

# Lambdachrome® Laser Dyes

Ulrich Brackmann



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# Lambdachrome® Laser Dyes

3rd Edition

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All laser dyes mentioned in this book are immediately available from Lambda Physik. For more information please call or write. All major credit cards accepted.

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Third Edition (January 2000)

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# Preface to the 3rd Edition

Dyes and solvents from Lambda Physik periodically change in availability and composition. In order to ensure that researchers and users of Lambda Physik products have the most up-to-date information possible, this 3rd Edition of Lambdachrome® Laser Dyes is provided as an update to previous editions.

Lambda Physik reserves the right to modify any information given herein. Every effort is made to ensure utmost accuracy; however, no liability is assumed for errors occurring in the tables.

January 2000

# Preface to the 2nd Edition

The second edition of Lambdachrome® Laser Dyes has been updated with additional dyes, dye laser measurements, and evaluation of the latest literature. In this way it should be possible for any dye laser user to get a quick overview of the most important information regarding the use of laser dyes.

Meanwhile, the standard dye laser technique has been completed by several solid state lasers showing very interesting features such as high operating lifetime of the active medium and high output power in a small IR spectral range (compared to the accessible fundamental tuning range of an excimer laser pumped dye). However, the use of laser dyes still guarantees:

- easy tunability over a wide range of frequencies or wavelengths without the need of changing gratings or mirrors
- hyperfine tuning
- high average power in pulsed and cw operation mode
- high peak power
- · ideal light source for the generation of ultrashort pulses

February 1994

# Preface to the 1st Edition

The intention of this book is to give information on the most frequently asked questions about commercially available dyes, their chemical formulas, lasing wavelengths, solvents, pump sources, performance, and literature references. The following topics are discussed:

- Solvents frequently used for laser dyes and their general and physical properties, safety precautions, and waste disposal procedures
- General considerations about the choice of a solvent for laser dyes, solvent handling, preparation of dye solutions, and the cleaning of the dye circulation system
- Precautions for the use of laser dyes and their solutions
- All Lambdachrome® laser dyes are listed, including more than one hundred dyes along with their chemical names, chemical structures, trivial names, and trade names
- Absorption and fluorescence data in addition to appearance
- The absorption spectrum of each dye
- The application of the dyes in a variety of solvents under the most common pumping conditions as well as the dye laser characteristics (peak, range, efficiency, concentration, and solvent)
- Tuning curves for excimer, nitrogen, and ion laser pumped dye lasers
- Several hundred literature references

This book is written for general users of dye lasers. Researchers of laser dyes may refer to the mentioned publications for more detailed information.

July 1985

# Dye Lasers and Laser Dyes

Dye lasers are "the fulfillment of an experimenter's pipe dream that was as old as the laser itself: To have a laser that is easily tunable over a wide range of frequencies or wavelengths" (Schaefer, 1977). Dye lasers can be pumped by incoherent or laser sources, both pulsed and continuous wave (CW), and offer the possibilities of broad wavelength control, multijoule pulsed operation, ultranarrow linewidths, or ultrashort pulses. They are conveniently divided into three broad technological categories:

- Continuous-wave jet-streamed dye lasers can provide narrow CW bandwidths and can be synchronously pumped or passively modelocked to generate short pulses.
- Flashlamp-pumped dye lasers have a larger bandwidth and less wavelength stability than CW laser-pumped dye lasers, but they have the advantage that large volumes of active dye medium can be pumped, yielding large output pulse energies and active powers.
- Dye lasers that are pumped by Cu vapor or nitrogen, excimer lasers, and the frequency-doubled or -tripled output from pulsed Nd:YAG lasers provide high peak powers. Also, the pulse duration of 4 to 60 nsec makes narrow bandwidths and a high spectral purity possible.

Typical output characteristics from commercial dye lasers are shown in Table 1.

# Continuous-Wave Dye Lasers

CW laser systems consist of three major elements: the optical resonator, the dye flow system, and the tuning element. The resonator is responsible for maintaining a rigid optical alignment of all cavity components. The dye

	Commercial Dy	Table 1. e Laser Outpu	t Charact	eristics	
Pump Source	Argon/Krypton	Ion Las	er Flashl	amp Nd:YA0	Excimer
	CW	Mode-Locked	/	Laser	Laser
		Cavity Dumpe	d		
Tuning Range [nr	n] 380-950	580-880	335-850	410-880	320-1,024
Average Power [V	V] 5	0.1	3	2	10
Repetition Rate [	Hz] CW	3.8M	2-30	10-40	1-500
Peak Power [kW] 10,000	-	10	7,00	00 20,	000
Energy/pulse [m.	] -	0.01	3,500	10-120	40-120
Pulsewidth [nsec	] -	0.0001	260-	600 5-10	7-250
Linewidth [GHz]	0.0005 to 40	Various	2	0.6	0.15

flow system, consisting of dye, circulation pump, and dye jet nozzle, must provide an optically flat stream of dye across the laser's optical axis. The tuning element allows the user to continuously tune to the required output wavelength.

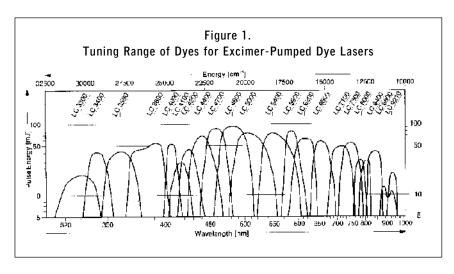
Excitation in a CW dye laser is provided by an input pump beam from another CW light source, typically an ion laser. This beam is focused onto the dye stream and causes an extremely high level of fluorescence. The fluorescence is focused between two highly reflective concave mirrors that feed back the fluorescent light, initiating the lasing process.

The dye laser emission described above is broadband, typically exhibiting a 40-GHz linewidth. Many applications require narrow-linewidth single-frequency operation. This is accomplished by inserting one or two etalons into the dye laser cavity. A dye laser with an etalon, when coupled with electronic stabilization, can narrow the linewidth to less than 1 MHz.

### Pulsed Dye Lasers

Of the various pulsed dye lasers, two types tend to dominate because of their versatility, broad spectral coverage and high output power. They are Nd:YAG and excimer pumped dye lasers.

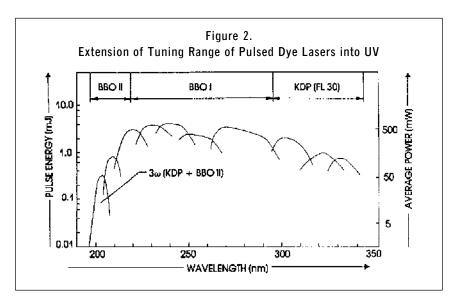
The use of Nd:YAG and excimer pump lasers allows dye laser pulses to be produced at energy levels up to 100 mJ or more. Their performance is wavelength-dependent. At the long wavelengths, more energy is provided by pumping with a frequency-doubled Q-switched Nd:YAG than with an excimer laser. However, UV-emitting excimer pumps allow dye lasers to operate directly at UV wavelengths down to 308 nm and at repetition rates of 500



Hz. The strongest Nd:YAG lasing line is at 1064 nm, which is unsuitable to pump dyes. Fortunately, its high peak power and near-diffraction-limited beam quality lend themselves to rather efficient frequency doubling, tripling and quadrupling. This provides suitable pump wavelengths of 532, 355, and 266 nm. Considerable engineering effort has gone into reducing the intensity fluctuations that are amplified by the inherent nonlinear frequency-doubling process. The tuning range can be covered by using one of these pump wavelengths and an appropriate dye.

The Nd:YAG laser has certain advantages with respect to the efficient pumping with certain dyes, especially when pumped longitudinally, i.e., collinearly to the laser beam. Rhodamine 6G, for example, can be excited very efficiently in this manner (>50 percent). Since excellent beam quality is achieved with the longitudinal arrangement, one also gets high conversion rates with all nonlinear frequency changes (by frequency mixing, 367 nm, and by frequency doubling, 280 nm).

The discovery of rare gas halide lasers, i.e., the excimer laser, in 1976, introduced a new and powerful tool for dye laser pumping. It combines the advantages of the nitrogen system (high repetition rate, ease of operation, and low cost) with those of the Nd:YAG system (high peak power and sufficiently long pulse duration). Furthermore, the excimer laser is scalable to very high repetition rates (>1 kHz) and high pulse energies (>1 J).

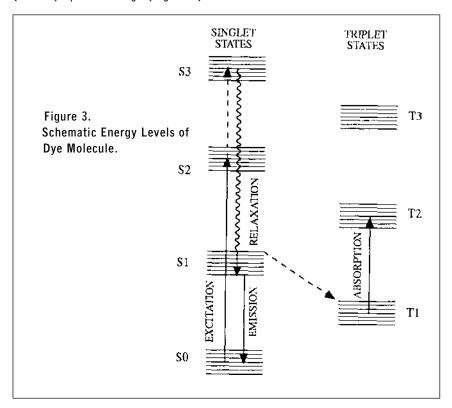


Excimer laser wavelengths range from 193 nm (ArF) to 351 nm (XeF). The strongest lines are at 248 nm (KrF) and 308 nm (XeCl). Most dyes have an absorption band at 308 nm. The result is that nearly the entire range of commercially available dyes can be pumped with a XeCl laser (see Figure 1).

A major advantage of pulsed dye lasers over CW is the ease with which the output can be frequency-doubled, thus extending the tuning range into the UV. Using a combination of KDP and BBO (beta barium borate) crystals, wavelengths from 207 to 350 nm can be produced. Further extension down to 197 nm can be achieved by frequency mixing (see Figure 2).

### Dyes

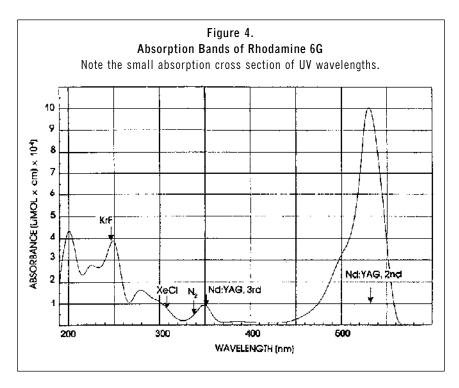
Dyes, either as solutions or vapors, are the active medium in pulsed and CW dye lasers as well as ultrafast shutters for Q-switching and passive modelocking. They emit in a comparatively narrow spectral region (typically 30 nm); thus a variety of dyes is necessary in order to cover the entire (visible) spectral range (Figure 1).



The optical excitation of dyes corresponds to transitions of molecules in the singlet state, with the absorption  $S_0 \dashrightarrow S_1$  being the strongest (see Figure 3), and is specific for each dye molecule. For optimum pumping  $(S_0 \dashrightarrow S_1)$  of the various dyes, one would therefore need a number of pump-laser wavelengths.

Fortunately, nearly all dyes have additional absorption bands in the UV range (see Figure 4). These absorptions correspond to transitions to higher singlet states (Figure 3) from which fast internal relaxation processes lead to the upper laser level ( $S_1$ ) with high quantum efficiency. This is the reason most dyes can be pumped by a single UV laser. However, the attractive excitation scheme of one pump laser for all dyes brings other problems:

(a) The inner efficiency of dye lasers is lower as a result of excitation in higher S-states because a considerable part of the excitation energy is converted into heat (large Stokes shift). However, the high efficiency of pulsed lasers more than compensates for this disadvantage.



(b) A multiphoton excitation can lead to destruction of the cell and the solvent molecules. In this process, a previously excited molecule absorbs additional photons (sequential absorption), or a molecule absorbs several photons at the same time. In these absorption processes, the molecule can absorb so much energy that the binding energy is surpassed, and the molecule dissociates, or at least changes, its structure. This process is more probable during excitation with UV light than with visible light. Thus, one must expect a reduced photostability of the dye when pumping with UV light.

The photostability of the most common dyes is summarized in Table 2 (Antonov and Hohla, 1983). For example, the value of 50 Wh/l corresponds to  $5 \times 10^5$  total shots as the point at which the dye laser energy falls to 50 percent of the initial value, i.e., when the dye solution (one liter) is excited with 360 mJ per pulse at 10 Hz.

(c) Another problem results from the small absorption cross section at short wavelengths (Figure 4). To excite as many molecules as possible, a very high pump power density  $I_p$  ( $I_p$  being inversely proportional to the absorption cross section), or high dye concentrations is required.  $I_p$  is limited to values <30 MW/cm² due to the stability of most solvents. To

Table 2. Photochemical Stablility of Laser Dyes			
Dye	Center of	Excimer-pumped	CW laser-pumped
	Emission [nm]	] [Wh]	[Wh]
p-Terphenyl	340	451	-
QUI	380	1457	-
Polyphenyl 1	380	870	-
Stilbene 1	410	10	200
Stilbene 3	430	14	300
Coumarin 2	450	31	100
Coumarin 47	470	45	-
Coumarin 102	480	244	100
Coumarin 30	515	-	100
Rhodamine 6G	590	316	1000
Rhodamine B	610	144	200
DCM	650	348	500
Rhodamine 700	700	80	1000
Styryl 9	840	73	500
HITCI	875	12	100
IR 140	960	10	100

keep  $\boldsymbol{I}_{_{\boldsymbol{p}}}$  as low as possible, high-power dye lasers have to be pumped transversely.

Transverse pumping configurations have been used for some time, and in contrast to the end-on pumped configurations, they lead to a non-Gaussian energy distribution, for which rising spatial filters can compsensate.

There are two types of transverse-pumped dye laser cells: the longitudinal and the transverse flowing. The transverse configuration, in which the dye flows vertically to the dye laser, results in a high repetition rate, whereas longitudinal flowing, in which dye flows in the direction of the dye laser, is characterized by a symmetrical energy distribution but small repetition rates.

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# Dye Solvents

The heart of any dye laser is the laser dye. Either dissolved in an appropriate solvent or in a vapor state, these very often highly colored substances play the major role in the overall performance of any dye laser. Both pulsed and continuous operation is possible. In addition, their unique photophysical properties make them ideal candidates for the generation of ultrashort light pulses. With mode-locking, pulses shorter than 100 femtoseconds have been obtained. Hyperfine tuning of the output has permitted many exciting experiments in spectroscopy.

# Solvents for Laser Dyes

Prepared laser dye solutions usually contain very small quantities of dye. Typical dye concentrations are  $10^2$  to  $10^5$  molar. For this reason, the solvent in which the dye is dissolved plays an important role when defining physical properties and potential hazards.

Lasing wavelength and energy are very sensitive to the choice of solvent. Most laser dyes are polar molecules, and excitation into their lowest-lying singlet state is accompanied by an increase in the dipole moment. Accordingly, solvent polarity plays an important role in shifting the lasing wavelength. In a majority of circumstances, increasing solvent polarity will shift the gain curve toward longer wavelength. In the case of more polar dyes, the shift can be as high as 20-60 nm. Table 3 gives an impression of this effect (Drexhage, 1973).

Some solvents cannot be used with longer wavelength lasing dyes because the solvents have vibrational overtones that interfere with the lasing process. Solvents such as water, methanol and ethanol, which would appear to be optimal for many dyes, are often not useful solvents for near-IR and IR dyes

Maximur	n of Main Absorption	ole 3. on Band in Diffe	rent Solvents
RHOD	AMINE 6G	COUM	ARIN 102
Solvent	Labs [nm]	SolventLabs [nm]	
HFIP	514	HFIP	418
TFE	516	TFE	405
Et0H	530	MeOH	390
DPA	537	NMP	383
DMSO	540		

because of the presence of hydroxyl group overtones in this spectral range. Accordingly, the solvent DMSO or polychlorinated aromatics, which lack hydroxyl resonances, are commonly used for dyes that lase in the near-IR and IR regions of the spectrum. Unfortunately these solvents are toxic. DMSO especially facilitates the transfer of toxic dyes through the skin and into the body if accidentally spilled. Therefore, it is strongly recommended that all laser dyes and solutions containing laser dyes are handled in well ventilated environments. All individuals handling the solutions should wear rubber gloves.

A summary is given below of the general and physical properties, application, and safety of the most frequently used solvents for laser dyes. Most of this information has been taken from the literature cited at the end of this article.

### Choice of Solvent

Although very often a specific solvent is recommended for use with a particular dye, it is important to recognize that other solvents can also be used, particularly if the user is interested in shifting the gain curve to different wavelengths.

The following are criteria for choosing appropriate solvents for laser dyes:

- a) The solvent must be transparent at the pump wavelength and the emission wavelength of the dye laser.
- b) The dye should be soluble in the solvent under consideration. In all cases, the rule "similia similibus solvuntur" applies, meaning that the solubility of nonpolar dyes, e.g. PTP, in nonpolar solvents, e.g. cyclohexane, is greater than in polar solvents, e.g. methanol, and vice versa.
- c) The solvent must be photochemically stable when exposed to the pump light. In particular, solvents containing chlorine, such as chloroform, and secondary alcohols, such as isopropanol, are not useful as solvents for laser dyes because of their low photochemical stability.
- d) Some solvents are often not useful for near-IR and IR dyes due to the presence of hydroxyl group overtones in this spectral region (see above).

The following solvents are recommended for different pumping conditions:

- a) Those appropriate for pumping with a pump source emitting below 300 nm (e.g., KrF: 248 nm, Nd:YAG 4th harmonic: 266 nm):
  - Cyclohexane
  - Ethylene glycol
  - Glycerol
  - Trifluoroethanol
  - p-Dioxane

- Ethanol
- Methanol
- Hexafluoroisopropanol
- b) Solvents appropriate for pumping with a pump source emitting between 300 and 400 nm (e.g., XeCl: 308 nm, nitrogen: 337 nm, Nd:YAG 3rd harmonic: 355 nm), in addition to those listed in (a):
- N,N-Dipropylacetamide
- Dimethylsulfoxide (DMSO)
- N,N-Dimethylformamide (DMF)
- Tetrahydrofurane
- 1-Methyl-2-pyrrolidinone (NMP)
- Tetrahydrothiophenedioxide (sulfolane)
- c) Appropriate solvents for pumping in the visible and near-IR spectral range (e.g., Nd:YAG 2nd harmonic: 532 nm, CW-ion lasers), in addition to those listed in (a) and (b):
- Toluene
- Chlorobenzene
- Chloroform
- Benzylalcohol
- 1,2-Dichloroethane
- 1,1,1-Trichloroethane
- o-Dichlorobenzene
- Dichloromethane

# Purity of Solvent

The output power of dye lasers is strongly dependent on the purity of the solvent. Impurities and additives may strongly affect upper state lifetime of the dye or may catalyse photochemical reactions. Therefore, for best results, only high quality solvents are to be recommended.

Very often we are asked whether it is necessary to use spectroscopic grade solvents. Our experience is that it is NOT necessary. Commercially available qualities of the grade "pro analysis" (p.a.) or "for synthesis" are sufficient for dye laser applications. Some suppliers will call them HPLC.

However, bulk quantities of these solvents are very often of poor quality and are not offered as p.a. grade. Under such circumstances, it is highly recommended that the transparency of the solvent is checked in a 1 cm cuvette at the pump wavelength of the dye laser with a simple absorption spectrometer. The measured value should be greater than 98 percent.

# Solvent Handling

With the exception of water, all solvents should be considered hazardous. In many instances, the solvent in which the dye is dissolved plays a major role in the hazard presented by the final solution. Some of the solvents listed above are highly toxic, irritants, narcotics, and/or anaesthetics. These hazards must be addressed carefully in dye handling and solution preparation.

Nearly all solvents are highly flammable. Therefore, a small fire extinguisher should be installed near the laser in a readily accessible and unobstructed area.

A particular fire hazard that is not commonly known occurs with nonpolar and, hence, nonconductive solvents. If these solvents are circulated at a high speed through plastic tubings, the pump unit acts as a van de Graff generator, producing up to 100 kV, and sparks may pierce the tubing and ignite the solvent. The dye selectors use grounding wires inside the plastic tubings to eliminate these problems. However, when using such solvents, check first for static electricity before opening the reservoir. Static electricity is present when hair on the back of your hand or forearm is attracted to the plastic tubing. Do not circulate dye solutions made with such solvents for more than a minute, unless the cuvette has been placed into the crate and is grounded.

# Preparation of the Dye Solution

As a rule of thumb, the dye concentration is selected to absorb 90 percent of the pump light within 0.5 mm, or the dye solution has to have an absorbance of OD = 2/mm for the wavelength of the pump light.

When in doubt about the concentration of the dye solution, measure the absorbance of the dye solution used for the oscillator in a spectrophotometer, using either a 1 mm cuvette or, after diluting by a factor of 10, a 10 mm standard cuvette. The cuvettes must be of fused-silica type if you measure the absorbance for a pump wavelength smaller than 300 nm. The absorbance of your oscillator dye solution should be within the range of OD = 1.8 to 2.2/mm. For an unknown dye, dissolve a known amount (a few milligrams) in a known volume and measure the absorbance at the pump wavelength in use; the dye concentration (in g/I) is then calculated according Beer's law:

The measurements should, of course, be made against pure solvent in a cuvette of the same pathlength in the reference beam of the spectrophotometer.

A concentration of OD = 2/mm is convenient for the oscillator solution. The amplifier needs only 1/3 this concentration. The amplifier solution is prepared by filling the reservoir bottle of the amplifier circulator with one part stock solution and then adding two parts pure solvent.

To prepare the dye solution, weigh out the amount of dye and transfer it into a 500 ml (or 1 liter) glass bottle. If some are available, use brown bottles. Make sure that the entire measure of dye is transferred to the bottle and be careful not to spill it. Most errors occur at this step. Fill the bottle to the 500 ml (or 1 liter) mark. Adding 10 percent more or less solvent does not affect the operation appreciably. Make the stock solution "fatter" than recommended, since solvent can always be added directly to the reservoir of the circulators later on. Some dyes do not dissolve instantly. Use of an ultrasonic bath is recommended.

#### NOTICE:

Do not use the dye solution until it is completely translucent and no floating dye particles are observed.

Some dyes have to be used close to their saturation level. It may take some dyes up to 30 minutes to dissolve completely. If necessary, heat gently and with caution.

# Optimizing the Dye Solution

There is an optimum concentration for a given dye gain, i.e. for a given dye, wavelength, and input power. This optimum concentration may vary from that giving an optical density of OD = 2/mm. Higher concentrations will cause a slightly red shift the tuning curves, while lower concentrations will result in a blue shift.

Optimization of the dye concentration is accomplished by adding either pure solvent or solution of higher concentration than that recommended in small increments to the solution in the dye circulation system until power is at maximum.

The concentrations for the amplifier solutions of excimer and Nd:YAG-pumped dye lasers are about 1/3 and 1/6 of the oscillator solutions, respectively.

# Cleaning the Dye Circulation System

Should it be necessary to change the dye solution, it is only necessary to rinse three times with 100 to 200 ml of solvent. It is recommended that the dye filter be changed. If it is not possible to change the dye selector when changing the dye, especially when going from a longer wavelength to a shorter one, the cuvette, the reservoir, the pump, and the tubings must be carefully cleaned and

the filter must be exchanged. The cleaning process is complete only when no fluorescence is observed after rinsing for approximately 15 minutes. It is advisable to use small quantities of solvent and few rinses. The rinsing process can take many hours. Low output power on changing the dyes can often be traced to dye residue in the pump and the tubings.

# Precautions for the Use of Laser Dyes

Cautious handling of dyes and dye solutions is advised, especially because those used in the infrared may be toxic or because solvents such as DMSO, Methanol, Dioxane, and Benzyl Alcohol have the ability to carry their solutes through the skin and into the body (Mosovsky, 1983; Kues, 1975).

In most cases the exact toxicity of laser dyes is not well known, but they should, like all chemicals, be considered dangerous until proven otherwise. The safest precaution is to use butyl rubber gloves when handling the dye solutions and to immediately clean any skin that comes into contact with the dye solutions or the dye itself.

Solvents should be kept away from heat, sparks, and open flames because they are extremely flammable or combustible. They should be handled in a hood due to their stench or potential danger if inhaled.

Lambda Physik provides, on request, a complete set of Material Safety Data Sheets (MSDS). These data sheets give more information on laser dye toxicity, hazards, and recommended controls. As already mentioned above, the exact toxicity of laser dyes is not well known in most cases. Therefore, it is important to know that the MSDS only describes general aspects of dye toxicity.

#### NOTICE:

The responsibility for the safe use of our Lambdachrome laser dyes rests with the user.

### Hazards

Solvents should be handled only by qualified people trained in laboratory procedures and familiar with their potential hazards. Some solvents are highly toxic, irritants, narcotics, and/or anaesthetics. Some form hazardous compounds upon decomposition; others are highly reactive. In the following tables, hazard warnings and literature references, such as Sax (Dangerous Properties of Industrial Materials) and the Registry of Toxic Effects of Chemical Substances (RTECS) are provided, so that the information about possible hazards are available to the trained technical person using the dye, solvent, and/or dye solution.

The absence of a warning must not be interpreted as an indication of safety. In several cases information is not available on the possible hazards of many compounds.

# Waste Disposal Procedures

The disposal methods outlined below are intended as guides to the users of laser dye solutions or solvents. Careful consideration must be given to the chemical and physical properties of the substances. In addition, local laws and regulations may preclude the use of these methods which are primarily designed for quantities of one to five liters. All federal, state, and local laws concerning health and pollution must be observed.

### **Definitions**

#### **Boiling point**

The temperature at which the vapor pressure of the liquid is equal to the opposing pressure. Values listed in the tables refer to an opposing pressure of 760 torr unless otherwise stated.

#### Density

The density of a substance is defined as the mass per unit of volume.

#### Dielectric constant

A measure of the relative effect a solvent has on the electronic force with which two oppositely charged plates are attracted to each other.

### Flash point

The flash point is usually not considered a common physical property. It is included because of its widespread use in classifying solvents for storage and shipping.

# Ionization potential

The work (expressed in electron volts) required to remove a given electron from its atomic orbit and place it at rest at an infinite distance.

### Melting point

The temperature at which a solid compound changes into the liquid state.

### Minimum ignition temperature

The minimum temperature at which, under certain conditions, the mixture combined with air may ignite.

#### Molecular weight

The sum of the atomic weights of all atoms in a molecule.

#### Refractive index

The the ratio of the velocity of light in a particular substance to the velocity of light in vacuum. Values usually reported refer to the ratio of the velocity in air to that in the substance saturated with air.

#### Threshold Limit Value (TLV)

The maximum permissible concentration of a chemical that is permissable for prolonged exposure. The TLV gives a conentration of vapors to which an average sized person can safely be exposed for 8 hours per day, 5 days per week.

#### Viscosity

The coefficient of viscosity is defined as the force per unit area necessary to maintain a unit velocity gradient between two parallel planes a unit distance apart.

# Benzyl Alcohol

Phenylcarbinol

# General Properties

Colorless liquid with aromatic odor. It is only slightly soluble in water.

# Physical Properties

Molecular weight: 108.14
Freezing point (°C): -15.3
Boiling point (°C): 205.45
Flash point (°C): 100
Min. ignition temp. (°C): 435

Density (g/cm<sup>3</sup>):  $1.0493^{15}$   $1.0413^{25}$   $1.03765^{30}$ 

Refractive Index: 1.54035<sup>20</sup> 1.453837<sup>25</sup>

Viscosity (cPoise): 7.76<sup>15</sup> 4.65<sup>35</sup>

Dielectric constant: 13.1 Ionization potential (eV): 9.14

Solubility: 0.08% in water, organic solvents

Optical properties:

380 Wavelength (nm) 290 300 310 320 340 350 Transmission (%) 4 23 70 85 90 95 98

# Application

Benzyl alcohol, due to its high viscosity, is a frequently used solvent in jet stream dye lasers. Its dissolving capacity makes it suitable for polar dyes like Coumarins.

# Safety

# RTECS # DN3150000; Sax 5, 409 · TLV:

#### Hazards

Harmful by inhalation and if swallowed. Benzyl alcohol is a toxic solvent. It is believed that benzyl alcohol that is present in a poorly ventilated area is the cause of violent headaches, vertigo, nausea, and other symptoms.

### Safety Precautions

In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. Don't smoke, eat, or drink when handling the solvent. Keep all containers and solutions tightly closed.

### WasteDisposal Procedures

Do not dispose of Benzyl Alcohol in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Cyclohexane

Hexahydrobenzene

# General Properties

Colorless, water insoluble, highly flammable liquid. Sweet, aromatic odor.

# Physical Properties

Molecular weight: 84.16
Freezing point (°C): 6.54
Boiling point (°C): 80.73
Flash point (°C): -18
Min. ignition temp. (°C): 260

Densitiy (g/cm³): 0.7785<sup>25</sup> 0.7788<sup>25</sup> Refractive Index: 1.4262<sup>20</sup> 1.4235<sup>25</sup> Viscosity (cPoise): 0.98<sup>20</sup> 0.898<sup>25</sup>

Dielectric constant: 2.023 Ionization potential (eV): 9.8

Solubility: 0.01% in water, nonpolar organic solvents

Optical properties:

Wavelength (nm) 200 210 220 230 240 250 300 Transmission (%) 5 15 45 75 90 98 100

# **Application**

Suitable solvent for nonpolar laser dyes like p-Terphenyl. Its high optical transparency in the UV allows the application in dye lasers pumped below 300 nm. The photochemical stability of Cyclohexane is poor.

# Safety

# RTECS # GU6300000; Sax <u>6</u>, 831 · TLV: 300 ppm or 1050 mg/m<sup>3</sup> Hazards

Highly flammable. Absorbed by inhalation. The vapor is mildly irritating to the mucous membranes. The liquid is a fat solvent and thus irritates the skin.

### Safety Precautions

Keep container in a well-ventilated place. Keep away from source of ignition. Take precautionary measures against static dis-charges. Avoid skin contact. Do not smoke, eat, or drink when handling the solvent. Keep all containers tightly closed away from sparks and open flames.

### Waste Disposal Procedures

Do not dispose of Cyclohexane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Dichloroethane

Ethylene chloride · 1,2-Dichloroethane

# General Properties

Dichloroethane is a colorless, water-insoluble liquid with a sweet odor.

### Physical Properties

Molecular weight: 98.96
Freezing point (°C): -35.7
Boiling point (°C): 83.5
Flash point (°C): 13
Min. ignition temp. (°C): 440

Density (g/cm $^3$ ): 1.2531 $^{20}$  1.2458 $^{25}$  1.2383 $^{30}$ 

Refractive Index: 1.4448<sup>20</sup> 1.4421<sup>25</sup> Viscosity (cPoise): 0.887<sup>15</sup> 0.73<sup>30</sup>

Dielectric constant: 10.36 Ionization potential (eV): 11.12

Solubility: 0.81% in water, organic solvents

Optical properties:

Wavelength (nm) 270 300 225 230 240 250 260 Transmission (%) 10 30 80 95 98 98 98

# **Application**

See Dichloromethane.

# Safety

# RTECS # KI0525000; Sax 6, 944 · TLV: 50 ppm or 200 mg/m<sup>3</sup>

Highly flammable. Harmful by inhalation. One can become adapted to the odor of Dichloroethane at low concentrations, therefore it cannot be considered as a reliable warning. The acute and chronic effects of the solvent can be significant.

### Safety Precautions

Keep container tightly closed. Keep away from sources of ignition. Do not smoke. Take precautionary measures against static discharges.

#### Waste Disposal Procedures

Do not dispose of Dichloroethane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Dichloromethane

Methylene chloride

# General Properties

Dichloromethane is a colorless, water-insoluble liquid with a sweet odor.

# Physical Properties

Molecular weight: 84.93
Freezing point (°C): -95.1
Boiling point (°C): 39.8
Flash point (°C): 0
Min. ignition temp. (°C): 605

Density (g/cm $^3$ ): 1.3348 $^{15}$  1.3168 $^{25}$  1.3078 $^{30}$ 

Refractive Index: 1.4242<sup>20</sup> 1.4212<sup>25</sup> Viscosity (cPoise): 0.449<sup>15</sup> 0.393<sup>30</sup>

Dielectric constant: 8.93 Ionization potential (eV): 11.35

Solubility: 1.3 % in water, organic solvents

Optical properties:

Wavelength (nm) 300 230 240 250 255 270 290 Transmission (%) 10 70 95 98 98 98 98

# **Application**

Dichloromethane is used as solvent for laser dyes and saturable absorbers absorbing in the near infrared spectral region.

# Safety

# RTECS # PA8050000; Sax <u>6</u>, 1763 · TLV: 500 ppm or 1750 mg/m<sup>3</sup> Hazards

Harmful by inhalation. The toxic effect of Dichloromethane is predominately narcosis. It is mildly irritating to the skin on repeat contact if free to evaporate. It is painful to the eyes but no permanent damage may be expected.

### **Safety Precautions**

Avoid contact with skin. Do not smoke, eat, or drink when handling the solvent.

#### Waste Disposal Procedures

Do not dispose of Dichloromethane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Dioxane

p-Dioxane

# **General Properties**

Colorless, volatile, and very hygroscopic liquid with slightly aromatic taste.

Physical Properties

Molecular weight: 88.11
Freezing point (°C): 11.8
Boiling point (°C): 101.32
Flash point (°C): 12
Min. ignition temp. (°C): 375

Densitiy (g/cm³): 1.0336<sup>20</sup> 1.028<sup>25</sup> Refractive Index: 1.42241<sup>20</sup> 1.42025<sup>25</sup> Viscosity (cPoise): 1.439<sup>15</sup> 1.087<sup>30</sup>

Dielectric constant: 2.209 Ionization potential (eV): 9.13

Solubility: in water, organic solvents

Optical properties:

Wavelength (nm) 225 240 250 260 280 290 300 Transmission (%) 40 50 60 70 85 93 98

# Application

Its high photochemical stability and excellent dissolving capacity makes dioxane a versatile solvent for UV and Coumarin dyes.

# Safety

# RTECS # JG8225000; Sax <u>6</u>, 1227 · TLV: 100 ppm, 360 mg/m<sup>3</sup> Hazards

Highly flammable. May form explosive peroxides. Harmful by inhalation. Painful to the eyes and irritating to the skin upon prolonged contact. It can be absorbed through the skin in toxic amounts. Dioxane is insidious. Its vapors have poor warning properties; they are faint and inoffensive. Concentrations in air of 300 ppm cause irritation of the eyes, nose, and throat. The vapors can be inhaled in amounts that cause serious systemic injury.

### **Safety Precautions**

Keep container in a well-ventilated place. Keep away from sources of ignition. Take precautionary measures against static discharges. Avoid skin contact. Do not smoke, eat, or drink when handling the solvent. Keep all containers and solutions tightly closed.

### **Waste Disposal Procedures**

Do not dispose of Dioxane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# **DMSO**

Dimethyl Sulfoxide

# General Properties

Colorless, odorless, hygroscopic liquid with a slightly bitter taste. It has an equilibrium moisture content of 10 percent with air at 20°C.

# Physical Properties

Molecular weight: 78.13
Freezing point (°C): 18.54
Boiling point (°C): 189
Flash point (°C): 95
Min. ignition temp. (°C): 383

Density ( $q/cm^3$ ): 1.0958<sup>25</sup> 1.0816<sup>40</sup> 1.0616<sup>60</sup>

Refractive Index: 1.4783<sup>20</sup> 1.4773<sup>25</sup> Viscosity (cPoise): 1.996<sup>25</sup> 1.654<sup>35</sup>

Dielectric constant: 46.68

Ionization potential (eV):

Solubility: 25.3 % in water, organic solvents

Optical properties:

Wavelength (nm) 263 270 280 290 300 310 340 Transmission (%) 10 34 60 71 85 90 98

# **Application**

DMSO is an excellent solvent for polar dyes like Rhodamines.

# Safety

#### RTECS # PV6210000; Sax 6, 1201 · TLV:

#### Hazards

DMSO dehydrates and defats the skin, but seems to be relatively free from toxic effects. Its vapor, mixed with air, may explode above  $90^{\circ}$  C.

#### Safety Precautions

Dimethyl sulfoxide may produce eye, skin, and respiratory irritations. The solvent penetrates the skin and that toxic solutes are carried with it into the body fluid. Avoid contact with skin and eyes.

### Waste Disposal Procedures

Avoid mixing contaminated solvents because several substances have been reported to have produced an explosion when mixed with dimethyl sulfoxide. Do not dispose of DMSO in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Ethanol

Ethyl Alcohol

# General Properties

Colorless liquid which may be obtained as the water azeotrop containing about 5 percent water or as absolute alcohol containing 0.1 percent or less water.

# Physical Properties

Molecular weight: 46.07
Freezing point (°C): -114.1
Boiling point (°C): 78.3
Flash point (°C): 12
Min. ignition temp. (°C): 425

Density (g/cm $^3$ ): 0.7936 $^{15}$  0.7894 $^{20}$  0.785 $^{25}$ 

Refractive Index: 1.3614<sup>20</sup> 1.3594<sup>25</sup> Viscosity (cPoise): 1.078<sup>25</sup> 0.991<sup>35</sup>

Dielectric constant: 24.55 Ionization potential (eV): 10.49

Solubility: water, organic solvents

Optical properties:

Wavelength (nm) 200 210 220 230 240 250 270 Transmission (%) 5 35 55 72 85 90 98

# **Application**

Ethanol is the most frequently used solvent for laser dyes. As it is highly polar, its application is restricted to polar dyes such as the Rhodamines.

# Safety

RTECS # KQ6300000; Sax 6, 1316 • TLV: 1000 ppm or 1900 mg/m³ Hazards Highly flammable. It is practically impossible to produce any toxic effects by inhalation of pure ethanol vapors under usual lab conditions. The minimum identifiable odor is about 530 ppm. Concentrations of 6,000-9,000 ppm have an intense odor that may be practically intolerable at first, but one becomes acclimated soon. Concentrations of about 1,000 ppm cause slight irritation of mucous membranes and other symptoms.

### Safety Precautions

Keep container tightly closed. Keep away from sources of ignition. Do not smoke. Limit the quantity stored to foreseeable short-term requirements; large quantities should not be allowed to accumulate in the laboratory. If spillage of solvent or accidental release occurs, ventilate the whole laboratory as soon as possible.

### Waste Disposal Procedures

Do not dispose of Ethanol in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Ethylene Glycol

1,2-Ethanediol

# **General Properties**

Ethylene glycol is a colorless, odorless liquid with a bittersweet taste. It is very hygroscopic.

# Physical Properties

Molecular weight: 62.07
Freezing point (°C): -13
Boiling point (°C): 197.3
Flash point (°C): 110
Min. ignition temp. (°C): 410

Density (g/cm $^3$ ): 1.1135 $^{20}$  1.11 $^{25}$  1.1066 $^{30}$ 

Refractive Index: 1.4318<sup>20</sup> 1.4306<sup>25</sup> Viscosity (cPoise): 26.09<sup>15</sup> 13.55<sup>30</sup>

Dielectric constant: 37.7 Ionization potential (eV): 10.49

Solubility: water, alcohols

Optical properties:

Wavelength (nm) 210 220 230 240 250 260 280 Transmission (%) 20 35 40 60 75 90 98

# Application

Ethylene glycol, due to its high viscosity, is a frequently used solvent in jet stream dye lasers. Its dissolving capacity makes it suitable for polar dyes like Coumarins, Rhodamines, and Cyanines.

# Safety

# RTECS # KW2975000; Sax 6, 1343 · TLV: 100 ppm or 274 mg/m³ Hazards

Harmful if swallowed. Ethylene glycol has a low vapor pressure at normal temperature. It presents negligible hazards to health except, possibly, when being used at elevated temperature. It has a low, acute oral toxicity. There does not appear to be any significant irritation from skin contact.

### Safety Precautions

Avoid skin contact. Store in closed container away from heat, sparks and open flame.

#### Waste Disposal Procedures

Place Ethylene Glycol in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Hexane

n-Hexane

### **General Properties**

Hexane is a colorless, water-insoluble, and highly flammable liquid.

### Physical Properties

Molecular weight: 86.18
Freezing point (°C): -95.4
Boiling point (°C): 68.74
Flash point (°C): -26
Min. ignition temp. (°C): 240

Density (g/cm³): 0.659420 0.654825 Refractive Index: 0.374920 1.372325 Viscosity (cPoise): 0.312620 0.298525

Dielectric constant: 1.8799 Ionization potential (eV): 10.18

Solubility: 0.00095 % in water, nonpolar organic solvents

Optical properties:

Wavelength (nm) 220 250 190 200 210 230 240 Transmission (%) 10 30 60 80 94 98 98

### **Application**

Suitable solvent for nonpolar laser dyes. Its high optical transparency in the UV range allows the application in dye lasers pumped below 300 nm.

### Safety

# RTECS # MN9275000; Sax <u>6</u>, 1523 · TLV: 500 ppm or 1800 mg/m<sup>3</sup> Hazards

Highly flammable. Harmful by inhalation and in contact with the skin. Possible risk of irreversible effects. Hexane is a fat solvent and thus irritates the skin.

