



# Lambdachrome® Laser Dyes

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Lambdachrome®  
**Laser Dyes**

3rd Edition

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All laser dyes mentioned in this book are immediately available from Lambda Physik. For more information please call or write. All major credit cards accepted.

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Third Edition (January 2000)

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# Contents

Preface to the 3rd Edition .....	ix
Preface to the 2nd Edition .....	xi
Preface to the First Edition .....	xiii

<b>Dye Lasers and Laser Dyes .....</b>	<b>1</b>
--	----------

<b>Dye Solvents .....</b>	<b>9</b>
Benzyl Alcohol .....	17
Cyclohexane .....	18
Dichloroethane .....	19
Dichloromethane .....	20
Dioxane .....	21
DMSO .....	23
Ethanol .....	24
Ethylene Glycol .....	26
Hexane .....	27
Methanol .....	28
Toluene .....	30

<b>Laser Dyes .....</b>	<b>32</b>
BM-Terphenyl (LC 3300) .....	34
p-Terphenyl (LC 3400) .....	36
TMQ (LC 3500) .....	38
BMQ (LC 3570) .....	40
DMQ (LC 3590) .....	42
Butyl-PBD (LC 3600) .....	44
PBD (LC 3640) .....	46
TMI (LC 3650) .....	48
QUI (LC 3690) .....	50
PPO (LC 3700) .....	52
PPF (LC 3720) .....	54
p-Quaterphenyl (LC 3740) .....	56
BBD (LC 3780) .....	58
Polyphenyl 1 (LC 3800) .....	60
Polyphenyl 2 (LC 3810) .....	62
BiBuQ (LC 3860) .....	64
Quinolon 390 (LC 3900) .....	66
TBS (LC 3930) .....	68

$\alpha$ -NPO (LC 3950) .....	70
Furan 2 (LC 3990) .....	72
PBBO (LC 4000) .....	74
DPS (LC 4090) .....	76
Stilbene 1 (LC 4100) .....	78
BBO (LC 4150) .....	80
Stilbene 3 (LC 4200) .....	82
Carbostyryl 7 (LC 4220) .....	84
POPOP (LC 4230) .....	86
Coumarin 4 (LC 4240) .....	88
Bis-MSB (LC 4250) .....	90
Furan 1 (LC 4260) .....	92
Carbostyryl 3 (LC 4350) .....	94
Coumarin 120 (LC 4400) .....	96
Coumarin 2 (LC 4500) .....	98
DASPI (LC 4650) .....	100
Coumarin 466 (LC 4660) .....	102
Coumarin 47 (LC 4700) .....	104
Coumarin 102 (LC 4800) .....	106
Coumarin 152A (LC 4810) .....	108
Coumarin 152 (LC 4850) .....	110
Coumarin 151 (LC 4900) .....	112
Coumarin 6H (LC 4910) .....	114
Coumarin 307 (LC 5000) .....	116
Coumarin 500 (LC 5010) .....	118
Coumarin 314 (LC 5040) .....	120
Coumarin 510 (LC 5100) .....	122
Coumarin 30 (LC 5150) .....	124
Coumarin 334 (LC 5210) .....	126
Coumarin 522 (LC 5220) .....	128
DASBTI (LC 5280) .....	130
Coumarin 7 (LC 5350) .....	132
Brillantsulfaflavin (LC 5360) .....	134
Coumarin 6 (LC 5370) .....	136
Coumarin 153 (LC 5400) .....	138
DOCI (LC 5410) .....	140
Pyrromethene 546 (LC 5450) .....	142
DMETCI (LC 5460) .....	144
Uranin (LC 5520) .....	146
Fluorescein 27 (LC 5530) .....	148
Fluorol 7GA (LC 5550) .....	150
Pyrromethene 556 (LC 5560) .....	152

Pyrromethene 567 (LC 5670) .....	154
Rhodamine 110 (LC 5700) .....	156
Rhodamine 19 (LC 5750) .....	158
Pyrromethene 580 (LC 5805) .....	160
Rhodamine 6G (LC 5900) .....	162
DQOCI (LC 5920) .....	166
DCI-2 (LC 5950) .....	168
Pyrromethene 597 (LC 5970) .....	170
Rhodamine B (LC 6100) .....	172
Sulforhodamine B (LC 6200) .....	174
Malachit Green (LC 6220) .....	176
DTCI (LC 6250) .....	178
DQTCI (LC 6290) .....	180
Rhodamine 101 (LC 6400) .....	182
DCM (LC 6500) .....	184
Pyrromethene 650 (LC 6505) .....	186
DODCI (LC 6550) .....	188
Sulforhodamine 101 (LC 6600) .....	190
Cresyl Violet (LC 6700) .....	192
Phenoxazone 9 (LC 6750) .....	194
Nile Blue (LC 6900) .....	196
Oxazine 4 (LC 6950) .....	198
Rhodamine 700 (LC 7000) .....	200
Pyridine 1 (LC 7100) .....	202
Carbazine 122 (LC 7200) .....	204
Oxazine 170 (LC 7210) .....	206
Oxazine 1 (LC 7250) .....	208
DTDCI (LC 7260) .....	210
Oxazine 750 (LC 7270) .....	212
Pyridine 2 (LC 7300) .....	214
HIDCI (LC 7400) .....	216
Cryptocyanine (LC 7450) .....	218
Styryl 6 (LC 7500) .....	220
Styryl 8 (LC 7550) .....	222
DDI (LC 7700) .....	224
Pyridine 4 (LC 7710) .....	226
Methyl-DOTCI (LC 7800) .....	228
Styryl 11 (LC 7950) .....	230
Rhodamine 800 (LC 8000) .....	232
Styryl 9M (LC 8400) .....	234
HITCI (LC 8500) .....	238
IR 125 (LC 8630) .....	240

DTTCI (LC 8760) .....	242
IR 144 (LC 8800) .....	244
Styryl 15 (LC 8810) .....	246
DNTTCI (LC 8850) .....	248
HDITCI (LC 9200) .....	250
DDTTCI (LC 9280) .....	252
DDCI-4 (LC 9300) .....	254
IR 140 (LC 9310) .....	256
Styryl 14 (LC 9450) .....	258
IR 132 (LC 9500) .....	260
Styryl 20 (LC 9940) .....	262
IR 26 (LC 1080) .....	264
IR 5 (LC 1090) .....	266
Saturable Absorbers .....	268
Reference List .....	271

<b>Tuning Curves .....</b>	<b>274</b>
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# Preface to the 3rd Edition

Dyes and solvents from Lambda Physik periodically change in availability and composition. In order to ensure that researchers and users of Lambda Physik products have the most up-to-date information possible, this 3rd Edition of Lambdachrome® Laser Dyes is provided as an update to previous editions.

Lambda Physik reserves the right to modify any information given herein. Every effort is made to ensure utmost accuracy; however, no liability is assumed for errors occurring in the tables.

January 2000



# Preface to the 2nd Edition

The second edition of *Lambdachrome® Laser Dyes* has been updated with additional dyes, dye laser measurements, and evaluation of the latest literature. In this way it should be possible for any dye laser user to get a quick overview of the most important information regarding the use of laser dyes.

Meanwhile, the standard dye laser technique has been completed by several solid state lasers showing very interesting features such as high operating lifetime of the active medium and high output power in a small IR spectral range (compared to the accessible fundamental tuning range of an excimer laser pumped dye). However, the use of laser dyes still guarantees:

- easy tunability over a *wide range* of frequencies or wavelengths without the need of changing gratings or mirrors
- hyperfine tuning
- high average power in pulsed and cw operation mode
- high peak power
- ideal light source for the generation of ultrashort pulses

February 1994

# Preface to the 1st Edition

The intention of this book is to give information on the most frequently asked questions about commercially available dyes, their chemical formulas, lasing wavelengths, solvents, pump sources, performance, and literature references.

The following topics are discussed:

- Solvents frequently used for laser dyes and their general and physical properties, safety precautions, and waste disposal procedures
- General considerations about the choice of a solvent for laser dyes, solvent handling, preparation of dye solutions, and the cleaning of the dye circulation system
- Precautions for the use of laser dyes and their solutions
- All Lambdachrome® laser dyes are listed, including more than one hundred dyes along with their chemical names, chemical structures, trivial names, and trade names
- Absorption and fluorescence data in addition to appearance
- The absorption spectrum of each dye
- The application of the dyes in a variety of solvents under the most common pumping conditions as well as the dye laser characteristics (peak, range, efficiency, concentration, and solvent)
- Tuning curves for excimer, nitrogen, and ion laser pumped dye lasers
- Several hundred literature references

This book is written for general users of dye lasers. Researchers of laser dyes may refer to the mentioned publications for more detailed information.

July 1985

# Dye Lasers and Laser Dyes

Dye lasers are "the fulfillment of an experimenter's pipe dream that was as old as the laser itself: To have a laser that is easily tunable over a wide range of frequencies or wavelengths" (Schaefer, 1977). Dye lasers can be pumped by incoherent or laser sources, both pulsed and continuous wave (CW), and offer the possibilities of broad wavelength control, multijoule pulsed operation, ultranarrow linewidths, or ultrashort pulses. They are conveniently divided into three broad technological categories:

- *Continuous-wave* jet-streamed dye lasers can provide narrow CW bandwidths and can be synchronously pumped or passively modelocked to generate short pulses.
- *Flashlamp*-pumped dye lasers have a larger bandwidth and less wavelength stability than CW laser-pumped dye lasers, but they have the advantage that large volumes of active dye medium can be pumped, yielding large output pulse energies and active powers.
- Dye lasers that are pumped by Cu vapor or nitrogen, excimer lasers, and the frequency-doubled or -tripled output from pulsed Nd:YAG lasers provide high peak powers. Also, the pulse duration of 4 to 60 nsec makes narrow bandwidths and a high spectral purity possible.

Typical output characteristics from commercial dye lasers are shown in Table 1.

## Continuous-Wave Dye Lasers

CW laser systems consist of three major elements: the optical resonator, the dye flow system, and the tuning element. The resonator is responsible for maintaining a rigid optical alignment of all cavity components. The dye

**Table 1.**  
**Commercial Dye Laser Output Characteristics**

Pump Source	Argon/Krypton CW	Ion Laser Mode-Locked/ Cavity Dumped	FlashlampNd:YAG Laser	Excimer Laser	
Tuning Range [nm]	380-950	580-880	335-850	410-880	320-1,024
Average Power [W]	5	0.1	3	2	10
Repetition Rate [Hz]	CW	3.8M	2-30	10-40	1-500
Peak Power [kW]	-	10	7,000	20,000	
10,000					
Energy/pulse [mJ]	-	0.01	3,500	10-120	40-120
Pulsewidth [nsec]	-	0.0001	260-600	5-10	7-250
Linewidth [GHz]	0.0005 to 40	Various	2	0.6	0.15

flow system, consisting of dye, circulation pump, and dye jet nozzle, must provide an optically flat stream of dye across the laser's optical axis. The tuning element allows the user to continuously tune to the required output wavelength.

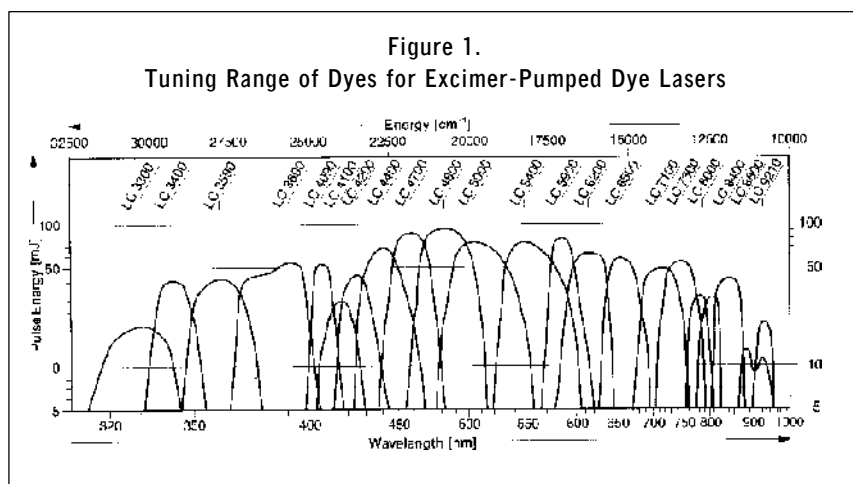
Excitation in a CW dye laser is provided by an input pump beam from another CW light source, typically an ion laser. This beam is focused onto the dye stream and causes an extremely high level of fluorescence. The fluorescence is focused between two highly reflective concave mirrors that feed back the fluorescent light, initiating the lasing process.

The dye laser emission described above is broadband, typically exhibiting a 40-GHz linewidth. Many applications require narrow-linewidth single-frequency operation. This is accomplished by inserting one or two etalons into the dye laser cavity. A dye laser with an etalon, when coupled with electronic stabilization, can narrow the linewidth to less than 1 MHz.

## Pulsed Dye Lasers

Of the various pulsed dye lasers, two types tend to dominate because of their versatility, broad spectral coverage and high output power. They are Nd:YAG and excimer pumped dye lasers.

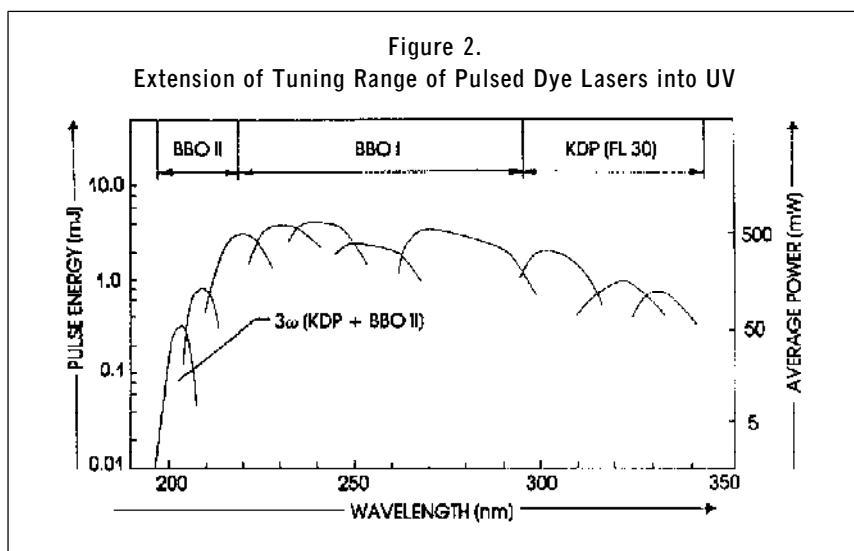
The use of Nd:YAG and excimer pump lasers allows dye laser pulses to be produced at energy levels up to 100 mJ or more. Their performance is wavelength-dependent. At the long wavelengths, more energy is provided by pumping with a frequency-doubled Q-switched Nd:YAG than with an excimer laser. However, UV-emitting excimer pumps allow dye lasers to operate directly at UV wavelengths down to 308 nm and at repetition rates of 500



Hz. The strongest Nd:YAG lasing line is at 1064 nm, which is unsuitable to pump dyes. Fortunately, its high peak power and near-diffraction-limited beam quality lend themselves to rather efficient frequency doubling, tripling and quadrupling. This provides suitable pump wavelengths of 532, 355, and 266 nm. Considerable engineering effort has gone into reducing the intensity fluctuations that are amplified by the inherent nonlinear frequency-doubling process. The tuning range can be covered by using one of these pump wavelengths and an appropriate dye.

The Nd:YAG laser has certain advantages with respect to the efficient pumping with certain dyes, especially when pumped longitudinally, i.e., collinearly to the laser beam. Rhodamine 6G, for example, can be excited very efficiently in this manner (>50 percent). Since excellent beam quality is achieved with the longitudinal arrangement, one also gets high conversion rates with all nonlinear frequency changes (by frequency mixing, 367 nm, and by frequency doubling, 280 nm).

The discovery of rare gas halide lasers, i.e., the excimer laser, in 1976, introduced a new and powerful tool for dye laser pumping. It combines the advantages of the nitrogen system (high repetition rate, ease of operation, and low cost) with those of the Nd:YAG system (high peak power and sufficiently long pulse duration). Furthermore, the excimer laser is scalable to very high repetition rates (>1 kHz) and high pulse energies (>1 J).

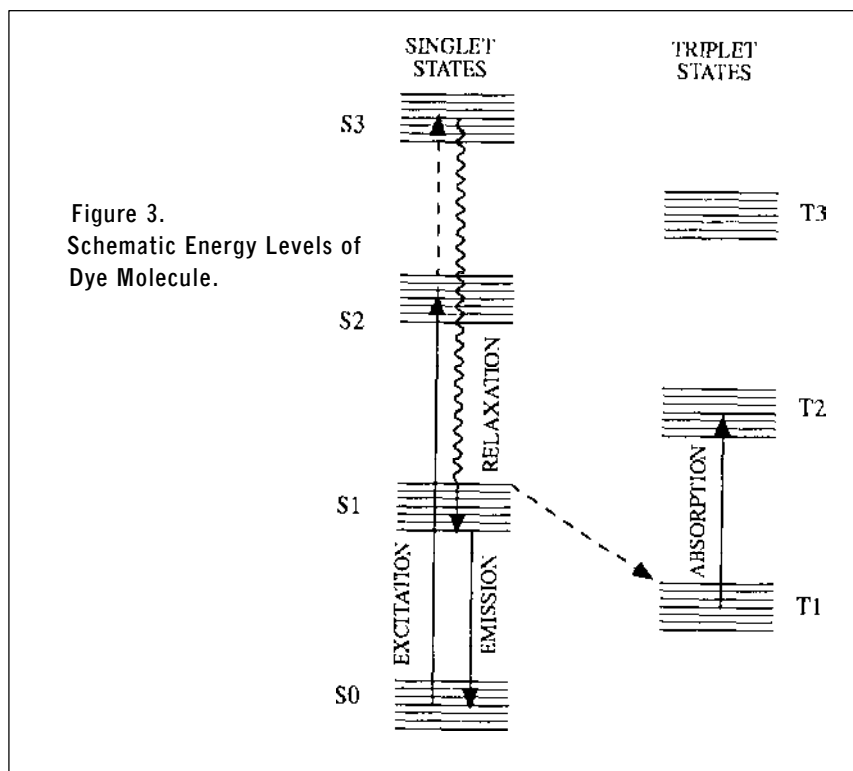


Excimer laser wavelengths range from 193 nm (ArF) to 351 nm (XeF). The strongest lines are at 248 nm (KrF) and 308 nm (XeCl). Most dyes have an absorption band at 308 nm. The result is that nearly the entire range of commercially available dyes can be pumped with a XeCl laser (see Figure 1).

A major advantage of pulsed dye lasers over CW is the ease with which the output can be frequency-doubled, thus extending the tuning range into the UV. Using a combination of KDP and BBO (beta barium borate) crystals, wavelengths from 207 to 350 nm can be produced. Further extension down to 197 nm can be achieved by frequency mixing (see Figure 2).

## Dyes

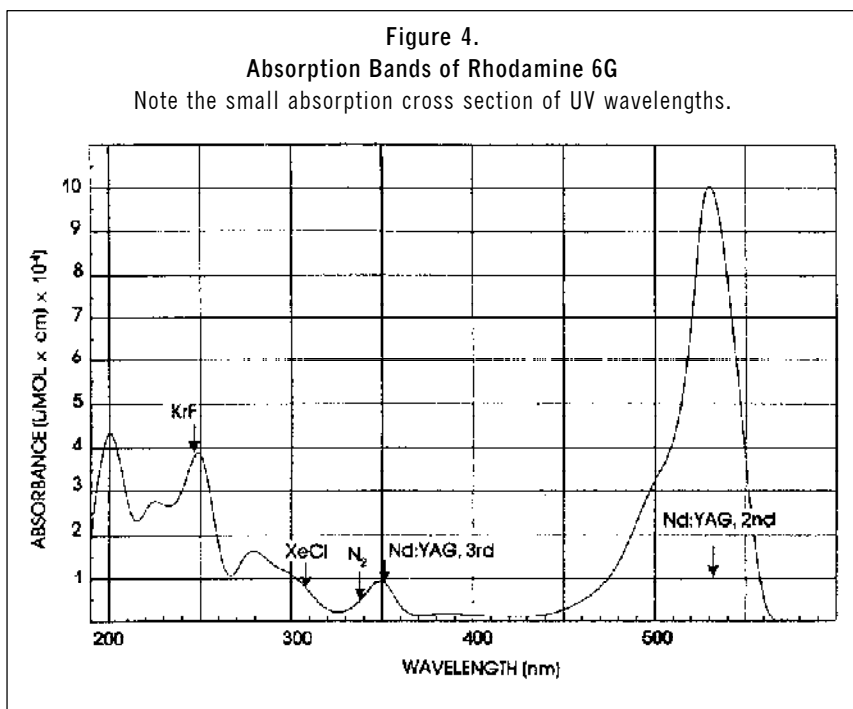
Dyes, either as solutions or vapors, are the active medium in pulsed and CW dye lasers as well as ultrafast shutters for Q-switching and passive modelocking. They emit in a comparatively narrow spectral region (typically 30 nm); thus a variety of dyes is necessary in order to cover the entire (visible) spectral range (Figure 1).



The optical excitation of dyes corresponds to transitions of molecules in the singlet state, with the absorption  $S_0 \rightarrow S_1$  being the strongest (see Figure 3), and is specific for each dye molecule. For optimum pumping ( $S_0 \rightarrow S_1$ ) of the various dyes, one would therefore need a number of pump-laser wavelengths.

Fortunately, nearly all dyes have additional absorption bands in the UV range (see Figure 4). These absorptions correspond to transitions to higher singlet states (Figure 3) from which fast internal relaxation processes lead to the upper laser level ( $S_1$ ) with high quantum efficiency. This is the reason most dyes can be pumped by a single UV laser. However, the attractive excitation scheme of one pump laser for all dyes brings other problems:

- (a) The inner efficiency of dye lasers is lower as a result of excitation in higher S-states because a considerable part of the excitation energy is converted into heat (large Stokes shift). However, the high efficiency of pulsed lasers more than compensates for this disadvantage.



- (b) A multiphoton excitation can lead to destruction of the cell and the solvent molecules. In this process, a previously excited molecule absorbs additional photons (sequential absorption), or a molecule absorbs several photons at the same time. In these absorption processes, the molecule can absorb so much energy that the binding energy is surpassed, and the molecule dissociates, or at least changes, its structure. This process is more probable during excitation with UV light than with visible light. Thus, one must expect a reduced photostability of the dye when pumping with UV light.

The photostability of the most common dyes is summarized in Table 2 (Antonov and Hohla, 1983). For example, the value of 50 Wh/l corresponds to  $5 \times 10^5$  total shots as the point at which the dye laser energy falls to 50 percent of the initial value, i.e., when the dye solution (one liter) is excited with 360 mJ per pulse at 10 Hz.

- (c) Another problem results from the small absorption cross section at short wavelengths (Figure 4). To excite as many molecules as possible, a very high pump power density  $I_p$  ( $I_p$  being inversely proportional to the absorption cross section), or high dye concentrations is required.  $I_p$  is limited to values  $<30 \text{ MW/cm}^2$  due to the stability of most solvents. To

**Table 2.**  
**Photochemical Stability of Laser Dyes**

Dye	Center of Emission [nm]	Excimer-pumped [Wh]	CW laser-pumped [Wh]
p-Terphenyl	340	451	-
QUI	380	1457	-
Polyphenyl 1	380	870	-
Stilbene 1	410	10	200
Stilbene 3	430	14	300
Coumarin 2	450	31	100
Coumarin 47	470	45	-
Coumarin 102	480	244	100
Coumarin 30	515	-	100
Rhodamine 6G	590	316	1000
Rhodamine B	610	144	200
DCM	650	348	500
Rhodamine 700	700	80	1000
Styryl 9	840	73	500
HITCI	875	12	100
IR 140	960	10	100



keep  $I_p$  as low as possible, high-power dye lasers have to be pumped transversely.

Transverse pumping configurations have been used for some time, and in contrast to the end-on pumped configurations, they lead to a non-Gaussian energy distribution, for which rising spatial filters can compensate.

There are two types of transverse-pumped dye laser cells: the longitudinal and the transverse flowing. The transverse configuration, in which the dye flows vertically to the dye laser, results in a high repetition rate, whereas longitudinal flowing, in which dye flows in the direction of the dye laser, is characterized by a symmetrical energy distribution but small repetition rates.

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# Dye Solvents

The heart of any dye laser is the laser dye. Either dissolved in an appropriate solvent or in a vapor state, these very often highly colored substances play the major role in the overall performance of any dye laser. Both pulsed and continuous operation is possible. In addition, their unique photophysical properties make them ideal candidates for the generation of ultrashort light pulses. With mode-locking, pulses shorter than 100 femtoseconds have been obtained. Hyperfine tuning of the output has permitted many exciting experiments in spectroscopy.

## Solvents for Laser Dyes

Prepared laser dye solutions usually contain very small quantities of dye. Typical dye concentrations are  $10^2$  to  $10^5$  molar. For this reason, the solvent in which the dye is dissolved plays an important role when defining physical properties and potential hazards.

Lasing wavelength and energy are very sensitive to the choice of solvent. Most laser dyes are polar molecules, and excitation into their lowest-lying singlet state is accompanied by an increase in the dipole moment. Accordingly, solvent polarity plays an important role in shifting the lasing wavelength. In a majority of circumstances, increasing solvent polarity will shift the gain curve toward longer wavelength. In the case of more polar dyes, the shift can be as high as 20-60 nm. Table 3 gives an impression of this effect (Drexhage, 1973).

Some solvents cannot be used with longer wavelength lasing dyes because the solvents have vibrational overtones that interfere with the lasing process. Solvents such as water, methanol and ethanol, which would appear to be optimal for many dyes, are often not useful solvents for near-IR and IR dyes

**Table 3.**  
**Maximum of Main Absorption Band in Different Solvents**

RHODAMINE 6G		COUMARIN 102	
Solvent	Labs [nm]	Solvent	Labs [nm]
HFIP	514	HFIP	418
TFE	516	TFE	405
EtOH	530	MeOH	390
DPA	537	NMP	383
DMSO	540		

because of the presence of hydroxyl group overtones in this spectral range. Accordingly, the solvent DMSO or polychlorinated aromatics, which lack hydroxyl resonances, are commonly used for dyes that lase in the near-IR and IR regions of the spectrum. Unfortunately these solvents are toxic. DMSO especially facilitates the transfer of toxic dyes through the skin and into the body if accidentally spilled. Therefore, it is strongly recommended that all laser dyes and solutions containing laser dyes are handled in well ventilated environments. All individuals handling the solutions should wear rubber gloves.

A summary is given below of the general and physical properties, application, and safety of the most frequently used solvents for laser dyes. Most of this information has been taken from the literature cited at the end of this article.

## Choice of Solvent

Although very often a specific solvent is recommended for use with a particular dye, it is important to recognize that other solvents can also be used, particularly if the user is interested in shifting the gain curve to different wavelengths.

The following are criteria for choosing appropriate solvents for laser dyes:

- a) The solvent must be transparent at the pump wavelength and the emission wavelength of the dye laser.
- b) The dye should be soluble in the solvent under consideration. In all cases, the rule "similia similibus solvuntur" applies, meaning that the solubility of nonpolar dyes, e.g. PTP, in nonpolar solvents, e.g. cyclohexane, is greater than in polar solvents, e.g. methanol, and vice versa.
- c) The solvent must be photochemically stable when exposed to the pump light. In particular, solvents containing chlorine, such as chloroform, and secondary alcohols, such as isopropanol, are not useful as solvents for laser dyes because of their low photochemical stability.
- d) Some solvents are often not useful for near-IR and IR dyes due to the presence of hydroxyl group overtones in this spectral region (see above).

The following solvents are recommended for different pumping conditions:

- a) Those appropriate for pumping with a pump source emitting below 300 nm (e.g., KrF: 248 nm, Nd:YAG 4th harmonic: 266 nm):
  - Cyclohexane
  - Ethylene glycol
  - Glycerol
  - Trifluoroethanol
  - p-Dioxane

- Ethanol
  - Methanol
  - Hexafluoroisopropanol
- b) Solvents appropriate for pumping with a pump source emitting between 300 and 400 nm (e.g., XeCl: 308 nm, nitrogen: 337 nm, Nd:YAG 3rd harmonic: 355 nm), in addition to those listed in (a):
- N,N-Dipropylacetamide
  - Dimethylsulfoxide (DMSO)
  - N,N-Dimethylformamide (DMF)
  - Tetrahydrofurane
  - 1-Methyl-2-pyrrolidinone (NMP)
  - Tetrahydrothiophenedioxide (sulfolane)
- c) Appropriate solvents for pumping in the visible and near-IR spectral range (e.g., Nd:YAG 2nd harmonic: 532 nm, CW-ion lasers), in addition to those listed in (a) and (b):
- Toluene
  - Chlorobenzene
  - Chloroform
  - Benzylalcohol
  - 1,2-Dichloroethane
  - 1,1,1-Trichloroethane
  - o-Dichlorobenzene
  - Dichloromethane

## Purity of Solvent

The output power of dye lasers is strongly dependent on the purity of the solvent. Impurities and additives may strongly affect upper state lifetime of the dye or may catalyse photochemical reactions. Therefore, for best results, only high quality solvents are to be recommended.

Very often we are asked whether it is necessary to use spectroscopic grade solvents. Our experience is that it is NOT necessary. Commercially available qualities of the grade "pro analysis" (p.a.) or "for synthesis" are sufficient for dye laser applications. Some suppliers will call them HPLC.

However, bulk quantities of these solvents are very often of poor quality and are not offered as p.a. grade. Under such circumstances, it is highly recommended that the transparency of the solvent is checked in a 1 cm cuvette at the pump wavelength of the dye laser with a simple absorption spectrometer. The measured value should be greater than 98 percent.

## Solvent Handling

*With the exception of water, all solvents should be considered hazardous.* In many instances, the solvent in which the dye is dissolved plays a major role in the hazard presented by the final solution. Some of the solvents listed above are highly toxic, irritants, narcotics, and/or anaesthetics. These hazards must be addressed carefully in dye handling and solution preparation.

Nearly all solvents are highly flammable. Therefore, a small fire extinguisher should be installed near the laser in a readily accessible and unobstructed area.

A particular fire hazard that is not commonly known occurs with nonpolar and, hence, nonconductive solvents. If these solvents are circulated at a high speed through plastic tubings, the pump unit acts as a van de Graff generator, producing up to 100 kV, and sparks may pierce the tubing and ignite the solvent. The dye selectors use grounding wires inside the plastic tubings to eliminate these problems. However, when using such solvents, check first for static electricity before opening the reservoir. Static electricity is present when hair on the back of your hand or forearm is attracted to the plastic tubing. Do not circulate dye solutions made with such solvents for more than a minute, unless the cuvette has been placed into the crate and is grounded.

## Preparation of the Dye Solution

As a rule of thumb, the dye concentration is selected to absorb 90 percent of the pump light within 0.5 mm, or the dye solution has to have an absorbance of  $OD = 2/\text{mm}$  for the wavelength of the pump light.

When in doubt about the concentration of the dye solution, measure the absorbance of the dye solution used for the oscillator in a spectrophotometer, using either a 1 mm cuvette or, after diluting by a factor of 10, a 10 mm standard cuvette. The cuvettes must be of fused-silica type if you measure the absorbance for a pump wavelength smaller than 300 nm. The absorbance of your oscillator dye solution should be within the range of  $OD = 1.8$  to  $2.2/\text{mm}$ . For an unknown dye, dissolve a known amount (a few milligrams) in a known volume and measure the absorbance at the pump wavelength in use; the dye concentration (in g/l) is then calculated according Beer's law:

$$\text{g/l} = \frac{20}{OD(\text{solution}/10\text{ mm})} \times \frac{\text{g}(\text{sample})}{\text{Vol}(\text{sample})}$$

The measurements should, of course, be made against pure solvent in a cuvette of the same pathlength in the reference beam of the spectrophotometer.

A concentration of  $OD = 2/\text{mm}$  is convenient for the oscillator solution. The amplifier needs only  $1/3$  this concentration. The amplifier solution is prepared by filling the reservoir bottle of the amplifier circulator with one part stock solution and then adding two parts pure solvent.

To prepare the dye solution, weigh out the amount of dye and transfer it into a 500 ml (or 1 liter) glass bottle. If some are available, use brown bottles. Make sure that the entire measure of dye is transferred to the bottle and be careful not to spill it. Most errors occur at this step. Fill the bottle to the 500 ml (or 1 liter) mark. Adding 10 percent more or less solvent does not affect the operation appreciably. Make the stock solution "fatter" than recommended, since solvent can always be added directly to the reservoir of the circulators later on. Some dyes do not dissolve instantly. Use of an ultrasonic bath is recommended.

**NOTICE:**

**Do not use the dye solution until it is completely translucent and no floating dye particles are observed.**

Some dyes have to be used close to their saturation level. It may take some dyes up to 30 minutes to dissolve completely. If necessary, heat gently and with caution.

## Optimizing the Dye Solution

There is an optimum concentration for a given dye gain, i.e. for a given dye, wavelength, and input power. This optimum concentration may vary from that giving an optical density of  $OD = 2/\text{mm}$ . Higher concentrations will cause a slightly red shift the tuning curves, while lower concentrations will result in a blue shift.

Optimization of the dye concentration is accomplished by adding either pure solvent or solution of higher concentration than that recommended in small increments to the solution in the dye circulation system until power is at maximum.

The concentrations for the amplifier solutions of excimer and Nd:YAG-pumped dye lasers are about  $1/3$  and  $1/6$  of the oscillator solutions, respectively.

## Cleaning the Dye Circulation System

Should it be necessary to change the dye solution, it is only necessary to rinse three times with 100 to 200 ml of solvent. It is recommended that the dye filter be changed. If it is not possible to change the dye selector when changing the dye, especially when going from a longer wavelength to a shorter one, the cuvette, the reservoir, the pump, and the tubings must be carefully cleaned and

the filter must be exchanged. The cleaning process is complete only when no fluorescence is observed after rinsing for approximately 15 minutes. It is advisable to use small quantities of solvent and few rinses. The rinsing process can take many hours. Low output power on changing the dyes can often be traced to dye residue in the pump and the tubings.

## Precautions for the Use of Laser Dyes

Cautious handling of dyes and dye solutions is advised, especially because those used in the infrared may be toxic or because solvents such as DMSO, Methanol, Dioxane, and Benzyl Alcohol have the ability to carry their solutes through the skin and into the body (Mosovsky, 1983; Kues, 1975).

In most cases the exact toxicity of laser dyes is not well known, but they should, like all chemicals, be considered dangerous until proven otherwise. The safest precaution is to use butyl rubber gloves when handling the dye solutions and to immediately clean any skin that comes into contact with the dye solutions or the dye itself.

Solvents should be kept away from heat, sparks, and open flames because they are extremely flammable or combustible. They should be handled in a hood due to their stench or potential danger if inhaled.

Lambda Physik provides, on request, a complete set of Material Safety Data Sheets (MSDS). These data sheets give more information on laser dye toxicity, hazards, and recommended controls. As already mentioned above, the exact toxicity of laser dyes is not well known in most cases. Therefore, it is important to know that the MSDS only describes general aspects of dye toxicity.

### **NOTICE:**

**The responsibility for the safe use of our Lambdachrome laser dyes rests with the user.**

## Hazards

Solvents should be handled only by qualified people trained in laboratory procedures and familiar with their potential hazards. Some solvents are highly toxic, irritants, narcotics, and/or anaesthetics. Some form hazardous compounds upon decomposition; others are highly reactive. In the following tables, hazard warnings and literature references, such as Sax (Dangerous Properties of Industrial Materials) and the Registry of Toxic Effects of Chemical Substances (RTECS) are provided, so that the information about possible hazards are available to the trained technical person using the dye, solvent, and/or dye solution.



The absence of a warning must not be interpreted as an indication of safety. In several cases information is not available on the possible hazards of many compounds.

## **Waste Disposal Procedures**

The disposal methods outlined below are intended as guides to the users of laser dye solutions or solvents. Careful consideration must be given to the chemical and physical properties of the substances. In addition, local laws and regulations may preclude the use of these methods which are primarily designed for quantities of one to five liters. All federal, state, and local laws concerning health and pollution must be observed.

## **Definitions**

### **Boiling point**

The temperature at which the vapor pressure of the liquid is equal to the opposing pressure. Values listed in the tables refer to an opposing pressure of 760 torr unless otherwise stated.

### **Density**

The density of a substance is defined as the mass per unit of volume.

### **Dielectric constant**

A measure of the relative effect a solvent has on the electronic force with which two oppositely charged plates are attracted to each other.

### **Flash point**

The flash point is usually not considered a common physical property. It is included because of its widespread use in classifying solvents for storage and shipping.

### **Ionization potential**

The work (expressed in electron volts) required to remove a given electron from its atomic orbit and place it at rest at an infinite distance.

### **Melting point**

The temperature at which a solid compound changes into the liquid state.

### **Minimum ignition temperature**

The minimum temperature at which, under certain conditions, the mixture combined with air may ignite.

**Molecular weight**

The sum of the atomic weights of all atoms in a molecule.

**Refractive index**

The the ratio of the velocity of light in a particular substance to the velocity of light in vacuum. Values usually reported refer to the ratio of the velocity in air to that in the substance saturated with air.

**Threshold Limit Value (TLV)**

The maximum permissible concentration of a chemical that is permissible for prolonged exposure. The TLV gives a concentration of vapors to which an average sized person can safely be exposed for 8 hours per day, 5 days per week.

**Viscosity**

The coefficient of viscosity is defined as the force per unit area necessary to maintain a unit velocity gradient between two parallel planes a unit distance apart.

# Benzyl Alcohol

Phenylcarbinol

## General Properties

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Colorless liquid with aromatic odor. It is only slightly soluble in water.

## Physical Properties

Molecular weight:	108.14						
Freezing point (°C):	-15.3						
Boiling point (°C):	205.45						
Flash point (°C):	100						
Min. ignition temp. (°C):	435						
Density (g/cm <sup>3</sup> ):	1.0493 <sup>15</sup>	1.0413 <sup>25</sup>	1.03765 <sup>30</sup>				
Refractive Index:	1.54035 <sup>20</sup>	1.453837 <sup>25</sup>					
Viscosity (cPoise):	7.76 <sup>15</sup>	4.65 <sup>35</sup>					
Dielectric constant:	13.1						
Ionization potential (eV):	9.14						
Solubility:	0.08 % in water, organic solvents						
Optical properties:							
Wavelength (nm)	290	300	310	320	340	350	380
Transmission (%)	4	23	70	85	90	95	98

## Application

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Benzyl alcohol, due to its high viscosity, is a frequently used solvent in jet stream dye lasers. Its dissolving capacity makes it suitable for polar dyes like Coumarins.

## Safety

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**RTECS # DN3150000; Sax 5, 409 • TLV:**

### Hazards

Harmful by inhalation and if swallowed. Benzyl alcohol is a toxic solvent. It is believed that benzyl alcohol that is present in a poorly ventilated area is the cause of violent headaches, vertigo, nausea, and other symptoms.

### Safety Precautions

In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. Don't smoke, eat, or drink when handling the solvent. Keep all containers and solutions tightly closed.

### Waste Disposal Procedures

Do not dispose of Benzyl Alcohol in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Cyclohexane

Hexahydrobenzene

## General Properties

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Colorless, water insoluble, highly flammable liquid. Sweet, aromatic odor.

## Physical Properties

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Molecular weight:	84.16						
Freezing point (°C):	6.54						
Boiling point (°C):	80.73						
Flash point (°C):	-18						
Min. ignition temp. (°C):	260						
Density (g/cm <sup>3</sup> ):	0.7785 <sup>25</sup>	0.7738 <sup>25</sup>					
Refractive Index:	1.4262 <sup>20</sup>	1.4235 <sup>25</sup>					
Viscosity (cPoise):	0.98 <sup>20</sup>	0.898 <sup>25</sup>					
Dielectric constant:	2.023						
Ionization potential (eV):	9.8						
Solubility:	0.01 % in water, nonpolar organic solvents						
Optical properties:							
Wavelength (nm)	200	210	220	230	240	250	300
Transmission (%)	5	15	45	75	90	98	100

## Application

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Suitable solvent for nonpolar laser dyes like p-Terphenyl. Its high optical transparency in the UV allows the application in dye lasers pumped below 300 nm. The photochemical stability of Cyclohexane is poor.

## Safety

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RTECS # GU6300000; Sax 6, 831 · TLV: 300 ppm or 1050 mg/m<sup>3</sup>

### Hazards

Highly flammable. Absorbed by inhalation. The vapor is mildly irritating to the mucous membranes. The liquid is a fat solvent and thus irritates the skin.

### Safety Precautions

Keep container in a well-ventilated place. Keep away from source of ignition. Take precautionary measures against static dis-charges. Avoid skin contact. Do not smoke, eat, or drink when handling the solvent. Keep all containers tightly closed away from sparks and open flames.

### Waste Disposal Procedures

Do not dispose of Cyclohexane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Dichloroethane

Ethylene chloride · 1,2-Dichloroethane

## General Properties

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Dichloroethane is a colorless, water-insoluble liquid with a sweet odor.

## Physical Properties

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Molecular weight:	98.96						
Freezing point (°C):	-35.7						
Boiling point (°C):	83.5						
Flash point (°C):	13						
Min. ignition temp. (°C):	440						
Density (g/cm <sup>3</sup> ):	1.2531 <sup>20</sup>	1.2458 <sup>25</sup>	1.2383 <sup>30</sup>				
Refractive Index:	1.4448 <sup>20</sup>	1.4421 <sup>25</sup>					
Viscosity (cPoise):	0.887 <sup>15</sup>	0.73 <sup>30</sup>					
Dielectric constant:	10.36						
Ionization potential (eV):	11.12						
Solubility:	0.81 % in water, organic solvents						
Optical properties:							
Wavelength (nm)	225	230	240	250	260	270	300
Transmission (%)	10	30	80	95	98	98	98

## Application

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See Dichloromethane.

## Safety

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**RTECS #** KI0525000; **Sax** 6, 944 · **TLV:** 50 ppm or 200 mg/m<sup>3</sup>

### Hazards

Highly flammable. Harmful by inhalation. One can become adapted to the odor of Dichloroethane at low concentrations, therefore it cannot be considered as a reliable warning. The acute and chronic effects of the solvent can be significant.

### Safety Precautions

Keep container tightly closed. Keep away from sources of ignition. Do not smoke. Take precautionary measures against static discharges.

### Waste Disposal Procedures

Do not dispose of Dichloroethane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Dichloromethane

Methylene chloride

## General Properties

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Dichloromethane is a colorless, water-insoluble liquid with a sweet odor.

## Physical Properties

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Molecular weight:	84.93						
Freezing point (°C):	-95.1						
Boiling point (°C):	39.8						
Flash point (°C):	0						
Min. ignition temp. (°C):	605						
Density (g/cm <sup>3</sup> ):	1.3348 <sup>15</sup>	1.3168 <sup>25</sup>	1.3078 <sup>30</sup>				
Refractive Index:	1.4242 <sup>20</sup>	1.4212 <sup>25</sup>					
Viscosity (cPoise):	0.449 <sup>15</sup>	0.393 <sup>30</sup>					
Dielectric constant:	8.93						
Ionization potential (eV):	11.35						
Solubility:	1.3 % in water, organic solvents						
Optical properties:							
Wavelength (nm)	230	240	250	255	270	290	300
Transmission (%)	10	70	95	98	98	98	98

## Application

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Dichloromethane is used as solvent for laser dyes and saturable absorbers absorbing in the near infrared spectral region.

## Safety

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**RTECS # PA8050000; Sax 6, 1763 · TLV: 500 ppm or 1750 mg/m<sup>3</sup>**

### Hazards

Harmful by inhalation. The toxic effect of Dichloromethane is predominately narcosis. It is mildly irritating to the skin on repeat contact if free to evaporate. It is painful to the eyes but no permanent damage may be expected.

### Safety Precautions

Avoid contact with skin. Do not smoke, eat, or drink when handling the solvent.

### Waste Disposal Procedures

Do not dispose of Dichloromethane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Dioxane

p-Dioxane

## General Properties

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Colorless, volatile, and very hygroscopic liquid with slightly aromatic taste.

## Physical Properties

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Molecular weight:	88.11						
Freezing point (°C):	11.8						
Boiling point (°C):	101.32						
Flash point (°C):	12						
Min. ignition temp. (°C):	375						
Density (g/cm <sup>3</sup> ):	1.0336 <sup>20</sup>		1.028 <sup>25</sup>				
Refractive Index:	1.4224 <sup>120</sup>		1.42025 <sup>25</sup>				
Viscosity (cPoise):	1.439 <sup>15</sup>		1.087 <sup>30</sup>				
Dielectric constant:	2.209						
Ionization potential (eV):	9.13						
Solubility:	in water, organic solvents						
Optical properties:							
Wavelength (nm)	225	240	250	260	280	290	300
Transmission (%)	40	50	60	70	85	93	98

## Application

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Its high photochemical stability and excellent dissolving capacity makes dioxane a versatile solvent for UV and Coumarin dyes.

## Safety

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**RTECS #** JG8225000; **Sax 6**, 1227 · **TLV:** 100 ppm, 360 mg/m<sup>3</sup>

### Hazards

Highly flammable. May form explosive peroxides. Harmful by inhalation. Painful to the eyes and irritating to the skin upon prolonged contact. It can be absorbed through the skin in toxic amounts. Dioxane is insidious. Its vapors have poor warning properties; they are faint and inoffensive. Concentrations in air of 300 ppm cause irritation of the eyes, nose, and throat. The vapors can be inhaled in amounts that cause serious systemic injury.

### Safety Precautions

Keep container in a well-ventilated place. Keep away from sources of ignition. Take precautionary measures against static discharges. Avoid skin contact. Do not smoke, eat, or drink when handling the solvent. Keep all containers and solutions tightly closed.

**Waste Disposal Procedures**

Do not dispose of Dioxane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.



# DMSO

Dimethyl Sulfoxide

## General Properties

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Colorless, odorless, hygroscopic liquid with a slightly bitter taste. It has an equilibrium moisture content of 10 percent with air at 20°C.

## Physical Properties

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Molecular weight:	78.13						
Freezing point (°C):	18.54						
Boiling point (°C):	189						
Flash point (°C):	95						
Min. ignition temp. (°C):	383						
Density (g/cm³):	1.0958 <sup>25</sup>	1.0816 <sup>40</sup>	1.0616 <sup>60</sup>				
Refractive Index:	1.4783 <sup>20</sup>	1.4773 <sup>25</sup>					
Viscosity (cPoise):	1.996 <sup>25</sup>	1.654 <sup>35</sup>					
Dielectric constant:	46.68						
Ionization potential (eV):							
Solubility:	25.3 % in water, organic solvents						
Optical properties:							
Wavelength (nm)	263	270	280	290	300	310	340
Transmission (%)	10	34	60	71	85	90	98

## Application

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DMSO is an excellent solvent for polar dyes like Rhodamines.

## Safety

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**RTECS #** PV6210000; **Sax 6, 1201 · TLV:**

### Hazards

DMSO dehydrates and defats the skin, but seems to be relatively free from toxic effects. Its vapor, mixed with air, may explode above 90° C.

### Safety Precautions

Dimethyl sulfoxide may produce eye, skin, and respiratory irritations. The solvent penetrates the skin and that toxic solutes are carried with it into the body fluid. Avoid contact with skin and eyes.

### Waste Disposal Procedures

Avoid mixing contaminated solvents because several substances have been reported to have produced an explosion when mixed with dimethyl sulfoxide. Do not dispose of DMSO in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Ethanol

Ethyl Alcohol

## General Properties

Colorless liquid which may be obtained as the water azeotrop containing about 5 percent water or as absolute alcohol containing 0.1 percent or less water.

## Physical Properties

Molecular weight:	46.07						
Freezing point (°C):	-114.1						
Boiling point (°C):	78.3						
Flash point (°C):	12						
Min. ignition temp. (°C):	425						
Density (g/cm <sup>3</sup> ):	0.7936 <sup>15</sup>	0.7894 <sup>20</sup>	0.785 <sup>25</sup>				
Refractive Index:	1.3614 <sup>20</sup>	1.3594 <sup>25</sup>					
Viscosity (cPoise):	1.078 <sup>25</sup>	0.991 <sup>35</sup>					
Dielectric constant:	24.55						
Ionization potential (eV):	10.49						
Solubility:	water, organic solvents						
Optical properties :							
Wavelength (nm)	200	210	220	230	240	250	270
Transmission (%)	5	35	55	72	85	90	98

## Application

Ethanol is the most frequently used solvent for laser dyes. As it is highly polar, its application is restricted to polar dyes such as the Rhodamines.

## Safety

**RTECS # KQ6300000; Sax 6, 1316 · TLV: 1000 ppm or 1900 mg/m<sup>3</sup>**

**Hazards** Highly flammable. It is practically impossible to produce any toxic effects by inhalation of pure ethanol vapors under usual lab conditions. The minimum identifiable odor is about 530 ppm. Concentrations of 6,000-9,000 ppm have an intense odor that may be practically intolerable at first, but one becomes acclimated soon. Concentrations of about 1,000 ppm cause slight irritation of mucous membranes and other symptoms.

### Safety Precautions

Keep container tightly closed. Keep away from sources of ignition. Do not smoke. Limit the quantity stored to foreseeable short-term requirements; large quantities should not be allowed to accumulate in the laboratory. If spillage of solvent or accidental release occurs, ventilate the whole laboratory as soon as possible.

**Waste Disposal Procedures**

Do not dispose of Ethanol in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Ethylene Glycol

1,2-Ethanediol

## General Properties

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Ethylene glycol is a colorless, odorless liquid with a bittersweet taste. It is very hygroscopic.

## Physical Properties

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Molecular weight:	62.07						
Freezing point (°C):	-13						
Boiling point (°C):	197.3						
Flash point (°C):	110						
Min. ignition temp. (°C):	410						
Density (g/cm <sup>3</sup> ):	1.1135 <sup>20</sup>	1.11 <sup>25</sup>	1.1066 <sup>30</sup>				
Refractive Index:	1.4318 <sup>20</sup>	1.4306 <sup>25</sup>					
Viscosity (cPoise):	26.09 <sup>15</sup>	13.55 <sup>30</sup>					
Dielectric constant:	37.7						
Ionization potential (eV):	10.49						
Solubility:	water, alcohols						
Optical properties:							
Wavelength (nm)	210	220	230	240	250	260	280
Transmission (%)	20	35	40	60	75	90	98

## Application

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Ethylene glycol, due to its high viscosity, is a frequently used solvent in jet stream dye lasers. Its dissolving capacity makes it suitable for polar dyes like Coumarins, Rhodamines, and Cyanines.

## Safety

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**RTECS #** KW2975000; **Sax 6**, 1343 · **TLV: 100 ppm or 274 mg/m<sup>3</sup>**

### Hazards

Harmful if swallowed. Ethylene glycol has a low vapor pressure at normal temperature. It presents negligible hazards to health except, possibly, when being used at elevated temperature. It has a low, acute oral toxicity. There does not appear to be any significant irritation from skin contact.

### Safety Precautions

Avoid skin contact. Store in closed container away from heat, sparks and open flame.

### Waste Disposal Procedures

Place Ethylene Glycol in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Hexane

n-Hexane

## General Properties

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Hexane is a colorless, water-insoluble, and highly flammable liquid.

## Physical Properties

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Molecular weight:	86.18						
Freezing point (°C):	-95.4						
Boiling point (°C):	68.74						
Flash point (°C):	-26						
Min. ignition temp. (°C):	240						
Density (g/cm <sup>3</sup> ):	0.6594 <sup>20</sup>	0.6548 <sup>25</sup>					
Refractive Index:	0.3749 <sup>20</sup>	1.3723 <sup>25</sup>					
Viscosity (cPoise):	0.3126 <sup>20</sup>	0.2985 <sup>25</sup>					
Dielectric constant:	1.8799						
Ionization potential (eV):	10.18						
Solubility:	0.00095 % in water, nonpolar organic solvents						
Optical properties:							
Wavelength (nm)	190	200	210	220	230	240	250
Transmission (%)	10	30	60	80	94	98	98

## Application

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Suitable solvent for nonpolar laser dyes. Its high optical transparency in the UV range allows the application in dye lasers pumped below 300 nm.

## Safety

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**RTECS #** MN9275000; **Sax** 6, 1523• **TLV:** 500 ppm or 1800 mg/m<sup>3</sup>

### Hazards

Highly flammable. Harmful by inhalation and in contact with the skin. Possible risk of irreversible effects. Hexane is a fat solvent and thus irritates the skin.

### Safety Precautions

Keep container in a well-ventilated place. Keep away from source of ignition. Do not inhale gas/fumes/vapor/spray. Avoid skin contact. Do not smoke, eat, or drink when handling the solvent. Keep all containers tightly closed away from sparks and open flames.

### Waste Disposal Procedures

Do not dispose of Hexane in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Methanol

Methyl Alcohol

## General Properties

Methanol is a colorless hygroscopic liquid usually containing 0.01 - 0.04 percent water. It is highly inflammable and toxic.

## Physical Properties

Molecular weight:	32.04						
Freezing point (°C):	-97.7						
Boiling point (°C):	64.7						
Flash point (°C):	11						
Min. ignition temp. (°C):	455						
Density (g/cm <sup>3</sup> ):	0.7961 <sup>15</sup>	0.7913 <sup>20</sup>	0.7866 <sup>25</sup>				
Refractive Index:	1.3284 <sup>20</sup>	1.3265 <sup>25</sup>					
Viscosity (cPoise):	0.5506 <sup>25</sup>	0.5445 <sup>35</sup>					
Dielectric constant:	32.7						
Ionization potential (eV):	10.84						
Solubility:	water, organic solvents						
Optical properties:							
Wavelength (nm)	200	210	220	230	240	250	260
Transmission (%)	2	20	50	75	85	95	98

## Application

Methanol is a polar, protic solvent frequently used to dissolve laser dyes like Coumarins, Rhodamines, and Cyanines. Its excellent optical transparency makes it the ideal solvent for UV-pumped dye lasers.

## Safety

**RTECS # PC1400000; Sax 6, 1764 · TLV: 200 ppm or 260 mg/m<sup>3</sup>**

### Hazards

Highly flammable. Toxic by inhalation and if swallowed. Methanol does not have suitable warning or irritating properties except at high concentrations. Ingestion of methanol can cause blindness and death.

### Safety Precautions

Keep container tightly closed. Keep away from sources of ignition. Do not smoke. Avoid contact with skin. Methanol vapor/air mixtures may produce explosible mixtures. Keep containers tightly closed. If spillage of solvent or accidental release occurs, ventilate the whole laboratory as soon as possible.

**Waste Disposal Procedures**

Do not dispose of Methanol in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.

# Toluene

Methylbenzene

## General Properties

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Colorless, water-insoluble, and highly flammable liquid with aromatic odor.

## Physical Properties

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Molecular weight:	92.14						
Freezing point (°C):	-95						
Boiling point (°C):	110.6						
Flash point (°C):	7						
Min. ignition temp. (°C):	535						
Density (g/cm <sup>3</sup> ):	0.867 <sup>25</sup>	0.8623 <sup>25</sup>	0.8577 <sup>30</sup>				
Refractive Index:	1.4969 <sup>20</sup>	1.4941 <sup>25</sup>					
Viscosity (cPoise):	0.5866 <sup>20</sup>	0.5516 <sup>25</sup>					
Dielectric constant:	2.379						
Ionization potential (eV):	8.82						
Solubility:	0.0515 % in water, nonpolar organic solvents						
Optical properties:							
Wavelength (nm)	285	290	300	310	320	340	350
Transmission (%)	10	50	80	89	93	95	98

## Application

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Toluene is a suitable solvent for nonpolar laser dyes like p-Terphenyl. Its low optical transparency in the UV range restricts the application on dye lasers pumped above 320 nm.

## Safety

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**RTECS #** XS5250000; **Sax** 6, 2588 · **TLV**:200 ppm or 7500 mg/m<sup>3</sup>

### Hazards

Highly flammable. Harmful by inhalation. Exposure of humans to Toluene vapor produces mild fatigue, weakness, confusion, and paresthesia of the skin. Toluene is most dangerous by inhalation. It is irritating to the skin, and contact should be avoided when possible.

### Safety Precautions

Keep away from source of ignition. Do not smoke. Take precautionary measures against static discharges.

### Waste Disposal Procedures

Do not dispose of Toluene in the sink. Place it in appropriately labelled, suitable containers. Only trained and licensed waste disposal experts should dispose of accumulated waste material in accordance with governmental regulations.



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Riddick, John A. and William B. Bunger. "Organic Solvents." *Techniques of Chemistry*, Vol. 3, 1970. New York, New York, USA: Wiley-Interscience.

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Sax, N. I. *Dangerous Properties of Industrial Materials*. New York, New York, USA: Van Nostrand Reinhold, 1975.

Schneider, R. L. "Physical Properties of Some Organic Solvents." *Eastman Organic Chemical Bulletin*. Vol. 47, No. 1, 1975.

# Laser Dyes

## Properties, Application, and Absorption Spectra

The output power of dye lasers depends on the quality of the dye used. To overcome reduced quantum efficiency and instability due to impurities, Lambdachrome® laser dyes are synthesized and examined by experienced chemists for their chemical and spectral properties. Finally, they are purified by specially developed techniques.

The composition of all dyes is guaranteed with spectrophotometric and chromatographic analysis. Wavelength ranges are given in the following tables to provide assistance in choosing the correct dye for a given application. The ranges were measured by Lambda Physik, taken from Coherent CW Dye Laser Fact Sheets, or taken from the pertinent literature. The exact spectral range depends on the solvent and the concentration as well as on the method of pumping.

Lambda Physik reserves the right to modify any information given herein. Every effort is made to ensure utmost accuracy; however, no liability is assumed for errors occurring in the tables.

## Abbreviations Used

MW	Molecular Weight
CAS	Chemical Abstracts Service
Effic.	Efficiency, defined as the ratio of optical input to output
Conc.	Concentration of dye, given in grams per liter solvent
Ref.	Reference
Cyclohex.	Cyclohexane
DCE	1,2-Dichloroethane
DMF	Dimethylformamide
EtOH	Ethanol
MeOH	Methanol
Eg	Ethyleneglycol
Bz	Benzylalcohol
DMSO	Dimethylsulfoxide
Pc	Propylenecarbonate
Tol	Toluene

## BEER's Law

Within the absorption spectra, the intensity is expressed as a molar decadic extinction coefficient,  $\epsilon$ . The amount of light absorbed depends on the extinction coefficient and the number of molecules in the light path. The latter amount depends on the concentration of the dye in solution and the path length of the absorption cell. The amount of light that passes through a solution (transmittance) is given by BEER's law:

$$\log I_0/I = \epsilon \cdot c \cdot d$$

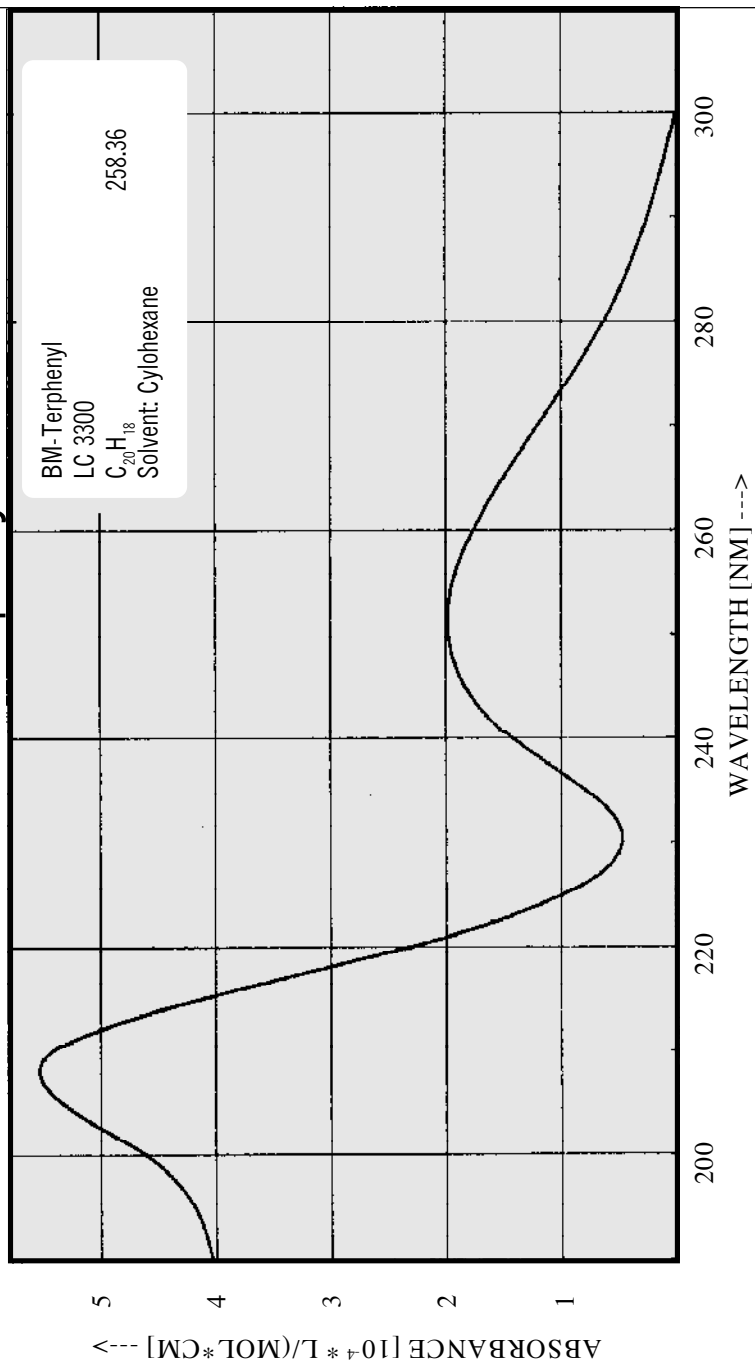
where  $I_0$  is the intensity of the light before it encounters the cell,  $I$  is the intensity of the light emerging from the cell,  $c$  is the concentration in moles per liter, and  $d$  is the path length in centimeters.

The absorption cross-section  $\sigma$  can be determined from the extinction coefficient  $\epsilon$  by:

$$\sigma = 0.385 \cdot 10^{-20} \epsilon.$$

Here  $\sigma$  is given in  $\text{cm}^2$  ( $\epsilon$  measured in  $\text{liter}/(\text{mole} \cdot \text{cm})$ ).

# BM-Terphenyl



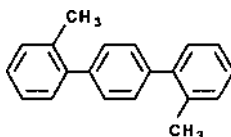
# BM-Terphenyl (LC 3300)

## Constitution

2,2''-Dimethyl-p-terphenyl

DMT

$C_{20}H_{18}$  · MW: 258.36



## Characteristics

Lambdachrome® number:	3300
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in cyclohexane):	251 nm
Molar absorptivity:	$1.98 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in cyclohexane):	335 nm
For research and development purposes only.	

## Lasing Performance

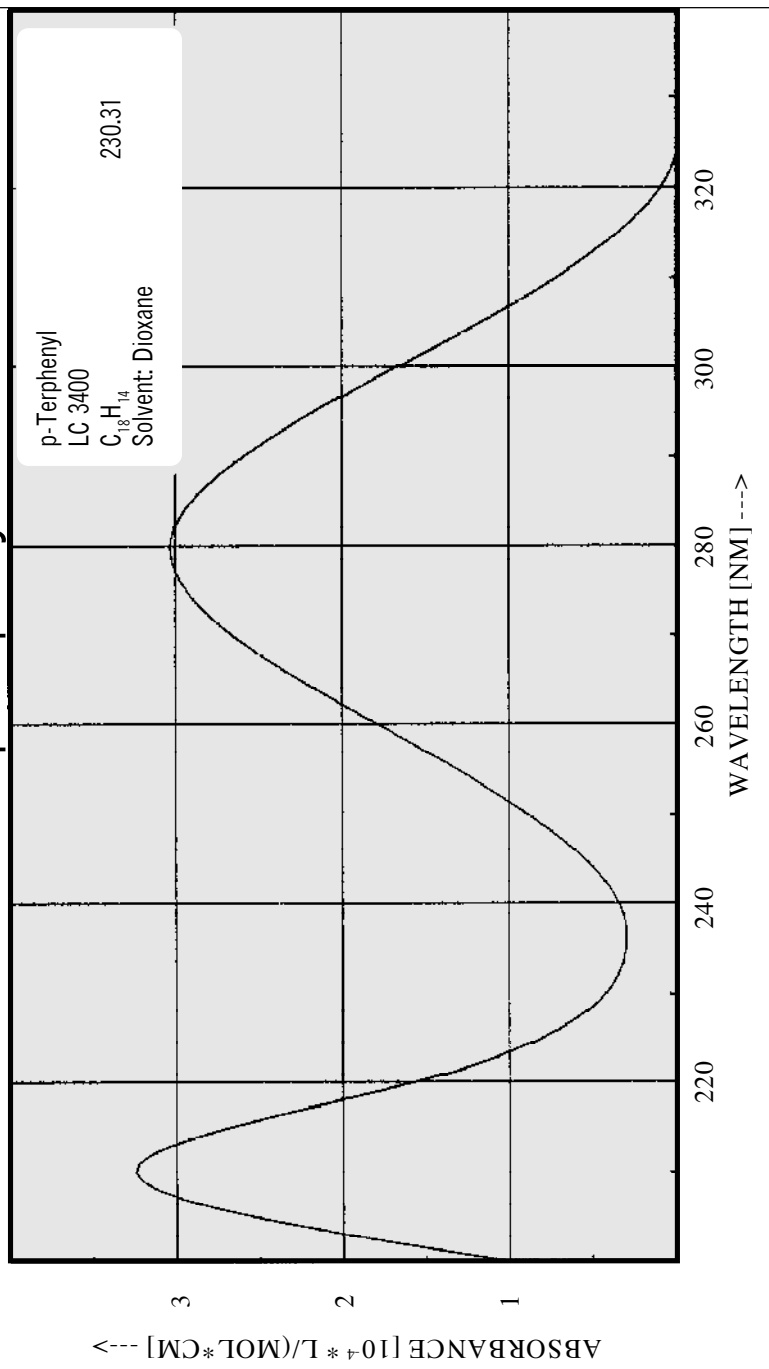
Shortest tunable laser dye for pulsed operation; tunable around 336 nm.

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
KrF-Excimer	248		334	312 - 343	4	0.50	Cyclohex	1, 2, 3

## References

1. Lambda Physik, *Wall Chart* 1996.
2. W. Zapka, U. Brackmann *Appl. Phys.* **20**, 283 (1979).
3. F.-G. Zhang, F. P. Schäfer *Appl. Phys.* **B26**, 211 (1981).

