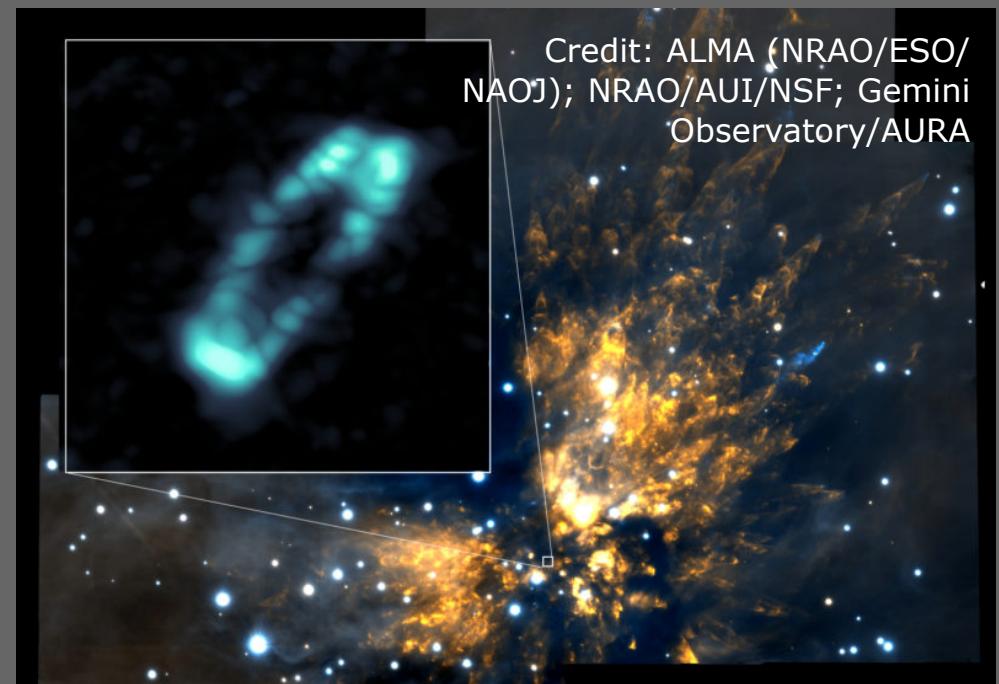
$(\text{CH}_3)_2\text{CO}$



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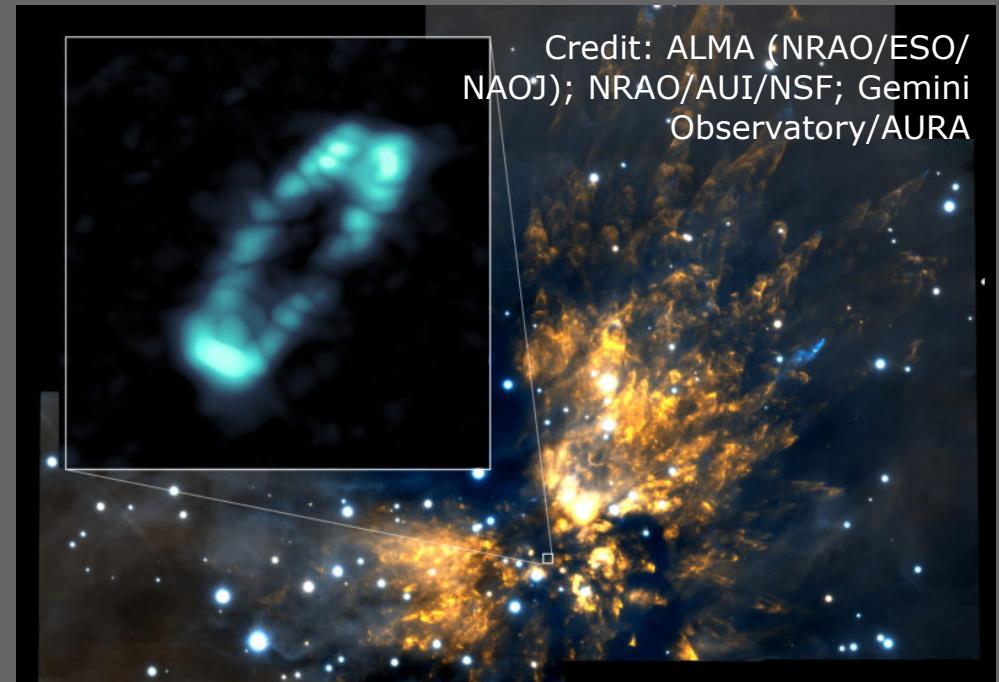
H_2O	water
CO	carbon monoxide
NaCl	salt
N_2O	laughing gas
NH_3	ammonia
$\text{CH}_3\text{CH}_2\text{OH}$	grain alcohol
$(\text{CH}_3)_2\text{CO}$	