#### **Safety Precautions**

Keep container in a well-ventilated place. Keep away from source of ignition. Do not inhale gas/fumes/vapor/spray. Avoid skin contact. Do not smoke, eat, or drink when handling the solvent. Keep all containers tightly closed away from sparks and open flames.

### Waste Disposal Procedures

Do not dispose of Hexane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Methanol

Methyl Alcohol

### General Properties

Methanol is a colorless hygroscopic liquid usually containing 0.01 - 0.04 percent water. It is highly inflammable and toxic.

### Physical Properties

Molecular weight: 32.04
Freezing point (°C): -97.7
Boiling point (°C): 64.7
Flash point (°C): 11
Min. ignition temp. (°C): 455

Density  $(g/cm^3)$ :  $0.7961^{15} \ 0.7913^{20} \ 0.7866^{25}$ 

Refractive Index: 1.3284<sup>20</sup> 1.3265<sup>25</sup> Viscosity (cPoise): 0.5506<sup>25</sup> 0.5445<sup>35</sup>

Dielectric constant: 32.7 Ionization potential (eV): 10.84

Solubility: water, organic solvents

Optical properties:

200 260 Wavelength (nm) 210 220 230 240 250 Transmission (%) 2 20 50 75 85 95 98

### **Application**

Methanol is a polar, protic solvent frequently used to dissolve laser dyes like Coumarins, Rhodamines, and Cyanines. Its excellent optical transparency makes it the ideal solvent for UV-pumped dye lasers.

### Safety

# RTECS # PC1400000; Sax <u>6</u>, 1764 · TLV: 200 ppm or 260 mg/m<sup>3</sup> Hazards

Highly flammable. Toxic by inhalation and if swallowed. Methanol does not have suitable warning or irritating properties except at high concentrations. Ingestion of methanol can cause blindness and death.

### **Safety Precautions**

Keep container tightly closed. Keep away from sources of ignition. Do not smoke. Avoid contact with skin. Methanol vapor/air mixtures may produce explosible mixtures. Keep containers tightly closed. If spillage of solvent or accidental release occurs, ventilate the whole laboratory as soon as possible.

### **Waste Disposal Procedures**

Do not dispose of Methanol in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Toluene

Methylbenzene

### General Properties

Colorless, water-insoluble, and highly flammable liquid with aromatic odor.

### Physical Properties

Molecular weight: 92.14
Freezing point (°C): -95
Boiling point (°C): 110.6
Flash point (°C): 7
Min. ignition temp. (°C): 535

Density (g/cm $^3$ ): 0.867 $^{25}$  0.8623 $^{25}$  0.8577 $^{30}$ 

Refractive Index: 1.4969<sup>20</sup> 1.4941<sup>25</sup> Viscosity (cPoise): 0.5866<sup>20</sup> 0.5516<sup>25</sup>

Dielectric constant: 2.379 Ionization potential (eV): 8.82

Solubility: 0.0515 % in water, nonpolar organic solvents

Optical properties:

Wavelength (nm) 285 290 300 310 320 340 350 50 80 89 93 95 98 Transmission (%) 10

### **Application**

Toluene is a suitable solvent for nonpolar laser dyes like p-Terphenyl. Its low optical transparency in the UV range restricts the application on dye lasers pumped above 320 nm.

### Safety

# RTECS # XS5250000; Sax <u>6</u>, 2588 · TLV:200 ppm or 7500 mg/m<sup>3</sup> Hazards

Highly flammable. Harmful by inhalation. Exposure of humans to Toluene vapor produces mild fatigue, weakness, confusion, and paresthesia of the skin. Toluene is most dangerous by inhalation. It is irritating to the skin, and contact should be avoided when possible.

### **Safety Precautions**

Keep away from source of ignition. Do not smoke. Take precautionary measures against static discharges.

### Waste Disposal Procedures

Do not dispose of Toluene in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

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# Laser Dyes

# Properties, Application, and Absorption Spectra

The output power of dye lasers depends on the quality of the dye used. To overcome reduced quantum efficiency and instability due to impurities, Lambdachrome® laser dyes are synthesized and examined by experienced chemists for their chemical and spectral properties. Finally, they are purified by specially developed techniques.

The composition of all dyes is guaranteed with spectrophotometric and chromatographic analysis. Wavelength ranges are given in the following tables to provide assistance in choosing the correct dye for a given application. The ranges were measured by Lambda Physik, taken from Coherent CW Dye Laser Fact Sheets, or taken from the pertinent literature. The exact spectral range depends on the solvent and the concentration as well as on the method of pumping.

Lambda Physik reserves the right to modify any information given herein. Every effort is made to ensure utmost accuracy; however, no liability is assumed for errors occurring in the tables.

#### Abbreviations Used

MW Molecular Weight

CAS Chemical Abstracts Service

Effic. Efficiency, defined as the ratio of optical input to output Conc. Concentration of dye, given in grams per liter solvent

Ref. Reference
Cyclohex. Cyclohexane
DCE 1,2-Dichloroethane
DMF Dimethylformamide

EtOH Ethanol

MeOH Methanol

Eg Ethyleneglycol

Bz Benzylalcohol

DMSO Dimethylsulfoxide

Pc Propylenecarbonate

Tol Toluene

#### BEER's Law

Within the absorption spectra, the intensity is expressed as a molar decadic extinction coefficient,  $\epsilon$ . The amount of light absorbed depends on the extinction coefficient and the number of molecules in the light path. The latter amount depends on the concentration of the dye in solution and the path length of the absorption cell. The amount of light that passes through a solution (transmittance) is given by BEER's law:

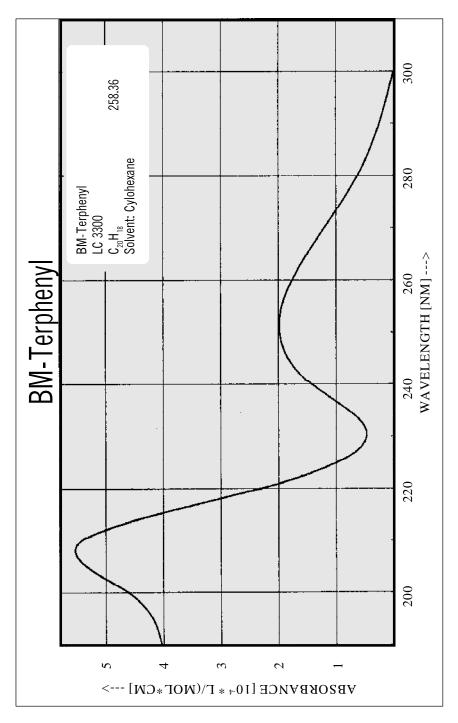
$$\log I_{o}/I = \varepsilon \cdot c \cdot d$$

where  $\mathbf{I}_{o}$  is the intensity of the light before it encounters the cell,  $\mathbf{I}$  is the intensity of the light emerging from the cell,  $\mathbf{c}$  is the concentration in moles per liter, and  $\mathbf{d}$  is the path length in centimeters.

The absorption cross-section  $\sigma$  can be determined from the extinction coefficient  $\epsilon$  by:

$$\sigma = 0.385 \cdot 10^{-20} \, \varepsilon.$$

Here  $\sigma$  is given in cm<sup>2</sup> ( $\varepsilon$  measured in liter/(mole  $\cdot$  cm)).



# BM-Terphenyl (LC 3300)

### Constitution

2,2''-Dimethyl-p-terphenyl

C<sub>20</sub>H<sub>18</sub> · MW: 258.36

#### Characteristics

Lambdachrome® number: 3300

CAS registry number:

Appearance:

white, crystalline solid

Absorption maximum (in cyclohexane): 251 nm

Molar absorptivity: 1.98 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in cyclohexane): 335 nm

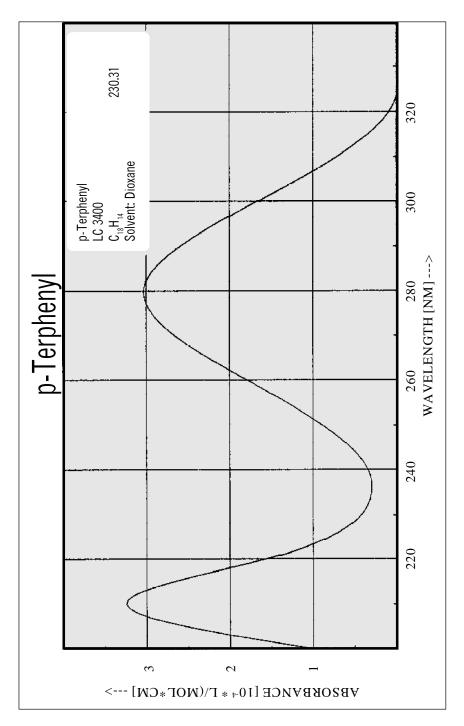
For research and development purposes only.

### Lasing Performance

Shortest tunable laser dye for pulsed operation; tunable around 336 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]		Conc. [g/l]	Solvent	Ref.
KrF-Excimer	248	334	312 - 343	4	0.50	Cyclohex	1, 2, 3

- 1. Lambda Physik, Wall Chart 1996.
- 2. W. Zapka, U. Brackmann *Appl. Phys.* 20, 283 (1979).
- 3. F.-G. Zhang, F. P. Schäfer *Appl. Phys.* <u>B26</u>, 211 (1981).

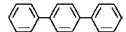


# p-Terphenyl (LC 3400)

### Constitution

PTF

C<sub>10</sub>H<sub>14</sub> · MW: 230.31



#### Characteristics

Lambdachrome® number: 3400 CAS registry number: 92-94-4

Appearance: white, crystalline solid

Absorption maximum (in cyclohexane): 275 nm

Molar absorptivity:  $3.21 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in cyclohexane): 339 nm

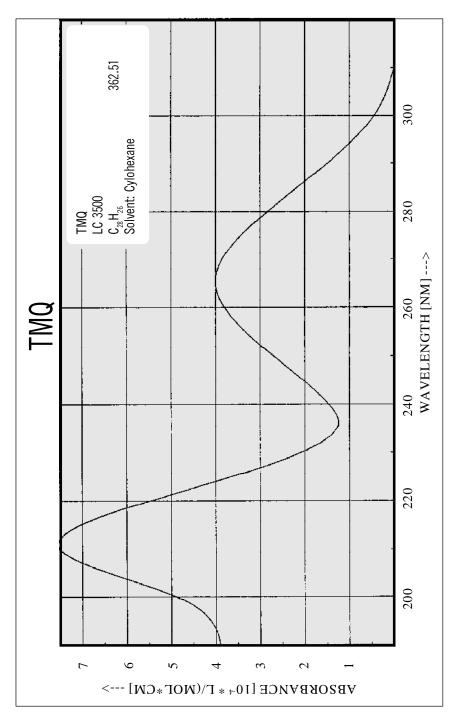
For research and development purposes only.

### Lasing Performance

Efficient laser dye for pulsed operation; tunable around 340 nm.

Pump	)	D	Dye Laser Characteristics				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
KrF-Excimer XeCl-Excimer	248 308	339 343	322 - 365 332 - 360	14 8	0.02 0.24	Cyclohex Dioxane	1 2, 3
Nd:YAG, 4th	266	340	-	5	0.23	Cyclohex.	4, 5

- 1. H. Bücher, W. Chow, *Appl. Phys.* 13, 267 (1977).
- 2. Lambda Physik, Wall Chart 1996.
- 3. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 402 (1981).
- 4. G. A. Abakumov et al., *Opto-Electron.* <u>1</u>, 205 (1969).
- 5. D. Huppert, P. M. Rentzepis, *J. Appl. Phys.* <u>49(2)</u>, 543 (1978).



# TMQ (LC 3500)

### Constitution

3,3',2"',3""-Tetramethyl-p-quaterphenyl

 $C_{28}H_{26} \cdot MW: 362,51$ 

#### Characteristics

Lambdachrome® number: 3500 CAS registry number: -

Appearance: white, crystalline solid

Absorption maximum (in cyclohexane): 266 nm

Molar absorptivity: 3.99 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum: - For research and development purposes only.

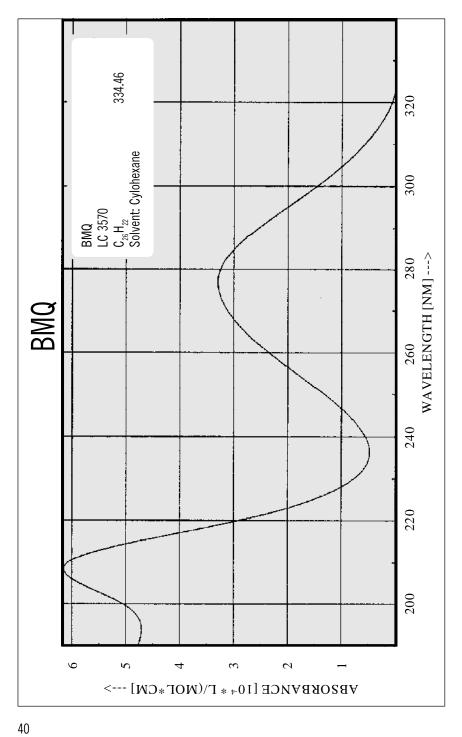
# Lasing Performance

Laser dye for pulsed operation; tunable around 350 nm.

Pump	)	Dye Laser Characteristics						
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.	
	[nm]	[nm]	[nm]	[%]	[g/l]			_
Nd:YAG, 4th	266	350	338 - 361	1.2	0.72	Cyclohex	1	

### References

1. L. D. Ziegler, B. S. Hudson, *Opt. Commun.* 32(1), 119 (1980).



# BMQ (LC 3570)

### Constitution

2,2'''-Dimethyl-p-quaterphenyl

C<sub>26</sub>H<sub>22</sub> - MW: 334.46

### Characteristics

Lambdachrome® number: 3570 CAS registry number: -

Appearance: white, crystalline solid

Absorption maximum (in cyclohexane): 275 nm

Molar absorptivity: 3.35 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum:
For research and development purposes only.

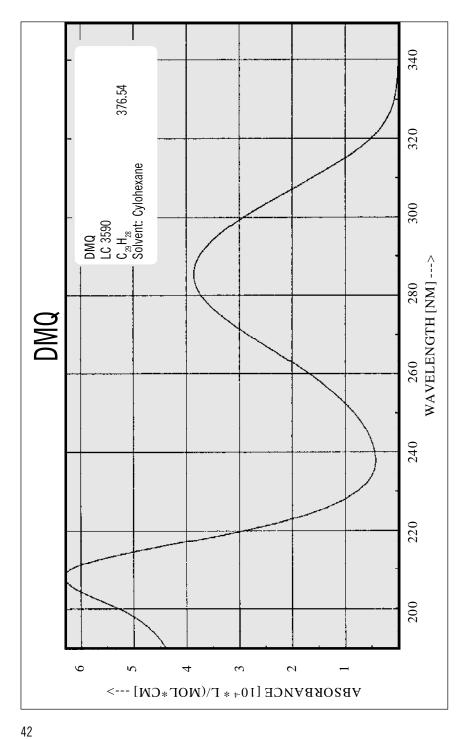
### Lasing Performance

Efficient laser dye for pulsed operation; tunable around 357 nm.

Pump	)	Dye Laser Characteristics							
Source	Wavelength [nm]	Peak [nm]	Range [nm]		Conc. [g/l]	Sol	vent	Ref.	
XeCI-Excimer	308	357	335 - 3	375	9 <i>0.</i>	60	Dioxa	ne 1	

### References

1. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).



# DMQ (LC 3590)

### Constitution

2-Methyl-5-t-butyl-p-quaterphenyl

C<sub>29</sub>H<sub>28</sub> · MW: 376.54

### Characteristics

Lambdachrome® number: 3590 CAS registry number: -

Appearance: white, crystalline solid

Absorption maximum (in cyclohexane): 285 nm

Molar absorptivity: 3.86 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

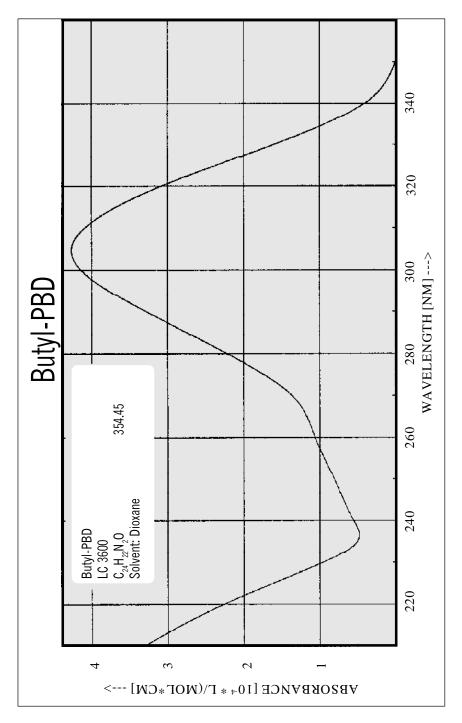
Fluorescence maximum: - For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed operation; tunable around 360 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	360	346 - 377	9	0.23	Dioxane	1, 2

- 1. Lambda Physik, Wall Chart 1996
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).



# Butyl-PBD (LC 3600)

### Constitution

2-(4-Biphenylyl)-5-(4-t-butylphenyl)-1,3,4-oxadiazol BPBD-365

 $C_{24}H_{22}N_2O \cdot MW: 354.45$ 

#### Characteristics

Lambdachrome® number: 3600

CAS registry number: 15082-28-7

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 302 nm

Molar absorptivity:  $4.35 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in cyclohexane): 368 nm

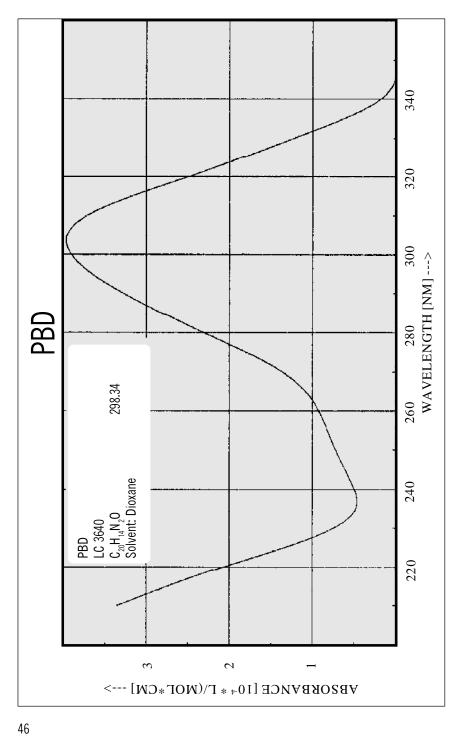
For research and development purposes only.

### Lasing Performance

Efficient laser dye for pulsed operation; tunable around 360 nm.

Pump	)	D	Dye Laser Characteristics				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nitrogen Nd:YAG, 4th	308 337 266	363 362 362	356 - 385 356 - 390 354-388	5 rel. 4.7	0.30 1.60 1.75	Dioxane Dioxane Cyclohex.	1, 2, 3 3, 4 5

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 402 (1981).
- 3. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. L. D. Ziegler, B. S. Hudson, *Opti. Commun.* <u>32(1)</u>, 119 (1980).



# PBD (LC 3640)

### Constitution

2-(4-Biphenylyl)-5-phenyl-1,3,4-oxadiazol

C<sub>20</sub>H<sub>14</sub>N<sub>2</sub>O · MW: 298.34

#### Characteristics

Lambdachrome® number: 3640

CAS registry number:

Appearance: white, crystalline solid Absorption maximum (in dioxane): 302 nm

Molar absorptivity: 3.90 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in toluene): 360 nm

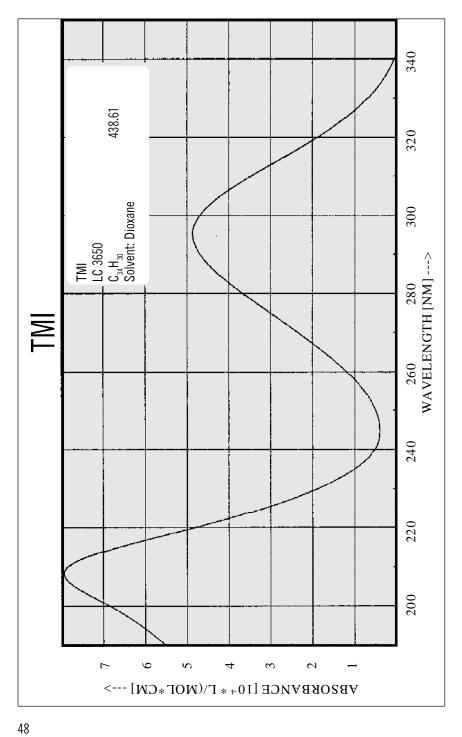
For research and development purposes only.

### Lasing Performance

Efficient laser dye for pulsed operation; tunable around 360 nm.

Pump	)	D	Dye Laser Characteristics				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Flashlamp	308 337 -	363 362 362	355 - 390 357 - 390 -	4 rel. -	0.11 1.20 0.08	Cyclohex. Dioxane Ethanol	1 2 3, 4

- 1. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 402 (1981).
- 2. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 3. H. W. Furumoto, H. L. Ceccon, IEEE J. Quant. Electron. QE-6, 262 (1970).
- 4. T. Morrow, H. T. W. Price, *Opt. Commun.* <u>10(2)</u>, 133 (1974).



# TMI (LC 3650)

### Constitution

2,5,2"",5""-Tetramethyl-p-quinquephenyl

C<sub>34</sub>H<sub>30</sub> · MW: 438.61

#### Characteristics

Lambdachrome® number: 3650

CAS registry number: Appearance: white, crystalline solid

Absorption maximum (in dioxane): 295 nm

Molar absorptivity: 4.86 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

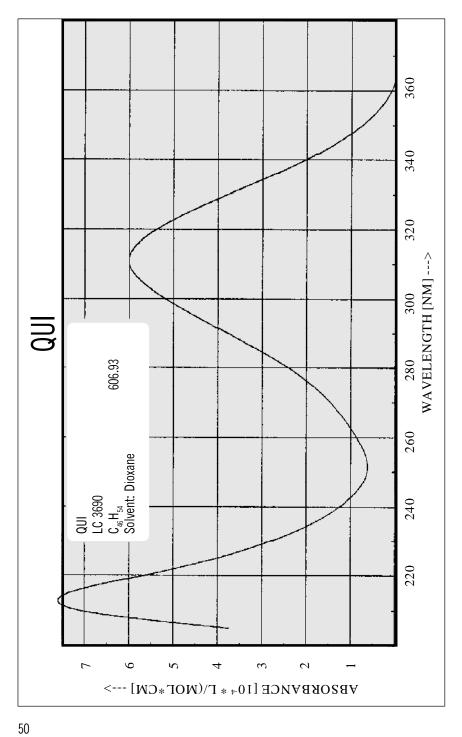
Fluorescence maximum:
For research and development purposes only.

### Lasing Performance

Efficient laser dye for pulsed operation; tunable around 370 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range			Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	372	355 - 395	11	0.18	Dioxane	1, 2

- 1. Lambda Physik, Data Sheet.
- 2. V. S. Antonov, K. L. Hohla, Appl. Phys. <u>B32</u>, 9 (1983).

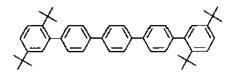


# QUI (LC 3690)

# Constitution

3,5,3"",5""-Tetra-t-butyl-p-quinquephenyl

C<sub>46</sub>H<sub>54</sub> · MW: 606.93



#### Characteristics

Lambdachrome® number: 3690 CAS registry number: -

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 310 nm

Molar absorptivity: 6.00 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

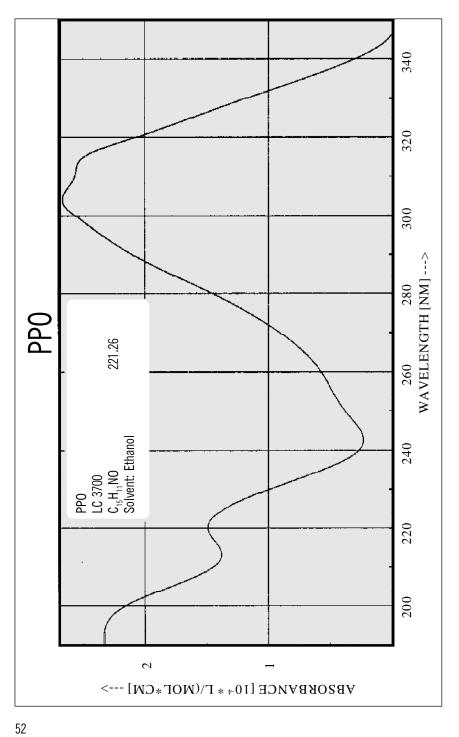
Fluorescence maximum:
For research and development purposes only.

### Lasing Performance

Efficient laser dye for pulsed operation; tunable around 390 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nitrogen	308 337	390 387	368 - 402 372-412	11 rel.	0.20 0.52	Dioxane Dioxane	1, 2 3

- 1. Lambda Physik, Wall Chart 1996.
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).
- 3. Lambda Physik, Data Sheet.

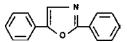


# PPO (LC 3700)

### Constitution

2,5-Diphenyloxazol

C<sub>15</sub>H<sub>11</sub>NO · MW: 221.26



### Characteristics

Lambdachrome® number: 3700 CAS registry number: 92-71-7

Appearance: white, crystalline solid

Absorption maximum (in ethanol): 303 nm

Molar absorptivity:  $2.80 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 365 nm

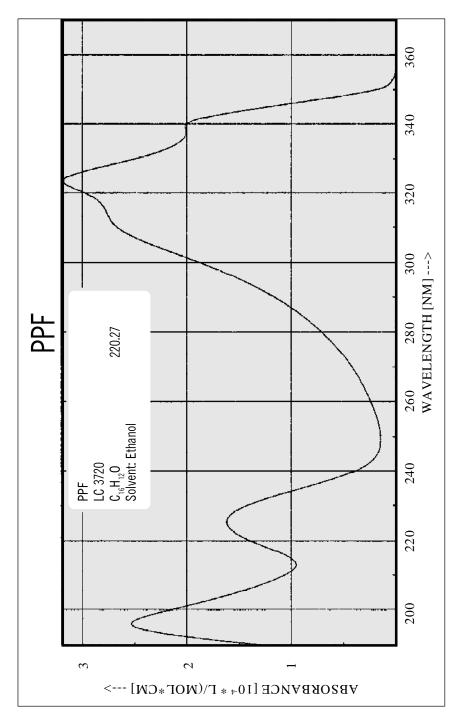
For research and development purposes only.

# Lasing Performance

Laser dye for pulsed operation; tunable around 380 nm.

Pump	)	D	ye Laser Ch				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Nd:YAG, 4th Flashlamp	308 337 266	377 375 375 381	- - 368-382 -	2.3 rel. 1.6	0.22 0.39 1.10 1.54	Methanol Dioxane Cylohex. Dioxane	1, 2 1 3 4

- 1. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 2. O. Uchino et al., *Appl. Phys.* 19, 35 (1979).
- 3. L. D. Ziegler, B. S. Hudson, *Opt. Commun.* 32(1), 119 (1980).
- 4. H. W. Furumoto, H. L. Ceccon, IEEE J. Quant. Electron. QE-6, 262 (1970).



# PPF (LC 3720)

### Constitution

2,5-Diphenylfuran

C<sub>16</sub>H<sub>12</sub>O · MW: 220.27

### Characteristics

Lambdachrome® number: 3720
CAS registry number: -

Appearance: white, crystalline solid

Absorption maximum (in ethanol): 324 nm

Molar absorptivity: 3.78 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 368 nm

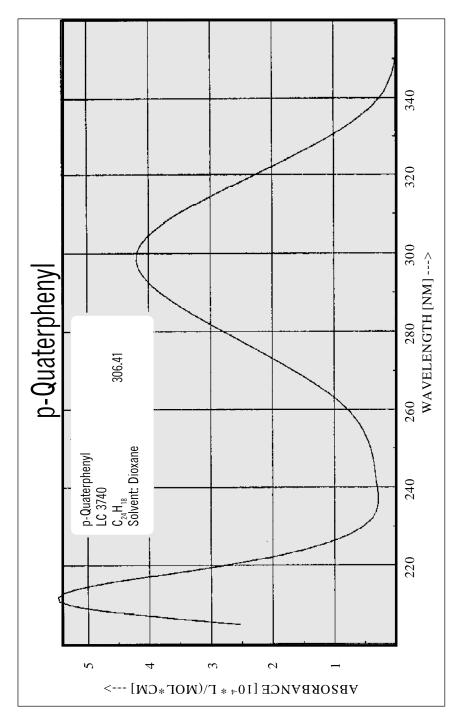
For research and development purposes only.

### Lasing Performance

Laser dye for pulsed operation; tunable around 370 nm.

Pun	пр	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Nitrogen	337	373	369 - 379	11	0.44	Dioxane	1,2

- 1. H. P. Broida, S. C. Haydon, *Appl. Phys. Letters* <u>16(3</u>), 142 (1970).
- 2. M. Maeda, Y. Miyazoe, Jpn. J. Appl. Phys. 13(5), 827 (1974).

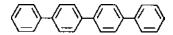


# p-Quaterphenyl (LC 3740)

### Constitution

POP

C<sub>24</sub>H<sub>18</sub> · MW: 306.41



#### Characteristics

Lambdachrome® number: 3740
CAS registry number: 135-70-6

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 297 nm

Molar absorptivity: 4.28 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in toluene): 374 nm

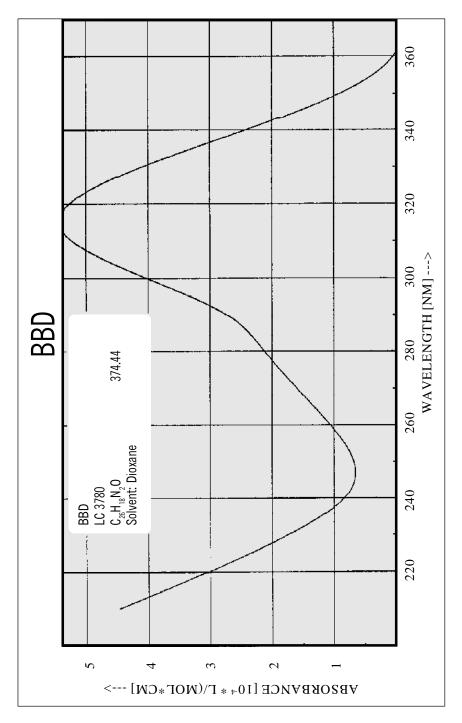
For research and development purposes only.

### Lasing Performance

Laser dye for pulsed operation; tunable around 370 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Flashlamp	308 337 -	371 370 372	- 362 - 390 -	20 11 -	0.09 sat. 0.08	Dioxane DMF DMF	1 2, 3 4

- 1. P. Cassard et al., Opt. Commun. 38(2), 131 (1981).
- 2. M. Maeda, Y. Miyazoe, Jpn. J. Appl. Phys. 13(5), 827 (1974).
- 3. J. A. Myer, I. Itzkan, E. Kierstead, *Nature* <u>225</u>, 544 (1970).
- 4. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* <u>11(5</u>), 692 (1972).

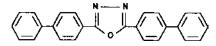


# BBD (LC 3780)

### Constitution

2,5-Bis-(4-biphenylyl)-1,3,4-oxadiazol

 $C_{26}H_{18}N_2O \cdot MW: 374.44$ 



### Characteristics

Lambdachrome® number: 3780
CAS registry number: -

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 314 nm

Molar absorptivity: 5.30 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in dioxane): 373 nm

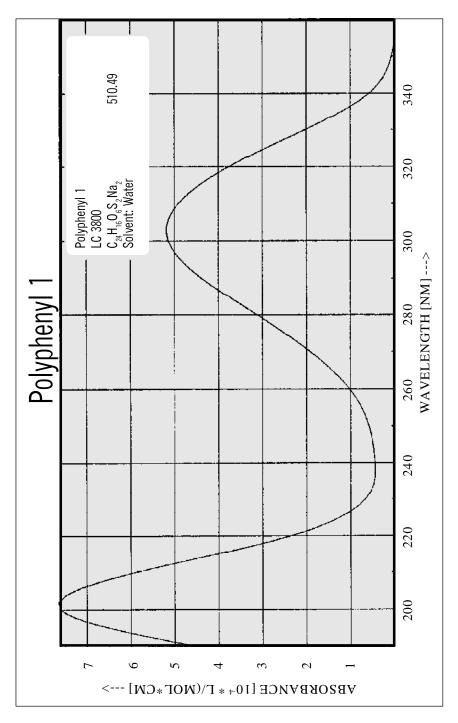
For research and development purposes only.

# Lasing Performance

Laser dye for pulsed operation; tunable around 380 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer Nitrogen Flashlamp	308 337 -	378 375 377	368 - 399 372 - 405 -	14 19 -	0.37 0.74 0.74	Dioxane Dioxane Dioxane	1, 2 2, 3 4

- 1. P. Cassard et al., Opt. Commun. 38(2), 131 (1981).
- 2. O. Uchino et al., *Appl. Phys.* <u>19</u>, 35 (1979).
- 3. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* <u>13(5</u>), 827 (1974).
- 4. M. Maeda et al., *IEEE J. Quant. Elec tron.* <u>QE-13</u>, 65 (1977).



# Polyphenyl 1 (LC 3800)

### Constitution

p-Quaterphenyl-4,4'''-disulfonicacid Disodiumsalt

 $C_{24}H_{16}O_6S_2Na_2 \cdot MW: 510.49$ 

### Characteristics

Lambdachrome® number: 3800 CAS registry number: -

Appearance: white, crystalline solid

Absorption maximum (in water): 308 nm

Molar absorptivity: 5.19 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

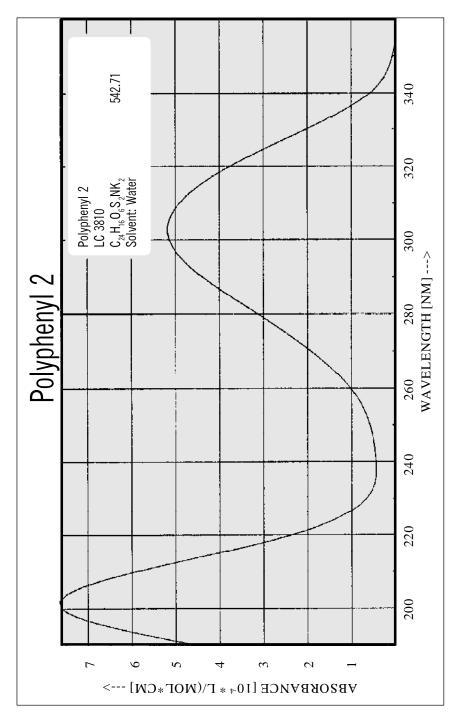
Fluorescence maximum:
For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 380 nm.

Pump		D	ye Laser Ch				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nitrogen CW, Ar <sup>+</sup>	308 337 UV	381 380 382	363 - 408 362 - 411 362 - 412	12 rel. -	0.20 0.36 1.25	Eg Eg Eg	1, 2, 3 3 3

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 402 (1981).
- 3. W. Hüffer et al., *Opt. Commun.* 33(1), 85 (1980).



# Polyphenyl 2 (LC 3810)

# Constitution

p-Quaterphenyl-4,4'''-disulfonicacid Dipotassiumsalt

 $C_{24}H_{16}O_6S_2NK_2 \cdot MW: 542.71$ 

#### Characteristics

Lambdachrome® number: 3810 CAS registry number: -

Appearance: white, crystalline solid

Absorption maximum (in water): 308 nm

Molar absorptivity: 5.19 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum: - For research and development purposes only.

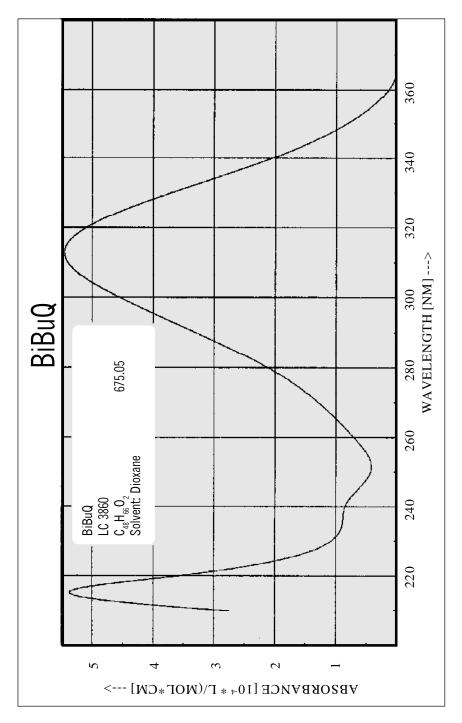
# Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 390 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer CW, Ar+	308 UV	382 384	363 - 400 370 - 406	10 -	0.25 2.0	Eg Eg	1 1

# References

1. Lambda Physik, Wall Chart 1996.



# BiBuQ (LC 3860)

#### Constitution

4,4"'-Bis-(2-butyloctyloxy)-p-quaterphenyl

BBQ • Pilot 386

C<sub>48</sub>H<sub>66</sub>O<sub>2</sub> · MW: 675.05

#### Characteristics

Lambdachrome® number: 3860 CAS registry number: 18434-08-7

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 313 nm

Molar absorptivity: 5.45 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

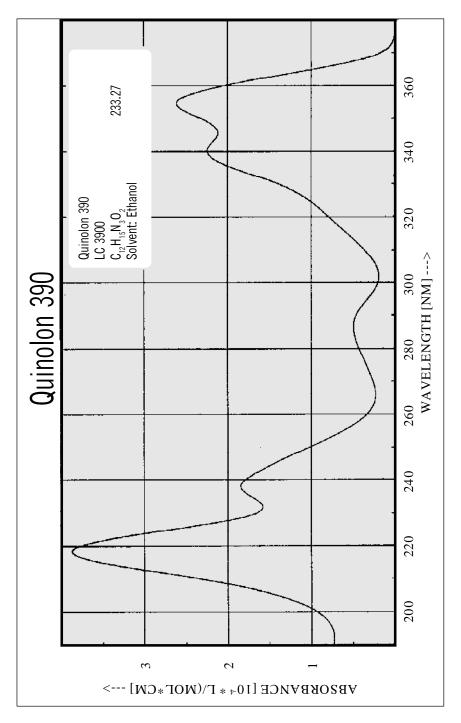
Fluorescence maximum:
For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed operation; tunable around 390 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range			Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	388	367 - 405	11	0.25	Dioxane	1, 2, 3
Nitrogen	<i>337</i>	383	364 - 405	rel.	0.41	Dioxane	3, 4
Nd:YAG, 3rd	<i>355</i>	<i>392</i>	380 - 410	-	1.34	EtOH/Tol	5
Flashlamp	-	-	389 - 395	-	-	DMF	6

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 402 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. K. Azuma et al., Jpn. J. Appl. Phys. 18(1), 209 (1979).
- 6. P. R. Hammond et al., *Appl. Phys.* <u>9</u>, 67 (1976).



# Quinolon 390 (LC 3900)

# Constitution

7-Dimethylamino-1-methyl-4-methoxy-8-azaquinolone-2

 $C_{12}H_{15}N_3O_2 \cdot MW: 233.27$ 

#### Characteristics

Lambdachrome® number: 3900

CAS registry number: 119883-58-8

white, crystalline solid Appearance:

Absorption maximum (in ethanol): 355 nm

Molar absorptivity: 2.52 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum:

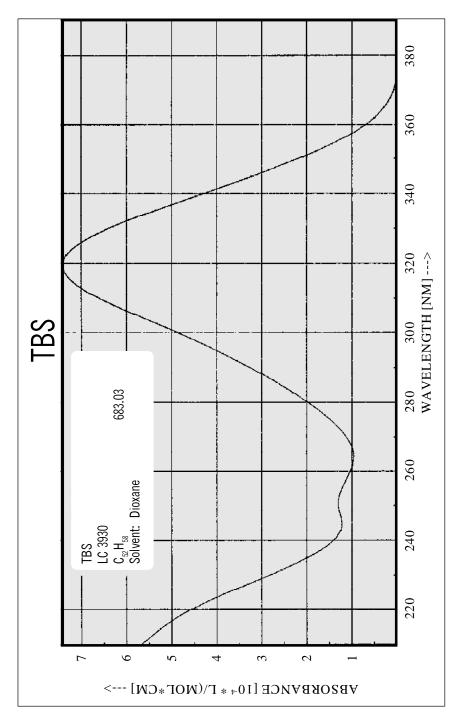
For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed operation; tunable around 390 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
Nd:YAG, 3rd Flashlamp	355 -	390 390	384 - 394 -	4	0.25 1.15	Methanol Ethanol	1 2

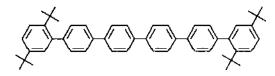
- 1. Lambda Physik, Wall Chart 1996.
- 2. P. R. Hammond et al., Appl. Phys. 8, 315 (1975).