# p-Terphenyl

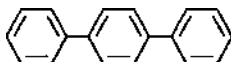


# p-Terphenyl (LC 3400)

## Constitution

PTP

C<sub>18</sub>H<sub>14</sub> · MW: 230.31



## Characteristics

Lambdachrome® number:	3400
CAS registry number:	92-94-4
Appearance:	white, crystalline solid
Absorption maximum (in cyclohexane):	275 nm
Molar absorptivity:	3.21 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum (in cyclohexane):	339 nm
For research and development purposes only.	

## Lasing Performance

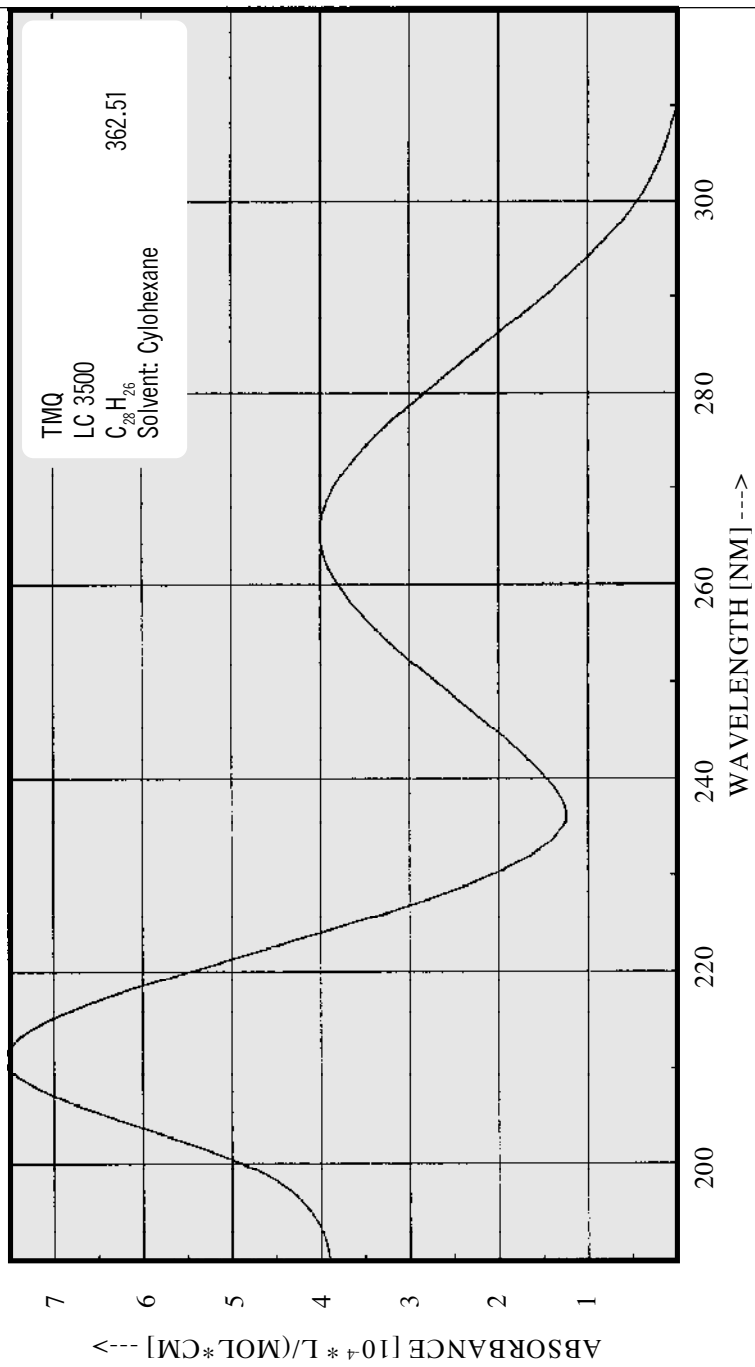
Efficient laser dye for pulsed operation; tunable around 340 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
KrF-Excimer	248	339	322 - 365	14	0.02	Cyclohex	1
XeCl-Excimer	308	343	332 - 360	8	0.24	Dioxane	2, 3
Nd:YAG, 4th	266	340	-	5	0.23	Cyclohex.	4, 5

## References

1. H. Bücher, W. Chow, *Appl. Phys.* **13**, 267 (1977).
2. Lambda Physik, *Wall Chart* 1996.
3. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38(5,6)**, 402 (1981).
4. G. A. Abakumov et al., *Opto-Electron.* **1**, 205 (1969).
5. D. Huppert, P. M. Rentzepis, *J. Appl. Phys.* **49(2)**, 543 (1978).

# TMQ



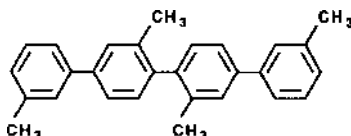


# TMQ (LC 3500)

## Constitution

3,3',2'',3'''-Tetramethyl-p-quaterphenyl

C<sub>28</sub>H<sub>26</sub> · MW: 362,51



## Characteristics

Lambdachrome® number:	3500
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in cyclohexane):	266 nm
Molar absorptivity:	3.99 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum:	-
For research and development purposes only.	

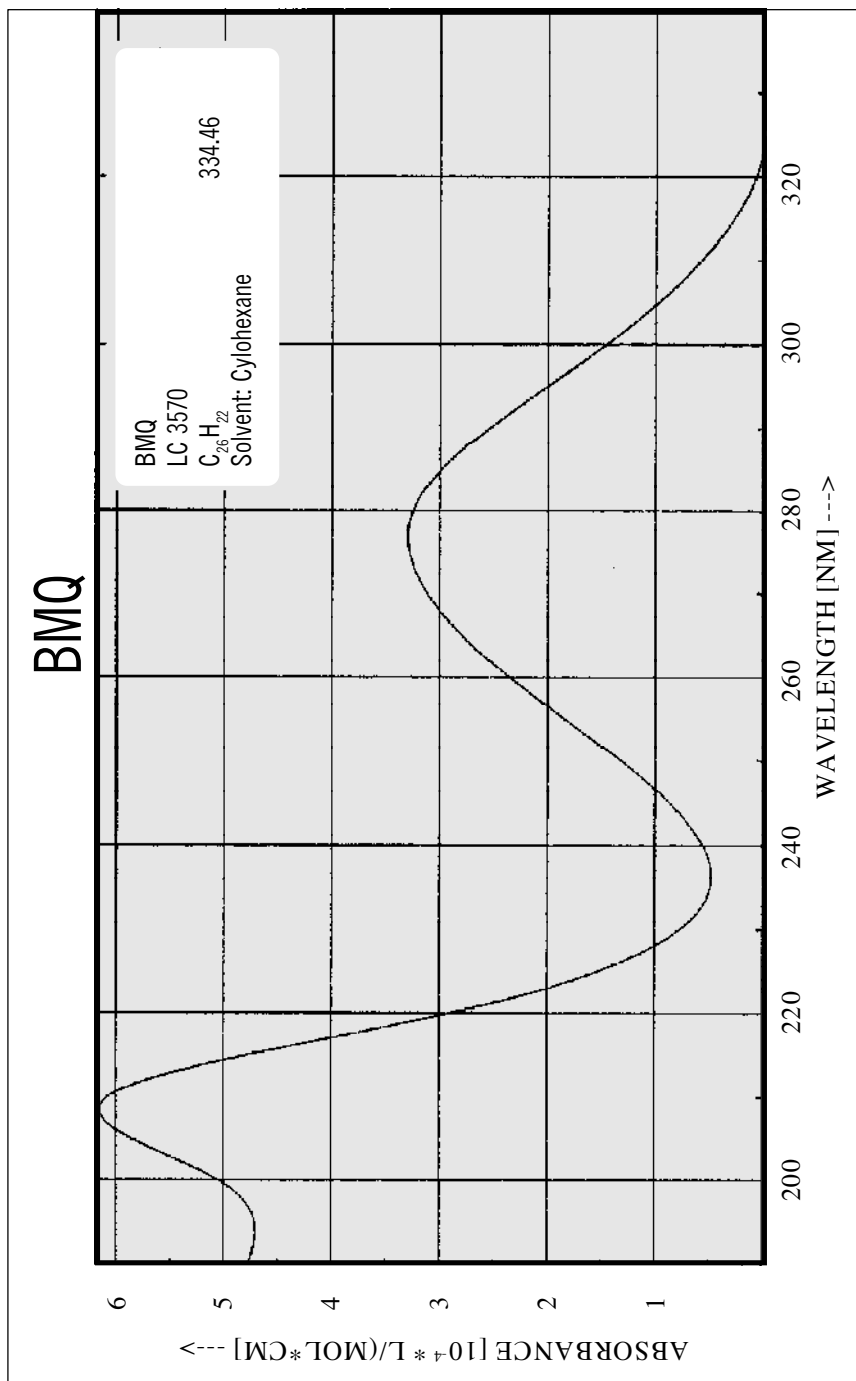
## Lasing Performance

Laser dye for pulsed operation; tunable around 350 nm.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
Nd:YAG, 4th	266	350	338 - 361	1.2	0.72	Cyclohex	1

## References

1. L. D. Ziegler, B. S. Hudson, *Opt. Commun.* **32**(1), 119 (1980).

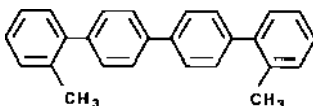


# BMQ (LC 3570)

## Constitution

2,2'''-Dimethyl-p-quaterphenyl

C<sub>26</sub>H<sub>22</sub> - MW: 334.46



## Characteristics

Lambdachrome® number:	3570
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in cyclohexane):	275 nm
Molar absorptivity:	$3.35 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

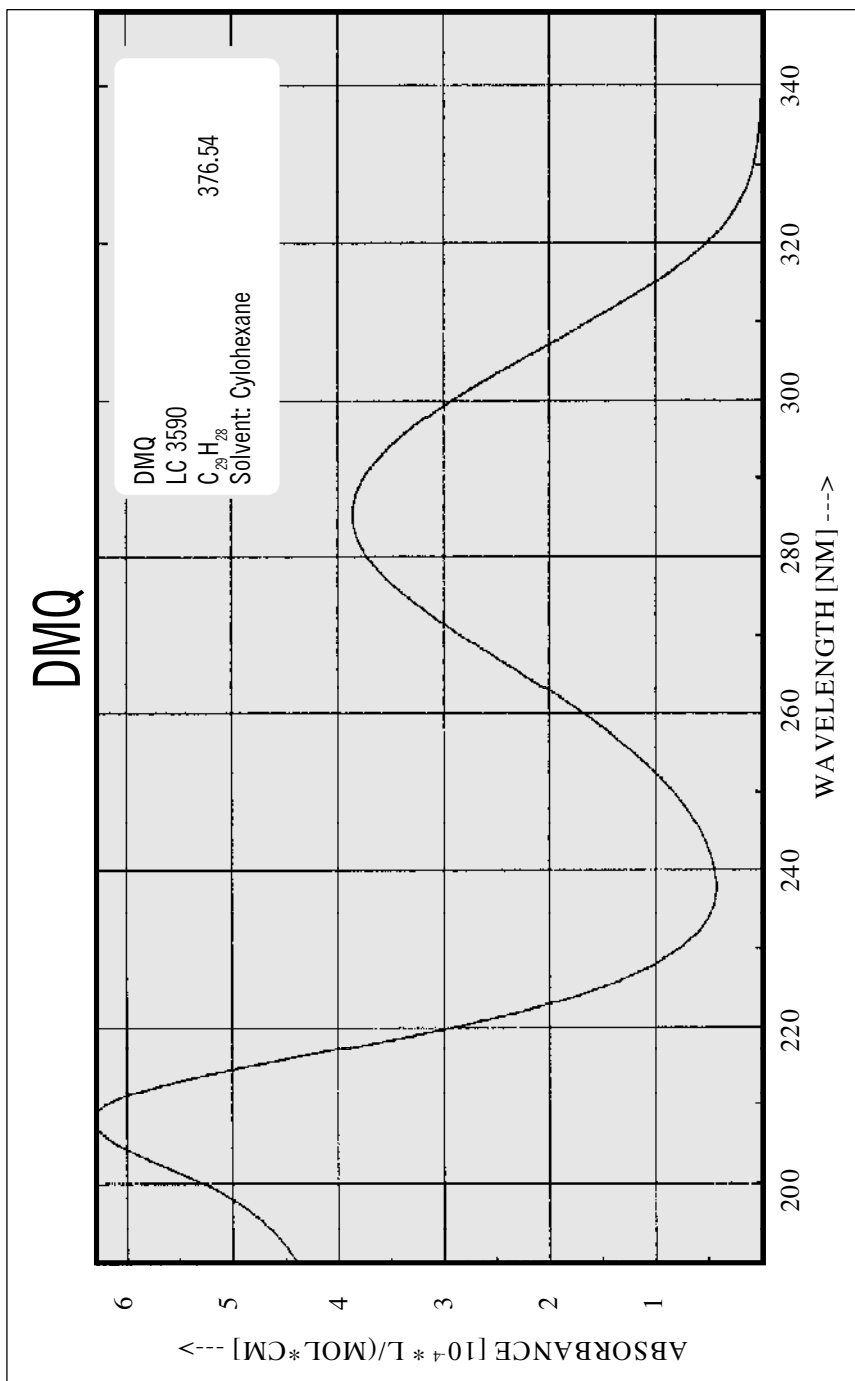
## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 357 nm.

<i>Pump</i>		<i>Dye Laser Characteristics</i>					<i>Ref.</i>
<i>Source</i>	<i>Wavelength</i> [nm]	<i>Peak</i> [nm]	<i>Range</i> [nm]	<i>Effic.</i> [%]	<i>Conc.</i> [g/l]	<i>Solvent</i>	
<i>XeCl-Excimer</i>	<i>308</i>	<i>357</i>	<i>335 - 375</i>	<i>9</i>	<i>0.60</i>	<i>Dioxane</i>	<i>1</i>

## References

1. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).

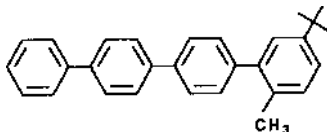


# DMQ (LC 3590)

## Constitution

2-Methyl-5-*t*-butyl-*p*-quaterphenyl

C<sub>29</sub>H<sub>28</sub> · MW: 376.54



## Characteristics

Lambdachrome® number:	3590
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in cyclohexane):	285 nm
Molar absorptivity:	3.86 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

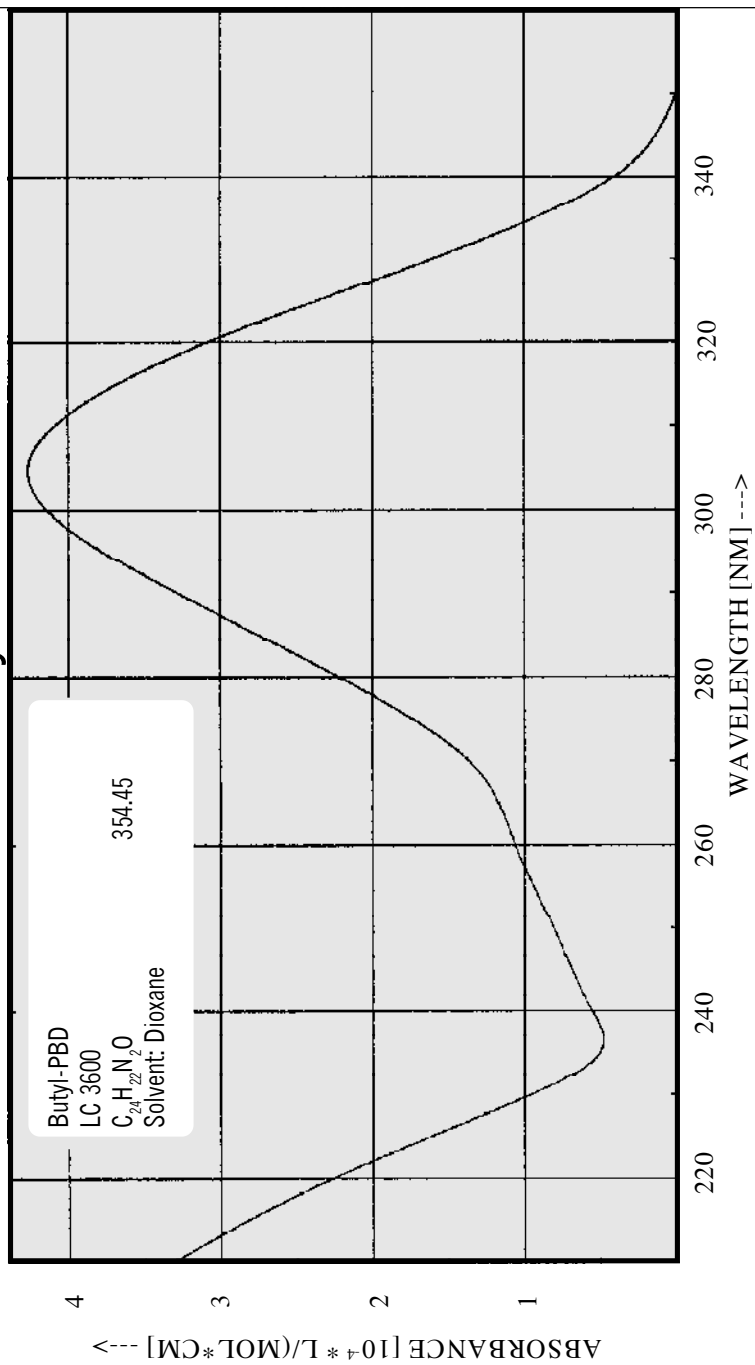
Efficient laser dye for pulsed operation; tunable around 360 nm.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	360	346 - 377	9	0.23	Dioxane	1, 2

## References

1. Lambda Physik, *Wall Chart* 1996
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).

## Butyl-PBD

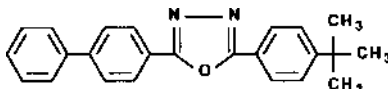


# Butyl-PBD (LC 3600)

## Constitution

2-(4-Biphenyl)-5-(4-t-butylphenyl)-1,3,4-oxadiazol  
BPBD-365

$C_{24}H_{22}N_2O$  · MW: 354.45



## Characteristics

Lambdachrome® number:	3600
CAS registry number:	15082-28-7
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	302 nm
Molar absorptivity:	$4.35 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in cyclohexane):	368 nm
For research and development purposes only.	

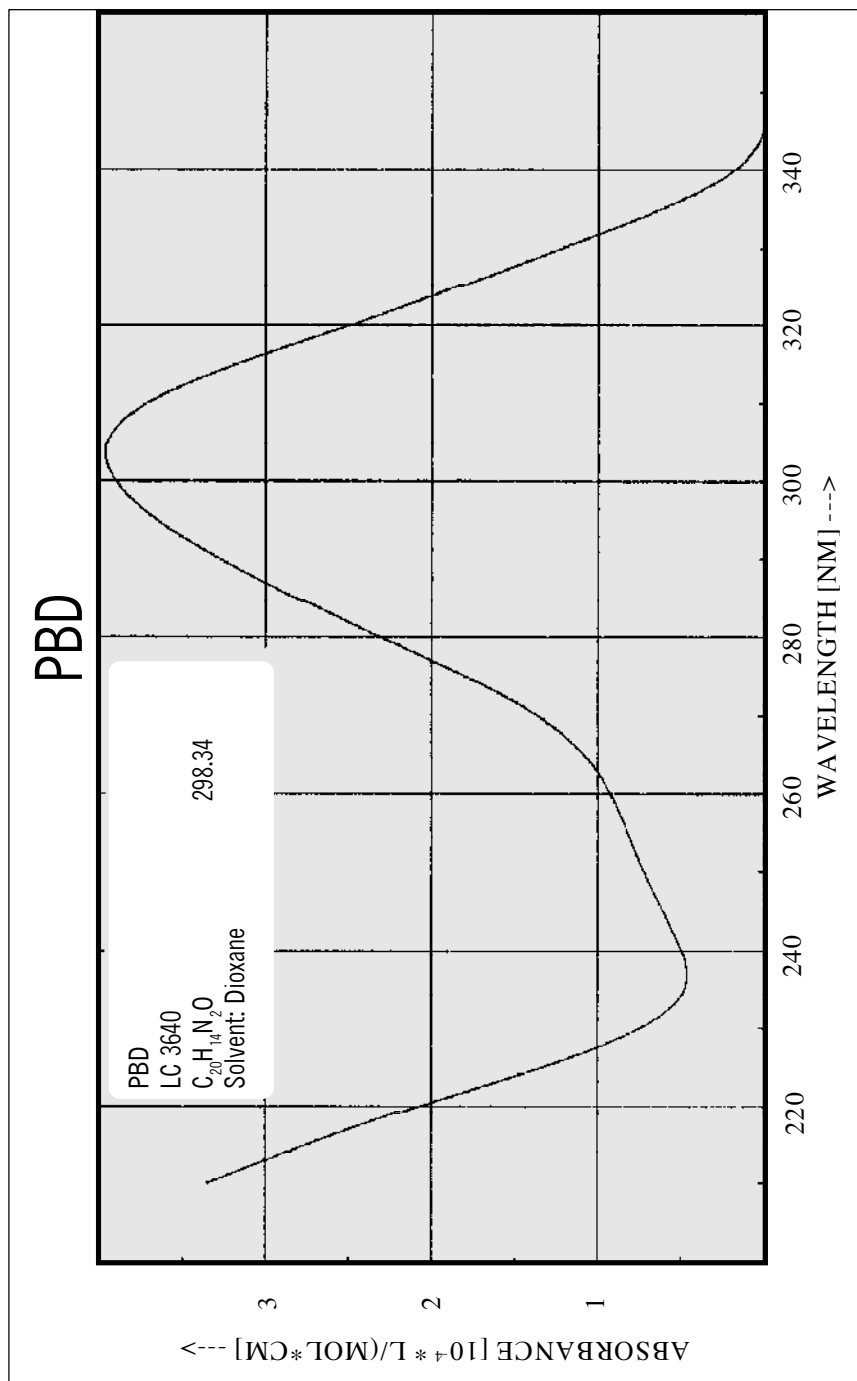
## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 360 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	363	356 - 385	5	0.30	Dioxane	1, 2, 3
Nitrogen	337	362	356 - 390	rel.	1.60	Dioxane	3, 4
Nd:YAG, 4th	266	362	354-388	4.7	1.75	Cyclohex.	5

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 402 (1981).
3. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. L. D. Ziegler, B. S. Hudson, *Opti. Commun.* **32**(1), 119 (1980).



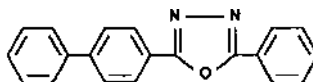


# PBD (LC 3640)

## Constitution

2-(4-Biphenyl)-5-phenyl-1,3,4-oxadiazol

C<sub>20</sub>H<sub>14</sub>N<sub>2</sub>O · MW: 298.34



## Characteristics

Lambdachrome® number:	3640
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	302 nm
Molar absorptivity:	3.90 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum (in toluene):	360 nm
For research and development purposes only.	

## Lasing Performance

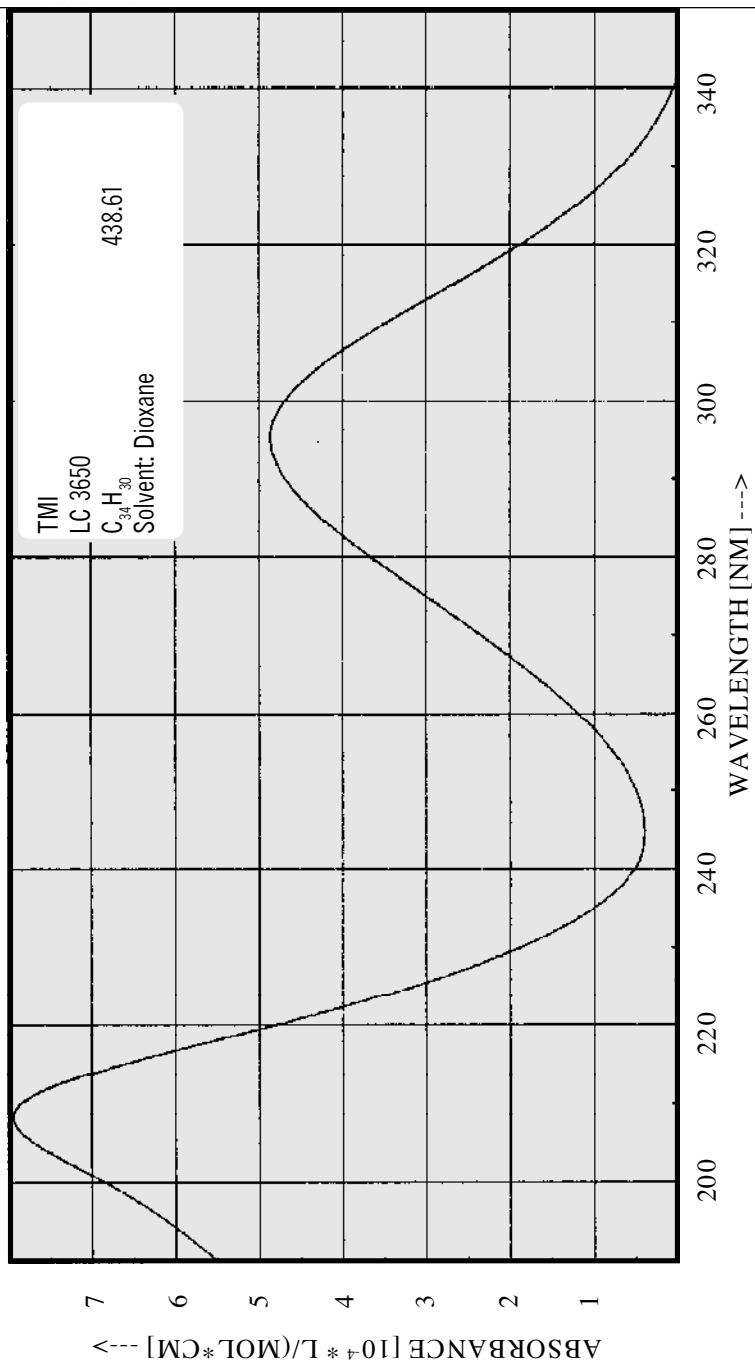
Efficient laser dye for pulsed operation; tunable around 360 nm.

Source	Pump		Dye Laser Characteristics					Ref.
	Wavelength		Peak	Range	Effic.	Conc.	Solvent	
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		363	355 - 390	4	0.11	Cyclohex.	1
Nitrogen	337		362	357 - 390	rel.	1.20	Dioxane	2
Flashlamp	-		362	-	-	0.08	Ethanol	3, 4

## References

1. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 402 (1981).
2. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
3. H. W. Furumoto, H. L. Cecon, *IEEE J. Quant. Electron.* **QE-6**, 262 (1970).
4. T. Morrow, H. T. W. Price, *Opt. Commun.* **10**(2), 133 (1974).

# TMI

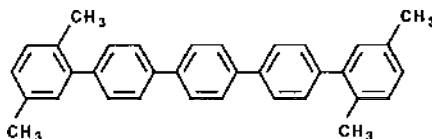


# TMI (LC 3650)

## Constitution

2,5,2''',5'''-Tetramethyl-p-quinquephenyl

C<sub>34</sub>H<sub>30</sub> · MW: 438.61



## Characteristics

Lambdachrome® number:	3650
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	295 nm
Molar absorptivity:	$4.86 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-

For research and development purposes only.

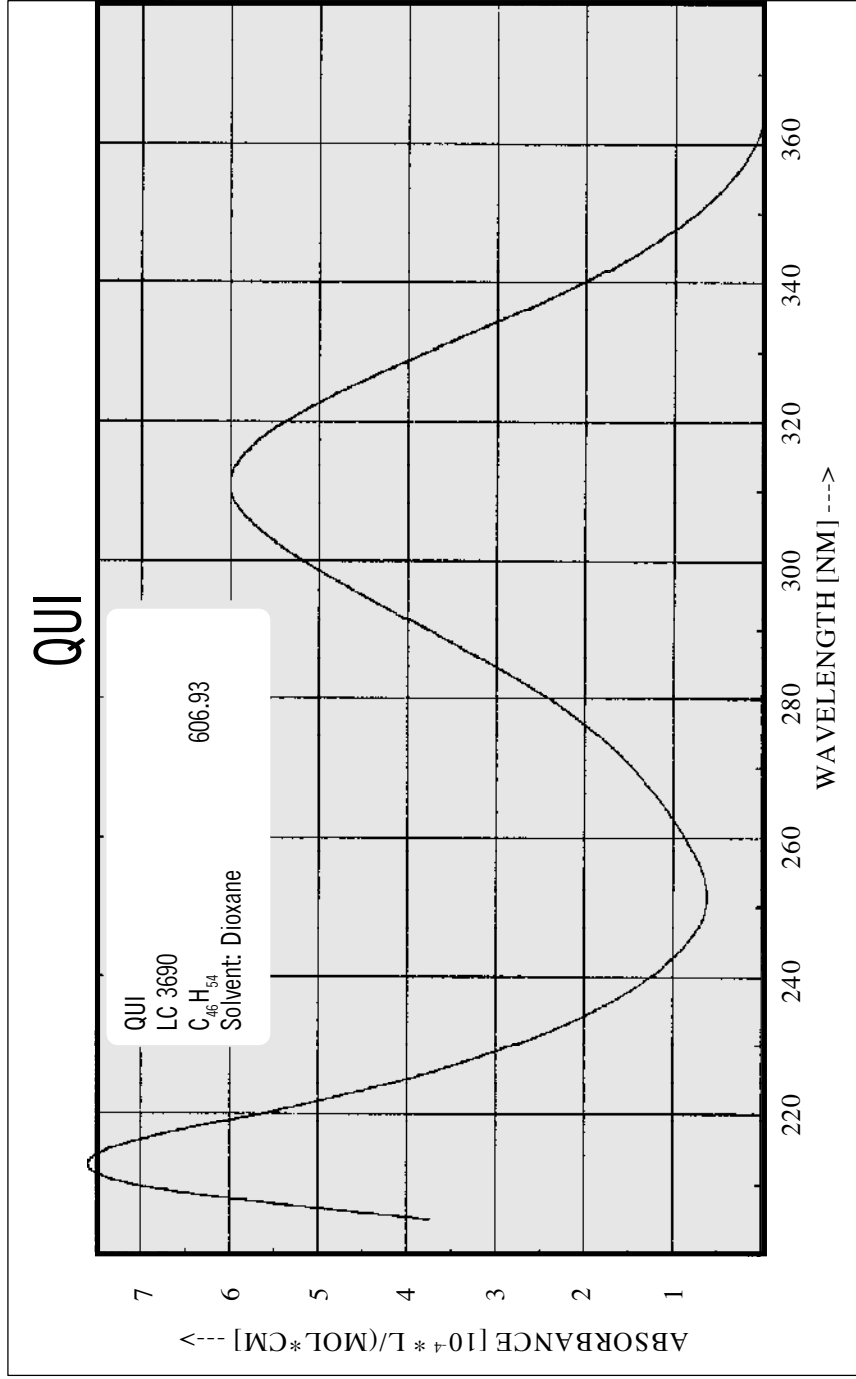
## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 370 nm.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. Conc. [%][g/l]			
XeCl-Excimer	308	372	355 - 395	11	0.18	Dioxane	1, 2

## References

1. Lambda Physik, *Data Sheet*.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).

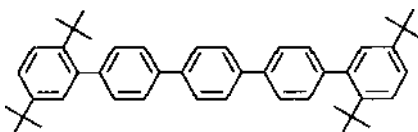


# QUI (LC 3690)

## Constitution

3,5,3''',5'''-Tetra-*t*-butyl-*p*-quinquephenyl

C<sub>46</sub>H<sub>54</sub> · MW: 606.93



## Characteristics

Lambdachrome® number:	3690
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	310 nm
Molar absorptivity:	6.00 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum:	-
For research and development purposes only.	

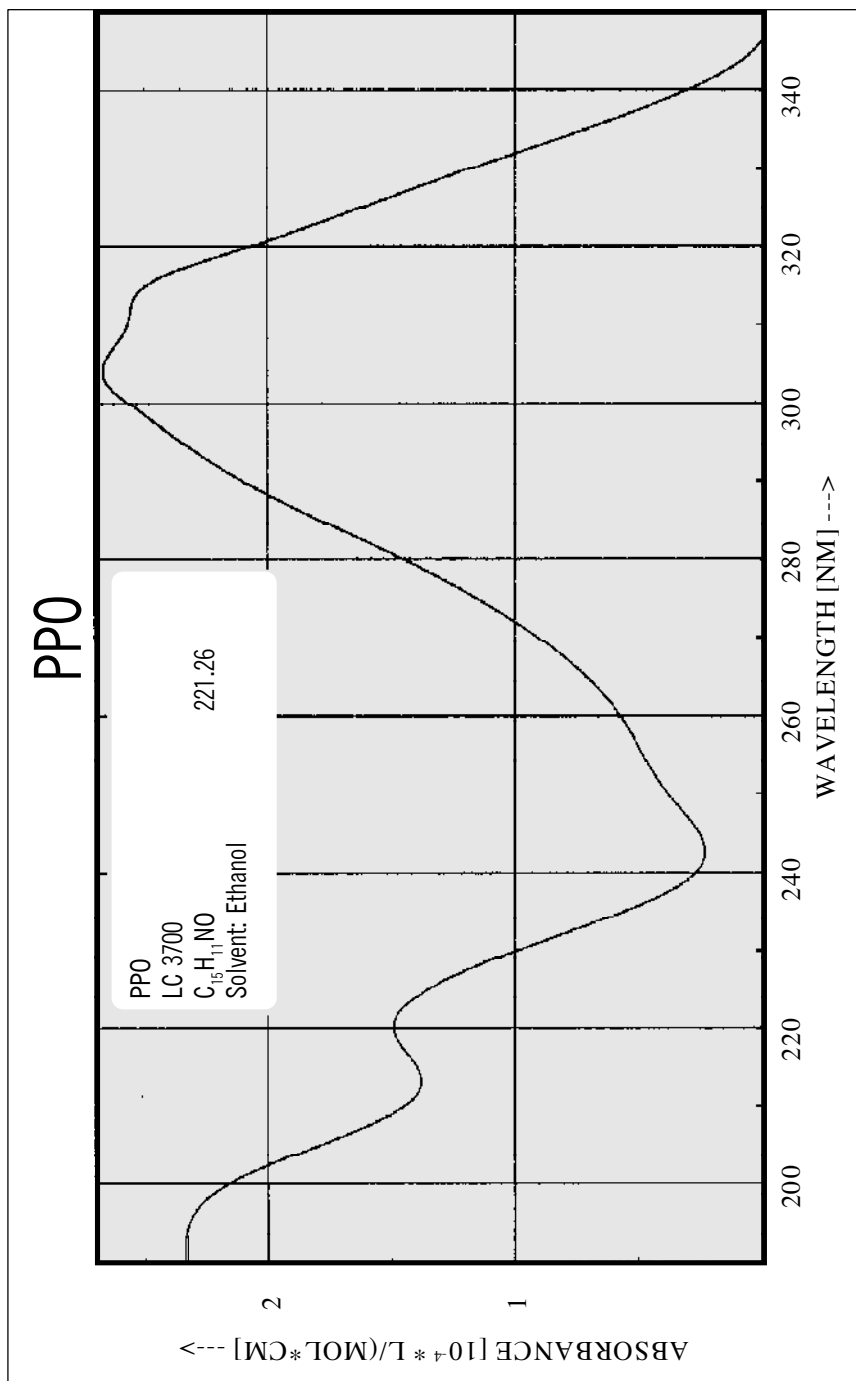
## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 390 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	390	368 - 402	11	0.20	Dioxane	1, 2
Nitrogen	337	387	372-412	rel.	0.52	Dioxane	3

## References

1. Lambda Physik, *Wall Chart* 1996.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).
3. Lambda Physik, *Data Sheet*.

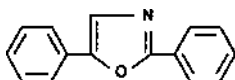


# PPO (LC 3700)

## Constitution

2,5-Diphenyloxazol

C<sub>15</sub>H<sub>11</sub>NO · MW: 221.26



## Characteristics

Lambdachrome® number:	3700
CAS registry number:	92-71-7
Appearance:	white, crystalline solid
Absorption maximum (in ethanol):	303 nm
Molar absorptivity:	$2.80 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	365 nm
For research and development purposes only.	

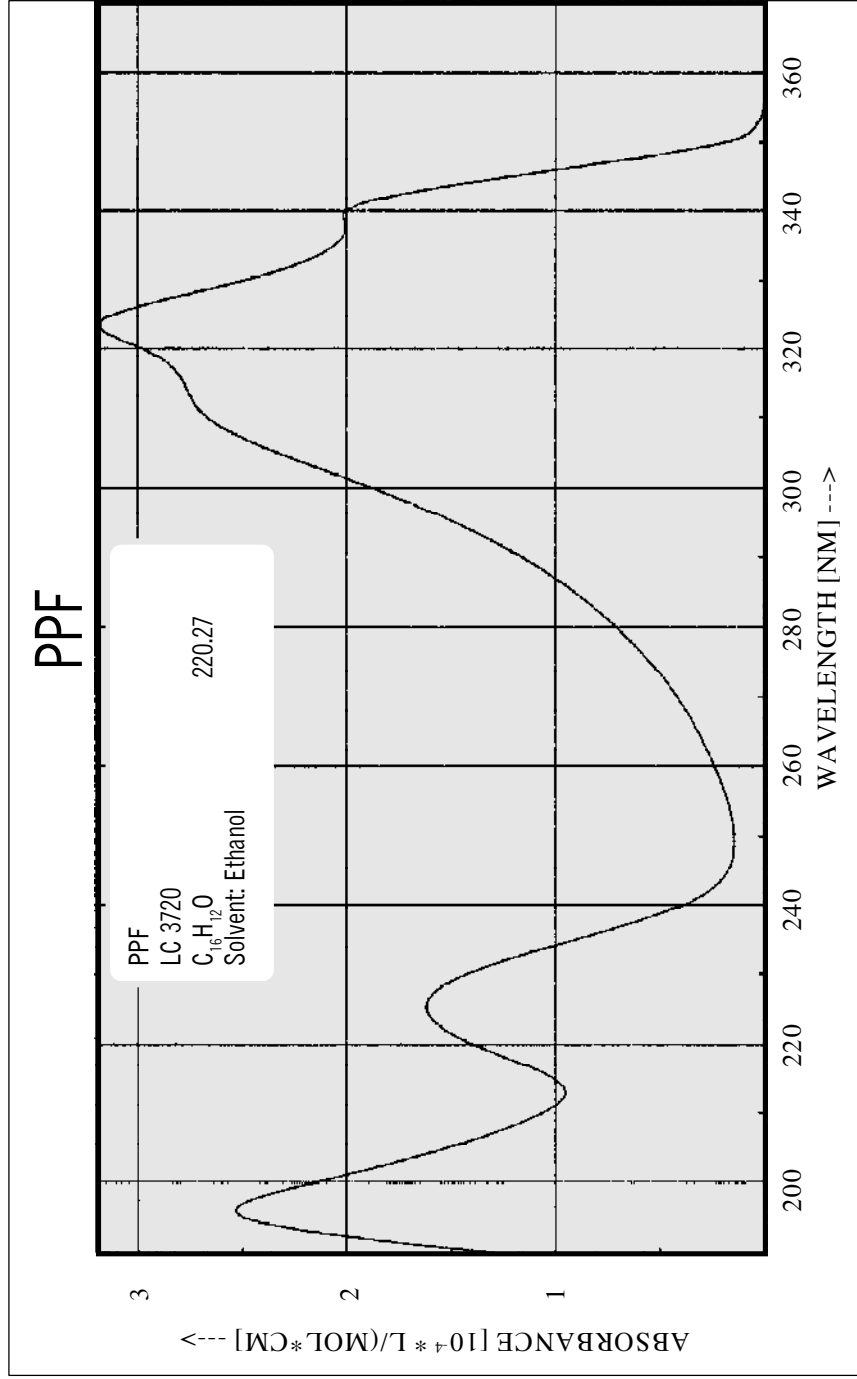
## Lasing Performance

Laser dye for pulsed operation; tunable around 380 nm.

Pump		Dye Laser Characteristics				Solvent	Ref.
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	377	-	2.3	0.22	Methanol	1, 2
Nitrogen	337	375	-	rel.	0.39	Dioxane	1
Nd:YAG, 4th	266	375	368-382	1.6	1.10	Cylohex.	3
Flashlamp	-	381	-	-	1.54	Dioxane	4

## References

1. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
2. O. Uchino et al., *Appl. Phys.* **19**, 35 (1979).
3. L. D. Ziegler, B. S. Hudson, *Opt. Commun.* **32**(1), 119 (1980).
4. H. W. Furumoto, H. L. Cecon, *IEEE J. Quant. Electron.* **QE-6**, 262 (1970).



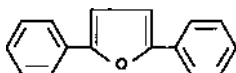


# PPF (LC 3720)

## Constitution

2,5-Diphenylfuran

C<sub>16</sub>H<sub>12</sub>O · MW: 220.27



## Characteristics

Lambdachrome® number:	3720
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in ethanol):	324 nm
Molar absorptivity:	$3.78 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	368 nm
For research and development purposes only.	

## Lasing Performance

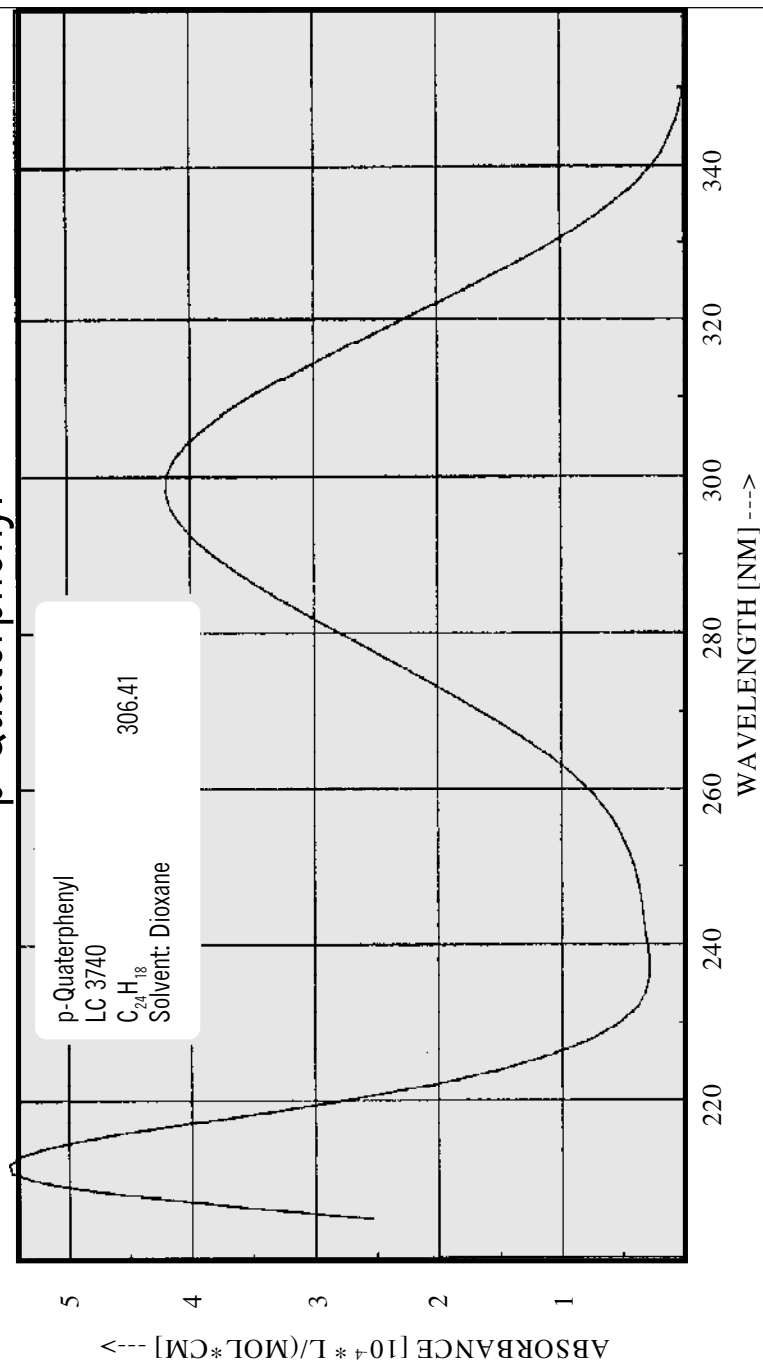
Laser dye for pulsed operation; tunable around 370 nm.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
Nitrogen	337	373	369 - 379	11	0.44	Dioxane	1,2

## References

1. H. P. Broida, S. C. Haydon, *Appl. Phys. Letters* **16**(3), 142 (1970).
2. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* **13**(5), 827 (1974).

## p-Quaterphenyl

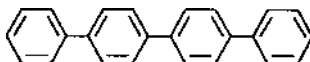


# p-Quaterphenyl (LC 3740)

## Constitution

PQP

C<sub>24</sub>H<sub>18</sub> · MW: 306.41



## Characteristics

Lambdachrome® number:	3740
CAS registry number:	135-70-6
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	297 nm
Molar absorptivity:	4.28 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum (in toluene):	374 nm
For research and development purposes only.	

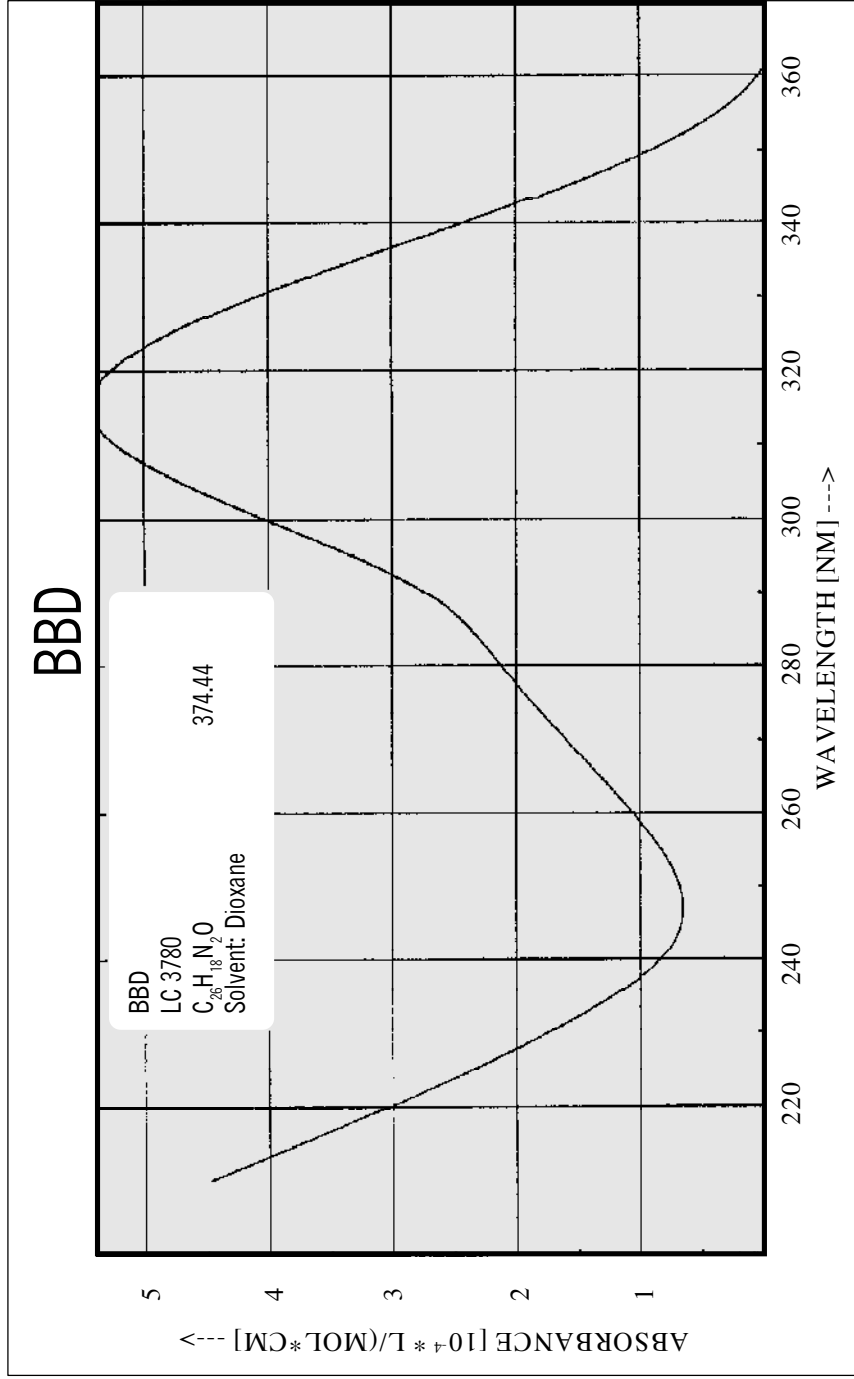
## Lasing Performance

Laser dye for pulsed operation; tunable around 370 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	371	-	20	0.09	Dioxane	1
Nitrogen	337	370	362 - 390	11	sat.	DMF	2, 3
Flashlamp	-	372	-	-	0.08	DMF	4

## References

1. P. Cassard et al., *Opt. Commun.* **38**(2), 131 (1981).
2. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* **13**(5), 827 (1974).
3. J. A. Myer, I. Itzkan, E. Kierstead, *Nature* **225**, 544 (1970).
4. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* **11**(5), 692 (1972).

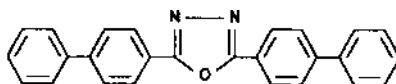


# BBD (LC 3780)

## Constitution

2,5-Bis-(4-biphenyl)-1,3,4-oxadiazol

C<sub>26</sub>H<sub>18</sub>N<sub>2</sub>O · MW: 374.44



## Characteristics

Lambdachrome® number:	3780
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	314 nm
Molar absorptivity:	5.30 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum (in dioxane):	373 nm
For research and development purposes only.	

## Lasing Performance

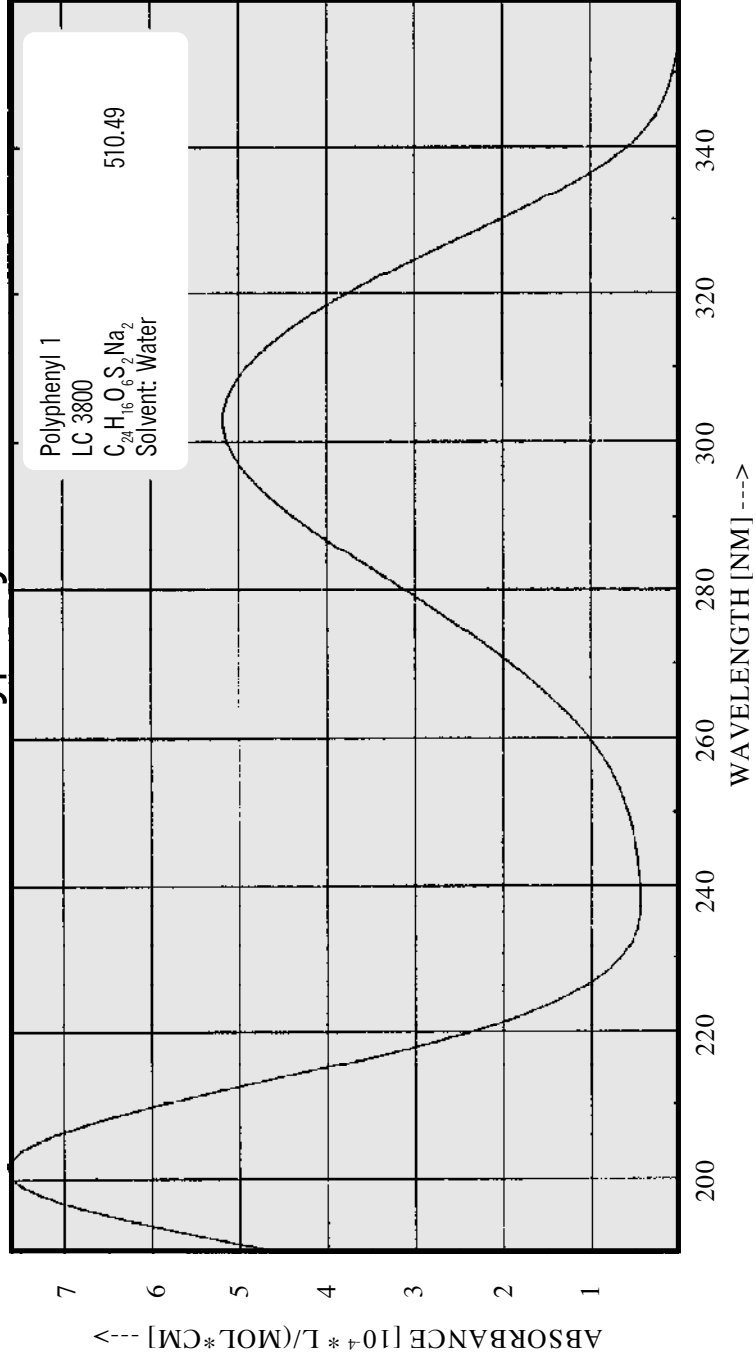
Laser dye for pulsed operation; tunable around 380 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	378	368 - 399	14	0.37	Dioxane	1, 2
Nitrogen	337	375	372 - 405	19	0.74	Dioxane	2, 3
Flashlamp	-	377	-	-	0.74	Dioxane	4

## References

1. P. Cassard et al., *Opt. Commun.* **38**(2), 131 (1981).
2. O. Uchino et al., *Appl. Phys.* **19**, 35 (1979).
3. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* **13**(5), 827 (1974).
4. M. Maeda et al., *IEEE J. Quant. Elec tron.* **QE-13**, 65 (1977).

# Polyphenyl 1

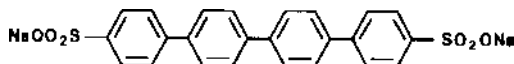


# Polyphenyl 1 (LC 3800)

## Constitution

p-Quaterphenyl-4,4'''-disulfonicacid Disodiumsalt

$C_{24}H_{16}O_6S_2Na_2$  · MW: 510.49



## Characteristics

Lambdachrome® number:	3800
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in water):	308 nm
Molar absorptivity:	$5.19 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

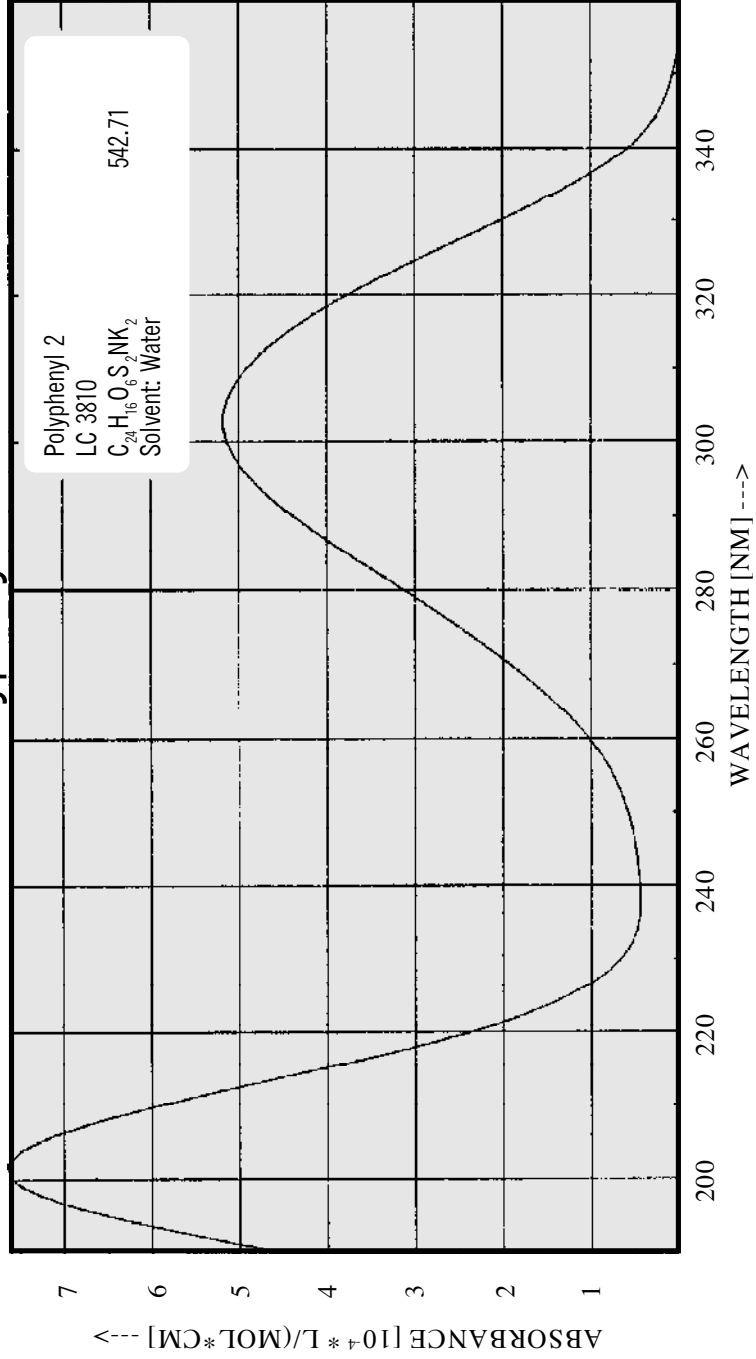
Efficient laser dye for pulsed and CW operation; tunable around 380 nm.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	381	363 - 408	12	0.20	Eg	1, 2, 3
Nitrogen	337	380	362 - 411	rel.	0.36	Eg	3
CW, Ar <sup>+</sup>	UV	382	362 - 412	-	1.25	Eg	3

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 402 (1981).
3. W. Hüffer et al., *Opt. Commun.* **33**(1), 85 (1980).

# Polyphenyl 2



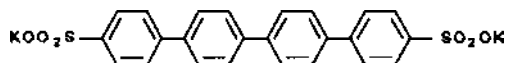


# Polyphenyl 2 (LC 3810)

## Constitution

p-Quaterphenyl-4,4'''-disulfonicacid Dipotassiumsalt

$C_{24}H_{16}O_6S_2NK_2$  · MW: 542.71



## Characteristics

Lambdachrome® number:	3810
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in water):	308 nm
Molar absorptivity:	$5.19 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

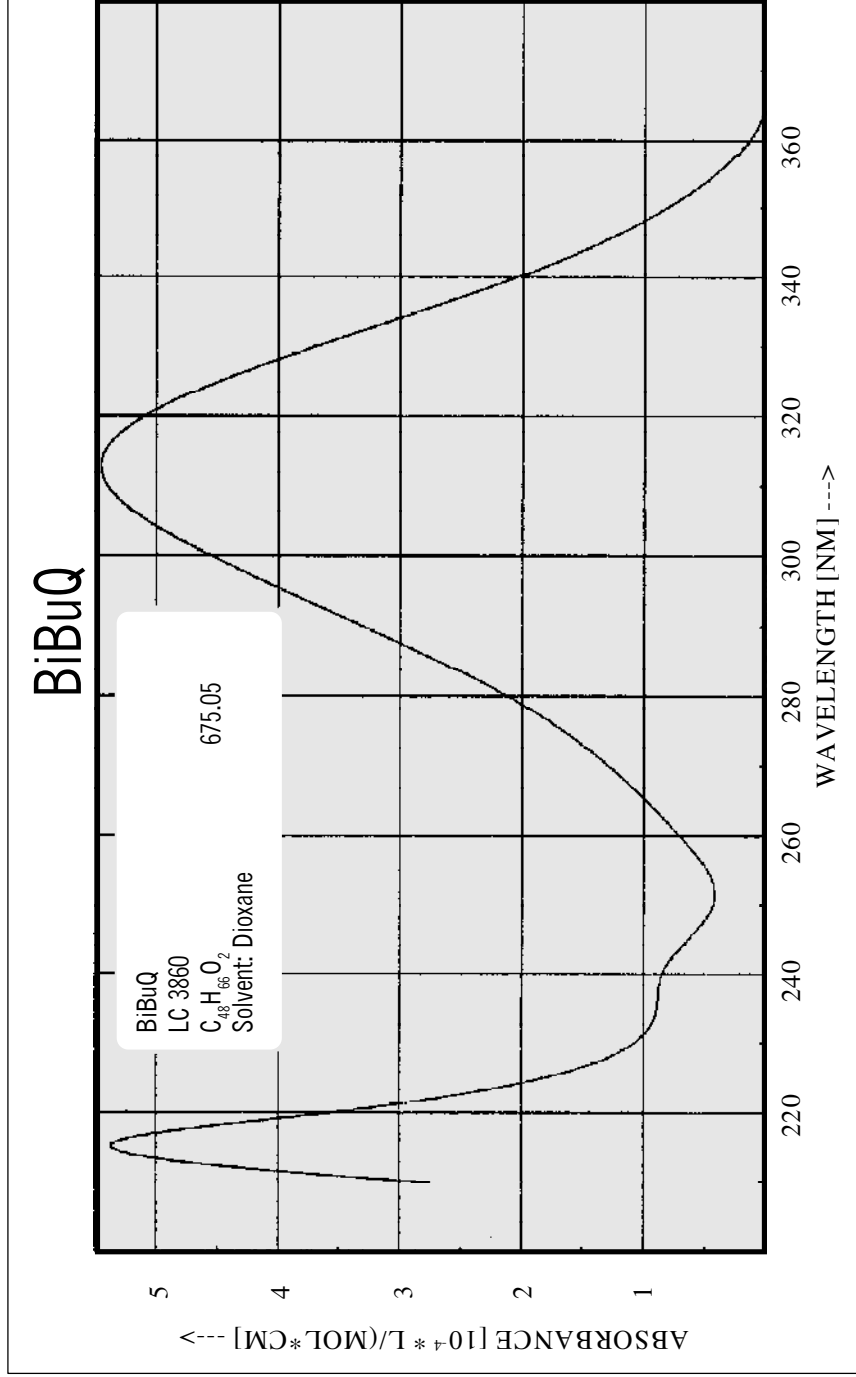
## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 390 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	382	363 - 400	10	0.25	Eg	1
CW, Ar <sup>+</sup>	UV	384	370 - 406	-	2.0	Eg	1

## References

1. Lambda Physik, *Wall Chart* 1996.



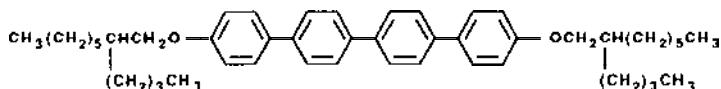
# BiBuQ (LC 3860)

## Constitution

4,4'''-Bis-(2-butyloctyloxy)-p-quaterphenyl

BBQ • Pilot 386

$C_{48}H_{66}O_2$  · MW: 675.05



## Characteristics

Lambdachrome® number:	3860
CAS registry number:	18434-08-7
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	313 nm
Molar absorptivity:	$5.45 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-

For research and development purposes only.

## Lasing Performance

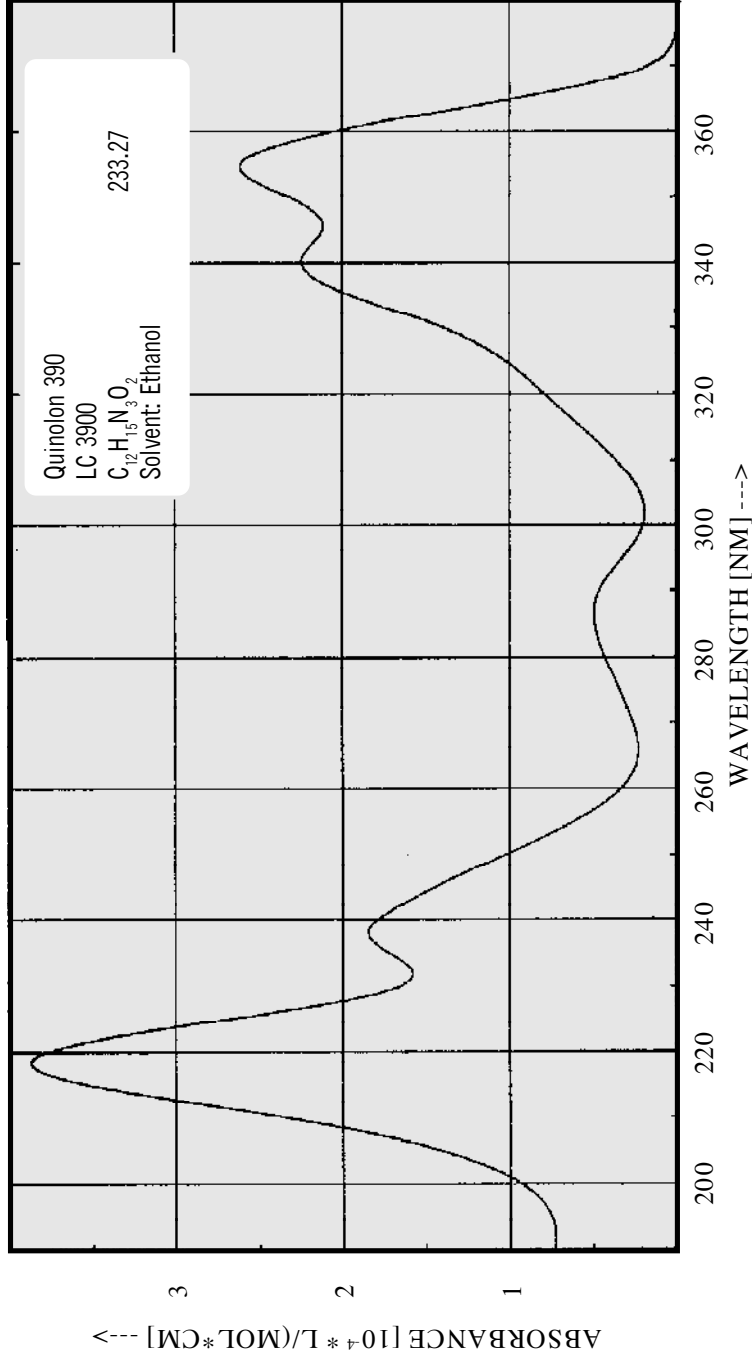
Efficient laser dye for pulsed operation; tunable around 390 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	388	367 - 405	11	0.25	Dioxane	1, 2, 3
Nitrogen	337	383	364 - 405	rel.	0.41	Dioxane	3, 4
Nd:YAG, 3rd	355	392	380 - 410	-	1.34	EtOH/Tol	5
Flashlamp	-	-	389 - 395	-	-	DMF	6

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38(5,6)**, 402 (1981).
3. F. Bos, *Appl. Optics* **20(20)**, 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. K. Azuma et al., *Jpn. J. Appl. Phys.* **18(1)**, 209 (1979).
6. P. R. Hammond et al., *Appl. Phys.* **9**, 67 (1976).