Where are molecules found?

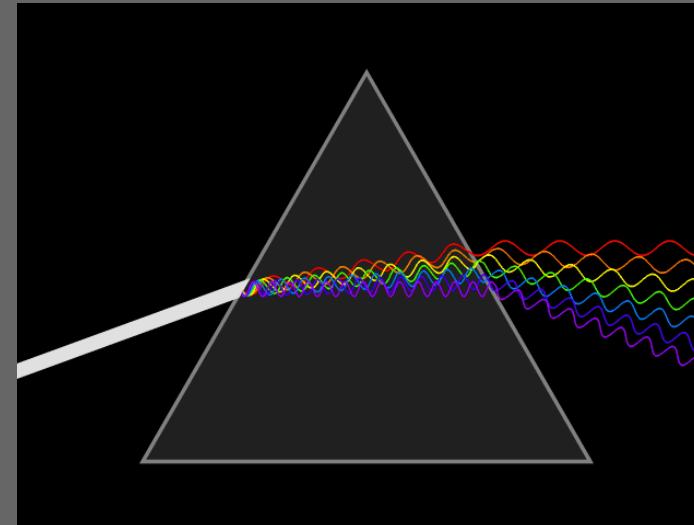
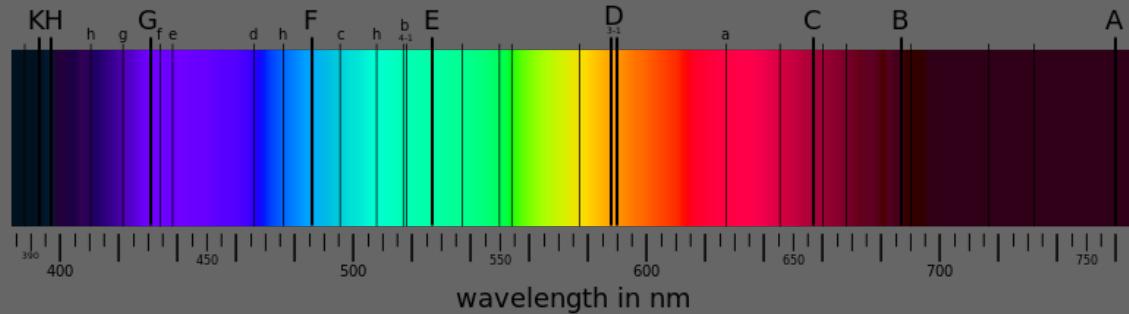
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CO	carbon monoxide
NaCl	salt
N_2O	laughing gas
NH_3	ammonia
$\text{CH}_3\text{CH}_2\text{OH}$	grain alcohol
$(\text{CH}_3)_2\text{CO}$	acetone



How do we detect atoms/molecules in space?