# TBS (LC 3930)

 $\frac{Constitution}{3,5,3^{\prime\prime\prime\prime\prime},5^{\prime\prime\prime\prime\prime}\text{-Tetra-t-butyl-p-sexiphenyl}}$ 

C<sub>52</sub>H<sub>58</sub> · MW: 683.03



#### Characteristics

Lambdachrome® number: 3930

CAS registry number: white, crystalline solid Appearance:

Absorption maximum (in dioxane): 320 nm

7.49 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup> Molar absorptivity:

Fluorescence maximum: For research and development purposes only.

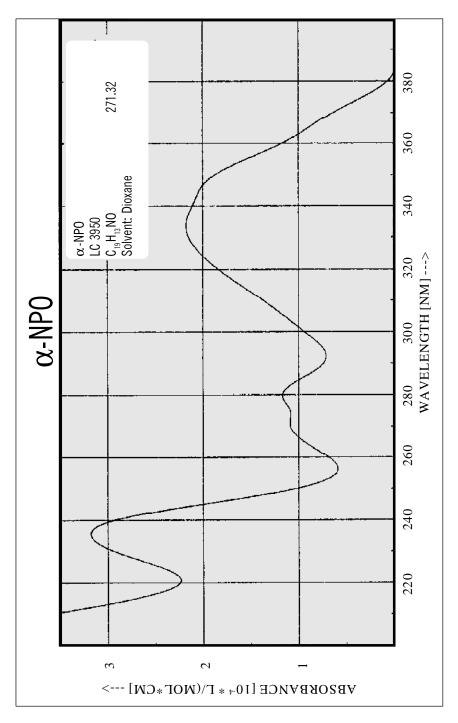
# Lasing Performance

Very stable and efficient laser dye for pulsed operation; tunable around 390 nm.

Pump	)	Dye Laser Characteristics						
Source	Wavelength [nm]	Peak [nm]	Range [nm]		Conc. [g/l]	Solvent	Ref.	
XeCI-Excimer	308	393	365 - 410	11	0.16	Cyclohex	1	

# References

1. Lambda Physik, Data Sheet.

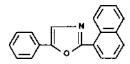


# $\alpha$ -NPO (LC 3950)

# Constitution

2-(1-Naphthyl)-5-phenyloxazol

C<sub>19</sub>H<sub>13</sub>NO · MW: 271.32



#### Characteristics

Lambdachrome® number: 3950

CAS registry number:

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 333 nm

Molar absorptivity:  $2.15 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in dioxane): 396 nm

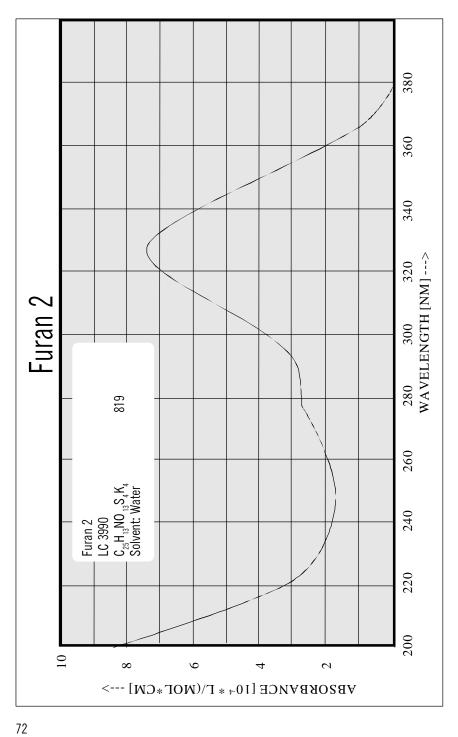
For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed operation; tunable around 400 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nitrogen Nd:YAG, 3rd Flashlamp	308 337 355 -	393 400 400 400	387 - 400 391 - 425 - -	6 rel. 6.5 -	0.32 0.67 1.00 0.07	Cyclohex. Dioxane Toluene Ethanol	1 2, 3 4 5

- 1. Lambda Physik, Data Sheet.
- 2. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 3. A. Dienes, *Appl. Phys.* <u>7</u>, 135 (1975).
- 4. G. A. Abakumov et al. *JETP Letters* <u>9</u>, 9 (1969).
- 5. H. W. Furumoto et al. *IEEE J. Quant. Electron.* QE-6(5), 262 (1970).



# Furan 2 (LC 3990)

#### Constitution

2-(4-Biphenylyl)-6-phenylbenzoxazotetrasulfonicacid Potassium Salt

 $C_{25}H_{13}NO_{13}S_{4}K_{4} \cdot MW: 819$ 

#### Characteristics

Lambdachrome® number: 3990 CAS registry number: -

Appearance: light yellow, crystalline solid

Absorption maximum (in water): 330 nm

Molar absorptivity: 7.79 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in dioxane): 396 nm

For research and development purposes only.

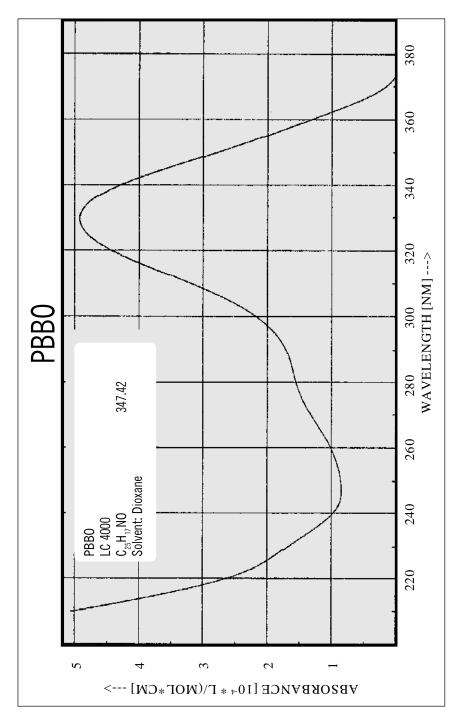
# Lasing Performance

Efficient laser dye for pulsed operation; tunable around 400 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nd:YAG, 3rd	308 355	399 402	388 - 426 392 - 422	8 15	0.50 0.5	Methanol Methanol	1 1

# References

1. Lambda Physik, Wall Chart 1996.



# PBBO (LC 4000)

# Constitution

2-(4-Biphenylyl)-6-phenylbenzoxazol-1,3

C<sub>25</sub>H<sub>17</sub>NO · MW: 347.42

#### Characteristics

Lambdachrome® number: 4000 CAS registry number: 17064-47-0

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 327 nm

Molar absorptivity:  $4.89 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 403 nm

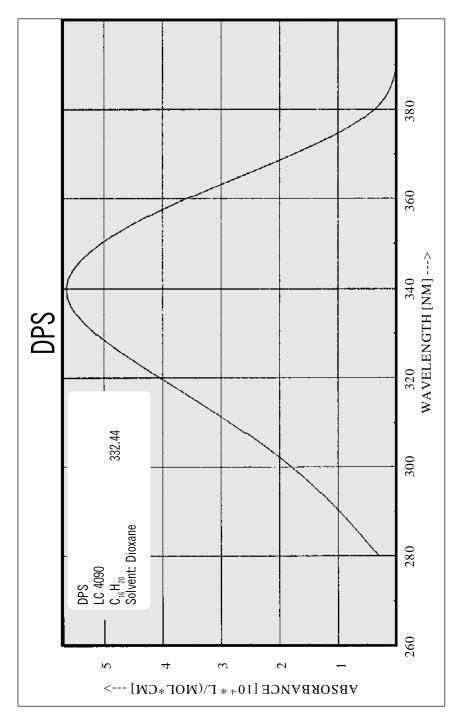
For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed operation; tunable around 400 nm.

Pump	)	D	Dye Laser Characteristics				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nitrogen	308 337	396 395	386 - 420 385 - 420	7 rel.	0.40 0.15	Dioxane Dioxane	1, 2 2, 3

- 1. Lambda Physik, Wall Chart 1996.
- 2. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 3. Lambda Physik, Data Sheet.

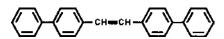


# DPS (LC 4090)

#### Constitution

4,4'-Diphenylstilbene Pilot 409

 $C_{16}H_{20}$  · MW: 332.44



#### Characteristics

Lambdachrome® number: 4090 CAS registry number: 26569-48-2

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 340 nm

Molar absorptivity: 5.65 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

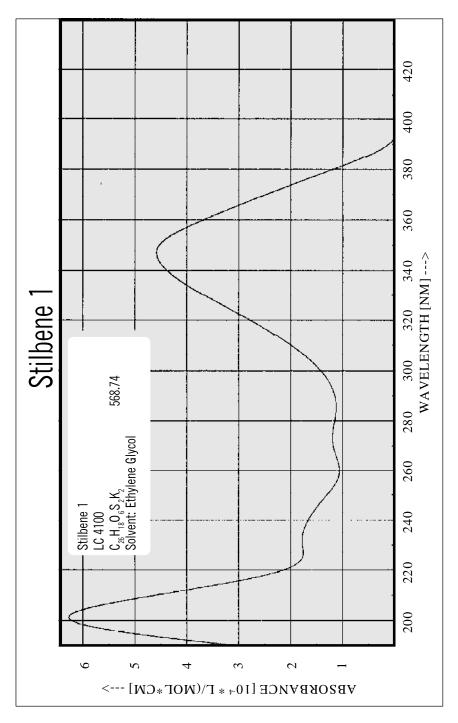
Fluorescence maximum: For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed operation; tunable around 400 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	406	399 - 415	11	0.25	Dioxane	1, 2
Nitrogen	<i>337</i>	404	394 - 416	rel.	0.12	Dioxane	3, 4
Nd:YAG, 3rd	<i>355</i>	408	-	-	-	Tol.	5
Flashlamp	-	409	406 - 411	-	sat.	DMF	<i>6, 7</i>

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 402 (1981).
- 3. Lambda Physik, Data Sheet.
- 4. A. Dienes, *Appl. Phys.* <u>7</u>, 135 (1975).
- 5. V. D. Kotzubanov et al., *Opt. Spectrosc.* <u>25</u>, 406 (1968).
- 6. P. R. Hammond et al., *Appl. Phys.* 9, 67 (1976).
- 7. H. W. Furumoto et al., *IEEE J. Quant. Electron.* QE-6, 262 (1970).



# Stilbene 1 (LC 4100)

### Constitution

[1,1'-Biphenyl]-4-sulfonic acid, 4',4''-1,2-ethene-diylbis-, dipotassium salt

 $C_{26}H_{18}O_6S_2K_2 \cdot MW: 568.74$ 

#### Characteristics

Lambdachrome® number: 4100 CAS registry number: 74758-59-1

Appearance: slightly yellow, crystalline solid

Absorption maximum (in Eg.): 350 nm

Molar absorptivity:  $4.55 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

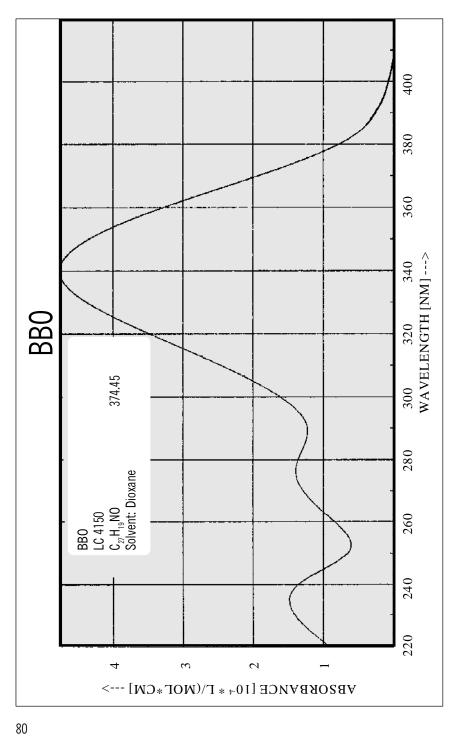
Fluorescence maximum: For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 415 nm.

Pum	p	D	Dye Laser Characteristics				
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
						_	
XeCI-Excime	r 308	416	405 - 428	6	0.50	Eg.	1, 2, 3
Nitrogen	<i>337</i>	417	405 - 446	rel.	0.20	Eg.	4
CW, Ar+	UV	415	403 - 428	-	0.75	Eg.	1, 4, 5, 6

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 402 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 4. Lambda Physik, *Data Sheet*.
- 5. T. F. Johnston, R. H. Brady, W. Proffitt, *Appl. Optics* <u>21(13)</u>, 2307 (1982).
- 6. W. Hüffer et al., Opt. Commun. 28(3), 353 (1979).

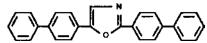


# BBO (LC 4150)

# Constitution

2,5-Bis-(4-biphenylyl)-oxazol

C<sub>27</sub>H<sub>10</sub>NO · MW: 374.45



#### Characteristics

Lambdachrome® number: 4150 CAS registry number: 2083-09-2

Appearance: white, crystalline solid

Absorption maximum (in dioxane): 340 nm

Molar absorptivity:  $4.76 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in toluene): 412 nm

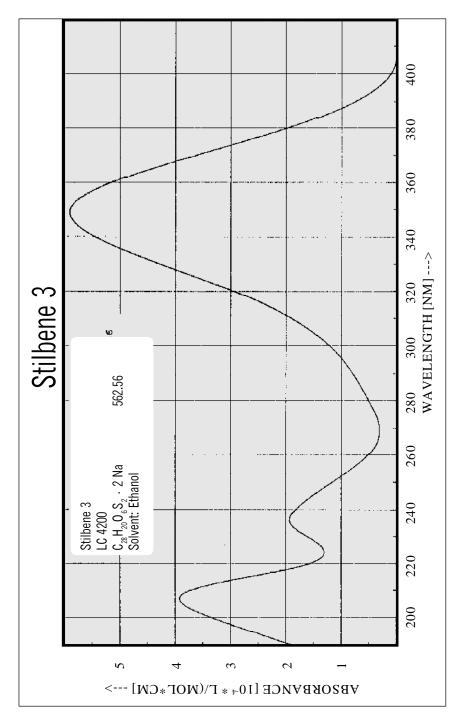
For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed operation; tunable around 410 nm.

Pum	D	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Nitrogen Nd:YAG, 3rd Flashlamp	337 355 -	408 409 410	401 - 419 - -	14 8 -	0.74 0.96 0.37	Dioxane Tol. Dioxane	1, 2 3 4, 5

- 1. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* <u>13(5)</u>, 827 (1974).
- 2. H. P. Broida, S. C. Haydon, *Appl. Phys. Letters* <u>16(3)</u>, 142 (1970).
- 3. G. A. Abakumov et al., *JETP Letters* 9, 9 (1969).
- 4. H. W. Furumoto et al., IEEE J. Quant. Electron. QE-6(5), 262 (1970).
- 5. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* <u>11(5)</u>, 692 (1972).



# Stilbene 3 (LC 4200)

#### Constitution

2,2'-([1,1'-Biphenyl]-4,4'-diyldi-2,1-ethenediyl)-bis-benzenesulfonic acid Disodium Salt Stilbene 420

 $C_{28}H_{20}O_6S_2 \cdot 2 \text{ Na} \cdot \text{MW: } 562.56$ 

#### Characteristics

Lambdachrome® number: 4200 CAS registry number: 27344-41-8

Appearance: yellow, crystalline solid

Absorption maximum (in Ethanol.): 350 nm

Molar absorptivity: 5.89 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum:

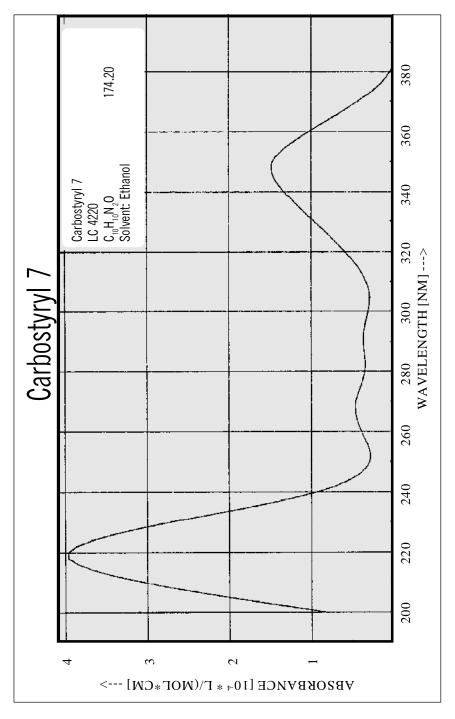
For research and development purposes only.

#### Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 425 nm.

Pum	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
				_			
XeCI-Excimer	308	425	412 - 443	9	0.65	Methanol	1, 2, 3
Nitrogen	<i>337</i>	424	408 - 457	rel.	0.22	Methanol	3, 4
Nd:YAG, 3rd	<i>355</i>	428	415 - 439	15	0.25	Methanol	1, 5
CW, Ar+	UV	435	410 - 485	-	1.0	Eg.	1, 4, 6, 7

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 402 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 4. Lambda Physik, *Data Sheet.*
- 5. D. M. Guthals, J. W. Nibler, Opt. Commun. 29(3), 322 (1979).
- 6. T. F. Johnston, R. H. Brady, W. Proffitt, *Appl. Optics* <u>21(13)</u>, 2307 (1982).
- 7. J. Kuhl et al., *Opt. Commun.* <u>24(3)</u>, 251 (1978).



# Carbostyryl 7 (LC 4220)

# Constitution

7-Amino-4-methylcarbostyryl Carbostyryl 124

C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O · MW: 174.20

#### Characteristics

Lambdachrome® number: 4220

CAS registry number: Appearance: white, crystalline solid

Absorption maximum (in ethanol): 350 nm

Molar absorptivity: 1.46 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 400 nm

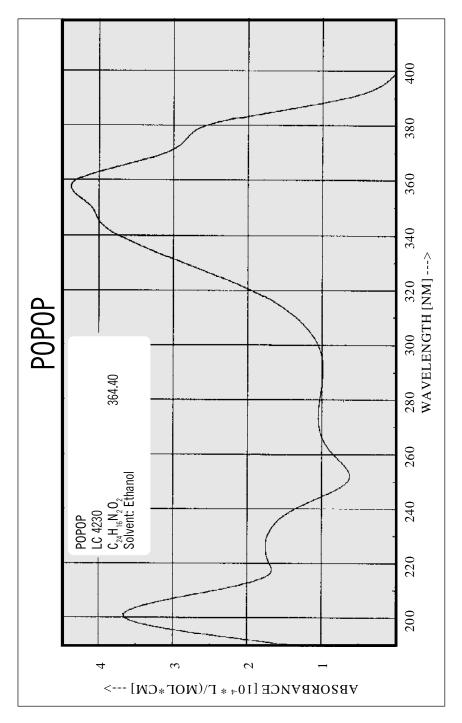
For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed operation; tunable around 410 nm.

Pum	р	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Flashlamp	-	413	408 - 420	-	-	Ethanol	1, 3

- 1. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).
- 2. R. Srinivasan, IEEE J. Quant. Electron. QE-5, 552 (1969).

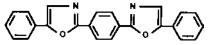


# POPOP (LC 4230)

#### Constitution

1,4-Di[2-(5-phenyloxazolyl)]benzene Pilot 423

 $C_{24}H_{16}N_2O_2 \cdot MW: 364.40$ 



#### Characteristics

Lambdachrome® number: 4230 CAS registry number: 1806-34-4

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 358 nm

Molar absorptivity:  $4.43 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 425 nm

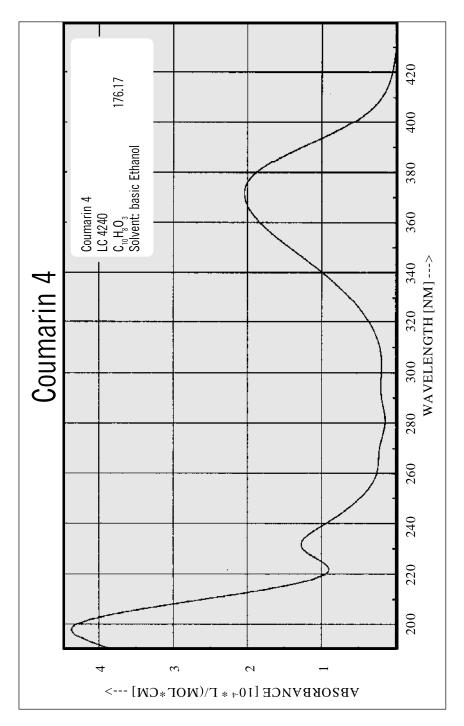
For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed operation; tunable around 420 nm.

Pum	р	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excime	r 308	421	411 - 446	6.5	0.42	Dioxane	1
Nitrogen	<i>337</i>	421	412 - 454	rel.	0.39	EtOH/Tol.	1,2
Nd:YAG, 3rd	<i>355</i>	417	-	-	-	Tol.	3
Flashlamp	-	419	-	-	0.12	Ethanol	4, 5

- 1. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 2. A. Dienes, *Appl. Phys.* <u>7</u>, 135 (1975).
- 3. V. D. Kotzubanov et al., Opt. Spectrosc. 25, 406 (1968).
- 4. H. W. Furumoto, H. L. Ceccon, J Appl Phys. 40, 4204 (1969).
- 5. P. R. Hammond et al., *Appl. Phys.* <u>9</u>, 67 (1976).



# Coumarin 4 (LC 4240)

#### Constitution

7-Hydroxy-4-methylcoumarin Umbelliferon 47

C<sub>10</sub>H<sub>o</sub>O<sub>2</sub> · MW: 176.17

#### Characteristics

Lambdachrome® number: 4240 CAS registry number: 90-33-5

Appearance: colorless, crystalline solid

Absorption maximum (in bas. ethanol): 372 nm

Molar absorptivity: 2.10 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 445 nm

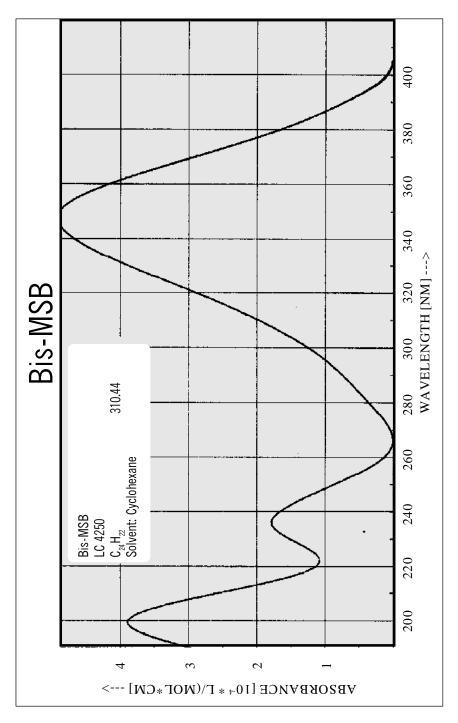
For research and development purposes only.

# Lasing Performance

Laser dye for pulsed and CW operation; tunable around 450 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Nitrogen Flashlamp CW, Ar+	337 - UV	- 454 460	370 - 580 - 460 - 560	- - -	0.51 0.51	various Ethanol Eg.	1,2,3 4,5 6

- 1. A. Dienes, *Appl. Phys.* 7, 135 (1975).
- 2. C. V. Shank et al., Appl. Phys. Letters 16(10), 405 (1970).
- 3. A. Dienes et al., IEEE J. Quant. Electr. QE-9, 833 (1973).
- 4. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).
- 5. B. B. Snavely et al., *Appl. Phys. Letters* 11(9), 275 (1967).
- 6. J. M. Yarborough, *Appl. Phys. Letters* 24(12), 629 (1974).



# Bis-MSB (LC 4250)

# Constitution

p-Bis(o-methylstyryl)-benzene

C<sub>24</sub>H<sub>22</sub> · MW: 310.44

#### Characteristics

Lambdachrome® number: 4250 CAS registry number: 13280-61-0

Appearance: yellow, crystalline solid

Absorption maximum (in cyclohexane): 350 nm

Molar absorptivity:  $4.88 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 418 nm

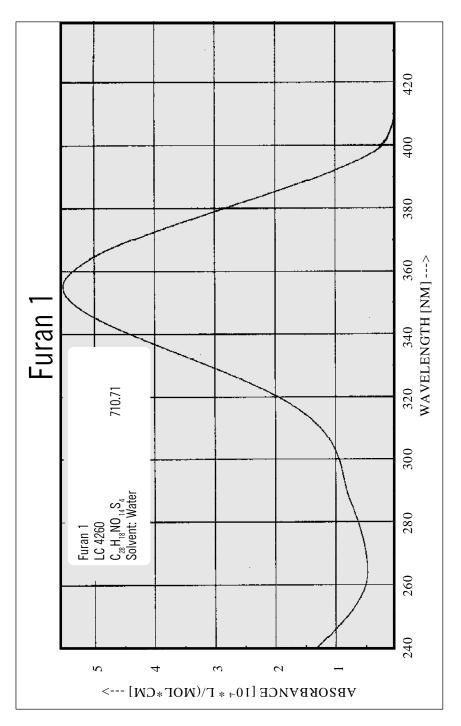
For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed operation; tunable around 400 nm.

Pump	)	D	ye Laser Ch				
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	<i>423</i>	414 - 428	8.3	0.24	Dioxane	1
Nitrogen	<i>337</i>	421	412 - 435	rel.	0.14	Dioxane	2, 3
Nd:YAG, 3rd	<i>355</i>	-	-	-	1.08	Dioxane	4
Flashlamp	-	420	-	-	0.12	Toluene	5

- 1. Lambda Physik, Wall Chart 6/83.
- 2. O. Uchino et al., Appl. Phys. 19, 35 (1979).
- 3. Lambda Physik, *Data Sheet.*
- 4. A. J. Cox et al., Appl. Phys. Letters 31(6), 389 (1977).
- 5. H. W. Furumoto et al. *J. Appl. Phys.* <u>40</u>, 4204 (1969).



# Furan 1 (LC 4260)

#### Constitution

Benzofuran,2,2'-[1,1'-biphenyl]-4,4'-diyl-bis-tetrasulfonic acid (tetrasodium salt)

 $C_{28}H_{18}NO_{14}S_4 \cdot MW: 710.71$ 

#### Characteristics

Lambdachrome® number: 4260 CAS registry number: -

Appearance: colorless, crystalline solid

Absorption maximum (in water): 355 nm

Molar absorptivity: 5.51 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

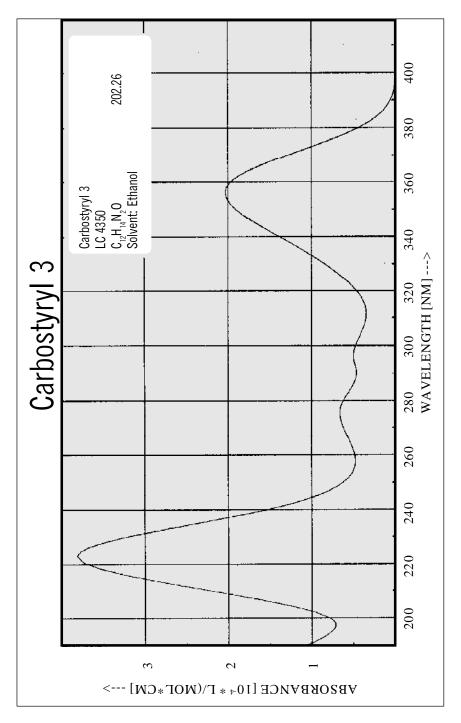
Fluorescence maximum:
For research and development purposes only.

# Lasing Performance

Laser dye for pulsed operation; tunable around 420 nm.

Pum	ס	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Nd:YAG, 3rd	308 337 355	421 425 421	410 - 435 414 - 445 410 - 435	10 3.5 10	0.60 1.35 0.26	Methanol Methanol Methanol	1 1 2

- 1. Lambda Physik, Data Sheet.
- 2. Lambda Physik, *Wall Chart* 1996.



# Carbostyryl 3 (LC 4350)

# Constitution

7-Dimethylamino-4-methylquinolon-2

C<sub>12</sub>H<sub>14</sub>N<sub>2</sub>O · MW: 202.26

#### Characteristics

Lambdachrome® number: 4350 CAS registry number: -

Appearance: colorless, crystalline solid

Absorption maximum (in ethanol): 360 nm

Molar absorptivity: 2.03 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 425 nm

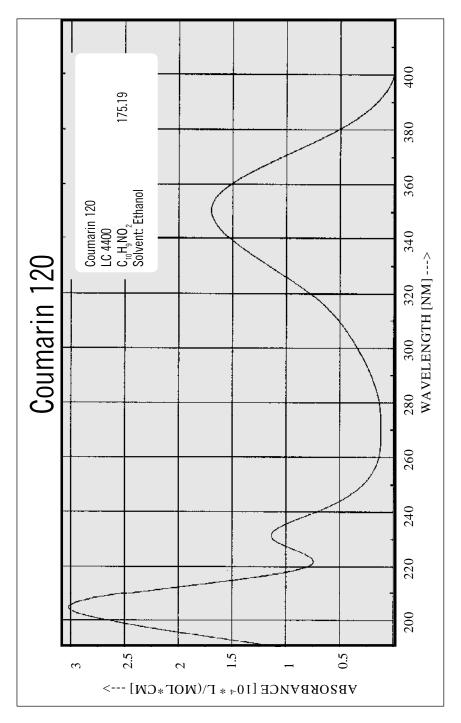
For research and development purposes only.

# Lasing Performance

Laser dye for pulsed and CW operation; tunable around 440 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range			Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
Flashlamp CW, Ar+	- UV	430 440	420 - 440 415 - 490	poor 3	0.04 0.6	Methanol Eg.	1, 2 3

- 1. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).
- 2. J. B. Marling et al., *Appl. Optics* <u>13(10)</u>, 2317 (1974).
- 3. J. M. Yarborough, *Appl. Phys. Letters* 24(12), 629 (1974).



# Coumarin 120 (LC 4400)

### Constitution

7-Amino-4-methylcoumarin Coumarin 440

C<sub>10</sub>H<sub>9</sub>NO<sub>2</sub> · MW: 175.19

#### Characteristics

Lambdachrome® number: 4400 CAS registry number: 26093-31-2

Appearance: slightly yellow, crystalline solid

Absorption maximum (in ethanol): 354 nm

Molar absorptivity:  $1.81 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 435 nm

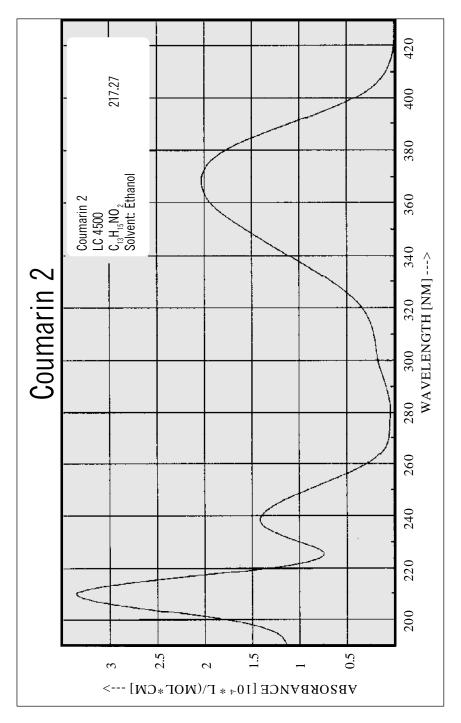
For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 440 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	<i>308</i>	441	423 - 462	15	0.82	Methanol	1,2
Nitrogen	<i>337</i>	438	418 - 465	rel.	0.25	Methanol	<i>2, 3</i>
Nd:YAG, 3rd	<i>355</i>	440	420 - 470	16	0.25	Methanol	1, 4
Flashlamp	-	440	420 - 470	-	0.04	Methanol	5
CW, Ar+	UV	450	425 - 475	-	0.52	Eg.	6

- 1. Lambda Physik, Wall Chart 1996.
- 2. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 3. Lambda Physik, Data Sheet.
- 4. K. Kato, IEEE J. Quantum Electr. QE-11, 373 (1975).
- 5. J. B. Marling et al., Appl. Optics 13(10), 2317 (1974).
- 6. J. M. Yarborough, Appl. Phys. Letters 24(12), 629 (1974).



# Coumarin 2 (LC 4500)

### Constitution

7-Amino-4-methylcoumarin Coumarin 450

 $C_{13}H_{15}NO_2 \cdot MW: 217.27$ 

#### Characteristics

Lambdachrome® number: 4500
CAS registry number: 26078-25-1

Appearance: slightly yellow, crystalline solid

Absorption maximum (in ethanol): 366 nm

Molar absorptivity: 2.02 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 443 nm

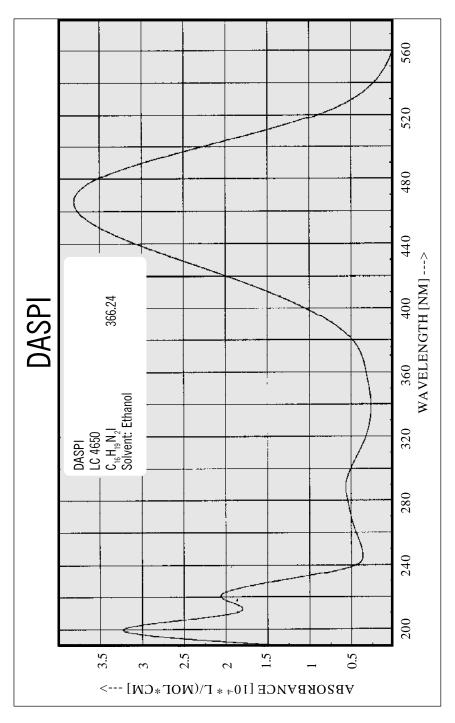
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 450 nm.

Pump	)	Dye Laser Characteristics						
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.	
	[nm]	[nm]	[nm]	[%]	[g/l]			
XeCI-Excimer	308	448	<i>432- 475</i>	15	1.50	Methanol	1, 2	
Nitrogen	<i>337</i>	444	<i>426 - 475</i>	rel.	0.40	Methanol	3, 4	
Nd:YAG, 3rd	<i>355</i>	450	435 - 467	10	0.30	Methanol	<i>3, 5</i>	
Flashlamp	-	450	440 - 458	-	-	Ethanol	6	
CW, Ar+	UV	450	430 - 480	-	0.59	Bz./Eg.	7	

- 1. Lambda Physik, Wall Chart 1996.
- 2. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 3. Lambda Physik, Data Sheet.
- 4. R. J. von Gutfeld et al., IEEE J. Quantum Electron. QE-6, 332 (1970).
- 5. D. M. Guthals, J. W. Nibler, Opt. Commun. 29(3), 322 (1979).
- 6. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).
- 7. Coherent, CW Dye Laser Fact Sheets.



# DASPI (LC 4650)

## Constitution

2-(p-Dimethylaminostyryl)-pyridylmethyl lodide

 $C_{16}H_{19}N_{2}I \cdot MW: 366.24$ 

### Characteristics

Lambdachrome® number: 4650

CAS registry number:

Appearance: orange, crystalline solid

Absorption maximum (in ethanol): 472 nm

Molar absorptivity: 3.83 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum: -

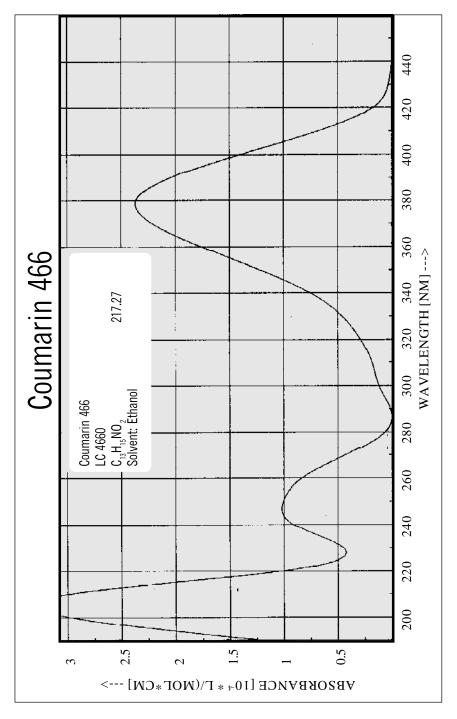
 $For \ research \ and \ development \ purposes \ only.$ 

## Lasing Performance

Saturable absorber for flashlamp pumped Coumarin 1, 102, 466, and 6H dye lasers<sup>1.)</sup>.

### References

1. W. Sibbett, J. R. Taylor, Opt. Commun. 46(1), 32 (1983).



# Coumarin 466 (LC 4660)

## Constitution

7-Diethylaminocoumarin LD 466 · C1H

 $C_{13}H_{15}NO_2 \cdot MW: 217.27$ 

### Characteristics

Lambdachrome® number: 4660

CAS registry number:

Appearance: yellow, crystalline solid

Absorption maximum (ethanol): 380 nm

Molar absorptivity: 2.38 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

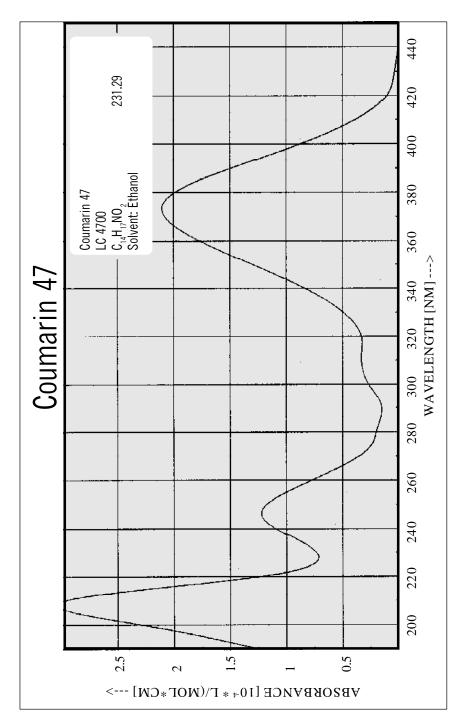
Fluorescence maximum: For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 465 nm.

Pum	p	Dy	Dye Laser Characteristics				
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
Flashlamp	-	466	-	-	0.16	Ethanol	1, 2

- 1. E. J. Schimitschek et al., *Opt. Commun.* <u>16(3)</u>, 313 (1976).
- 2. E. A. Stappaerts, Appl. Optics 16(12), 3079 (1977).



# Coumarin 47 (LC 4700)

### Constitution

7-Diethylamino-4-methylcoumarin Coumarin 460 · Coumarin 1

C<sub>14</sub>H<sub>17</sub>NO<sub>2</sub> · MW: 231.29

#### Characteristics

Lambdachrome® number: 4700 CAS registry number: 99-44-1

Appearance: slightly yellow, crystalline solid

Absorption maximum (in ethanol): 373 nm

Molar absorptivity: 2.10 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 450 nm

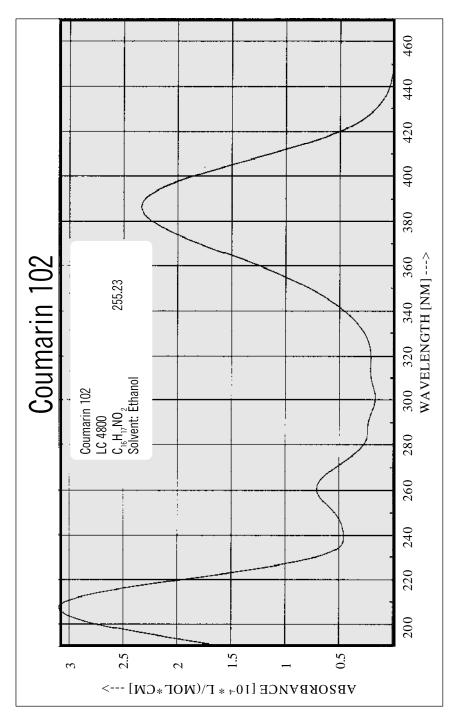
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 450 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	<i>308</i>	456	440 - 484	18	1.60	Methanol	<i>1, 2, 3</i>
Nitrogen	<i>337</i>	453	436 - 486	rel.	0.66	Methanol	3, 4
Nd:YAG, 3rd	<i>355</i>	460	444 - 476	15	0.3	Methanol	1, 5
Flashlamp	-	460	435 - 490	-	0.02	Ethanol	6, 7
CW, Ar+	UV	470	450 - 500	-	1.76	MeOH/Eg.	8

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* <u>38(5,6)</u>, 403 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. D. M. Guthals, J. W. Nibbler, Opt. Commun. 29(3), 322 (1977).
- 6. J. B. Marling et al., *Appl. Optics* 13(10), 2317 (1974).
- 7. J. B. Marling et al., *Appl. Phys. Letters* <u>17(12)</u>, 527 (1970).
- 8. Coherent, CW Dye Laser Fact Sheets.