## Quinolol 390

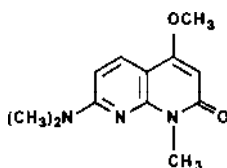


# Quinolon 390 (LC 3900)

## Constitution

7-Dimethylamino-1-methyl-4-methoxy-8-azaquinolone-2

$C_{12}H_{15}N_3O_2$  · MW: 233.27



## Characteristics

Lambdachrome® number:	3900
CAS registry number:	119883-58-8
Appearance:	white, crystalline solid
Absorption maximum (in ethanol):	355 nm
Molar absorptivity:	$2.52 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

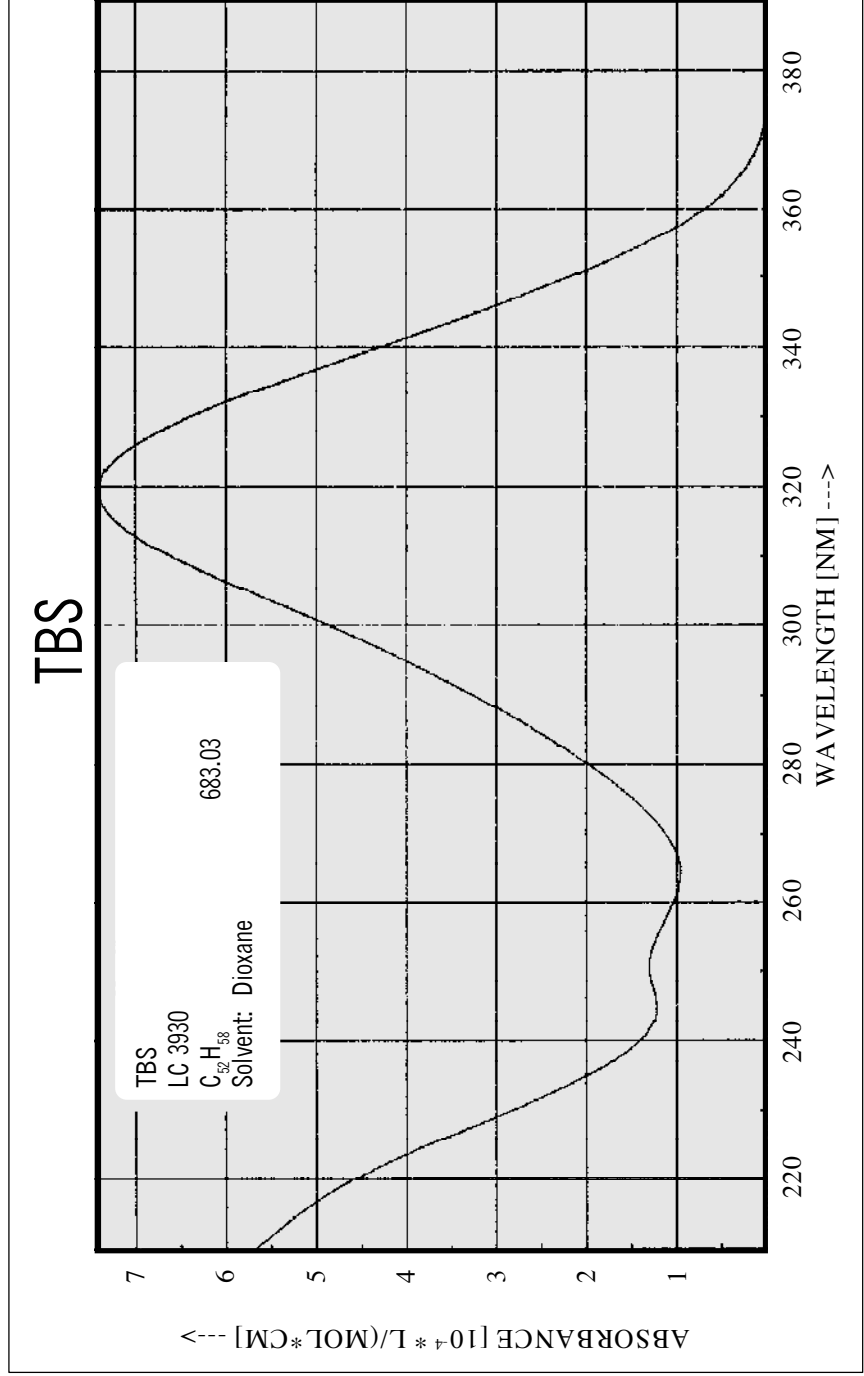
## Lasing Performance

Laser dye for pulsed operation; tunable around 390 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nd:YAG, 3rd	355	390	384 - 394	4	0.25	Methanol	1
Flashlamp	-	390	-	-	1.15	Ethanol	2

## References

1. Lambda Physik, *Wall Chart* 1996.
2. P. R. Hammond et al., *Appl. Phys.* **8**, 315 (1975).

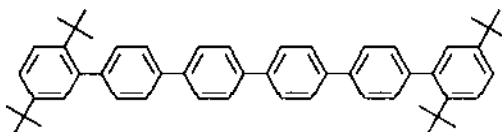


# TBS (LC 3930)

## Constitution

3,5,3',5'-Tetra-*t*-butyl-*p*-sexiphenyl

C<sub>52</sub>H<sub>58</sub> · MW: 683.03



## Characteristics

Lambdachrome® number:	3930
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	320 nm
Molar absorptivity:	7.49 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum:	-

For research and development purposes only.

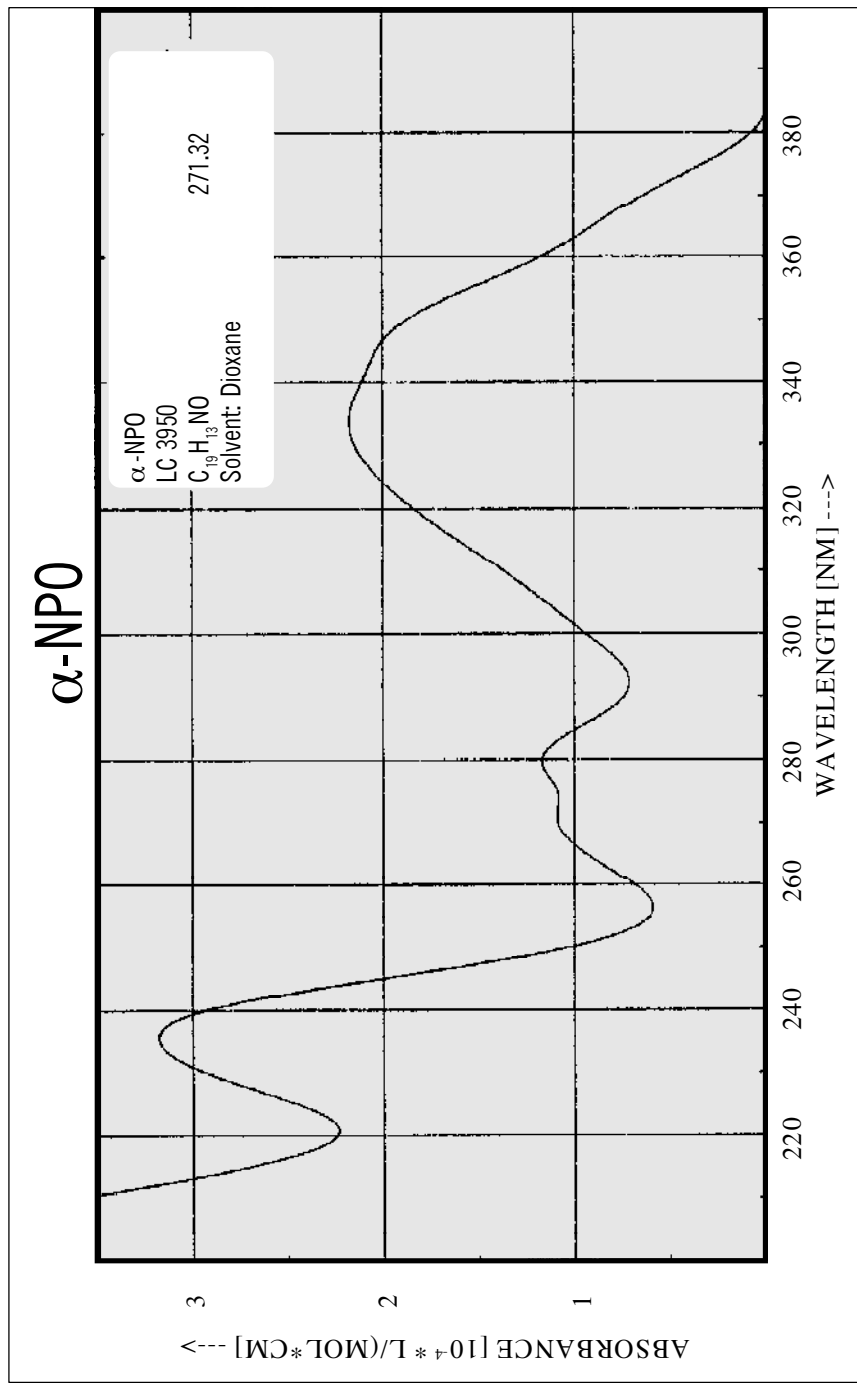
## Lasing Performance

Very stable and efficient laser dye for pulsed operation; tunable around 390 nm.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	393	365 - 410	11	0.16	Cyclohex	1

## References

1. Lambda Physik, *Data Sheet*.



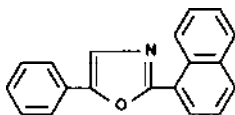


# $\alpha$ -NPO (LC 3950)

## Constitution

2-(1-Naphthyl)-5-phenyloxazol

$C_{19}H_{13}NO$  · MW: 271.32



## Characteristics

Lambdachrome® number:	3950
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	333 nm
Molar absorptivity:	$2.15 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in dioxane):	396 nm
For research and development purposes only.	

## Lasing Performance

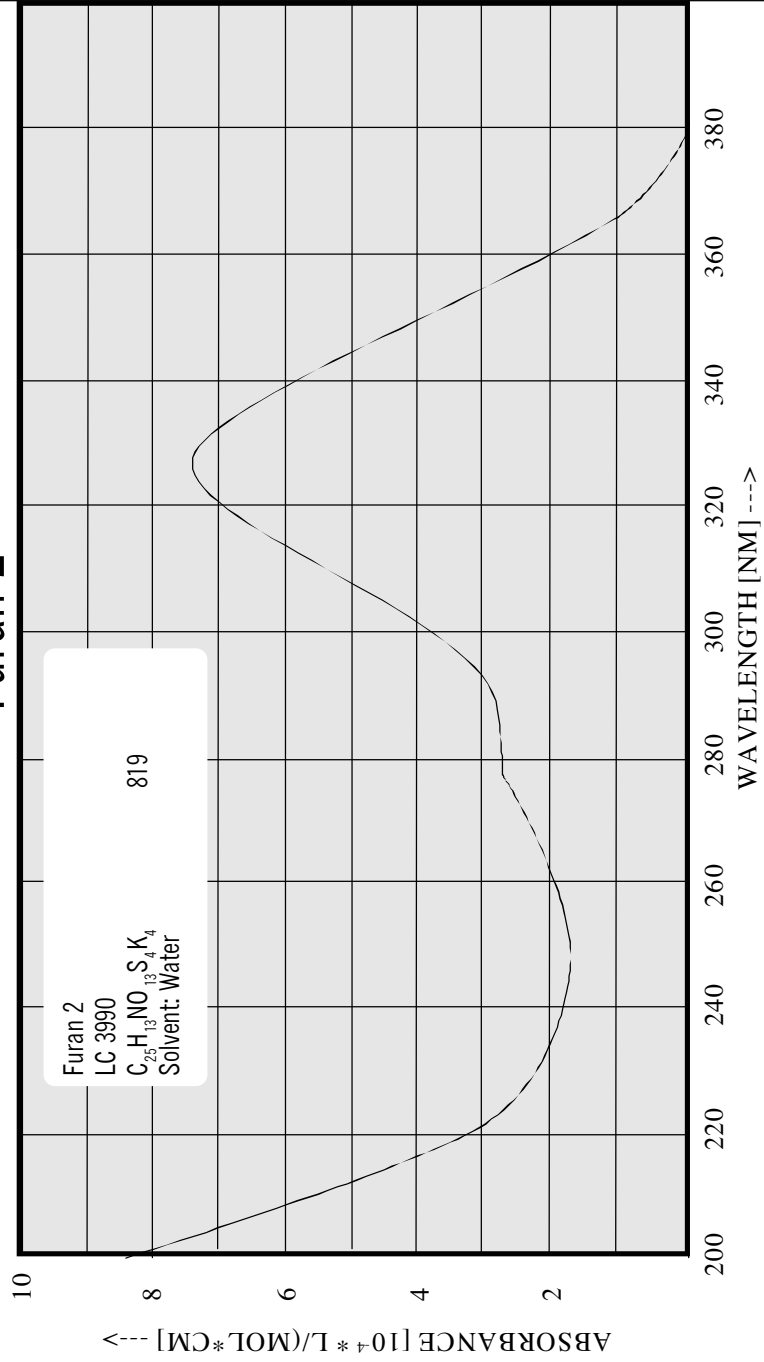
Efficient laser dye for pulsed operation; tunable around 400 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	393	387 - 400	6	0.32	Cyclohex.	1
Nitrogen	337	400	391 - 425	rel.	0.67	Dioxane	2, 3
Nd:YAG, 3rd	355	400	-	6.5	1.00	Toluene	4
Flashlamp	-	400	-	-	0.07	Ethanol	5

## References

1. Lambda Physik, *Data Sheet*.
2. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
3. A. Dienes, *Appl. Phys.* **7**, 135 (1975).
4. G. A. Abakumov et al. *JETP Letters* **9**, 9 (1969).
5. H. W. Furumoto et al. *IEEE J. Quant. Electron.* **QE-6**(5), 262 (1970).

## Furan 2

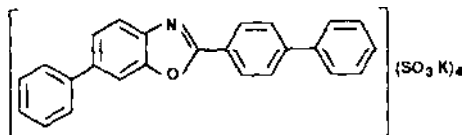


# Furan 2 (LC 3990)

## Constitution

2-(4-Biphenyl)-6-phenylbenzoxazotetrasulfonicacid Potassium Salt

$C_{25}H_{13}NO_{13}S_4K_4$  · MW: 819



## Characteristics

Lambdachrome® number:	3990
CAS registry number:	-
Appearance:	light yellow, crystalline solid
Absorption maximum (in water):	330 nm
Molar absorptivity:	$7.79 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in dioxane):	396 nm
For research and development purposes only.	

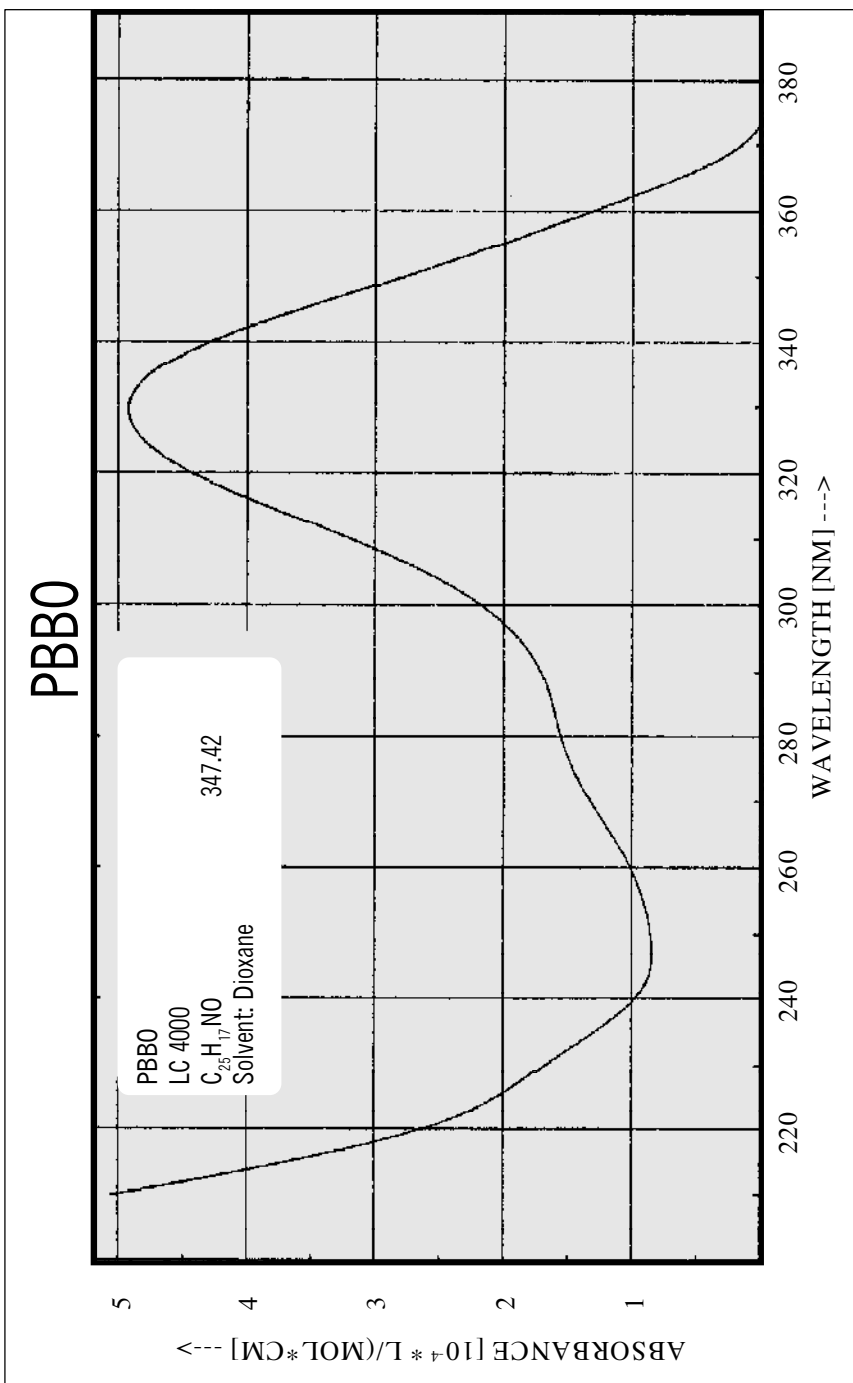
## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 400 nm.

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		399	388 - 426	8	0.50	Methanol	1
Nd:YAG, 3rd	355		402	392 - 422	15	0.5	Methanol	1

## References

1. Lambda Physik, *Wall Chart* 1996.

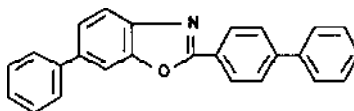


# PBBO (LC 4000)

## Constitution

2-(4-Biphenyl)-6-phenylbenzoxazol-1,3

C<sub>25</sub>H<sub>17</sub>NO · MW: 347.42



## Characteristics

Lambdachrome® number:	4000
CAS registry number:	17064-47-0
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	327 nm
Molar absorptivity:	$4.89 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	403 nm
For research and development purposes only.	

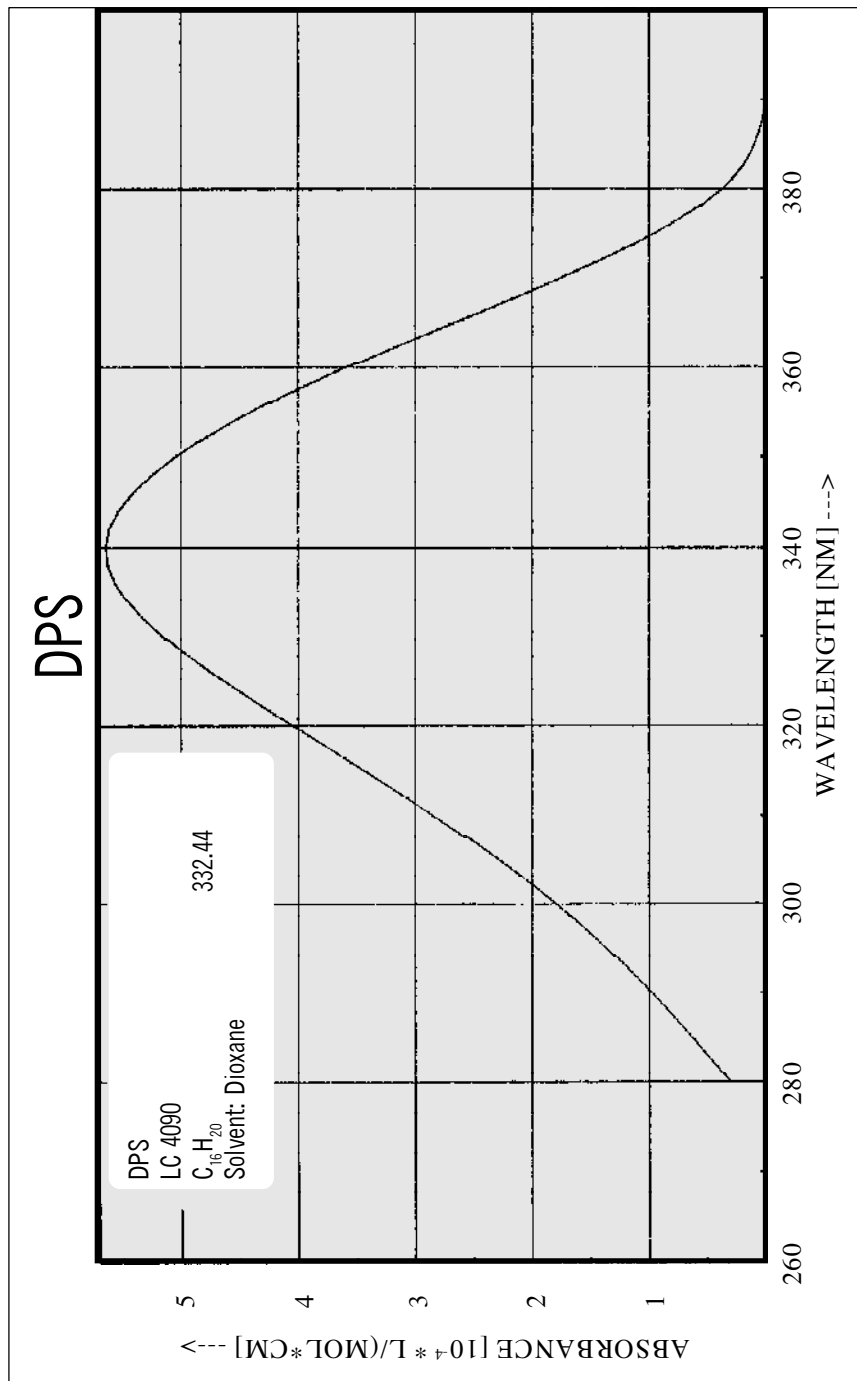
## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 400 nm.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	396	386 - 420	7	0.40	Dioxane	1, 2
Nitrogen	337	395	385 - 420	rel.	0.15	Dioxane	2, 3

## References

1. Lambda Physik, *Wall Chart* 1996.
2. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
3. Lambda Physik, *Data Sheet*.



# DPS (LC 4090)

## Constitution

4,4'-Diphenylstilbene

Pilot 409

$C_{16}H_{20}$  · MW: 332.44



## Characteristics

Lambdachrome® number:	4090
CAS registry number:	26569-48-2
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	340 nm
Molar absorptivity:	$5.65 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

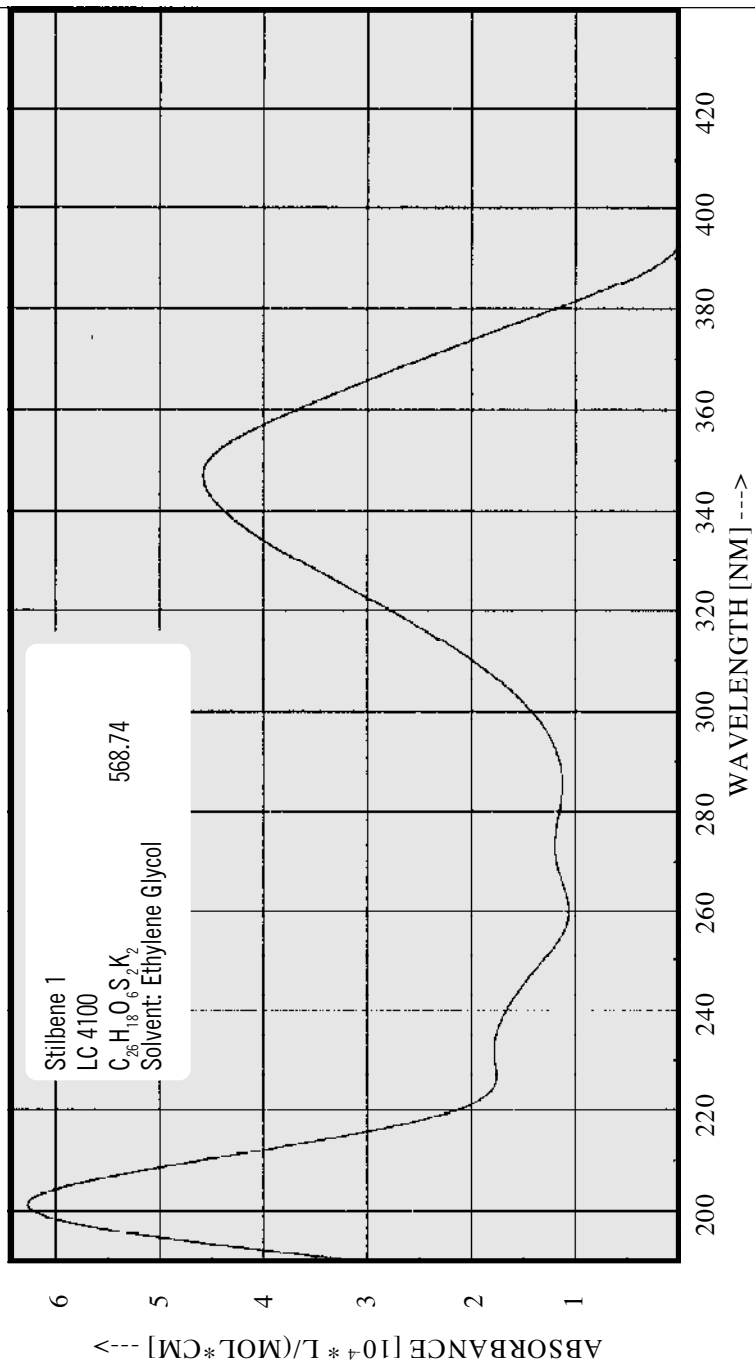
Efficient laser dye for pulsed operation; tunable around 400 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	406	399 - 415	11	0.25	Dioxane	1, 2
Nitrogen	337	404	394 - 416	rel.	0.12	Dioxane	3, 4
Nd:YAG, 3rd	355	408	-	-	-	Tol.	5
Flashlamp	-	409	406 - 411	-	sat.	DMF	6, 7

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 402 (1981).
3. Lambda Physik, *Data Sheet*.
4. A. Dienes, *Appl. Phys.* **7**, 135 (1975).
5. V. D. Kotzubanov et al., *Opt. Spectrosc.* **25**, 406 (1968).
6. P. R. Hammond et al., *Appl. Phys.* **9**, 67 (1976).
7. H. W. Furumoto et al., *IEEE J. Quant. Electron.* **QE-6**, 262 (1970).

# Stilbene 1





# Stilbene 1 (LC 4100)

## Constitution

[1,1'-Biphenyl]-4-sulfonic acid, 4',4''-1,2-ethene-diylbis-, dipotassium salt

$C_{26}H_{18}O_6S_2K_2$  · MW: 568.74



## Characteristics

Lambdachrome® number:	4100
CAS registry number:	74758-59-1
Appearance:	slightly yellow, crystalline solid
Absorption maximum (in Eg.):	350 nm
Molar absorptivity:	$4.55 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

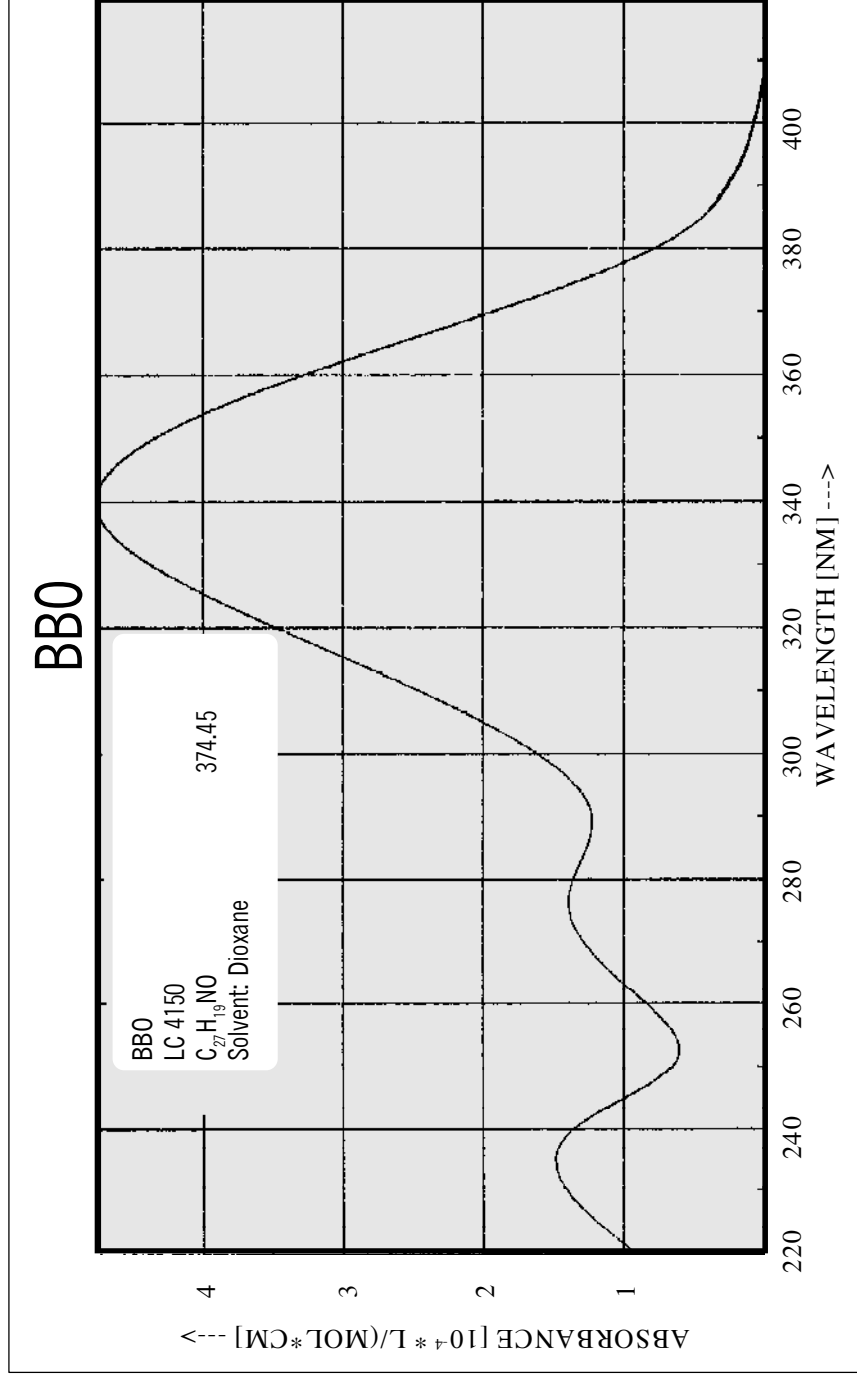
## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 415 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	416	405 - 428	6	0.50	Eg.	1, 2, 3
Nitrogen	337	417	405 - 446	rel.	0.20	Eg.	4
CW, Ar <sup>+</sup>	UV	415	403 - 428	-	0.75	Eg.	1, 4, 5, 6

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38(5,6)**, 402 (1981).
3. F. Bos, *Appl. Optics* **20(20)**, 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. T. F. Johnston, R. H. Brady, W. Proffitt, *Appl. Optics* **21(13)**, 2307 (1982).
6. W. Hüffer et al., *Opt. Commun.* **28(3)**, 353 (1979).

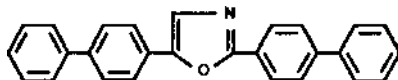


# BBO (LC 4150)

## Constitution

2,5-Bis-(4-biphenyl)-oxazol

C<sub>27</sub>H<sub>19</sub>NO · MW: 374.45



## Characteristics

Lambdachrome® number:	4150
CAS registry number:	2083-09-2
Appearance:	white, crystalline solid
Absorption maximum (in dioxane):	340 nm
Molar absorptivity:	$4.76 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in toluene):	412 nm
For research and development purposes only.	

## Lasing Performance

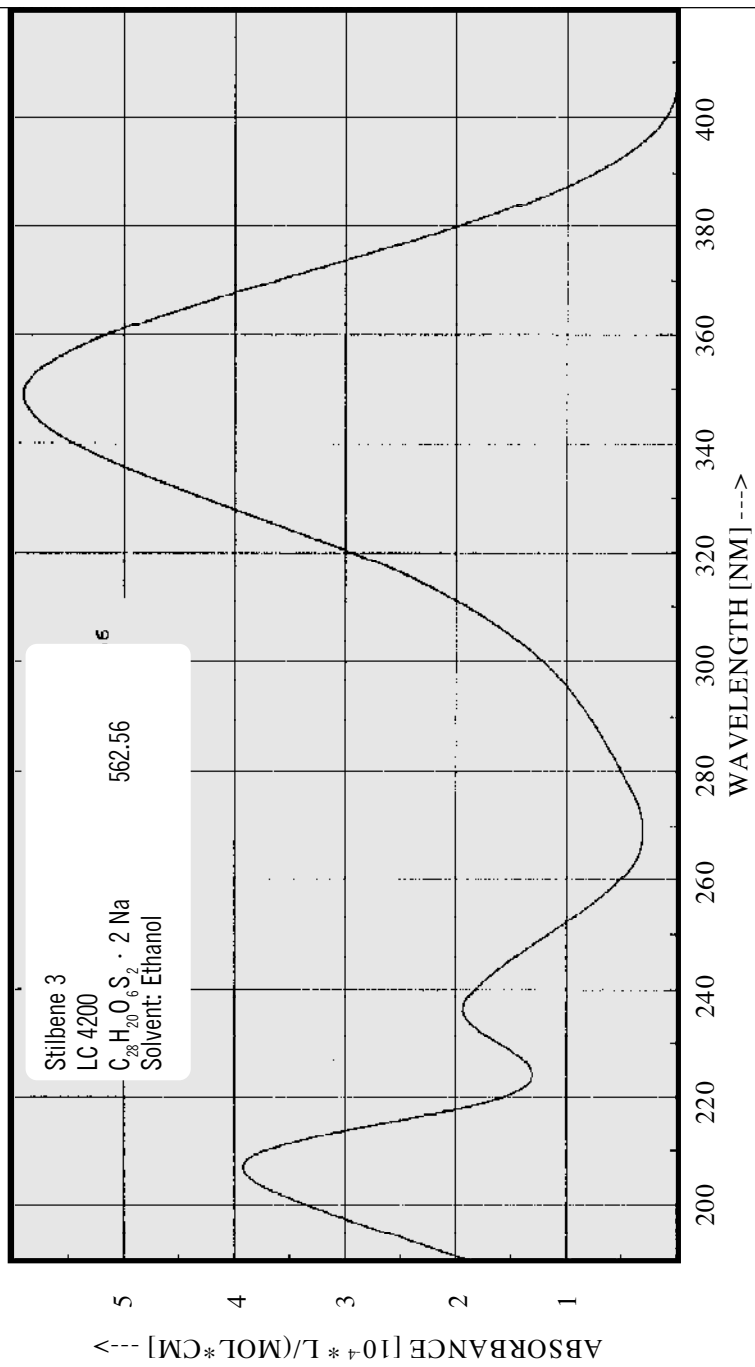
Laser dye for pulsed operation; tunable around 410 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nitrogen	337	408	401 - 419	14	0.74	Dioxane	1, 2
Nd:YAG, 3rd	355	409	-	8	0.96	Tol.	3
Flashlamp	-	410	-	-	0.37	Dioxane	4, 5

## References

1. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* **13**(5), 827 (1974).
2. H. P. Broida, S. C. Haydon, *Appl. Phys. Letters* **16**(3), 142 (1970).
3. G. A. Abakumov et al., *JETP Letters* **9**, 9 (1969).
4. H. W. Furumoto et al., *IEEE J. Quant. Electron.* **QE-6**(5), 262 (1970).
5. M. Maeda, Y. Miyazoe, *Jpn. J. Appl. Phys.* **11**(5), 692 (1972).

## Stilbene 3

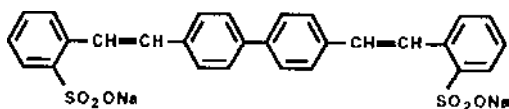


# Stilbene 3 (LC 4200)

## Constitution

2,2'-([1,1'-Biphenyl]-4,4'-diyl-di-2,1-ethenediyl)-bis-benzenesulfonic acid  
Disodium Salt  
Stilbene 420

$C_{28}H_{20}O_6S_2 \cdot 2 Na$  · MW: 562.56



## Characteristics

Lambdachrome® number:	4200
CAS registry number:	27344-41-8
Appearance:	yellow, crystalline solid
Absorption maximum (in Ethanol.):	350 nm
Molar absorptivity:	$5.89 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

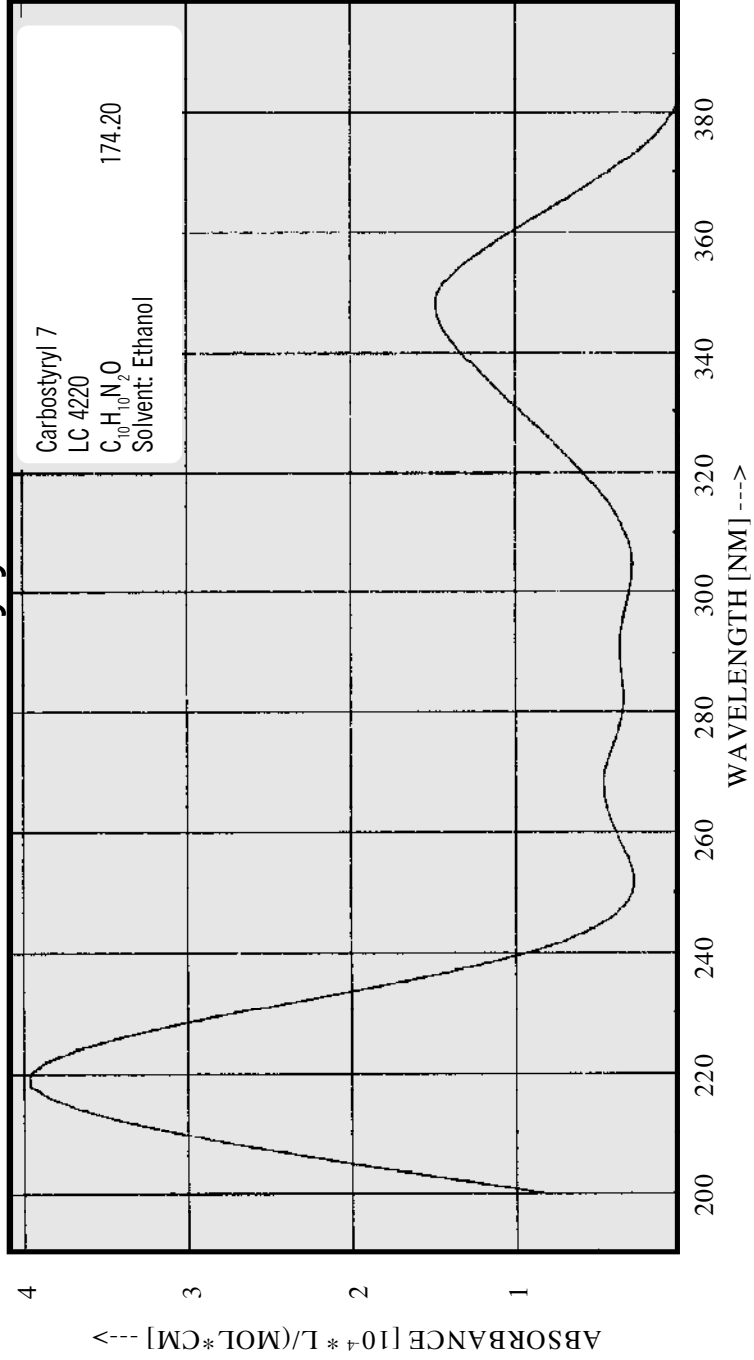
Efficient laser dye for pulsed and CW operation; tunable around 425 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	425	412 - 443	9	0.65	Methanol	1, 2, 3
Nitrogen	337	424	408 - 457	rel.	0.22	Methanol	3, 4
Nd:YAG, 3rd	355	428	415 - 439	15	0.25	Methanol	1, 5
CW, Ar <sup>+</sup>	UV	435	410 - 485	-	1.0	Eg.	1, 4, 6, 7

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 402 (1981).
3. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. D. M. Guthals, J. W. Nibler, *Opt. Commun.* **29**(3), 322 (1979).
6. T. F. Johnston, R. H. Brady, W. Proffitt, *Appl. Optics* **21**(13), 2307 (1982).
7. J. Kuhl et al., *Opt. Commun.* **24**(3), 251 (1978).

## Carbostyryl 7



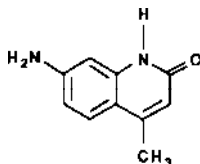
# Carbostyryl 7 (LC 4220)

## Constitution

7-Amino-4-methylcarbostyryl

Carbostyryl 124

$C_{10}H_{10}N_2O$  · MW: 174.20



## Characteristics

Lambdachrome® number:	4220
CAS registry number:	-
Appearance:	white, crystalline solid
Absorption maximum (in ethanol):	350 nm
Molar absorptivity:	$1.46 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	400 nm
For research and development purposes only.	

## Lasing Performance

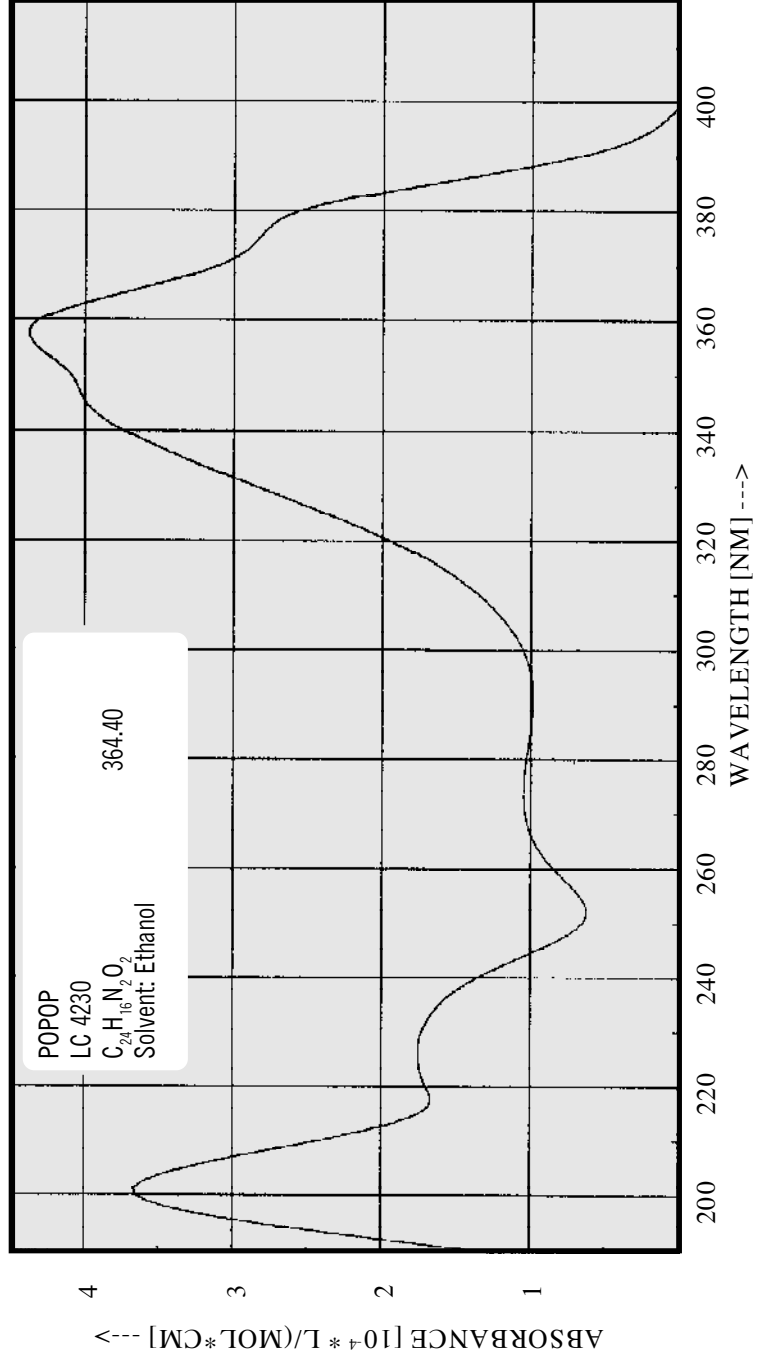
Laser dye for pulsed operation; tunable around 410 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	413	408 - 420	-	-	Ethanol	1, 3

## References

1. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).
2. R. Srinivasan, *IEEE J. Quant. Electron.* **QE-5**, 552 (1969).

# POPOP



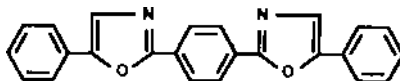


# POPOP (LC 4230)

## Constitution

1,4-Di[2-(5-phenyloxazolyl)]benzene  
Pilot 423

$C_{24}H_{16}N_2O_2$  · MW: 364.40



## Characteristics

Lambdachrome® number: 4230  
CAS registry number: 1806-34-4  
Appearance: yellow, crystalline solid  
Absorption maximum (in ethanol): 358 nm  
Molar absorptivity:  $4.43 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in ethanol): 425 nm  
For research and development purposes only.

## Lasing Performance

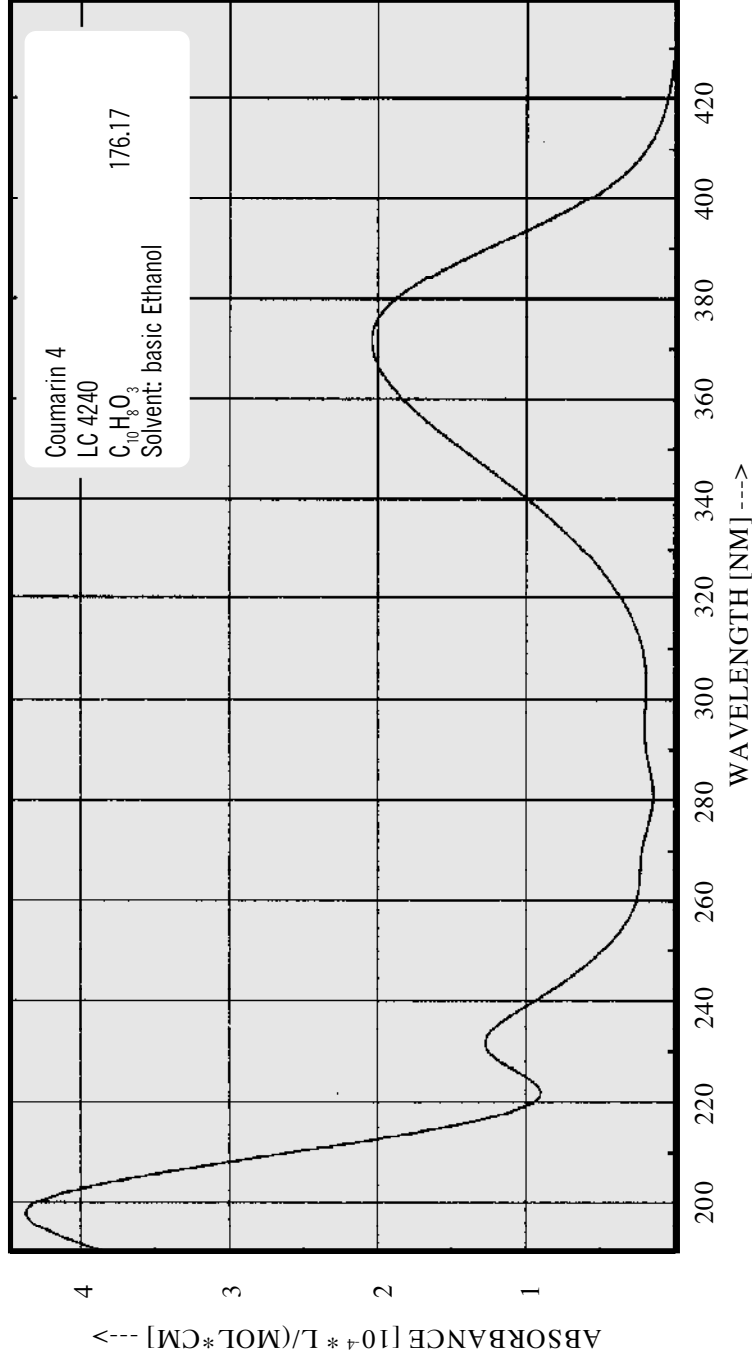
Efficient laser dye for pulsed operation; tunable around 420 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	421	411 - 446	6.5	0.42	Dioxane	1
Nitrogen	337	421	412 - 454	rel.	0.39	EtOH/Tol.	1,2
Nd:YAG, 3rd	355	417	-	-	-	Tol.	3
Flashlamp	-	419	-	-	0.12	Ethanol	4, 5

## References

1. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
2. A. Dienes, *Appl. Phys.* **7**, 135 (1975).
3. V. D. Kotzubanov et al., *Opt. Spectrosc.* **25**, 406 (1968).
4. H. W. Furumoto, H. L. Ceccon, *J Appl Phys.* **40**, 4204 (1969).
5. P. R. Hammond et al., *Appl. Phys.* **9**, 67 (1976).

## Coumarin 4



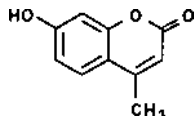
# Coumarin 4 (LC 4240)

## Constitution

7-Hydroxy-4-methylcoumarin

Umbelliferon 47

$C_{10}H_8O_3$  · MW: 176.17



## Characteristics

Lambdachrome® number:	4240
CAS registry number:	90-33-5
Appearance:	colorless, crystalline solid
Absorption maximum (in bas. ethanol):	372 nm
Molar absorptivity:	$2.10 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	445 nm
For research and development purposes only.	

## Lasing Performance

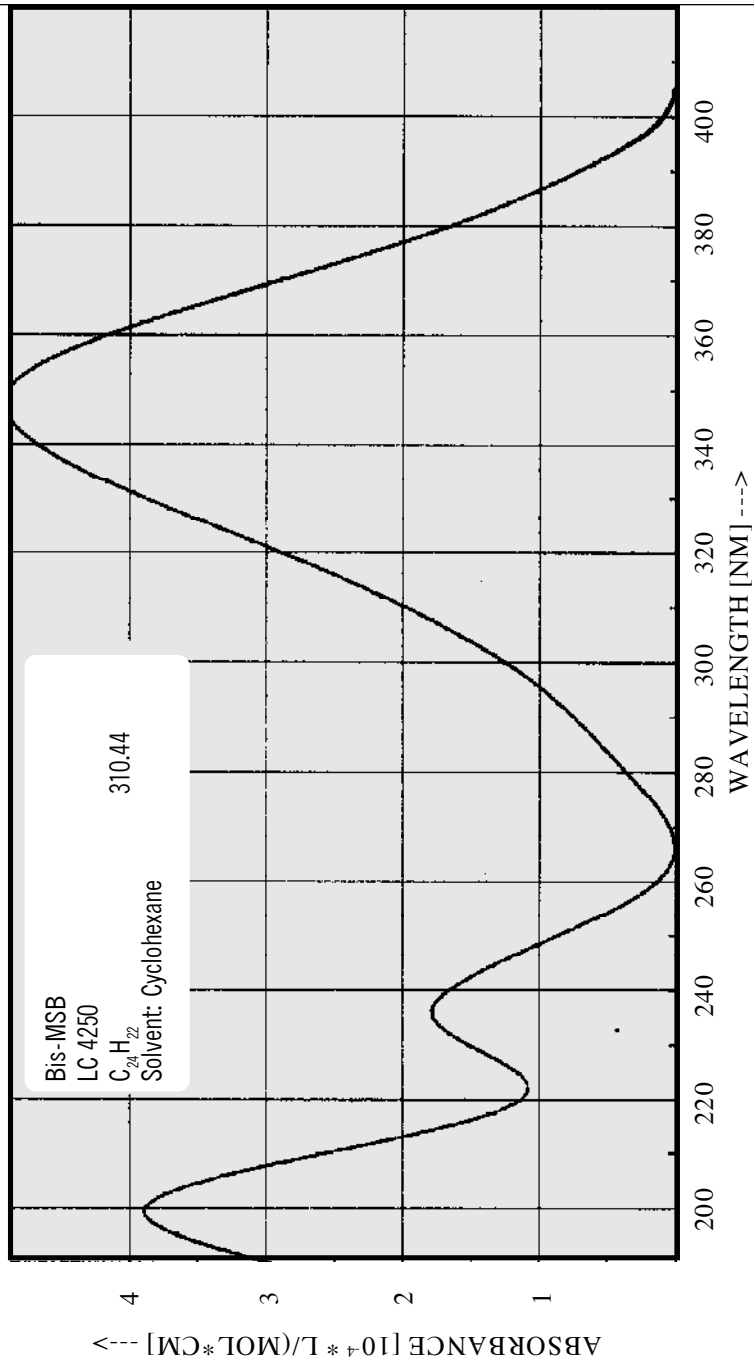
Laser dye for pulsed and CW operation; tunable around 450 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
Nitrogen	337	-	370 - 580	-		various	1,2,3
Flashlamp	-	454	-	-	0.51	Ethanol	4,5
CW, Ar <sup>+</sup>	UV	460	460 - 560	-	0.51	Eg.	6

## References

1. A. Dienes, *Appl. Phys.* **7**, 135 (1975).
2. C. V. Shank et al., *Appl. Phys. Letters* **16**(10), 405 (1970).
3. A. Dienes et al., *IEEE J. Quant. Electr.* **QE-9**, 833 (1973).
4. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).
5. B. B. Snavey et al., *Appl. Phys. Letters* **11**(9), 275 (1967).
6. J. M. Yarborough, *Appl. Phys. Letters* **24**(12), 629 (1974).

# Bis-MSB

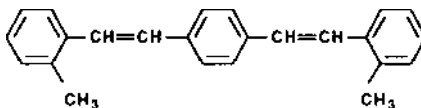


# Bis-MSB (LC 4250)

## Constitution

p-Bis(o-methylstyryl)-benzene

C<sub>24</sub>H<sub>22</sub> · MW: 310.44



## Characteristics

Lambdachrome® number:	4250
CAS registry number:	13280-61-0
Appearance:	yellow, crystalline solid
Absorption maximum (in cyclohexane):	350 nm
Molar absorptivity:	$4.88 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	418 nm
For research and development purposes only.	

## Lasing Performance

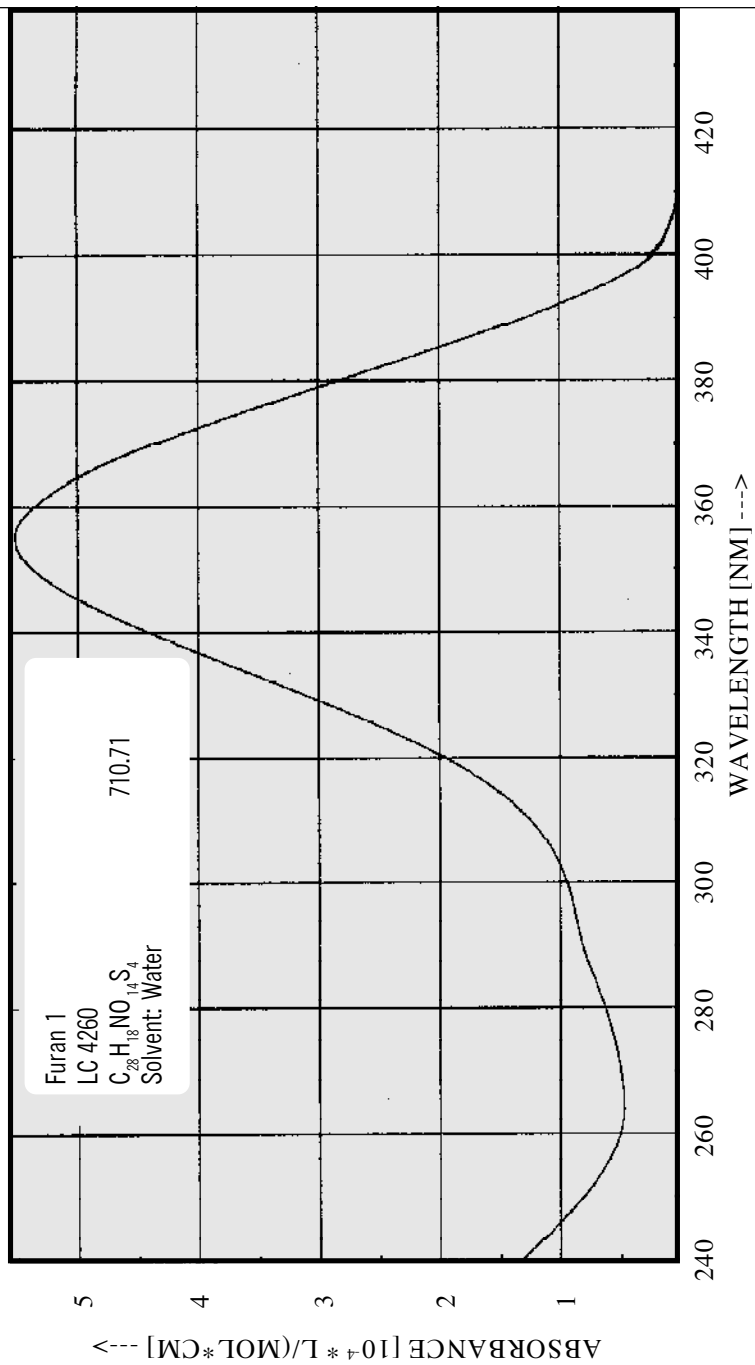
Efficient laser dye for pulsed operation; tunable around 400 nm.

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		423	414 - 428	8.3	0.24	Dioxane	1
Nitrogen	337		421	412 - 435	rel.	0.14	Dioxane	2, 3
Nd:YAG, 3rd	355		-	-	-	1.08	Dioxane	4
Flashlamp	-		420	-	-	0.12	Toluene	5

## References

1. Lambda Physik, *Wall Chart* 6/83.
2. O. Uchino et al., *Appl. Phys.* **19**, 35 (1979).
3. Lambda Physik, *Data Sheet*.
4. A. J. Cox et al., *Appl. Phys. Letters* **31**(6), 389 (1977).
5. H. W. Furumoto et al. *J. Appl. Phys.* **40**, 4204 (1969).

# Furan 1

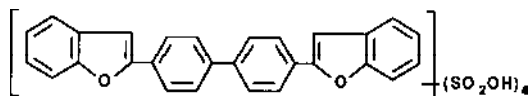


# Furan 1 (LC 4260)

## Constitution

Benzofuran,2,2'-[1,1'-biphenyl]-4,4'-diyl-bis-tetrasulfonic acid (tetrasodium salt)

$C_{28}H_{18}NO_{14}S_4$  · MW: 710.71



## Characteristics

Lambdachrome® number:	4260
CAS registry number:	-
Appearance:	colorless, crystalline solid
Absorption maximum (in water):	355 nm
Molar absorptivity:	$5.51 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

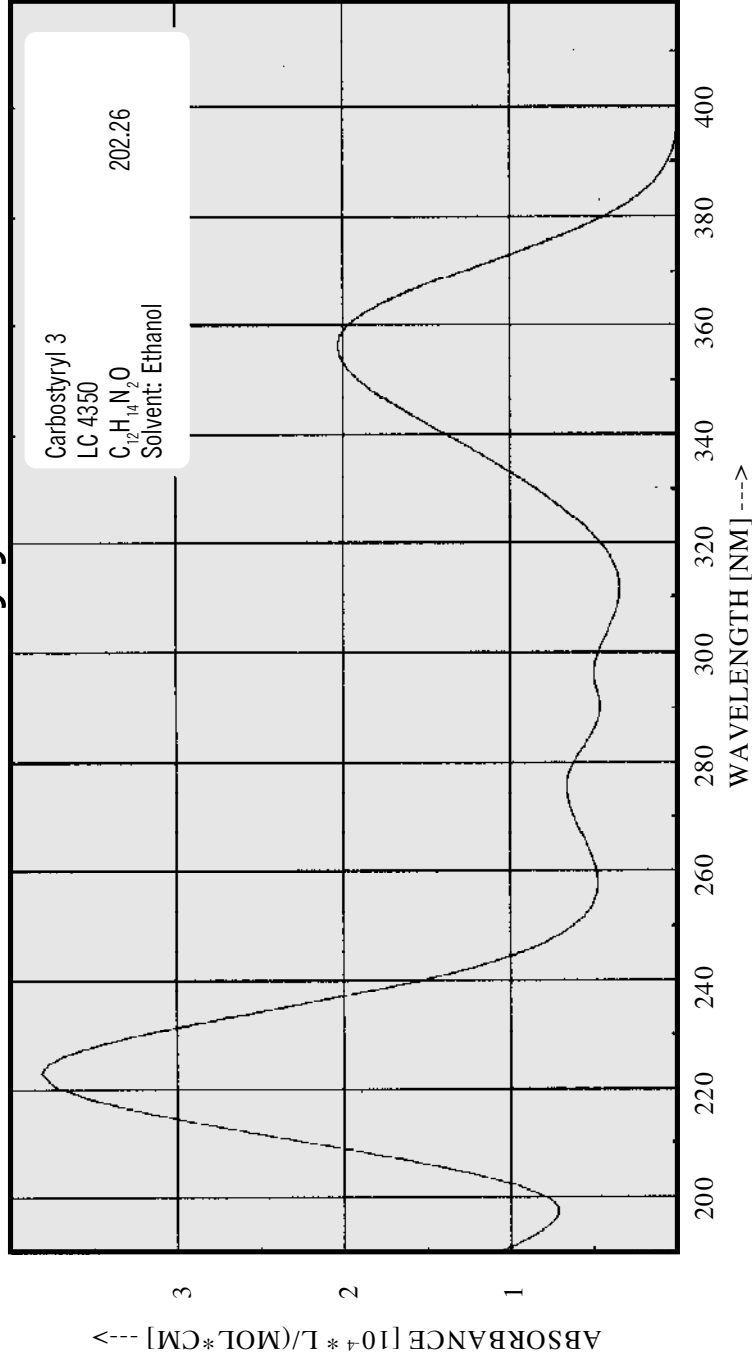
Laser dye for pulsed operation; tunable around 420 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	421	410 - 435	10	0.60	Methanol	1
Nitrogen	337	425	414 - 445	3.5	1.35	Methanol	1
Nd:YAG, 3rd	355	421	410 - 435	10	0.26	Methanol	2

## References

1. Lambda Physik, *Data Sheet*.
2. Lambda Physik, *Wall Chart* 1996.

## Carbostyryl 3



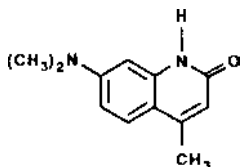


# Carbostyryl 3 (LC 4350)

## Constitution

7-Dimethylamino-4-methylquinolon-2

$C_{12}H_{14}N_2O$  · MW: 202.26



## Characteristics

Lambdachrome® number:	4350
CAS registry number:	-
Appearance:	colorless, crystalline solid
Absorption maximum (in ethanol):	360 nm
Molar absorptivity:	$2.03 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	425 nm

For research and development purposes only.

## Lasing Performance

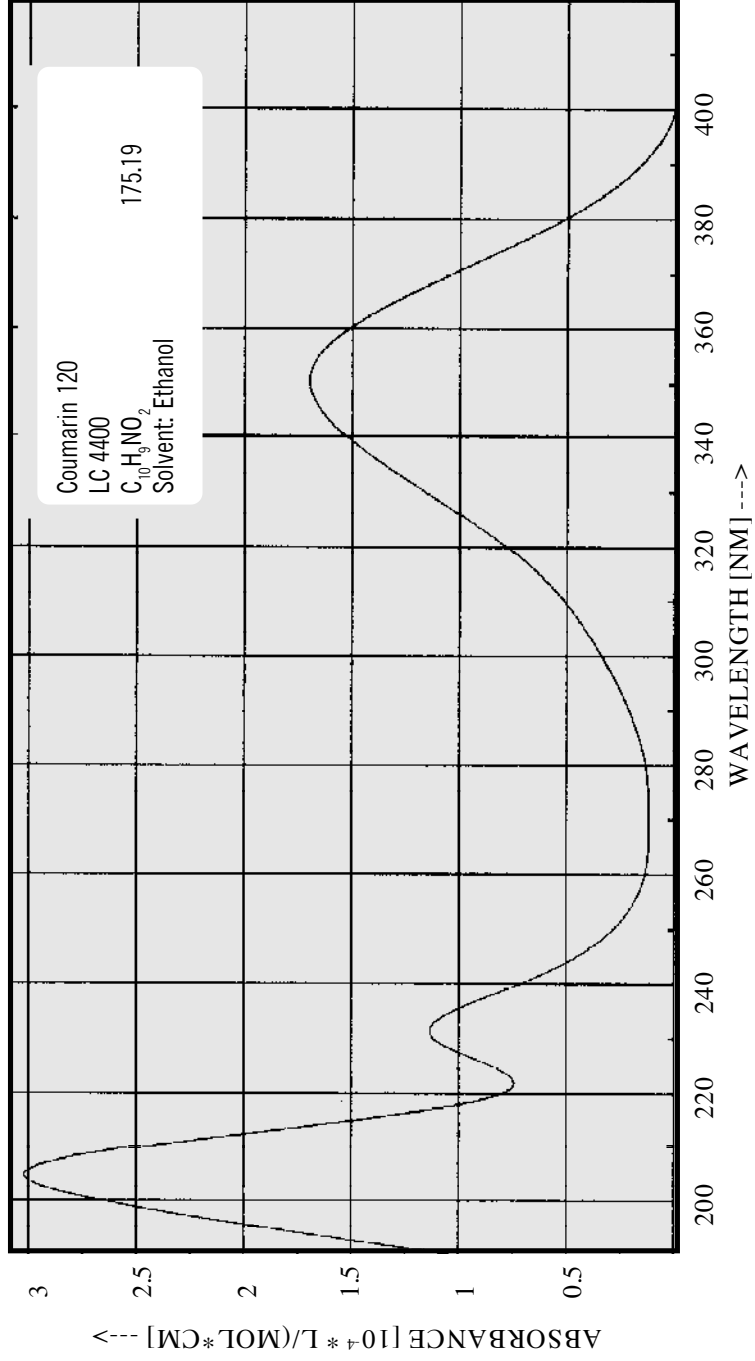
Laser dye for pulsed and CW operation; tunable around 440 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	430	420 - 440	poor	0.04	Methanol	1, 2
CW, Ar <sup>+</sup>	UV	440	415 - 490	3	0.6	Eg.	3

## References

1. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).
2. J. B. Marling et al., *Appl. Optics* **13**(10), 2317 (1974).
3. J. M. Yarborough, *Appl. Phys. Letters* **24**(12), 629 (1974).