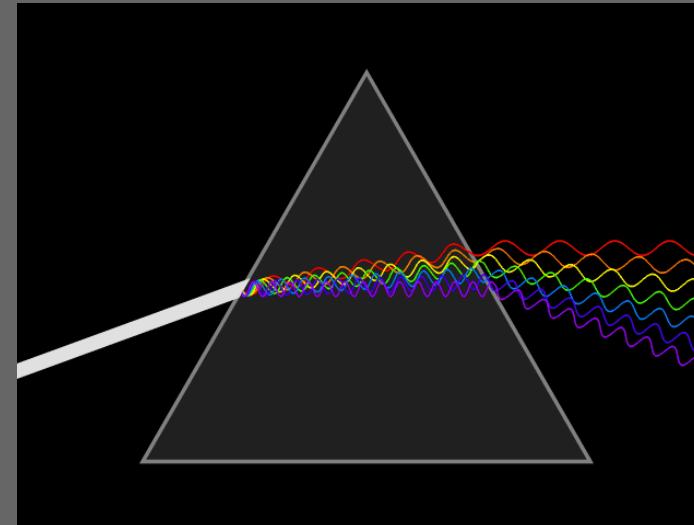
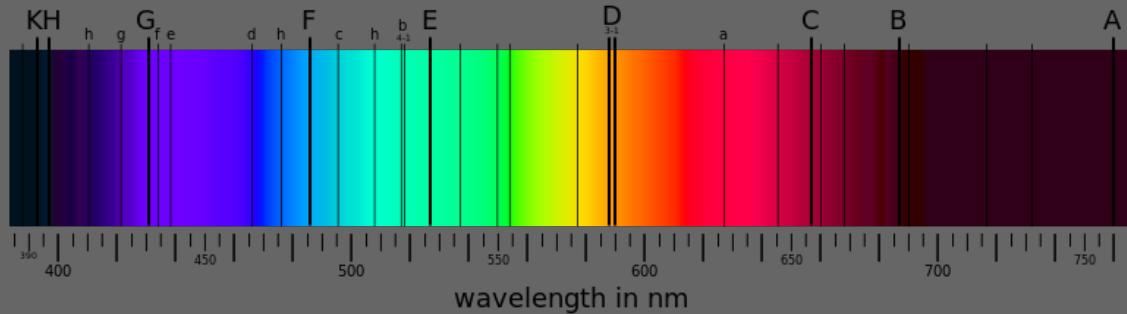
- Light is an astronomer's go-to tool
- Like molten metal, atoms and molecules in space emit light when they are hot (called a **spectrum**)
- The light emitted by a certain compound is unique to that compound