## Coumarin 102 (LC 4800)

#### Constitution

2,3,5,6-1H,4H-Tetrahydro-8-methylquinolizino-[9,9a,1-gh]-coumarin

C<sub>16</sub>H<sub>17</sub>NO<sub>2</sub> · MW: 255.23

#### Characteristics

Lambdachrome® number: 4800 CAS registry number: 41267-76-9

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 389 nm

Molar absorptivity:  $2.15 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 465 nm

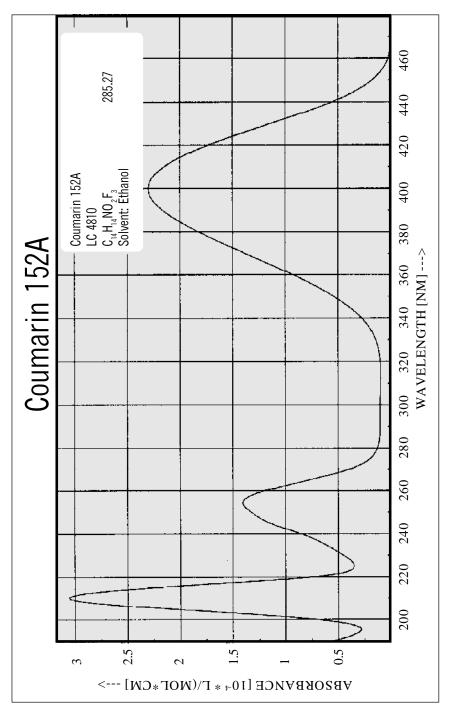
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 480 nm.

Pump	Pump Dye Laser Characteristics						
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
							_
XeCI-Excimer	308	480	460 - 510	18	2.30	Methanol	1, 2
Nitrogen	<i>337</i>	470	454 - 506	rel.	1.44	Methanol	3
Nd:YAG, 3rd	<i>355</i>	480	462 - 497	15	0.40	Methanol	1, 4
Flashlamp	-	480	460 - 530		0.05	Methanol	5
CW, Ar+	UV	482	463 - 515	-	2.0	Bz./Eg.	1, 6

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 403 (1981).
- 3. Lambda Physik, *Data Sheet*.
- 4. K. Kato, IEEE J. Quant. Electron. QE-11, 373 (1975).
- 5. J. B. Marling et al., *Appl. Optics* 13(10), 2317 (1974).
- 6. Coherent, CW Dye Laser Fact Sheets.



# Coumarin 152A (LC 4810)

## Constitution

7-Diethylamino-4-trifluormethylcoumarin Coumarin 481 · C1F

C<sub>14</sub>H<sub>14</sub>NO<sub>2</sub>F<sub>3</sub> · MW: 285.27

#### Characteristics

Lambdachrome® number: 4810 CAS registry number: 41934-47-8

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 405 nm

Molar absorptivity:  $2.16 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 510 nm

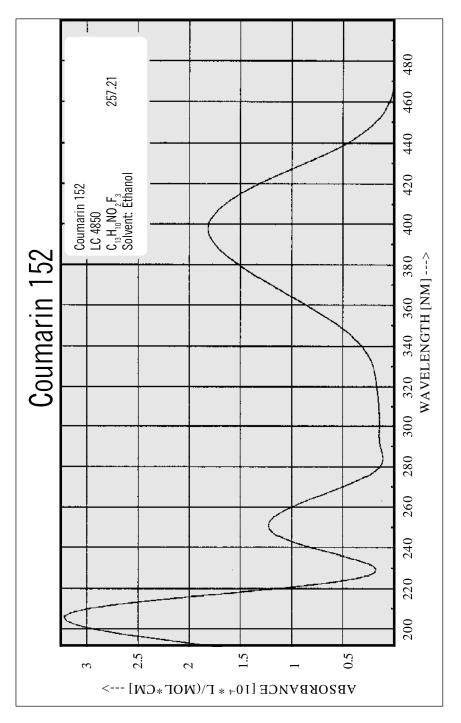
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 500 nm.

Pump	)	D	ye Laser Ch				
Source	Wavelength	Peak	Range		Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer Nitrogen Nd:YAG, 3rd Flashlamp	308 337 355 -	517 490 520 481	491 - 553 461 - 549 500 - 540 -	4 rel. -	1.84 1.71 2.85 0.21	Ethanol Dioxane Ethanol Dioxane	1, 2 2, 3 4 5

- 1. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 403 (1981).
- 2. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 3. J. A. Halstead, R. R. Reeves, Opt. Commun. 27(2), 273 (1978).
- 4. D. M. Guthals, J. W. Nibbler, Opt. Commun. 29(3), 322 (1977).
- 5. E. J. Schimitschek et al., IEEE J. Quantum Electron. QE-9, 781 (1973).



# Coumarin 152 (LC 4850)

### Constitution

7-Dimethylamino-4-trifluormethylcoumarin Coumarin 485 · C2F

 $C_{13}H_{10}NO_{2}F_{3} \cdot MW: 257.21$ 

### Characteristics

Lambdachrome® number: 4850 CAS registry number: 53518-14-2

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 397 nm

Molar absorptivity:  $1.97 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 510 nm

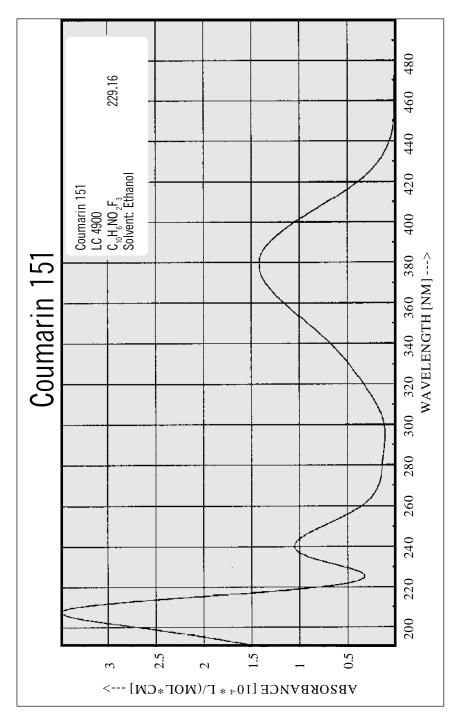
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 520 nm.

Pump	)	D	ye Laser Cha				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Flashlamp	308 337	520 532 -	490 - 570 495 - 560 518 - 531	5.5 - -	2.00 - -	Ethanol Ethanol Ethanol	1 2 3

- 1. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* <u>38(5,6)</u>, 403 (1981).
- 2. Lambda Physik, Data Sheet.
- 3. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).



# Coumarin 151 (LC 4900)

## Constitution

7-Amino-4-trifluormethylcoumarin Coumarin 490 · C3F

C<sub>10</sub>H<sub>6</sub>NO<sub>2</sub>F<sub>3</sub> · MW: 229.16

### Characteristics

Lambdachrome® number: 4900 CAS registry number: 53518-13-3

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 382 nm

Molar absorptivity:  $1.70 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 480 nm

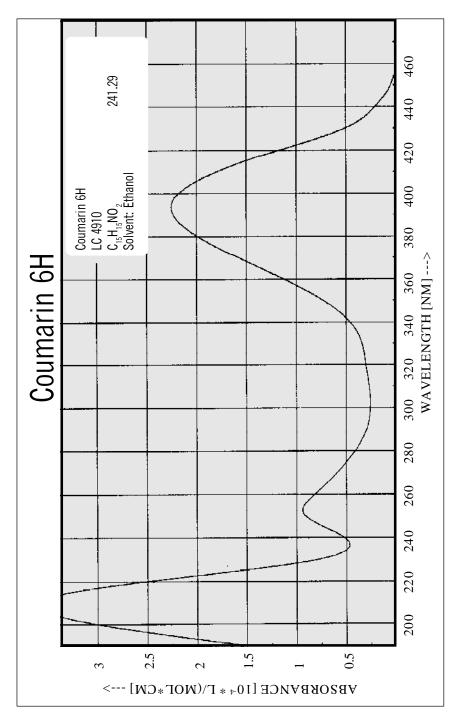
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 490 nm.

Pum	р	D	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.	
	[nm]	[nm]	[nm]	[%]	[g/l]			_
Flashlamp	-	490	481 - 493	-	-	Ethanol	1, 2	

- 1. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).
- 2. G. A. Reynolds, K. H. Drexhage, Opt. Commun. 13(3), 222 (1975).

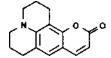


# Coumarin 6H (LC 4910)

### Constitution

2,3,5,6-1H,4H-Tetrahydroquinolizino-[9,9a,1-gh]coumarin

C<sub>15</sub>H<sub>15</sub>NO<sub>2</sub> · MW: 241.29



### Characteristics

Lambdachrome® number: 4910 CAS registry number: 58336-35-9

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 396 nm

Molar absorptivity:  $2.50 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

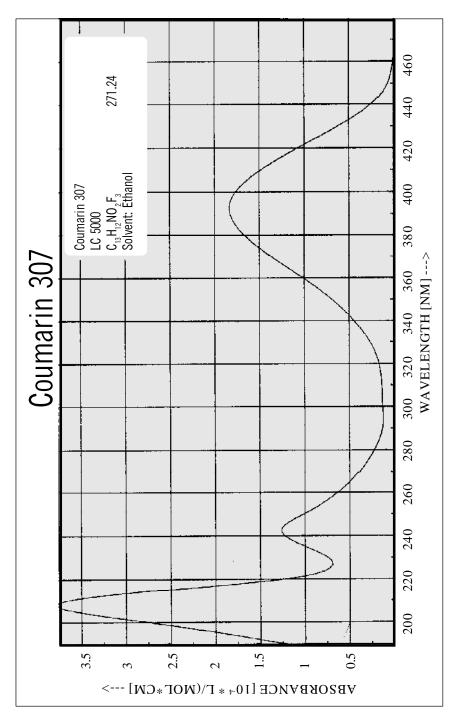
Fluorescence maximum: - For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 490 nm.

Pump	)	D	ye Laser Cha				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Flashlamp CW, Kr+	308 - VIO	491 - 490	463 - 522 477 - 493 476 - 515	5.7 - -	1.33 - 1.20	Etahnol Ethanol Eg./Bz.	1 2 3

- 1. Lambda Physik.
- 2. A. N. Fletcher, Appl. Phys. 14, 295 (1977).
- 3. Lambda Physik, Data Sheet.



# Coumarin 307 (LC 5000)

### Constitution

7-Ethylamino-6-methyl-4-trifluormethylcoumarin Coumarin 503

C<sub>13</sub>H<sub>12</sub>NO<sub>2</sub>F<sub>3</sub> · MW: 271.24

#### Characteristics

Lambdachrome® number: 5000 CAS registry number: 55804-66-5

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 395 nm

Molar absorptivity: 1.85 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 490 nm

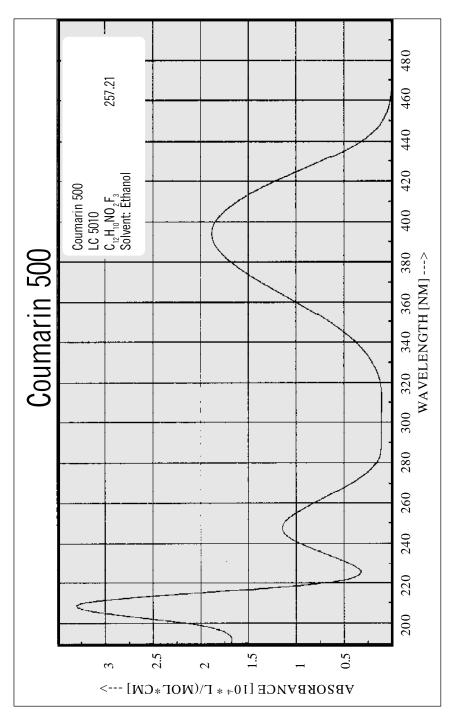
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 500 nm.

Pump	1	Dye Laser Characteristics					
Source	Wavelength	Peak	Range			Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer Nitrogen Nd:YAG, 3rd Flashlamp	308 337 355 -	500 504 508	479 - 553 478 - 547 485 - 546 490 - 510	16 rel. 15 -	3.40 1.60 0.70 -	Methanol Methanol Methanol Ethanol	1, 2 3, 4 1 6

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 403 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. A. N. Fletcher, *Appl Phys.* <u>14</u>, 295 (1977).



# Coumarin 500 (LC 5010)

## Constitution

7-Ethylamino-4-trifluormethylcoumarin

C<sub>12</sub>H<sub>10</sub>NO<sub>2</sub>F<sub>3</sub> · MW: 257.21

### Characteristics

Lambdachrome® number:

CAS registry number:

Appearance:

Absorption maximum (in ethanol):

Molar absorptivity:

Fluorescence maximum:

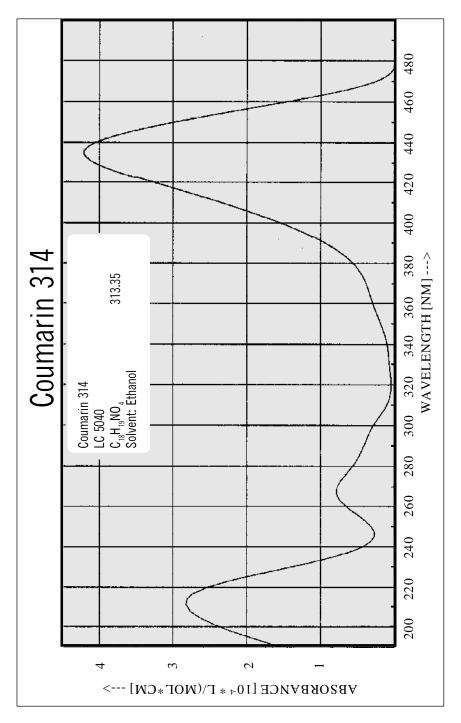
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 500 nm.

Pump	)	D	ye Laser Ch	aracteri	stics		
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Nd:YAG, 3rd Flashlamp	308 337 355	502 503 518 -	480 - 520 473 - 562 498 - 546 -	12 rel. 10 -	2.40 1.40 0.70 -	Methanol Ethanol Methanol -	1, 2 2 3, 6 4, 5

- 1. Lambda Physik
- 2. F. Bos, *Appl. Optics* <u>20(20)</u>, 3553 (1981).
- 3. D. M. Guthals, J. W. Nibler, Opt. Commun. 29(3), 322(1979).
- 4. Th. Varghese, Opt. Commun. 44(5), 353(1983).
- 5. A. N. Fletcher, *Appl. Phys.* 14, 295 (1977).
- 6. Lambda Physik, Wall Chart 1996.



# Coumarin 314 (LC 5040)

## Constitution

2,3,5,6-1H,4H-Tetrahydro-9-carboethoxyquinolizino-[9,9a,1-gh]coumarin

 $C_{18}H_{19}NO_4 \cdot MW: 313.35$ 

### Characteristics

Lambdachrome® number: 5040

CAS registry number: 55804-66-5

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 436 nm

Molar absorptivity:  $4.70 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 480 nm

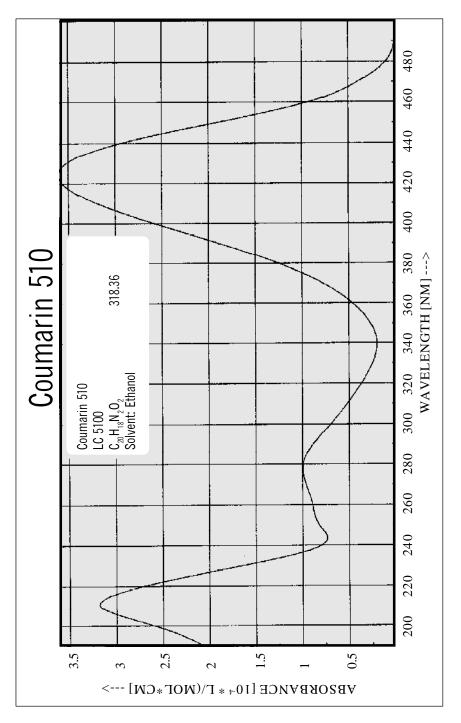
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 505 nm.

Pum	р	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Flashlamp	-	506	490 - 504	-	-	Ethanol	1, 2

- 1. K. H. Drexhage et al., IEEE J. Quantum Electron. QE-10, 695 (1974).
- 2. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).



# Coumarin 510 (LC 5100)

## Constitution

2,3,5,6-1H,4H-Tetrahydro-9-(3-pyridyl)-quinolizino-[9,9a,1-gh]coumarin

 $C_{20}H_{18}N_{2}O_{2} \cdot MW: 318.36$ 

#### Characteristics

Lambdachrome® number: 5100 CAS registry number: -

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 425 nm

Molar absorptivity: 3.70 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

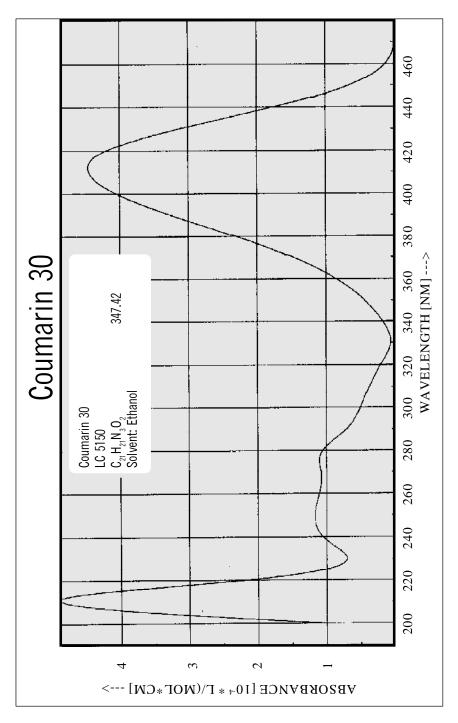
Fluorescence maximum:
For research and development purposes only.

Lasing Performance

Efficient laser dye for pulsed operation; tunable around 510 nm.

Pum	р	D	ye Laser Ch	aracteri	stics		
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Flashlamp CW, Ar+	- VIO	511 525	504 - 511 495 - 565	-	- 1.0	Ethanol Bz./Eg	1 2

- 1. A. N. Fletcher et al., *Opt. Commun.* <u>47(1)</u>, 57 (1983).
- 2. Lambda Physik.



# Coumarin 30 (LC 5150)

## Constitution

3-(2'-N-Methylbenzimidazolyl)-7-N,N-diethylaminocoumarin Coumarin 515

 $C_{21}H_{21}N_3O_2 \cdot MW: 347.42$ 

### Characteristics

Lambdachrome® number: 5150 CAS registry number: 41044-12-6

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 412 nm

Molar absorptivity:  $4.45 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 488 nm

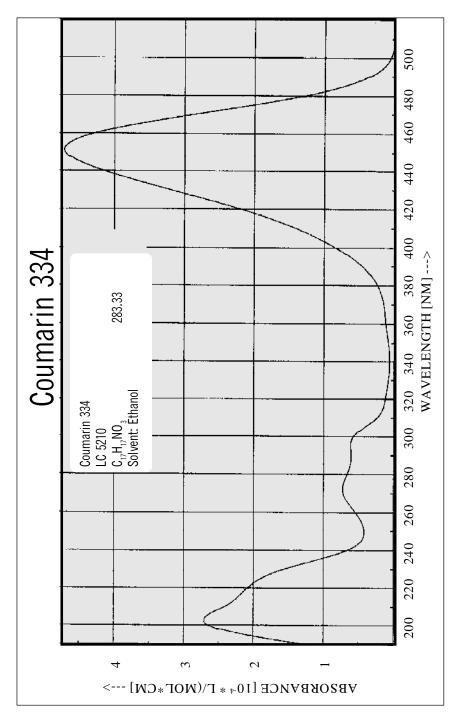
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 510 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak Гпт1	Range [nm]	Effic. 「%1	Conc. [q/l]	Solvent	Ref.
	["""]	ניייין	L''''	[/0]	[97]		
Flashlamp	-	508	480 - 540	-	0.07	Methanol	1, 2
CW, Kr+	VIO	510	480 - 555	12	0.26	MeOH/Eg	3

- 1. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).
- 2. J. B. Marling et al., Appl. Optics. 13(10), 2317 (1974).
- 3. Coherent, CW Dye Laser Fact Sheets.



# Coumarin 334 (LC 5210)

### Constitution

2,3,5,6-1H,4H-Tetrahydro-9-acetylquinolizino-[9,9a,1-gh]-coumarin Coumarin 521

C<sub>17</sub>H<sub>17</sub>NO<sub>3</sub> · MW: 283.33

#### Characteristics

Lambdachrome® number: 5210

CAS registry number: 55804-67-6
Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 450 nm

Molar absorptivity: 4.73 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 495 nm

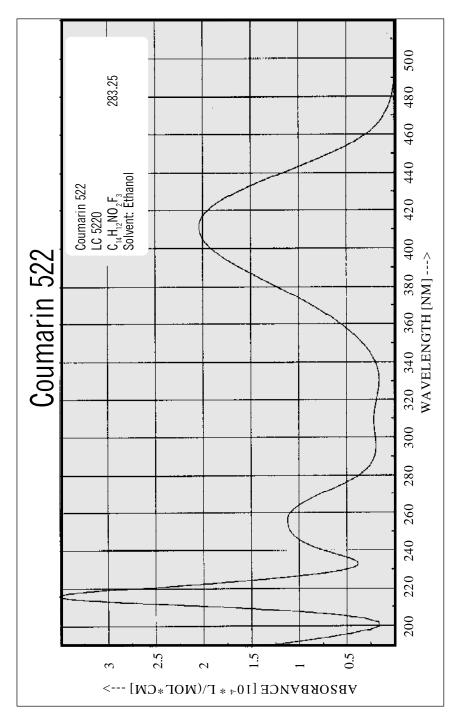
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 480 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer Flashlamp	308 -	<i>520</i> -	506 - 537 507 - 512	12	1.50 -	Methanol Methanol	1, 2 3, 4

- 1. Lambda Physik, Wall Chart 6/83.
- 2. V. S. Antonov, K. L. Hohla, Appl. Phys. <u>B32</u>, 9(1983).
- 3. A. N. Fletcher *Appl. Phys.* <u>14</u>, 295 (1977).
- 4. G. A. Reynolds, K. H. Drexhage, Opt. Commun. 13(3), 222(1975).



# Coumarin 522 (LC 5220)

### Constitution

N-Methyl-4-trifluormethylpiperidino-[3,2-g]-coumarin C8F

C<sub>14</sub>H<sub>12</sub>NO<sub>2</sub>F<sub>3</sub> · MW: 283.25

#### Characteristics

Lambdachrome® number: 5220

CAS registry number: 55318-19-7

Appearance: yellow, crystalline solid
Absorption maximum (in ethanol): 410 nm

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Molar absorptivity: 2.06 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 516 nm

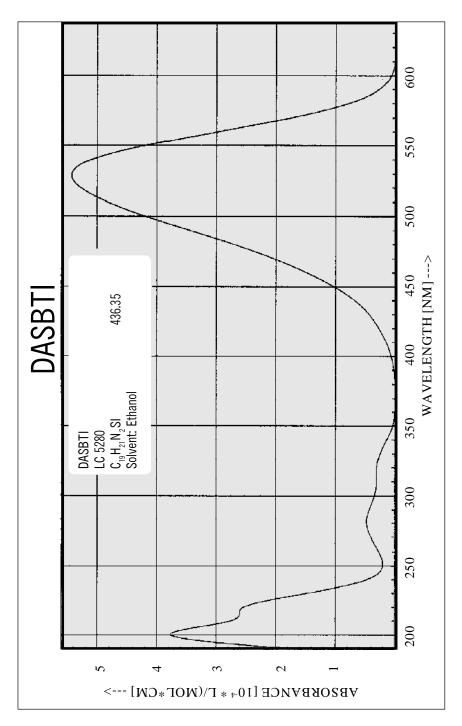
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 520 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Nitrogen Nd:YAG, 3rd Flashlamp	337 355 -	520 525 -	495 - 575 505 - 550 518 - 528	14 - -	0.04 1.41 -	DMF Ethanol Ethanol	1 2 3

- 1. P. R. Hammond, *IEEE J. Quantum Electron.* <u>QE-15(7)</u>, 624(1979).
- 2. D. M. Guthals, J. W. Nibler, Opt. Commun. 29(3), 322(1979).
- 3. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295 (1977).



## DASBTI (LC 5280)

### Constitution

2-(p-Dimethylaminostyryl)-benzothiazolylethyl lodide

 $C_{19}H_{21}N_2SI \cdot MW: 436.35$ 

$$CH = CH - (CH_3)_2$$

$$CH = CH - (CH_3)_2$$

### Characteristics

Lambdachrome® number: 5280 CAS registry number: -

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 530nm

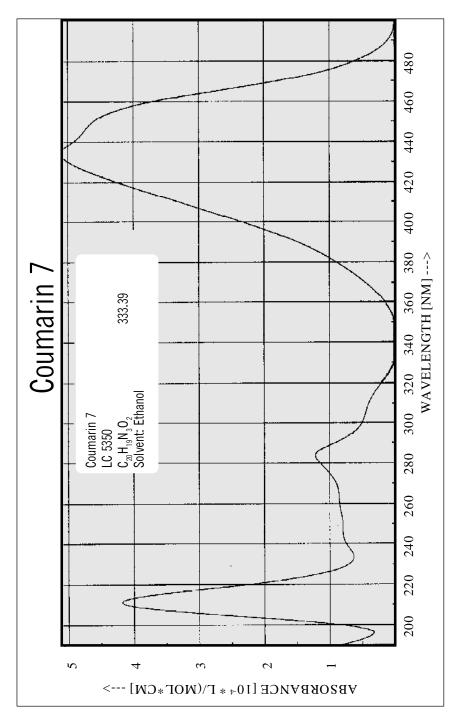
Molar absorptivity:  $5.49 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum: - For research and development purposes only.

## Lasing Performance

Saturable absorber for flashlamp pumped Coumarin 6, 153, 522, ands Rhodamine 110 dye lasers <sup>1,,2,)</sup>. Applicable around 530 nm.

- 1. W. Sibbett, J. R. Taylor, *Appl. Phys.* <u>B29</u>, 191(1982).
- 2. W. Sibbett, J. R. Taylor, *IEEE J. Quantum Electron*. <u>QE-19(4)</u>, 558(1983).



# Coumarin 7 (LC 5350)

### Constitution

3-(2'-Benzimidazolyl)-7-N,N-diethylaminocoumarin Coumarin 535

 $C_{20}H_{19}N_3O_2 \cdot MW: 333.39$ 

### Characteristics

Lambdachrome® number: 5350 CAS registry number: 27425-55-4

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 433 nm

Molar absorptivity: 5.05 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 493 nm

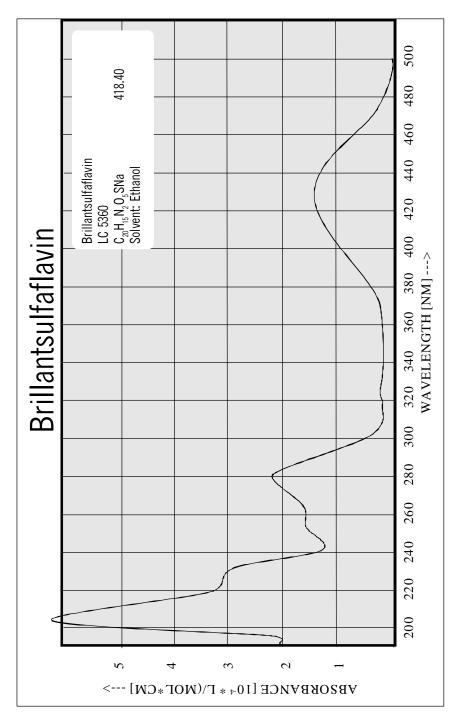
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 530 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. 「%1	Conc. [q/l]	Solvent	Ref.
Flashlamp CW, Ar+	- VIO	- 530	517 - 527 495 - 570	- 9	- 1.00	Ethanol Bz.	1 2

- 1. A. N. Fletcher *Appl. Phys.* <u>14</u>, 295 (1977).
- 2. J. M. Yarborough, *Appl. Phys. Lett.* 24(12), 629 (1974).



## Brillantsulfaflavin (LC 5360)

#### Constitution

Pilot 512

 $C_{20}H_{15}N_{2}O_{5}SNa \cdot MW: 418.40$ 

#### Characteristics

Lambdachrome® number: 5360 CAS registry number: 2391-30-2

Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 423 nm

Molar absorptivity:  $1.41 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 517 nm

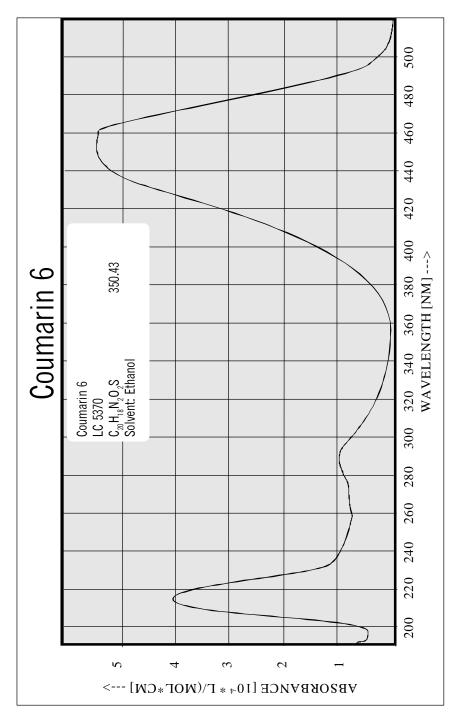
For research and development purposes only.

### Lasing Performance

Laser dye for pulsed operation; tunable around 540 nm.

Pum	р	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Flashlamp	-	540	508 - 574	-	0.21	Ethanol	1, 2

- 1. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* <u>11(5)</u>, 692 (1972).
- 2. J. B. Marling et al., IEEE J. Quantum Electr. QE-7, 498 (1971).



## Coumarin 6 (LC 5370)

#### Constitution

3-(2'-Benzothiazolyl)-7-diethylaminocoumarin Coumarin 540

 $C_{20}H_{18}N_{2}O_{2}S \cdot MW: 350.43$ 

#### Characteristics

Lambdachrome® number: 5370

CAS registry number: 38215-35-0

Appearance: orange, crystalline solid

Absorption maximum (in ethanol): 458 nm

Molar absorptivity: 5.4 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 505 nm

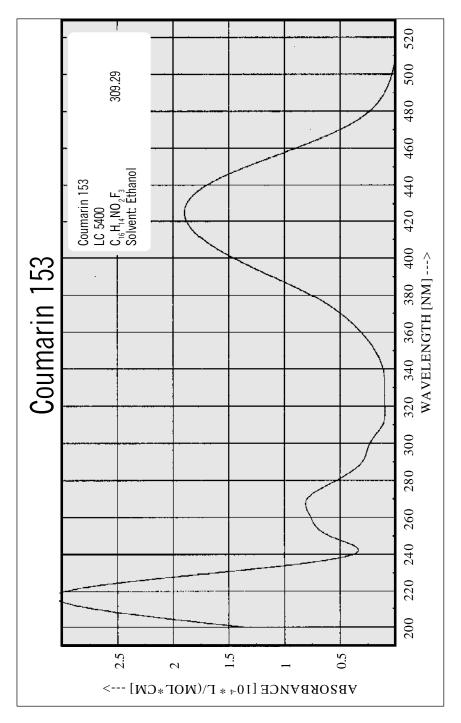
For research and development purposes only.

#### Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 540 nm.

Pump	)	D	Dye Laser Characteristics				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Flashlamp CW, Ar+	308 - 488	534 - 535	515 - 558 530 - 539 510 - 550	9 - -	0.84 - 2.0	DMSO Ethanol Bz./Eg.	1, 2 3 4, 5

- 1. Lambda Physik.
- 2. O. Uchino et al., Appl. Phys. 19, 35(1979).
- 3. A. N. Fletcher, *Appl. Phys.* <u>14</u>, 295(1977).
- 4. Lambda Physik, Wall Chart 1996.
- 5. T. F. Johnston at al., *Appl. Optics* 21(13), 2307(1982).



## Coumarin 153 (LC 5400)

#### Constitution

2,3,5,6-1H,4H-Tetrahydro-8-trifluormethylquinolizino-[9,9a,1-gh]coumarin Coumarin 540A, C6F

 $C_{16}H_{14}NO_{2}F_{3} \cdot MW: 309.29$ 

#### Characteristics

Lambdachrome® number: 5400 CAS registry number: 53518-18-6

CAS registry number: 53518-18-6
Appearance: yellow, crystalline solid

Absorption maximum (in ethanol): 423 nm

Molar absorptivity: 1.89 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 530 nm

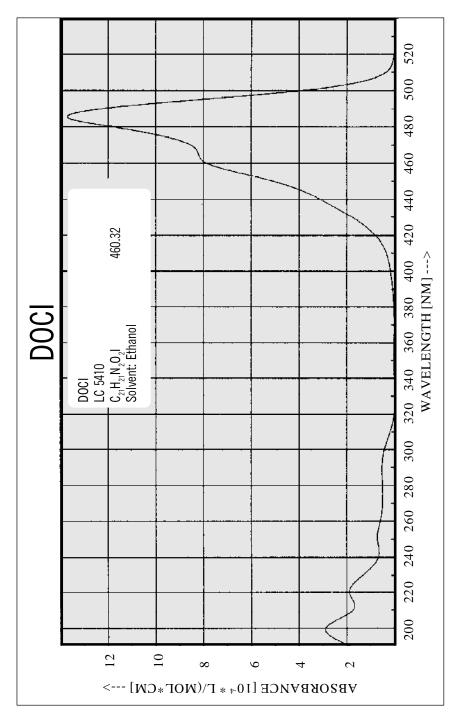
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 500 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range			Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer Nd:YAG, 3rd Nitrogen Flashlamp	308 355 337 -	540 540 540 -	522 - 600 516 - 575 517 - 590 528 - 547	15 18 rel. -	4.20 2.36 3.10	Methanol Methanol Methanol Methanol	1, 2, 3 1 4, 5 6

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 403 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3552 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. R. E. Drullinger, *Opt. Commun.* 39(4), 263 (1981).
- 6. A. N. Fletcher *Appl. Phys.* <u>14</u>, 295 (1977).



## DOCI (LC 5410)

### Constitution

3,3'-Diethyloxacarbocyanine lodide

 $C_{21}H_{21}N_{2}O_{2}I \cdot MW: 460.32$ 

#### Characteristics

Lambdachrome® number: 5410

CAS registry number:

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 485 nm

Molar absorptivity:  $12.6 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

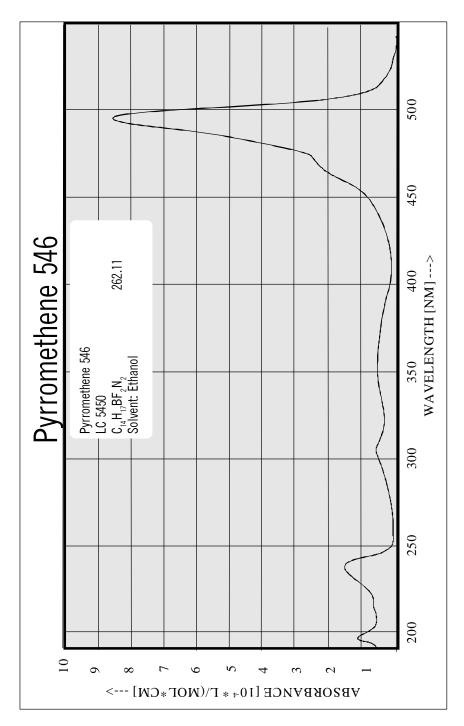
Fluorescence maximum: For research and development purposes only.

## Lasing Performance

Laser dye for pulsed operation; tunable around 540 nm. Saturable absorber for Coumarin 102 dye lasers; applicable around 480 nm <sup>2.,3.)</sup>.

Pum	р	Dy	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.	
Flashlamp	-	541	-	-	0.09	Glyzerine	1	

- 1. M Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* 11(5), 692 (1972).
- 2. J. C. Mialocq, P. Goujon, *Appl. Phys. Letters* 33(9), 819 (1978).
- 3. R. Wyatt, *Opt. Commun.* 38(1), 64 (1981).



# Pyrromethene 546 (LC 5450)

#### Constitution

4,4-Difluoro-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene 1,3,5,7,8-Pentamethylpyrromethenedifluoroborate Complex

C<sub>14</sub>H<sub>17</sub>BF<sub>2</sub>N<sub>2</sub>· MW: 262.11

#### Characteristics

Lambdachrome® number: 5450

CAS registry number: 121207-31-6

Appearance: orange, crystalline solid

Absorption maximum (in ethanol): 494 nm

Molar absorptivity: 8.58 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in methanol): 519 nm

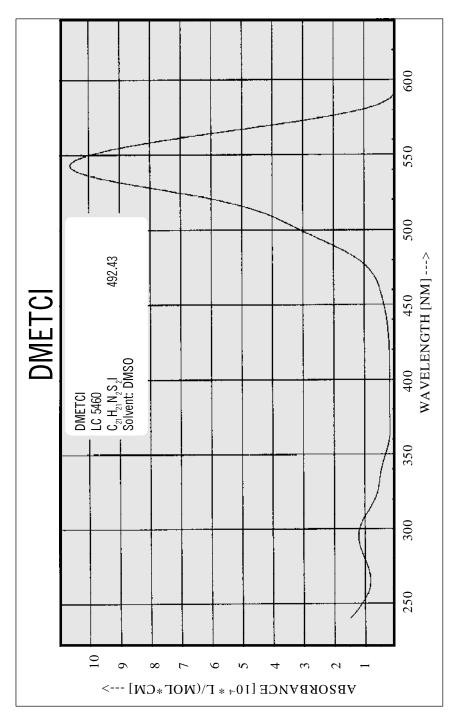
For research and development purposes only.

### Lasing Performance

Laser dye for flashlamp pumped dye lasers; tunable around 542 nm.

Pum	р	D	Dye Laser Characteristics				
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Flashlamp Flashlamp	- -	542 546	523-580 -	- -	0.066 0.039	Methanol Ethanol	1 2

- 1. M. Shah et al., *Heteroatom Chem.* <u>1(5)</u>, 389(1990).
- 2. Th. G. Pavlopoulos, M. Shah, J. H. Boyer, *Opt. Commun.* 70(5), 425 (1989).



## DMETCI (LC 5460)

### Constitution

3,3'-Dimethyl-9-ethylthiacarbocyanine lodide

C<sub>21</sub>H<sub>21</sub>N<sub>2</sub>S<sub>2</sub>I · MW: 492.43

#### Characteristics

Lambdachrome® number: 5460

CAS registry number:

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 540 nm

Molar absorptivity: 10.6 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum:

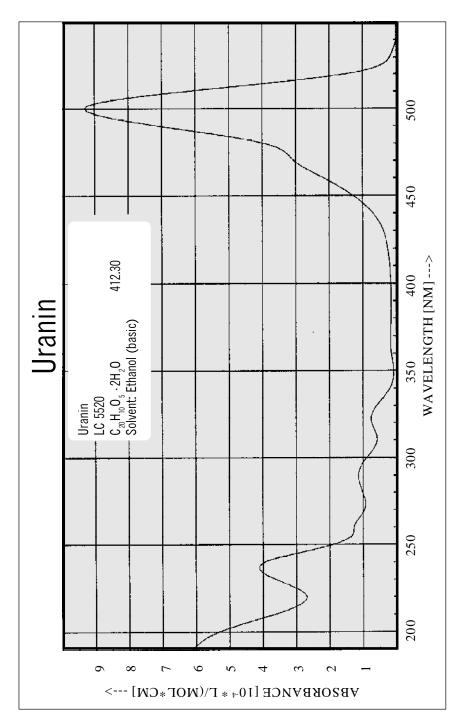
For research and development purposes only.

## Lasing Performance

Saturable absorber for Coumarin 153 dye lasers; applicable around 540 nm <sup>1.)</sup>.