# Coumarin 120



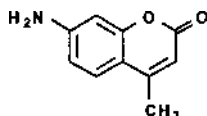
# Coumarin 120 (LC 4400)

## Constitution

7-Amino-4-methylcoumarin

Coumarin 440

$C_{10}H_9NO_2$  · MW: 175.19



## Characteristics

Lambdachrome® number:	4400
CAS registry number:	26093-31-2
Appearance:	slightly yellow, crystalline solid
Absorption maximum (in ethanol):	354 nm
Molar absorptivity:	$1.81 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	435 nm
For research and development purposes only.	

## Lasing Performance

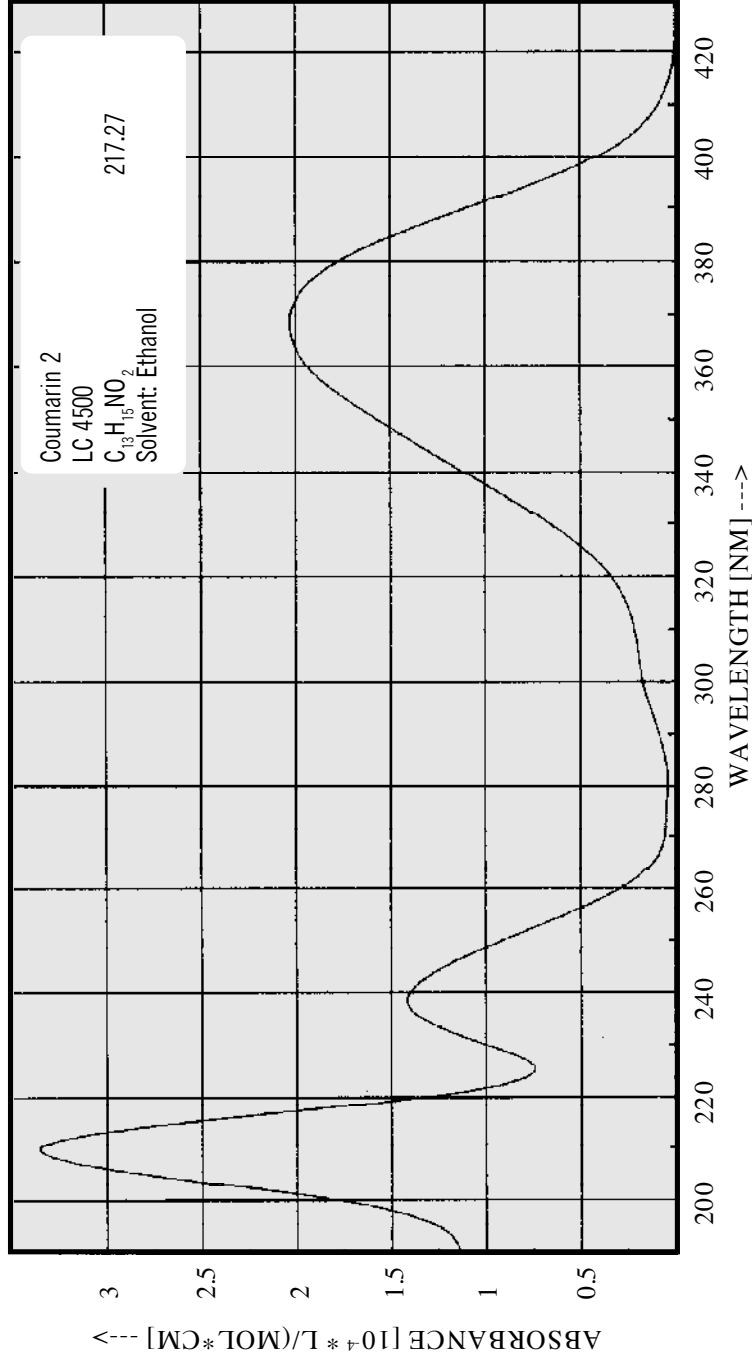
Efficient laser dye for pulsed and CW operation; tunable around 440 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	441	423 - 462	15	0.82	Methanol	1,2
Nitrogen	337	438	418 - 465	rel.	0.25	Methanol	2, 3
Nd:YAG, 3rd	355	440	420 - 470	16	0.25	Methanol	1, 4
Flashlamp	-	440	420 - 470	-	0.04	Methanol	5
CW, Ar <sup>+</sup>	UV	450	425 - 475	-	0.52	Eg.	6

## References

1. Lambda Physik, *Wall Chart* 1996.
2. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
3. Lambda Physik, *Data Sheet*.
4. K. Kato, *IEEE J. Quantum Electr.* **QE-11**, 373 (1975).
5. J. B. Marling et al., *Appl. Optics* **13**(10), 2317 (1974).
6. J. M. Yarborough, *Appl. Phys. Letters* **24**(12), 629 (1974).

## Coumarin 2



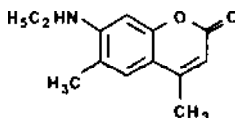
# Coumarin 2 (LC 4500)

## Constitution

7-Amino-4-methylcoumarin

Coumarin 450

$C_{13}H_{15}NO_2$  · MW: 217.27



## Characteristics

Lambdachrome® number:	4500
CAS registry number:	26078-25-1
Appearance:	slightly yellow, crystalline solid
Absorption maximum (in ethanol):	366 nm
Molar absorptivity:	$2.02 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	443 nm
For research and development purposes only.	

## Lasing Performance

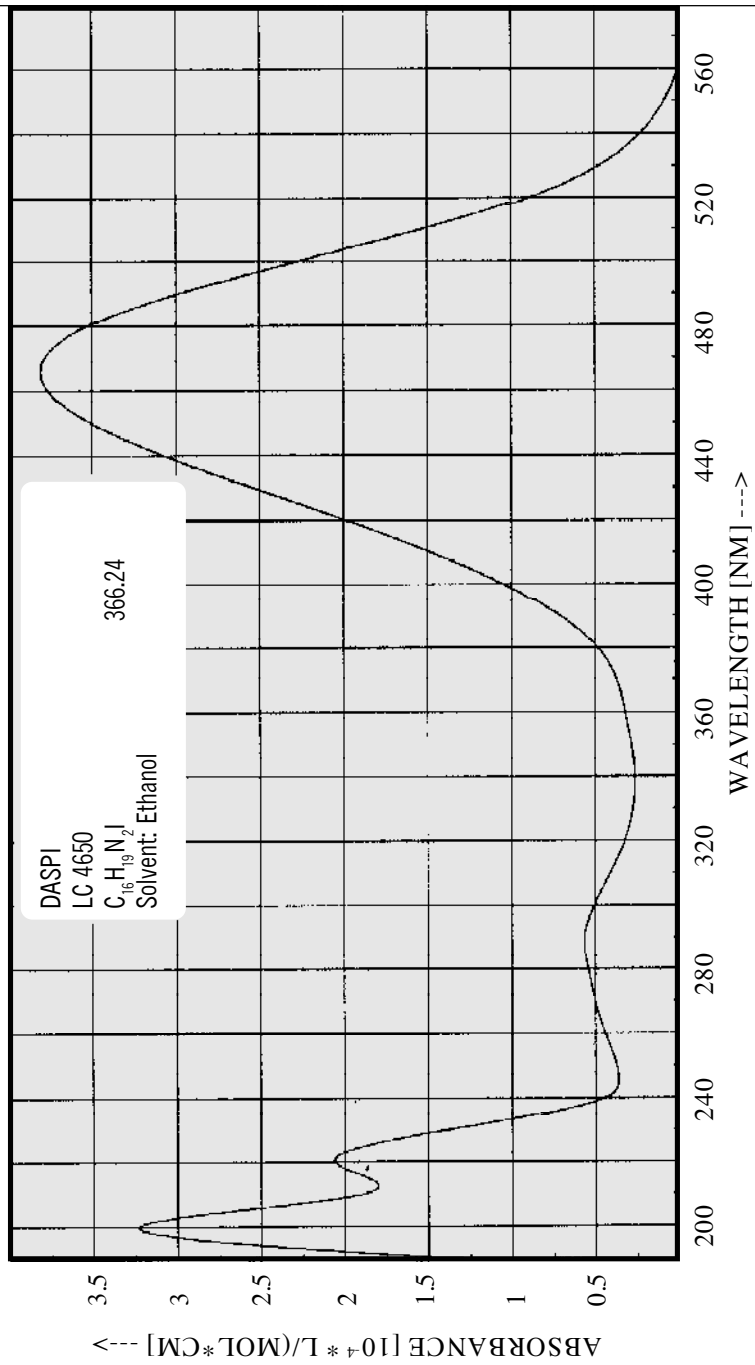
Efficient laser dye for pulsed and CW operation; tunable around 450 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	448	432-475	15	1.50	Methanol	1, 2
Nitrogen	337	444	426-475	rel.	0.40	Methanol	3, 4
Nd:YAG, 3rd	355	450	435-467	10	0.30	Methanol	3, 5
Flashlamp	-	450	440-458	-	-	Ethanol	6
CW, Ar <sup>+</sup>	UV	450	430-480	-	0.59	Bz./Eg.	7

## References

1. Lambda Physik, *Wall Chart* 1996.
2. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
3. Lambda Physik, *Data Sheet*.
4. R. J. von Gutfeld et al., *IEEE J. Quantum Electron.* QE-6, 332 (1970).
5. D. M. Guthals, J. W. Nibler, *Opt. Commun.* 29(3), 322 (1979).
6. A. N. Fletcher, *Appl. Phys.* 14, 295 (1977).
7. Coherent, *CW Dye Laser Fact Sheets*.

# DASPI



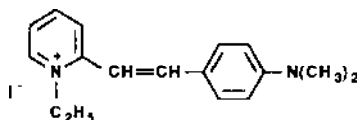
# DASPI (LC 4650)

## Constitution

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2-(p-Dimethylaminostyryl)-pyridylmethyl iodide

$C_{16}H_{19}N_2I$  · MW: 366.24



## Characteristics

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Lambdachrome® number:	4650
CAS registry number:	-
Appearance:	orange, crystalline solid
Absorption maximum (in ethanol):	472 nm
Molar absorptivity:	$3.83 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

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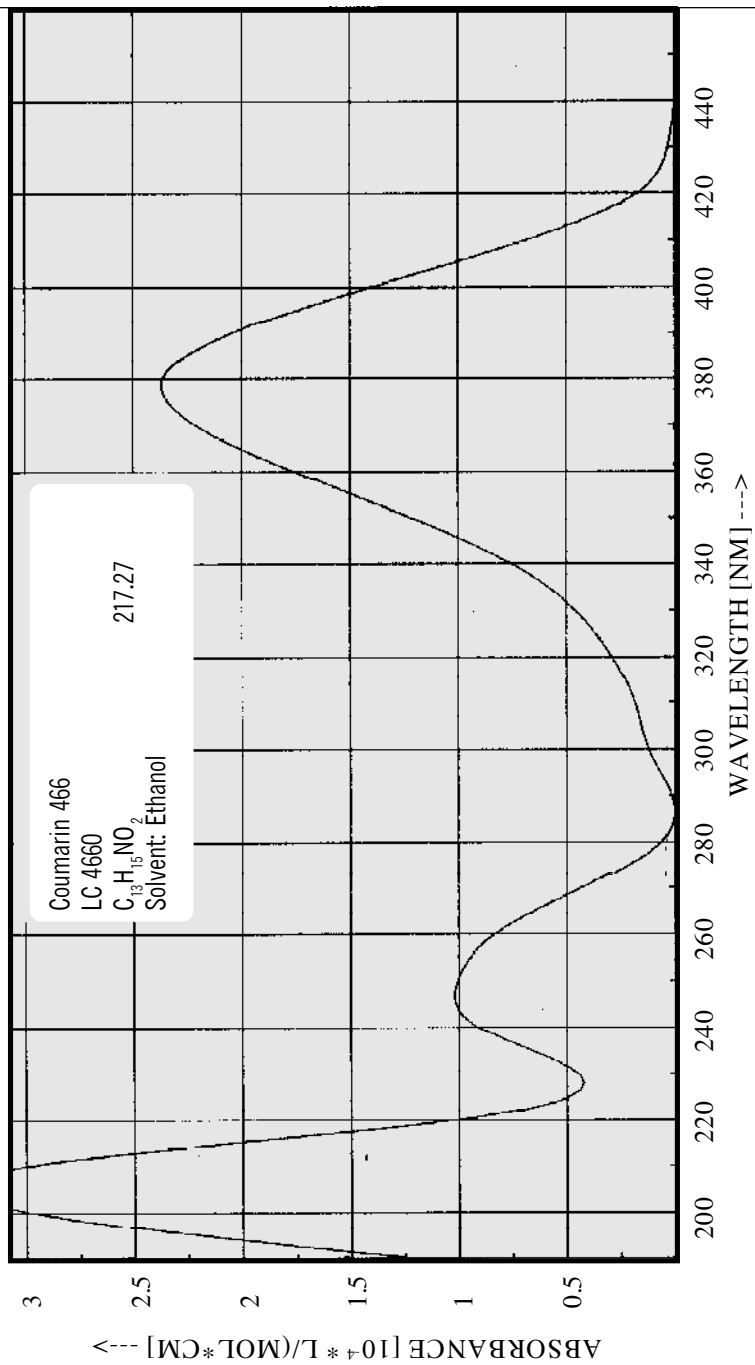
Saturable absorber for flashlamp pumped Coumarin 1, 102, 466, and 6H dye lasers<sup>1,2</sup>.

## References

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1. W. Sibbett, J. R. Taylor, *Opt. Commun.* **46**(1), 32 (1983).

# Coumarin 466





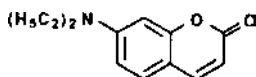
# Coumarin 466 (LC 4660)

## Constitution

7-Diethylaminocoumarin

LD 466 · C<sub>1</sub>H

C<sub>13</sub>H<sub>15</sub>NO<sub>2</sub> · MW: 217.27



## Characteristics

Lambdachrome® number:	4660
CAS registry number:	-
Appearance:	yellow, crystalline solid
Absorption maximum (ethanol):	380 nm
Molar absorptivity:	2.38 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

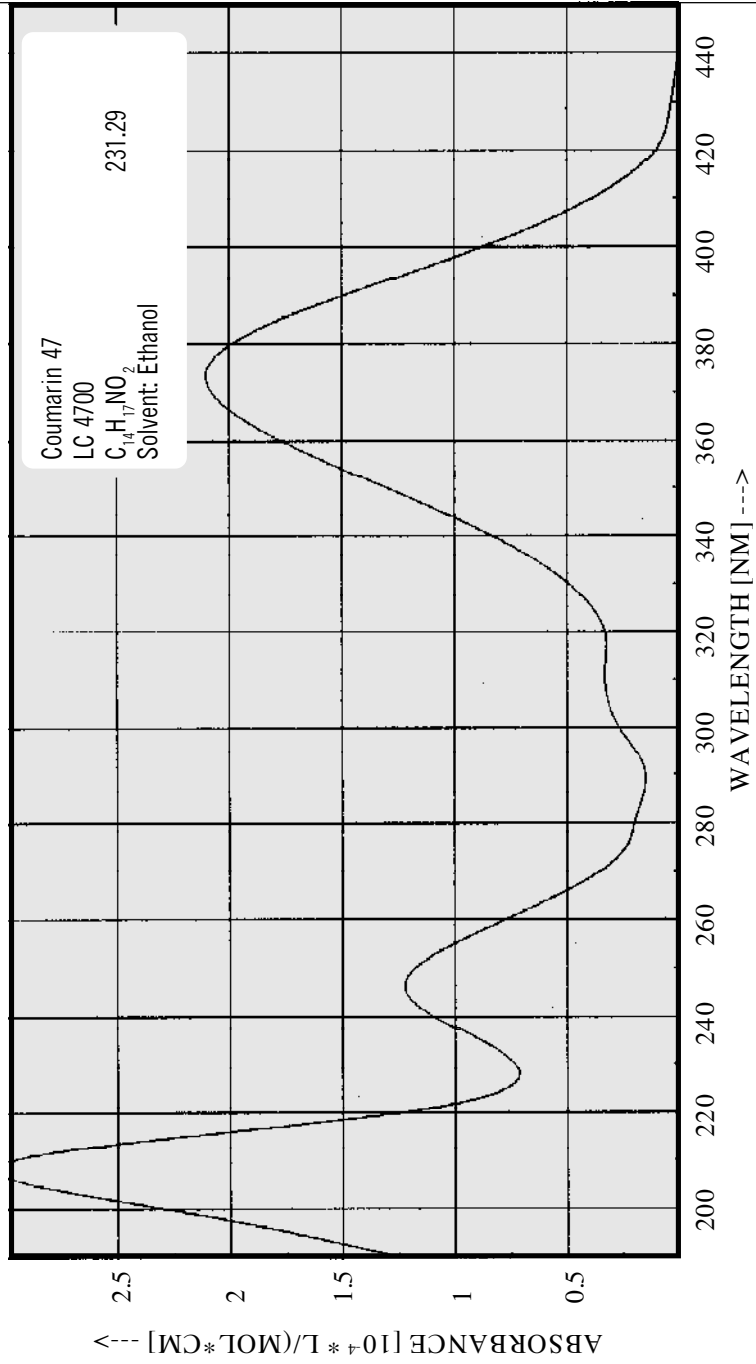
Efficient laser dye for pulsed operation; tunable around 465 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	466	-	-	0.16	Ethanol	1, 2

## References

1. E. J. Schimitschek et al., *Opt. Commun.* **16**(3), 313 (1976).
2. E. A. Stappaerts, *Appl. Optics* **16**(12), 3079 (1977).

## Coumarin 47



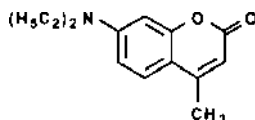
# Coumarin 47 (LC 4700)

## Constitution

7-Diethylamino-4-methylcoumarin

Coumarin 460 · Coumarin 1

$C_{14}H_{17}NO_2$  · MW: 231.29



## Characteristics

Lambdachrome® number:	4700
CAS registry number:	99-44-1
Appearance:	slightly yellow, crystalline solid
Absorption maximum (in ethanol):	373 nm
Molar absorptivity:	$2.10 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	450 nm
For research and development purposes only.	

## Lasing Performance

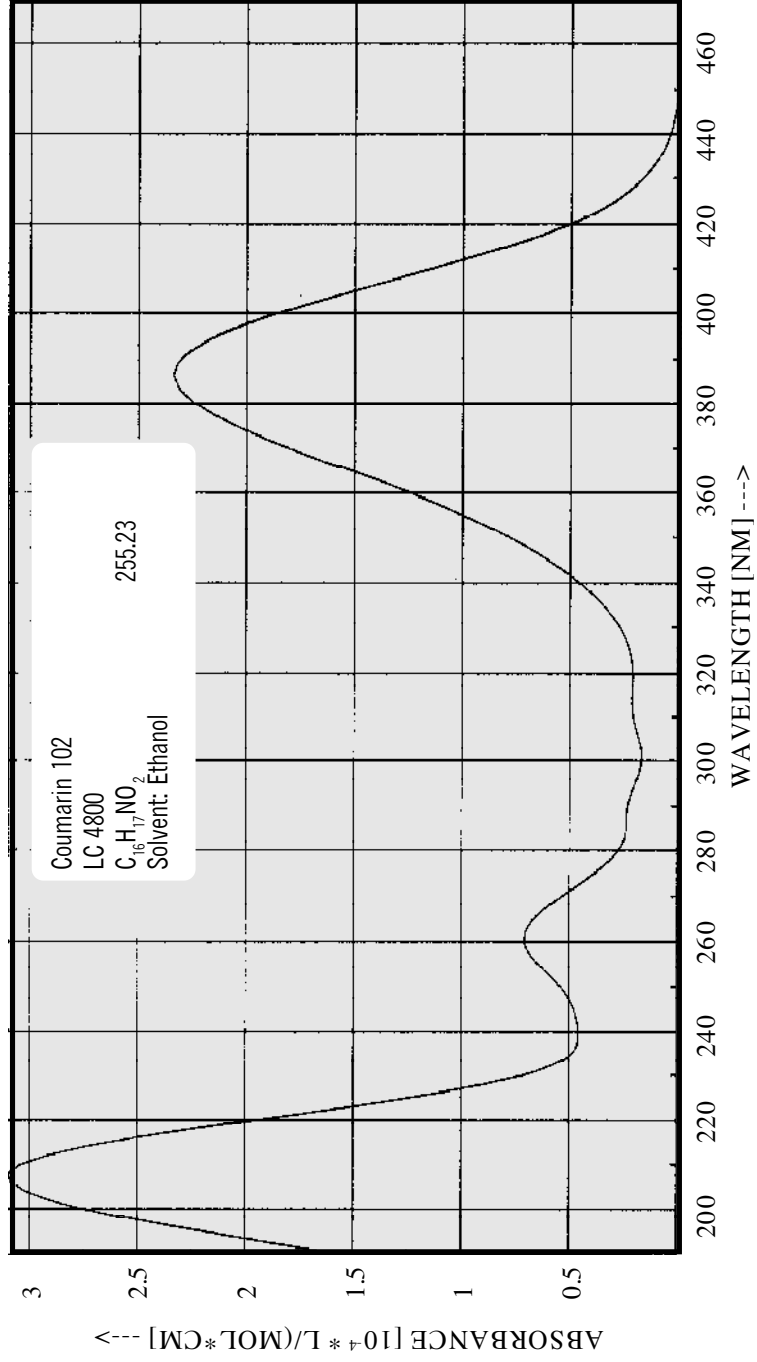
Efficient laser dye for pulsed and CW operation; tunable around 450 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	456	440 - 484	18	1.60	Methanol	1, 2, 3
Nitrogen	337	453	436 - 486	rel.	0.66	Methanol	3, 4
Nd:YAG, 3rd	355	460	444 - 476	15	0.3	Methanol	1, 5
Flashlamp	-	460	435 - 490	-	0.02	Ethanol	6, 7
CW, Ar <sup>+</sup>	UV	470	450 - 500	-	1.76	MeOH/Eg.	8

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 403 (1981).
3. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. D. M. Guthals, J. W. Nibbler, *Opt. Commun.* **29**(3), 322 (1977).
6. J. B. Marling et al., *Appl. Optics* **13**(10), 2317 (1974).
7. J. B. Marling et al., *Appl. Phys. Letters* **17**(12), 527 (1970).
8. Coherent, *CW Dye Laser Fact Sheets*.

# Coumarin 102

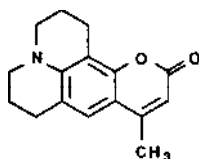


# Coumarin 102 (LC 4800)

## Constitution

2,3,5,6-1H,4H-Tetrahydro-8-methylquinolizino-[9,9a,1-g]-coumarin  
Coumarin 480

$C_{16}H_{17}NO_2$  · MW: 255.23



## Characteristics

Lambdachrome® number:	4800
CAS registry number:	41267-76-9
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	389 nm
Molar absorptivity:	$2.15 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	465 nm
For research and development purposes only.	

## Lasing Performance

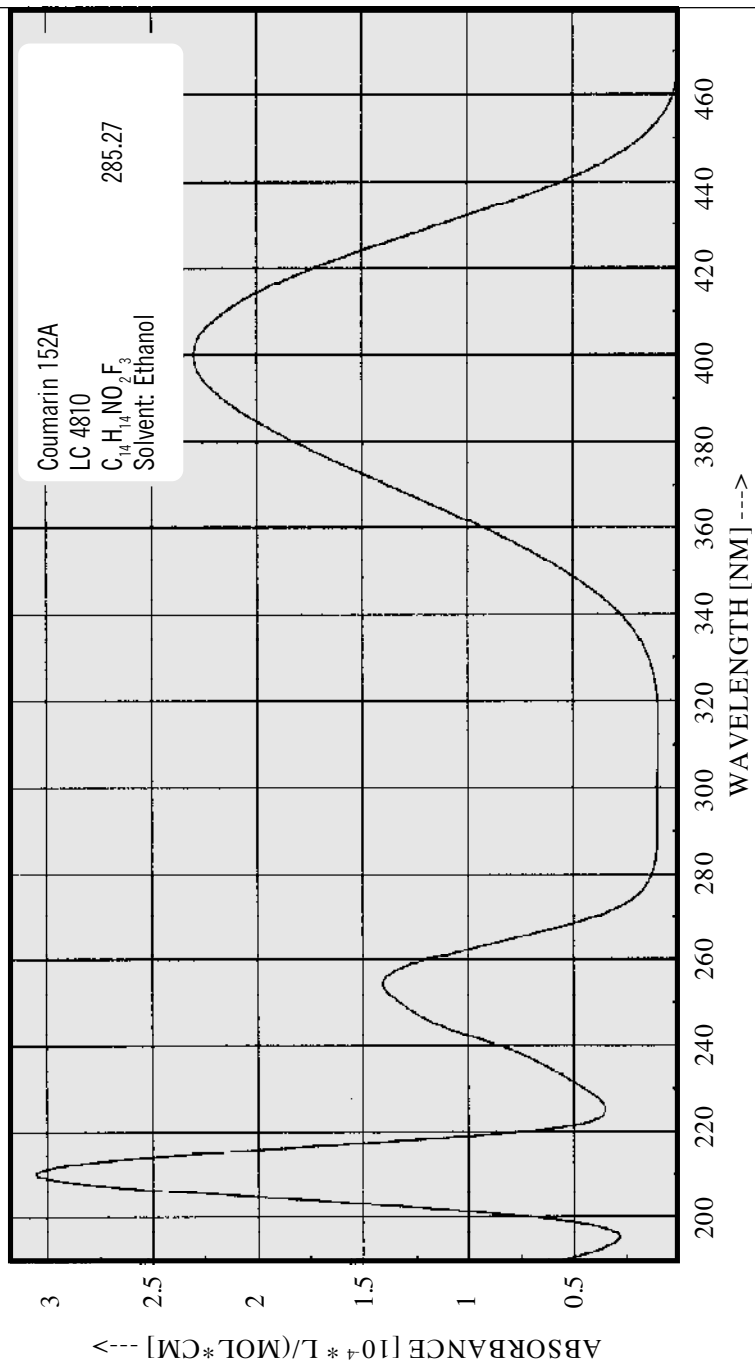
Efficient laser dye for pulsed and CW operation; tunable around 480 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	480	460 - 510	18	2.30	Methanol	1, 2
Nitrogen	337	470	454 - 506	rel.	1.44	Methanol	3
Nd:YAG, 3rd	355	480	462 - 497	15	0.40	Methanol	1, 4
Flashlamp	-	480	460 - 530		0.05	Methanol	5
CW, Ar <sup>+</sup>	UV	482	463 - 515	-	2.0	Bz./Eg.	1, 6

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38(5,6)**, 403 (1981).
3. Lambda Physik, *Data Sheet*.
4. K. Kato, *IEEE J. Quant. Electron.* **QE-11**, 373 (1975).
5. J. B. Marling et al., *Appl. Optics* **13(10)**, 2317 (1974).
6. Coherent, *CW Dye Laser Fact Sheets*.

## Coumarin 152A



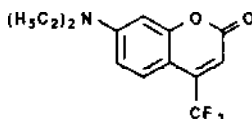
# Coumarin 152A (LC 4810)

## Constitution

7-Diethylamino-4-trifluormethylcoumarin

Coumarin 481 · C1F

$C_{14}H_{14}NO_2F_3$  · MW: 285.27



## Characteristics

Lambdachrome® number:	4810
CAS registry number:	41934-47-8
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	405 nm
Molar absorptivity:	$2.16 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	510 nm
For research and development purposes only.	

## Lasing Performance

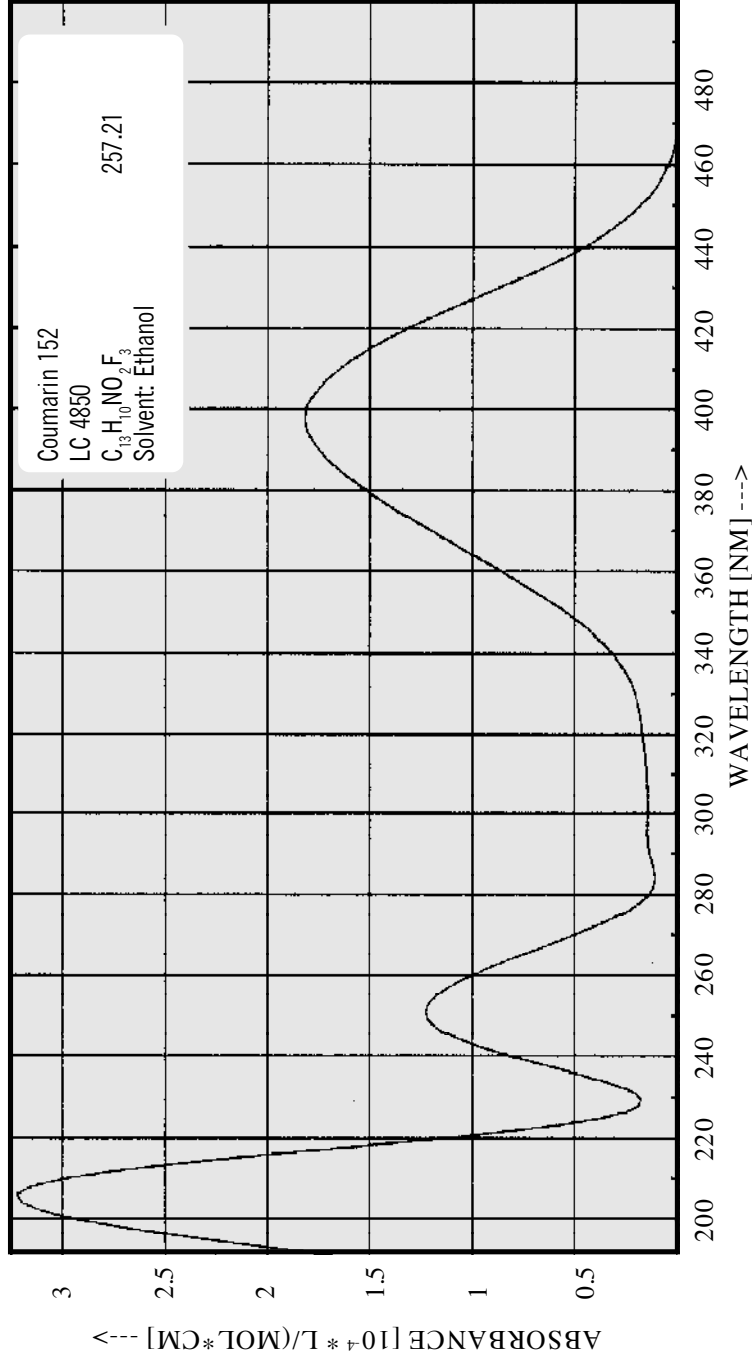
Efficient laser dye for pulsed operation; tunable around 500 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	517	491 - 553	4	1.84	Ethanol	1, 2
Nitrogen	337	490	461 - 549	rel.	1.71	Dioxane	2, 3
Nd:YAG, 3rd	355	520	500 - 540	-	2.85	Ethanol	4
Flashlamp	-	481	-	-	0.21	Dioxane	5

## References

1. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 403 (1981).
2. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
3. J. A. Halstead, R. R. Reeves, *Opt. Commun.* **27**(2), 273 (1978).
4. D. M. Guthals, J. W. Nibbler, *Opt. Commun.* **29**(3), 322 (1977).
5. E. J. Schimitschek et al., *IEEE J. Quantum Electron.* **QE-9**, 781 (1973).

# Coumarin 152





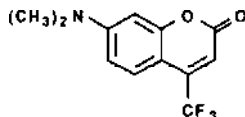
# Coumarin 152 (LC 4850)

## Constitution

7-Dimethylamino-4-trifluoromethylcoumarin

Coumarin 485 · C2F

$C_{13}H_{10}NO_2F_3$  · MW: 257.21



## Characteristics

Lambdachrome® number:	4850
CAS registry number:	53518-14-2
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	397 nm
Molar absorptivity:	$1.97 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	510 nm
For research and development purposes only.	

## Lasing Performance

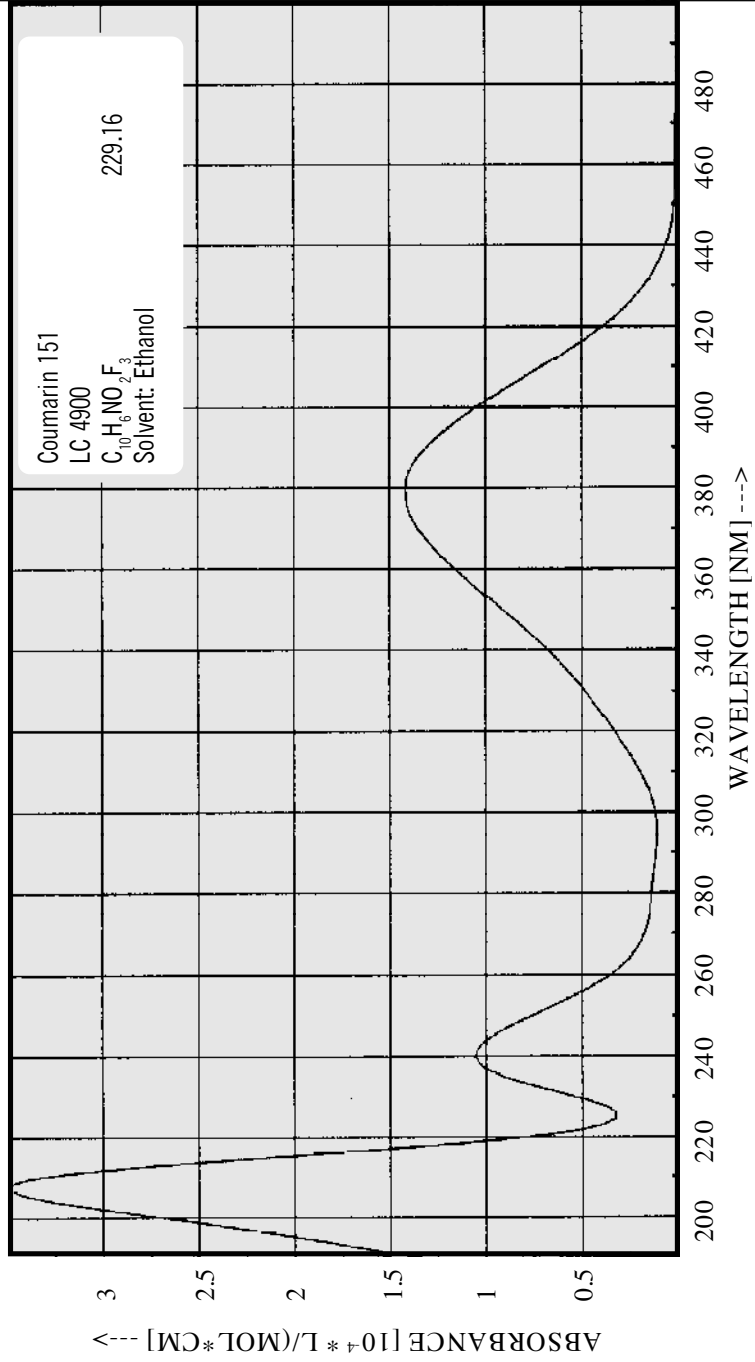
Efficient laser dye for pulsed operation; tunable around 520 nm.

Source	Pump		Dye Laser Characteristics			Solvent	Ref.
	Wavelength	Peak	Range	Effic.	Conc.		
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308	520	490 - 570	5.5	2.00	Ethanol	1
Nitrogen	337	532	495 - 560	-	-	Ethanol	2
Flashlamp	-	-	518 - 531	-	-	Ethanol	3

## References

1. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 403 (1981).
2. Lambda Physik, *Data Sheet*.
3. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).

# Coumarin 151



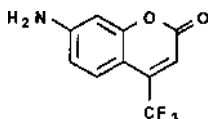
# Coumarin 151 (LC 4900)

## Constitution

7-Amino-4-trifluoromethylcoumarin

Coumarin 490 · C3F

$C_{10}H_6NO_2F_3$  · MW: 229.16



## Characteristics

Lambdachrome® number:	4900
CAS registry number:	53518-13-3
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	382 nm
Molar absorptivity:	$1.70 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	480 nm
For research and development purposes only.	

## Lasing Performance

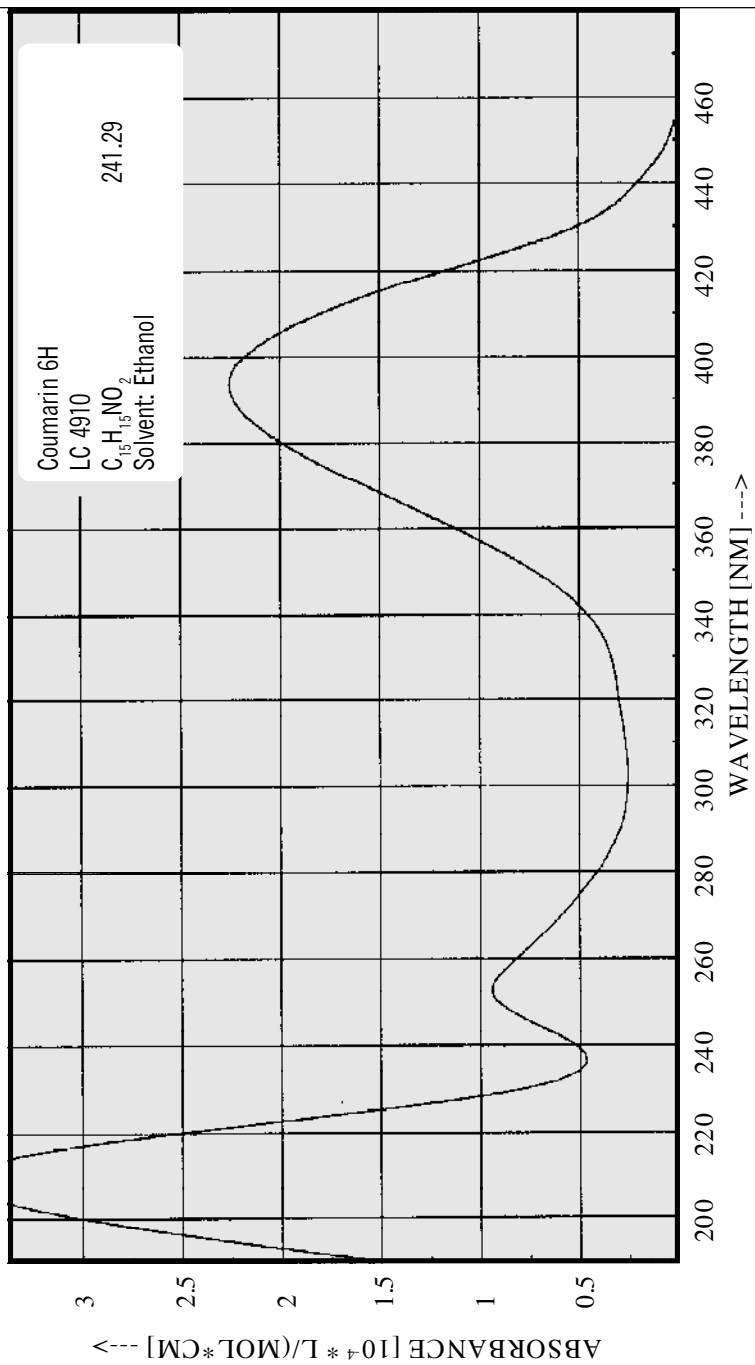
Efficient laser dye for pulsed operation; tunable around 490 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
Flashlamp	-	490	481 - 493	-	-	Ethanol	1, 2

## References

1. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).
2. G. A. Reynolds, K. H. Drexhage, *Opt. Commun.* **13(3)**, 222 (1975).

# Coumarin 6H



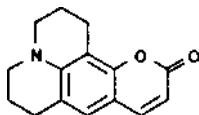
# Coumarin 6H (LC 4910)

## Constitution

2,3,5,6-1H,4H-Tetrahydroquinolizino-[9,9a,1-gh]coumarin

LD 490 · C<sub>6</sub>H

C<sub>15</sub>H<sub>15</sub>NO<sub>2</sub> · MW: 241.29



## Characteristics

Lambdachrome® number:	4910
CAS registry number:	58336-35-9
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	396 nm
Molar absorptivity:	$2.50 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

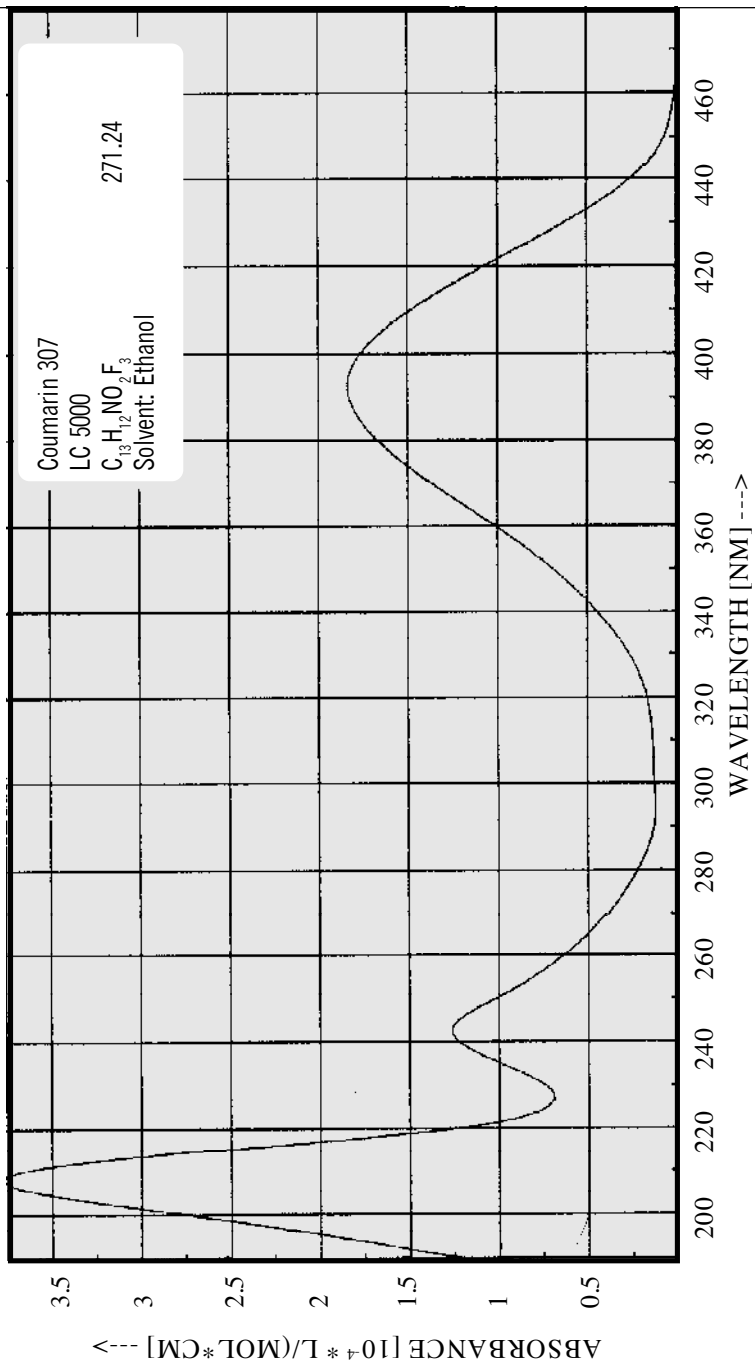
Efficient laser dye for pulsed operation; tunable around 490 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	491	463 - 522	5.7	1.33	Ethanol	1
Flashlamp	-	-	477 - 493	-	-	Ethanol	2
CW, Kr <sup>+</sup>	VIO	490	476 - 515	-	1.20	Eg./Bz.	3

## References

1. Lambda Physik.
2. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).
3. Lambda Physik, *Data Sheet*.

# Coumarin 307



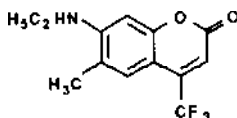
# Coumarin 307 (LC 5000)

## Constitution

7-Ethylamino-6-methyl-4-trifluormethylcoumarin

Coumarin 503

$C_{13}H_{12}NO_2F_3$  · MW: 271.24



## Characteristics

Lambdachrome® number:	5000
CAS registry number:	55804-66-5
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	395 nm
Molar absorptivity:	$1.85 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	490 nm
For research and development purposes only.	

## Lasing Performance

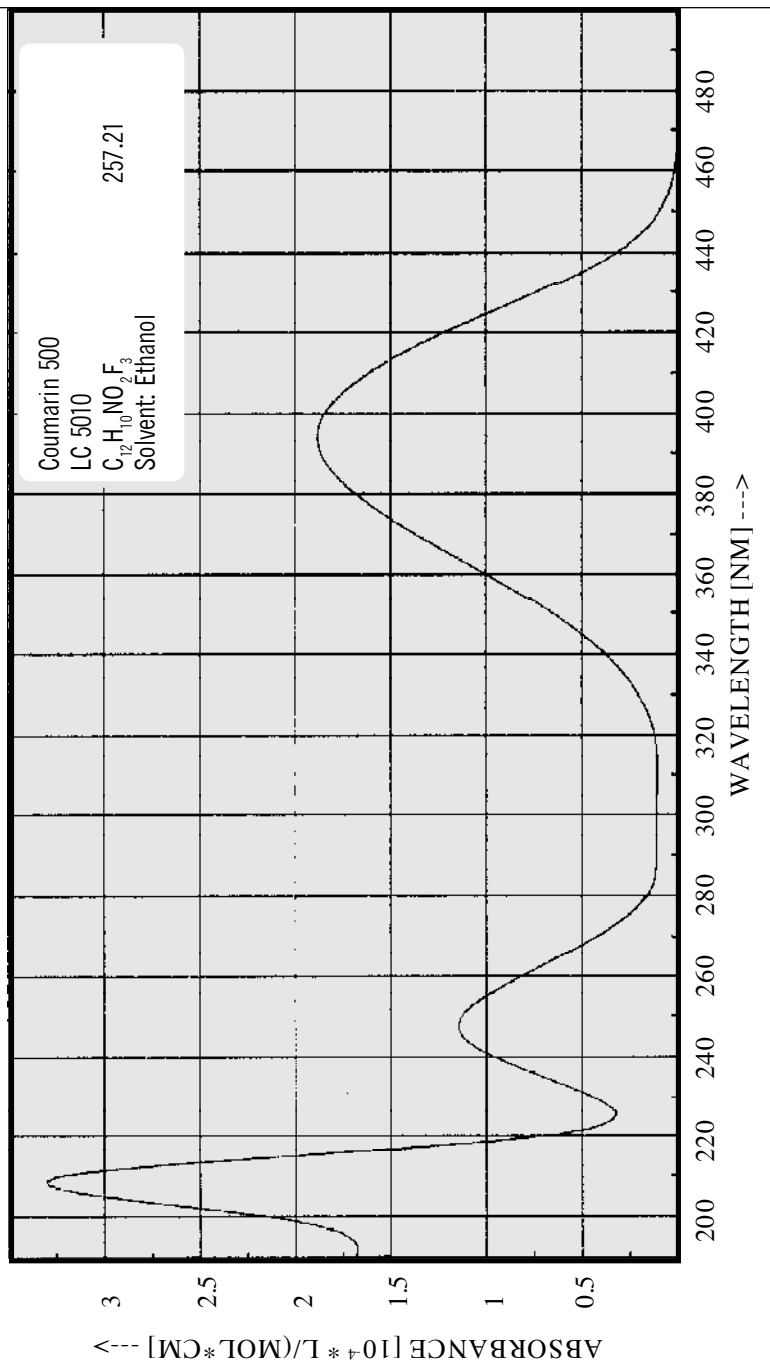
Efficient laser dye for pulsed and CW operation; tunable around 500 nm.

Source	Pump		Dye Laser Characteristics			Solvent	Ref.
	Wavelength	Peak	Range	Effic.	Conc.		
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308	500	479 - 553	16	3.40	Methanol	1, 2
Nitrogen	337	504	478 - 547	rel.	1.60	Methanol	3, 4
Nd:YAG, 3rd	355	508	485 - 546	15	0.70	Methanol	1
Flashlamp	-	-	490 - 510	-	-	Ethanol	6

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38(5,6)**, 403 (1981).
3. F. Bos, *Appl. Optics* **20(20)**, 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. A. N. Fletcher, *Appl Phys.* **14**, 295 (1977).

# Coumarin 500



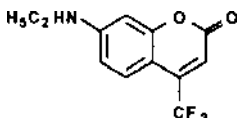


# Coumarin 500 (LC 5010)

## Constitution

7-Ethylamino-4-trifluormethylcoumarin

$C_{12}H_{10}NO_2F_3$  · MW: 257.21



## Characteristics

Lambdachrome® number:	5010
CAS registry number:	-
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	395 nm
Molar absorptivity:	$1.85 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

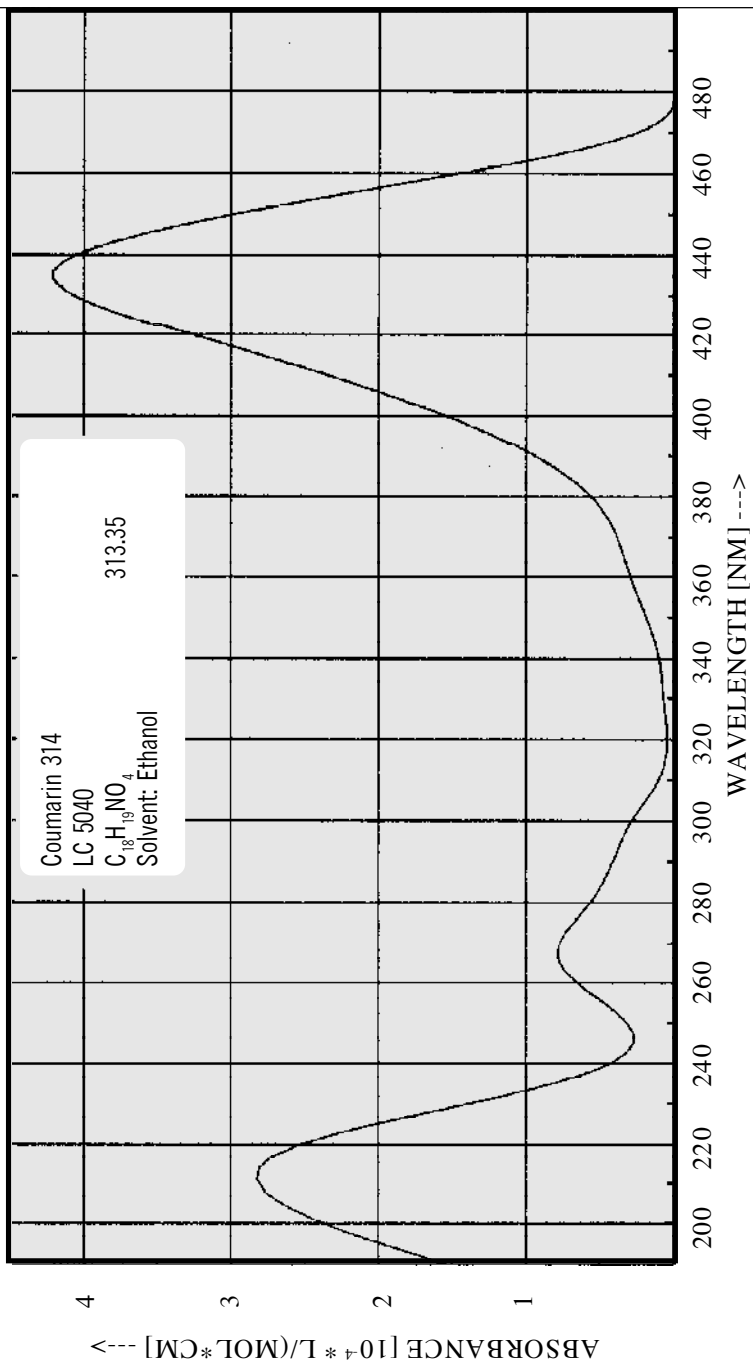
Efficient laser dye for pulsed operation; tunable around 500 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	502	480 - 520	12	2.40	Methanol	1, 2
Nitrogen	337	503	473 - 562	rel.	1.40	Ethanol	2
Nd:YAG, 3rd	355	518	498 - 546	10	0.70	Methanol	3, 6
Flashlamp	-	-	-	-	-	-	4, 5

## References

1. Lambda Physik
2. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
3. D. M. Guthals, J. W. Nibler, *Opt. Commun.* **29**(3), 322(1979).
4. Th. Varghese, *Opt. Commun.* **44**(5), 353(1983).
5. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).
6. Lambda Physik, *Wall Chart* **1996**.

# Coumarin 314

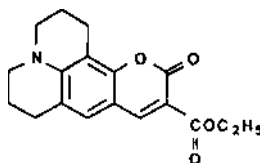


# Coumarin 314 (LC 5040)

## Constitution

2,3,5,6-1H,4H-Tetrahydro-9-carboethoxyquinolizino-[9,9a,1-gh]coumarin  
Coumarin 504

$C_{18}H_{19}NO_4$  · MW: 313.35



## Characteristics

Lambdachrome® number:	5040
CAS registry number:	55804-66-5
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	436 nm
Molar absorptivity:	$4.70 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	480 nm
For research and development purposes only.	

## Lasing Performance

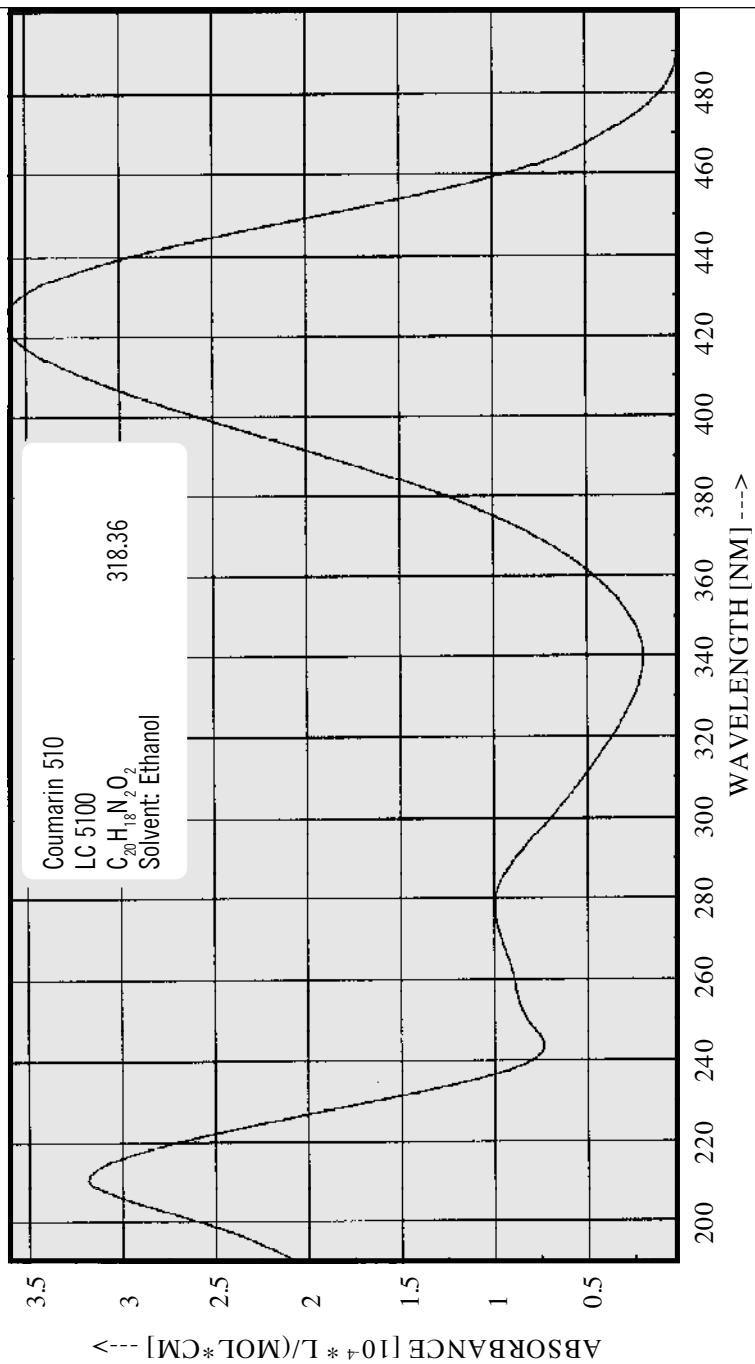
Efficient laser dye for pulsed operation; tunable around 505 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
Flashlamp	-	506	490 - 504	-	-	Ethanol	1, 2

## References

1. K. H. Drexhage et al., *IEEE J. Quantum Electron.* **QE-10**, 695 (1974).
2. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).

# Coumarin 510

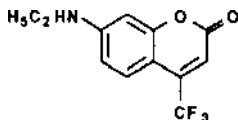


# Coumarin 510 (LC 5100)

## Constitution

2,3,5,6-1H,4H-Tetrahydro-9-(3-pyridyl)-quinolizino-[9,9a,1-gh]coumarin

$C_{20}H_{18}N_2O_2$  · MW: 318.36



## Characteristics

Lambdachrome® number:	5100
CAS registry number:	-
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	425 nm
Molar absorptivity:	$3.70 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

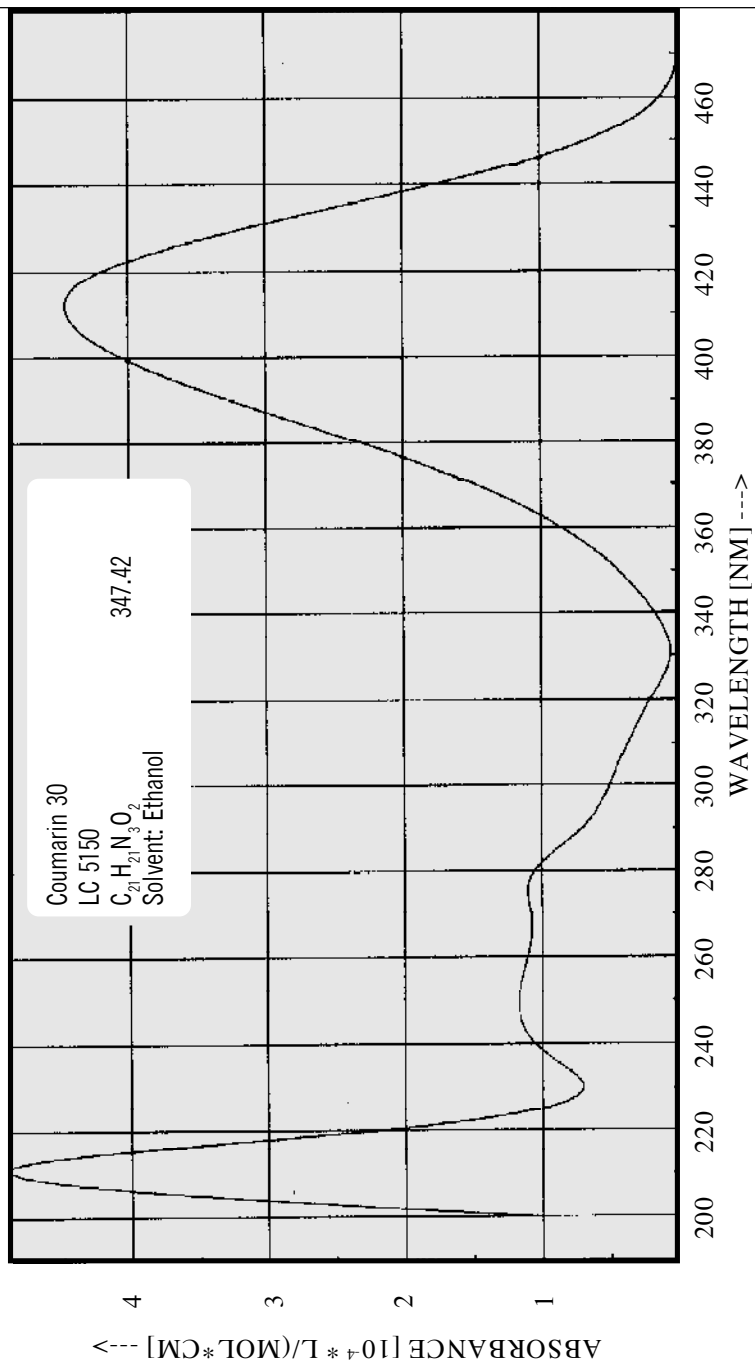
Efficient laser dye for pulsed operation; tunable around 510 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	511	504 - 511	-	-	Ethanol	1
CW, Ar <sup>+</sup>	VIO	525	495 - 565	-	1.0	Bz./Eg	2

## References

1. A. N. Fletcher et al., *Opt. Commun.* **47**(1), 57 (1983).
2. Lambda Physik.

## Coumarin 30

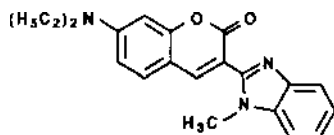


# Coumarin 30 (LC 5150)

## Constitution

3-(2'-N-Methylbenzimidazolyl)-7-N,N-diethylaminocoumarin  
Coumarin 515

$C_{21}H_{21}N_3O_2$  · MW: 347.42



## Characteristics

Lambdachrome® number:	5150
CAS registry number:	41044-12-6
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	412 nm
Molar absorptivity:	$4.45 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	488 nm
For research and development purposes only.	

## Lasing Performance

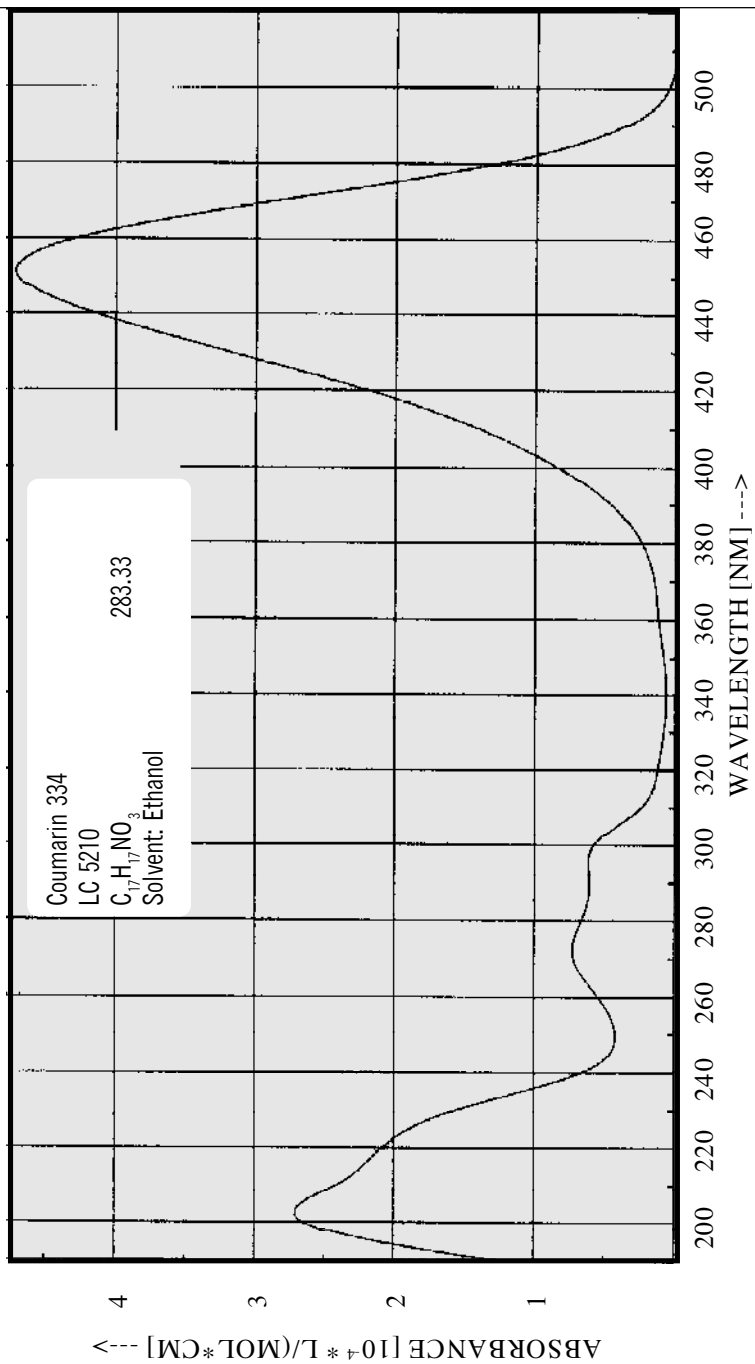
Efficient laser dye for pulsed operation; tunable around 510 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	508	480 - 540	-	0.07	Methanol	1, 2
CW, Kr <sup>+</sup>	VIO	510	480 - 555	12	0.26	MeOH/Eg	3

## References

1. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).
2. J. B. Marling et al., *Appl. Optics*. **13**(10), 2317 (1974).
3. Coherent, *CW Dye Laser Fact Sheets*.

# Coumarin 334



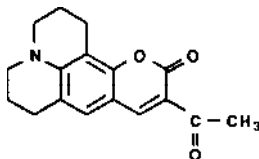


# Coumarin 334 (LC 5210)

## Constitution

2,3,5,6-1H,4H-Tetrahydro-9-acetylquinolizino-[9,9a,1-gh]-coumarin  
Coumarin 521

$C_{17}H_{17}NO_3$  · MW: 283.33



## Characteristics

Lambdachrome® number:	5210
CAS registry number:	55804-67-6
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	450 nm
Molar absorptivity:	$4.73 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	495 nm
For research and development purposes only.	