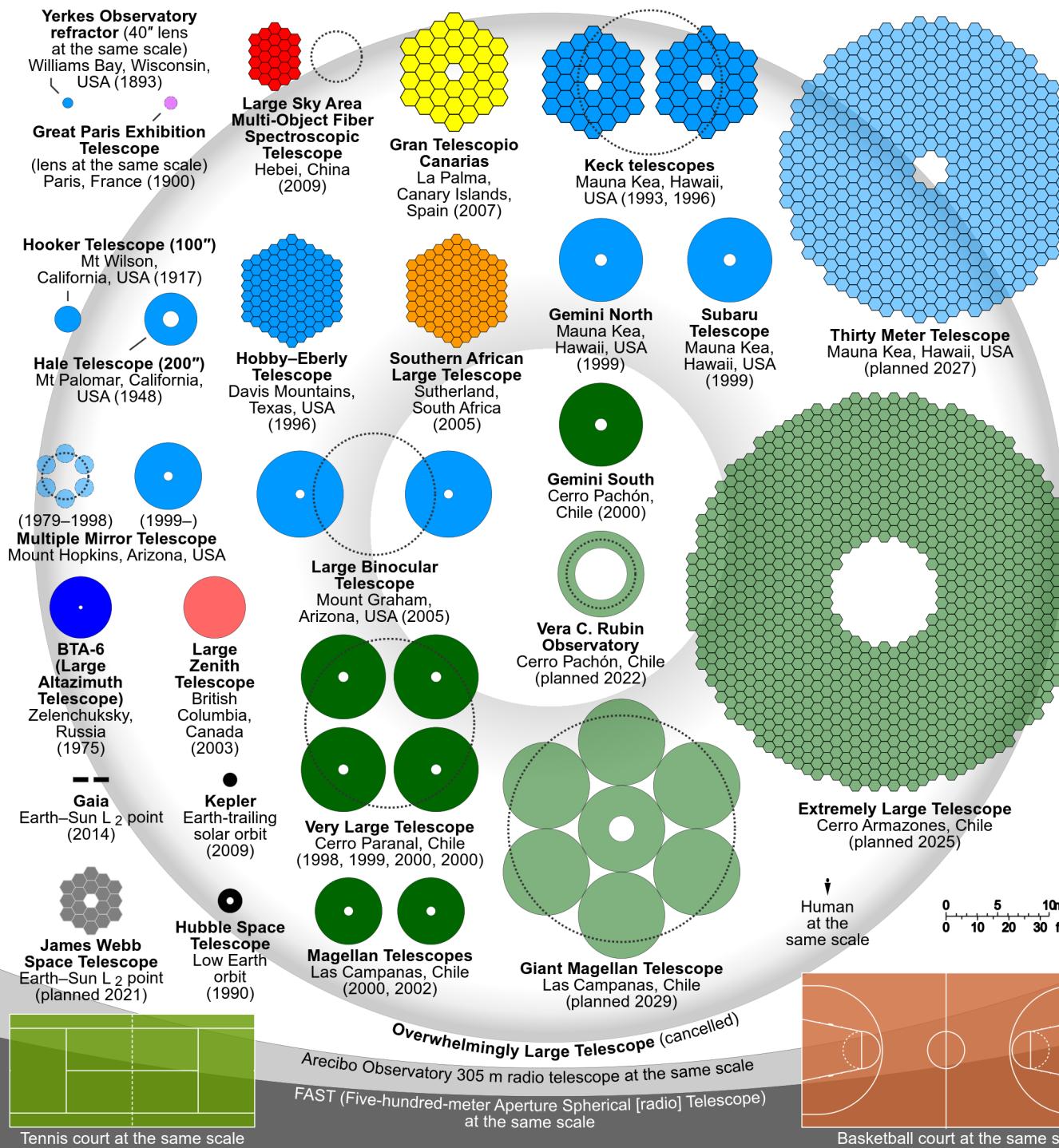
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Telescopes (the best part of my job)



Source: Wikipedia (Pachango)



Source: Wikipedia (Cmglee)



Me and astronomy

- My interest in astronomy was piqued while watching the TV show *Cosmos* (2014)
- I liked doing physics problems more than writing papers (but I didn't like math), and looking at pretty pictures of space is a nice part of the job!
- Useful classes: physics, geometry, computer science



A day in my life as a graduate student

- I use the Atacama Large Millimeter/submillimeter Array (ALMA) to figure out which atoms and molecules are present at different stages of a newborn star's early life
- In reality...
 - *Homework and projects*
 - *Writing code*
 - *Making and editing images from raw telescope data*
 - *Teaching/TAing*
 - *Meetings (on Zoom)*
 - *Observing*
 - *Outreach (like this presentation!)*



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Takeaways

- Elements are formed in a variety of places throughout the universe (and throughout time).
- Atoms can bond together to form molecules, which make up the materials (living and nonliving) that we encounter every day.
- Astronomers use light released by hot atoms and molecules to identify them.



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Learn more:
bit.ly/SEFSsite



Thank you!

I am happy to answer any
questions you have about
physics, astronomy, college,
or more!

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- Periodic table: https://en.wikipedia.org/wiki/Atom#/media/File:Nucleosynthesis_periodic_table.svg
- Covalent bonding: https://en.wikipedia.org/wiki/Covalent_bond#/media/File:Covalent_bond_hydrogen.svg
- Salt in the Orion Source I disk: <https://public.nrao.edu/gallery/alma-image-of-salt-in-orion-source-i/>
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