#### References

1. W. Sibbett, J. R. Taylor, *Opt. Commun.* 43(1), 50 (1982).



## Uranin (LC 5520)

#### Constitution

Disodium Fluorescein  $C_{20}H_{10}O_5 \cdot 2H_2O \cdot MW: 412.30$ 

#### Characteristics

Lambdachrome® number: 5520 CAS registry number: 518-47-8

Appearance: red, crystalline solid

Absorption maximum (in basic ethanol): 500 nm

Molar absorptivity: 9.92 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in bas. ethanol): 521 nm

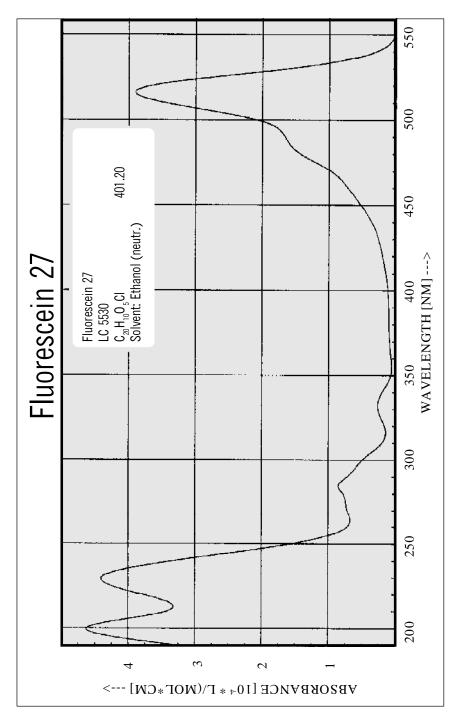
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 550 nm.

Pum	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	<i>540</i>	<i>532 - 561</i>	9	1.35	Methanol	1
Nitrogen	<i>337</i>	<i>538</i>	-	rel.	1.10	Methanol	<i>2, 3</i>
Nd:YAG, 3rd	<i>355</i>	<i>550</i>	<i>536 - 568</i>	-	2.06	Ethanol	4
Cu-vapor	510	<i>528</i>	-	-	0.42	Ethanol	5
Flashlamp	-	-	549 - 574	-	0.12	Methanol	6
CW, Ar+	all	560	530 - 590	7	1.76	MeOh/Eg.	7

- 1. Lambda Physik, Wall Chart 6/83.
- 2. A. Dienes, *Appl. Phys.* <u>7</u>, 135 (1975).
- 3. G. Capelle, D. Phillips, *Appl. Optics* <u>9(12)</u>, 2742 (1970).
- 4. D, M. Guthals, J. W. Nibler, Opt. Commun. 29(3), 322 (1979).
- 5. L. Masarnovskii et al., *Sov. J. Quantum Electr.* <u>9(7)</u>, 900 (1979).
- 6. D. A. Jennings, A. J. Varga, *J. Appl. Phys.* <u>42(12)</u>, 5171 (1971).
- 7. Coherent, CW Dye Laser Fact Sheets.



## Fluorescein 27 (LC 5530)

#### Constitution

9-(o-Carboxyphenyl)-2,7-dichloro-6-hydroxy-3H-xanthen-3-on 2,7-Dichlorofluorescein • Fluorescein 548

C<sub>20</sub>H<sub>10</sub>O<sub>5</sub>CI · MW: 401.20

#### Characteristics

Lambdachrome® number: 5530
CAS registry number: 76-54-0

Appearance: red, crystalline solid

Absorption maximum (in basic ethanol): 512 nm

Molar absorptivity: 11.0 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in bas. ethanol): 530 nm

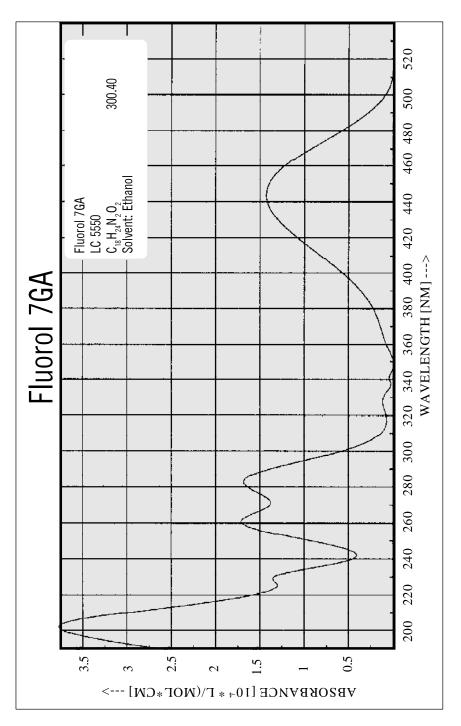
For research and development purposes only.

## Lasing Performance

Laser dye for pulsed and CW operation; tunable around 550 nm.

Pum	ס	Dye Laser Characteristics					
Source	Wavelength	Peak	Range			Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer Nitrogen Nd:YAG, 2nd Flashlamp	337	553 558 550 -	540 - 587 546 - 589 540 - 575 557 - 581	12 rel. 28 -	1.40 1.00 0.64 0.20	Methanol Ethanol Methanol Methanol	1 1, 2 3 4

- 1. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 2. E. D. Stokes et al., Opt. Commun. 5(4), 267 (1972).
- 3. Lambda Physik, Wall Chart 1996.
- 4. D. A. Jennings, A. J. Varga, J. Appl. Phys. 42(12), 5171 (1971).



## Fluorol 7GA (LC 5550)

### Constitution

Fluorol 555

 $C_{20}H_{24}N_2O_2 \cdot MW: 324.40$ 

#### Characteristics

Lambdachrome® number: 5550 CAS registry number: -

Appearance: red, crystalline solid

Absorption maximum (in methanol): 440 nm

Molar absorptivity: 1.40 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

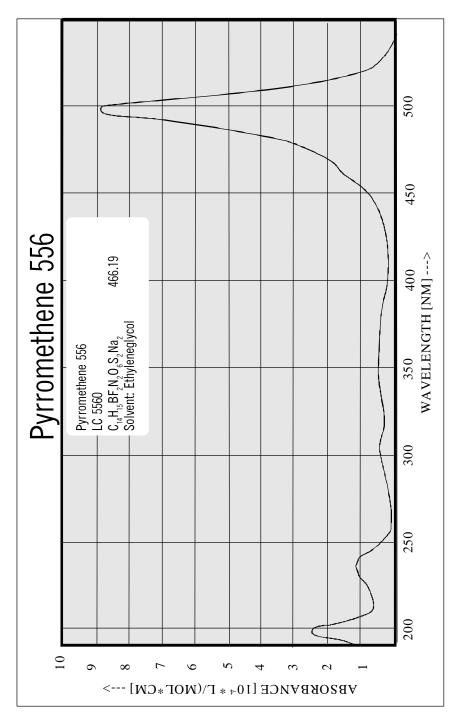
Fluorescence maximum: - For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsedoperation; tunable around 520 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	560	530 - 590	5		Methanol	1, 2
Flashlamp	-	-	530 - 600	-	0.10	Methanol	3

- 1. Lambda Physik.
- 2. F. Bos, Appl. Optics 20(20), 3553, (1981).
- 3. M. Lambropoulos, *Opt. Commun.* 15(1), 35 (1975).



# Pyrromethene 556 (LC 5560)

## Constitution

Disodium-1,3,5,7,8-pentamethylpyrromethene-2,6-disulfonate-difluoroborate complex

 $C_{14}H_{17}BF_2N_2O_6S_2Na_2 \cdot MW: 466.19$ 

#### Characteristics

Lambdachrome® number: 5560 CAS registry number: 121461-69-6

Appearance: yellow/orange, crystalline solid

Absorption maximum (in ethyleneglycol): 498 nm

Molar absorptivity: 8.88 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in methanol): 533 nm

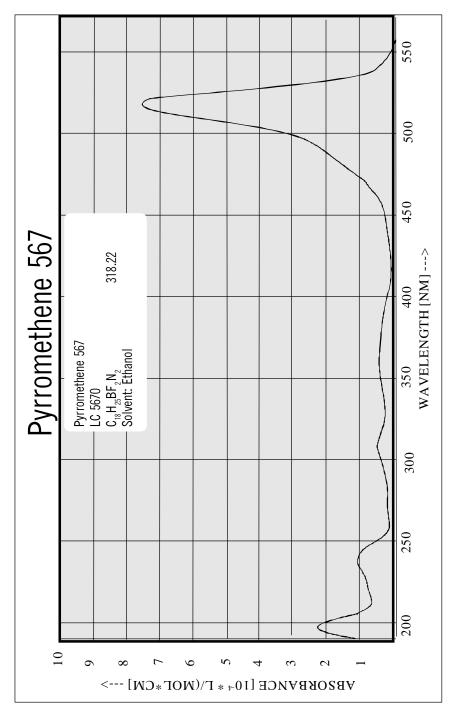
For research and development purposes only.

### Lasing Performance

Laser dye for pulsed and CW operation; tunable around 550 nm.

Pum	р	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Flashlamp CW, Ar+	- 458-514	561 553	540-580 530-624	- 45	0.03 0.93	Methanol Eg.	1 2

- 1. M. Shah et al., *Heteroatom Chem.* <u>1(5)</u>, 389(1990).
- 2. S. G. Guggenheimer et al., *Appl. Optics* <u>32(21)</u>, 3942 (1993).



## Pyrromethene 567 (LC 5670)

#### Constitution

4,4-Difluoro-2,6-diethyl-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene 2,6-Diethyl-1,3,5,7,8-pentamethylpyrromethenedifluoroborate Complex

C<sub>18</sub>H<sub>25</sub>BF<sub>2</sub>N<sub>2</sub> · MW: 318.22

#### Characteristics

Lambdachrome® number: 5670
CAS registry number: 131083-16-4

Appearance: orange/red, crystalline solid

Absorption maximum (in ethanol): 518 nm

Molar absorptivity: 7.73 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 547 nm

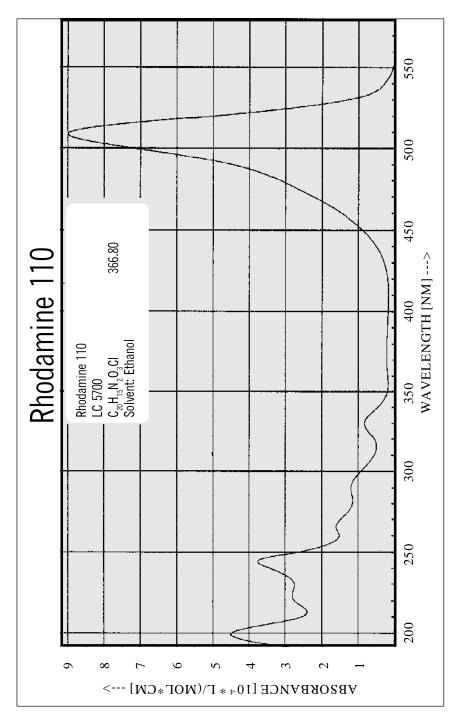
For research and development purposes only.

#### Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 570 nm. Alternative to Rhodamine 6G.

Pum	p	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Flashlamp CW, Ar <sup>+</sup> Nd:YAG	- all 532	567 571 566	- 552-608 549-592	- 36 44	0.06 0.45 0.31	Ethanol PC PC	1, 2 3 4

- 1. M. Shah et al., *Heteroatom Chem.* <u>1(5)</u>, 389(1990).
- 2. T. G. Pavlopoulos et al., Appl. Optics 29(27), 3885 (1990).
- 3. S. G. Guggenheimer et al., Appl. Optics 32(21), 3942 (1993).
- 4. M. P. O'Neil, *Optics Letters* 18(1), 37 (1993).
- 5. R. E. Hermes et al., *Appl. Phys. Letters* <u>63(7)</u>, 877 (1993).



## Rhodamine 110 (LC 5700)

#### Constitution

o-(6-Amino-3-imino-3H-xanthen-9-yl)-benzoic acid Rhodamine 560

C<sub>20</sub>H<sub>15</sub>N<sub>2</sub>O<sub>3</sub>CI · MW: 366.80

#### Characteristics

Lambdachrome® number: 5700

CAS registry number: 13558-31-1

Appearance: red, crystalline solid
Absorption maximum (in ethanol): 510 nm

Molar absorptivity: 8.99 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 535 nm

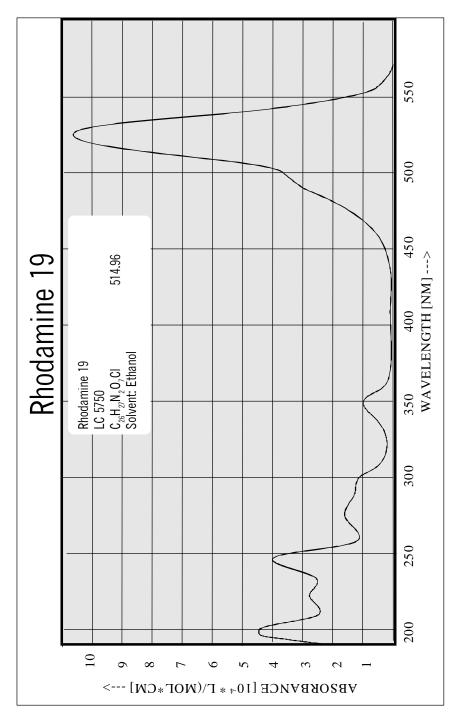
For research and development purposes only.

#### Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 570 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
v 0/ 5 ·	222	570	547 500	_	4.00		
XeCI-Excimer	<i>308</i>	<i>572</i>	547 - 592	5	1.22	Ethanol	1
Cu-vapor	510	<i>550</i>	<i>528 - 574</i>	9	0.09	Methanol	2
Flashlamp	-	-	551 <i>- 583</i>	-	0.07	Ethanol	3
CW, Ar⁺	all	560	530 - 580	-	0.75	Eg.	4, 5, 6

- 1. Lambda Physik.
- 2. M. Broyer et al., Appl. Phys. <u>B35</u>, 31 (1984).
- 3. W. Sibbett, J. R. Taylor, *IEEE J. Quantum Electron*. <u>QE-19(4)</u>, 558 (1983).
- 4. Coherent, CW Dye Laser Fact Sheets.
- 5. T. F. Johnston et al., *Appl. Optics* <u>21(13)</u>, 2307 (1982).
- 6. Lambda Physik, Wall Chart 1996.



## Rhodamine 19 (LC 5750)

#### Constitution

Benzoic Acid, 2-[6-(ethylamino)-3-(ethylimino)-2,7-dimethyl-3H-xanthen-9-yl], perchlorate Rhodamine 575

C<sub>26</sub>H<sub>27</sub>N<sub>2</sub>O<sub>7</sub>CI · MW: 514.96

#### Characteristics

Lambdachrome® number: 5750 CAS registry number: 62669-66-3

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 528 nm

Molar absorptivity: 10.9 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

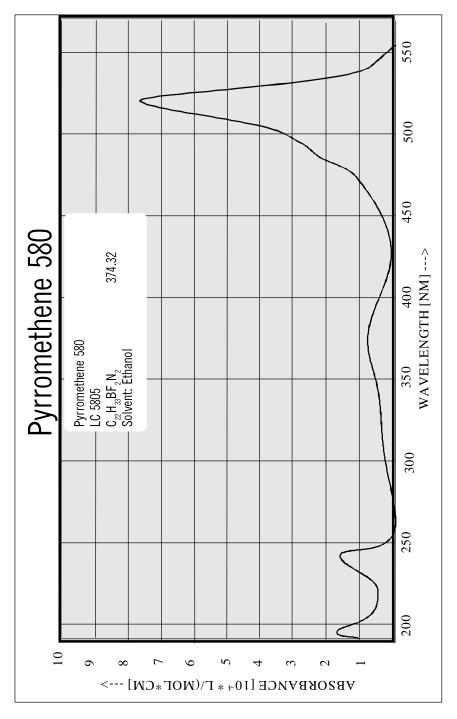
Fluorescence maximum: - For research and development purposes only.

#### Lasing Performance

Efficient laser dye for pulsed operation; tunable around 560 nm.

Pump	)	D	Dye Laser Characteristics				
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
Nd:YAG, 3rd Nd:YAG, 2nd	355 532	562 567	552 - 582 556 - 586	- 31	0.21 0.22	Ethanol Methanol	1 2

- 1. D. M. Guthals, J. W. Nibler, *Opt. Commun.* <u>29(3)</u>, 322 (1979).
- 2. Lambda Physik, Wall Chart 1996.



## Pyrromethene 580 (LC 5805)

#### Constitution

4,4-Difluoro-2,6-di-n-butyl-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene 2,6-Di-n-butyl-1,3,5,7,8-pentamethylpyrromethenedifluoroborate Complex

C<sub>22</sub>H<sub>33</sub>BF<sub>2</sub>N<sub>2</sub> · MW: 374.32

#### Characteristics

Lambdachrome® number: 5805 CAS registry number: N/A

Appearance: orange/red, crystalline solid

Absorption maximum (in ethanol): 519 nm

Molar absorptivity: 7.68 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 550 nm

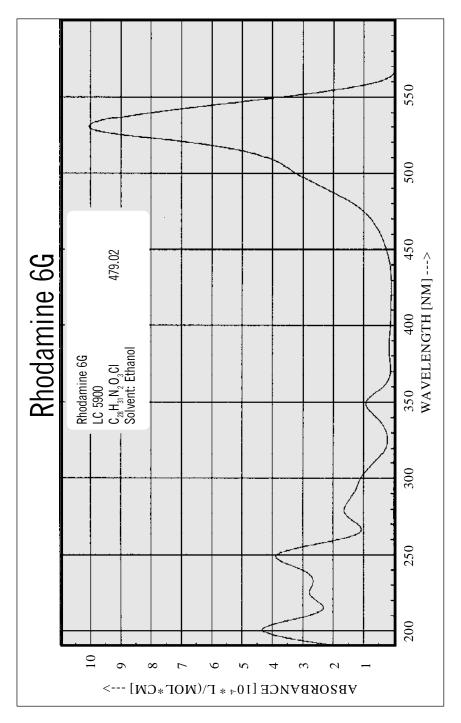
For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed operation; tunable around 580 nm.

Pump		Dye Laser Characteristics						
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.	
Flashlamp	-	580	-	-	0.08	Ethanol	1	

- 1. M. Shah et al., *Heteroatom Chem.* <u>1(5)</u>, 389(1990).
- 2. R. E. Hermes et al., Appl. Phys. Letters 63(7), 877 (1993).



## Rhodamine 6G (LC 5900)

#### Constitution

Benzoic Acid, 2-[6-(ethylamino)-3-(ethylimino)-2,7-dimethyl-3H-xanthen-9-yl]-ethyl ester, monohydrochloride Rhodamine 590

 $C_{28}H_{31}N_2O_3CI \cdot MW: 479.02$ 

#### Characteristics

Lambdachrome® number: 5900 CAS registry number: 989-38-8

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 530 nm

Molar absorptivity: 10.50 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 556 nm

For research and development purposes only.

## Lasing Performance

The laser dye "per se" Rhodamine 6G is by far the most frequently used and most widely investigated laser dye. Very efficient laser dye for pulsed and CW operation; tunable around 590 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	581	569 - 608	16	1.20	Methanol	1, 2, 3
Nitrogen	<i>337</i>	581	573 - 618	rel.	1.63	Methanol	3, 4, 5
Nd:YAG, 2nd	<i>532</i>	<i>566</i>	<i>555 - 585</i>	<i>32</i>	0.10	Methanol	1, 6, 7
Flashlamp	-	600	<i>555 - 620</i>	-	1.20	Ethanol	9, 10
CW, Ar+	all	<i>575</i>	560 - 625	-	0.75	Eg. 1,	11, 12, 13

## References

**S**ee page 164.

## References (RHODAMINE 6G)

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 403 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. A. Dienes, *Appl. Phys.* 7, 135 (1975).
- 6. D. M. Guthals, J. W. Nibler, Opt. Commun. 29(3), 322 (1979).
- 7. C. A. Moore, C. D. Decker, J. Appl. Phys. 49(1), 47 (1978).
- 8. M. Broyer et al., Appl. Phys. <u>B35</u>, 31 (1984).
- 9. P. R. Hammond, *Opt. Commun.* 29(3), 331 (1979).
- 10. J. Jethwa, F. P. Schäfer, *Appl. Phys.* <u>4</u>, 299 (1974).
- 11. Coherent, CW Dye Laser Fact Sheets.
- 12. H. J. Baving et al., *Appl. Phys.* <u>B29</u>, 19 (1982).
- 13. T. F. Johnston et al., Appl. Optics 21(13), 2307 (1982).

## References (RHODAMINE B)

- 1. Lambda Physik, Wall Chart 1996.
- 2. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 3. Lambda Physik, Data Sheet.
- 4. A. Dienes, Appl. Phys. 7, 135 (1975).
- 5. I. A. Stenhaouse, D. R. Williams, *Appl. Spectrosc.* 33(2), 175 (1979).
- 6. Q. H. F. Vrehen, Opt. Commun. 3(3), 144 (1971).
- 7. L. Masarnovskii et al., Sov. J. Quantum Electron. 9(7), 900 (1979).
- 8. R. S. Hargrove, T. Kan, IEEE J. Quantum Electron. QE-13, 28D (1977).
- 9. J. M. Drake et al., *Chem. Phys. Letters* <u>35(2)</u>, 181 (1975).
- 10. P. R. Hammond, *Opt. Commun.* <u>29(3)</u>, 331 (1979).
- 11. Coherent, CW Dye Laser Fact Sheets.

## References (SULFORHODAMINE B)

- 1. Lambda Physik, Wall Chart 1996.
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).
- 3. Lambda Physik, Data Sheet.
- 4. M. Broyer et al., Appl. Phys. <u>B35</u>, 31 (1984).
- 5. P. R. Hammond, Opt. Commun. 29(3), 331 (1979).
- 6. R. M. Schotland, *Appl. Optics* <u>19(1)</u>, 124 (1980).
- 7. J. M. Yarborough, *Appl. Phys. Letters* 24(12), 629 (1974).

### References (RHODAMINE 101)

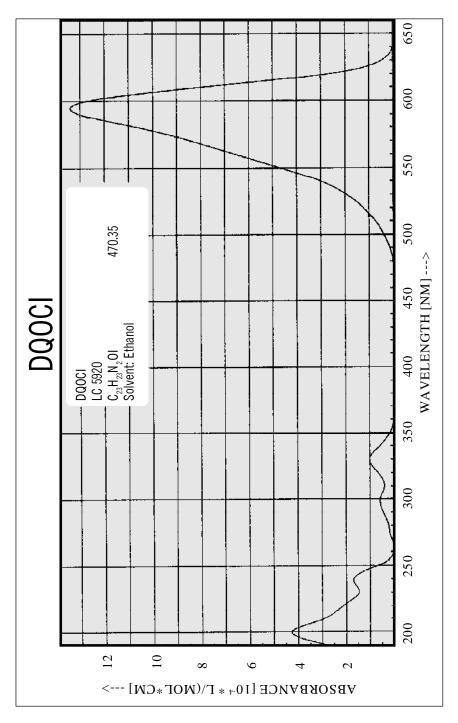
- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 403 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. Lambda Physik.
- 6. Bos, Appl. Optics 20(10), 1886 (1981).
- 7. M. Broyer et al., *Appl. Phys.* <u>B35</u>, 31 (1984).
- 8. T. J. Negran, A. M. Glass, *Appl. Optics* <u>17(17)</u>, 2812 (1978).
- 9. Coherent, CW Dye Laser Fact Sheets.

### References (DCM)

- 1. Lambda Physik, Wall Chart 1996.
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).
- 3. Lambda Physik, Data Sheet.
- 4. Lambda Physik.
- 5. M. Broyer et al., *Appl. Phys.* B35, 31 (1984).
- 6. P. R. Hammond, Opt. Commun. 29(3), 331 (1979).
- 7. G. P. Weber, *IEEE J. Quantum Electron.* QE-19(7), 1200 (1983).
- 8. Coherent. Data Sheet.
- 9. E. G. Marason, Opt. Commun. 37(1), 56 (1981).
- 10. T. F. Johnston et al., *Appl. Optics* <u>21(13)</u>, 2307 (1982).

## References (CRESYL VIOLET)

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- 2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6) 403 (1981).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 4. F. Castelli, *Appl. Phys. Letters* 26(1), 18 (1975).
- 5. A. Dienes, *Appl. Phys.* <u>7</u>, 135(1975).
- 6. I. A. Stenhouse, D. R. Williams, *Appl. Spectrosc.* <u>33(2)</u>, 175 (1979).
- 7. C. A. Moore, C. D. Decker, J. Appl. Phys. 49(1), 47 (1978).
- 8. W. Schmidt, W. Appt, N. Wittekindt, Z. Naturforsch. 27a, 37 (1972).
- 9. J. B. Marling et al., *Appl. Optics*. <u>13(10)</u>, 2317 (1974).
- 10. J. M. Yarborough, *Appl. Phys. Letters* 24(12), 629 (1974).



## **DQOCI (LC 5920)**

### Constitution

1,3'-Diethyl-4,2'-quinolyloxacarbocyanine lodide

 $C_{23}H_{23}N_2OI \cdot MW: 470.35$ 

$$H_{5}C_{2}-N_{4}$$

$$CH=CH-CH=0$$

$$C_{2}H_{5}$$

#### Characteristics

Lambdachrome® number: 5920

CAS registry number:

Appearance: violet, crystalline solid

Absorption maximum (in ethanol): 592 nm

Molar absorptivity:  $13.5 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

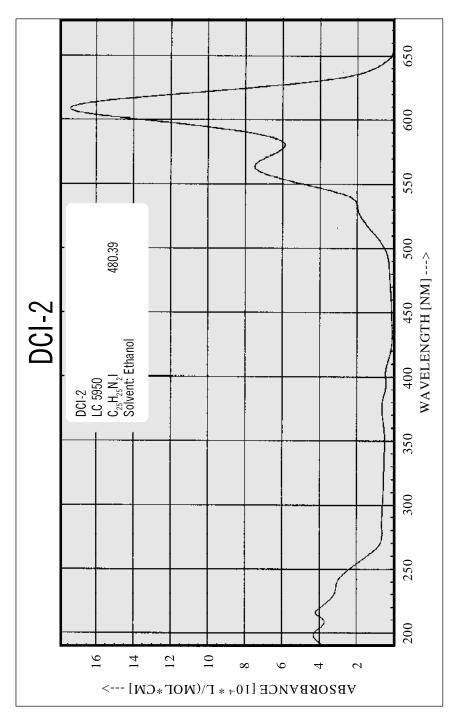
Fluorescence maximum: -

For research and development purposes only.

## Lasing Performance

Saturable absorber for flashlamp pumped Fluorol 7GA and Rhodamine 6G dye lasers <sup>1,,2)</sup>. Applicable around 590 nm.

- 1. E. Lill, S. Schneider, F. Dörr, *Opt. Commun.* <u>20(2)</u>, 223 (1977).
- 2. R. S. Adrain et al., Opt. Commun. 12(2), 140 (1974).



# DCI-2 (LC 5950)

### Constitution

1,1'-Diethyl-2,2'-carbocyanine lodide Pinacyanol lodide · Chinaldinblau

 $C_{25}H_{25}N_2I \cdot MW: 480.39$ 

$$CH = CH - CH$$

$$C_2H_5$$

$$C_2H_5$$

#### Characteristics

Lambdachrome® number: 5950 CAS registry number: -

Appearance: violet, crystalline solid

Absorption maximum (in ethanol): 606 nm

Molar absorptivity: 17.0 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

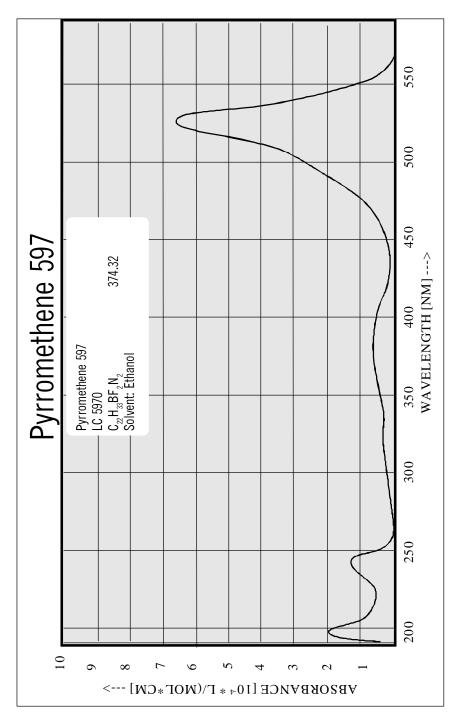
Fluorescence maximum: For research and development purposes only.

## Lasing Performance

Saturable absorber for flashlamp pumped Rhodamine 6G dye lasers <sup>1.)</sup>. Applicable around 606 nm.

## References

1. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* 13(1), 193 (1974).



# Pyrromethene 597 (LC 5970)

## Constitution

4,4-Difluoro-2,6-di-t-butyl-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene 2,6-Di-t-butyl-1,3,5,7,8-pentamethylpyrromethenedifluoroborate Complex

 $C_{22}H_{33}BF_{2}N_{2} \cdot MW: 374.32$ 

### Characteristics

Lambdachrome® number: 5970

CAS registry number: 137829-79-9

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 524 nm

Molar absorptivity: 6.76 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 557 nm

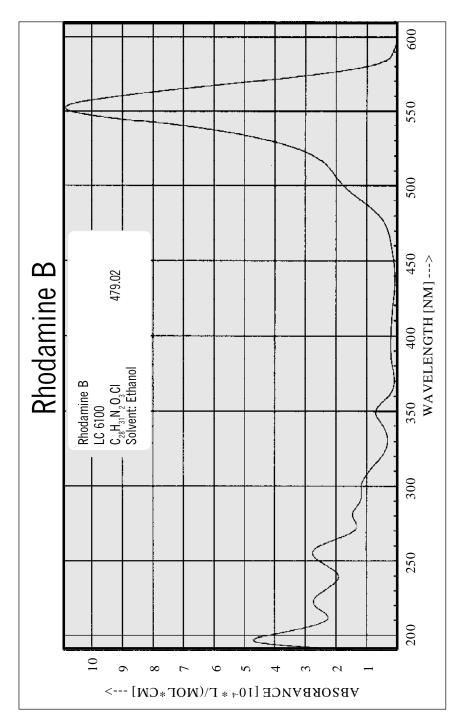
For research and development purposes only.

## Lasing Performance

Laser dye for pulsed operation; tunable around 590 nm.

Pum	р	Dye Laser Characteristics						
Source	Wavelength [nm]	Peak [nm]	Range [nm]		Conc. [g/l]	Solvent	Ref.	
Flashlamp	-	593	-	-	0.08	Ethanol	1, 2	

- 1. J. H. Boyer et al., *Appl. Optics* 30(27), 3788 (1991).
- 2. J. H. Boyer et al., *Heteroatom Chem.* 4(1), 39 (1993).



# Rhodamine B (LC 6100)

## Constitution

2-[6-(Diethylamino)-3-(diethylimino)-3H-xanthen-9-yl] benzoic acid Rhodamine 610

C<sub>28</sub>H<sub>31</sub>N<sub>2</sub>O<sub>3</sub>CI · MW: 479.02

#### Characteristics

Lambdachrome® number: 6100 CAS registry number: 81-88-9

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 552 nm

Molar absorptivity: 10.7 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 580 nm

For research and development purposes only.

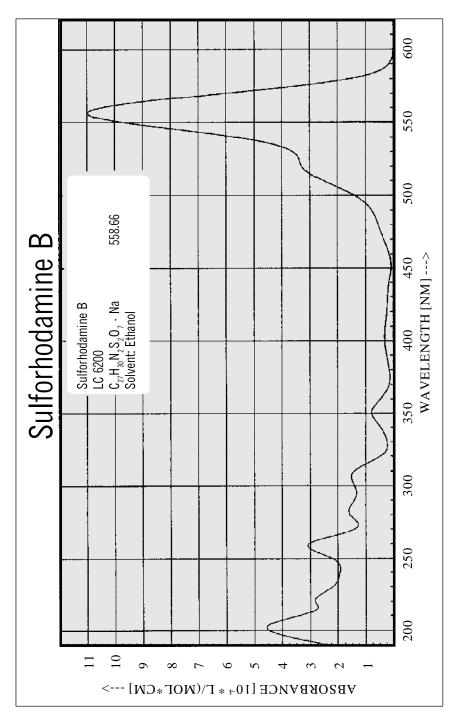
## Lasing Performance

Very efficient and frequently used laser dye for pulsed and CW operation; tunable around 610 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	600	<i>588 - 644</i>	12	0.91	Methanol	1, 2
Nitrogen	<i>337</i>	622	599 - 650	rel.	2.13	Methanol	2, 3, 4
Nd:YAG, 2nd	<i>532</i>	594	584 - 619	<i>2</i> 9	0.22	Methanol	1, 5, 6
Cu-vapo	510	591	<i>582 - 618</i>	21	0.62	Ethanol	7, 8
Flashlamp	-	618	590 - 640	-	0.05	Ethanol	9, 10
CW, Ar+	all	640	605 - 675	-	3.53	MeOH/Eg.	11

### References

see page 164.



# Sulforhodamine B (LC 6200)

## Constitution

Ethanaminium, N-[6-diethylamino)-9-(2,4-disulfophenyl)-3H-xanthen-3-ylidene]-N-ethylhydroxid, inner salt, sodium salt Kiton Red 620 · Kiton Red S

 $C_{27}H_{30}N_{2}S_{2}O_{7}$  · Na · MW: 558.66

## Characteristics

Lambdachrome® number: 6200 CAS registry number: 3520-42-1

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 556 nm

Molar absorptivity: 11.1 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 575 nm

For research and development purposes only.

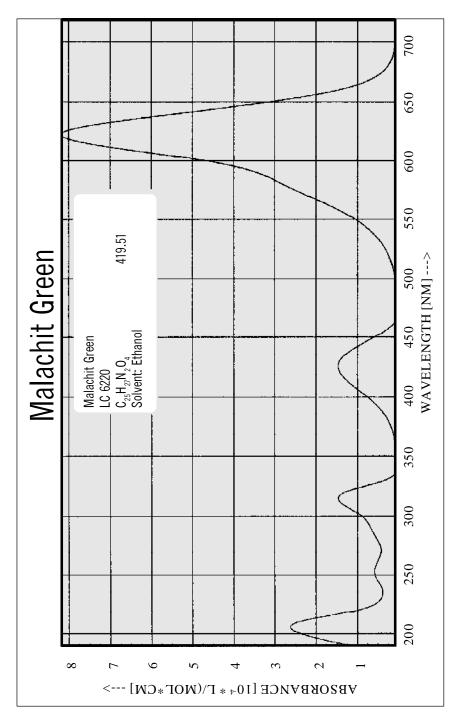
## Lasing Performance

Efficient and frequently used laser dye for pulsed and CW operation; performance similar to Rhodamin B; tunable around 620 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
VaCI Fyaimar	200	COE	E04 C42	10	0.00	Mothanal	1 0
XeCI-Excimer		605	<i>594 - 642</i>	12	0.90	Methanol	1, 2
Nitrogen	<i>337</i>	622	<i>600 - 646</i>	rel.	2.85	Methanol	3
Nd:YAG, 2nd	<i>532</i>	<i>588</i>	<i>579 - 600</i>	<i>2</i> 9	0.27	Methanol	1
Cu-vapo	510	<i>620</i>	598 - 645	14	1.74	Methanol	4
Flashlamp	-	<i>629</i>	600 - 650	-	3.91	Methanol	5, 6
CW, Ar+	all	<i>625</i>	<i>598 - 650</i>	-	2.50	MeOH/Eg.	1, 7

## References

See page 164.



# Malachit Green (LC 6220)

## Constitution

Malachit Grün Oxalat

C<sub>25</sub>H<sub>27</sub>N<sub>2</sub>O<sub>4</sub> · MW: 419.51

#### Characteristics

Lambdachrome® number: 6220

CAS registry number:

Appearance: green, crystalline solid

Absorption maximum (ethanol): 622 nm

Molar absorptivity: 8.07 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

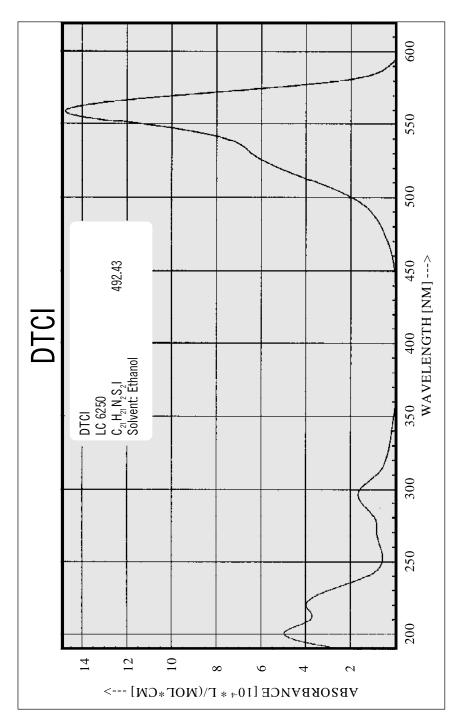
Fluorescence maximum (in bas. ethanol): 530 nm

For research and development purposes only.

## Lasing Performance

Additive for CW pumped, passivlely mode locked Rhodamine 6G dye lasers.

- 1. M. Young, *Appl. Optics* 18(19), 3212 (1979).
- 2. A. Watanabe et al., *IEEE J. Quantum Electron.* <u>QE-19(4)</u>, 533 (1983).



# DTCI (LC 6250)

## Constitution

3,3'-Diethylthiacarbocyanine Iodide

C<sub>21</sub>H<sub>21</sub>N<sub>2</sub>S<sub>2</sub>I · MW: 492.43

$$CH = CH - CH = \frac{1}{1}$$

$$C_2H_5$$

$$C_2H_5$$

#### Characteristics

Lambdachrome® number: 6250

CAS registry number:

Appearance: red, crystalline solid

Absorption maximum (in methanol): 557 nm

Molar absorptivity: 14.60 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

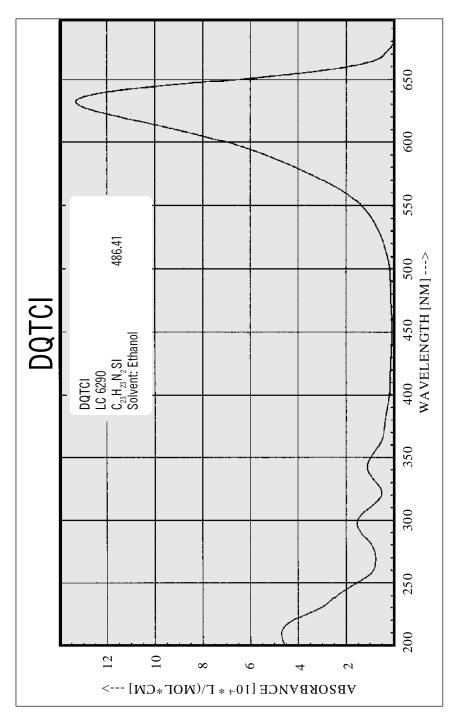
Fluorescence maximum: - For research and development purposes only.

# Lasing Performance

Laser dye for pulsed operation; tunable around 625 nm. Saturable absorber for flashlamp pumped Rhodamine 6G dye lasers; applicable around 560 nm<sup>2.</sup>).

Pum	р	D	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.	
Flashlamp	-	625	-	_	0.10	Glyzerine	1	

- 1. M. Maeda, Y. Miyazoe, Jap. J. Appl. Phys. 11(5), 692 (1972).
- 2. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* 13(1), 193 (1974).



# **DQTCI (LC 6290)**

## Constitution

1,3'-Diethyl-4,2'-quinolylthiacarbocyanine lodide

C23H23N2SI · MW: 486.41

$$H_5C_2 - N \longrightarrow CH = CH - CH = \begin{cases} S \\ I \\ C_2H_5 \end{cases}$$

#### Characteristics

Lambdachrome® number: 6290

CAS registry number:

Appearance: green, crystalline solid

Absorption maximum (in methanol): 629 nm

Molar absorptivity: 13.1 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum:

For research and development purposes only.

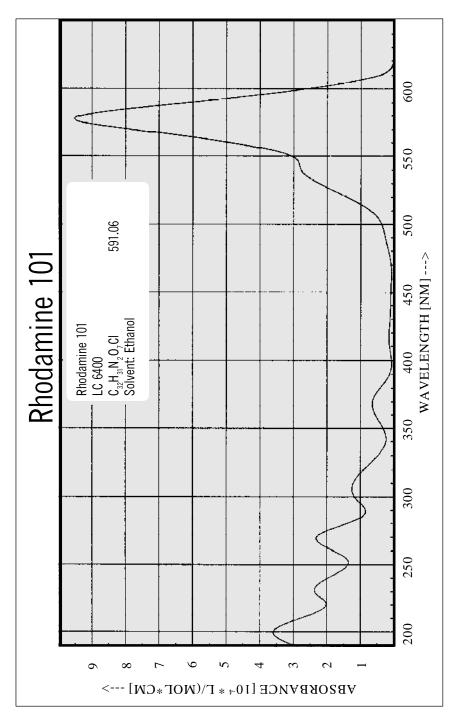
# Lasing Performance

Saturable absorber for flashlamp pumped Rhodamine B dye lasers; applicable around 630 nm <sup>1.,2.)</sup>.