## Lasing Performance

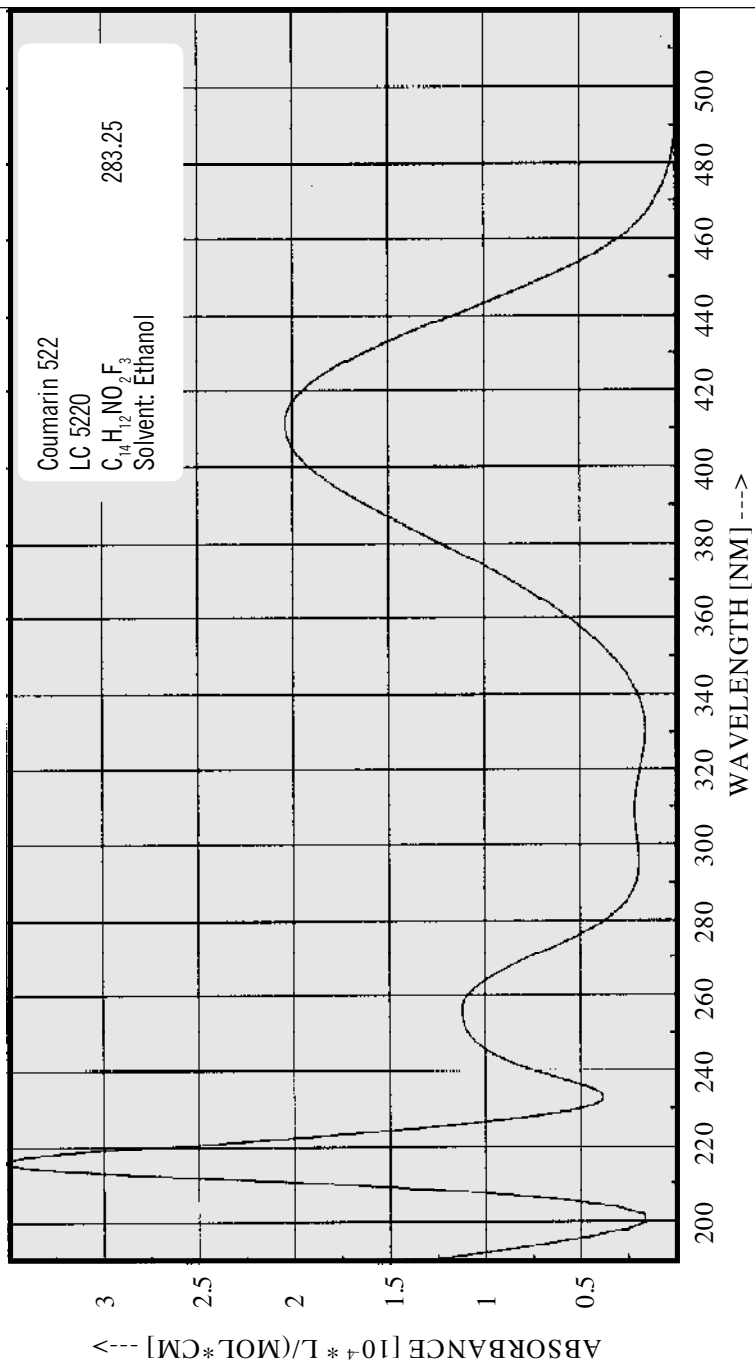
Efficient laser dye for pulsed and CW operation; tunable around 480 nm.

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		520	506 - 537	12	1.50	Methanol	1, 2
Flashlamp	-		-	507 - 512		-	Methanol	3, 4

## References

1. Lambda Physik, *Wall Chart* 6/83.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9(1983).
3. A. N. Fletcher *Appl. Phys.* **14**, 295 (1977).
4. G. A. Reynolds, K. H. Drexhage, *Opt. Commun.* **13(3)**, 222(1975).

# Coumarin 522

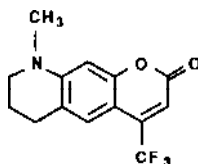


# Coumarin 522 (LC 5220)

## Constitution

N-Methyl-4-trifluormethylpiperidino-[3,2-g]-coumarin  
C8F

$C_{14}H_{12}NO_2F_3$  · MW: 283.25



## Characteristics

Lambdachrome® number:	5220
CAS registry number:	55318-19-7
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	410 nm
Molar absorptivity:	$2.06 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	516 nm
For research and development purposes only.	

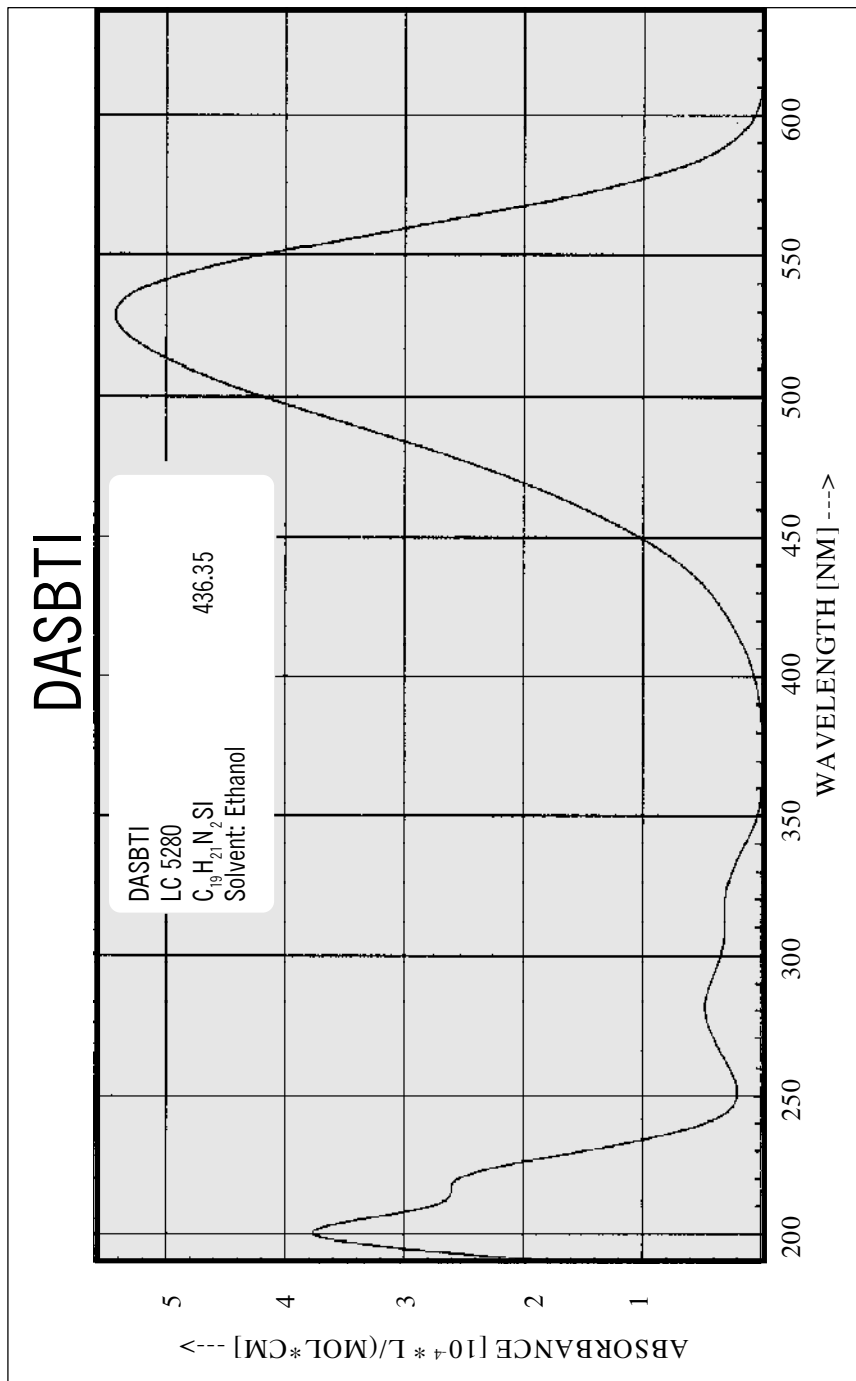
## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 520 nm.

Source	Pump		Dye Laser Characteristics			Solvent	Ref.
	Wavelength	Peak	Range	Effic.	Conc.		
	[nm]	[nm]	[nm]	[%]	[g/l]		
Nitrogen	337	520	495 - 575	14	0.04	DMF	1
Nd:YAG, 3rd	355	525	505 - 550	-	1.41	Ethanol	2
Flashlamp	-	-	518 - 528	-	-	Ethanol	3

## References

1. P. R. Hammond, *IEEE J. Quantum Electron.* **QE-15**(7), 624(1979).
2. D. M. Guthals, J. W. Nibler, *Opt. Commun.* **29**(3), 322(1979).
3. A. N. Fletcher, *Appl. Phys.* **14**, 295 (1977).



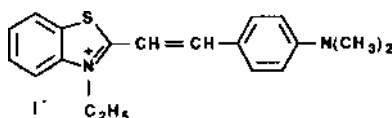
# DASBTI (LC 5280)

## Constitution

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2-(p-Dimethylaminostyryl)-benzothiazolylethyl iodide

$C_{19}H_{21}N_2SI$  · MW: 436.35



## Characteristics

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Lambdachrome® number:	5280
CAS registry number:	-
Appearance:	red, crystalline solid
Absorption maximum (in ethanol):	530nm
Molar absorptivity:	$5.49 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

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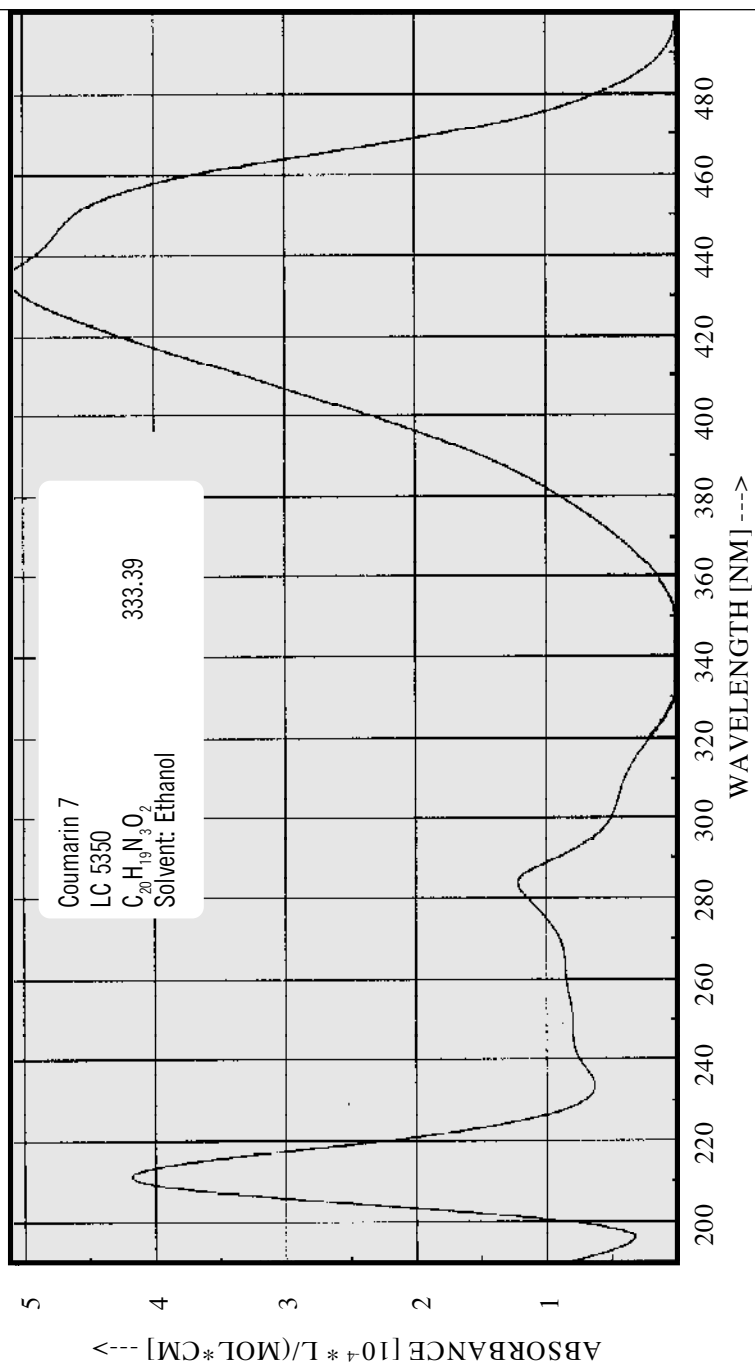
Saturable absorber for flashlamp pumped Coumarin 6, 153, 522, and Rhodamine 110 dye lasers <sup>1,2</sup>. Applicable around 530 nm.

## References

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1. W. Sibbett, J. R. Taylor, *Appl. Phys.* **B29**, 191(1982).
2. W. Sibbett, J. R. Taylor, *IEEE J. Quantum Electron.* **QE-19(4)**, 558(1983).

# Coumarin 7



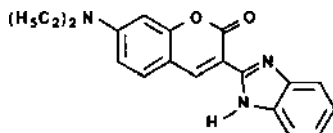
# Coumarin 7 (LC 5350)

## Constitution

3-(2'-Benzimidazolyl)-7-N,N-diethylaminocoumarin

Coumarin 535

$C_{20}H_{19}N_3O_2$  · MW: 333.39



## Characteristics

Lambdachrome® number:	5350
CAS registry number:	27425-55-4
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	433 nm
Molar absorptivity:	$5.05 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	493 nm
For research and development purposes only.	

## Lasing Performance

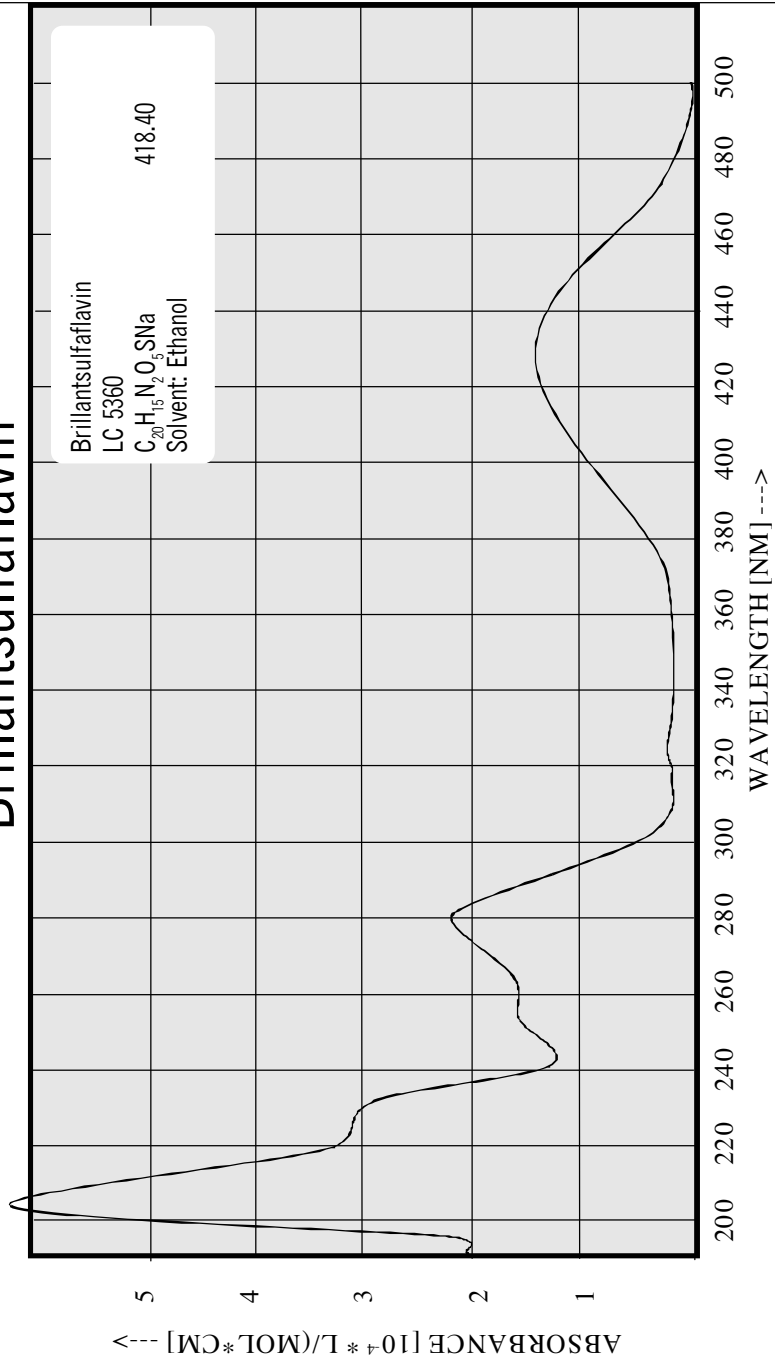
Efficient laser dye for pulsed and CW operation; tunable around 530 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	-	517 - 527	-	-	Ethanol	1
CW, Ar <sup>+</sup>	VIO	530	495 - 570	9	1.00	Bz.	2

## References

1. A. N. Fletcher *Appl. Phys.* **14**, 295 (1977).
2. J. M. Yarborough, *Appl. Phys. Lett.* **24**(12), 629 (1974).

# Brillantsulfaflavin



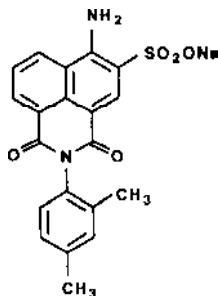


# Brillantsulfaflavin (LC 5360)

## Constitution

Pilot 512

$C_{20}H_{15}N_2O_5SNa$  · MW: 418.40



## Characteristics

Lambdachrome® number:	5360
CAS registry number:	2391-30-2
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	423 nm
Molar absorptivity:	$1.41 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	517 nm
For research and development purposes only.	

## Lasing Performance

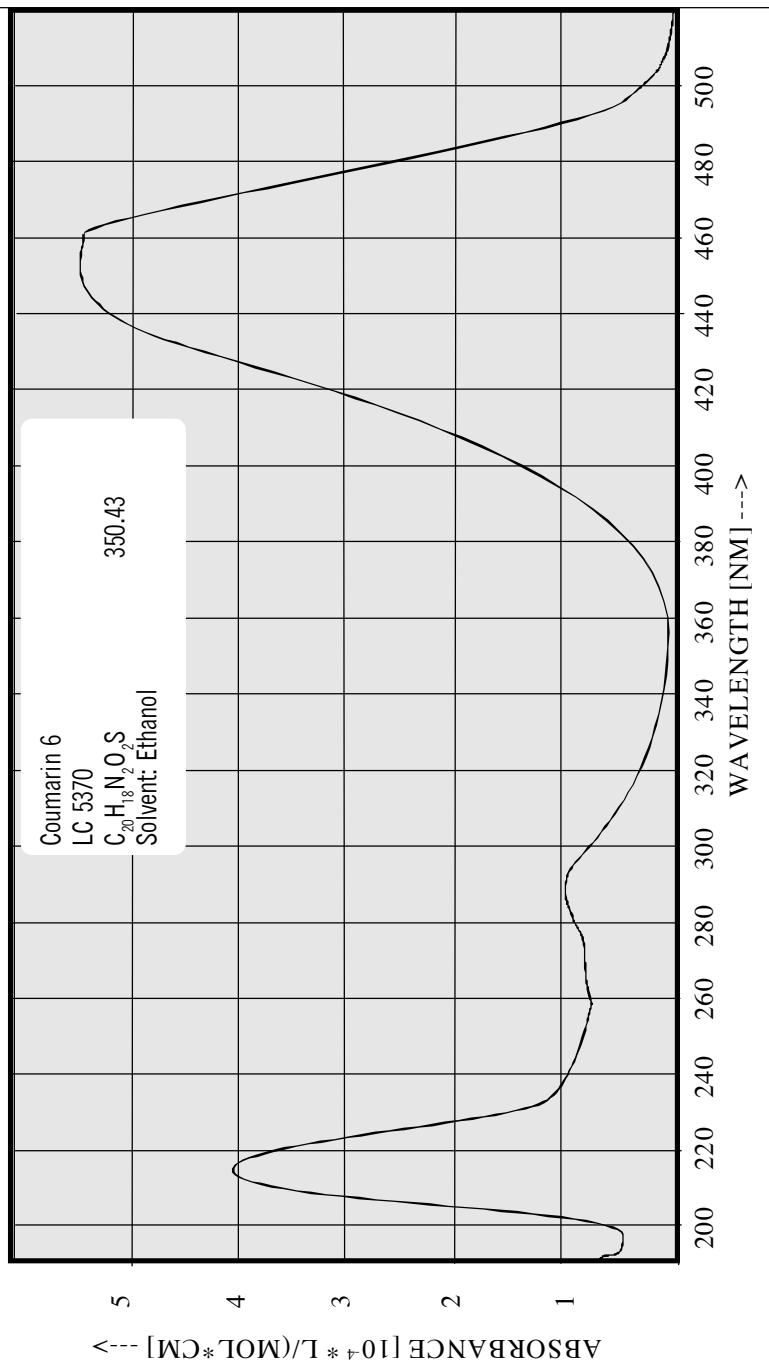
Laser dye for pulsed operation; tunable around 540 nm.

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
Flashlamp	-		540	508 - 574	-	0.21	Ethanol	1, 2

## References

1. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **11**(5), 692 (1972).
2. J. B. Marling et al., *IEEE J. Quantum Electr.* **QE-7**, 498 (1971).

# Coumarin 6



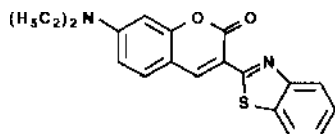
# Coumarin 6 (LC 5370)

## Constitution

3-(2'-Benzothiazolyl)-7-diethylaminocoumarin

Coumarin 540

$C_{20}H_{18}N_2O_2S$  · MW: 350.43



## Characteristics

Lambdachrome® number:	5370
CAS registry number:	38215-35-0
Appearance:	orange, crystalline solid
Absorption maximum (in ethanol):	458 nm
Molar absorptivity:	$5.4 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	505 nm
For research and development purposes only.	

## Lasing Performance

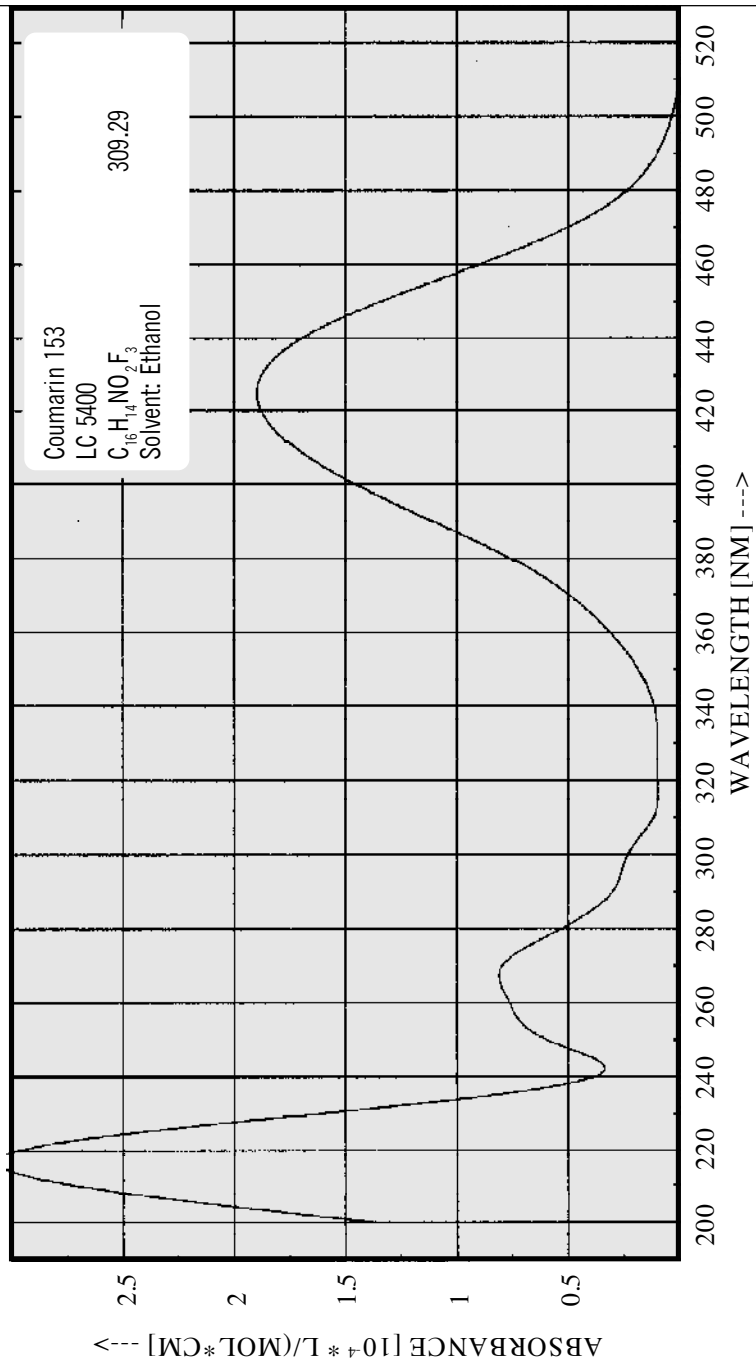
Efficient laser dye for pulsed and CW operation; tunable around 540 nm.

Source	Pump		Dye Laser Characteristics			Solvent	Ref.
	Wavelength	Peak	Range	Effic.	Conc.		
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308	534	515 - 558	9	0.84	DMSO	1, 2
Flashlamp	-	-	530 - 539	-	-	Ethanol	3
CW, Ar <sup>+</sup>	488	535	510 - 550	-	2.0	Bz./Eg.	4, 5

## References

1. Lambda Physik.
2. O. Uchino et al., *Appl. Phys.* **19**, 35(1979).
3. A. N. Fletcher, *Appl. Phys.* **14**, 295(1977).
4. Lambda Physik, *Wall Chart* 1996.
5. T. F. Johnston et al., *Appl. Optics* **21**(13), 2307(1982).

# Coumarin 153

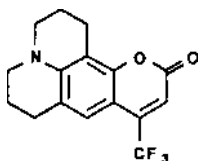


# Coumarin 153 (LC 5400)

## Constitution

2,3,5,6-1H,4H-Tetrahydro-8-trifluormethylquinolizino-[9,9a,1-g]coumarin  
Coumarin 540A, C6F

$C_{16}H_{14}NO_2F_3$  · MW: 309.29



## Characteristics

Lambdachrome® number:	5400
CAS registry number:	53518-18-6
Appearance:	yellow, crystalline solid
Absorption maximum (in ethanol):	423 nm
Molar absorptivity:	$1.89 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	530 nm
For research and development purposes only.	

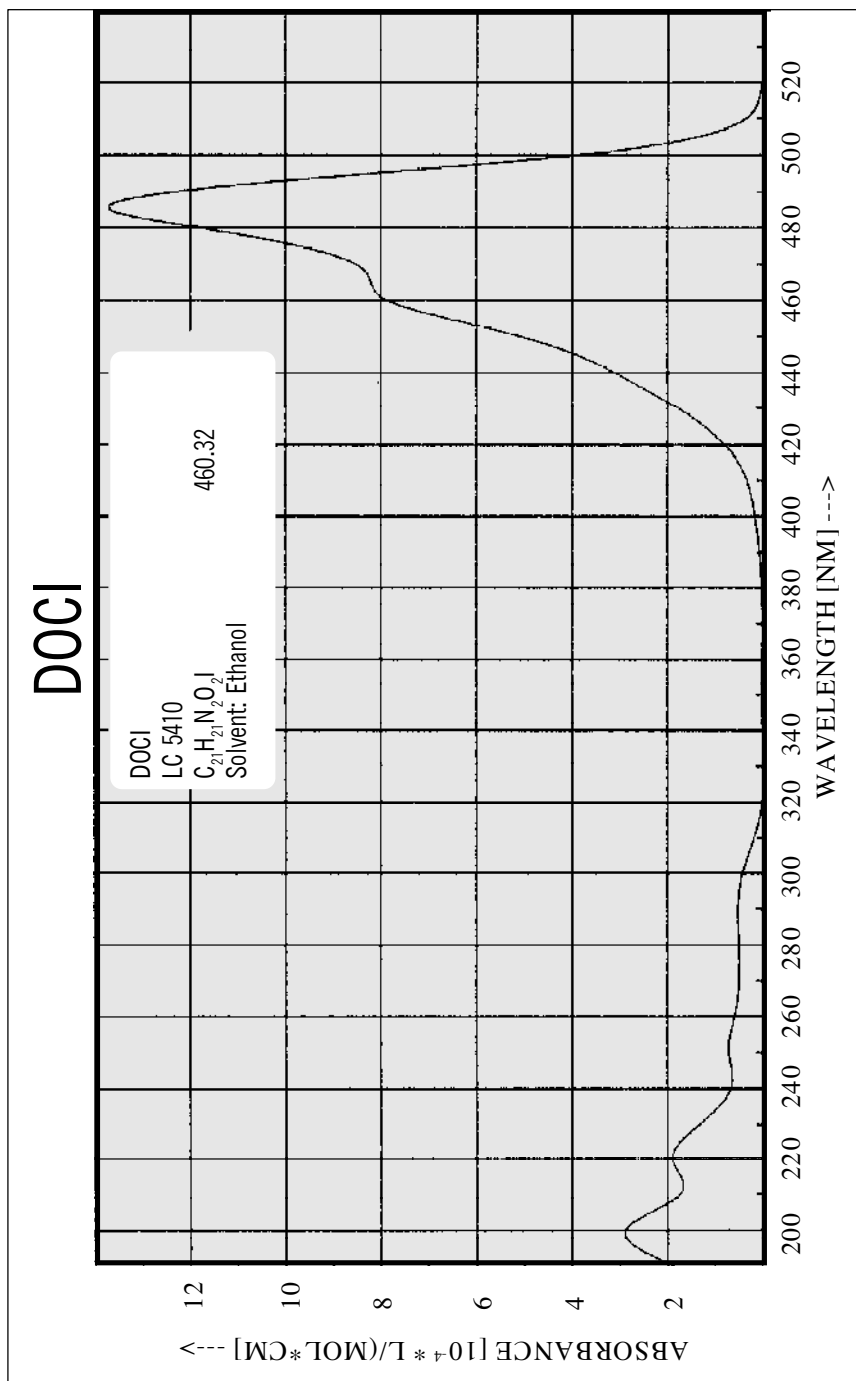
## Lasing Performance

Efficient laser dye for pulsed operation; tunable around 500 nm.

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		540	522 - 600	15	4.20	Methanol	1, 2, 3
Nd:YAG, 3rd	355		540	516 - 575	18	2.36	Methanol	1
Nitrogen	337		540	517 - 590	rel.	3.10	Methanol	4, 5
Flashlamp	-		-	528 - 547	-	-	Methanol	6

## References

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 403 (1981).
3. F. Bos, *Appl. Optics* **20**(20), 3552 (1981).
4. Lambda Physik, *Data Sheet*.
5. R. E. Drullinger, *Opt. Commun.* **39**(4), 263 (1981).
6. A. N. Fletcher *Appl. Phys.* **14**, 295 (1977).

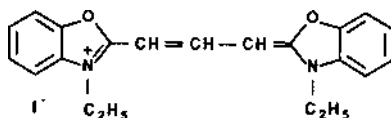


# DOCI (LC 5410)

## Constitution

3,3'-Diethyloxacarbocyanine iodide

$C_{21}H_{21}N_2O_2I$  · MW: 460.32



## Characteristics

Lambdachrome® number:	5410
CAS registry number:	-
Appearance:	red, crystalline solid
Absorption maximum (in ethanol):	485 nm
Molar absorptivity:	$12.6 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

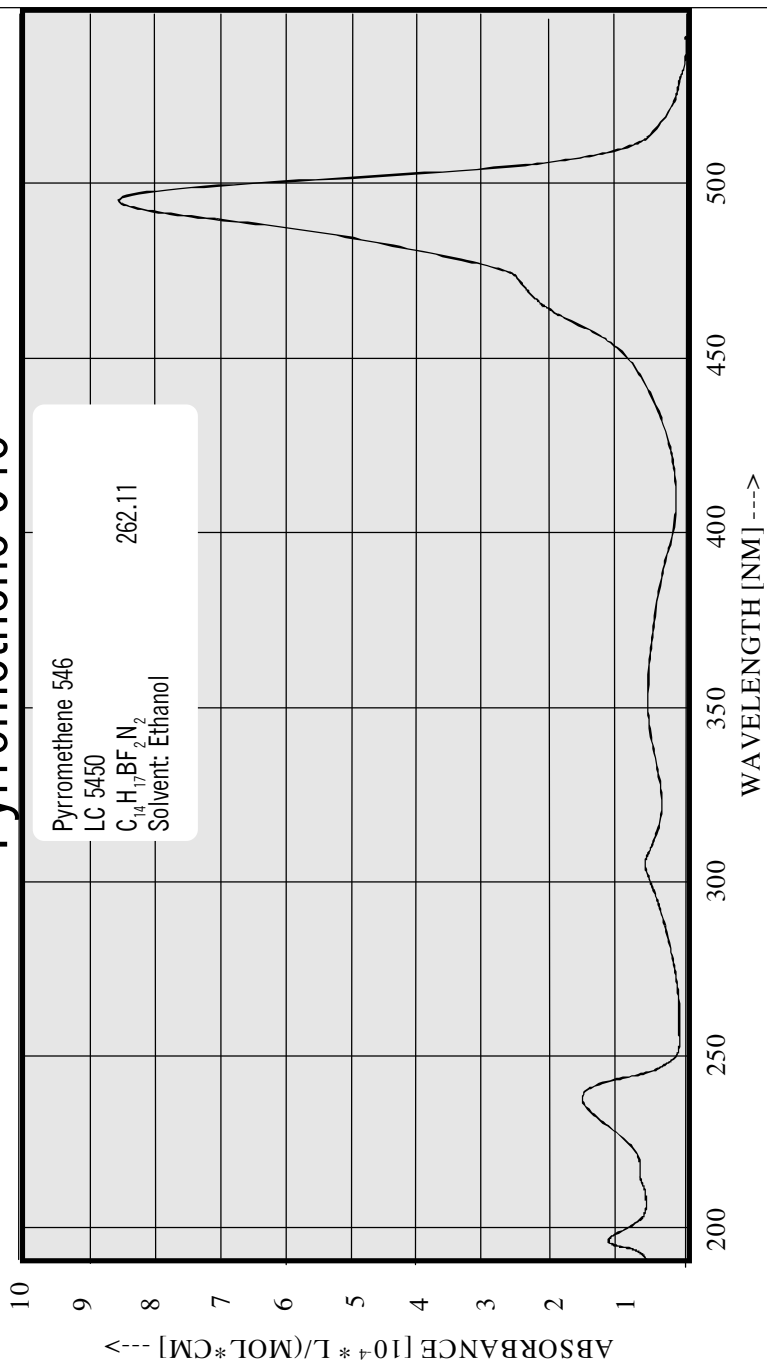
Laser dye for pulsed operation; tunable around 540 nm. Saturable absorber for Coumarin 102 dye lasers; applicalbel around 480 nm <sup>2.,3.)</sup>.

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
Flashlamp	-		541	-	-	0.09	Glyzerine	1

## References

1. M Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **11**(5), 692 (1972).
2. J. C. Mialocq, P. Goujon, *Appl. Phys. Letters* **33**(9), 819 (1978).
3. R. Wyatt, *Opt. Commun.* **38**(1), 64 (1981).

# Pyrromethene 546



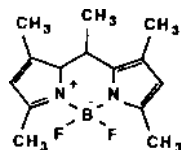


# Pyrromethene 546 (LC 5450)

## Constitution

4,4-Difluoro-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene  
1,3,5,7,8-Pentamethylpyrromethenedifluoroborate Complex

$C_{14}H_{17}BF_2N_2$  · MW: 262.11



## Characteristics

Lambdachrome® number: 5450  
CAS registry number: 121207-31-6  
Appearance: orange, crystalline solid  
Absorption maximum (in ethanol): 494 nm  
Molar absorptivity:  $8.58 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in methanol): 519 nm  
For research and development purposes only.

## Lasing Performance

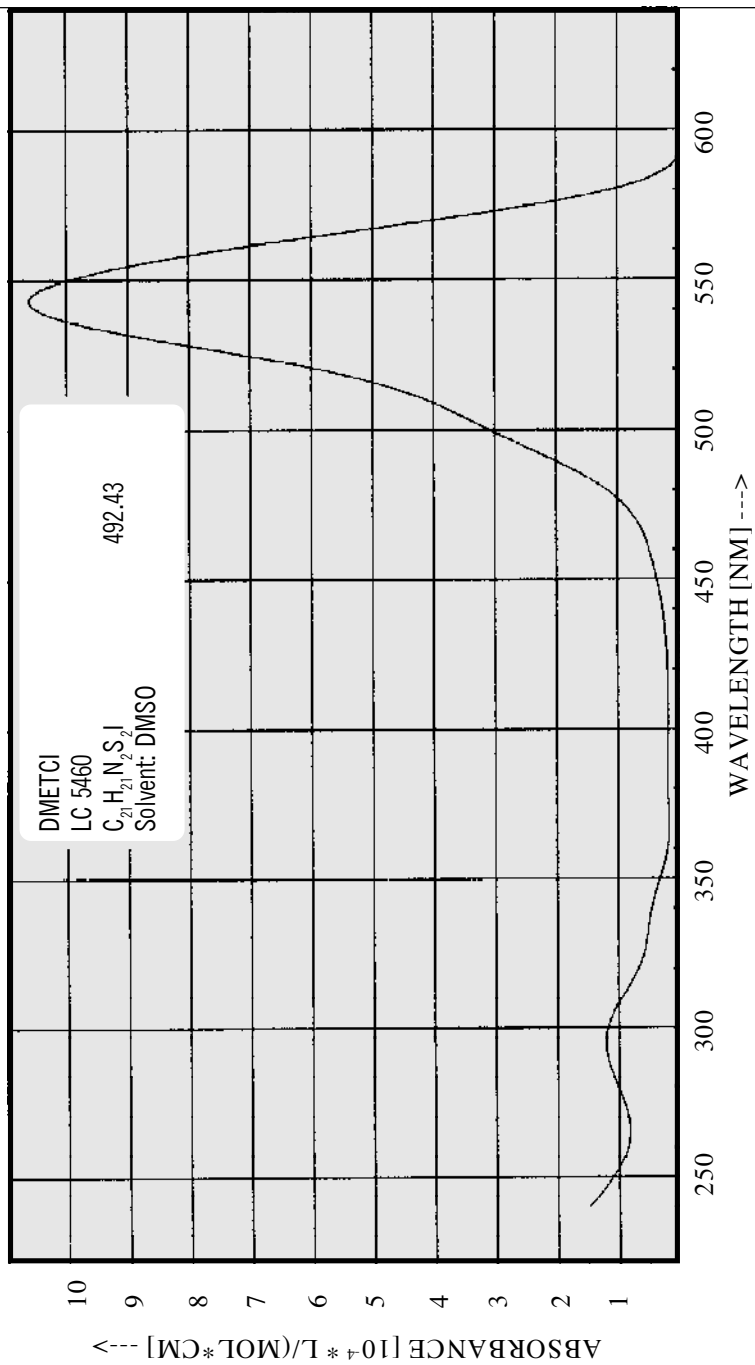
Laser dye for flashlamp pumped dye lasers; tunable around 542 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	542	523-580	-	0.066	Methanol	1
Flashlamp	-	546	-	-	0.039	Ethanol	2

## References

1. M. Shah et al., *Heteroatom Chem.* **1**(5), 389(1990).
2. Th. G. Pavlopoulos, M. Shah, J. H. Boyer, *Opt. Commun.* **70**(5), 425 (1989).

# DMETCI

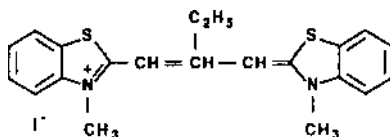


# DMETCI (LC 5460)

## Constitution

3,3'-Dimethyl-9-ethylthiacarbocyanine Iodide

$C_{21}H_{21}N_2S_2I$  · MW: 492.43



## Characteristics

Lambdachrome® number:	5460
CAS registry number:	-
Appearance:	red, crystalline solid
Absorption maximum (in ethanol):	540 nm
Molar absorptivity:	$10.6 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-

For research and development purposes only.

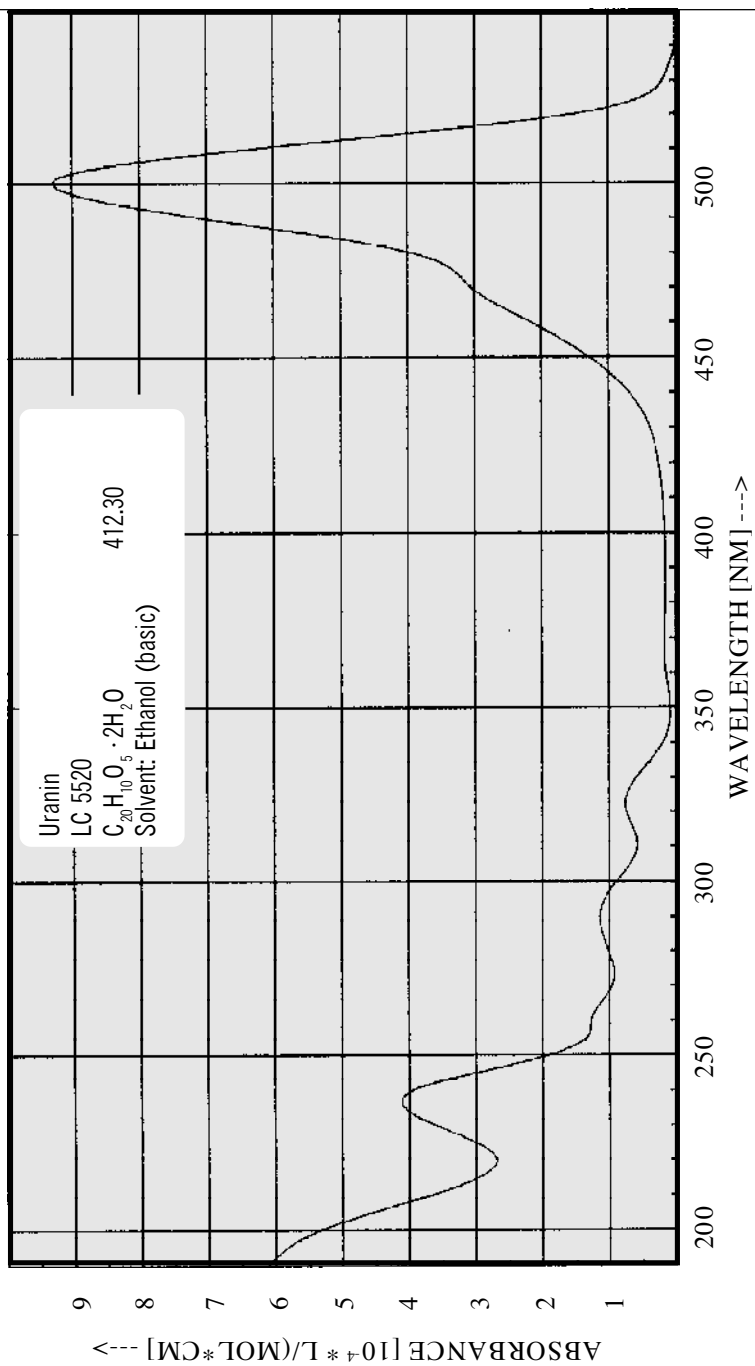
## Lasing Performance

Saturable absorber for Coumarin 153 dye lasers; applicable around 540 nm <sup>1,2</sup>.

## References

1. W. Sibbett, J. R. Taylor, *Opt. Commun.* **43**(1), 50 (1982).

# Uranin

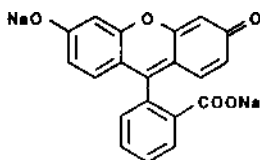


# Uranin (LC 5520)

## Constitution

Disodium Fluorescein

$C_{20}H_{10}O_5 \cdot 2H_2O$  · MW: 412.30



## Characteristics

Lambdachrome® number:	5520
CAS registry number:	518-47-8
Appearance:	red, crystalline solid
Absorption maximum (in basic ethanol):	500 nm
Molar absorptivity:	$9.92 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in bas. ethanol):	521 nm
For research and development purposes only.	

## Lasing Performance

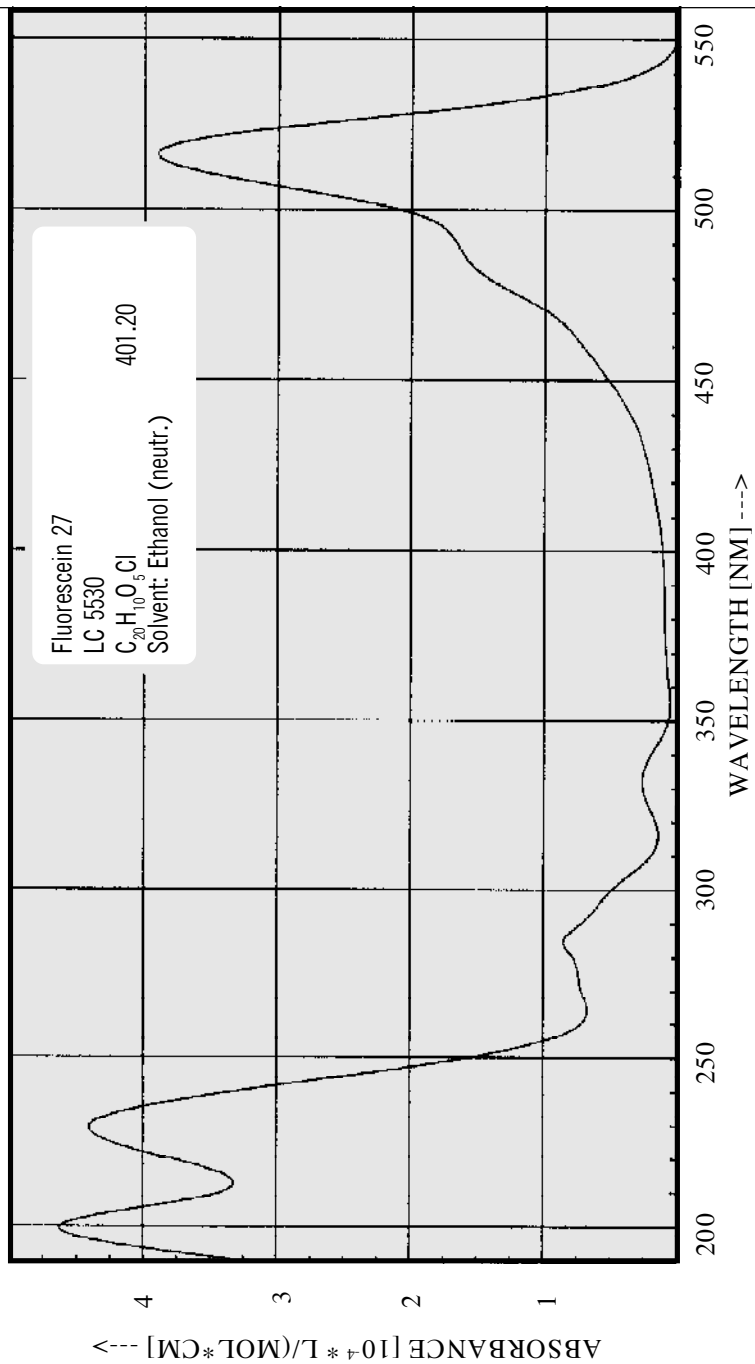
Efficient laser dye for pulsed and CW operation; tunable around 550 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	540	532 - 561	9	1.35	Methanol	1
Nitrogen	337	538	-	rel.	1.10	Methanol	2, 3
Nd:YAG, 3rd	355	550	536 - 568	-	2.06	Ethanol	4
Cu-vapor	510	528	-	-	0.42	Ethanol	5
Flashlamp	-	-	549 - 574	-	0.12	Methanol	6
CW, Ar <sup>+</sup>	all	560	530 - 590	7	1.76	MeOH/Eg.	7

## References

1. Lambda Physik, *Wall Chart* 6/83.
2. A. Dienes, *Appl. Phys.* **7**, 135 (1975).
3. G. Capelle, D. Phillips, *Appl. Optics* **9**(12), 2742 (1970).
4. D. M. Guthals, J. W. Nibler, *Opt. Commun.* **29**(3), 322 (1979).
5. L. Masarnovskii et al., *Sov. J. Quantum Electr.* **9**(7), 900 (1979).
6. D. A. Jennings, A. J. Varga, *J. Appl. Phys.* **42**(12), 5171 (1971).
7. Coherent, *CW Dye Laser Fact Sheets*.

# Fluorescein 27

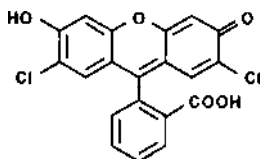


# Fluorescein 27 (LC 5530)

## Constitution

9-(o-Carboxyphenyl)-2,7-dichloro-6-hydroxy-3H-xanthen-3-on  
2,7-Dichlorofluorescein · Fluorescein 548

$C_{20}H_{10}O_5Cl$  · MW: 401.20



## Characteristics

Lambdachrome® number:	5530
CAS registry number:	76-54-0
Appearance:	red, crystalline solid
Absorption maximum (in basic ethanol):	512 nm
Molar absorptivity:	$11.0 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in bas. ethanol):	530 nm
For research and development purposes only.	

## Lasing Performance

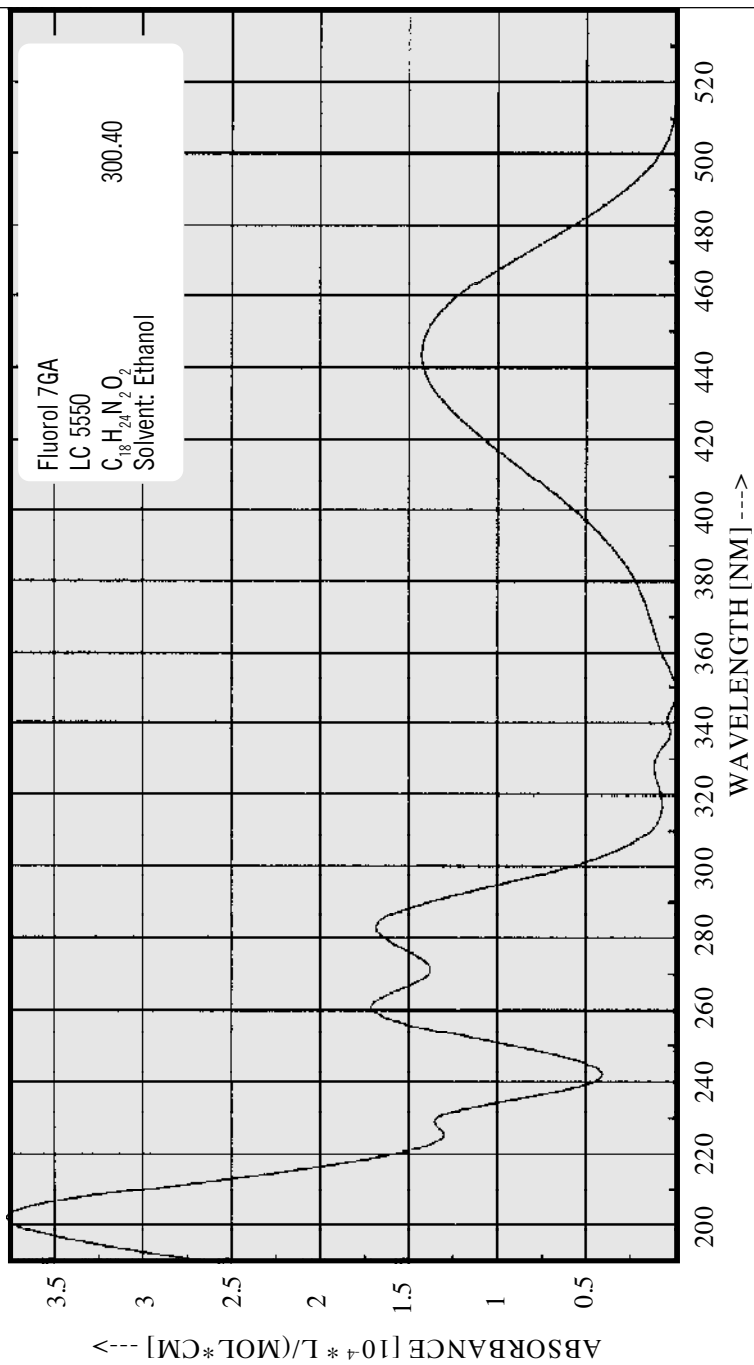
Laser dye for pulsed and CW operation; tunable around 550 nm.

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		553	540 - 587	12	1.40	Methanol	1
Nitrogen	337		558	546 - 589	rel.	1.00	Ethanol	1, 2
Nd:YAG, 2nd	532		550	540 - 575	28	0.64	Methanol	3
Flashlamp	-		-	557 - 581	-	0.20	Methanol	4

## References

1. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
2. E. D. Stokes et al., *Opt. Commun.* **5**(4), 267 (1972).
3. Lambda Physik, *Wall Chart* 1996.
4. D. A. Jennings, A. J. Varga, *J. Appl. Phys.* **42**(12), 5171 (1971).

## Fluorol 7GA



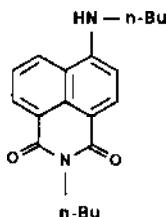


# Fluorol 7GA (LC 5550)

## Constitution

Fluorol 555

$C_{20}H_{24}N_2O_2$  · MW: 324.40



## Characteristics

Lambdachrome® number:	5550
CAS registry number:	-
Appearance:	red, crystalline solid
Absorption maximum (in methanol):	440 nm
Molar absorptivity:	$1.40 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

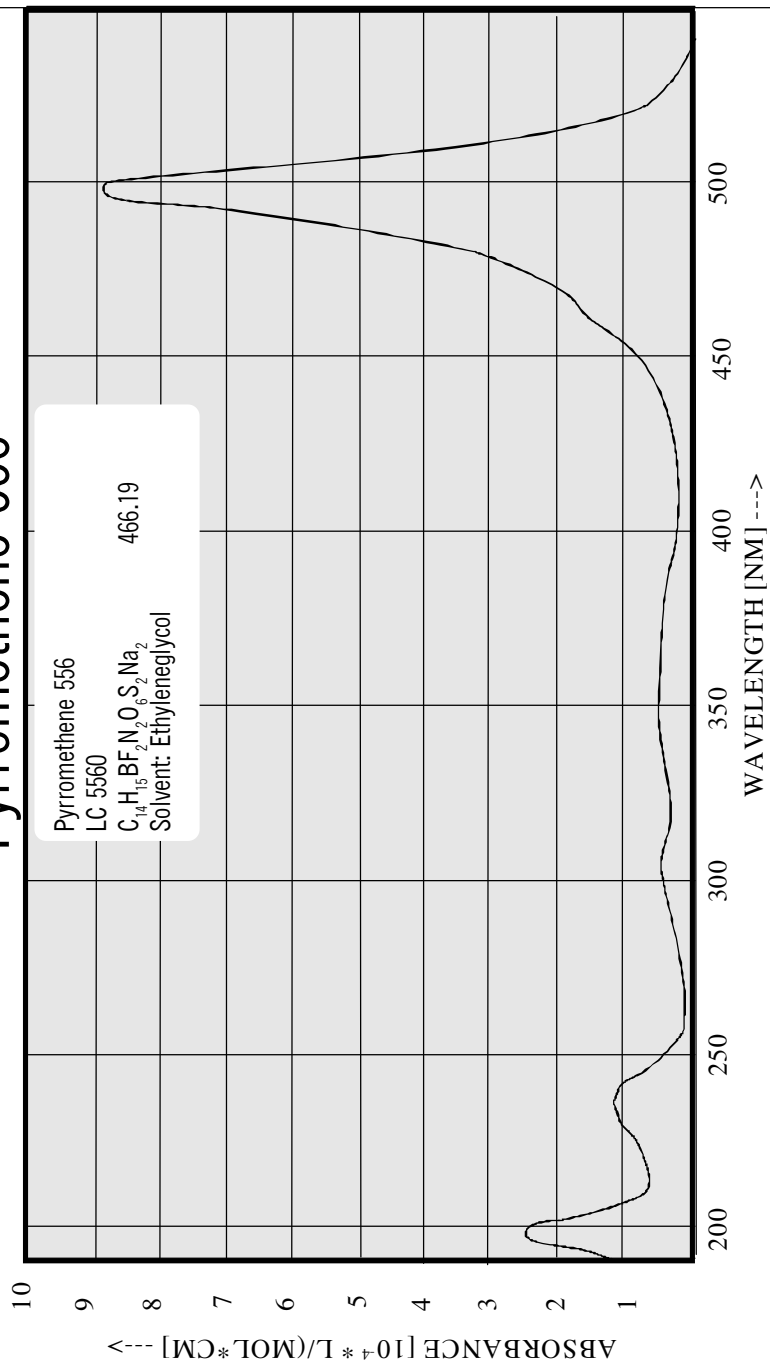
Efficient laser dye for pulsedoperation; tunable around 520 nm.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	560	530 - 590	5	1.80	Methanol	1, 2
Flashlamp	-	-	530 - 600	-	0.10	Methanol	3

## References

1. Lambda Physik.
2. F. Bos, *Appl. Optics* **20**(20), 3553, (1981).
3. M. Lambropoulos, *Opt. Commun.* **15**(1), 35 (1975).

# Pyrromethene 556

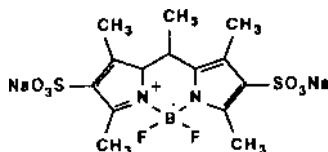


# Pyrromethene 556 (LC 5560)

## Constitution

Disodium-1,3,5,7,8-pentamethylpyrromethene-2,6-disulfonate-difluoroborate complex

$C_{14}H_{17}BF_2N_2O_6S_2Na_2$  · MW: 466.19



## Characteristics

Lambdachrome® number: 5560  
CAS registry number: 121461-69-6  
Appearance: yellow/orange, crystalline solid  
Absorption maximum (in ethyleneglycol): 498 nm  
Molar absorptivity:  $8.88 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in methanol): 533 nm  
For research and development purposes only.

## Lasing Performance

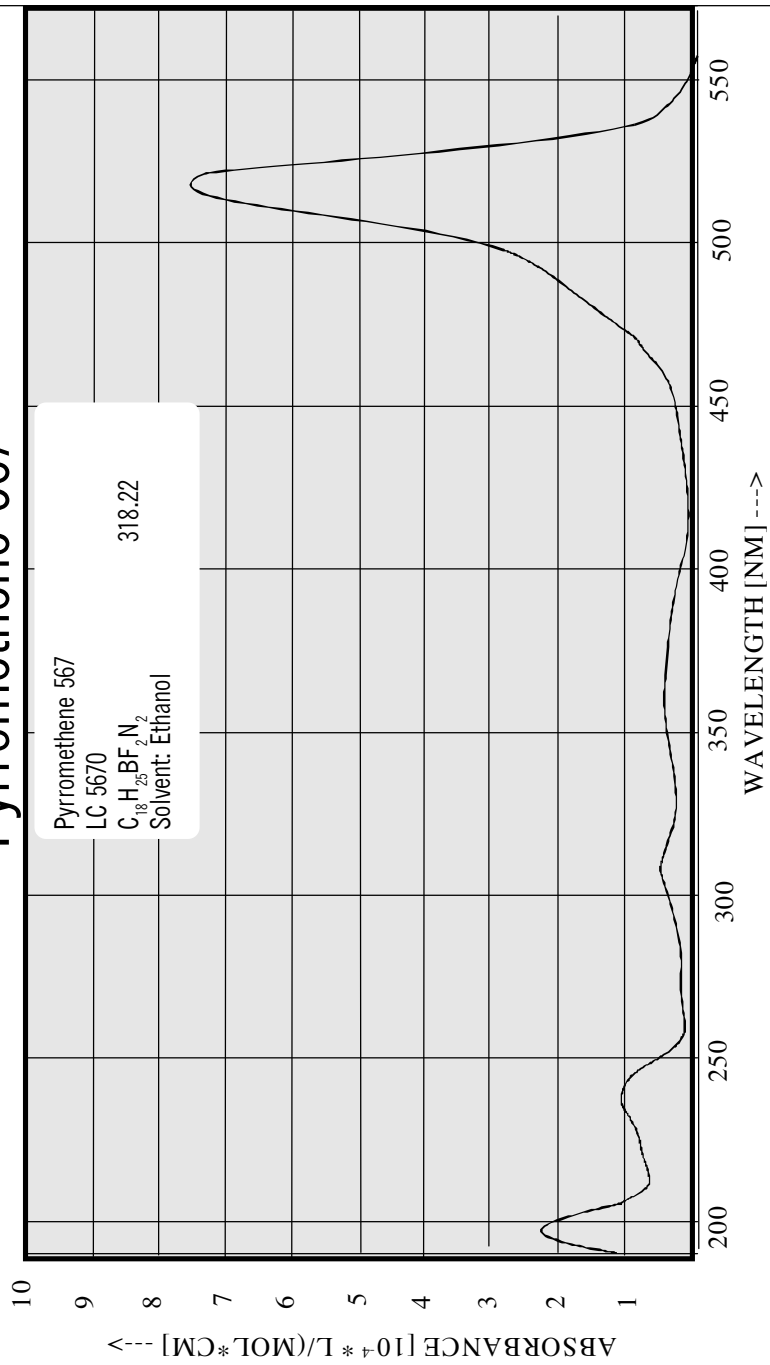
Laser dye for pulsed and CW operation; tunable around 550 nm.

Pump		Dye Laser Characteristics				Solvent	Ref.
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	561	540-580	-	0.03	Methanol	1
CW, Ar <sup>+</sup>	458-514	553	530-624	45	0.93	Eg.	2

## References

1. M. Shah et al., *Heteroatom Chem.* **1**(5), 389(1990).
2. S. G. Guggenheimer et al., *Appl. Optics* **32**(21), 3942 (1993).

# Pyrromethene 567



# Pyrromethene 567 (LC 5670)

## Constitution

4,4-Difluoro-2,6-diethyl-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene  
2,6-Diethyl-1,3,5,7,8-pentamethylpyrromethenedifluoroborate Complex

$C_{18}H_{25}BF_2N_2$  · MW: 318.22



## Characteristics

Lambdachrome® number:	5670
CAS registry number:	131083-16-4
Appearance:	orange/red, crystalline solid
Absorption maximum (in ethanol):	518 nm
Molar absorptivity:	$7.73 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	547 nm
For research and development purposes only.	

## Lasing Performance

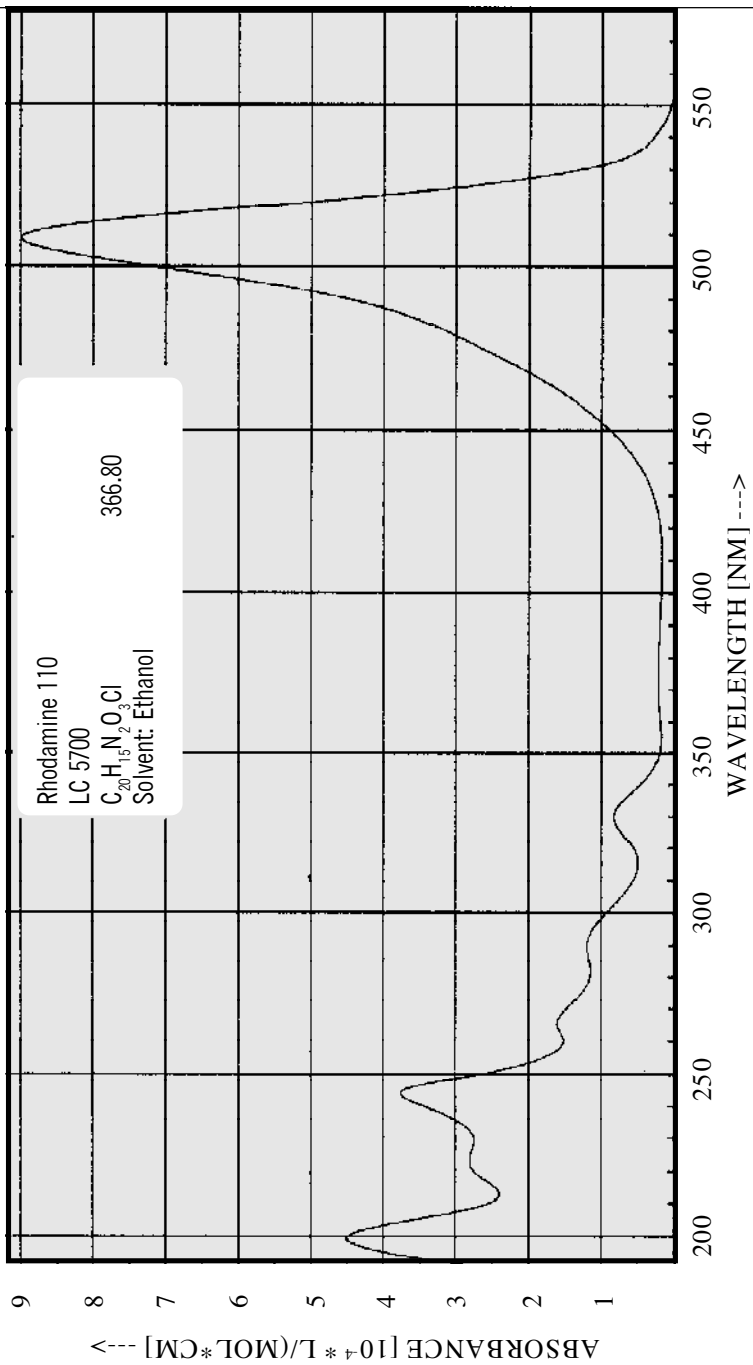
Efficient laser dye for pulsed and CW operation; tunable around 570 nm.  
Alternative to Rhodamine 6G.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	567	-	-	0.06	Ethanol	1, 2
CW, Ar <sup>+</sup>	all	571	552-608	36	0.45	PC	3
Nd:YAG	532	566	549-592	44	0.31	PC	4

## References

1. M. Shah et al., *Heteroatom Chem.* **1**(5), 389(1990).
2. T. G. Pavlopoulos et al., *Appl. Optics* **29**(27), 3885 (1990).
3. S. G. Guggenheimer et al., *Appl. Optics* **32**(21), 3942 (1993).
4. M. P. O'Neil, *Optics Letters* **18**(1), 37 (1993).
5. R. E. Hermes et al., *Appl. Phys. Letters* **63**(7), 877 (1993).

## Rhodamine 110



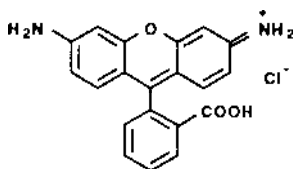
# Rhodamine 110 (LC 5700)

## Constitution

o-(6-Amino-3-imino-3H-xanthen-9-yl)-benzoic acid

Rhodamine 560

$C_{20}H_{15}N_2O_3Cl$  · MW: 366.80



## Characteristics

Lambdachrome® number:	5700
CAS registry number:	13558-31-1
Appearance:	red, crystalline solid
Absorption maximum (in ethanol):	510 nm
Molar absorptivity:	$8.99 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	535 nm
For research and development purposes only.	