# References

1. E. G. Arthurs et al., Appl. Phys. Lett. 20(3), 125(1972).

2. E. Lill, S. Schneider, F. Dörr, Opt. Commun. 22(1), 107(1977).



# Rhodamine 101 (LC 6400)

## Constitution

8-(2-Carboxyphenyl)-2,3,5,6,11,12,14,15-octahydro-1*H*,4*H*,10*H*,13*H*-diquinolizino[9,9a,1-*bc*:9',9a',1-*hi*]xanthylium Perchlorate Rhodamine 640

C<sub>22</sub>H<sub>21</sub>N<sub>2</sub>O<sub>7</sub>CI · MW: 591.06

## Characteristics

Lambdachrome® number: 6400 CAS registry number: 64339-18-0

Appearance: red, crystalline solid

Absorption maximum (in acidic ethanol): 576 nm

Molar absorptivity: 9.50 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum: - For research and development purposes only.

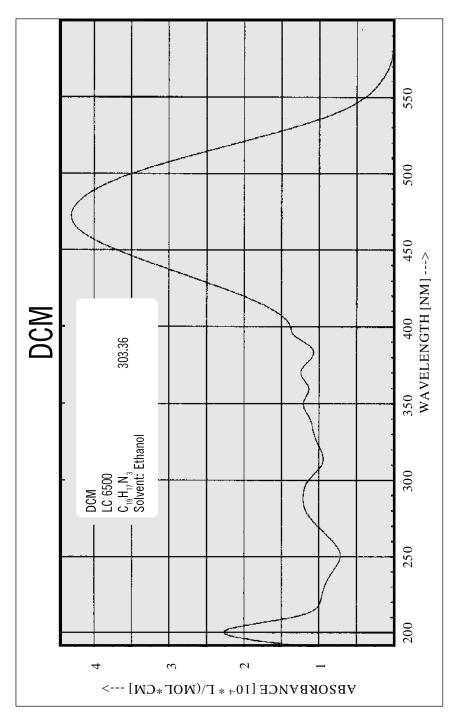
## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 640 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	<i>308</i>	623	614 - 672	12	0.75	Methanol	<i>1, 2, 3</i>
Nitrogen	<i>337</i>	648	<i>623 - 676</i>	rel.	2.36	Methanol	2, 4
Nd:YAG, 2nd	<i>532</i>	621	611 - 662	<i>26</i>	0.50	Methanol	1, 5, 6
Cu-vapor	510	630	607 - 659	14	-	Methanol	7
Flashlamp	-	-	-	-	-	Ethanol	8
CW, Ar+	VIS	<i>625</i>	610 - 695	-	2.5	MeOH/Eg.	1, 9

## References

See page 165.



# DCM (LC 6500)

## Constitution

4-Dicyanmethylene-2-methyl-6-(p-dimethylaminostyryl)-4H-pyran

 $C_{19}H_{17}N_3 \cdot MW: 303.36$ 

### Characteristics

Lambdachrome® number: 6500 CAS registry number: 51325-91-8

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 472 nm

Molar absorptivity: 4.25 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 644 nm

For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 650 nm. DCM Special gives higher efficiency due to better solubility.

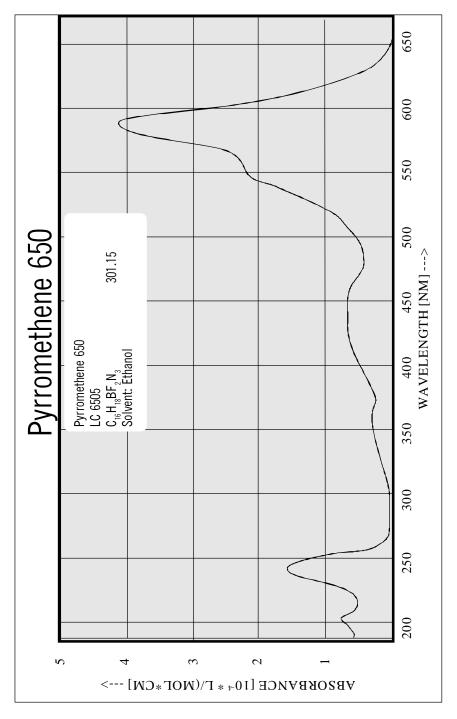
Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	658	632 - 690	12	0.71	DMS0	1, 2
Nitrogen	337	659	626 - 703	rel.	0.71	DMS0	3
Nd:YAG, 2nd	532	<i>639</i>	615 - 666	27	0.50	PC	1.4
Cu-vapor	510	644	598 - 677	14	0.61	Methanol	5
Flashlamp	-	655	610 - 710	-	0.76	<b>DMSO</b>	6, 7
CW, Ar+	VIS	660	-	-	0.45	Bz./Eg.	1, 8, 9, 10

### References

See page 165.

### P.S.

LC 6501 is a 1:1 mixture of LC 6500 and LC 6200 giving high absorption at the green line of an Ar-Ion laser. Its application should be restricted to this pump laser only.



# Pyrromethene 650 (LC 6505)

# Constitution

4,4-Difluoro-8-cyano-1,2,3,5,6,7-hexamethyl-4-bora-3a,4a-diaza-s-indacene 8-Cyano-1,2,3,5,6,7-hexamethylpyrromethenedifluoroborate Complex

 $C_{16}H_{18}BF_{2}N_{3} \cdot MW: 301.15$ 

### Characteristics

Lambdachrome® number: 6505 CAS registry number: -

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 590 nm

Molar absorptivity:  $4.05 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 612 nm

For research and development purposes only.

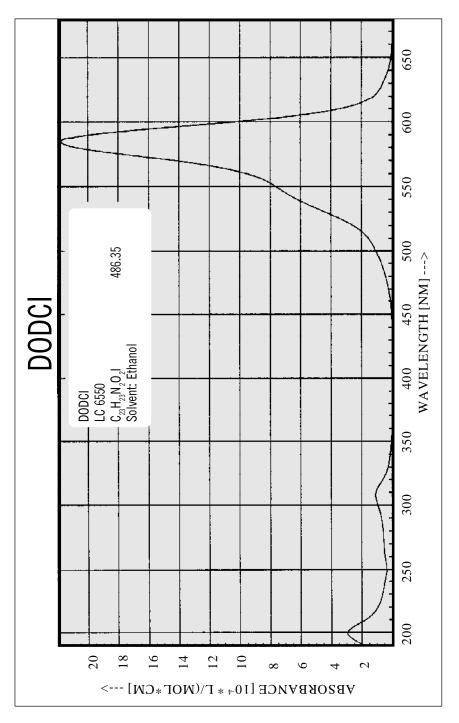
# Lasing Performance

Laser dye for pulsed operation; tunable around 630 nm.

Pun	пр	Dye Laser Characteristics						
Source	Wavelength [nm]	Peak [nm]	Range [nm]		Conc. [g/l]	Solvent	Ref.	
Nd:YAG	532	630	-	31	0.03	Ethanol	1	

## References

1. T. H. Allik et al., *SPIE Proceedings* 2115, 240 (1994).



# **DODCI (LC 6550)**

### Constitution

3,3'-Diethyloxadicarbocyanine lodide NK 1533

C<sub>22</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>I · MW: 486.35

$$C_{2}H_{5} = CH_{2} - CH = CH_{2}$$

$$C_{2}H_{5}$$

$$C_{2}H_{5}$$

#### Characteristics

Lambdachrome® number: 6550 CAS registry number: -

Appearance: blue, crystalline solid

Absorption maximum (in ethanol): 582 nm

Molar absorptivity: 22.3 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum:

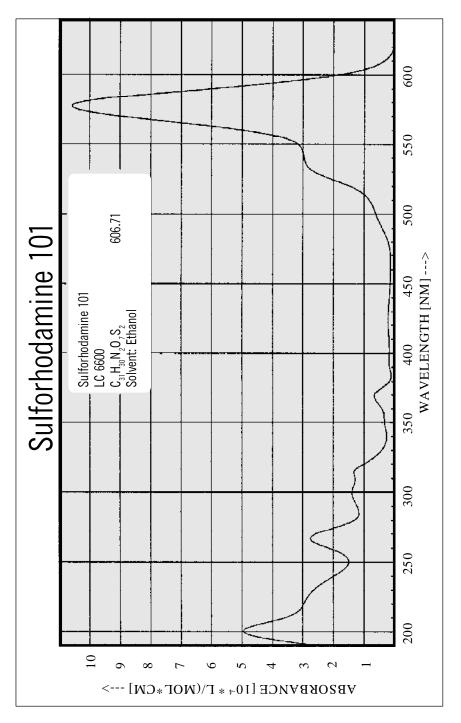
For research and development purposes only.

# Lasing Performance

Laser dye for pulsed operation; tunable around 660 nm. Most frequently used saturable absorber for flashlamp and CW pumped Rhodamine 6G und Rhodamin B dye lasers 3., 4., 5.). Applicable around 580 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range		Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
Nd:YAG, 2nd	<i>532</i>	-	-	-	0.49	Ethanol	1
Flashlamp	-	662	-	-	0.10	<i>DMSO</i>	2

- 1. C. Rulliere, *Chem. Phys. Letters* <u>43(2)</u>, 303 (1976).
- 2. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* <u>11(5)</u>, 692 (1972).
- 3. M. Young, Appl. Optics. 18(19), 3212 (1979).
- 4. A. Watanabe et al., IEEE J. Quantum Electron. QE-19(4), 533 (1983).
- 5. E. G. Arthurs et al., *Appl. Phys. Letters* 20(3), 125 (1972).



# Sulforhodamine 101 (LC 6600)

## Constitution

8-(2, 4-Disulfophenyl)-2,3,5,6,11,12,14,15-octahydro-1*H*,4*H*,10*H*,13*H*-diquinolizino[9,9a,1-*bc*:9',9a',1-*hi*]xanthene Sulforhodamine 640

 $C_{31}H_{30}N_2O_7S_2 \cdot MW: 606.71$ 

#### Characteristics

Lambdachrome® number: 6600

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 578 nm

Molar absorptivity: 10.6 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

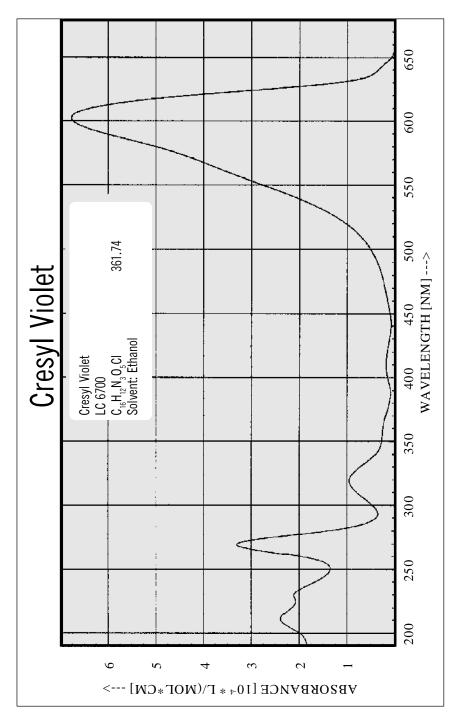
For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 650 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	652	616 - 667	15	0.78	Methanol	2, 3
Nd:YAG, 2nd	<i>532</i>	<i>628</i>	619 - 673	15	0.3	Methanol	1, 4
CW, Ar+	VIS	<i>625</i>	<i>598 - 650</i>	-	2.5	Eg.	5, 6

- 1. Lambda Physik, Wall Chart 1996.
- 2. V. S. Antonov, K. L. Hohla, Appl. Phys. <u>B32</u>, 9 (1983).
- 3. Lambda Physik.
- 4. K. Kato, IEEE J. Quantum Electron. QE-13(7), 544 (1977).
- 5. Lambda Physik, Wall Chart 6/90.
- 6. M. Yamashita et al., Opt. Commun. 26(3), 343 (1978).



# Cresyl Violet (LC 6700)

## Constitution

5,9-Diaminobenzo[a]phenoxazonium Perchlorate Cresyl Violet 670

C<sub>16</sub>H<sub>12</sub>N<sub>3</sub>O<sub>5</sub>CI · MW: 361.74

## Characteristics

Lambdachrome® number: 6700 CAS registry number: 41830-80-2

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 601 nm

Molar absorptivity:  $6.74 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): 632 nm

For research and development purposes only.

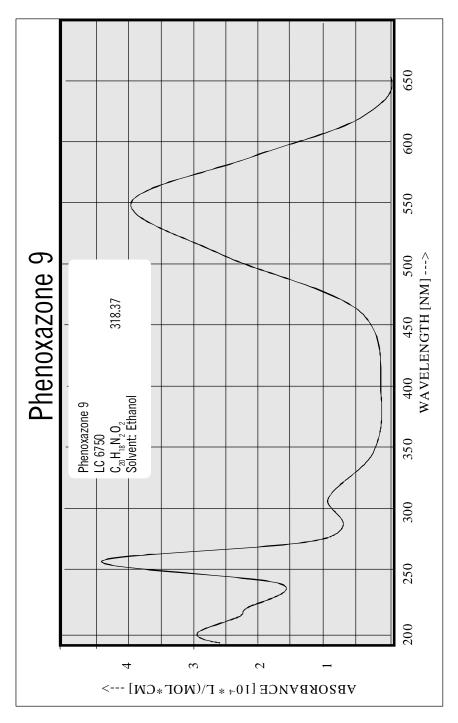
# Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 650 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
VaCI Fraiman	200	CEC	CA2 COO	г	0.50	[the area]	1 0 0
XeCI-Excimer		<i>656</i>	643 - 688	5	0.50	Ethanol	1 <i>, 2, 3</i>
Nitrogen	<i>337</i>	<i>650</i>	<i>630 - 680</i>	rel.	0.36	Ethanol	4, 5
Nd:YAG, 2nd	<i>532</i>	<i>630</i>	614 - 654	36	0.04	Methanol	6, 7
Flashlamp	-	<i>650</i>	630 - 680	-	0.01	Methanol	8, 9
CW, Ar+	VIS	670	650 - 695	4	1.09	Eg.	10

## References

See page 165.



# Phenoxazone 9 (LC 6750)

### Constitution

9-Diethylamino-5H-benzo(a)phenoxazin-5-one

 $C_{20}H_{18}N_2O_2 \cdot MW: 318.37$ 

### Characteristics

Lambdachrome® number: 6750 CAS registry number: 7385-67-3

Appearance: green, crystalline solid

Absorption maximum (ethanol): 550 nm

Molar absorptivity: 2.83 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in bas. ethanol): 650nm

For research and development purposes only.

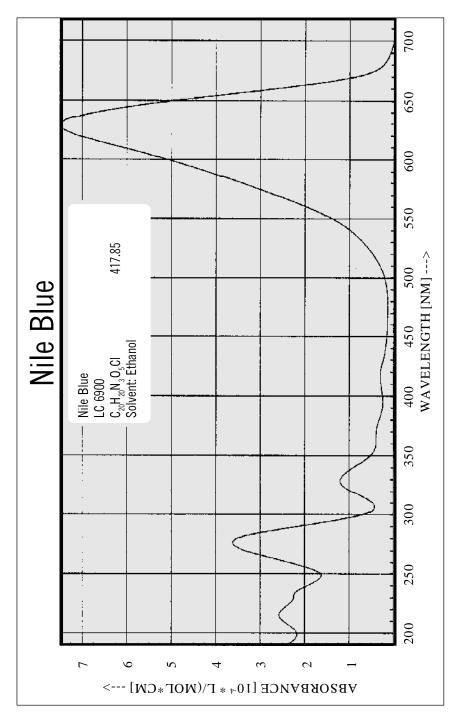
# Lasing Performance

Laser dye for pulsed operation; tunable around 620 nm.

Pum	р	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
Nitrogen Flashlamp	<i>337</i> -	620 -	560 - 700 -	rel. -	0.32 0.03	Ethanol various	1 1

## References

1. D. Basting, D. Ouw, F. P. Schäfer, *Opt. Commun.* 18(3), 260 (1976).



# Nile Blue (LC 6900)

## Constitution

5-Amino-9-diethyliminobenzo[a]phenoxazonium Perchlorate

C<sub>20</sub>H<sub>20</sub>N<sub>3</sub>O<sub>5</sub>CI · MW: 417.85

#### Characteristics

Lambdachrome® number: 6900 CAS registry number: 53340-16-2

Appearance: green, crystalline solid

Absorption maximum (ethanol): 633 nm
Molar absorptivity: 7.75 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in bas. ethanol): 672 nm

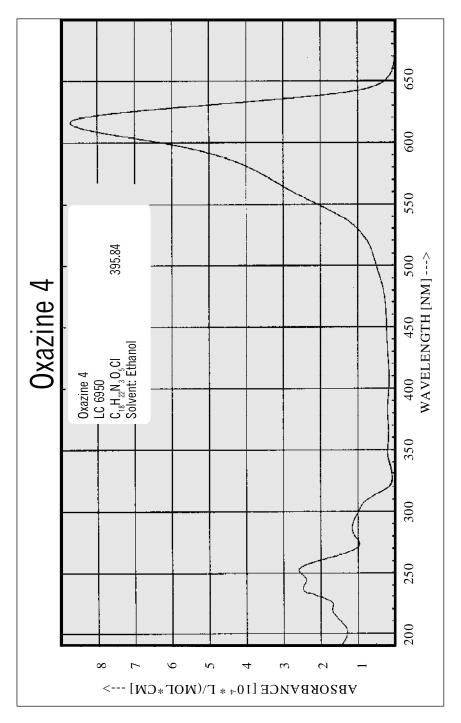
For research and development purposes only.

# Lasing Performance

Laser dye for pulsed operation; tunable around 690 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	703	688 - 747	5	0.47	Methanol	1. 2
Nitrogen	337	695	683 - 751	rel.	0.47	Methanol	2
Nd:YAG, 2nd	532	683	•	18	0.08	Methanol	3
Cu-vapor	510	695	<i>682 - 730</i>	4	0.84	Methanol	4
Flashlamp	-	710	690 - 750	poor	0.31	Methanol	5
CW, Kr+	red	730	690 - 780	7	1.2	Eg.	6

- 1. Lambda Physik.
- 2. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 3. K. Kato, Opt. Commun. 19(1), 19 (1976).
- 4. M. Broyer et al., Appl. Phys. <u>B35</u>, 31 (1984).
- 5. J. B. Marling et al., *Appl. Optics* 13(10), 2317 (1974).
- 6. J. M. Yarborough, *Appl. Phys. Letters* 24(12), 629 (1974).



# Oxazine 4 (LC 6950)

## Constitution

3-Ethylamino-7-ethylimino-2,8-dimethylphenoxazin-5-ium Perchlorate LD 690

 $C_{18}H_{22}N_3O_5CI \cdot MW: 395.84$ 

6950

## Characteristics

Lambdachrome® number:

CAS registry number: -

Appearance: green, crystalline solid

Absorption maximum (ethanol): 615 nm

Molar absorptivity: 10.9 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

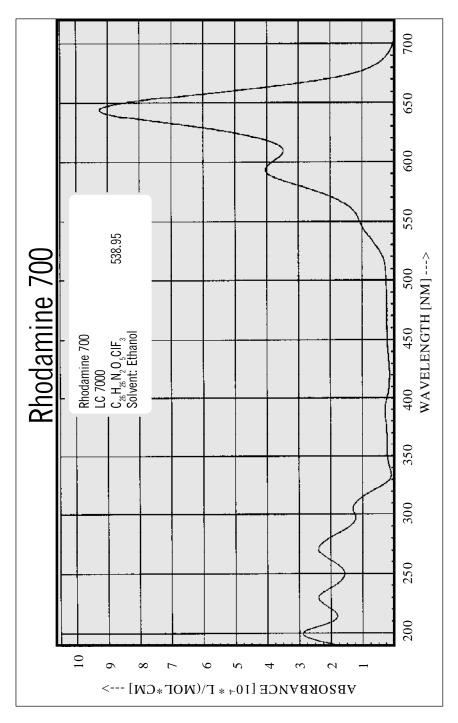
Fluorescence maximum:
For research and development purposes only.

# Lasing Performance

Laser dye for pulsed operation; tunable around 690 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nd:YAG, 2nd CW, Ar+	308 532 all	703 660 -	665 - 718 - -	5 - -	0.53 - + DCM	DMSO - -	1 2 3

- 1. Lambda Physik.
- 2. R. J. Hall et al., Opt. Letters 4(3), 87 (1979).
- 3. J. Heber, A. Szabo, IEEE J. Quantum Electron. QE-20(1), 9 (1984).



# Rhodamine 700 (LC 7000)

## Constitution

8-(Trifluoromethyl)-2,3,5,6,11,12,14,15-octahydro-1H,4H,10H,13H-diquinolizino[9,9a,1-bc:9',9a',1-hi]xanthylium Perchlorate LD 700

 $C_{26}H_{26}N_2O_5CIF_3 \cdot MW: 538.95$ 

### Characteristics

Lambdachrome® number: 7000

Appearance: brown, crystalline solid

Absorption maximum (in ethanol): 643 nm

Molar absorptivity: 9.25 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

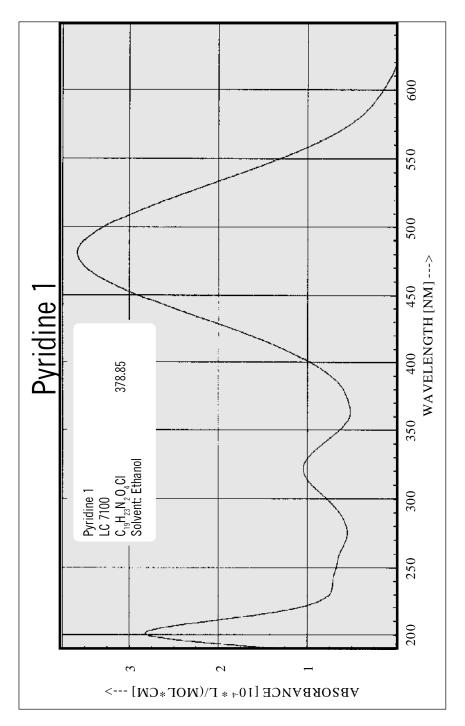
For research and development purposes only.

## Lasing Performance

Very efficient laser dye for pulsed and CW operation; tunable around 750 nm

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Flashlamp CW, Kr+	308 - red	723 - 740	701 - 768 705 - 798 690 - 785	11 - -	0.85 0.11 1.0	Methanol Ethanol Eg.	1, 2 3 1. 4. 5. 6

- 1. Lambda Physik, Wall Chart 1996.
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).
- 3. W. Sibbett, J. R. Taylor, *IEEE J. Quantum Electron.* <u>QE-20(2)</u>, 108 (1984).
- 4. T. F. Johnston, R. H. Brady, W. Proffitt, Appl. Optics. 21(13), 2307 (1982).
- 5. G. D. Aumiller, Appl. Optics 23(5), 651 (1984).
- 6. E. G. Marason, Opt. Commun. 40(3), 212 (1982).



# Pyridine 1 (LC 7100)

### Constitution

1-Ethyl-2-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-pyridinium Perchlorat LDS 698

 $C_{19}H_{23}N_2O_4CI \cdot MW: 378.85$ 

#### Characteristics

Lambdachrome® number: 7100

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 480 nm

Molar absorptivity: 3.80 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

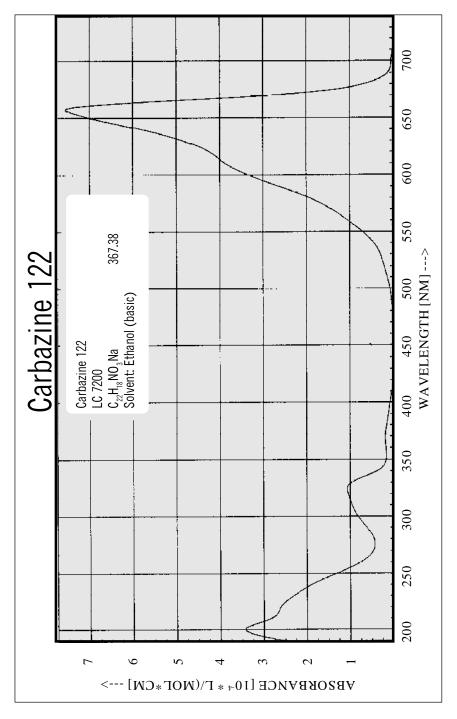
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 710 nm

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Nd:YAG, 2nd Cu-vapor CW, Ar+	308 337 532 510 VIS	710 703 697 684 710	670 - 760 675 - 750 667 - 736 661 - 724 670 - 780	10 rel. 32 6	0.84 0.88 0.36 1.17	DMSO DMSO PC Methanol Pc./Eg.	1, 2 3 1, 4 5 6, 7

- 1. Lambda Physik, Wall Chart 1996.
- 2. V. S. Antonov, K. L. Hohla, Appl. Phys. <u>B32</u>, 9 (1983).
- 3. Lambda Physik, Data Sheet.
- 4. Lambda Physik.
- 5. M. Broyer et al., Appl. Phys. <u>B35</u>, 31 (1984).
- 6. J. Hoffnagle et al., Opt. Commun. 42(4), 267 (1982).
- 7. Ph. Bado et al., Opt. Commun. 46(3,4), 241 (1983).



# Carbazine 122 (LC 7200)

## Constitution

Carbazine 720

C<sub>22</sub>H<sub>18</sub>NO<sub>3</sub>Na · MW: 367.38

## Characteristics

Lambdachrome® number: 7200 CAS registry number: -

Appearance: red, crystalline solid

Absorption maximum (in basic ethanol): 655 nm

Molar absorptivity:  $7.54 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

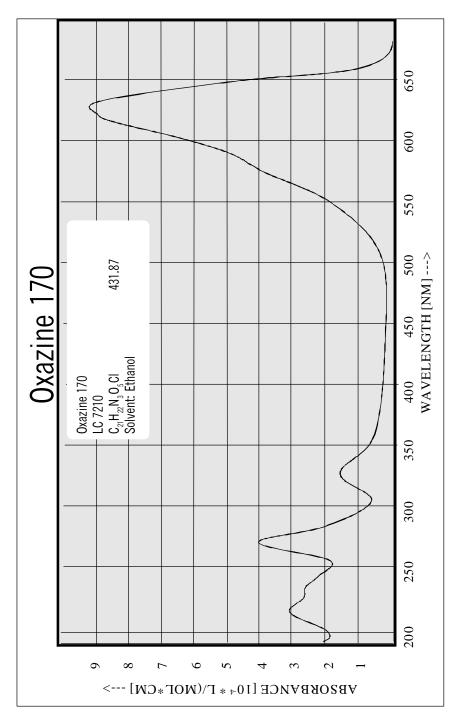
Fluorescence maximum: - For research and development purposes only.

# Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 640 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XNd:YAG, 2n	d 532	720	-	30	1.10	Water	1. 2. 3
Flashlamp	-	700	680 - 740	-	0.07	Methanol	4
CW, Kr+	red	<i>750</i>	690 - 820	-	0.72	Eg.	5

- 1. F. Bos, *Appl. Optics* 20(10), 1886 (1981).
- 2. K. Kato, Opt. Commun. 19(1), 18 (1976).
- 3. K. Kato, Opt. Commun. 18(4), 447 (1976).
- 4. J. B. Marling et al., *Appl. Optics* <u>13(10)</u>, 2317 (1974).
- 5. P. E. Jessop, A. Szabo, *IEEE J. Quantum Electr.* <u>QE-16(8)</u>, 812 (1980).



## Oxazine 170 (LC 7210)

### Constitution

9-Ethylamino-5-ethylimino-10-methyl-5H-benzo(a)phenoxazonium Perchlorate Oxazine 720

C<sub>21</sub>H<sub>22</sub>N<sub>3</sub>O<sub>5</sub>CI · MW: 431.87

#### Characteristics

Lambdachrome® number: 7210 CAS registry number: 62669-60-7

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 627 nm
Molar absorptivity: 9.20 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 650 nm

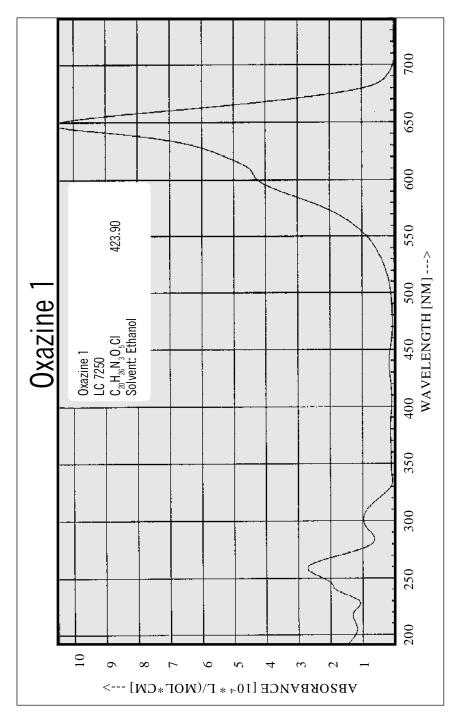
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 670 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	<i>308</i>	708	660 - 728	4	1.14	Methanol	1, 2, 3
Nitrogen	<i>337</i>	<i>705</i>	<i>672 - 727</i>	rel.	0.79	Methanol	4
Nd:YAG, 2nd	<i>532</i>	<i>672</i>	-	20	0.08	Methanol	5, 6
Cu-vapor	510	<i>675</i>	660 - 712	12	0.13	Methanol	7
CW, Kr+	red	<i>730</i>	670 - 740	-	0.86	Eg.	8

- 1. Lambda Physik, Wall Chart 6/83.
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).
- 3. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. K. Kato, *Opt. Commun.* 19(1), 18 (1976).
- 6. C. A. Moore, C. D. Decker, *J. Appl. Phys.* 49(1), 47 (1978).
- 7. M. Broyer et al., Appl. Phys. <u>B35</u>, 31 (1984).
- 8. P. E. Jessop, A. Szabo, *IEEE J. Quantum Electr.* QE-16(8), 812 (1980).



# Oxazine 1 (LC 7250)

#### Constitution

3-Diethylamino-7-diethyliminophenoxazonium Perchlorate Oxazine 725

 $C_{20}H_{26}N_3O_5CI \cdot MW: 423.90$ 

#### Characteristics

Lambdachrome® number: 7250 CAS registry number: 24796-94-9

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 646 nm

Molar absorptivity: 13.0 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 670 nm

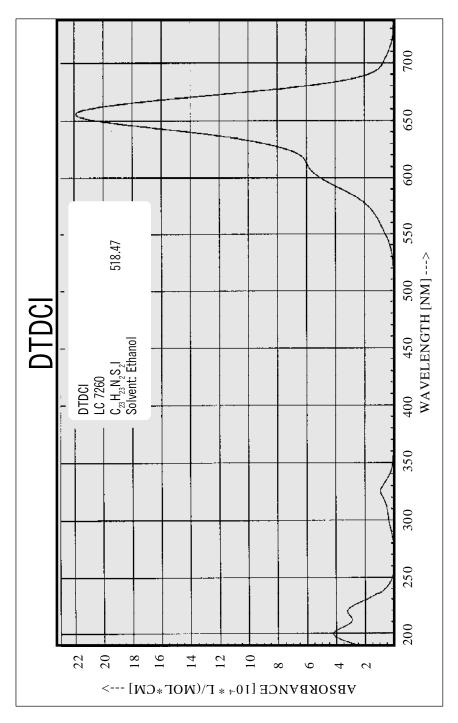
For research and development purposes only.

### Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 670 nm.

Pump	)	D	ye Laser Ch	aracteri	stics		
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. 「%1	Conc. [q/l]	Solvent	Ref.
	נווווון	Lilli	נווווו	[/0]	[9/1]		
XeCI-Excimer	308	734	692 - 768	6	0.85	Ethanol	1, 2, 3
Nitrogen	<i>337</i>	<i>730</i>	692 - 751	-	1.96	Ethanol	4
Nd:YAG, 2nd	<i>532</i>	695	-	18	0.07	Methanol	5, 6
Flashlamp	-	<i>720</i>	700 - 740	-	-	Methanol	7
CW, Kr+	red	<i>720</i>	695 - 800	-	1.20	Eg.	8

- 1. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 403 (1981).
- 2. V. S. Antonov, K. L. Hohla, Appl. Phys. <u>B32</u>, 9 (1983).
- 3. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 4. B. M. Pierce, R. R. Birge, *IEEE J. Quantum Electr.* <u>QE-18(7)</u>, 1164 (1982).
- 5. F. Bos, Appl. Optics 20(10), 1886 (1981).
- 6. C. A. Moore, C. D. Decker, J. Appl. Phys. 49(1), 47 (1978).
- 7. J. B. Marling et al., *Appl. Optics* <u>13(10)</u>, 2317 (1974).
- 8. Coherent, CW Dye Laser Fact Sheets.



# **DTDCI (LC 7260)**

#### Constitution

3-Diethylthiadicarbocyanine lodide NK 136

C<sub>23</sub>H<sub>23</sub>N<sub>2</sub>S<sub>2</sub>I · MW: 518.47

$$CH = CH_{12} - CH = \frac{S}{C_2H_5}$$

### Characteristics

Lambdachrome® number: 7260 CAS registry number: -

Appearance: blue, crystalline solid

Absorption maximum (in ethanol): 653 nm

Molar absorptivity: 22.3 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

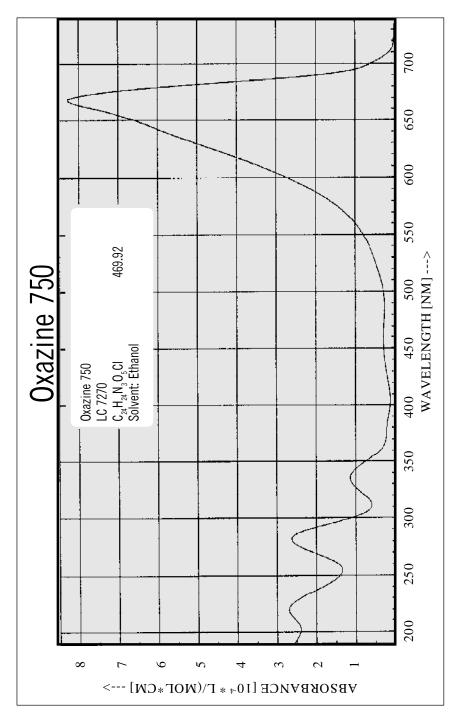
Fluorescence maximum: - For research and development purposes only.

## Lasing Performance

Laser dye for pulsed operation; tunable around 760 nm. Saturable absorber for flashlamp pumped Rhodamine 101 dye lasers; applicable around 650 nm <sup>1.)</sup>.

Pum	р	Dy	ve Laser Ch	aracteri	stics		
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Nitrogen Flashlamp	<i>337</i> -	695 760	-	-	0.13 0.10	Acetone DMSO	2 3

- 1. T. J. Negran, A. M. Glass, *Appl. Optics* <u>17(17)</u>, 2812 (1978).
- 2. Chinlon Lin, IEEE J. Quantum Electr. QE-11, 61 (1975).
- 3. A. Hirth, K. Vollrath, J. Faure, D. Lougnot, *Opt. Commun.* 7(4), 339 (1973).



# Oxazine 750 (LC 7270)

## Constitution

C24H24N2O5CI · MW: 469.92

#### Characteristics

Lambdachrome® number: 7270 CAS registry number: -

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 667 nm

Molar absorptivity: 8.25 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

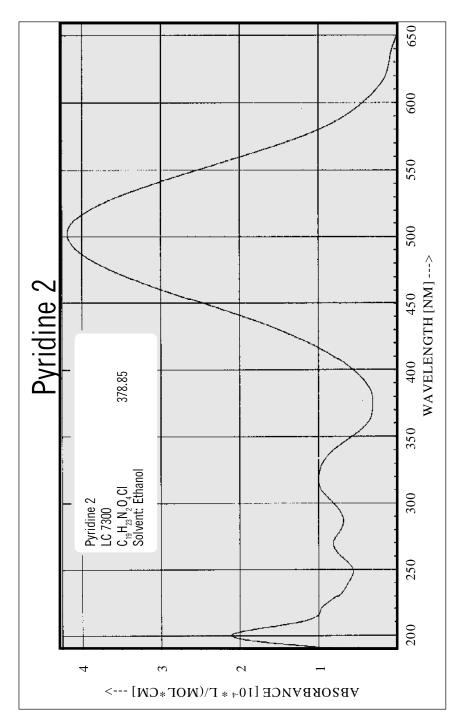
Fluorescence maximum: - For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 750 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen CW, Kr+	308 337 red	777 724 810	735 - 796 708 - 780 790 - 900	6 rel. -	1.25 0.50 0.62	DMSO Ethanol PC./Eg.	1, 2, 3 3, 4 5

- 1. Lambda Physik, Wall Chart 6/83.
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).
- 3. F. Bos, *Appl. Optics* 20(20), 3553 (1983).
- 4. B. M. Pierce, R. R. Birge, IEEE J. Quantum Electr. QE18(7), 1164 (1982).
- 5. G. D. Aumiller, *Opt. Commun.* 41(2), 115 (1982).



# Pyridine 2 (LC 7300)

### Constitution

1-Ethyl-4-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-pyridinium Perchlorat LDS 722

 $C_{19}H_{23}N_2O_4CI \cdot MW: 378.85$ 

$$H_5 C_2 - N_{+} \longrightarrow (CH = CH)_2 - (CH_3)_2$$

### Characteristics

Lambdachrome® number: 7300

Appearance: red, crystalline solid

Absorption maximum (in ethanol): 500 nm

Molar absorptivity: 4.22 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

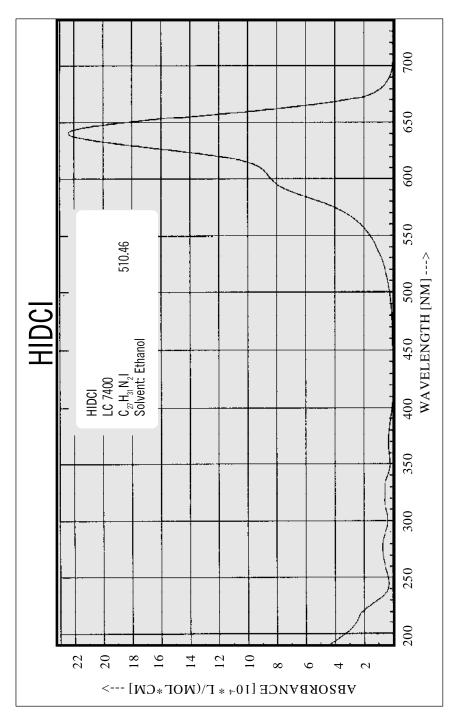
For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 740 nm

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer	308	740	695 - 790	11	0.72	<i>DMSO</i>	1
Nitrogen	<i>337</i>	<i>743</i>	710 - 790	rel.	0.85	<i>DMSO</i>	2
Nd:YAG	<i>532</i>	<i>750</i>	<i>725 - 776</i>	21	0.22	PC	3
Cu-vapor	510	722	687 - 755	4	1.00	Methanol	4
CW, Ar+	VIS	720	685 - 820	-	0.75	Pc./Eg.	1

- 1. Lambda Physik, Wall Chart 1996.
- 2. Lambda Physik, Data Sheet.
- 3. Lambda Physik.
- 3. M. Broyer et al., *Appl. Phys.* <u>B35</u>, 31 (1984).



# HIDCI (LC 7400)

#### Constitution

1,1',3,3,3',3'-Hexamethylindodicarbocyanine lodide Hexacyanine 2

C<sub>27</sub>H<sub>31</sub>N<sub>2</sub>I · MW: 510.46

#### Characteristics

Lambdachrome® number: 7400

CAS registry number: 36536-22-8

Appearance: blue, crystalline solid

Absorption maximum (in ethanol): 639 nm

Molar absorptivity: 22.5 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

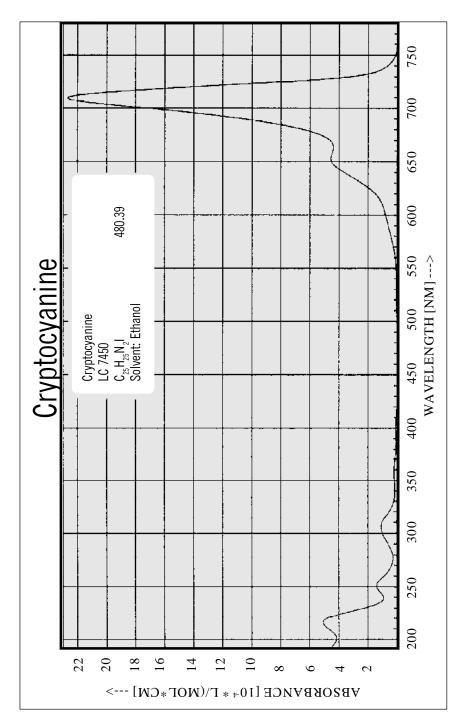
Fluorescence maximum:
For research and development purposes only.