## Lasing Performance

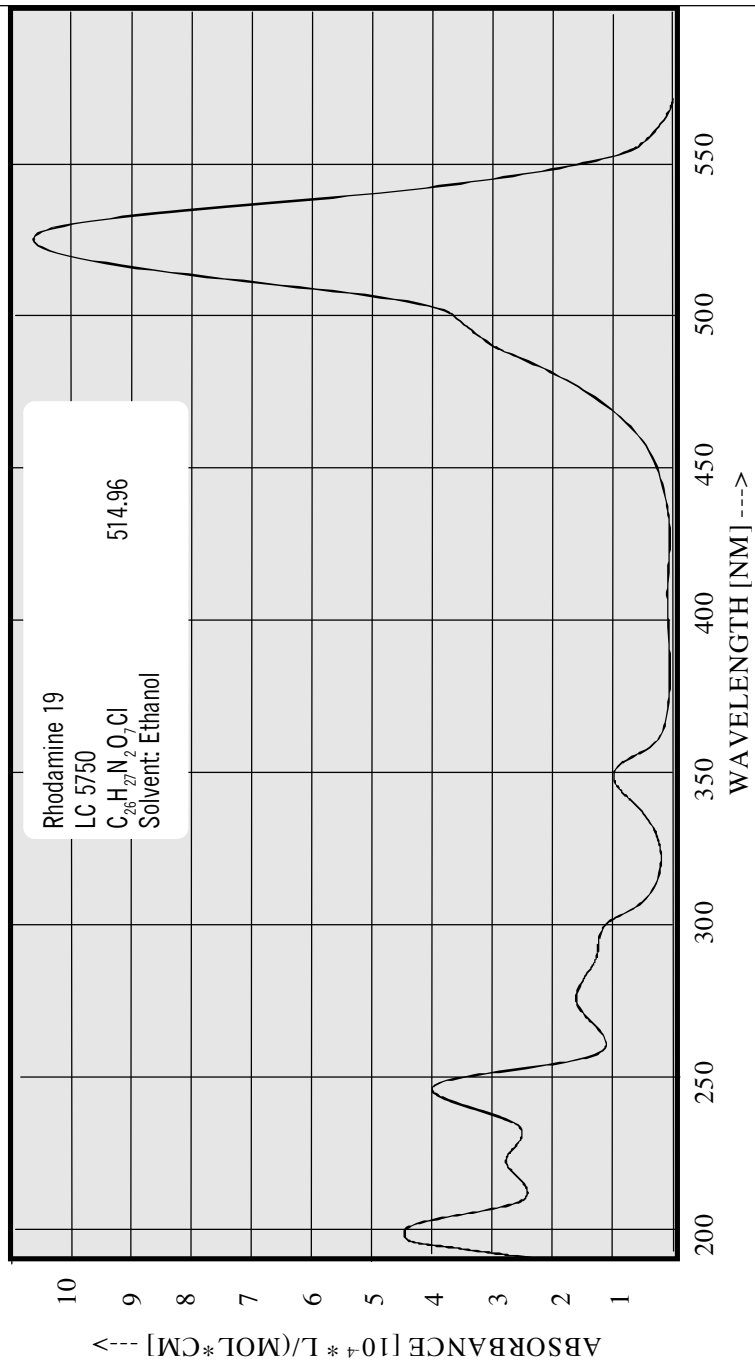
Efficient laser dye for pulsed and CW operation; tunable around 570 nm.

Source	Pump		Dye Laser Characteristics			Solvent	Ref.
	Wavelength	Peak	Range	Effic.	Conc.		
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308	572	547 - 592	5	1.22	Ethanol	1
Cu-vapor	510	550	528 - 574	9	0.09	Methanol	2
Flashlamp	-	-	551 - 583	-	0.07	Ethanol	3
CW, Ar <sup>+</sup>	all	560	530 - 580	-	0.75	Eg.	4, 5, 6

## References

1. Lambda Physik.
2. M. Broyer et al., *Appl. Phys.* **B35**, 31 (1984).
3. W. Sibbett, J. R. Taylor, *IEEE J. Quantum Electron.* **QE-19**(4), 558 (1983).
4. Coherent, *CW Dye Laser Fact Sheets*.
5. T. F. Johnston et al., *Appl. Optics* **21**(13), 2307 (1982).
6. Lambda Physik, Wall Chart 1996.

## Rhodamine 19



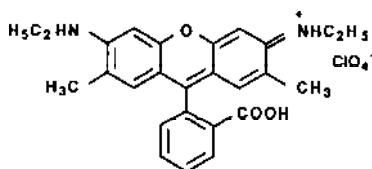


# Rhodamine 19 (LC 5750)

## Constitution

Benzoic Acid, 2-[6-(ethylamino)-3-(ethylimino)-2,7-dimethyl-3H-xanthen-9-yl], perchlorate  
Rhodamine 575

$C_{26}H_{27}N_2O_7Cl$  · MW: 514.96



## Characteristics

Lambdachrome® number: 5750  
CAS registry number: 62669-66-3  
Appearance: red, crystalline solid  
Absorption maximum (in ethanol): 528 nm  
Molar absorptivity:  $10.9 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum: -  
For research and development purposes only.

## Lasing Performance

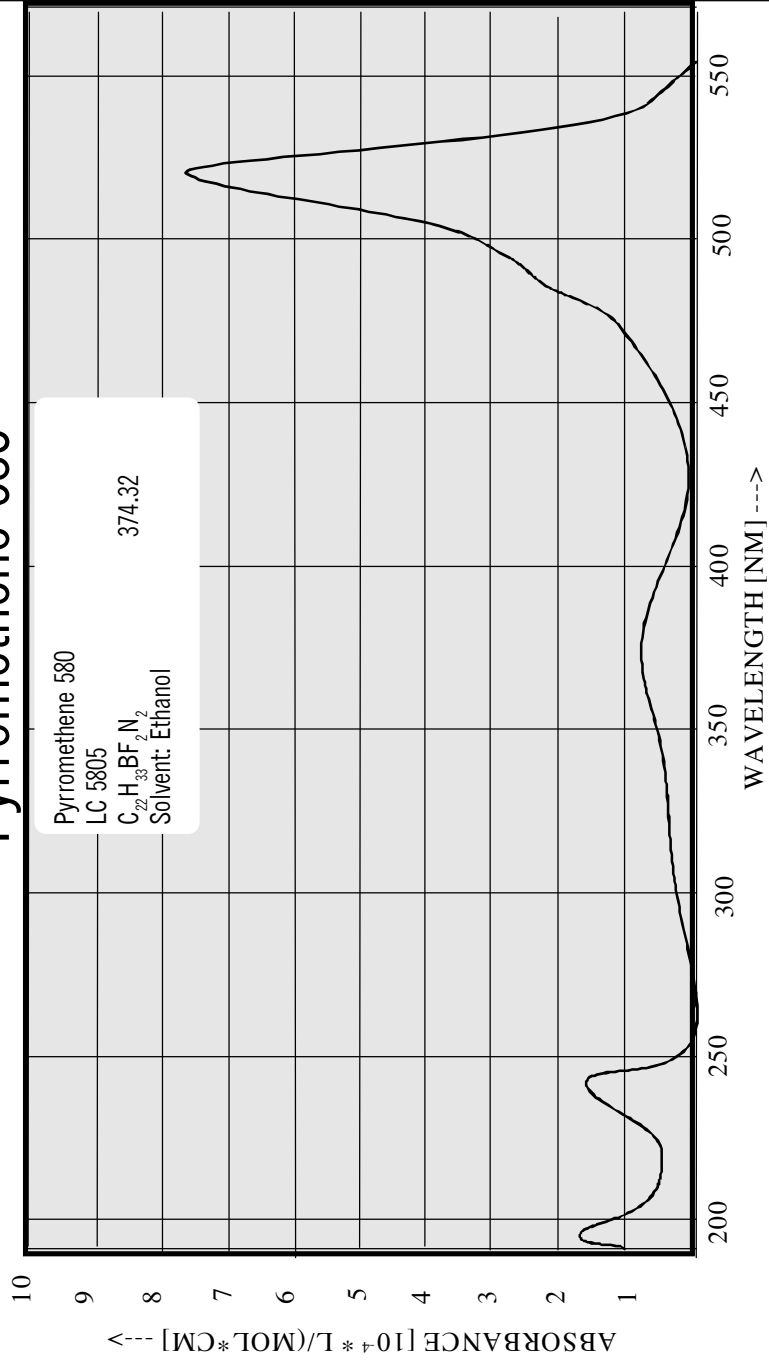
Efficient laser dye for pulsed operation; tunable around 560 nm.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nd:YAG, 3rd	355	562	552 - 582	-	0.21	Ethanol	1
Nd:YAG, 2nd	532	567	556 - 586	31	0.22	Methanol	2

## References

1. D. M. Guthals, J. W. Nibler, *Opt. Commun.* **29**(3), 322 (1979).
2. Lambda Physik, *Wall Chart* **1996**.

# Pyrrromethene 580

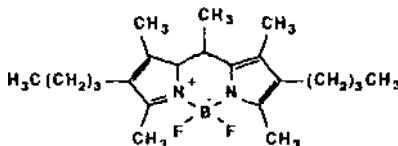


# Pyrromethene 580 (LC 5805)

## Constitution

4,4-Difluoro-2,6-di-n-butyl-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene  
2,6-Di-n-butyl-1,3,5,7,8-pentamethylpyrromethenedifluoroborate Complex

$C_{22}H_{33}BF_2N_2$  · MW: 374.32



## Characteristics

Lambdachrome® number:	5805
CAS registry number:	N/A
Appearance:	orange/red, crystalline solid
Absorption maximum (in ethanol):	519 nm
Molar absorptivity:	$7.68 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	550 nm
For research and development purposes only.	

## Lasing Performance

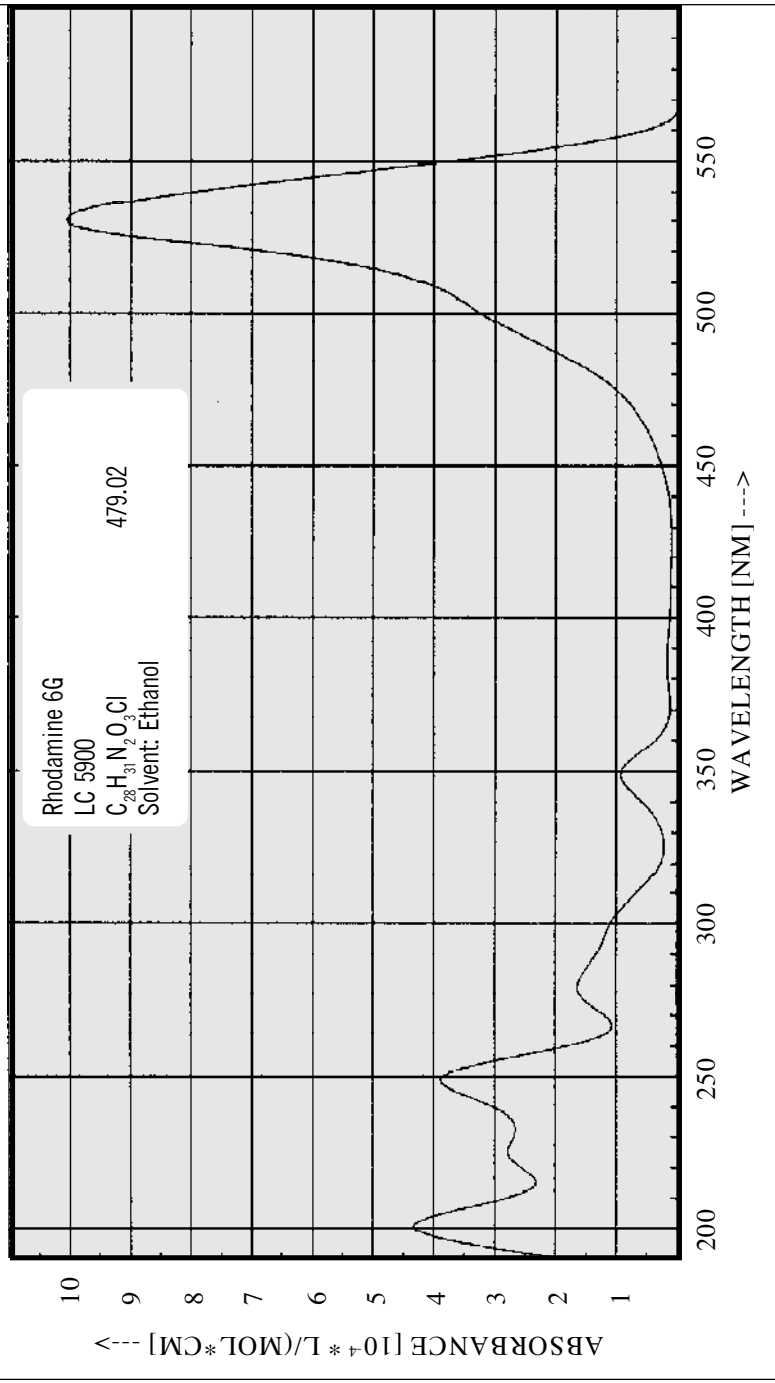
Laser dye for pulsed operation; tunable around 580 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	580	-	-	0.08	Ethanol	1

## References

1. M. Shah et al., *Heteroatom Chem.* **1**(5), 389(1990).
2. R. E. Hermes et al., *Appl. Phys. Letters* **63**(7), 877 (1993).

# Rhodamine 6G

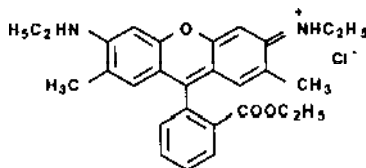


# Rhodamine 6G (LC 5900)

## Constitution

Benzoic Acid, 2-[6-(ethylamino)-3-(ethylimino)-2,7-dimethyl-3H-xanthen-9-yl]-ethyl ester, monohydrochloride  
Rhodamine 590

$C_{28}H_{31}N_2O_3Cl$  · MW: 479.02



## Characteristics

Lambdachrome® number: 5900  
CAS registry number: 989-38-8  
Appearance: red, crystalline solid  
Absorption maximum (in ethanol): 530 nm  
Molar absorptivity:  $10.50 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in ethanol): 556 nm  
For research and development purposes only.

## Lasing Performance

The laser dye "per se" Rhodamine 6G is by far the most frequently used and most widely investigated laser dye. Very efficient laser dye for pulsed and CW operation; tunable around 590 nm.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	581	569 - 608	16	1.20	Methanol	1, 2, 3
Nitrogen	337	581	573 - 618	rel.	1.63	Methanol	3, 4, 5
Nd:YAG, 2nd	532	566	555 - 585	32	0.10	Methanol	1, 6, 7
Flashlamp	-	600	555 - 620	-	1.20	Ethanol	9, 10
CW, Ar <sup>+</sup>	all	575	560 - 625	-	0.75	Eg.	1, 11, 12, 13

## References

See page 164.

## References (RHODAMINE 6G)

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 403 (1981).
3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. A. Dienes, *Appl. Phys.* 7, 135 (1975).
6. D. M. Guthals, J. W. Nibler, *Opt. Commun.* 29(3), 322 (1979).
7. C. A. Moore, C. D. Decker, *J. Appl. Phys.* 49(1), 47 (1978).
8. M. Broyer et al., *Appl. Phys.* B35, 31 (1984).
9. P. R. Hammond, *Opt. Commun.* 29(3), 331 (1979).
10. J. Jethwa, F. P. Schäfer, *Appl. Phys.* 4, 299 (1974).
11. Coherent, *CW Dye Laser Fact Sheets*.
12. H. J. Baving et al., *Appl. Phys.* B29, 19 (1982).
13. T. F. Johnston et al., *Appl. Optics* 21(13), 2307 (1982).

## References (RHODAMINE B)

1. Lambda Physik, *Wall Chart* 1996.
2. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
3. Lambda Physik, *Data Sheet*.
4. A. Dienes, *Appl. Phys.* 7, 135 (1975).
5. I. A. Stenhouse, D. R. Williams, *Appl. Spectrosc.* 33(2), 175 (1979).
6. Q. H. F. Vrehen, *Opt. Commun.* 3(3), 144 (1971).
7. L. Masarnovskii et al., *Sov. J. Quantum Electron.* 9(7), 900 (1979).
8. R. S. Hargrove, T. Kan, *IEEE J. Quantum Electron.* QE-13, 28D (1977).
9. J. M. Drake et al., *Chem. Phys. Letters* 35(2), 181 (1975).
10. P. R. Hammond, *Opt. Commun.* 29(3), 331 (1979).
11. Coherent, *CW Dye Laser Fact Sheets*.

## References (SULFORHODAMINE B)

1. Lambda Physik, *Wall Chart* 1996.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* B32, 9 (1983).
3. Lambda Physik, *Data Sheet*.
4. M. Broyer et al., *Appl. Phys.* B35, 31 (1984).
5. P. R. Hammond, *Opt. Commun.* 29(3), 331 (1979).
6. R. M. Schotland, *Appl. Optics* 19(1), 124 (1980).
7. J. M. Yarborough, *Appl. Phys. Letters* 24(12), 629 (1974).

## References (RHODAMINE 101)

1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 403 (1981).
3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. Lambda Physik.
6. Bos, *Appl. Optics* 20(10), 1886 (1981).
7. M. Broyer et al., *Appl. Phys.* B35, 31 (1984).
8. T. J. Negran, A. M. Glass, *Appl. Optics* 17(17), 2812 (1978).
9. Coherent, *CW Dye Laser Fact Sheets*.

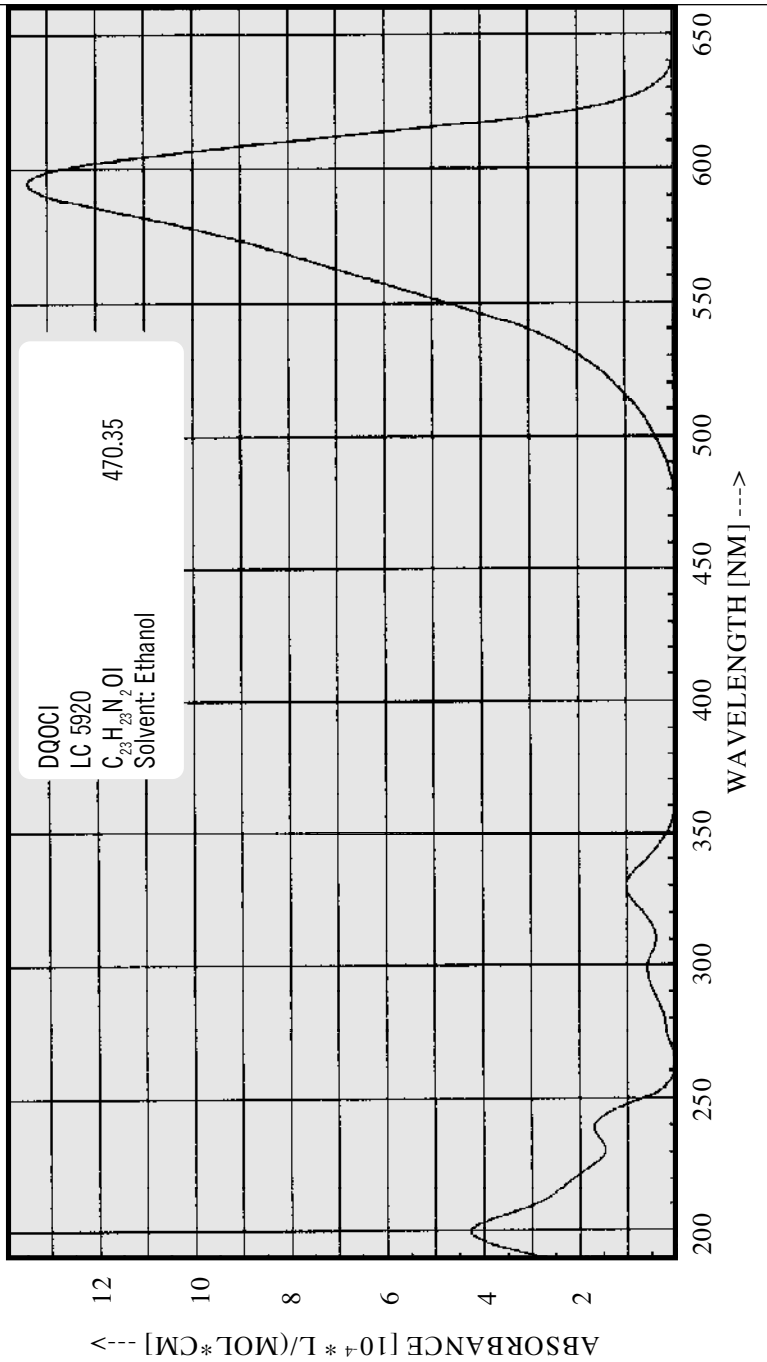
## References (DCM)

1. Lambda Physik, *Wall Chart* 1996.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* B32, 9 (1983).
3. Lambda Physik, *Data Sheet*.
4. Lambda Physik.
5. M. Broyer et al., *Appl. Phys.* B35, 31 (1984).
6. P. R. Hammond, *Opt. Commun.* 29(3), 331 (1979).
7. G. P. Weber, *IEEE J. Quantum Electron.* QE-19(7), 1200 (1983).
8. Coherent, *Data Sheet*.
9. E. G. Marason, *Opt. Commun.* 37(1), 56 (1981).
10. T. F. Johnston et al., *Appl. Optics* 21(13), 2307 (1982).

## References (CRESYL VIOLET)

1. Lambda Physik.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6) 403 (1981).
3. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
4. F. Castelli, *Appl. Phys. Letters* 26(1), 18 (1975).
5. A. Dienes, *Appl. Phys.* 7, 135(1975).
6. I. A. Stenhouse, D. R. Williams, *Appl. Spectrosc.* 33(2), 175 (1979).
7. C. A. Moore, C. D. Decker, *J. Appl. Phys.* 49(1), 47 (1978).
8. W. Schmidt, W. Appt, N. Wittekindt, *Z. Naturforsch.* 27a, 37 (1972).
9. J. B. Marling et al., *Appl. Optics* 13(10), 2317 (1974).
10. J. M. Yarborough, *Appl. Phys. Letters* 24(12), 629 (1974).

# DQOCI



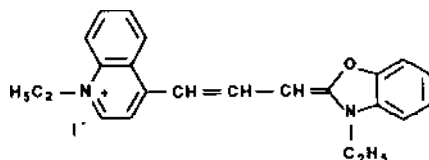


# DQOCI (LC 5920)

## Constitution

1,3'-Diethyl-4,2'-quinolyloxacarbocyanine Iodide

$C_{23}H_{23}N_2OI$  · MW: 470.35



## Characteristics

Lambdachrome® number:	5920
CAS registry number:	-
Appearance:	violet, crystalline solid
Absorption maximum (in ethanol):	592 nm
Molar absorptivity:	$13.5 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

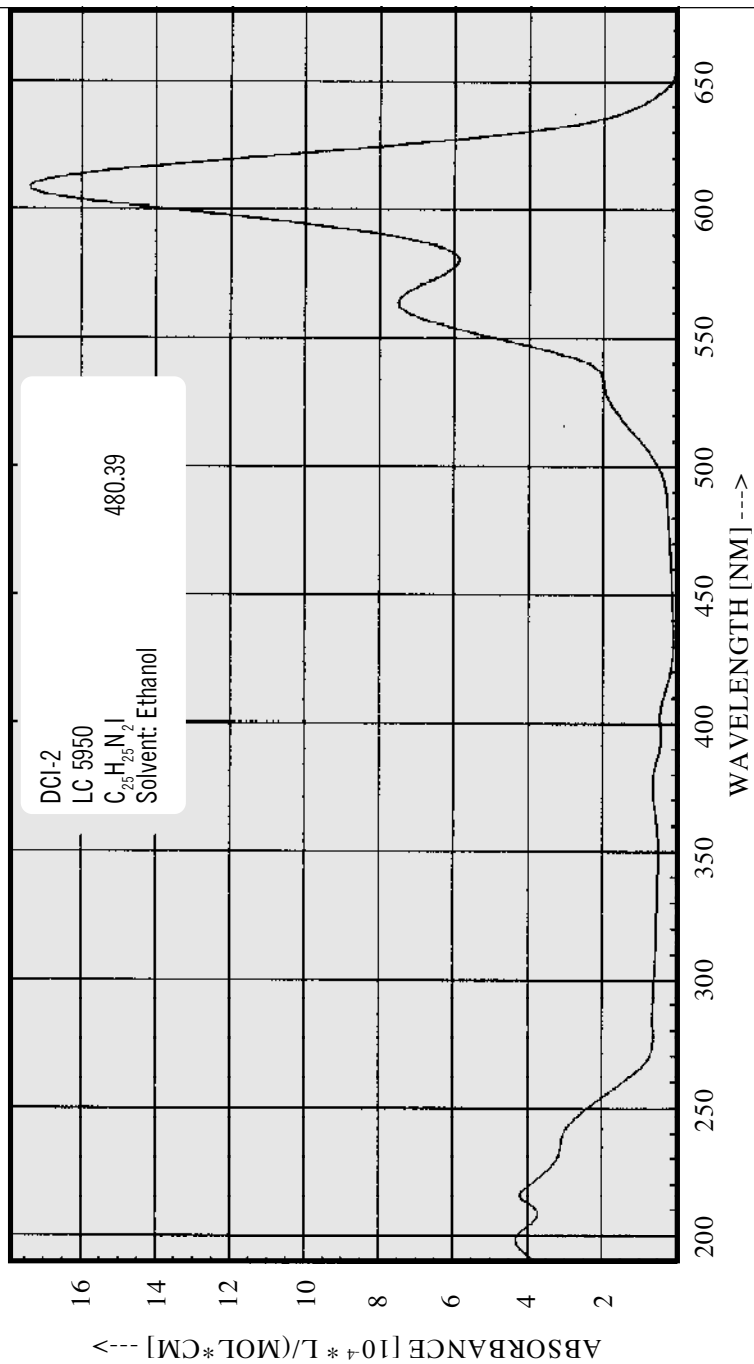
## Lasing Performance

Saturable absorber for flashlamp pumped Fluorol 7GA and Rhodamine 6G dye lasers <sup>1,2</sup>). Applicable around 590 nm.

## References

1. E. Lill, S. Schneider, F. Dörr, *Opt. Commun.* **20**(2), 223 (1977).
2. R. S. Adrain et al., *Opt. Commun.* **12**(2), 140 (1974).

# DCI-2



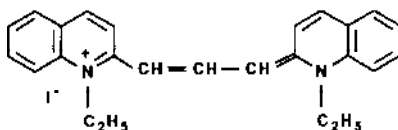
# DCI-2 (LC 5950)

## Constitution

---

1,1'-Diethyl-2,2'-carbocyanine Iodide  
Pinacyanol Iodide · Chinaldinblau

$C_{25}H_{25}N_2I$  · MW: 480.39



## Characteristics

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Lambdachrome® number:	5950
CAS registry number:	-
Appearance:	violet, crystalline solid
Absorption maximum (in ethanol):	606 nm
Molar absorptivity:	$17.0 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

---

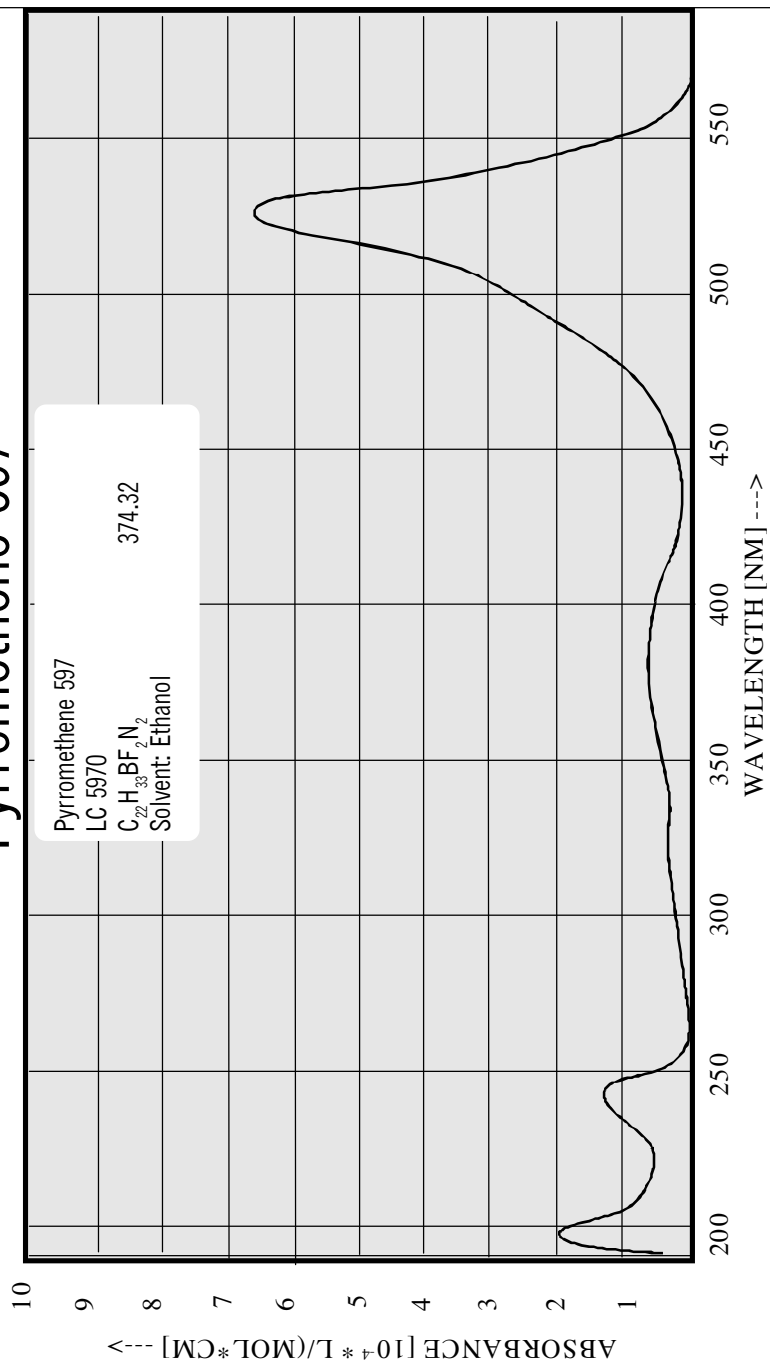
Saturable absorber for flashlamp pumped Rhodamine 6G dye lasers <sup>1).</sup> Applicable around 606 nm.

## References

---

1. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **13**(1), 193 (1974).

# Pyrromethene 597

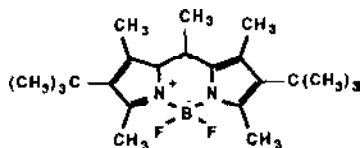


# Pyrromethene 597 (LC 5970)

## Constitution

4,4-Difluoro-2,6-di-*t*-butyl-1,3,5,7,8-pentamethyl-4-bora-3a,4a-diaza-s-indacene  
2,6-Di-*t*-butyl-1,3,5,7,8-pentamethylpyrromethenedifluoroborate Complex

$C_{22}H_{33}BF_2N_2$  · MW: 374.32



## Characteristics

Lambdachrome® number:	5970
CAS registry number:	137829-79-9
Appearance:	red, crystalline solid
Absorption maximum (in ethanol):	524 nm
Molar absorptivity:	$6.76 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	557 nm
For research and development purposes only.	

## Lasing Performance

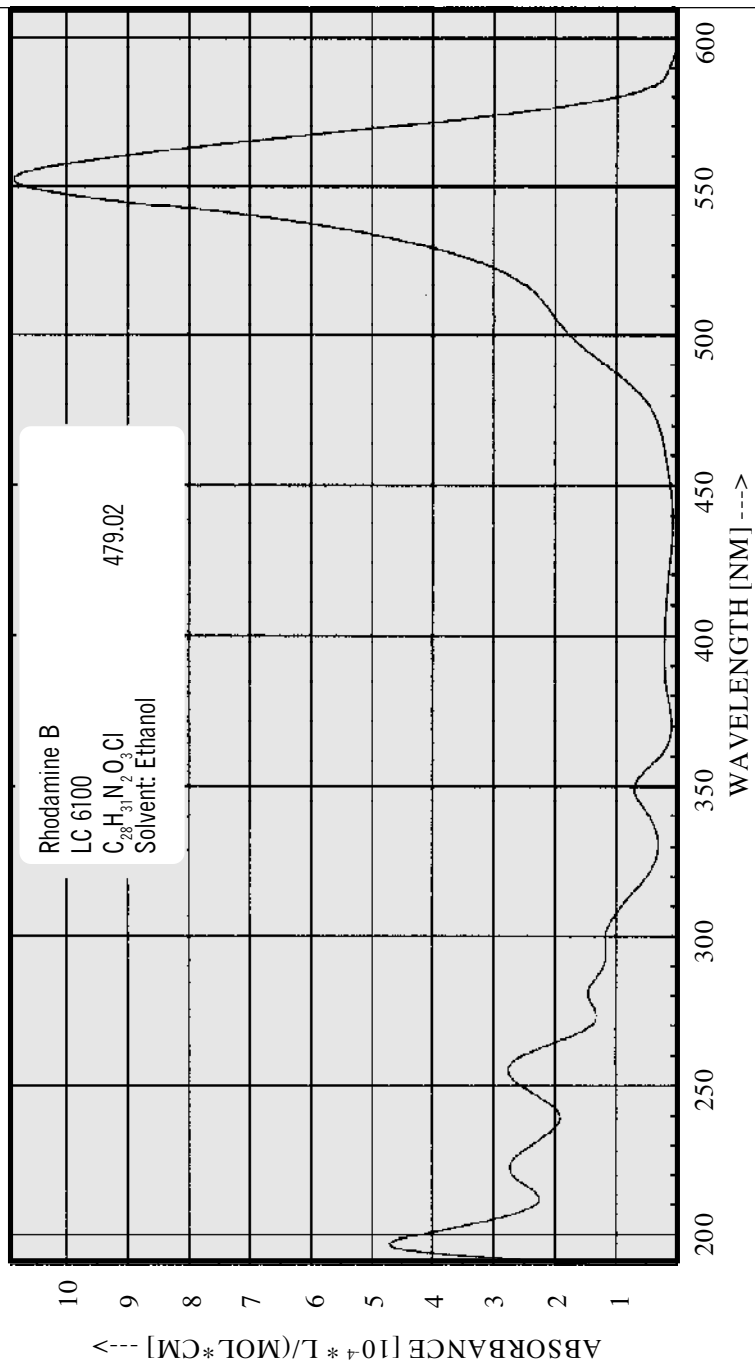
Laser dye for pulsed operation; tunable around 590 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
Flashlamp	-	593	-	-	0.08	Ethanol	1, 2

## References

1. J. H. Boyer et al., *Appl. Optics* **30**(27), 3788 (1991).
2. J. H. Boyer et al., *Heteroatom Chem.* **4**(1), 39 (1993).

# Rhodamine B

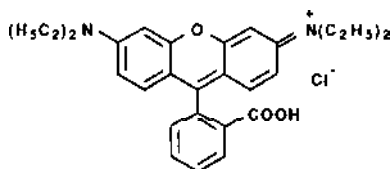


# Rhodamine B (LC 6100)

## Constitution

2-[6-(Diethylamino)-3-(diethylimino)-3H-xanthen-9-yl] benzoic acid  
Rhodamine 610

$C_{28}H_{31}N_2O_3Cl$  · MW: 479.02



## Characteristics

Lambdachrome® number: 6100  
CAS registry number: 81-88-9  
Appearance: green, crystalline solid  
Absorption maximum (in ethanol): 552 nm  
Molar absorptivity:  $10.7 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in ethanol): 580 nm  
For research and development purposes only.

## Lasing Performance

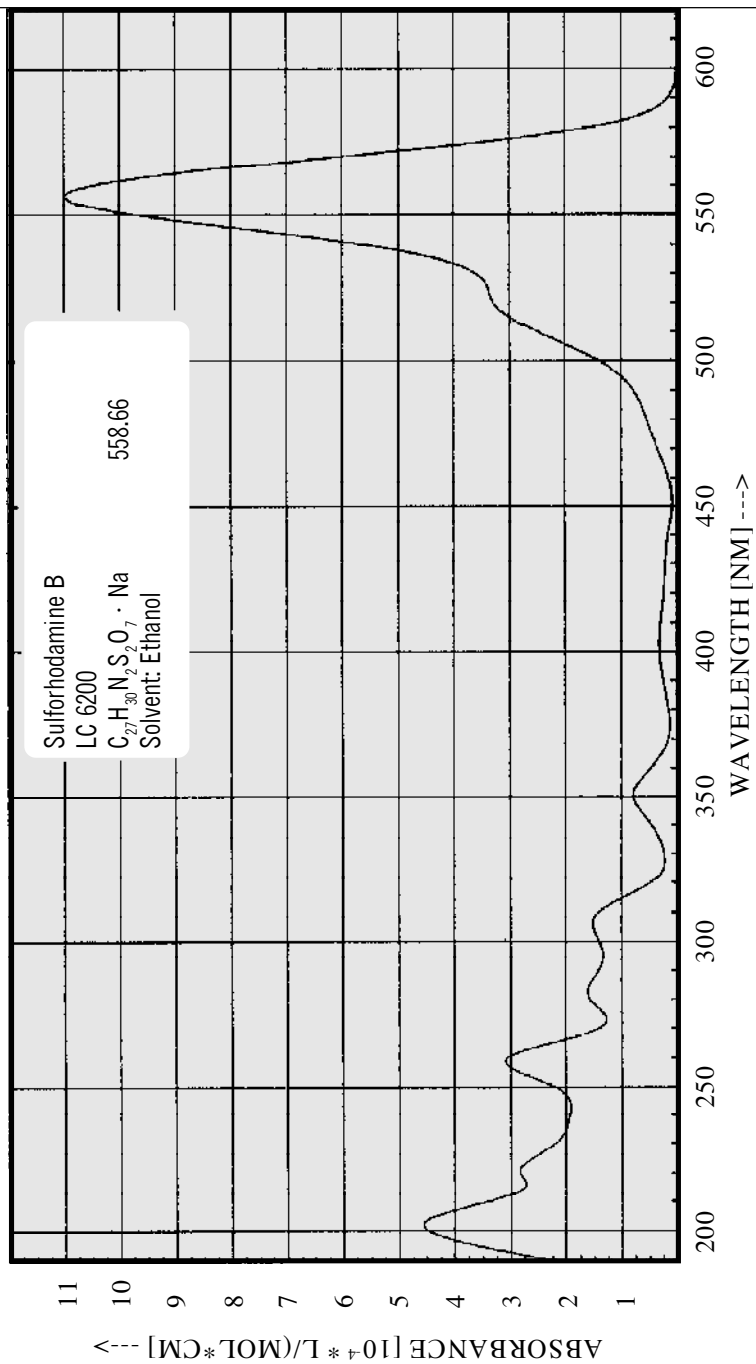
Very efficient and frequently used laser dye for pulsed and CW operation; tunable around 610 nm.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	600	588 - 644	12	0.91	Methanol	1, 2
Nitrogen	337	622	599 - 650	rel.	2.13	Methanol	2, 3, 4
Nd:YAG, 2nd	532	594	584 - 619	29	0.22	Methanol	1, 5, 6
Cu-vapo	510	591	582 - 618	21	0.62	Ethanol	7, 8
Flashlamp	-	618	590 - 640	-	0.05	Ethanol	9, 10
CW, Ar <sup>+</sup>	all	640	605 - 675	-	3.53	MeOH/Eq.	11

## References

see page 164.

# Sulforhodamine B



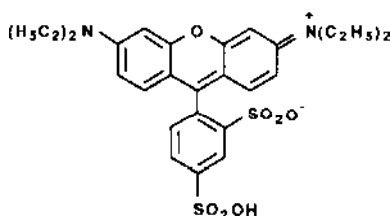


# Sulforhodamine B (LC 6200)

## Constitution

Ethanaminium, N-[6-diethylamino)-9-(2,4-disulfophenyl)-3H-xanthen-3-ylidene]-  
N-ethylhydroxid, inner salt, sodium salt  
Kiton Red 620 · Kiton Red S

$C_{27}H_{30}N_2S_2O_7 \cdot Na$  · MW: 558.66



## Characteristics

Lambdachrome® number: 6200  
CAS registry number: 3520-42-1  
Appearance: green, crystalline solid  
Absorption maximum (in ethanol): 556 nm  
Molar absorptivity:  $11.1 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in ethanol): 575 nm  
For research and development purposes only.

## Lasing Performance

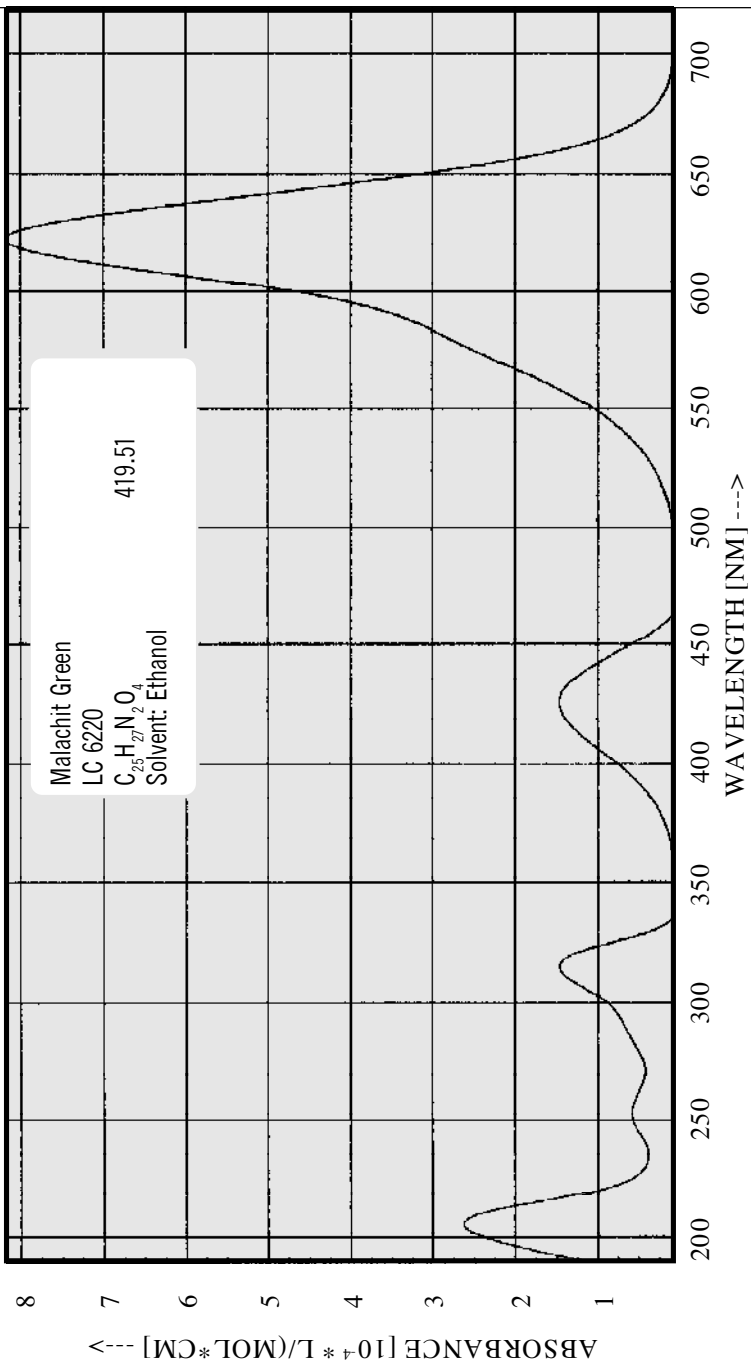
Efficient and frequently used laser dye for pulsed and CW operation; performance similar to Rhodamin B; tunable around 620 nm.

Pump		Dye Laser Characteristics					
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	Ref.
XeCl-Excimer	308	605	594 - 642	12	0.90	Methanol	1, 2
Nitrogen	337	622	600 - 646	rel.	2.85	Methanol	3
Nd:YAG, 2nd	532	588	579 - 600	29	0.27	Methanol	1
Cu-vapo	510	620	598 - 645	14	1.74	Methanol	4
Flashlamp	-	629	600 - 650	-	3.91	Methanol	5, 6
CW, Ar <sup>+</sup>	all	625	598 - 650	-	2.50	MeOH/Eg.	1, 7

## References

See page 164.

# Malachit Green



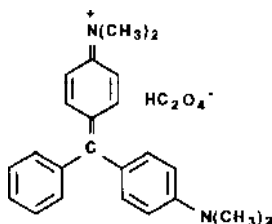
# Malachit Green (LC 6220)

## Constitution

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Malachit Grün Oxalat

$C_{25}H_{27}N_2O_4$  · MW: 419.51



## Characteristics

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Lambdachrome® number:	6220
CAS registry number:	-
Appearance:	green, crystalline solid
Absorption maximum (ethanol):	622 nm
Molar absorptivity:	$8.07 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in bas. ethanol):	530 nm
For research and development purposes only.	

## Lasing Performance

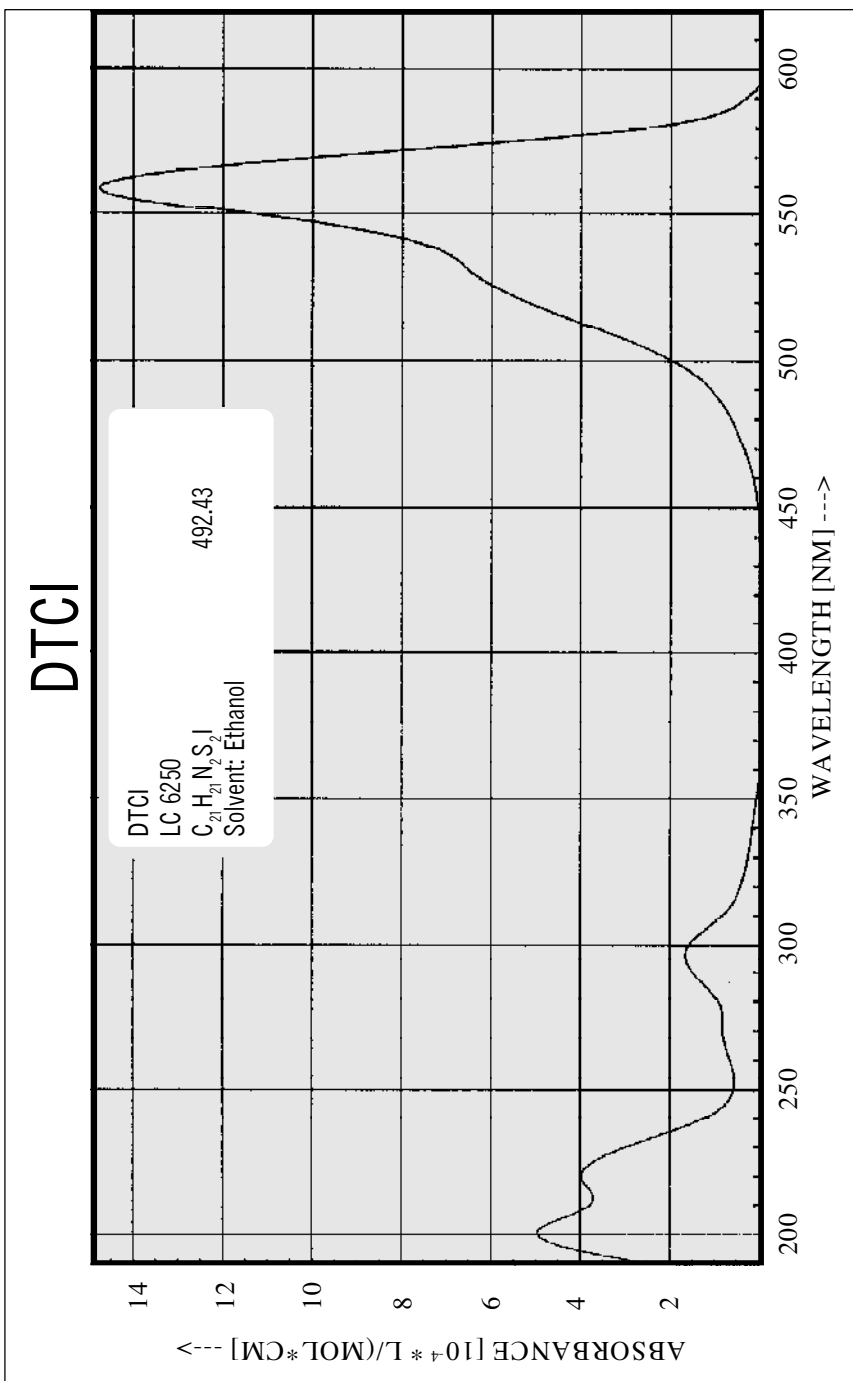
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Additive for CW pumped, passively mode locked Rhodamine 6G dye lasers.

## References

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1. M. Young, *Appl. Optics* **18**(19), 3212 (1979).
2. A. Watanabe et al., *IEEE J. Quantum Electron.* **QE-19**(4), 533 (1983).

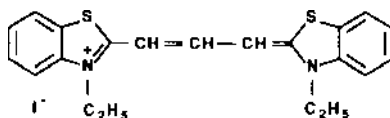


# DTCI (LC 6250)

## Constitution

3,3'-Diethylthiacarbocyanine Iodide

$C_{21}H_{21}N_2S_2I$  · MW: 492.43



## Characteristics

Lambdachrome® number:	6250
CAS registry number:	-
Appearance:	red, crystalline solid
Absorption maximum (in methanol):	557 nm
Molar absorptivity:	$14.60 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

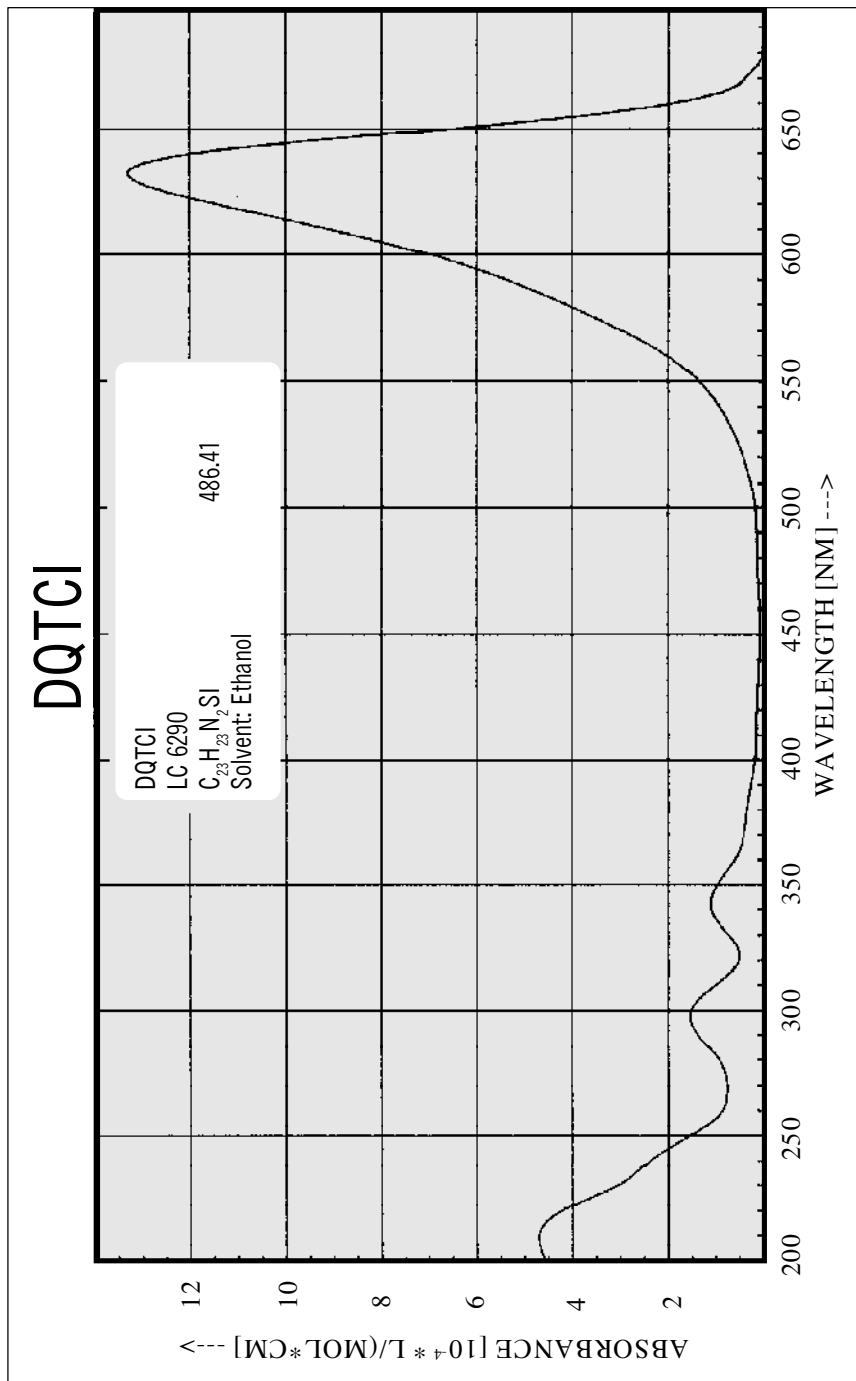
## Lasing Performance

Laser dye for pulsed operation; tunable around 625 nm. Saturable absorber for flashlamp pumped Rhodamine 6G dye lasers; applicable around 560 nm<sup>2,3</sup>).

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
Flashlamp	-		625	-	-	0.10	Glyzerine	1

## References

1. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **11**(5), 692 (1972).
2. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **13**(1), 193 (1974).



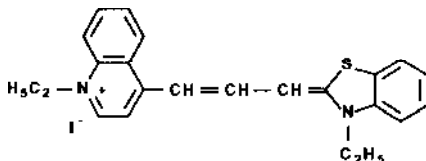
# DQTCI (LC 6290)

## Constitution

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1,3'-Diethyl-4,2'-quinolylthiacarbocyanine Iodide

$C_{23}H_{23}N_2SI$  · MW: 486.41



## Characteristics

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Lambdachrome® number:	6290
CAS registry number:	-
Appearance:	green, crystalline solid
Absorption maximum (in methanol):	629 nm
Molar absorptivity:	$13.1 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

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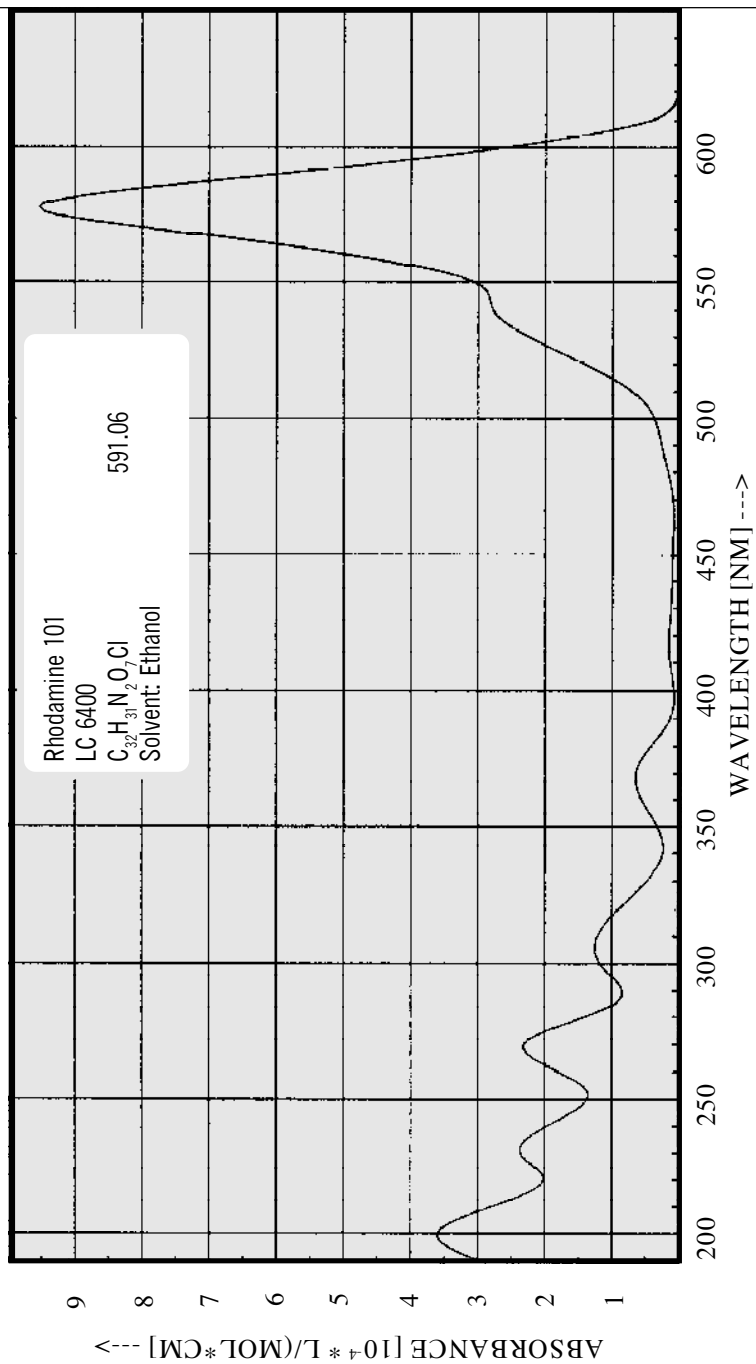
Saturable absorber for flashlamp pumped Rhodamine B dye lasers; applicable around 630 nm <sup>1.,2.)</sup>.

## References

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1. E. G. Arthurs et al., *Appl. Phys. Lett.* **20**(3), 125(1972).
2. E. Lill, S. Schneider, F. Dörr, *Opt. Commun.* **22**(1), 107(1977).

# Rhodamine 101



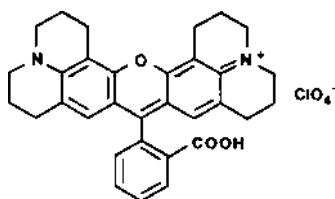


# Rhodamine 101 (LC 6400)

## Constitution

8-(2-Carboxyphenyl)-2,3,5,6,11,12,14,15-octahydro-1*H*,4*H*,10*H*,13*H*-diquinolizino[9,9*a*,1-*bc*:9',9*a'*,1-*h'*]xanthylium Perchlorate  
Rhodamine 640

C<sub>32</sub>H<sub>31</sub>N<sub>2</sub>O<sub>7</sub>Cl · MW: 591.06



## Characteristics

Lambdachrome® number:	6400
CAS registry number:	64339-18-0
Appearance:	red, crystalline solid
Absorption maximum (in acidic ethanol):	576 nm
Molar absorptivity:	$9.50 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

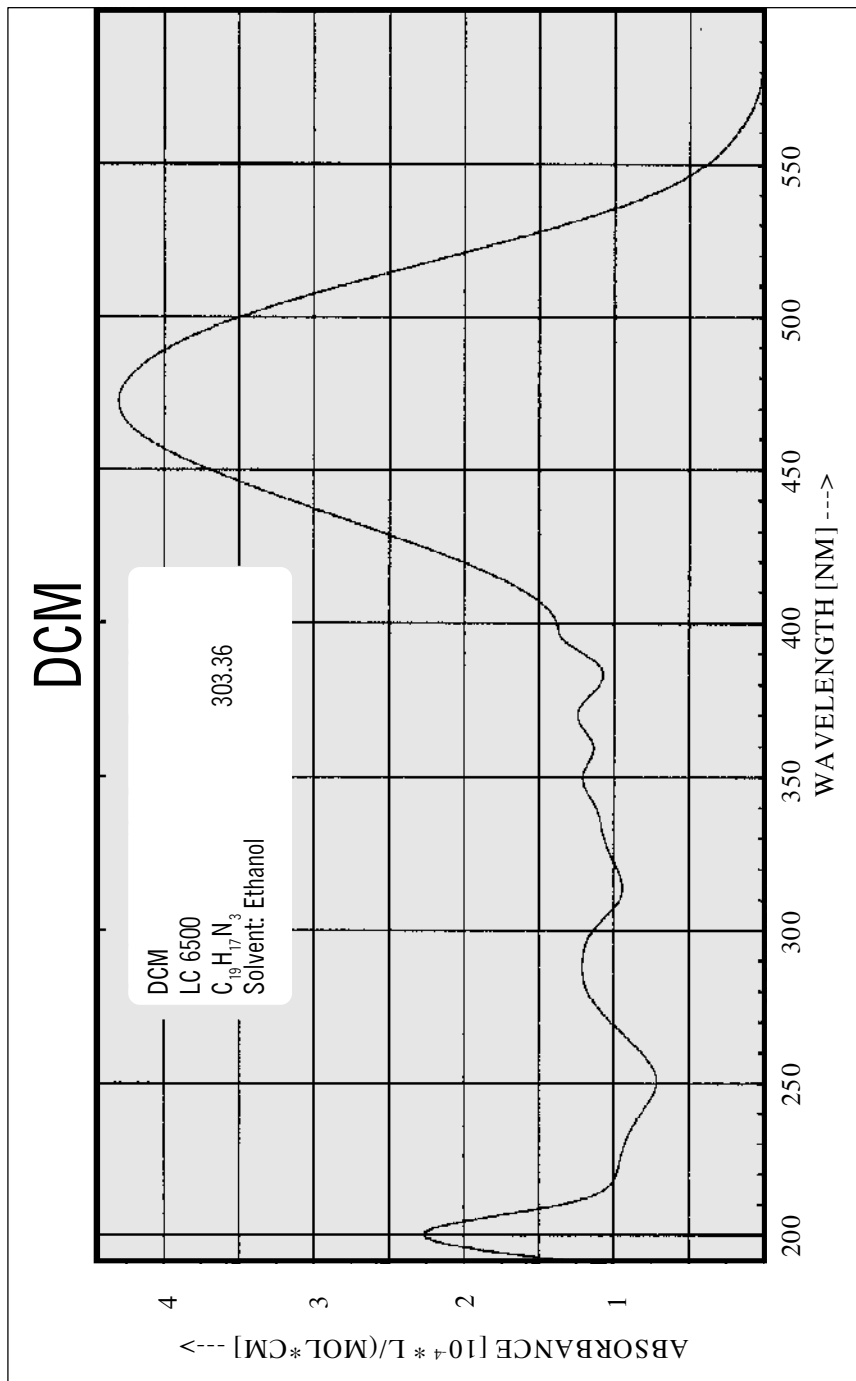
## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 640 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	623	614 - 672	12	0.75	Methanol	1, 2, 3
Nitrogen	337	648	623 - 676	rel.	2.36	Methanol	2, 4
Nd:YAG, 2nd	532	621	611 - 662	26	0.50	Methanol	1, 5, 6
Cu-vapor	510	630	607 - 659	14	-	Methanol	7
Flashlamp	-	-	-	-	-	Ethanol	8
CW, Ar <sup>+</sup>	VIS	625	610 - 695	-	2.5	MeOH/Eg.	1, 9

## References

See page 165.

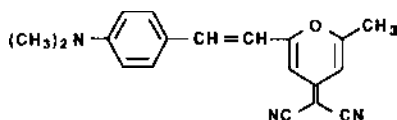


# DCM (LC 6500)

## Constitution

4-Dicyanmethyle-2-methyl-6-(p-dimethylaminostyryl)-4H-pyran

$C_{19}H_{17}N_3$  · MW: 303.36



## Characteristics

Lambdachrome® number:	6500
CAS registry number:	51325-91-8
Appearance:	red, crystalline solid
Absorption maximum (in ethanol):	472 nm
Molar absorptivity:	$4.25 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	644 nm
For research and development purposes only.	

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 650 nm. DCM Special gives higher efficiency due to better solubility.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	658	632 - 690	12	0.71	DMSO	1, 2
Nitrogen	337	659	626 - 703	rel.	0.50	DMSO	3
Nd:YAG, 2nd	532	639	615 - 666	27	0.50	PC	1, 4
Cu-vapor	510	644	598 - 677	14	0.61	Methanol	5
Flashlamp	-	655	610 - 710	-	0.76	DMSO	6, 7
CW, Ar <sup>+</sup>	VIS	660	-	-	0.45	Bz./Eg.	1, 8, 9, 10

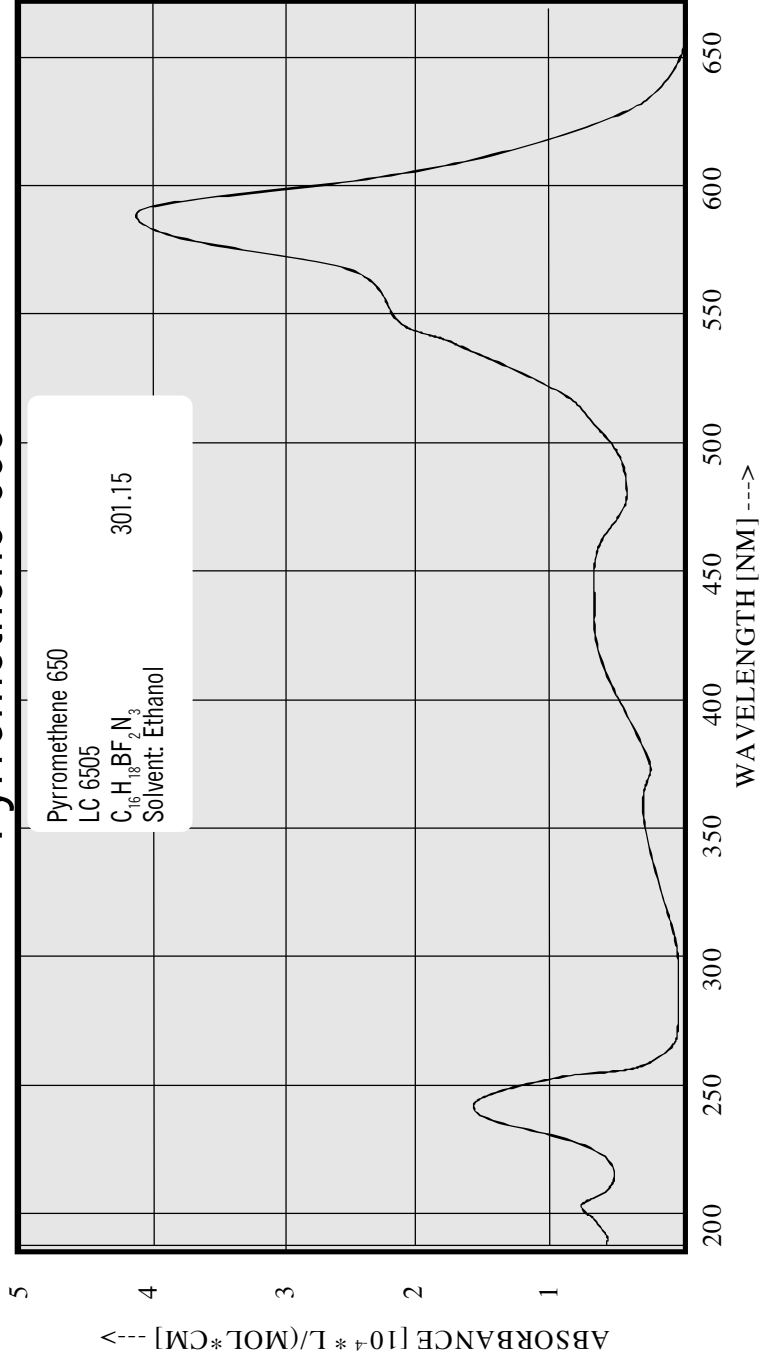
## References

See page 165.