### Lasing Performance

Laser dye for pulsed operation; tunable around 740 nm. Saturable absorber for flashlamp pumped Rhodamine 6G dye lasers; applicable around 630 nm<sup>1.3</sup>.

Pum	р	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Flashlamp	-	740	-	-	0.11	DMS0	2

- 1. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* 13(1), 193 (1974).
- 2. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* <u>11(5)</u>, 692 (1972).



# Cryptocyanine (LC 7450)

### Constitution

1,1'-Diethyl-4,4'-carbocyanine lodide DCI-4

 $C_{25}H_{25}N_{2}I \cdot MW: 480.39$ 

$$H_0C_2 - N$$
 $CH = CH - CH = N - C_2H_0$ 

### Characteristics

Lambdachrome® number: 7450 CAS registry number: 4727-50-8

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 708 nm

Molar absorptivity: 22.5 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

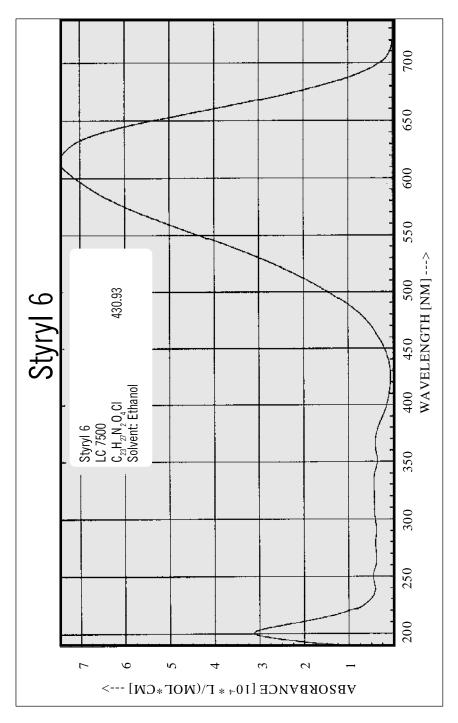
Fluorescence maximum: -

For research and development purposes only.

## Lasing Performance

Saturable absorber for the ruby laser; applicable around 700 nm<sup>1., 2., 3., 4.)</sup>.

- 1. M. L. Spaeth, W. R. Sooy, *J. Chem. Phys.* 48(5), 2315 (1968).
- 2. I. K. Krasyuk et al., *JETP Letters* 7(4), 89 (1968).
- 3. H. W. Mocker, R. J. Collins, Appl. Phys. Letters 7(10), 270 (1965).
- 4. V. I. Malyshev, A. S. Markin, A. A. Sychev, JETP Letters 6, 34 (1967).



# Styryl 6 (LC 7500)

## Constitution

2-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-1,3,3-trimethyl-3H-indolium Perchlorate LDS 730

 $C_{23}H_{27}N_2O_4CI \cdot MW: 430.93$ 

$$CH_3$$

$$CH_3$$

$$CH_3$$

$$CH_3$$

$$CH_3$$

$$CH_3$$

#### Characteristics

Lambdachrome® number: 7500 CAS registry number: -

Appearance: blue, crystalline solid

Absorption maximum (in ethanol): 615 nm

Molar absorptivity: 7.38 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

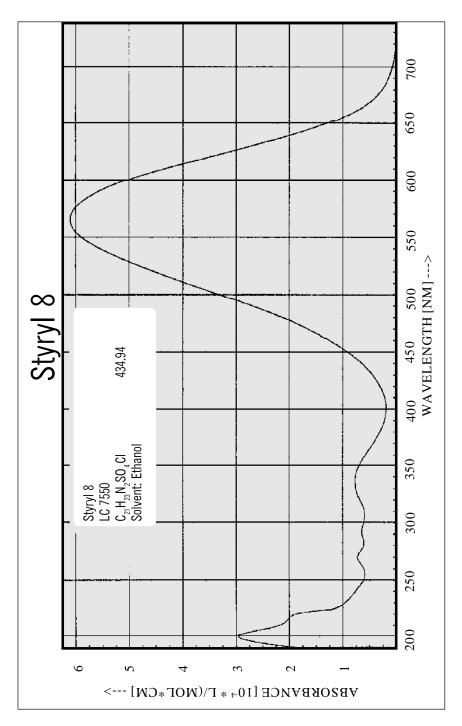
Fluorescence maximum: - For research and development purposes only.

## Lasing Performance

Laser dye for pulsed operation; tunable around 720 nm

Pump	)	D	ye Laser Ch	aracteri	stics			
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.	
	[nm]	[nm]	[nm]	[%]	[g/l]			
Nd:YAG, 2nd	532	721	708 - 735	16	0.28	PC	1	

- 1. Lambda Physik, Wall Chart 1996.
- 2. K. Kato, IEEE J. Quantaum Electr. QE-16(10), 1017 (1980).



# Styryl 8 (LC 7550)

### Constitution

2-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-3-ethylbenzothoazolium Perchlorat LDS 751

C<sub>21</sub>H<sub>23</sub>N<sub>2</sub>SO<sub>4</sub>CI · MW: 434.94

$$CIO_4 \xrightarrow{I} C_2H_5$$

$$CH = CH)_2 - N(CH_3)_5$$

#### Characteristics

Lambdachrome® number: 7550 CAS registry number: 76433-29-9

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 570 nm

Molar absorptivity:  $6.15 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

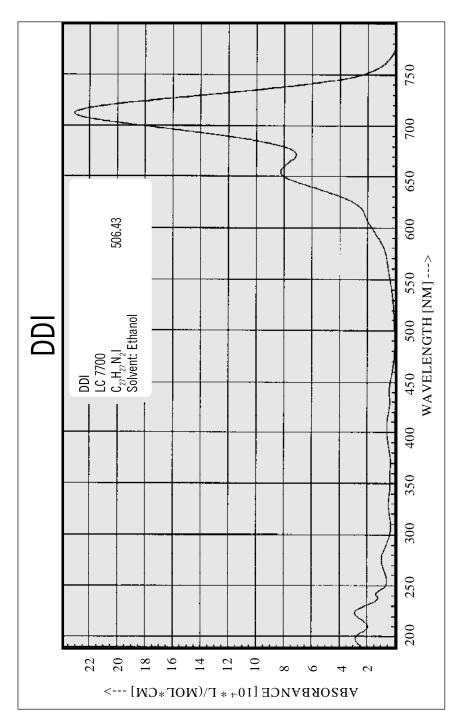
Fluorescence maximum: - For research and development purposes only.

## Lasing Performance

Laser dye for pulsed and CW operation; tunable around 750 nm

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Nd:YAG, 2nd Cu-vapor CW, Ar+	532 510 VIS	750 711 780	717 - 780 703 - 724 700 - 840	13 3 -	0.15 1.70 -	PC Methanol Pc./Eg.	1 2 3, 4

- 1. Lambda Physik, Wall Chart 1996.
- 2. M. Broyer et al., *Appl. Phys.* <u>B35</u>, 31 (1984).
- 3. J. Hoffnagle et al., *Opt. Commun.* 42(4), 267 (1982).
- 4. J. J. L. Mulders, L. W. G. Steenhuysen, *Opt. Commun.* <u>54(5)</u>, 295 (1985).



# DDI (LC 7700)

### Constitution

1,1'-Diethyl-2,2'-dicarbocyanine lodide

C<sub>27</sub>H<sub>27</sub>N<sub>2</sub>I · MW: 506.43

$$C_2H_3$$

$$C_2H_3$$

$$C_2H_3$$

$$C_2H_5$$

#### Characteristics

Lambdachrome® number: 7700 CAS registry number: 14187-31-6

Appearance: green, crystalline solid

Absorption maximum (in methanol): 710 nm

Molar absorptivity: 23.0 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 745 nm

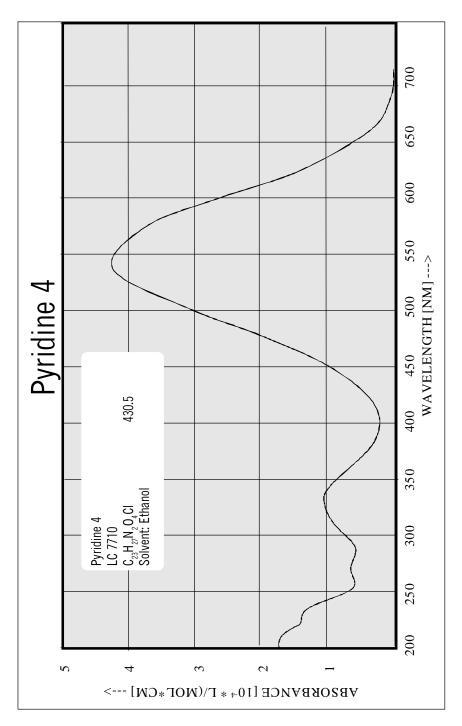
For research and development purposes only.

## Lasing Performance

Laser dye for pulsed operation; tunable around 800 nm. Saturable absorber for the Ruby laser and flashlamp pumped Cresyl Violet and Rhodamine 700 dye lasers; applicable around 710 nm<sup>1., 2., 3.)</sup>.

Pur	пр	D	ve Laser Ch	naracteri	stics			
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.	
Ruby	694	806	-	13	0.03	Ethanol	4	

- 1. E. G. Arthurs et al., *Appl. Phys. Letters* 20(3), 125 (1972).
- 2. M. E. Mack, IEEE J. Quantum Electr. QE-4, 1015 (1968).
- 3. W. Sibbett, J. R. Taylor, *IEEE J. Quantum Electr.* 20(2), 108 (1984).
- 4. A. M. Bonch-Bruevich, Opt. Spectr. 28, 51 (1970).



# Pyridine 4 (LC 7710)

### Constitution

1-Ethyl-4-(4-(9-(2,3,6,7-tetrahydro-1H,5H-benzo(i,j)-chinolizinium))-1,3-butadienyl)-pyridinium Perchlorate

 $C_{23}H_{27}N_2O_4CI \cdot MW: 430.5$ 

#### Characteristics

Lambdachrome® number: 7710

CAS Registry number:

Appearance: dark brown, crystalline solid

Absorption maximum (in ethanol): 550 nm

Molar absorptivity:  $4.26 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$ 

Fluorescence maximum (in ethanol): - For research and development purposes only.

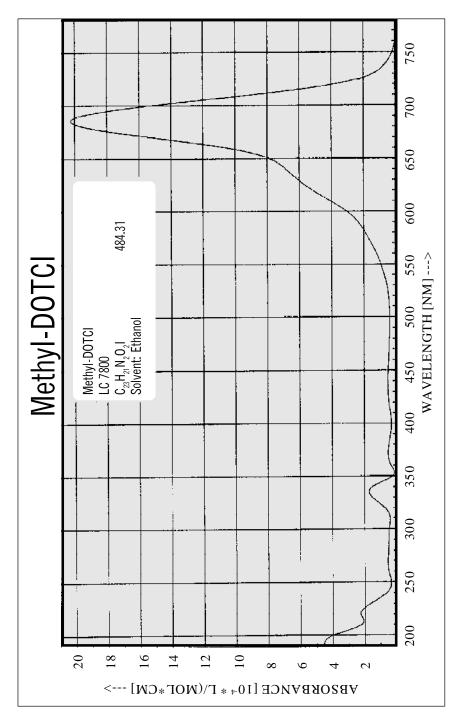
## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 770 nm.

Pump	)	Dye Laser Characteristics						
Source	Wavelength [nm]	Peak [nm]	Range [nm]		Conc. [g/l]	Solvent	Ref.	
XeCI-Excimer	308	771	744 - 812	7	0.75	DMS0	1	

## References

1. Lambda Physik, Wall Chart 1996.



## Methyl-DOTCI (LC 7800)

### Constitution

3,3'-Dimethyloxatricarbocyanine lodide DMOTCI · NK 199

C<sub>23</sub>H<sub>21</sub>N<sub>2</sub>O<sub>2</sub>I · MW: 484.31

$$(CH = CH)^3 - CH = CH^3$$

#### Characteristics

Lambdachrome® number: 7800

Appearance: blue, crystalline solid

Absorption maximum (in ethanol): 682 nm

Molar absorptivity: 19.8 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 718 nm

For research and development purposes only.

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 780 nm.

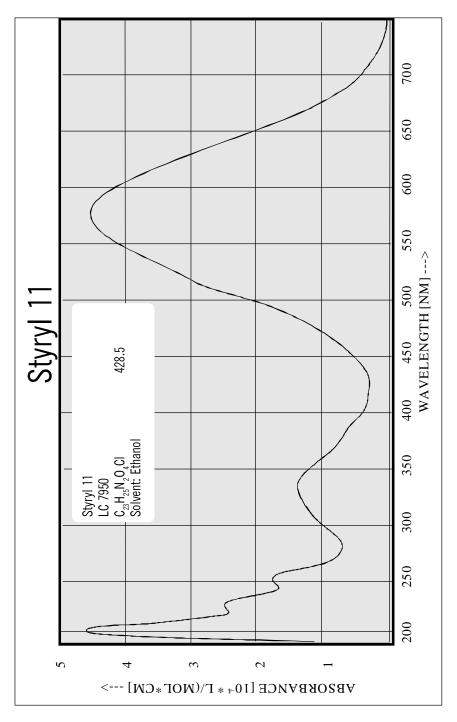
Pump	)	D	ye Laser Cha				
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	<i>792</i>	774 - 810	4	0.90	DMS0	1, 2, 3
Nitrogen	<i>337</i>	<i>780</i>	768 - 820	rel.	0.51	<i>DMSO</i>	3, 4
Nd:YAG, 2nd	<i>532</i>	<i>780</i>	-	rel.	-	<i>DMSO</i>	5
Flashlamp	-	810	-	-	0-07	<b>DMSO</b>	6
CW, Kr+	red	-	745 - 790	-	1.45	Eg.	7

### References

- 1. Lambda Physik, Wall Chart 6/83.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 403 (1981).
- 3. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. F. Bos, *Appl. Optics* 20(10), 1886 (1981).
- 6. C. Loth, P. Flamant, *Opt. Commun.* <u>21(1)</u>, 13 (1977).
- 7. J. M. Yarborough, *Appl. Phys. Letters* 24(12), 629 (1974).

#### P.S

The DOTCI (3,3'-Diethyloxatricarbocyanine lodide) shows identical performance, however, its photochemical stability is much lower.



# Styryl 11 (LC 7950)

### Constitution

1-Ethyl-4-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-quinolinium Perchlorate LDS 798

 $C_{23}H_{25}N_2O_4CI \cdot MW: 428.5$ 

$$H_5 C_2 - N_4$$
  $(CH = CH)_2 - N(CH_3)_2$ 

#### Characteristics

Lambdachrome® number: 7950 CAS registry number: 92479-59-9

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 575 nm

Molar absorptivity: 4.55 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum: - For research and development purposes only.

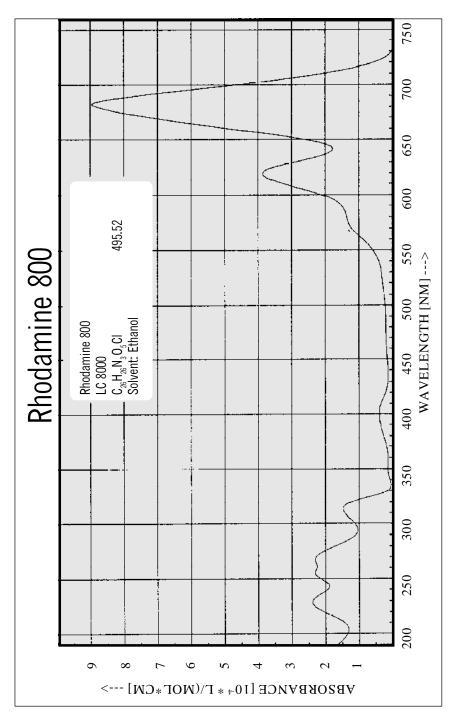
## Lasing Performance

Laser dye for pulsed operation; tunable around 800 nm

Pun	пр	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]		Conc. [g/l]	Solvent	Ref.
CW, Ar+	VIS	800	770 - 845	5	0.51	Pc./Eg.	1

## References

1. J. Hoffnagle et al., *Opt. Commun.* <u>42(4)</u>, 267 (1982).



# Rhodamine 800 (LC 8000)

### Constitution

8-Cyano-2,3,5,6,11,12,14,15-octahydro-1*H*,4*H*,10*H*,13*H*-diquinolizino[9,9a,1-*bc*:9',9a',1-*hi*]xanthylium Perchlorate

 $C_{26}H_{26}N_3O_5CI \cdot MW: 495.52$ 

#### Characteristics

Lambdachrome® number: 8000 CAS registry number: 101027-54-7

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 682 nm

Molar absorptivity: 8.95 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

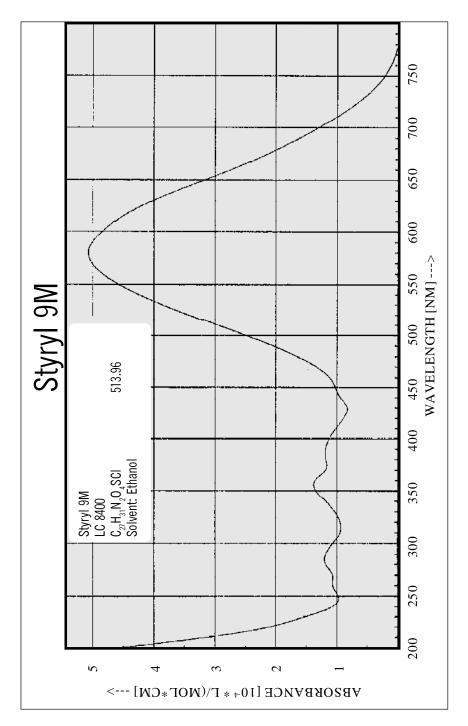
Fluorescence maximum:
For research and development purposes only.

## Lasing Performance

Very efficient laser dye for pulsed and CW operation; tunable around 810 nm

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer CW, Kr+	308 red	810 795	776 - 823 730 - 835	6 22	1.00 0.21	DMSO Eg.	1 2

- 1. Lambda Physik, Wall Chart 1996.
- 2. R. Raue, H. Harnisch, K. H. Drexhage, *Heterocycles* <u>21(1)</u>, 167 (1984).



# Styryl 9M (LC 8400)

### Constitution

2-(6-(4-Dimethylaminophenyl)-2,4-neopentylene-1,3,5-hexatrienyl)-3-methylbenzothiazolium Perchlorat LDS 821

 $C_{27}H_{31}N_2O_4SCI \cdot MW: 513.96$ 

#### Characteristics

Lambdachrome® number: 8400

CAS registry number: 120528-73-6

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 585 nm

Molar absorptivity: 5.05 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum: For research and development purposes only.

## Lasing Performance

Very efficient laser dye for pulsed and CW operation; tunable around 840 nm

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
VaCI Fyaimar	200	040	010 075	0	1 10	DMCO	1 2
XeCI-Excimer		840	810 - 875	9	1.10	DMS0	1,2
Nitrogen	<i>337</i>	840	<i>803 - 875</i>	rel.	1.03	<i>DMSO</i>	3
Nd:YAG, 2nd	<i>532</i>	824	797 - 851	15	0.26	Pc.	1, 4
Cu-vapor	510	815	793 - 845	14	0.67	Methanol	5
Flashlamp	-	840	810 - 860	-	0.01	Pc./Eg.	6, 7
CW, Ar+	VIS	830	785 - 900	-	2.0	Pc./Eg.	1, 8,9

## References

See page 236.

## P.S.

The 3-Ethyl-derivative (Styryl 9/LDS 820) shows similar performance. However, its photochemical stability in CW pumped dye lasers is slightly lower.

## References (STYRYL 9M)

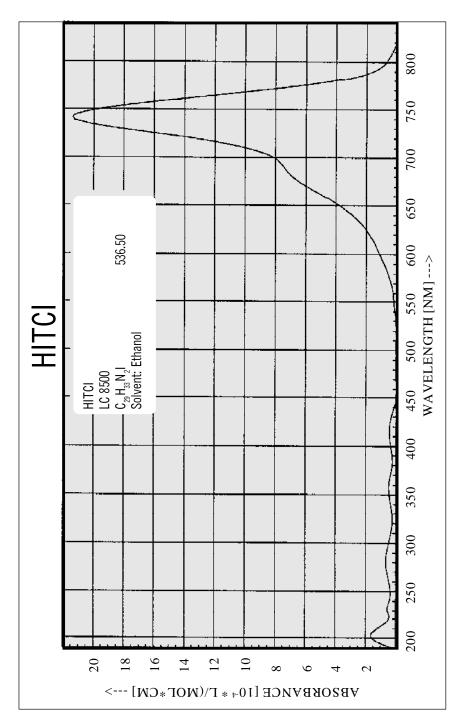
- 1. Lambda Physik, Wall Chart 1996.
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B32</u>, 9 (1983).
- 3. Lambda Physik, Data Sheet.
- 4. K. Kato, *IEEE J. Quantum Electr.* QE-16(19), 1017 (1980).
- 5. M. Broyer et al., *Appl. Phys.* B35, 31 (1984).
- 6. K. Smith, W. Sibbett, J. R. Taylor, *Opt. Commun.* 49(5), 359 (1984).
- 7. Cheng-Huei Lin, B. Marshall, *Appl. Optics* <u>23(14)</u>, 2228 (1984).
- 8. J. Hofnagle et al., *Opt. Commun.* 42(4), 267 (1982).
- 9. J. J. L. Mulders, L. W. G. Steenhuysen, *Opt. Commun.* 54(5), 295 (1985).

### References (HITCI)

- 1. Lambda Physik, Wall Chart 1996.
- 2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* <u>38(5,6)</u>, 403 (1981).
- 3. F. Bos, *Appl. Optics* 20(10), 3553 (1981).
- 4. Lambda Physik, Data Sheet.
- 5. F. Bos, *Appl. Optics* <u>20(20)</u>, 1886 (1981).
- 6. A. Hirth, K. Vollrath, J. Faure, D. Lougnot, *Opt. Commun.* 7(4), 339(1973).
- 7. Coherent, CW Dye Laser Fact Sheets.
- 8. T. F. Johnston, R. H. Brady, W. Proffitt, *Appl. Optics* <u>21(13)</u>, 2307 (1982).

## References (IR 140)

- 1. Lambda Physik, Wall Chart 6/83.
- 2. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 3. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B30</u>, 109 (1983).
- 4. Lambda Physik, Data Sheet.
- 5. F. Bos, *Appl. Optics* 20(10), 1886 (1981).
- 6. C. A. Moore, C. D. Decker, *J. Appl. Phys.* <u>49(1)</u>, 47 (1978).
- 7. C. D. Decker, *Appl. Phys. Letters* 27(11), 607 (1975).
- 8. J. P. Webb et al., *IEEE J. Quantum Electr.* <u>QE-11</u>, 114 (1975).
- 9. Coherent, CW Dye Laser Fact Sheets.
- 10. Lambda Physik.



# HITCI (LC 8500)

## Constitution

1,1',3,3,3',3'-Hexamethylindotricarbocyanine lodide Hexacyanine 3

C<sub>29</sub>H<sub>33</sub>N<sub>2</sub>I · MW: 536.50

### Characteristics

Lambdachrome® number: 8500 CAS registry number: 19764-96-6

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 741 nm

Molar absorptivity: 21.5 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 778 nm

For research and development purposes only.

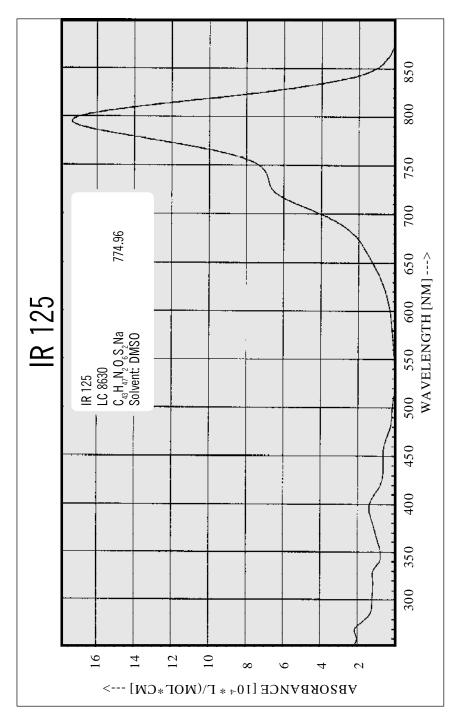
## Lasing Performance

Laser dye for pulsed and CW operation; tunable around 860 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	868	837 - 905	4	1.20	DMS0	1, 2, 3
Nitrogen	<i>337</i>	846	828 -891	rel.	1.06	<i>DMSO</i>	3, 4
Nd:YAG, 2nd	<i>532</i>	815	-	-	-	Ethanol	5
Flashlamp	-	<i>879</i>	-	-	0.11	<i>DMSO</i>	6
CW, Kr⁺	ir	880	815 - 920	10	0.3	DMSO/Eg.	7, 8

## References

See page 236.



# IR 125 (LC 8630)

## Constitution

 $C_{43}H_{47}N_2O_6S_2Na \cdot MW: 774.96$ 

### Characteristics

Lambdachrome® number: 8630 CAS registry number: 3599-32-4

Appearance: bronze, crystalline solid

Absorption maximum (in DMSO): 795 nm

Molar absorptivity: 17.3 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

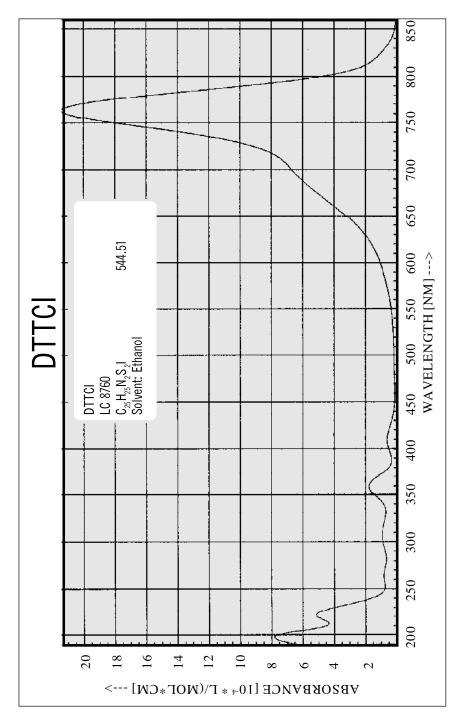
Fluorescence maximum (in chloroform): 838 For research and development purposes only.

## Lasing Performance

Laser dye for pulsed operation; tunable around 920 nm

Pump		Dye Laser Characteristics					
Source	Wavelength	Peak	Range		Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCI-Excimer	308	920	890 - 960	4	2.0	DMS0	1
Nitrogen	<i>337</i>	918	893 - 958	rel.	1.94	<i>DMSO</i>	2, 3
Nd:YAG, 2nd	<i>532</i>	913	-	3	0.39	<i>DMSO</i>	4
Flashlamp	-	940	-	-	0.08	<i>DMSO</i>	5

- 1. Lambda Physik, Wall Chart 6/90.
- 2. Lambda Physik, Data Sheet.
- 3. B. M. Pierce, R. R. Birge, *IEEE J. Quantum Electr.* <u>QE-18(7)</u>, 1164 (1982).
- 4. C. D. Decker, Appl. Phys. Letters 27(11), 607 (1975).
- 5. J. P. Webb et al., *IEEE J. Quantum Electr.* QE-11, 114 (1975).



# DTTCI (LC 8760)

#### Constitution

3,3'-Diethylthiatricarbocyanine lodide

C<sub>25</sub>H<sub>25</sub>N<sub>2</sub>S<sub>2</sub>I · MW: 544.51

$$\begin{array}{c|c} S & & & \\ & \downarrow & \\ & \downarrow$$

#### Characteristics

Lambdachrome® number: 8760 CAS registry number: 3071-70-3

Appearance: blue, crystalline solid

Absorption maximum (in ethanol): 760 nm

Molar absorptivity: 21.0 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

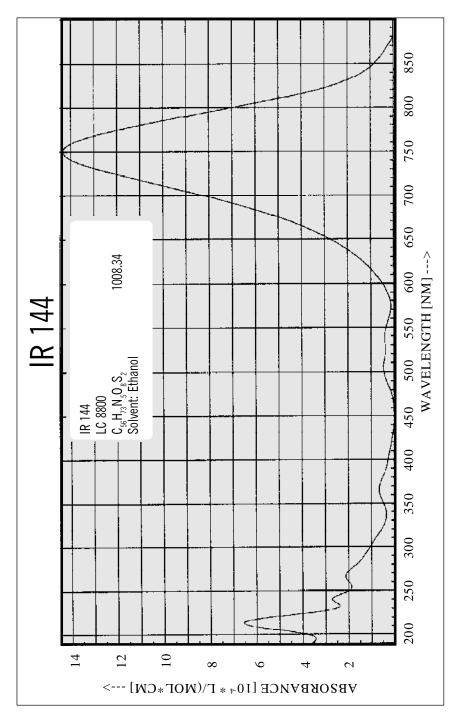
Fluorescence maximum (in chloroform): 815 For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed operation; tunable around 850 nm

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [q/l]	Solvent	Ref.
	[,,,,,]	Liiii	[,,,,,]	[/0]	ני 19		
XeCI-Excimer	308	849	828 - 883	1	0.42	DMS0	1
Nitrogen	<i>337</i>	<i>852</i>	834 - 892	rel.	0.60	<i>DMSO</i>	1, 2
Flashlamp	-	889	-	-	0.11	Methanol	3, 4

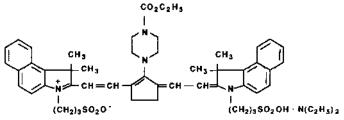
- 1. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
- 2. B. M. Pierce, R. R. Birge, *IEEE J. Quantum Electr.* <u>QE-18(7)</u>, 1164 (1982).
- 3. M. Maeda, Y. Miyazoe, Jap. J. Appl. Phys. 11(5), 692 (1972).
- 4. A. Hirth, K. Vollrath, J. Faure, D. Lougnot, *Opt. Commun.* 7(4), 339 (1973).



# IR 144 (LC 8800)

#### Constitution

 $C_{56}H_{72}N_5O_{\circ}S_{2} \cdot MW: 1008.34$ 



#### Characteristics

Lambdachrome® number: 8800 CAS registry number: 54849-69-3

Appearance: bronze, crystalline solid

Absorption maximum (in DMSO): 750 nm

Molar absorptivity: 14.1 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

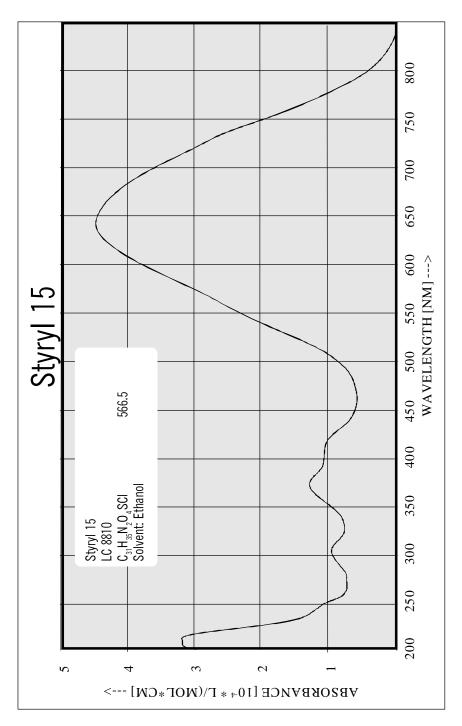
Fluorescence maximum (in ethanol): 848 For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed operation; tunable around 880 nm

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Nd:YAG, 2nd Flashlamp	308 337 532	869 874 867 880	856 - 879 862 - 892 -	3 rel. 6	1.08 1.61 0.30 0.10	DMSO DMSO DMSO DMSO	1, 2 1, 3, 4 5, 6 7

- 1. F. Bos, Appl. Optics 20(20), 3553 (1981).
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B30</u>, 109 (1983).
- 3. Lambda Physik, Data Sheet.
- 4. B. M. Pierce, R. R. Birge, IEEE J. Quantum Electr. QE-18(7), 1164 (1982).
- 5. F. Bos, *Appl. Optics*. <u>20(10)</u>, 1886 (1981).
- 6. C. A. Moore, C. D. Decker, J. Appl. Phys. 49(1), 47 (1978).
- 7. J. P. Webb et al., *IEEE J. Quantum Electr.* QE-11, 114 (1975).



# Styryl 15 (LC 8810)

#### Constitution

2-(6-(9-(2,3,6,7-Tetrahydro-1H,5H-benzo(i,j)-chinolizinium))-2,4-neopentylene-1,3,5-hexatrienyl)-3-methylbenzothiazolium Perchlorate

 $C_{31}H_{35}N_2O_4SCI \cdot MW: 566.5$ 

#### Characteristics

Lambdachrome® number: 8810 CAS Registry number: -

Appearance: dark green, crystalline solid

Absorption maximum (in ethanol): 648 nm

Molar absorptivity: 4.42 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): - For research and development purposes only.

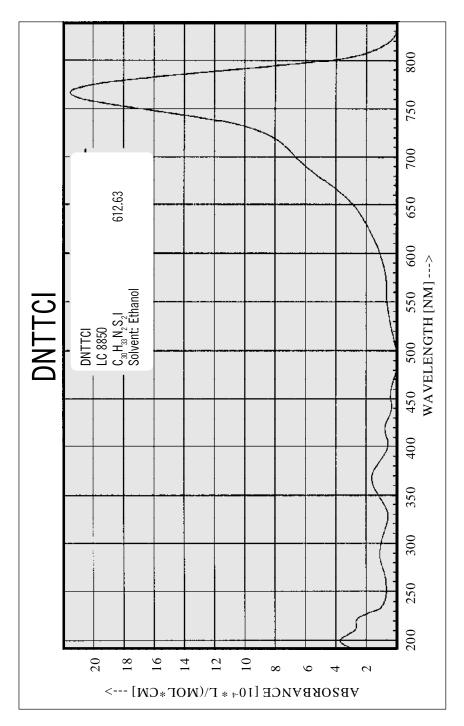
#### Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 880 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer Nd:YAG, 2nd	308 532	880 880	856 - 918 856 - 918	<i>7 7</i>	1.15 0.62	DMSO Pc.	1 1

#### References

1. Lambda Physik, Wall Chart 1996.



# **DNTTCI (LC 8850)**

#### Constitution

3,3'-Diethyl-9,11-neopentylenethiatricarbocyanine lodide

C<sub>30</sub>H<sub>33</sub>N<sub>2</sub>S<sub>2</sub>I · MW: 612.63

#### Characteristics

Lambdachrome® number: 8850 CAS registry number: -

Appearance: brass colored, crystalline solid

Absorption maximum (in ethanol): 765nm

Molar absorptivity: 22.5 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum : - For research and development purposes only.

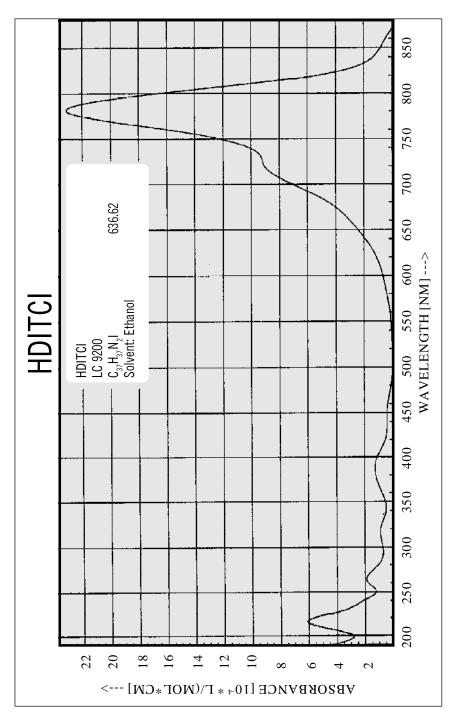
#### Lasing Performance

Laser dye for pulsed operation; tunable around 880 nm

Pum	р	Dye Laser Characteristics						
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.	
Flashlamp	-	880	-	-	0.12	DMS0	1	

#### References

1. A. Hirth, K. Vollrath, J. Faure, D. Lougnot, *Opt. Commun.* 7(4), 339 (1973).



# **HDITCI (LC 9200)**

#### Constitution

1,1',3,3,3',3'-Hexamethyl-4,4',5,5'-dibenzo-2,2'-indotricarbocyanine lodide Hexadibenzocyanin 3

C<sub>37</sub>H<sub>37</sub>N<sub>2</sub>I · MW: 636.62

#### Characteristics

Lambdachrome® number: 9200 CAS registry number: 23178-67-8

Appearance: bronze colored, crystalline solid

Absorption maximum (in ethanol): 780 nm

Molar absorptivity: 23.1 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

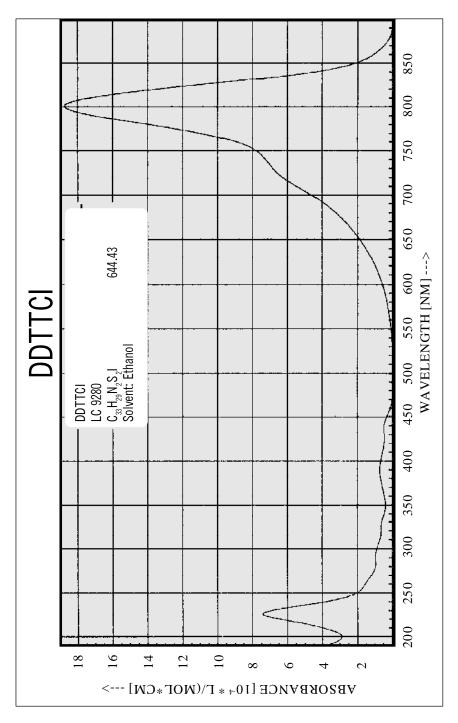
Fluorescence maximum (in chloroform): 824 For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed and CW operation; tunable around 920 nm. Saturable absorber for CW pumped Oxazine 170 dye lasers; applicable around 780 nm <sup>1.)</sup>.

Pump	)	Dye Laser Characteristics						
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.	
	[nm]	[nm]	[nm]	[%]	[g/l]			_
XeCl-Excimer CW, Kr+	308 red	932 920	899-975 880 - 960	4 2	0.90 0.95	DMSO DMSO/Eg.	1, 2 3	

- 1. G. W. Fehrenbach et al., *Appl. Phys. Letters* 33(2), 159 (1978).
- 2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* <u>B30</u>, 109 (1983).
- 3. K. M. Romanek et al., *Opt. Commun.* 21(1), 16 (1977).



# DDTTCI (LC 9280)

#### Constitution

3,3'-Diethyl-4,4',5,5'-dibenzothiatricarbocyanine lodide Hexadibenzocyaini 45

 $C_{33}H_{29}N_2S_2I \cdot MW: 644.43$ 

$$C_2H_5$$

$$C_2H_5$$

$$C_2H_5$$

#### Characteristics

Lambdachrome® number: 9280 CAS registry number: -

Appearance: bronze colored, crystalline solid

Absorption maximum (in ethanol): 798 nm

Molar absorptivity: 19.6 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

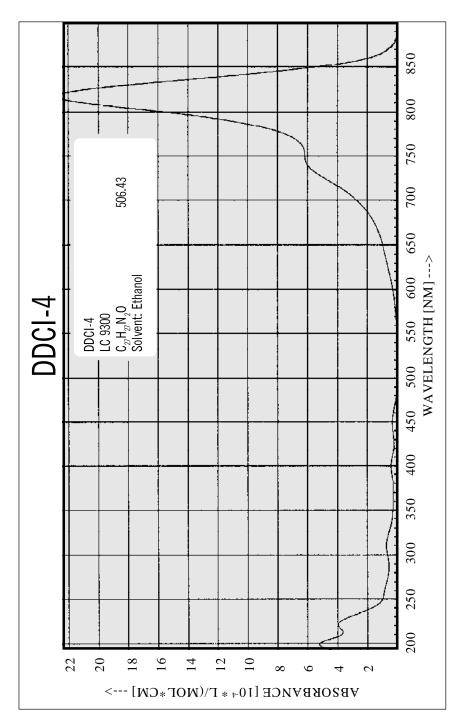
Fluorescence maximum:
For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed operation; tunable around 930 nm

Pum	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer Nitrogen Flashlamp	308 337 -	932 870 946	899 - 975 - -	5 rel. -	0.92 0.32	DMSO Acetone Pc./DMSO	1, 2 3 4

- 1. Lambda Physik.
- 2. H. Telle, W. Hüffer, D. Basting, Opt. Commun. 38(5,6), 402 (1981).
- 3. Chinlon Lin, IEEE J. Quantum Electr. QE-11, 61 (1975).
- 4. A. Hirth, J. Faure, D. Lougnot, *Opt. Commun.* <u>8(4)</u>, 318 (1973).