## P.S.

LC 6501 is a 1:1 mixture of LC 6500 and LC 6200 giving high absorption at the green line of an Ar-Ion laser. Its application should be restricted to this pump laser only.

# Pyrromethene 650

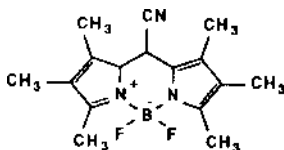


# Pyrromethene 650 (LC 6505)

## Constitution

4,4-Difluoro-8-cyano-1,2,3,5,6,7-hexamethyl-4-bora-3a,4a-diaza-s-indacene  
8-Cyano-1,2,3,5,6,7-hexamethylpyrromethenedifluoroborate Complex

$C_{16}H_{18}BF_2N_3$  · MW: 301.15



## Characteristics

Lambdachrome® number:	6505
CAS registry number:	-
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	590 nm
Molar absorptivity:	$4.05 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	612 nm
For research and development purposes only.	

## Lasing Performance

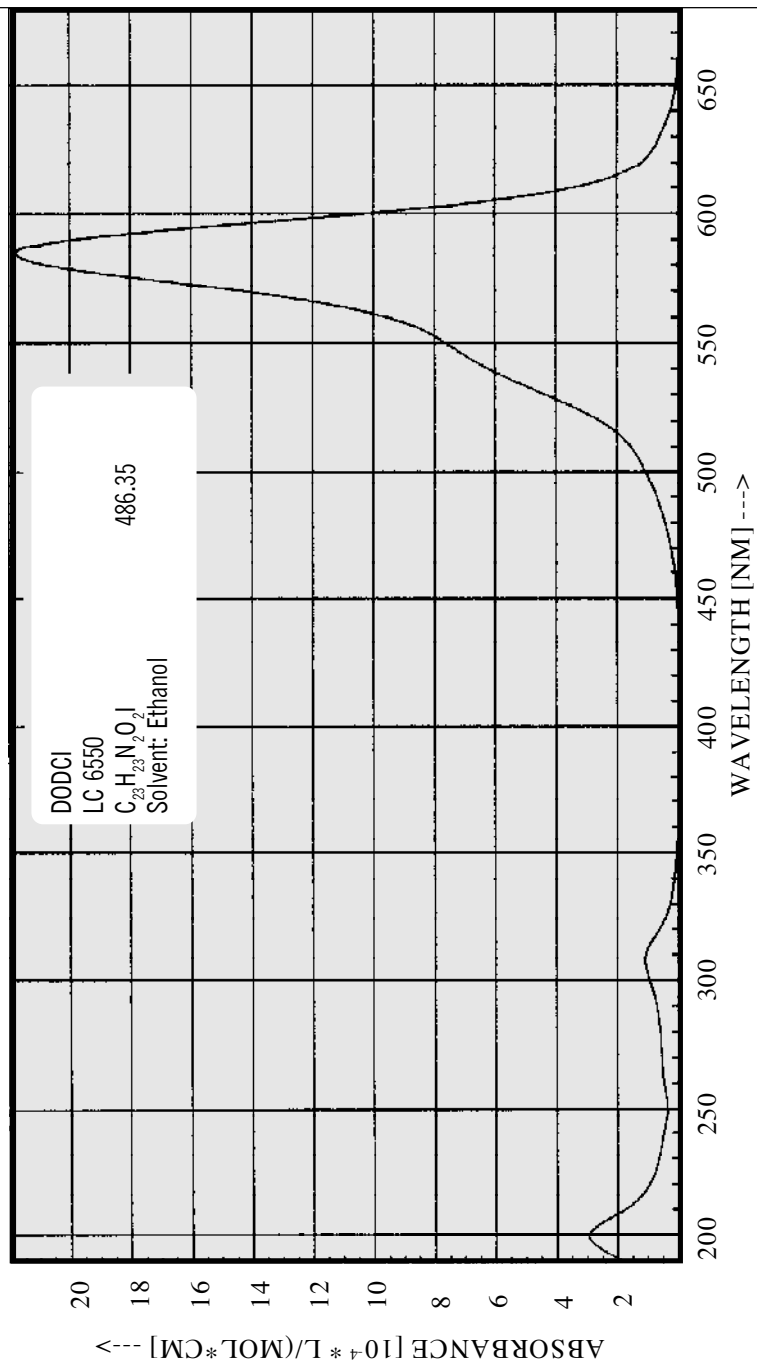
Laser dye for pulsed operation; tunable around 630 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nd:YAG	532	630	-	31	0.03	Ethanol	1

## References

1. T. H. Allik et al., *SPIE Proceedings* 2115, 240 (1994).

# DODCI

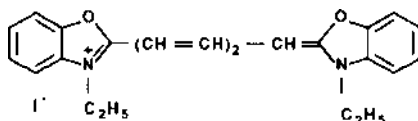


# DODCI (LC 6550)

## Constitution

3,3'-Diethyloxadicyanine Iodide  
NK 1533

$C_{23}H_{23}N_2O_2I$  · MW: 486.35



## Characteristics

Lambdachrome® number:	6550
CAS registry number:	-
Appearance:	blue, crystalline solid
Absorption maximum (in ethanol):	582 nm
Molar absorptivity:	$22.3 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

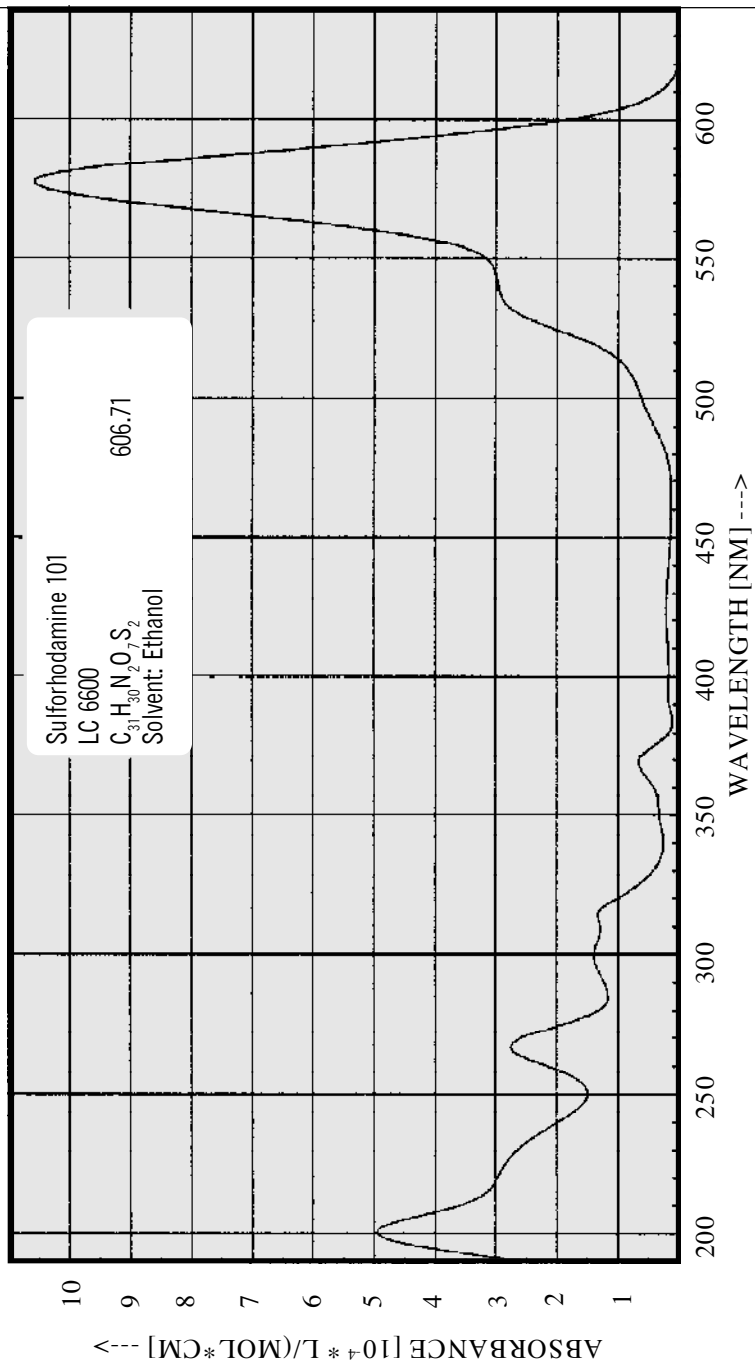
Laser dye for pulsed operation; tunable around 660 nm. Most frequently used saturable absorber for flashlamp and CW pumped Rhodamine 6G und Rhodamin B dye lasers <sup>3., 4., 5.)</sup>. Applicable around 580 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nd:YAG, 2nd	532	-	-	-	0.49	Ethanol	1
Flashlamp	-	662	-	-	0.10	DMSO	2

## References

1. C. Rulliere, *Chem. Phys. Letters* **43**(2), 303 (1976).
2. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **11**(5), 692 (1972).
3. M. Young, *Appl. Optics*. **18**(19), 3212 (1979).
4. A. Watanabe et al., *IEEE J. Quantum Electron.* **QE-19**(4), 533 (1983).
5. E. G. Arthurs et al., *Appl. Phys. Letters* **20**(3), 125 (1972).

# Sulforhodamine 101



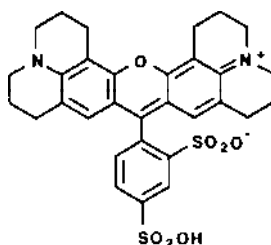


# Sulforhodamine 101 (LC 6600)

## Constitution

8-(2, 4-Disulfophenyl)-2,3,5,6,11,12,14,15-octahydro-1*H*,4*H*,10*H*,13*H*-diquinolizino[9,9*a*,1-*bc*:9',9*a'*,1-*hi*]xanthene  
Sulforhodamine 640

$C_{31}H_{30}N_2O_7S_2$  · MW: 606.71



## Characteristics

Lambdachrome® number: 6600  
Appearance: green, crystalline solid  
Absorption maximum (in ethanol): 578 nm  
Molar absorptivity:  $10.6 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
For research and development purposes only.

## Lasing Performance

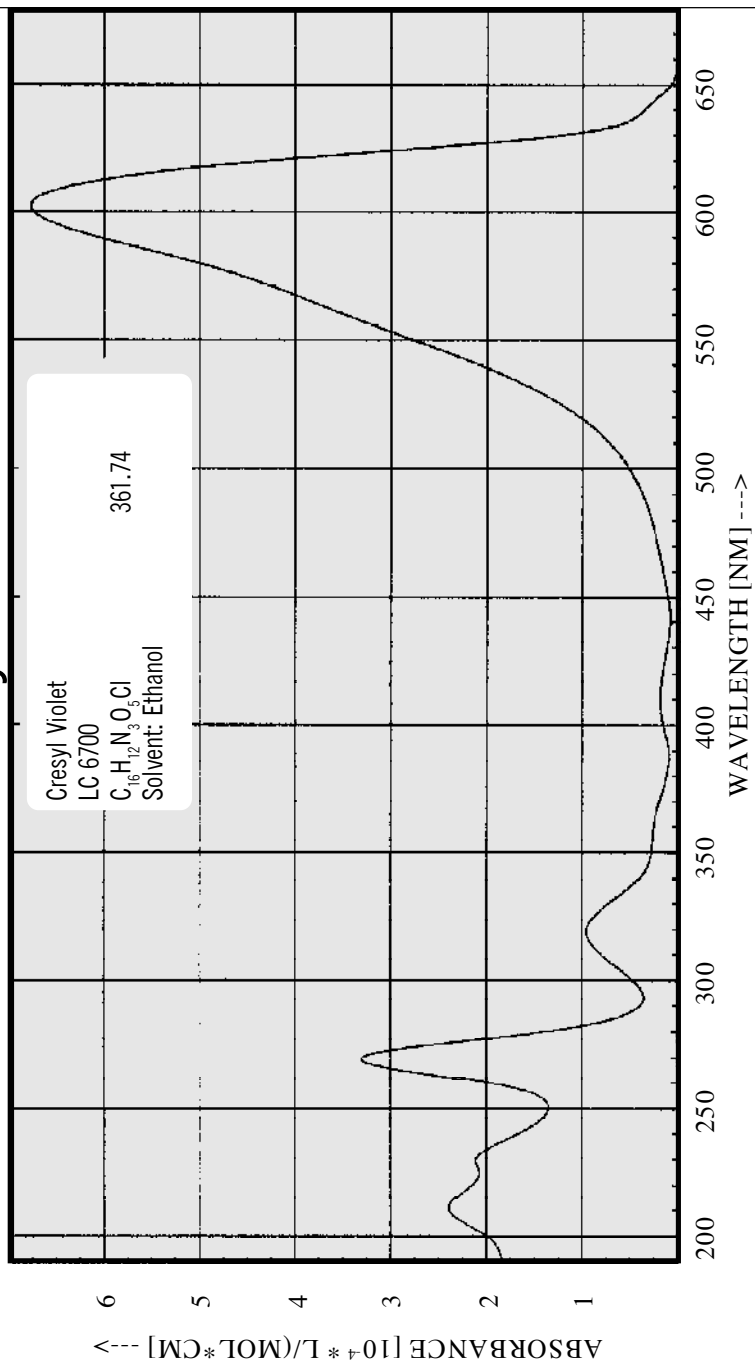
Efficient laser dye for pulsed and CW operation; tunable around 650 nm.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	652	616-667	15	0.78	Methanol	2, 3
Nd:YAG, 2nd	532	628	619-673	15	0.3	Methanol	1, 4
CW, Ar <sup>+</sup>	VIS	625	598-650	-	2.5	Eg.	5, 6

## References

1. Lambda Physik, *Wall Chart* 1996.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).
3. Lambda Physik.
4. K. Kato, *IEEE J. Quantum Electron.* **QE-13**(7), 544 (1977).
5. Lambda Physik, *Wall Chart* 6/90.
6. M. Yamashita et al., *Opt. Commun.* **26**(3), 343 (1978).

# Cresyl Violet



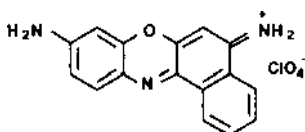
# Cresyl Violet (LC 6700)

## Constitution

5,9-Diaminobenzo[a]phenoxazonium Perchlorate

Cresyl Violet 670

$C_{16}H_{12}N_3O_5Cl$  · MW: 361.74



## Characteristics

Lambdachrome® number:	6700
CAS registry number:	41830-80-2
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	601 nm
Molar absorptivity:	$6.74 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	632 nm
For research and development purposes only.	

## Lasing Performance

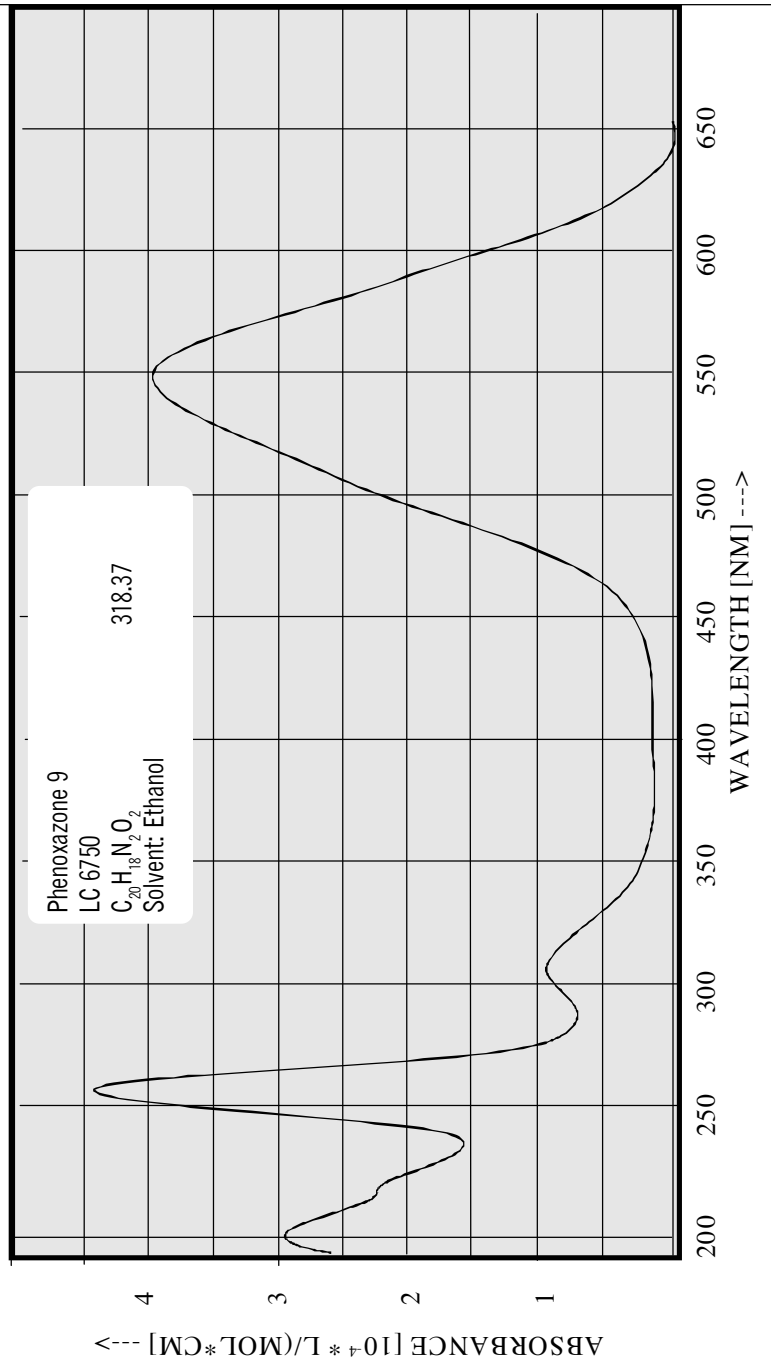
Efficient laser dye for pulsed and CW operation; tunable around 650 nm.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	656	643 - 688	5	0.50	Ethanol	1, 2, 3
Nitrogen	337	650	630 - 680	rel.	0.36	Ethanol	4, 5
Nd:YAG, 2nd	532	630	614 - 654	36	0.04	Methanol	6, 7
Flashlamp	-	650	630 - 680	-	0.01	Methanol	8, 9
CW, Ar <sup>+</sup>	VIS	670	650 - 695	4	1.09	Eg.	10

## References

See page 165.

## Phenoxazone 9

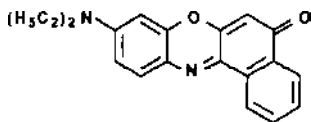


# Phenoxazone 9 (LC 6750)

## Constitution

9-Diethylamino-5H-benzo(a)phenoxazin-5-one

$C_{20}H_{18}N_2O_2$  · MW: 318.37



## Characteristics

Lambdachrome® number:	6750
CAS registry number:	7385-67-3
Appearance:	green, crystalline solid
Absorption maximum (ethanol):	550 nm
Molar absorptivity:	$2.83 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in bas. ethanol):	650nm
For research and development purposes only.	

## Lasing Performance

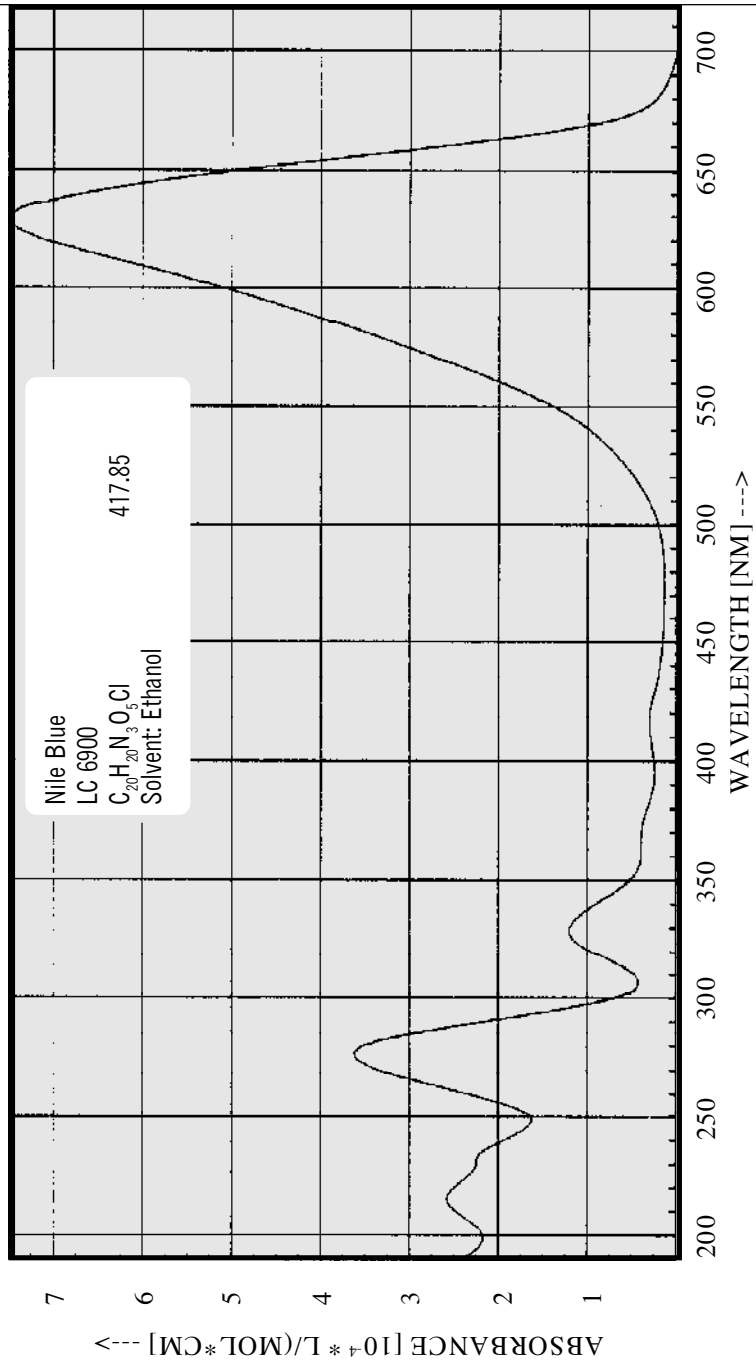
Laser dye for pulsed operation; tunable around 620 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nitrogen	337	620	560 - 700	rel.	0.32	Ethanol	1
Flashlamp	-	-	-	-	0.03	various	1

## References

1. D. Basting, D. Ouw, F. P. Schäfer, *Opt. Commun.* **18**(3), 260 (1976).

# Nile Blue

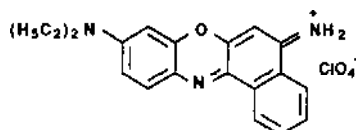


# Nile Blue (LC 6900)

## Constitution

5-Amino-9-diethyliminobenzo[a]phenoxazonium Perchlorate

$C_{20}H_{20}N_3O_5Cl$  · MW: 417.85



## Characteristics

Lambdachrome® number:	6900
CAS registry number:	53340-16-2
Appearance:	green, crystalline solid
Absorption maximum (ethanol):	633 nm
Molar absorptivity:	$7.75 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in bas. ethanol):	672 nm
For research and development purposes only.	

## Lasing Performance

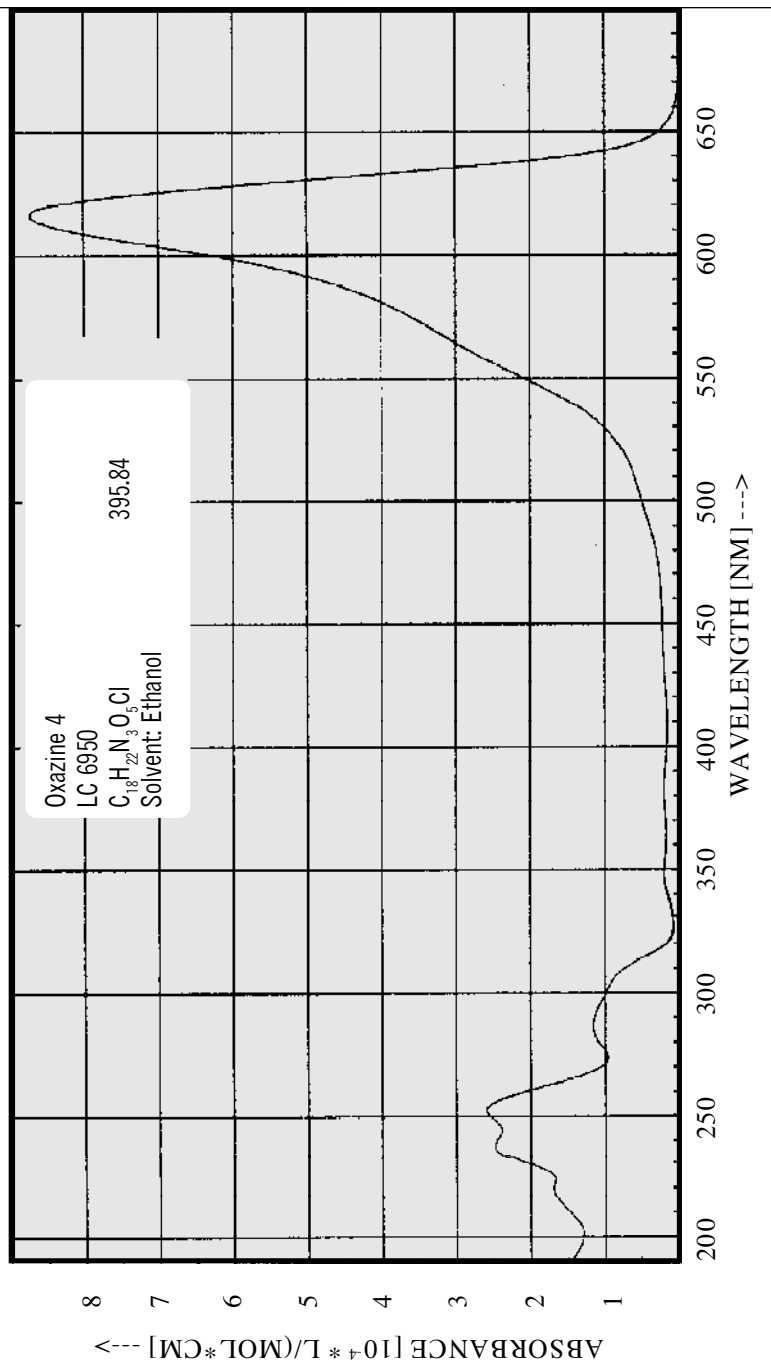
Laser dye for pulsed operation; tunable around 690 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	703	688 - 747	5	0.47	Methanol	1, 2
Nitrogen	337	695	683 - 751	rel.	0.45	Methanol	2
Nd:YAG, 2nd	532	683	-	18	0.08	Methanol	3
Cu-vapor	510	695	682 - 730	4	0.84	Methanol	4
Flashlamp	-	710	690 - 750	poor	0.31	Methanol	5
CW, Kr <sup>+</sup>	red	730	690 - 780	7	1.2	Eg.	6

## References

1. Lambda Physik.
2. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
3. K. Kato, *Opt. Commun.* **19**(1), 19 (1976).
4. M. Broyer et al., *Appl. Phys.* **B35**, 31 (1984).
5. J. B. Marling et al., *Appl. Optics* **13**(10), 2317 (1974).
6. J. M. Yarborough, *Appl. Phys. Letters* **24**(12), 629 (1974).

# Oxazine 4



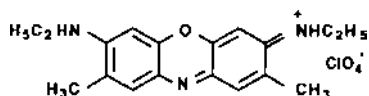


# Oxazine 4 (LC 6950)

## Constitution

3-Ethylamino-7-ethylimino-2,8-dimethylphenoxazin-5-ium Perchlorate  
LD 690

$C_{18}H_{22}N_3O_5Cl$  · MW: 395.84



## Characteristics

Lambdachrome® number:	6950
CAS registry number:	-
Appearance:	green, crystalline solid
Absorption maximum (ethanol):	615 nm
Molar absorptivity:	$10.9 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

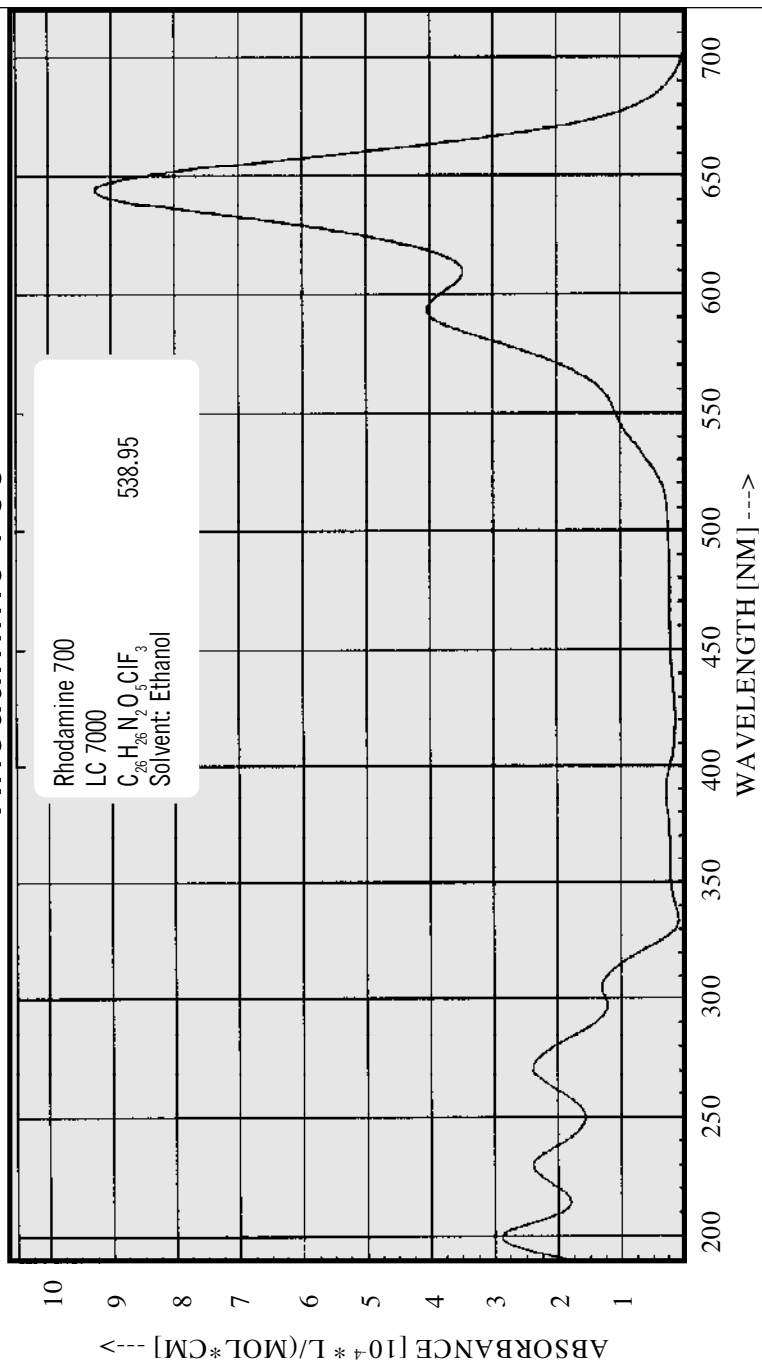
Laser dye for pulsed operation; tunable around 690 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	703	665 - 718	5	0.53	DMSO	1
Nd:YAG, 2nd	532	660	-	-	-	-	2
CW, Ar <sup>+</sup>	all	-	-	-	+ DCM	-	3

## References

1. Lambda Physik.
2. R. J. Hall et al., *Opt. Letters* **4**(3), 87 (1979).
3. J. Heber, A. Szabo, *IEEE J. Quantum Electron.* **QE-20**(1), 9 (1984).

# Rhodamine 700

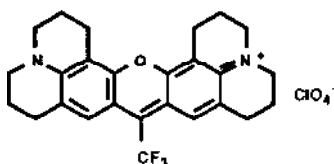


# Rhodamine 700 (LC 7000)

## Constitution

8-(Trifluoromethyl)-2,3,5,6,11,12,14,15-octahydro-1*H*,4*H*,10*H*,13*H*-diquinolizino[9,9*a*,1-*bc*:9',1-*hi*]xanthylum Perchlorate  
LD 700

$C_{26}H_{26}N_2O_5ClF_3$  · MW: 538.95



## Characteristics

Lambdachrome® number: 7000  
Appearance: brown, crystalline solid  
Absorption maximum (in ethanol): 643 nm  
Molar absorptivity:  $9.25 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
For research and development purposes only.

## Lasing Performance

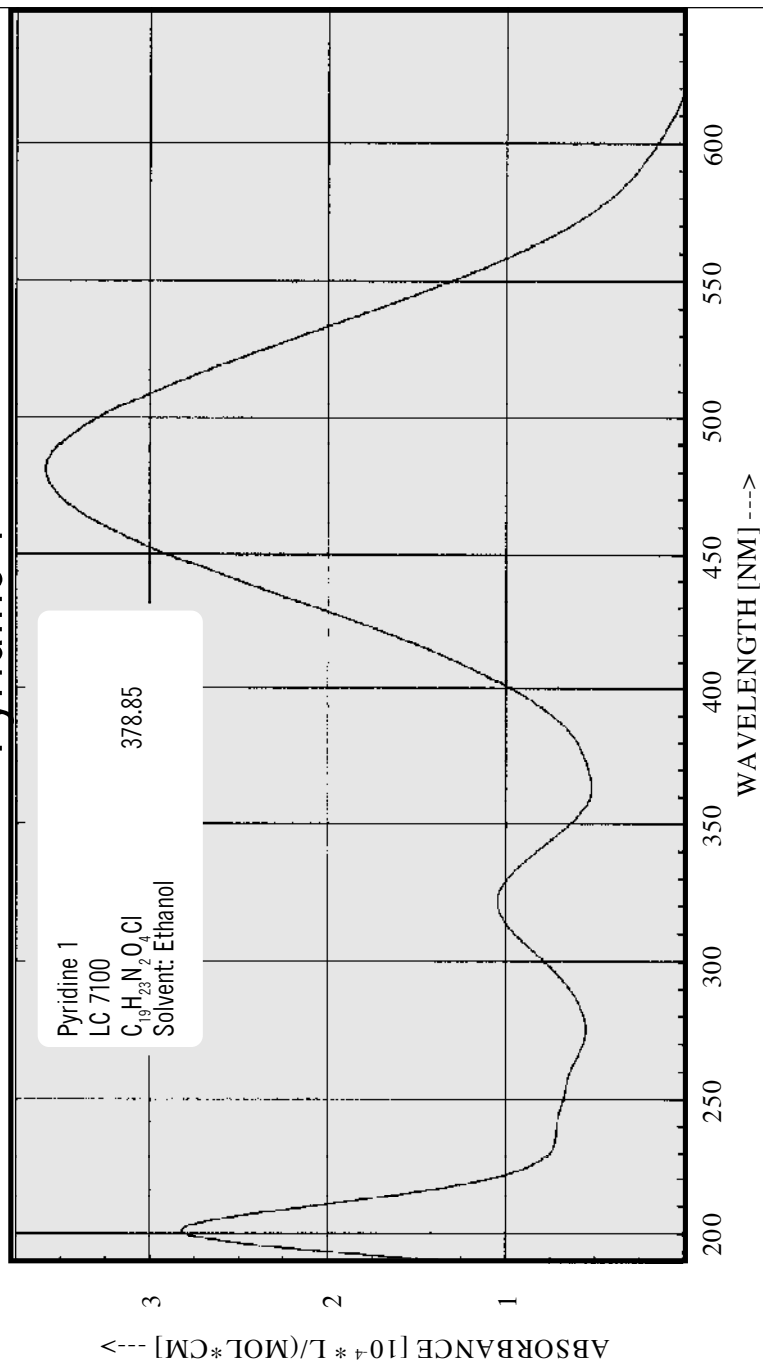
Very efficient laser dye for pulsed and CW operation; tunable around 750 nm

Pump		Dye Laser Characteristics				Solvent	Ref.
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	723	701 - 768	11	0.85	Methanol	1, 2
Flashlamp	-	-	705 - 798	-	0.11	Ethanol	3
CW, Kr <sup>+</sup>	red	740	690 - 785	-	1.0	Eg.	1, 4, 5, 6

## References

1. Lambda Physik, *Wall Chart* 1996.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).
3. W. Sibbett, J. R. Taylor, *IEEE J. Quantum Electron.* **QE-20**(2), 108 (1984).
4. T. F. Johnston, R. H. Brady, W. Proffitt, *Appl. Optics.* **21**(13), 2307 (1982).
5. G. D. Aumiller, *Appl. Optics* **23**(5), 651 (1984).
6. E. G. Marason, *Opt. Commun.* **40**(3), 212 (1982).

# Pyridine 1

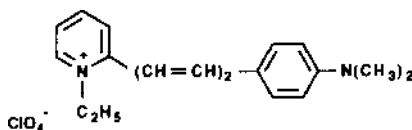


# Pyridine 1 (LC 7100)

## Constitution

1-Ethyl-2-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-pyridinium Perchlorat  
LDS 698

$C_{19}H_{23}N_2O_4Cl$  · MW: 378.85



## Characteristics

Lambdachrome® number: 7100  
Appearance: red, crystalline solid  
Absorption maximum (in ethanol): 480 nm  
Molar absorptivity:  $3.80 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
For research and development purposes only.

## Lasing Performance

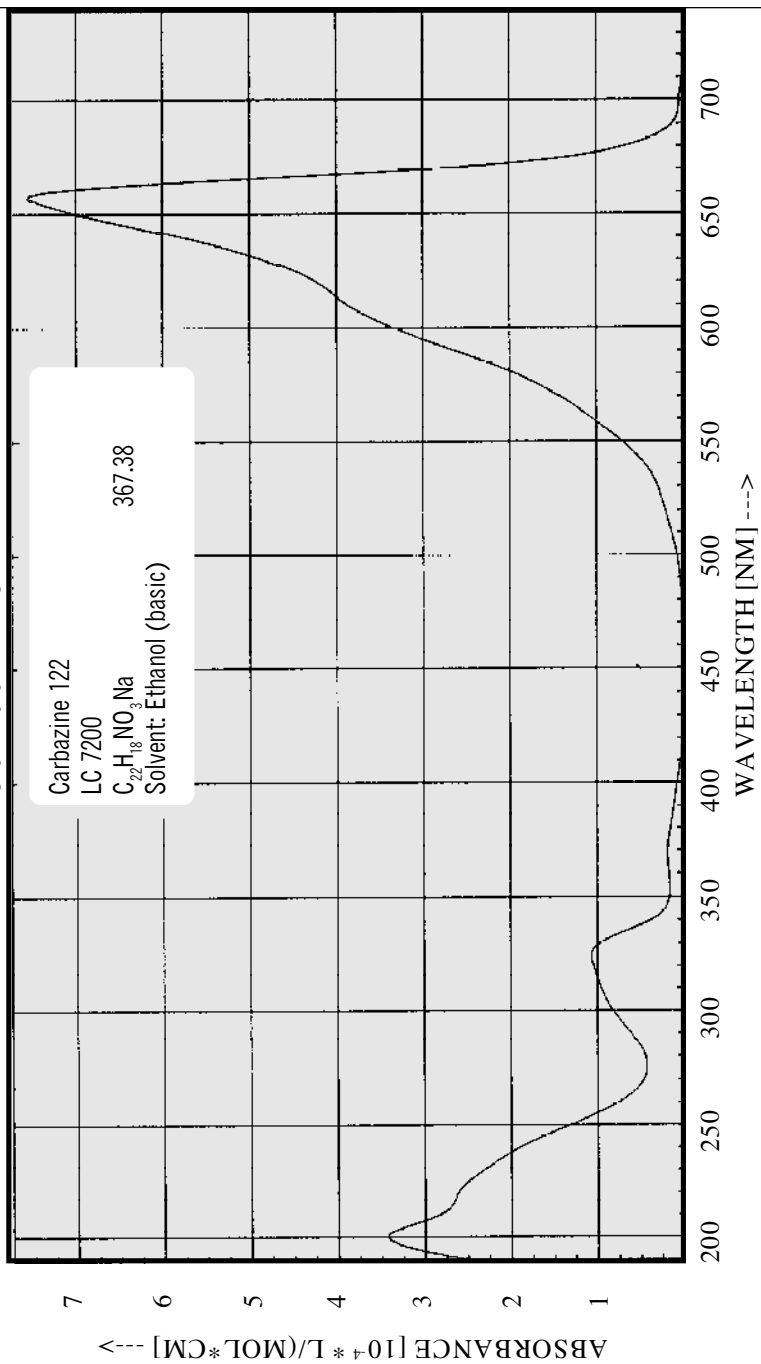
Efficient laser dye for pulsed and CW operation; tunable around 710 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	710	670 - 760	10	0.84	DMSO	1, 2
Nitrogen	337	703	675 - 750	rel.	0.88	DMSO	3
Nd:YAG, 2nd	532	697	667 - 736	32	0.36	PC	1, 4
Cu-vapor	510	684	661 - 724	6	1.17	Methanol	5
CW, Ar <sup>+</sup>	VIS	710	670 - 780	-	-	Pc./Eg.	6, 7

## References

1. Lambda Physik, *Wall Chart* 1996.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).
3. Lambda Physik, *Data Sheet*.
4. Lambda Physik.
5. M. Broyer et al., *Appl. Phys.* **B35**, 31 (1984).
6. J. Hoffnagle et al., *Opt. Commun.* **42(4)**, 267 (1982).
7. Ph. Bado et al., *Opt. Commun.* **46(3,4)**, 241 (1983).

# Carbazine 122

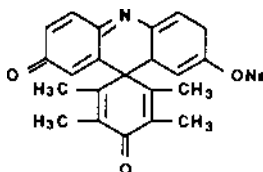


# Carbazine 122 (LC 7200)

## Constitution

Carbazine 720

$C_{22}H_{18}NO_3Na$  · MW: 367.38



## Characteristics

Lambdachrome® number:	7200
CAS registry number:	-
Appearance:	red, crystalline solid
Absorption maximum (in basic ethanol):	655 nm
Molar absorptivity:	$7.54 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

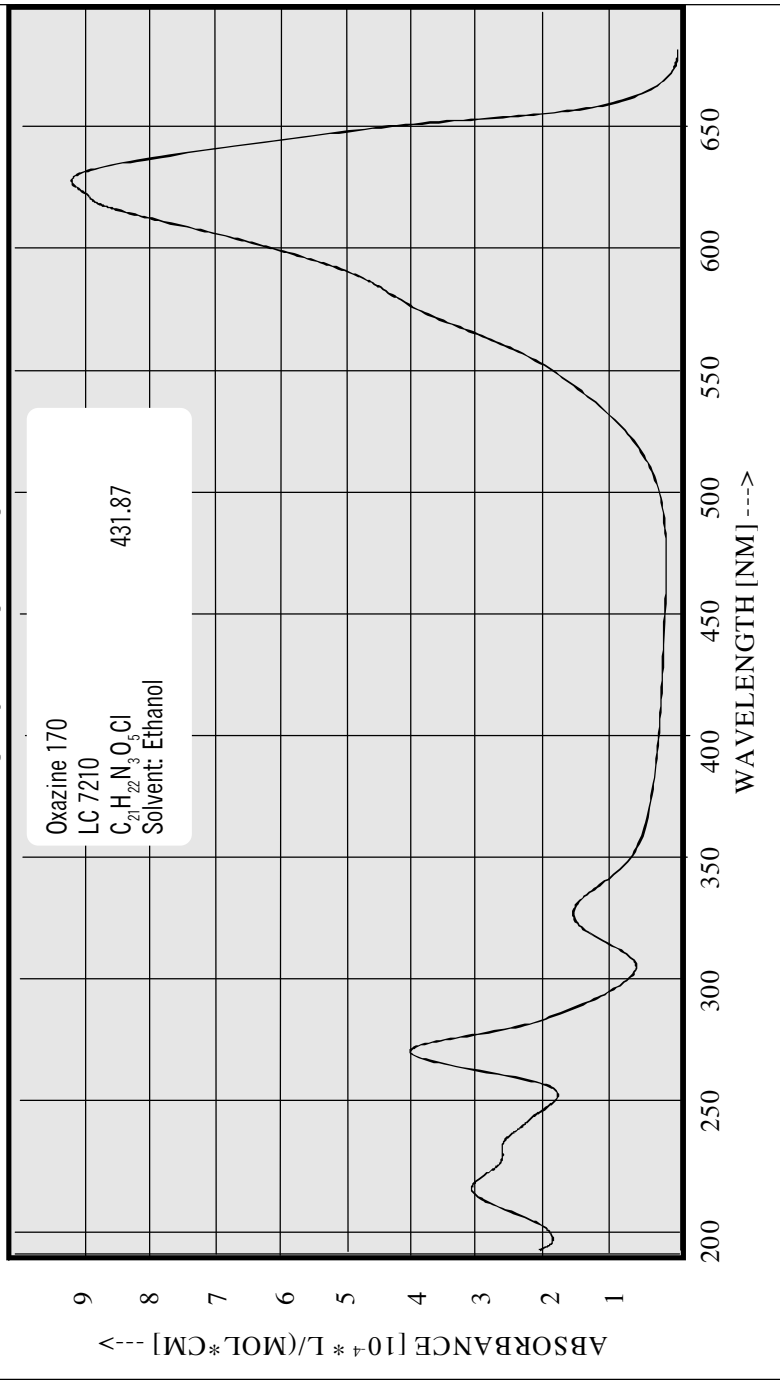
Efficient laser dye for pulsed and CW operation; tunable around 640 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XNd:YAG, 2nd	532	720	-	30	1.10	Water	1, 2, 3
Flashlamp	-	700	680 - 740	-	0.07	Methanol	4
CW, Kr <sup>+</sup>	red	750	690 - 820	-	0.72	Eg.	5

## References

1. F. Bos, *Appl. Optics* **20**(10), 1886 (1981).
2. K. Kato, *Opt. Commun.* **19**(1), 18 (1976).
3. K. Kato, *Opt. Commun.* **18**(4), 447 (1976).
4. J. B. Marling et al., *Appl. Optics* **13**(10), 2317 (1974).
5. P. E. Jessop, A. Szabo, *IEEE J. Quantum Electr.* **QE-16**(8), 812 (1980).

# Oxazine 170



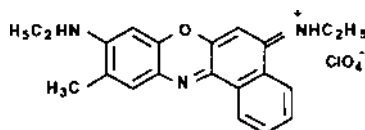


# Oxazine 170 (LC 7210)

## Constitution

9-Ethylamino-5-ethylimino-10-methyl-5H-benzo(a)phenoxazonium Perchlorate  
Oxazine 720

$C_{21}H_{22}N_3O_5Cl$  · MW: 431.87



## Characteristics

Lambdachrome® number: 7210  
CAS registry number: 62669-60-7  
Appearance: green, crystalline solid  
Absorption maximum (in ethanol): 627 nm  
Molar absorptivity:  $9.20 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in ethanol): 650 nm  
For research and development purposes only.

## Lasing Performance

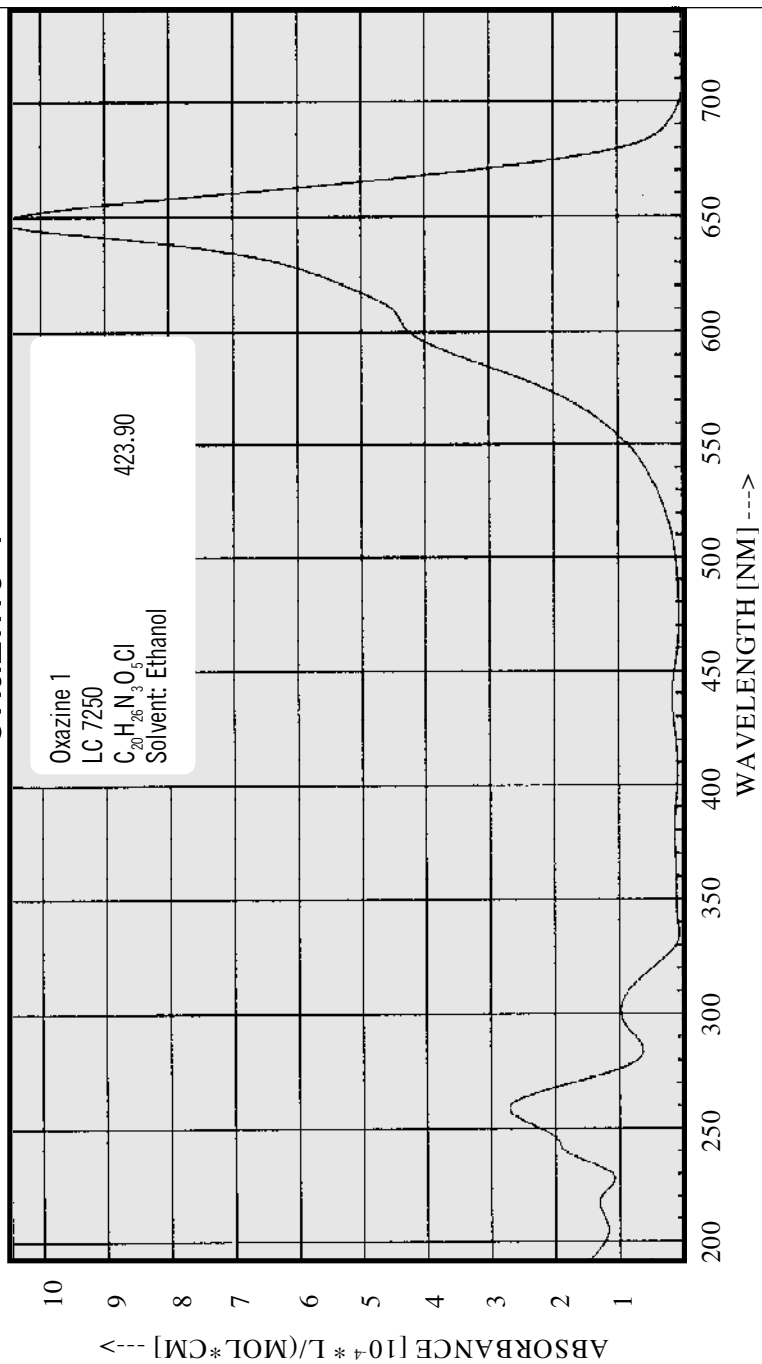
Efficient laser dye for pulsed and CW operation; tunable around 670 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	708	660 - 728	4	1.14	Methanol	1, 2, 3
Nitrogen	337	705	672 - 727	rel.	0.79	Methanol	4
Nd:YAG, 2nd	532	672	-	20	0.08	Methanol	5, 6
Cu-vapor	510	675	660 - 712	12	0.13	Methanol	7
CW, Kr <sup>+</sup>	red	730	670 - 740	-	0.86	Eg.	8

## References

1. Lambda Physik, *Wall Chart* 6/83.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).
3. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. K. Kato, *Opt. Commun.* **19**(1), 18 (1976).
6. C. A. Moore, C. D. Decker, *J. Appl. Phys.* **49**(1), 47 (1978).
7. M. Broyer et al., *Appl. Phys.* **B35**, 31 (1984).
8. P. E. Jessop, A. Szabo, *IEEE J. Quantum Electr.* **QE-16**(8), 812 (1980).

# Oxazine 1



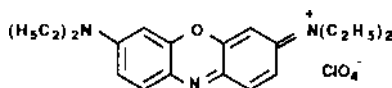
# Oxazine 1 (LC 7250)

## Constitution

3-Diethylamino-7-diethyliminophenoxazonium Perchlorate

Oxazine 725

$C_{20}H_{26}N_3O_5Cl$  · MW: 423.90



## Characteristics

Lambdachrome® number:	7250
CAS registry number:	24796-94-9
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	646 nm
Molar absorptivity:	$13.0 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	670 nm
For research and development purposes only.	

## Lasing Performance

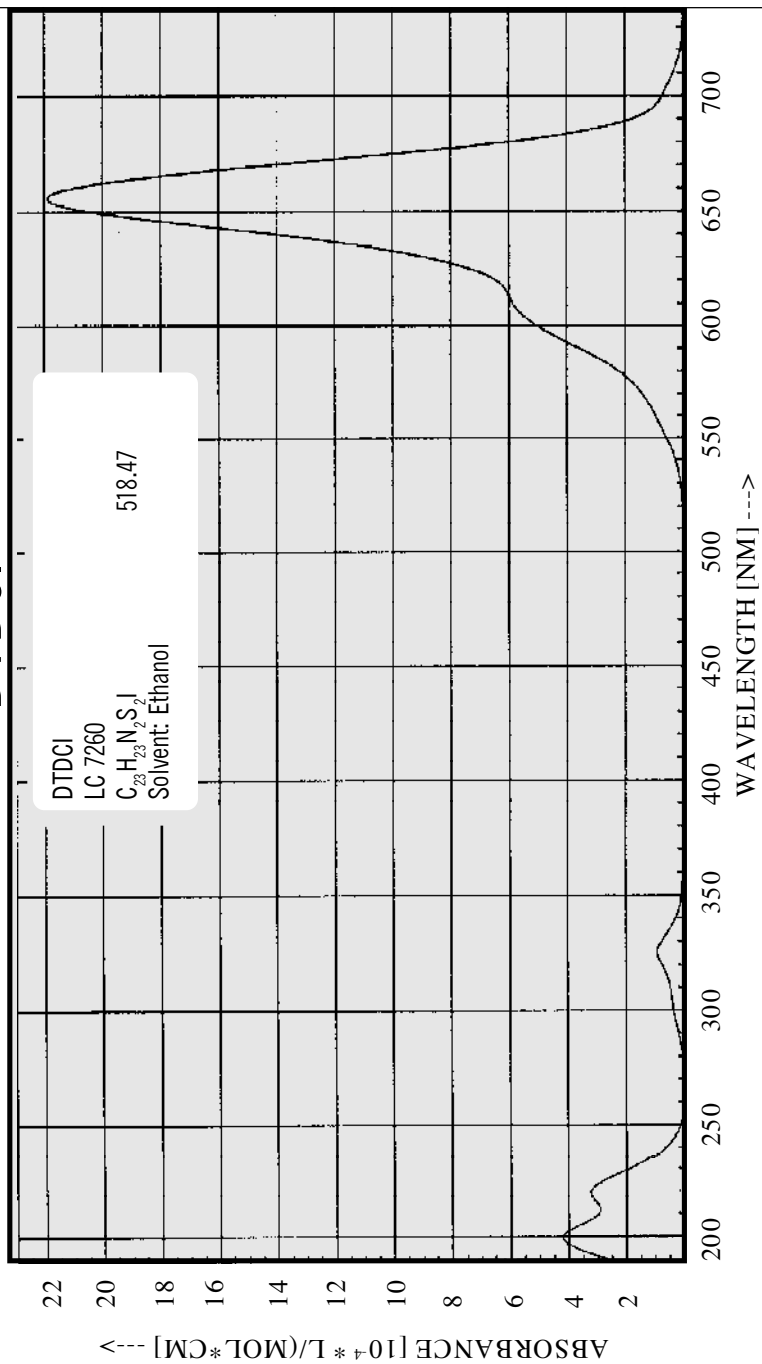
Efficient laser dye for pulsed and CW operation; tunable around 670 nm.

Source	Pump		Dye Laser Characteristics			Solvent	Ref.
	Wavelength	Peak	Range	Effic.	Conc.		
	[nm]	[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308	734	692 - 768	6	0.85	Ethanol	1, 2, 3
Nitrogen	337	730	692 - 751	-	1.96	Ethanol	4
Nd:YAG, 2nd	532	695	-	18	0.07	Methanol	5, 6
Flashlamp	-	720	700 - 740	-	-	Methanol	7
CW, Kr <sup>+</sup>	red	720	695 - 800	-	1.20	Eg.	8

## References

1. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 403 (1981).
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).
3. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
4. B. M. Pierce, R. R. Birge, *IEEE J. Quantum Electr.* **QE-18**(7), 1164 (1982).
5. F. Bos, *Appl. Optics* **20**(10), 1886 (1981).
6. C. A. Moore, C. D. Decker, *J. Appl. Phys.* **49**(1), 47 (1978).
7. J. B. Marling et al., *Appl. Optics* **13**(10), 2317 (1974).
8. Coherent, *CW Dye Laser Fact Sheets*.

# DTDCI

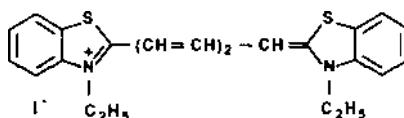


# DTDCI (LC 7260)

## Constitution

3-Diethylthiadicyanocyanine Iodide  
NK 136

$C_{23}H_{23}N_2S_2I \cdot MW: 518.47$



## Characteristics

Lambdachrome® number:	7260
CAS registry number:	-
Appearance:	blue, crystalline solid
Absorption maximum (in ethanol):	653 nm
Molar absorptivity:	$22.3 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

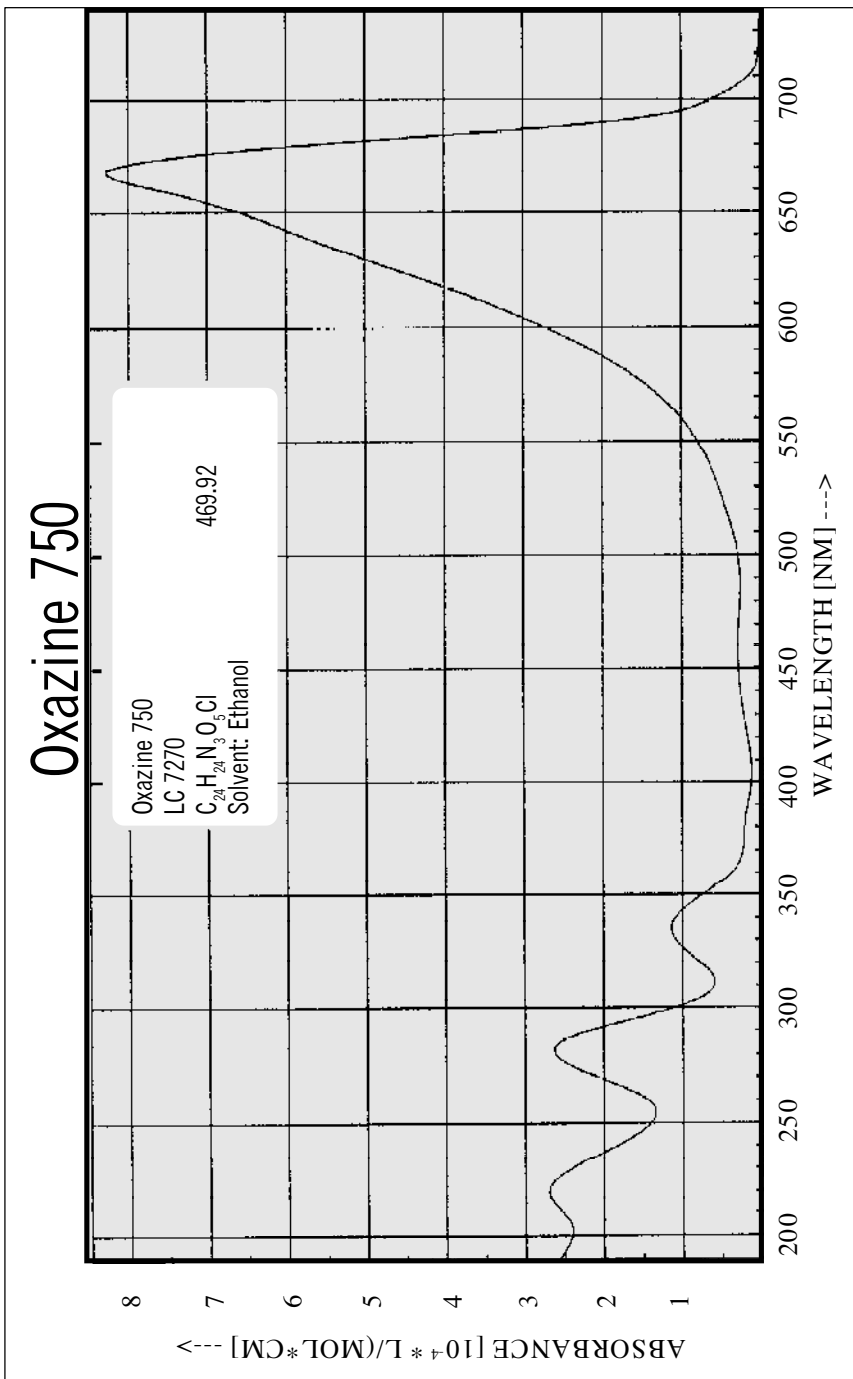
## Lasing Performance

Laser dye for pulsed operation; tunable around 760 nm. Saturable absorber for flashlamp pumped Rhodamine 101 dye lasers; applicable around 650 nm <sup>1-3</sup>.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nitrogen	337	695	-	-	0.13	Acetone	2
Flashlamp	-	760	-	-	0.10	DMSO	3

## References

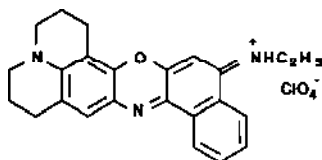
1. T. J. Negran, A. M. Glass, *Appl. Optics* **17**(17), 2812 (1978).
2. Chinlon Lin, *IEEE J. Quantum Electr.* **QE-11**, 61 (1975).
3. A. Hirth, K. Vollrath, J. Faure, D. Loughnot, *Opt. Commun.* **7**(4), 339 (1973).



# Oxazine 750 (LC 7270)

## Constitution

C<sub>24</sub>H<sub>24</sub>N<sub>3</sub>O<sub>5</sub>Cl · MW: 469.92



## Characteristics

Lambdachrome® number:	7270
CAS registry number:	-
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	667 nm
Molar absorptivity:	8.25 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

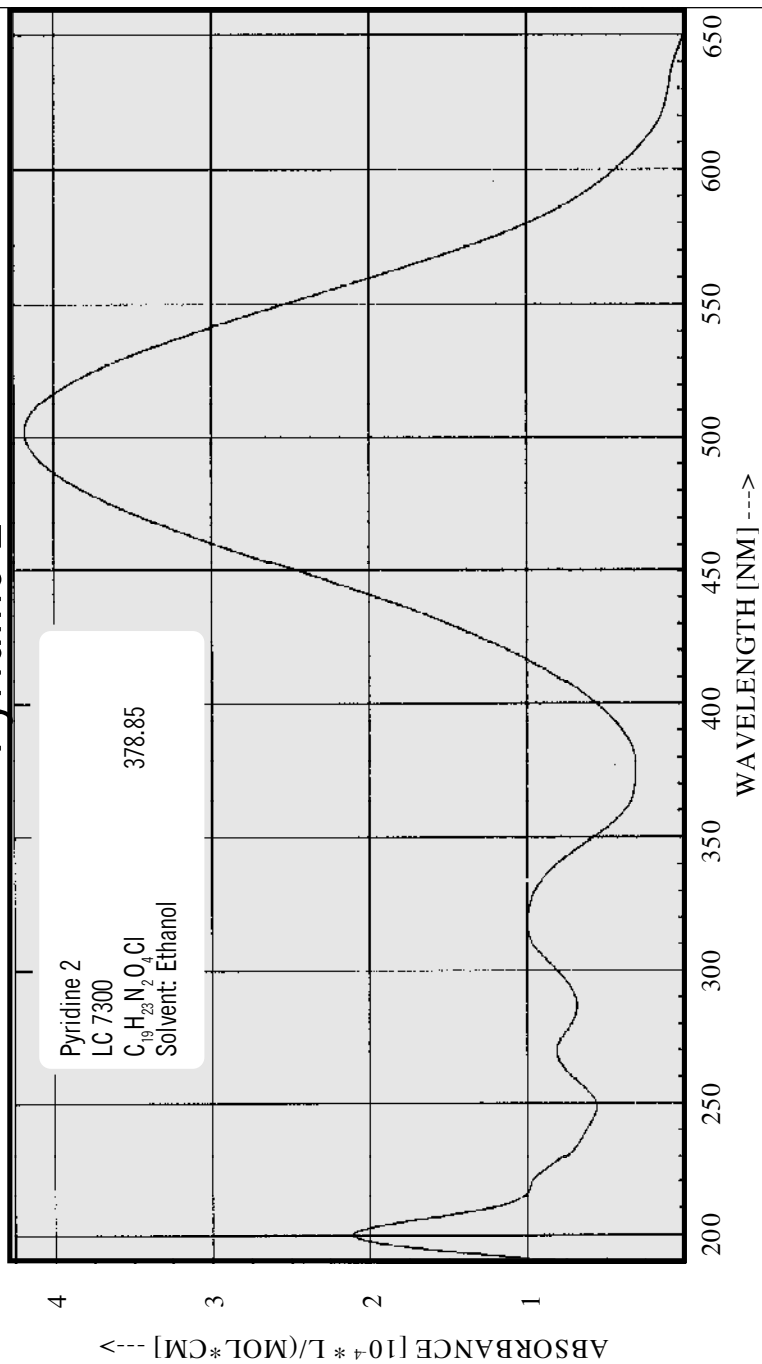
Efficient laser dye for pulsed and CW operation; tunable around 750 nm.

Source	Pump		Dye Laser Characteristics					Ref.
	Wavelength		Peak	Range	Effic.	Conc.	Solvent	
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		777	735 - 796	6	1.25	DMSO	1, 2, 3
Nitrogen	337		724	708 - 780	rel.	0.50	Ethanol	3, 4
CW, Kr <sup>+</sup>	red		810	790 - 900	-	0.62	PC./Eg.	5

## References

1. Lambda Physik, *Wall Chart* 6/83.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B32**, 9 (1983).
3. F. Bos, *Appl. Optics* **20**(20), 3553 (1983).
4. B. M. Pierce, R. R. Birge, *IEEE J. Quantum Electr.* **QE18**(7), 1164 (1982).
5. G. D. Aumiller, *Opt. Commun.* **41**(2), 115 (1982).

## Pyridine 2



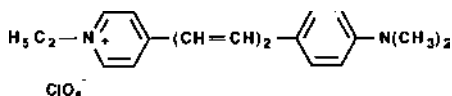


# Pyridine 2 (LC 7300)

## Constitution

1-Ethyl-4-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-pyridinium Perchlorat  
LDS 722

$C_{19}H_{23}N_2O_4Cl$  · MW: 378.85



## Characteristics

Lambdachrome® number: 7300  
Appearance: red, crystalline solid  
Absorption maximum (in ethanol): 500 nm  
Molar absorptivity:  $4.22 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
For research and development purposes only.