# DDCI-4 (LC 9300)

#### Constitution

1,2'-Diethyl-4,4'-dicarbocyanine lodide
NK 1144

 $C_{27}H_{27}N_2O \cdot MW: 506.43$ 

$$H_5C_2 - N_4$$
 (CH = CH)<sub>2</sub> - CH =  $N - C_2H_5$ 

#### Characteristics

Lambdachrome® number: 9300

CAS registry number:

Appearance: green, crystalline solid

Absorption maximum (in ethanol): 815 nm

Molar absorptivity: 23.6 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 850 nm

For research and development purposes only.

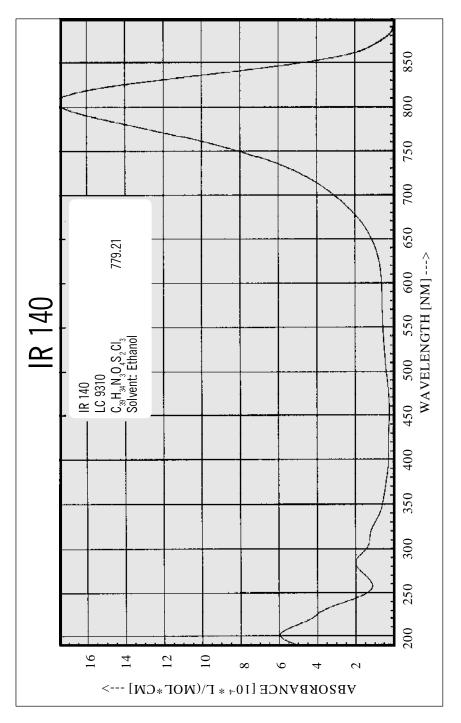
#### Lasing Performance

Laser dye for pulsed operation; tunable around 930 nm

Pun	ıp	Dye Laser Characteristics						
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.	
Nitrogen	337	930	-	rel.	0.50	Acetone	1	

#### References

1. Chinlon Lin, IEEE J. Quantum Electr. QE-11, 61 (1975).



# IR 140 (LC 9310)

#### Constitution

C<sub>39</sub>H<sub>34</sub>N<sub>3</sub>O<sub>4</sub>S<sub>2</sub>CI<sub>3</sub> · MW: 779.21

#### Characteristics

Lambdachrome® number: 9310
CAS registry number: 53655-17-7

Appearance: brown, crystalline solid

Absorption maximum (in ethanol): 810 nm

Molar absorptivity: 15.0 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): 860 nm

For research and development purposes only.

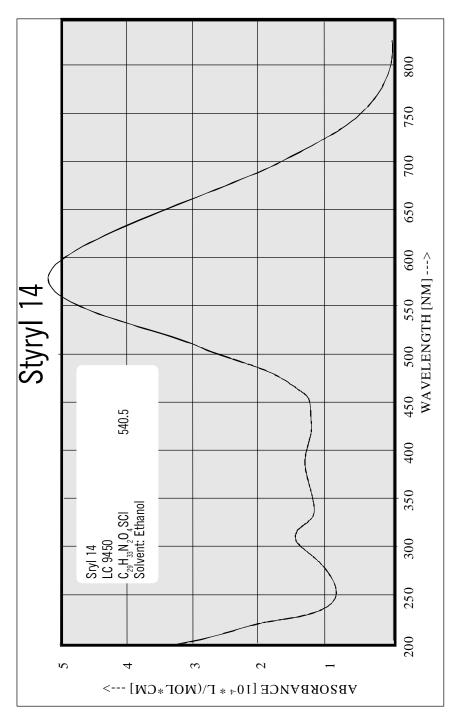
#### Lasing Performance

Laser dye for pulsed operation; tunable around 950 nm

Pump	)	Dye Laser Characteristics					
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
V 01 5 :	200	050	000 005	•	1.10	54400	1 0 0
XeCI-Excimer	308	950	<i>882 - 985</i>	3	1.10	<i>DMSO</i>	<i>1, 2, 3</i>
Nitrogen	<i>337</i>	910	900 - 936	rel.	0.78	<i>DMSO</i>	3, 4
Nd:YAG, 2nd	<i>532</i>	<i>890</i>	-	5	0.31	<i>DMSO</i>	5, 6, 7
Flashlamp	-	950	-	-	0.08	DMS0	8
CW, Kr+	VIS	970	880 - 1010	14	0.71	DMSO/Eg.	9, 10

#### References

See page 236.



# Styryl 14 (LC 9450)

#### Constitution

2-(8-(4-p-Dimetyhlaminophenyl)-2,4-neopentylene-1,3,5,7-octatetraenyl)-3-methylbenzothiazolium Perchlorate

 $C_{29}H_{33}N_{2}O_{4}SCI \cdot MW: 540.5$ 

#### Characteristics

Lambdachrome® number: 9450 CAS Registry number: -

Appearance: dark green, crystalline solid

Absorption maximum (in ethanol): 588 nm

Molar absorptivity: 5.17 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in ethanol): For research and development purposes only.

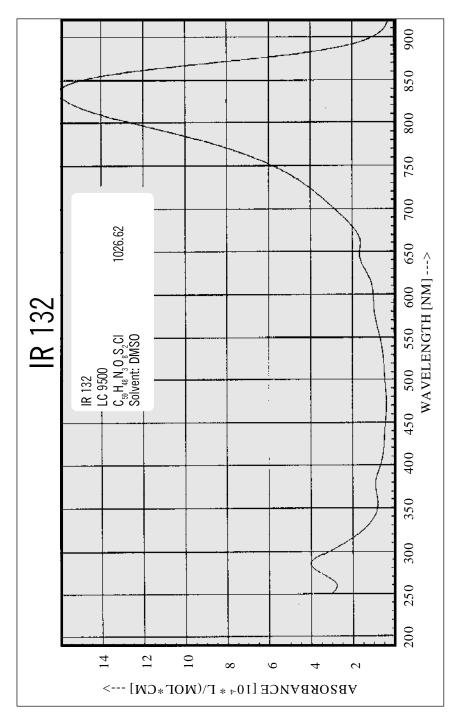
#### Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 950 nm.

Pump	)	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nd:YAG, 2nd	308 532	945 945	904 - 992 904 - 990	9 9	1.10 0.27	DMSO Pc.	1 1

#### References

1. Lambda Physik, Wall Chart 1996.



# IR 132 (LC 9500)

#### Constitution

C<sub>59</sub>H<sub>48</sub>N<sub>3</sub>O<sub>8</sub>S<sub>2</sub>CI · MW: 1026.62

#### Characteristics

Lambdachrome® number: 9500 CAS registry number: 62669-62-9

Appearance: red, crystalline solid

Absorption maximum (in DMSO): 830 nm

Molar absorptivity: 15.9 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in chloroform): 861 nm

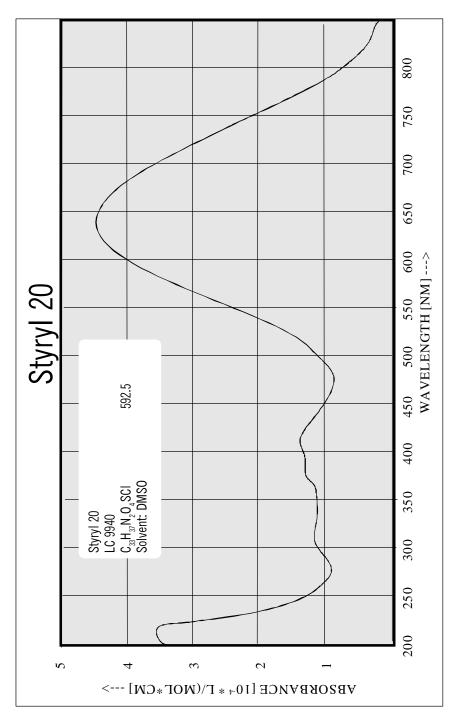
For research and development purposes only.

#### Lasing Performance

Laser dye for pulsed operation; tunable around 950 nm

Pump	)	D	ye Laser Cha	aracteri	stics		
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
Nd:YAG, 2nd	<i>532</i>	909	-	1	0.51	<i>DMSO</i>	1
Flashlamp	-	972	-	-	0.10	<i>DMSO</i>	2
ML, Kr+	VIS	-	863 - 1048	-	0.80	DMS0	3

- 1. C. D. Decker, *Appl. Phys. Letters* <u>27(11)</u>, 607 (1975).
- 2. J. P. Webb et al., *IEEE J. Quantum Electr.* <u>QE-11</u>, 114(1975).
- 3. M. Leduc, *Opt. Commun.* 31(1), 66(1979).



# Styryl 20 (LC 9940)

#### Constitution

2-(8-(9-(2,3,6,7-Tetrahydro-1H,5H-benzo(i,j)chinolizinium))-2,4-neopentylene-1,3,5,7-octatetraenyl)-3-methylbenzothiazolium Perchlorate

 $C_{33}H_{37}N_{2}O_{4}SCI \cdot MW: 592.5$ 

#### Characteristics

Lambdachrome® number: 9940 CAS registry number: -

Appearance: dark green, crystalline solid

Absorption maximum (in ethanol): 645 nm

Molar absorptivity: 4.70 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum:
For research and development purposes only.

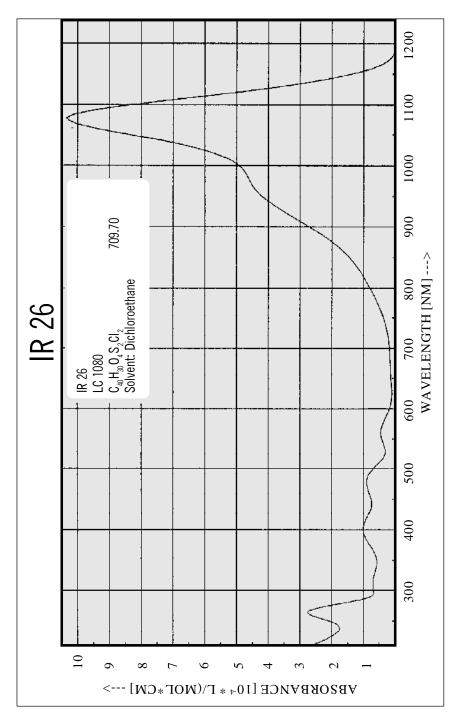
#### Lasing Performance

Efficient IR laser dye for pulsed and CW operation; tunable around 990 nm.

Pump	)	D	ye Laser Cha	aracteri	stics		
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCI-Excimer Nd:YAG, 2nd	308 532	994 994	970 - 1036 970 - 1036	4 4	1.10 0.68	DMSO Pc.	1 1

#### References

1. Lambda Physik, Wall Chart 1996.



# IR 26 (LC 1080)

#### Constitution

C<sub>40</sub>H<sub>30</sub>O<sub>4</sub>S<sub>2</sub>CI<sub>2</sub> · MW: 709.70

#### Characteristics

Lambdachrome® number: 1080 CAS registry number: 76871-75-5

Appearance: dark green, crystalline solid

Absorption maximum (in dichloroethane): 1080 nm

Molar absorptivity: 10.3 x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum (in benzyl alcohol): 1180 nm

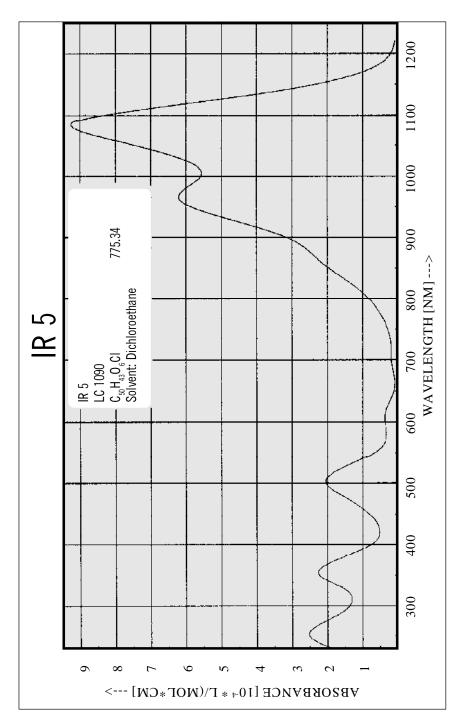
For research and development purposes only.

#### Lasing Performance

Efficient laser dye for synchronously pumped (Nd:YAG) dye lasers; tunable around 1030 nm. Saturable absorber for Nd:YAG lasers.

Pur	тр	Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]		Conc. [g/l]	Solvent	Ref.
Nd:YAG	1064	1290	1200 - 1320	4	0.71	Bz.	1, 2, 3, 4

- 1. W. Kranitzky et al., *Opt. Commun.* 36(2), 149 (1981).
- 2. A. Seilmeier et al., Optics. Letters 8(4), 205 (1983).
- 3. A. Seilmeier, *Opt. Quantum Electr.* <u>16</u>, 89 (1984).
- 4. K. Kato, *IEEE J. Quantum Electr.* <u>QE-20(7)</u>, 698 (1984).



# IR 5 (LC 1090)

#### Constitution

C<sub>50</sub>H<sub>42</sub>O<sub>6</sub>CI · MW: 775.34

#### Characteristics

Lambdachrome® number: 1090 CAS registry number: 61010-01-3

Appearance: dark green, crystalline solid

Absorption maximum (in dichloroethane): 1090 nm

Molar absorptivity: 9.30x 10<sup>4</sup> L mol<sup>-1</sup> cm<sup>-1</sup>

Fluorescence maximum:
For research and development purposes only.

#### Lasing Performance

Laser dye for synchronously pumped (Nd:YAG) dye lasers; tunable around 1300 nm. Saturable absorber for Nd:glass lasers<sup>2.</sup>).

Pun	<i>1</i> р	E	Dye Laser Cha	racteri	stics		
Source	Wavelength	Peak	Range	Effic.	Conc.	Solvent	Ref.
	[nm]	[nm]	[nm]	[%]	[g/l]		
Nd:YAG	1064	1320	1180 - 1400	10	0.64	DCE	1, 2,

- 1. T. Elsaesser et al., IEEE J. Quantum Electr. QE-20(3), 191 (1984).
- 2. R. R. Alfano et al., IEEE J. Quantum Electr. QE-17(3), 290 (1981).

### Saturable Absorbers

- a: Dye/LC#
- b: Application
- c: Literature
- a: DASPI/LC4660
- b: s.a. for FL-pumped C1-, C102-, C466-, C6H-dye laser applicable in the 480 500 nm range
- c: Sibbett et al., *Opt. Commun.* 46(1), 32 (1983)
- a: DASBTI/LC 5280
- b: s.a. for FL-pumped C6-, C522-, C153-, Rh110-dye laser applicable in the 500 540 nm range
- W. Sibbett et al., Opt. Commun. 44(2), 121 (1982)
   W. Sibbett et al., Appl. Phys. B29, 191 (1982)
   W. Sibbett et al., IEEE J. Quantum Electr. QE-19(4), 558 (1983)
- a: DOCI/LC 5410
- b: s.a. for FL-pumped C102-, C500-, XeCl-pumped C102-dye laser applicable in the 480 500 nm range
- c: J.C. Mialocq et al., *Appl. Phys. Lett.* 33(9), 819 (1978) R. Wyatt, *Opt. Commun.* 38(1), 64 (1981) Th. Varghese. *Opt. Commun.* 44(5), 353 (1983)
  - M. Watanabe et al., *Appl. Phys. Lett.* 45(9), 929 (1984)
- a: DMETCI/LC 5460
- b: s.a. for FL-pumped C153-dye laser applicable in the 530 550 nm range
- c: W. Sibbett et al., Opt. Commun. 43(1), 50 (1982)
- a: DQOCI/LC 5920
- b: s.a. for FL-pumped Fluorol 7GA-, Rh6G-dye laser applicable in the 550 590 nm range
- c: E. Lill et al., *Opt. Commun.* 20(2), 223 (1977) R.S. Adrain et al., *Opt. Commun.* 12(2), 140 (1974)
- a: DCI-2/LC 5950
- b: s.a. for FL-pumped Rh6G-dye laser applicable in the 560 600 nm range
- c: M. Maeda et al., Japan J. Appl. Phys. <u>13(1)</u>, 193 (1974)

- a: Malachit Green/LC 6220
- b: additive for CW-pumped Rh6G-dye laser applicable in combination with DODCI as s.a.
- c: M. Young, *Appl. Optics* 18(19), 3212 (1979) A. Watanabe et al., *IEEE J. Quant. Electr.* QE-19(4), 533 (1983)
- a: DTCI/LC 6250
- b: s.a. for FL-pumped Rh6G-dye laser applicable in the 580 nm range
- c: M. Maeda et al., *J. Appl. Phys.* 13(1), 193 (1974)
- a: DQTCI/LC 6290
- b: s.a. for FL-pumped RhB-dye laser applicable in the 600 620 nm range
- c: E. G. Arthurs et al., *Appl. Phys. Lett.* <u>20(3)</u>, 125 (1972) E. Lill et al., *Opt. Commun.* <u>22(1)</u>, 107 (1977)
- a: DODCI/LC 6550
- b: "state of the art"-s.a. for CW- and FL-pumped Rh6G-dye laser applicable in the 570 600 nm range
- c: elsewhere
- a: DTDCI/LC7260
- b: s.a. for FL-pumped Rh101-, CV/Rh6G-dye laser applicable in the 630 650 nm range
- c: J. Negran et al., *Appl. Optics* <u>17(17)</u>, 2812 (1978) E.G. Arthurs et al., *Appl. Phys. Lett.* <u>20(3)</u>, 125 (1972)
- a: Cryptocyanine/LC 7450
- b: s.a. for Ruby-lasers
- c: I.K. Krasyuk et al., *JETP Letters* 7(4), 89 (1968) H.W. Mocker et al., *Appl Phys. Lett.* 7(10), 270 (1965) V.I. Malyshev et al., *JETP Letters* 6(2), 34 (1967)
- a: DDI/LC 7700
- b: s.a. for FL-pumped CV/Rh6G-, Rh700-dye laser, Ruby-laser applicable in the 680 710 nm range
- c: E.G. Arthurs et al., Appl. Phys. Lett. 20(3), 125 (1972)
   M.E. Mack, IEEE J. Quant. Electr. QE-4, 1015 (1968)
   W. Sibbett et al., IEEE J. Quant. Electr. QE-20(2), 108 (1984)
- a: DOTCI/LC 7880
- b: s.a. for FL-pumped CV-dye laser applicable in the 680 700 nm range
- c: E.G. Arthurs et al., *Appl. Phys. Lett.* 20(3), 125 (1972)

- a: HITCI/LC 8500
- b: s.a. for FL-pumped DOTCI-dye laser applicable in the 750 800 nm range
- c: A. Hirth et al., *Opt. Commun.* 7(4), 339 (1973)
- a: HDITCI/LC 9200
- b: s.a. for CW-pumped Oxazine 750-dye laser applicable in the 750 830 nm range
- c: G.W. Fehrenbach et al., Appl. Phys. Lett. 33(2), 159 (1978)
- a: <u>IR 140/LC 9310</u>
- b: s.a. for FL- and CW-pumped Styryl 9-dye laser applicable in the 840 nm range
- c: K. Smith et al., Opt. Commun. 49(5), 359 (1984)
- a: IR 26/LC 1080
- b: s.a. for Nd:YAG-laser
- c: B. Kopainsky et al., *Appl. Phys.* <u>B-29</u>, 15 (1982)
- a: IR 5/LC 1090
- b: s.a. for Nd:glass-laser
- c: R.R. Alfano et al., IEEE J. Quant. Electr. QE-17(3), 290 (1981)

#### Abbreviations used:

- C Coumarin
- CV Cresyl Violet
- CW continuous wave
- FL flashlamp
- LC Lambdachrome®
- Rh Rhodamine
- s.a. saturable absorber

# Reference List

LC	Lambdachrome®	corresponds
No.	Dye	to
3300	BM-Terphenyl	DMT
3400	PTP	p-Terphenyl
3500	TMQ	TMQ
3570	BMQ	-
3590	DMQ	-
3600	Butyl-PBD	BPBD-365
3640	PBD	PBD
3650	TMI	-
3690	QUI	-
3700	PPO	PPO PPO
3720	PPF	-
3740	PQP	p-Quaterphenyl
3780	BBD	-
3800	Polyphenyl 1	-
3810	Polyphenyl 2	-
3860	BiBuQ	BBQ
3900	Quinolon 390	LD 390
3950	a-NPO	a-NPO
3990	Furan 2	-
4000	PBB0	PBB0
4090	DPS	DPS
4100	Stilbene 1	-
4150	BB0	BB0
4200	Stilbene 3	Stilbene 420
4220	Carbostyryl 7	Carbostyryl 124
4230	POPOP	POPOP
4240	Coumarin 4	Umbelliferon 47
4250	Bis-MSB	Bis-MSB
4300	BBOT	-
4350	Carbostyryl 3	Carbostyryl 165
4400	Coumarin 120	Coumarin 440
4500	Coumarin 2	Coumarin 450
4650	DASPI	-
4660	Coumarin 466	LD 466, C1H

LC	Lambdachrome®	corresponds
No.	Dye	to
4700	Coumarin 47	Coumarin 460, Coumarin 1
4800	Coumarin 102	Coumarin 480
4810	Coumarin 152A	Coumarin 481, Coumarin 35
4850	Coumarin 152	Coumarin 485
4900	Coumarin 151	Coumarin
4910	Coumarin 6H	LD 490
5000	Coumarin 307	Coumarin 503
5010	Coumarin 500	Coumarin 500
5040	Coumarin 314	Coumarin 504
5100	Coumarin 510	Coumarin 510
5150	Coumarin 30	Coumarin 515
5210	Coumarin 334	Coumarin 521
5220	Coumarin 522	Coumarin 522
5280	DASBTI	-
5350	Coumarin 7	Coumarin 535
5360	Brillant Sulfaflavine	Brillant Sulfaflavin
5370	Coumarin 6	Coumarin 540
5400	Coumarin 153	Coumarin 540A
5410	DOCI	DOC, NK 85
5460	DMETCI	-
5520	Uranin	Disodium Fluorescein
5530	Fluorescein 27	Fluorescein 548
5700	Rhodamine 110	Rhodamine 560
5750	Rhodamine 19	Rhodamine 575
5900	Rhodamine 6G	Rhodamine 590
5901	Rhodamine 6G (Perchl.)	Rhodamine 590 (Perchl.)
5920	DQOCI	-
5950	DCI-2	-
6100	Rhodamine B	Rhodamine 610
6101	Rhodamine B (Perchl.)	Rhodamine 610 (Perchl.)
6200	Sulforhodamine B	Kiton Red 620
6220	Malachit Green	-
6250	DTCI	NK 76
6290	DQTCI	-
6400	Rhodamine 101	Rhodamine 640
6500	DCM	DCM
	·	· · · · · · · · · · · · · · · · · · ·

LC	Lambdachrome®	corresponds
No.	Dye	to
6501	DCM-spec.	-
6550	DODCI	DODC-lodide
6600	Sulforhodamine 101	Sulforhodamine 640
6700	Cresyl Violet	Cresyl Violet 670
6900	Nile Blue	Nile Blue 690
6950	Oxazine 4	LD 690
7000	Rhodamine 700	LD 700
7100	Pyridin 1	LDS 698
7210	Oxazine 170	Oxazine 720
7250	Oxazine 1	Oxazine 725
7260	DTDCI	DTDC-lodide, NK 136
7270	Oxazine 750	Oxazine 750
7300	Pyridin 2	LDS 722
7400	HIDCI	Hexacyanine 2, NK 529
7500	Styryl 6	LDS 730
7550	Styryl 8	LDS 751
7700	DDI	NK 1456
7710	Pyridin 4	-
7800	Methyl-DOTCI	DMOTC-lodide, NK 199
7880	DOTCI	DEOTC-lodide
7950	Styryl 11	LDS 798
8000	Rhodamine 800	-
8410	Styryl 9 (M)	LDS 820 (821)
8500	HITCI	Hexacyanine 3, NK 125
8630	IR 125	IR 125
8760	DTTCI	DTTC-lodide, NK 126
8800	IR 144	IR 144
8810	Styryl 15	-
8850	DNTTCI	-
9300	DDCI-4	NK 1144
9310	IR 140	IR 140
9500	IR 132	IR 132
9940	Styryl 20	-
10600	IR 25	-
10800	IR 26	IR 26
10810	IR 26 (HFB)	-
10900	IR 5	IR 5

# Tuning Curves

Tuning curves of excimer-, nitrogen-, Nd:YAG-, and CW-laser pumped dye lasers are shown on the following pages. Parameters given are defined as follows:

#### Peak

Output maximum of the tuning curve, in nanometers.

#### **Tuning Range**

The tuning range is defined as the range, in nanometers, giving an efficiency larger than 10 percent of the maximum. All tuning ranges are restricted to broadband operation.

#### Efficiency

Dye laser output at the maximum of the tuning range relative to pump laser input, in percent. Efficiency may change in other configurations or pump power levels.

#### Pump Wavelength

Pump wavelength used in nanometers.

#### Solvent

BZ=Benzyl Alcohol, EG=Ethylene Glycol, CH=Cyclohexane, DI=Dioxane, ME=Methanol, DMSO=Dimethylsulfoxide, PC=Propylene Carbonate.

#### Concentration

Amount of dye, in grams, for 1 liter stock solution. There is an optimum concentration for a given dye, wavelength and input power. In case of continuously pumped dye lasers this optimum concentration generally lies between 60 percent and 80 percent absorption of the pump power, and in the case of transversally pumped pulsed dye lasers, at 99 percent absorption of the pump energy within 1 millimeters of the dye solution. Higher concentration causes the tuning curves to be shifted slightly to the red, while lower concentrations will result in blue shift. Optimization of the dye concentration is accomplished by adding either pure solvent or a solution of higher concentration than that recommended to the solution in the dye circulation system, until optimum power is at maximum.

#### **Stability**

The accumulated pump energy t, in Wh, causing a decrease in dye laser output to 50 percent of the initial value for 1 liter dye solution (resp. 1 liter amplifier solution in the case of excimer pumped dye laser), measured at 10 Hz. The following classification has been used in the case of excimer laser pumped dye lasers: -:  $\tau$  10-30 Wh, +:  $\tau$  30-50 Wh, ++:  $\tau$  > 50 Wh. In the case of Nd:YAG-pumped lasers: -:  $\tau$ < 50Wh, +:  $\tau$  50-100 Wh, ++:  $\tau$  100-300 Wh, +++:  $\tau$  > 300 Wh. The stability of dyes being continuously pumped is classified as: -:  $\tau$ < 100 Wh, +:  $\tau$  100-500Wh, ++:  $\tau$  > 500 Wh.

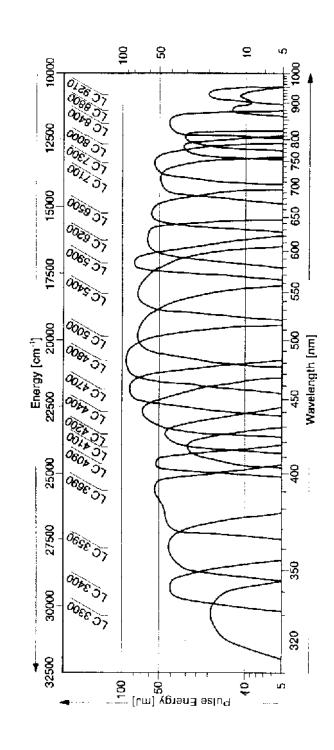
#### **General Remarks**

The output power of dye lasers is strongly dependent on the quality of the dye used. To overcome reduced quantum efficiency and instability due to impurities, Lambdachrome® laser dyes are examined by experienced chemist for their chemical and spectral properties and purified by specially developed techniques.

Lambda Physik reserves the right to modify any information given herein. Every effort is made to ensure utmost accuracy; no liability, however, is assumed for errors occurring. Nothing here is to be constructed as recommending any practice or any product in violation of any patent.

Cautious handling of dyes and dye solutions is advised, since the exact toxicity in most cases is not well known. The responsibility for the safe use of our laser dyes must rest in all cases with the user.

Dyes for EXCIMER LASER Pumped Dye Lasers



	Stability*		ı	+	++	++	++	++	++	+	+	++	ı	ı	+	+	+	+	ı	++	+	+			$\tau > 50 \text{ Wh}$
	Conc.	(g/I)	0.50	0.24	0.23	0.30	0.20	0.20	0.25	0.25	0.50	0.40	0.25	0.65	0.82	1.50	1.59	2.30	3.40	4.20	1.20	0.91	0.75	0.71	τ 30-50 Wh, ++:
	Solvent		K			Ճ	Ճ	EG		EG	ME	П		ME	ME	ME	ME	ME	ME	ME	ME	ME	ME	DMS0	
Char.	Energy	(m))	150	400	400	400	400	400	400	400	400	400	400	400	400	400	400	400	400	400	400	400	400	400	τ 10-30 Wh, +:
Pump Source	Wavelength	(nm)	248	308	308	308	308	308	308	308	308	308	308	308	308	308	308	308	308	308	308	308	308	308	as follows: -:
Effic.		(%)	4	∞	6	2	=	12	=	10	∞	7	Ξ	6	15	15	18	18	16	15	16	12	12	12	*Values are as fol
Tuning	Range	(nm)	312-343	332-360	346-377	356-385	368-402	363-408	367-405	386-418	388-426	386-420	399-415	412-443	423-462	432-475	440-484	460-510	479-553	522-600	269-608	588-644	614-672	632-690	
Peak		(nm)	334	343	360	363	330	381	388	397	333	336	406	425	441	448	456	480	200	540	581	009	623	658	
rome Dye			BM-Terphenyl	p-Terphenyl	DMQ	Butyl-PBD	dui	Polyphenyl 1	BiBUQ	Polyphenyl 2	Furan 2	PBB0	DPS	Stilbene 3	Coumarin 120	Coumarin 2	Coumarin 47	Coumarin 102	Coumarin 307	Coumarin 153	Rhodamine 6G	Rhodamine B	Rhodamine 101	DCM	
Lambdachrome												LC 4000													

# Dyes for EXCIMER LASER Pumped Dye Lasers (cont.)

Lambdach	ambdachrome Dye	Peak	Tuning	Effic.	Pump Source	Char.			
	•		Range		Wavelength	Energy	Solvent	Conc.	Stability*
		(mu)	(mu)	%)	(nn)	(Jm)		(g/I)	
C 7000	Rhodamine 700	723	701-768	11	308	400	ME	0.85	
LC 7100	Pyridine 1	710	092-029	10	308	400	DMS0	0.84	++
LC 7300	Pyridine 2	740	695-790	=	308	400	DMS0	0.72	+
LC 7710	Pyridine 4	771	744-812	7	308	400	DMS0	0.75	+++
0008 DT	Rhodamine 800	810	776-823	9	308	400	DWS0	1.00	+
LC 8400	Styryl 9	840	810-875	6	308	400	DMS0	1.10	
LC 8500	HITCI	898	837-905	4	308	400	DMS0	1.20	
LC 8810	Styryl 15	880	856-918	7	308	400	DMS0	1.15	
LC 9210	IR125	920	096-068	4	308	400	DIMS0	2.00	
LC 9450	Styryl 14	945	904-992	6	308	400	DMS0	1.10	
LC 9940	Styryl 20	994	970-1036	4	308	400	DMS0	1.10	
			*	Values are	"Values are as follows: -:	τ 10-30 Wh, +:	1 30-20 v	: 30-50 Wh, ++:	$\tau > 50 \text{ Wh}$

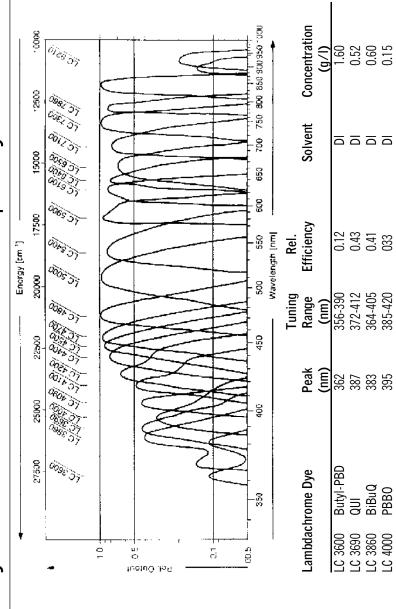
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	Stability**			+	+	ı	+	+	+	ı	ı	+	++	+ + +	+++	++++	++	++	++	++	++++	+ + +	$\tau > 300$ Wh
	Conc. Stak	(l/b)	0.25	0.50	0.26	0.25	0.25	0.30	0.40	0.70	0.70	2.36	0.64	0.22	0.10	0.22	0.27	0.50	0.30	0.50	0.36	0.22	Wh, +++:
	Solvent		ME	ME	ME	ME	ME	ME	ME	ME	ME	ME	ME	ME	ME	ME	ME	ME	<u>გ</u>	<u>გ</u>	PC	<u>S</u>	τ 100-300 Wh, +++
<u>ت</u>	Energy	(m)	200	200	200	200	200	200	200	200	200	200	200	200	200	200	200	200	200	200	200	200	0 Wh, ++:
Source Cha	Wavelength	(mu)	322	355	355	355	355	355	355	322	355	355	532	532	532	532	532	532	532	532	532	532	$=: \tau 50-100$
Pump Pump Source Char	geometry W		transv.			transv	transv.	transv	transv	transv			long.	long.	long.	long.	long.	long.	long.	long.	transv		: -: $\tau < 50$ ,
Effic.		%)	4	15	10	15	16	15	15	15	10	18	78	31	32	59	53	56	15	27	32	21	as follows:
Tuning	Range	(mm)	384-394	392-422	410-435	415-439	420-470	444-476	462-497	485-546	498-546	516-575	540-575	556-586	555-585	584-619	223-600	611-662	619-673	615-666	982-199	725-776	**Values are as follows:
Peak		(mm)	330	402	421	428	440	460	480	208	518	540	220	267	266	294	288	621	628	639	269	750	**
Lambdachrome Dye			Quinolon 390	Furan 2	Furan 1	Stilbene 3	Coumarin 120	Coumarin 47	Coumarin 102	Coumarin 307	Coumarin 500	Coumarin 153	Fluorescein 27	Rhodamine 19	Rhodamine 6G	Rhodamine B	Sulforhodamine B	Rhodamine 101	Sulforhodamine 101	DCM	Pyridine I	Pyridine 2	
Lambdacl			TC 3300	C 3990	LC 4260	LC 4200	LC 4400	LC 4700	LC 4800	C 5000	LC 5100	LC 5400	LC 5530	LC 5750	TC 2000	LC 6100	LC 6200	LC 6400	0099 JT	LC 6500	LC 7100	LC 7300	

Dyes for Nd:YAG LASER Pumped Dye Lasers (cont.)

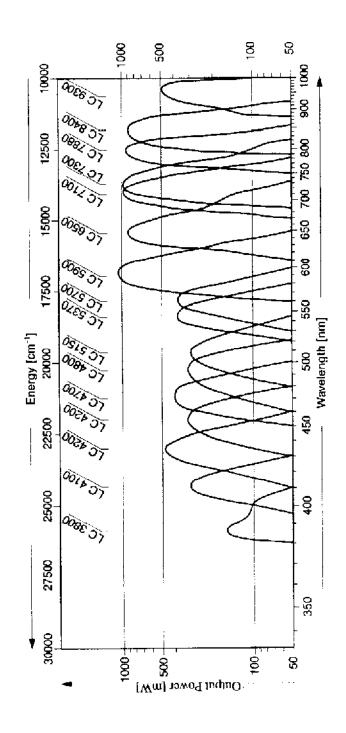
Lambdachrome	hrome Dye	Peak	Tuning	Effic.	Pump Pu	ump Source Cha	ت			
	•		Range		geometry	geometry Wavelength	Energy	Solvent	Conc. Stability**	billity**
		(nm)	(nm)	(%)		(nm)	(m)		(l/b)	
LC 7500	Styryl 6	721	708-735	16		532	200	PC	0.28	
TC 7600	Styryl 7	720	701-749	16	transv	532	200	ME	0.12	
LC 7550	Styryl 8	750	717-780	13	transv.	532	200	ME	0.15	
LC 8400	Styryl 9M	824	797-851	15	transv.	532	200	SC	0.26	
LC 8810	Styryl 15	880	856-918	7		532	200	2	0.62	
LC 9450	Styryl 14	945	904-990	6		532	200	2	0.27	
LC 9940	Styryl 20	994	970-1036 4	4		532	200	2	PC 0.68	++
		*	**Values are	as follows:		$\tau < 50$ . =: $\tau 50-100$ Wh. ++:	) Wh. ++:	τ 100-300 \	√h. +++:	

# Dyes for NITROGEN LASER Pumped Dye Lasers



	,		Tuning	Rel.		:
Lambdac	Lambdachrome Dye	Peak (nm)	Range (nm)	Efficiency	Solvent	Concentration (g/l)
LC 4090	DPS	404	394-416	0.43	Ы	0.12
LC 4100	Stilbene I	417	405-446	0.49	EG	0.20
LC 4200	Stilbene 3	424	408-457	99.0	ME	0.22
LC 4250	Bis-MSB	421	412-435	0.59	□	0.14
LC 4400	Coumarin 120	438	418-465	0.83	ME	0.25
LC 4500	Coumarin 2	444	426-475	0.94	ME	0.40
LC 4700	Coumarin 47	453	436-486	0.95	ME	99.0
LC 4800	Coumarin 102	470	454-506	1.00	ME	1.44
TC 2000	Coumarin 307	504	478-547	1.00	ME	1.60
LC 5400	Coumarin 153	537	517-590	0.87	ME	3.10
LC 5900	Rhodamine 6G	581	573-618	0.93	ME	1.63
LC 6100	Rhodamine B	622	600-646	0.91	ME	2.85
LC 6200	Sulforhodamine B	622	600-646	0.91	ME	2.85
LC 6400	Rhodamine 101	648	623-676	0.82	ME	2.36
LC 6500	DCM	629	626-703	69.0	DMSO	0.50
LC 7100	Pyridine I	703	675-750	0.78	DMISO	0.88
LC 7210	Oxazine 170	705	672-727	0.35	ME	0.79
LC 7300	Pyridine 2	743	710-790	1.00	DMSO	0.85
LC 7800	Methyl-DOTC	780	768-820	98.0	DMSO	0.51
TC 7880	DOTC/HITC	823	794-867	0.74	DMSO	1.23/0.03
LC 8400	Styryl 9	840	803-875	1.00	DMSO	1.03
C 8760	DTTC/IR 144	871	829-886	0.18	DMSO	0.65/2.52
C 8800	IR 144/IR 125	887	872-935	0.14	DMSO	2.52/1.94
LC 9210	IR 125	918	863-628	0.21	DMISO	1.94
LC 9301	IR 140	910	896-006	0.11	DIMSO	0.78

Dyes for ION LASER Pumped Dye Lasers



Lambdac	-ambdachrome Dye	Peak	Tuning	Pump source char.	Power	SolCent	Conc. Stability***	bility***
			Range	Wavelength				
		(mm)	(mu)	(mn)	8		(l/b)	
LC 3810	Polyphenyl 2	384	370-406	Ar+, UV, 300-336nm	2.0	EG	2.0	+
LC 4100	Stilbene 1	415	403-428	Ar+, all lines UV	3.0	EG	0.75	+
LC 4200	Stilbene 3	435	410-485	Ar+, all lines UV	2.0	EG	1.0	+
LC 4800	Coumarin 102	482	463-515	Ar+, UV, 350-386nm	3.0	BZ/EG	2.0	
LC 5370	Coumarin. 6	535	510-550	Ar+, 488 nm	0.9	BZ	2.0	+
C 5700	Rhodamine 110	220	530-580	Ar+, 514.5 nm	0.9	EG	0.75	+
LC 5900	Rhodamine 6G	575	560-625	Ar+, 514.5 nm	0.9	EG	0.75	++
LC 6200	Sulforhodamine B	625	298-650	Ar+, 514.5 nm	0.9	EG	2.5	++
TC 6500	DCM Special	645	610-695	Ar+, 514.5 nm	0.9	BZ/EG	2.0	+
C 7000	Rhodamine 700	740	982-069	Kr+, all lines red	4.6	EG	1.0	++
LC 7300	Pyridine 2	720	675-783	Ar+, 514.5 nm	7.5	PC/EG	1.5	++
LC 8400	Styryl 9	830	282-900	Ar+, 514.5 nm	0.9	PC/EG	2.0	+

 $\tau < \! 100$  Wh, +:  $\tau$  100-500 Wh, ++:  $\tau > \! 500$  Wh

\*\*\*Vaules are as follows: -:

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