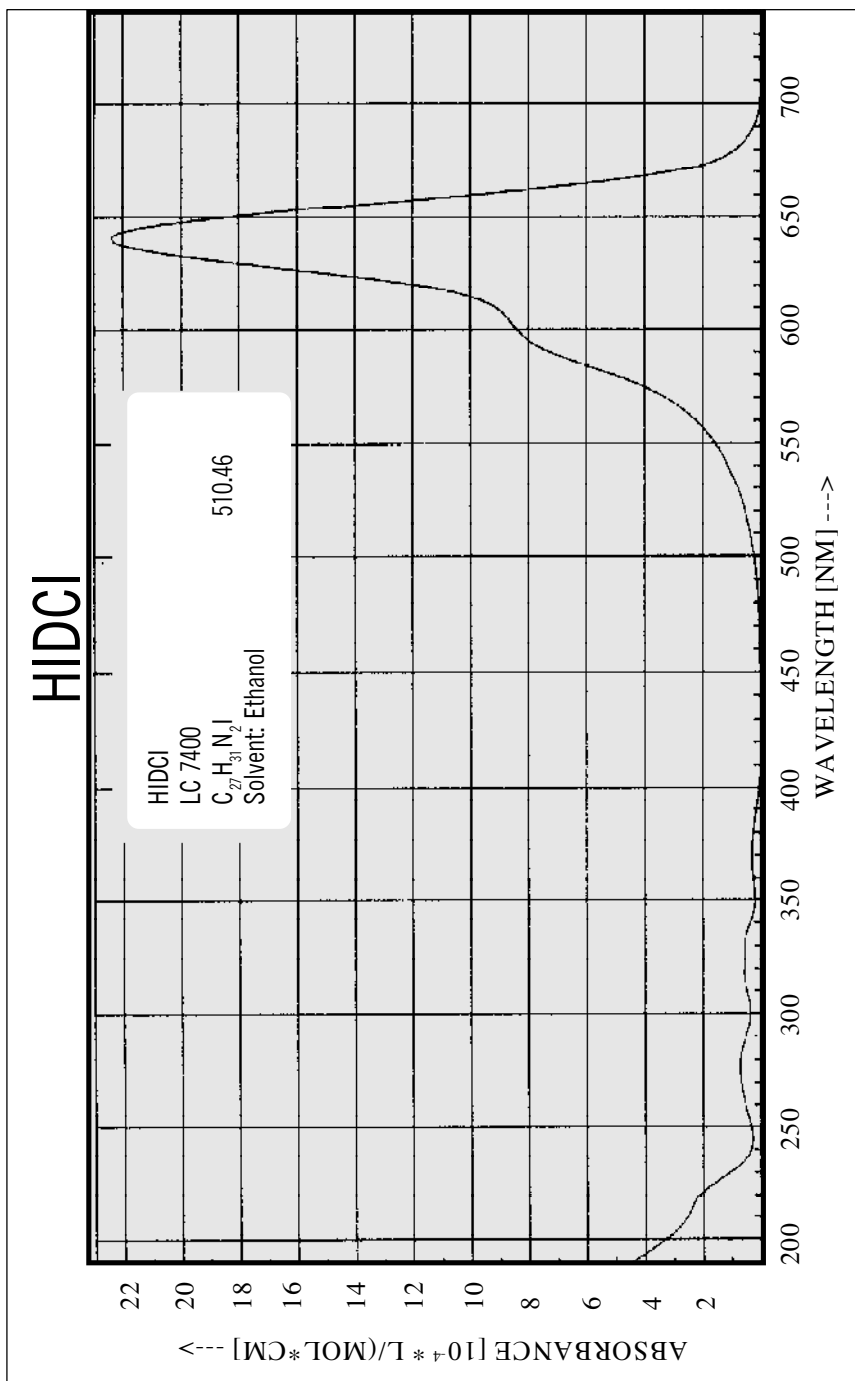
## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 740 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	740	695 - 790	11	0.72	DMSO	1
Nitrogen	337	743	710 - 790	rel.	0.85	DMSO	2
Nd:YAG	532	750	725 - 776	21	0.22	PC	3
Cu-vapor	510	722	687 - 755	4	1.00	Methanol	4
CW, Ar <sup>+</sup>	VIS	720	685 - 820	-	0.75	Pc./Eg.	1

## References

1. Lambda Physik, *Wall Chart* 1996.
2. Lambda Physik, *Data Sheet*.
3. *Lambda Physik*.
3. M. Broyer et al., *Appl. Phys.* **B35**, 31 (1984).

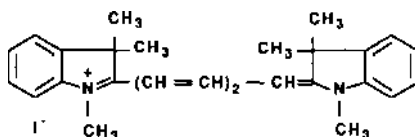


# HIDCI (LC 7400)

## Constitution

1,1',3,3',3'-Hexamethylindodicarbocyanine Iodide  
Hexacyanine 2

$C_{27}H_{31}N_2I$  · MW: 510.46



## Characteristics

Lambdachrome® number:	7400
CAS registry number:	36536-22-8
Appearance:	blue, crystalline solid
Absorption maximum (in ethanol):	639 nm
Molar absorptivity:	$22.5 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

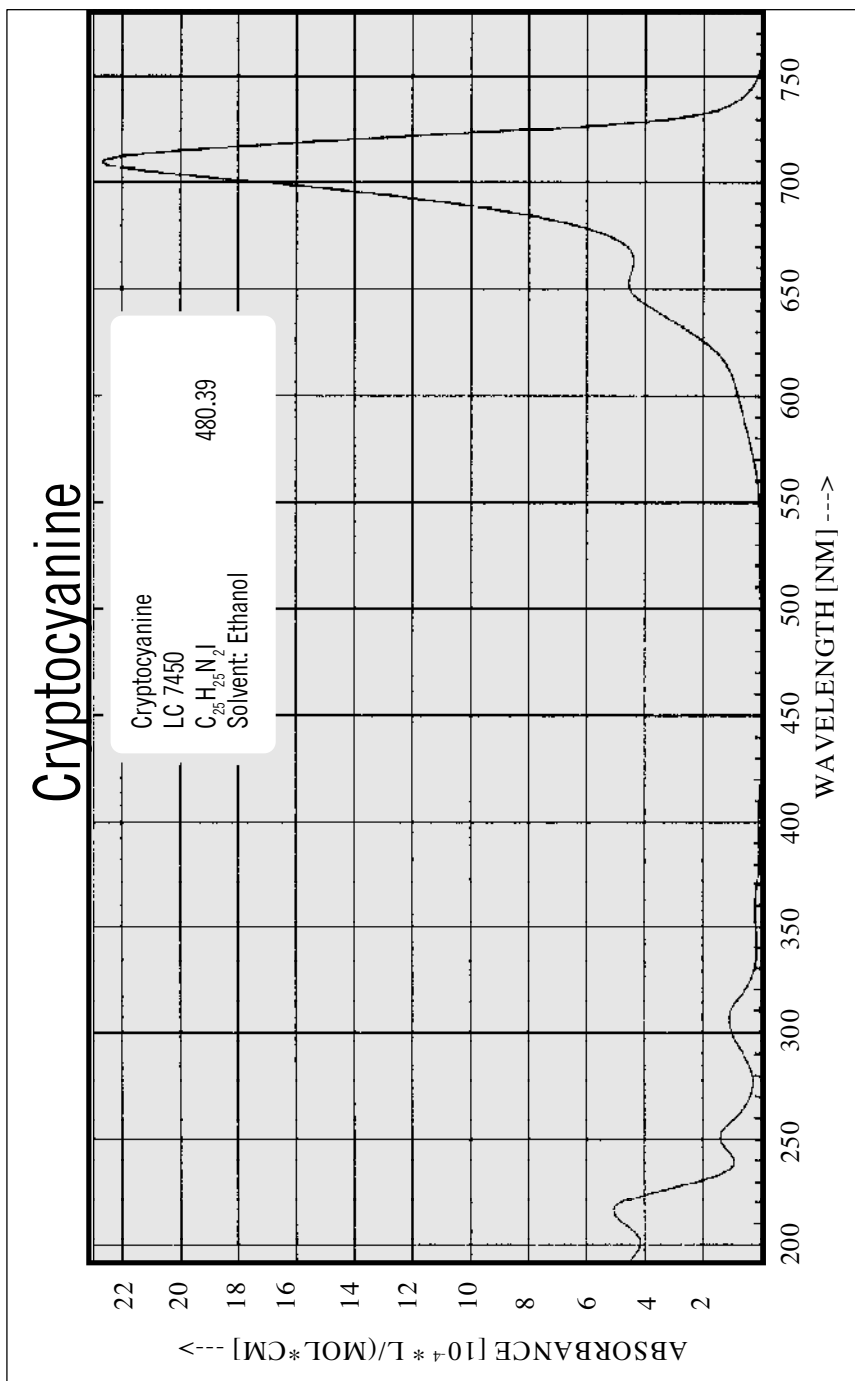
## Lasing Performance

Laser dye for pulsed operation; tunable around 740 nm. Saturable absorber for flashlamp pumped Rhodamine 6G dye lasers; applicable around 630 nm<sup>1,2</sup>.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Flashlamp	-	740	-	-	0.11	DMSO	2

## References

1. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **13**(1), 193 (1974).
2. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **11**(5), 692 (1972).

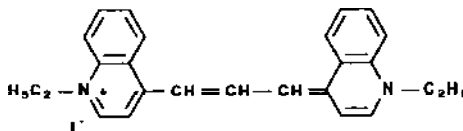


# Cryptocyanine (LC 7450)

## Constitution

1,1'-Diethyl-4,4'-carbocyanine Iodide  
DCI-4

$C_{25}H_{25}N_2I$  · MW: 480.39



## Characteristics

Lambdachrome® number:	7450
CAS registry number:	4727-50-8
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	708 nm
Molar absorptivity:	$22.5 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

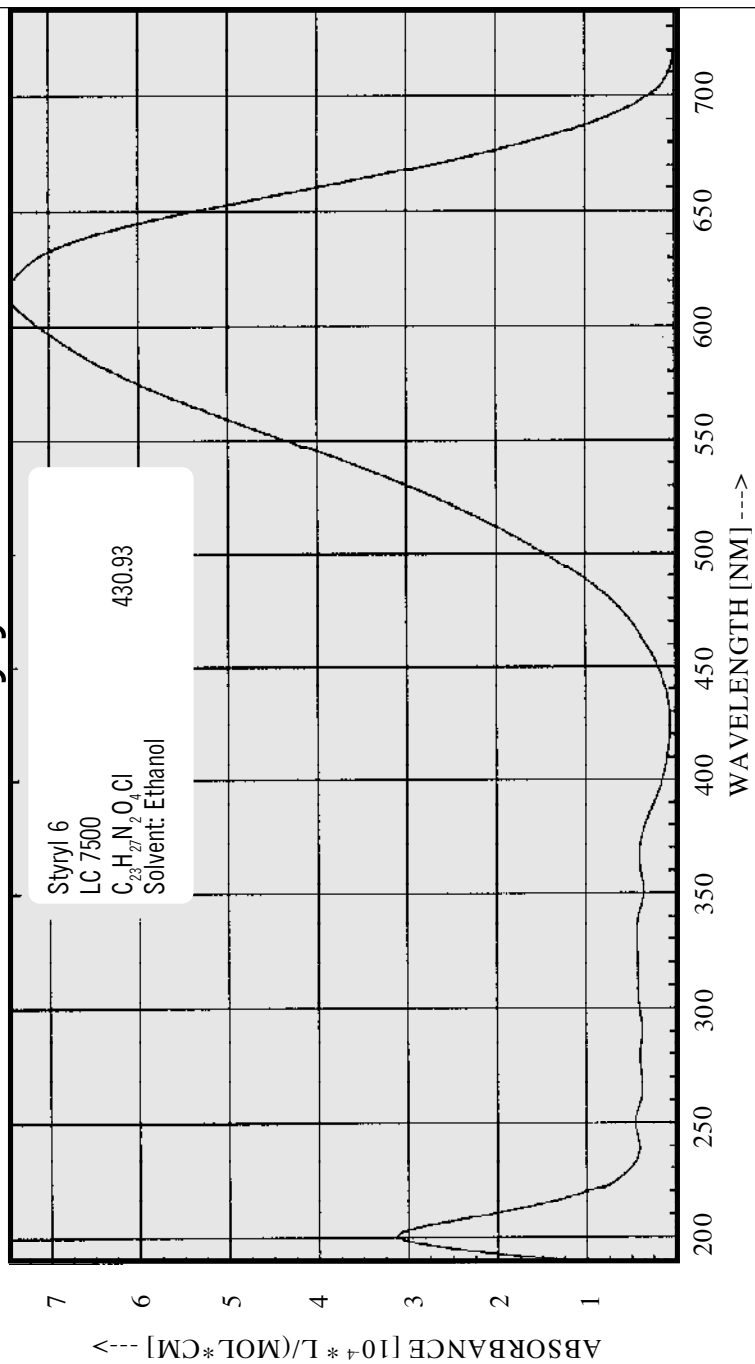
## Lasing Performance

Saturable absorber for the ruby laser; applicable around 700 nm<sup>1., 2., 3., 4.)</sup>.

## References

1. M. L. Spaeth, W. R. Sooy, *J. Chem. Phys.* **48**(5), 2315 (1968).
2. I. K. Krasnyuk et al., *JETP Letters* **7**(4), 89 (1968).
3. H. W. Mockler, R. J. Collins, *Appl. Phys. Letters* **7**(10), 270 (1965).
4. V. I. Malyshev, A. S. Markin, A. A. Sychev, *JETP Letters* **6**, 34 (1967).

# Styryl 6



# Styryl 6 (LC 7500)

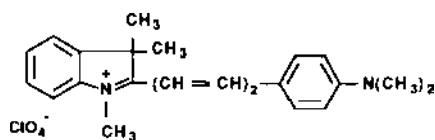
## Constitution

2-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-1,3,3-trimethyl-3H-indolium

Perchlorate

LDS 730

$C_{23}H_{27}N_2O_4Cl$  · MW: 430.93



## Characteristics

Lambdachrome® number:	7500
CAS registry number:	-
Appearance:	blue, crystalline solid
Absorption maximum (in ethanol):	615 nm
Molar absorptivity:	$7.38 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

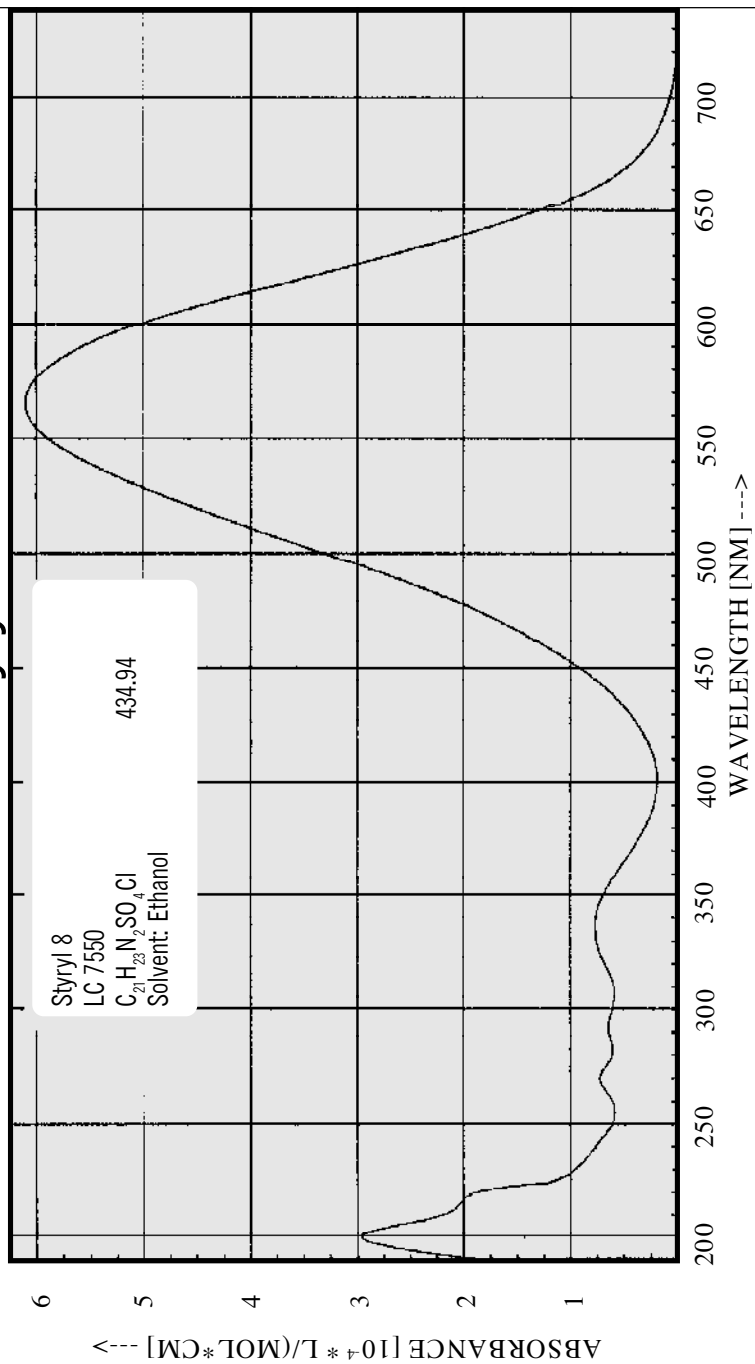
Laser dye for pulsed operation; tunable around 720 nm

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
Nd:YAG, 2nd	532		721	708 - 735	16	0.28	PC	1

## References

1. Lambda Physik, *Wall Chart* 1996.
2. K. Kato, *IEEE J. Quantaum Electr.* QE-16(10), 1017 (1980).

## Styryl 8



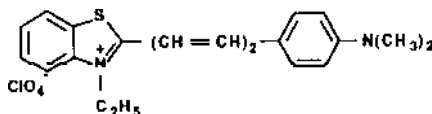


# Styryl 8 (LC 7550)

## Constitution

2-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-3-ethylbenzothiazolium Perchlorat  
LDS 751

$C_{21}H_{23}N_2SO_4Cl$  · MW: 434.94



## Characteristics

Lambdachrome® number:	7550
CAS registry number:	76433-29-9
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	570 nm
Molar absorptivity:	$6.15 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

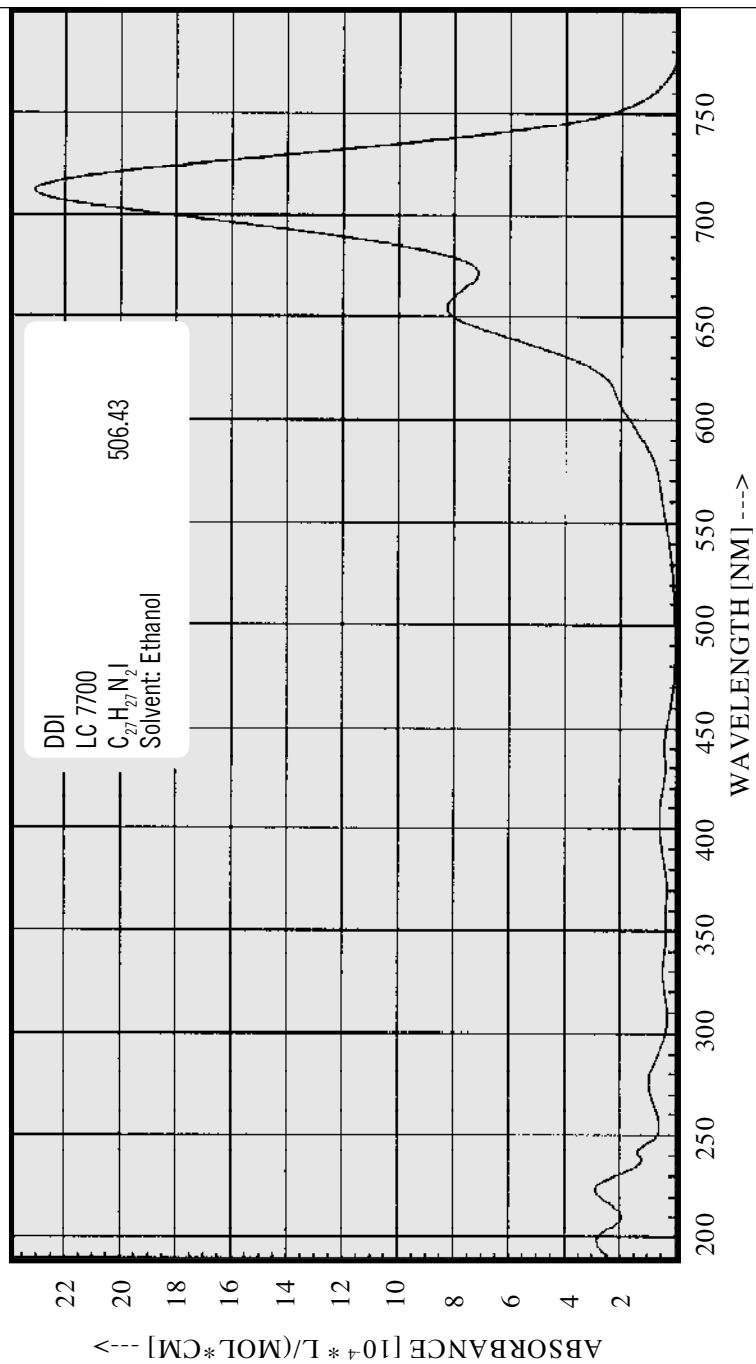
Laser dye for pulsed and CW operation; tunable around 750 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nd:YAG, 2nd	532	750	717 - 780	13	0.15	PC	1
Cu-vapor	510	711	703 - 724	3	1.70	Methanol	2
CW, Ar <sup>+</sup>	VIS	780	700 - 840	-	-	Pc./Eg.	3, 4

## References

1. Lambda Physik, *Wall Chart* 1996.
2. M. Broyer et al., *Appl. Phys.* **B35**, 31 (1984).
3. J. Hoffnagle et al., *Opt. Commun.* **42(4)**, 267 (1982).
4. J. J. L. Mulders, L. W. G. Steenhuysen, *Opt. Commun.* **54(5)**, 295 (1985).

## DDI

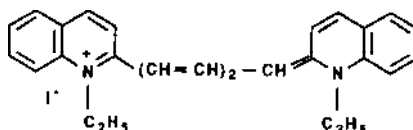


# DDI (LC 7700)

## Constitution

1,1'-Diethyl-2,2'-dicarbocyanine Iodide

$C_{27}H_{27}N_2I$  · MW: 506.43



## Characteristics

Lambdachrome® number:	7700
CAS registry number:	14187-31-6
Appearance:	green, crystalline solid
Absorption maximum (in methanol):	710 nm
Molar absorptivity:	$23.0 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	745 nm
For research and development purposes only.	

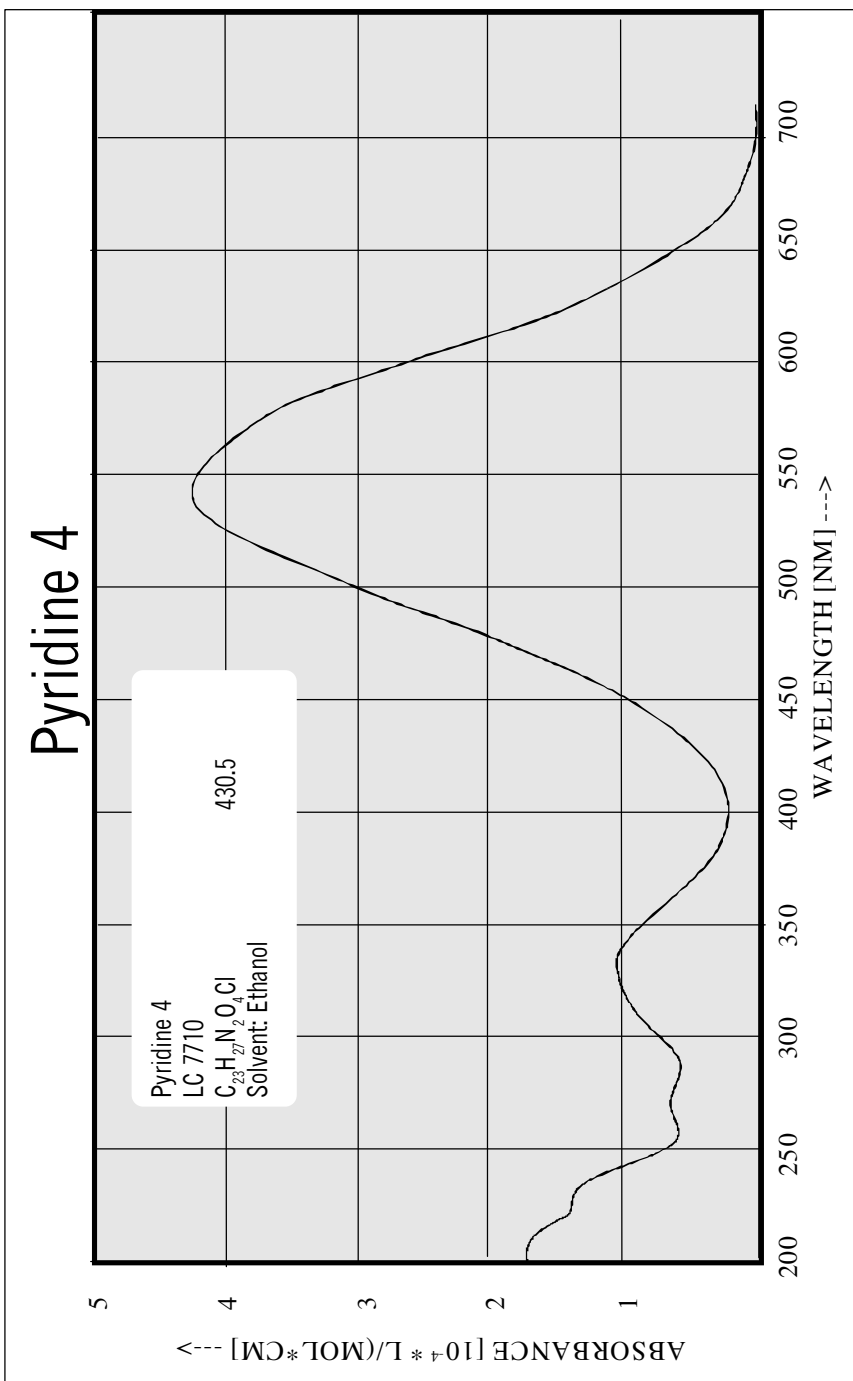
## Lasing Performance

Laser dye for pulsed operation; tunable around 800 nm. Saturable absorber for the Ruby laser and flashlamp pumped Cresyl Violet and Rhodamine 700 dye lasers; applicable around 710 nm<sup>1., 2., 3.)</sup>.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
Ruby	694	806	-	13	0.03	Ethanol	4

## References

1. E. G. Arthurs et al., *Appl. Phys. Letters* **20**(3), 125 (1972).
2. M. E. Mack, *IEEE J. Quantum Electr.* **QE-4**, 1015 (1968).
3. W. Sibbett, J. R. Taylor, *IEEE J. Quantum Electr.* **20**(2), 108 (1984).
4. A. M. Bonch-Bruевич, *Opt. Spectr.* **28**, 51 (1970).

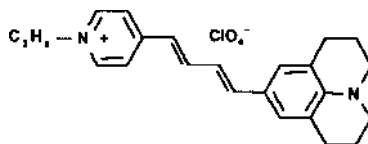


# Pyridine 4 (LC 7710)

## Constitution

1-Ethyl-4-(4-(9-(2,3,6,7-tetrahydro-1H,5H-benzo(i,j)-chinolizinium))-1,3-butadienyl)-pyridinium Perchlorate

$C_{23}H_{27}N_2O_4Cl$  · MW: 430.5



## Characteristics

Lambdachrome® number:	7710
CAS Registry number:	-
Appearance:	dark brown, crystalline solid
Absorption maximum (in ethanol):	550 nm
Molar absorptivity:	$4.26 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	-
For research and development purposes only.	

## Lasing Performance

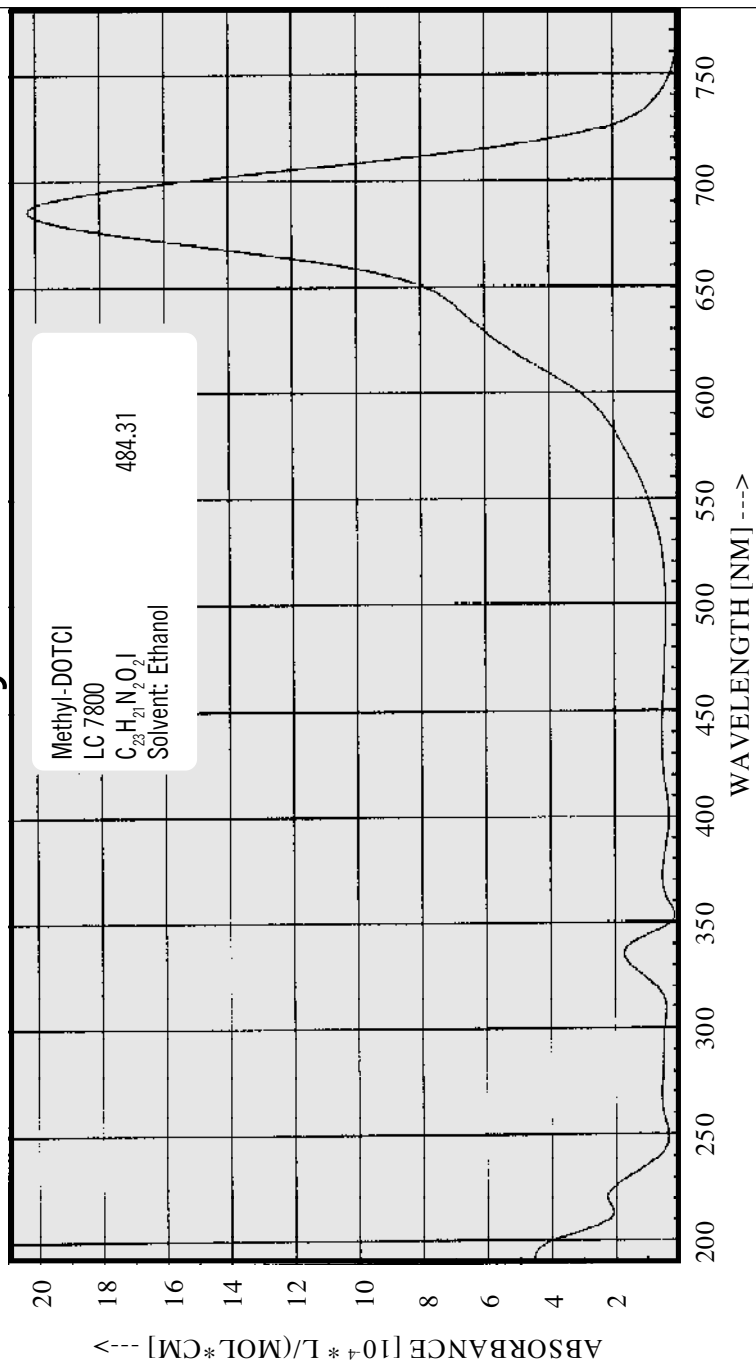
Efficient laser dye for pulsed and CW operation; tunable around 770 nm.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	771	744 - 812	7	0.75	DMSO	1

## References

1. Lambda Physik, *Wall Chart* 1996.

# Methyl-DOTCI

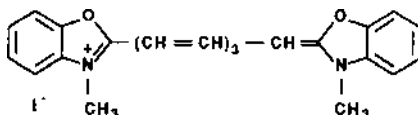


# Methyl-DOTCI (LC 7800)

## Constitution

3,3'-Dimethyloxatricarbocyanine Iodide DMOTCI · NK 199

$C_{23}H_{21}N_2O_2I$  · MW: 484.31



## Characteristics

Lambdachrome® number:	7800
Appearance:	blue, crystalline solid
Absorption maximum (in ethanol):	682 nm
Molar absorptivity:	$19.8 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	718 nm
For research and development purposes only.	

## Lasing Performance

Efficient laser dye for pulsed and CW operation; tunable around 780 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	792	774 - 810	4	0.90	DMSO	1, 2, 3
Nitrogen	337	780	768 - 820	rel.	0.51	DMSO	3, 4
Nd:YAG, 2nd	532	780	-	rel.	-	DMSO	5
Flashlamp	-	810	-	-	0-07	DMSO	6
CW, Kr <sup>+</sup>	red	-	745 - 790	-	1.45	Eg.	7

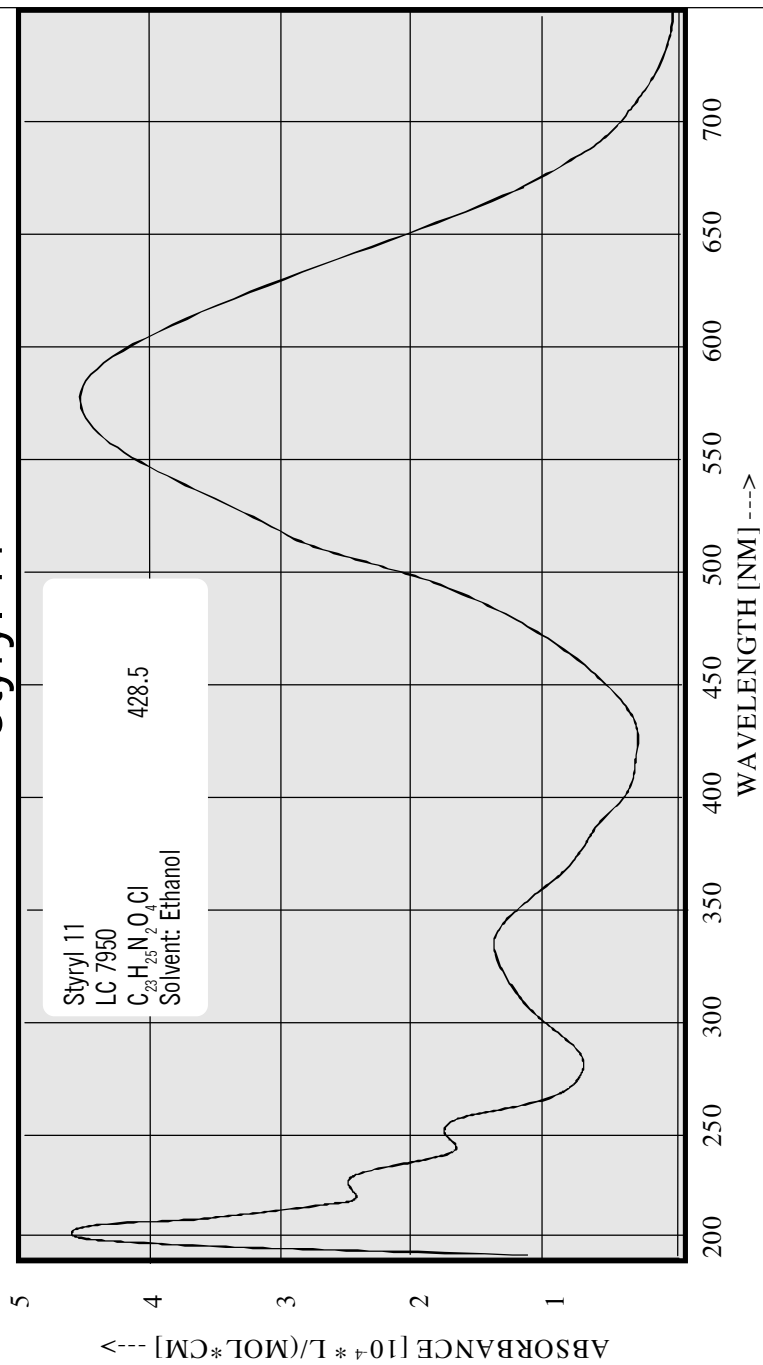
## References

1. Lambda Physik, *Wall Chart* 6/83.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 403 (1981).
3. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. F. Bos, *Appl. Optics* **20**(10), 1886 (1981).
6. C. Loth, P. Flamant, *Opt. Commun.* **21**(1), 13 (1977).
7. J. M. Yarborough, *Appl. Phys. Letters* **24**(12), 629 (1974).

## P.S.

The DOTCI (3,3'-Diethyloxatricarbocyanine Iodide) shows identical performance, however, its photochemical stability is much lower.

# Styryl 11



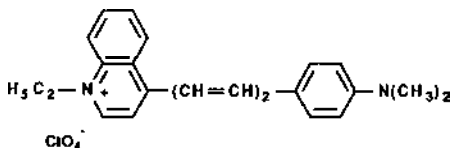


# Styryl 11 (LC 7950)

## Constitution

1-Ethyl-4-(4-(p-Dimethylaminophenyl)-1,3-butadienyl)-quinolinium Perchlorate  
LDS 798

$C_{23}H_{25}N_2O_4Cl$  · MW: 428.5



## Characteristics

Lambdachrome® number:	7950
CAS registry number:	92479-59-9
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	575 nm
Molar absorptivity:	$4.55 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

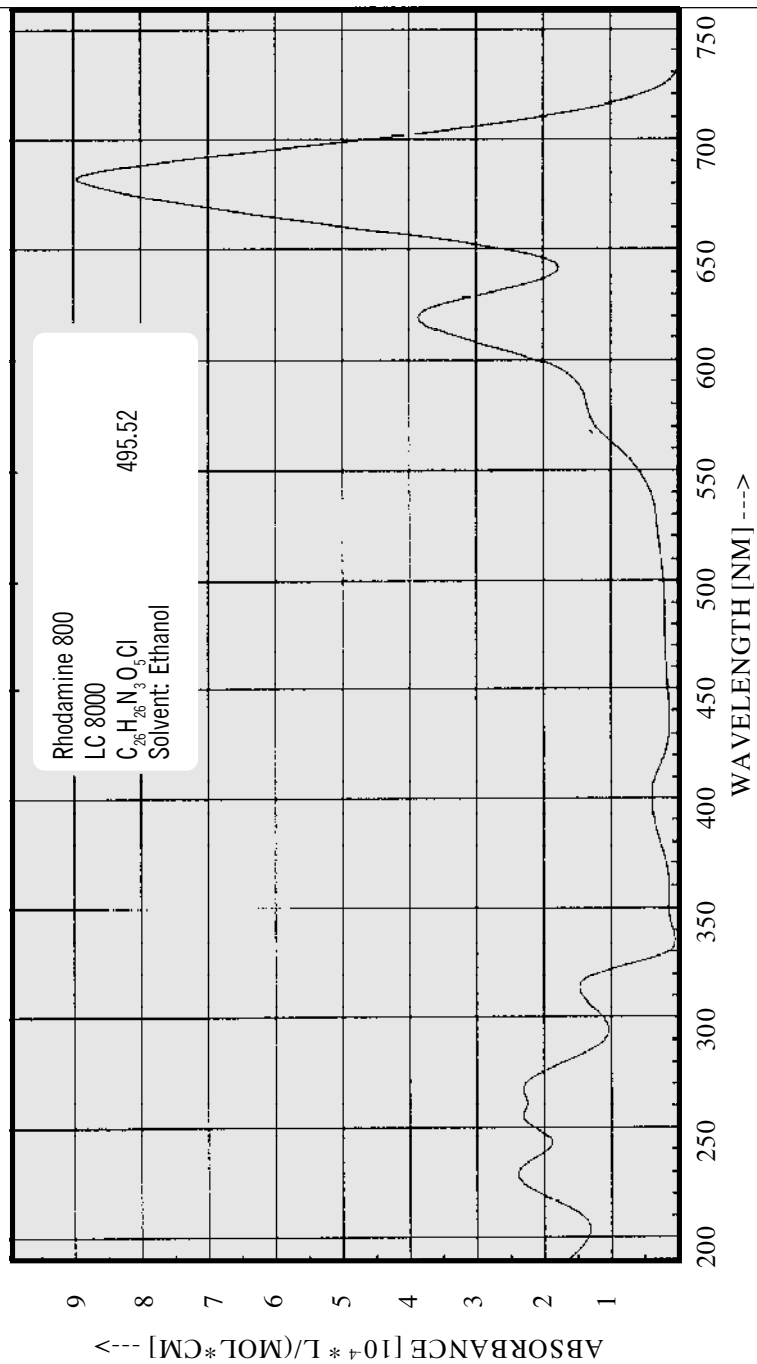
Laser dye for pulsed operation; tunable around 800 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
CW, Ar <sup>+</sup>	VIS	800	770 - 845	5	0.51	Pc./Eg.	1

## References

1. J. Hoffnagle et al., *Opt. Commun.* **42**(4), 267 (1982).

## Rhodamine 800

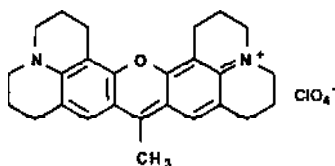


# Rhodamine 800 (LC 8000)

## Constitution

8-Cyano-2,3,5,6,11,12,14,15-octahydro-1*H*,4*H*,10*H*,13*H*-diquinolizino[9,9*a*,1-*bc*:9',9*a'*,1-*hi*]xanthylum Perchlorate

C<sub>26</sub>H<sub>26</sub>N<sub>3</sub>O<sub>5</sub>Cl · MW: 495.52



## Characteristics

Lambdachrome® number:	8000
CAS registry number:	101027-54-7
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	682 nm
Molar absorptivity:	8.95 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

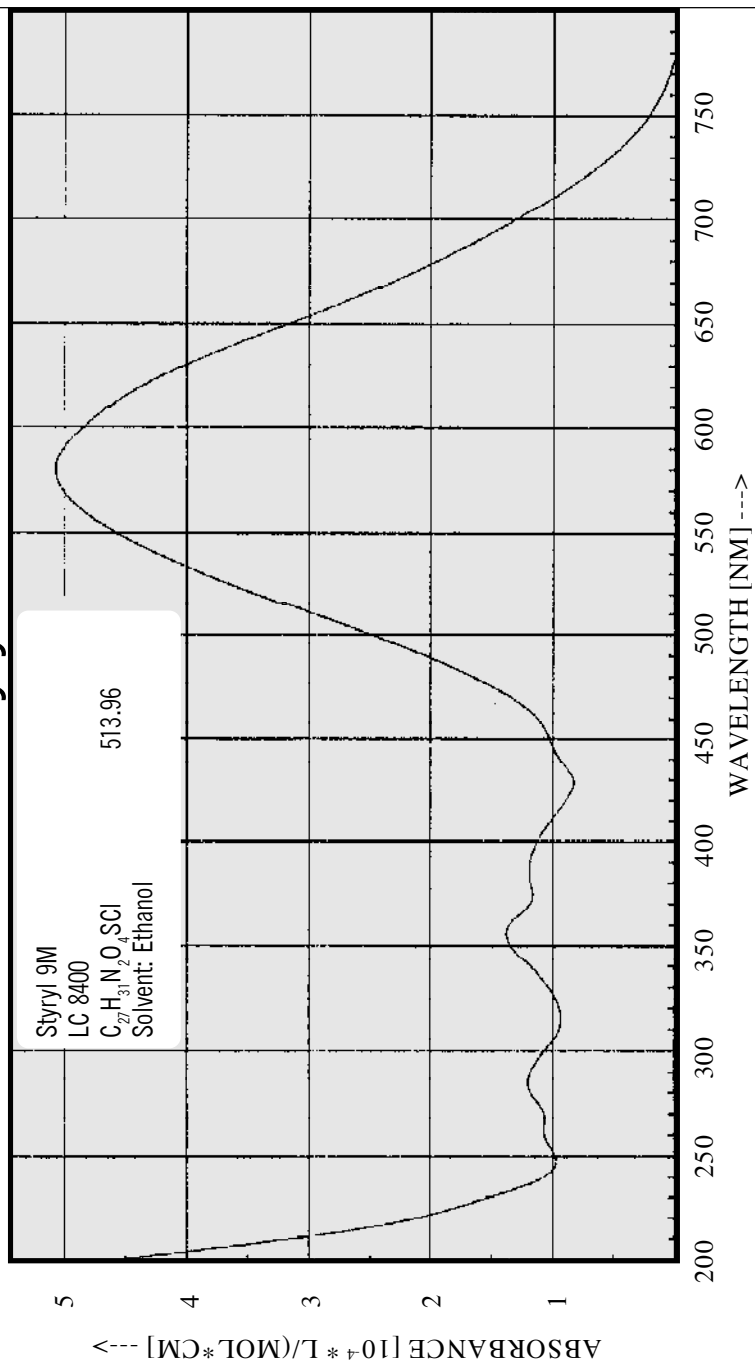
Very efficient laser dye for pulsed and CW operation; tunable around 810 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	810	776 - 823	6	1.00	DMSO	1
CW, Kr <sup>+</sup>	red	795	730 - 835	22	0.21	Eg.	2

## References

1. Lambda Physik, *Wall Chart* 1996.
2. R. Raue, H. Harnisch, K. H. Drexhage, *Heterocycles* **21**(1), 167 (1984).

# Styryl 9M

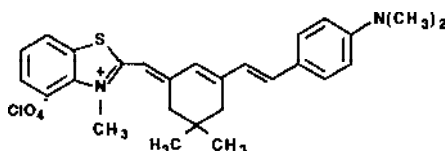


# Styryl 9M (LC 8400)

## Constitution

2-(6-(4-Dimethylaminophenyl)-2,4-neopentylene-1,3,5-hexatrienyl)-3-methyl-benzothiazolium Perchlorat  
LDS 821

$C_{27}H_{31}N_2O_4S \cdot Cl$  · MW: 513.96



## Characteristics

Lambdachrome® number:	8400
CAS registry number:	120528-73-6
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	585 nm
Molar absorptivity:	$5.05 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

Very efficient laser dye for pulsed and CW operation; tunable around 840 nm

Pump		Dye Laser Characteristics				Solvent	Ref.
Source	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	840	810 - 875	9	1.10	DMSO	1,2
Nitrogen	337	840	803 - 875	rel.	1.03	DMSO	3
Nd:YAG, 2nd	532	824	797 - 851	15	0.26	Pc.	1, 4
Cu-vapor	510	815	793 - 845	14	0.67	Methanol	5
Flashlamp	-	840	810 - 860	-	0.01	Pc./Eg.	6, 7
CW, Ar <sup>+</sup>	VIS	830	785 - 900	-	2.0	Pc./Eg.	1, 8,9

## References

See page 236.

## P.S.

The 3-Ethyl-derivative (Styryl 9/LDS 820) shows similar performance. However, its photochemical stability in CW pumped dye lasers is slightly lower.

## References (STYRYL 9M)

1. Lambda Physik, *Wall Chart* 1996.
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* B32, 9 (1983).
3. Lambda Physik, *Data Sheet*.
4. K. Kato, *IEEE J. Quantum Electr.* QE-16(19), 1017 (1980).
5. M. Broyer et al., *Appl. Phys.* B35, 31 (1984).
6. K. Smith, W. Sibbett, J. R. Taylor, *Opt. Commun.* 49(5), 359 (1984).
7. Cheng-Huei Lin, B. Marshall, *Appl. Optics* 23(14), 2228 (1984).
8. J. Hofnagle et al., *Opt. Commun.* 42(4), 267 (1982).
9. J. J. L. Mulders, L. W. G. Steenhuysen, *Opt. Commun.* 54(5), 295 (1985).

## References (HITCI)

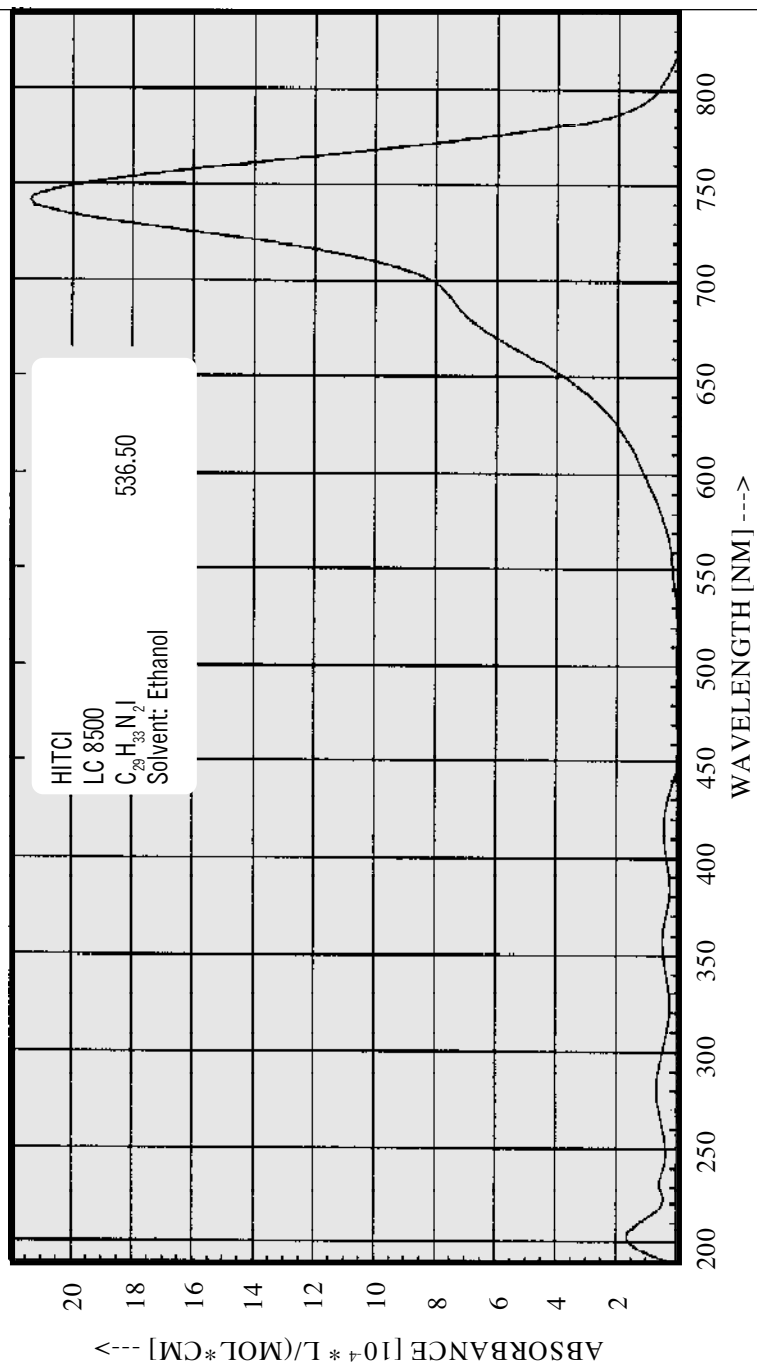
1. Lambda Physik, *Wall Chart* 1996.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* 38(5,6), 403 (1981).
3. F. Bos, *Appl. Optics* 20(10), 3553 (1981).
4. Lambda Physik, *Data Sheet*.
5. F. Bos, *Appl. Optics* 20(20), 1886 (1981).
6. A. Hirth, K. Vollrath, J. Faure, D. Loughnot, *Opt. Commun.* 7(4), 339 (1973).
7. Coherent, *CW Dye Laser Fact Sheets*.
8. T. F. Johnston, R. H. Brady, W. Proffitt, *Appl. Optics* 21(13), 2307 (1982).

## References (IR 140)

1. Lambda Physik, *Wall Chart* 6/83.
2. F. Bos, *Appl. Optics* 20(20), 3553 (1981).
3. V. S. Antonov, K. L. Hohla, *Appl. Phys.* B30, 109 (1983).
4. Lambda Physik, *Data Sheet*.
5. F. Bos, *Appl. Optics* 20(10), 1886 (1981).
6. C. A. Moore, C. D. Decker, *J. Appl. Phys.* 49(1), 47 (1978).
7. C. D. Decker, *Appl. Phys. Letters* 27(11), 607 (1975).
8. J. P. Webb et al., *IEEE J. Quantum Electr.* QE-11, 114 (1975).
9. Coherent, *CW Dye Laser Fact Sheets*.
10. Lambda Physik.



# HITCI



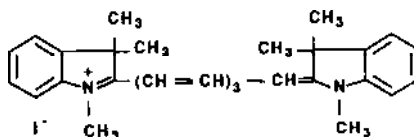


# HITCI (LC 8500)

## Constitution

1,1',3,3,3',3'-Hexamethylindotricarbocyanine Iodide  
Hexacyanine 3

$C_{29}H_{33}N_2I$  · MW: 536.50



## Characteristics

Lambdachrome® number: 8500  
CAS registry number: 19764-96-6  
Appearance: green, crystalline solid  
Absorption maximum (in ethanol): 741 nm  
Molar absorptivity:  $21.5 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in ethanol): 778 nm  
For research and development purposes only.

## Lasing Performance

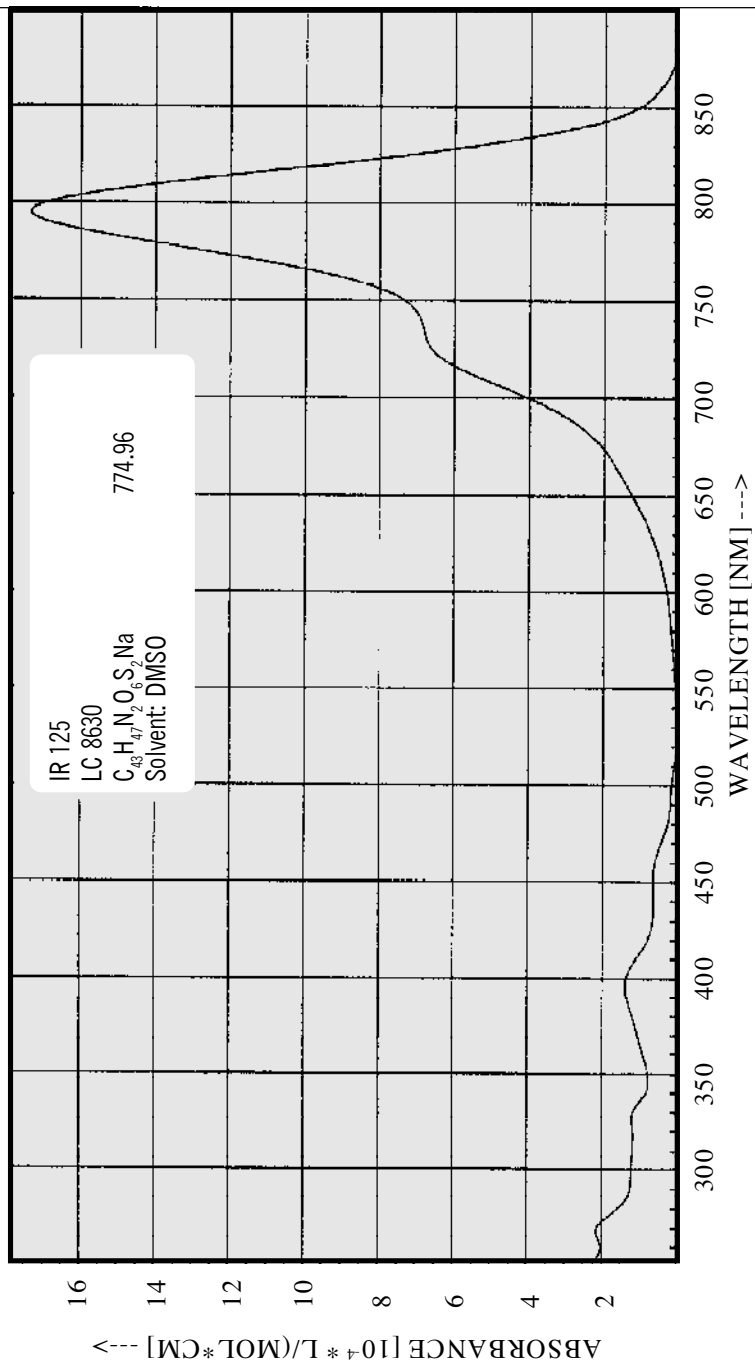
Laser dye for pulsed and CW operation; tunable around 860 nm.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	868	837 - 905	4	1.20	DMSO	1, 2, 3
Nitrogen	337	846	828 - 891	rel.	1.06	DMSO	3, 4
Nd:YAG, 2nd	532	815	-	-	-	Ethanol	5
Flashlamp	-	879	-	-	0.11	DMSO	6
CW, Kr <sup>+</sup>	ir	880	815 - 920	10	0.3	DMSO/Eg.	7, 8

## References

See page 236.

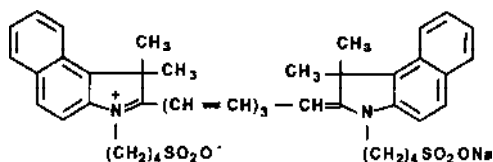
## IR 125



# IR 125 (LC 8630)

## Constitution

$C_{43}H_{47}N_2O_6S_2Na$  · MW: 774.96



## Characteristics

Lambdachrome® number:	8630
CAS registry number:	3599-32-4
Appearance:	bronze, crystalline solid
Absorption maximum (in DMSO):	795 nm
Molar absorptivity:	$17.3 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in chloroform):	838
For research and development purposes only.	

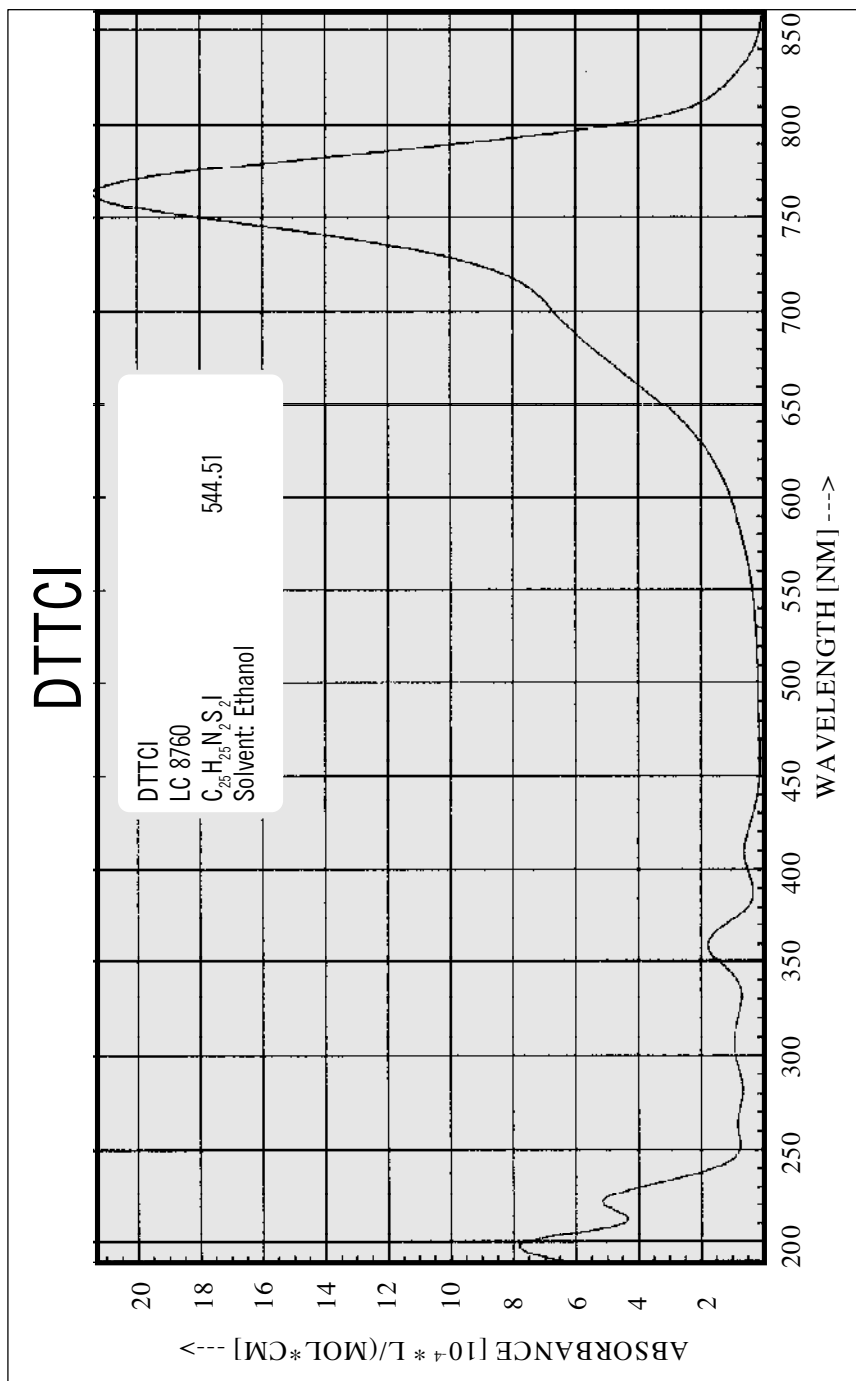
## Lasing Performance

Laser dye for pulsed operation; tunable around 920 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	920	890 - 960	4	2.0	DMSO	1
Nitrogen	337	918	893 - 958	rel.	1.94	DMSO	2, 3
Nd:YAG, 2nd	532	913	-	3	0.39	DMSO	4
Flashlamp	-	940	-	-	0.08	DMSO	5

## References

1. Lambda Physik, *Wall Chart* 6/90.
2. Lambda Physik, *Data Sheet*.
3. B. M. Pierce, R. R. Birge, *IEEE J. Quantum Electr.* **QE-18**(7), 1164 (1982).
4. C. D. Decker, *Appl. Phys. Letters* **27**(11), 607 (1975).
5. J. P. Webb et al., *IEEE J. Quantum Electr.* **QE-11**, 114 (1975).

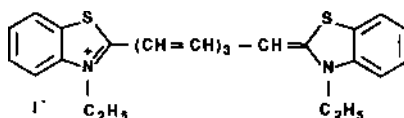


# DTTCI (LC 8760)

## Constitution

3,3'-Diethylthiatriccyanine iodide

$C_{25}H_{25}N_2S_2I$  · MW: 544.51



## Characteristics

Lambdachrome® number:	8760
CAS registry number:	3071-70-3
Appearance:	blue, crystalline solid
Absorption maximum (in ethanol):	760 nm
Molar absorptivity:	$21.0 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in chloroform):	815
For research and development purposes only.	

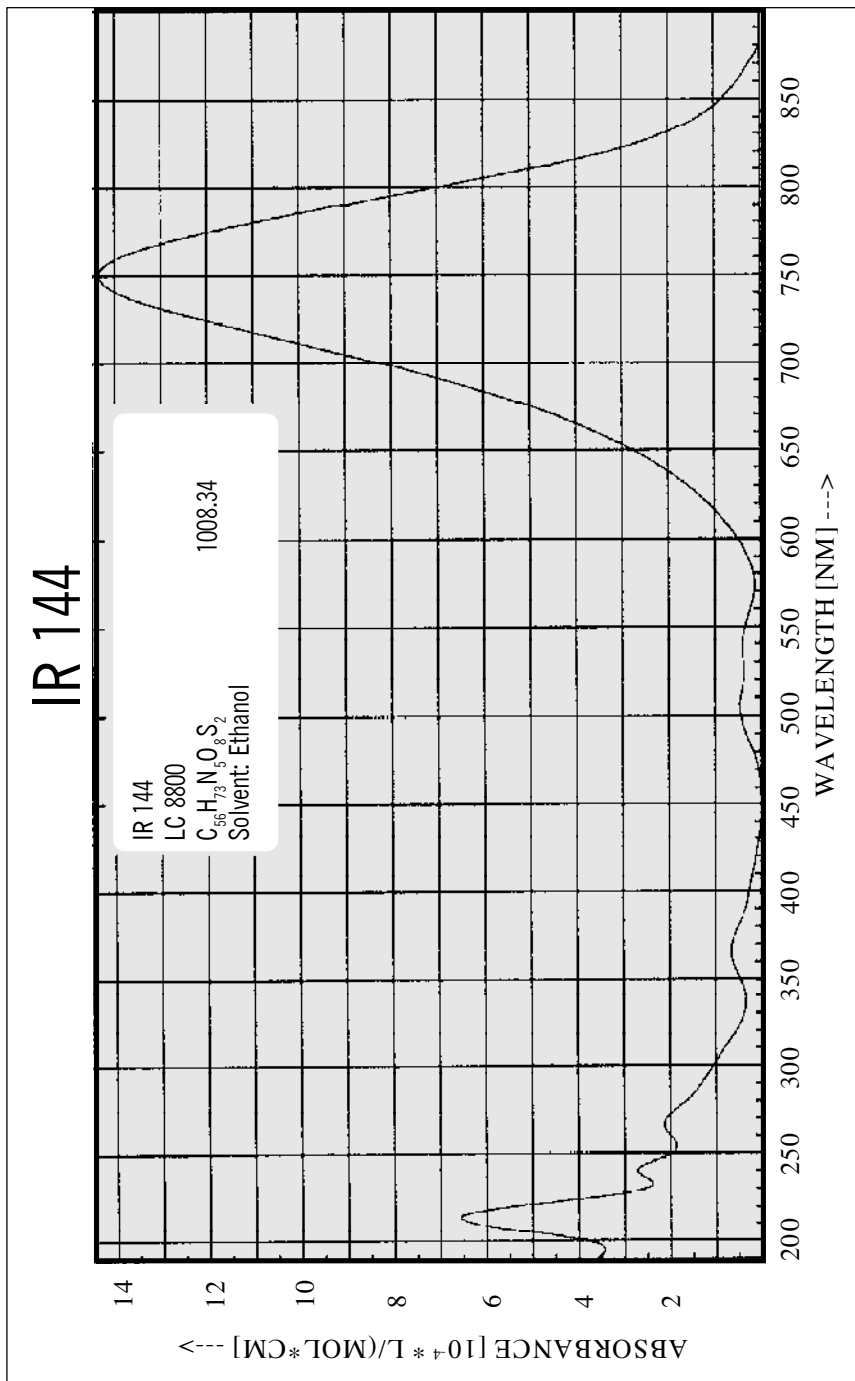
## Lasing Performance

Laser dye for pulsed operation; tunable around 850 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	849	828 - 883	1	0.42	DMSO	1
Nitrogen	337	852	834 - 892	rel.	0.60	DMSO	1, 2
Flashlamp	-	889	-	-	0.11	Methanol	3, 4

## References

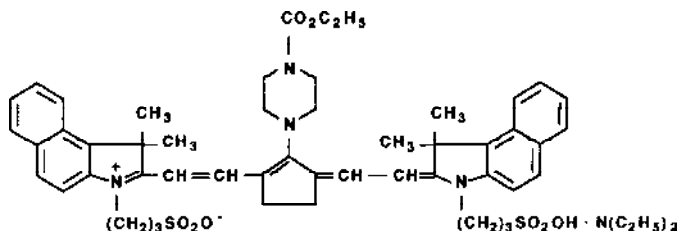
1. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
2. B. M. Pierce, R. R. Birge, *IEEE J. Quantum Electr.* **QE-18**(7), 1164 (1982).
3. M. Maeda, Y. Miyazoe, *Jap. J. Appl. Phys.* **11**(5), 692 (1972).
4. A. Hirth, K. Vollrath, J. Faure, D. Lounnot, *Opt. Commun.* **7**(4), 339 (1973).



# IR 144 (LC 8800)

## Constitution

$C_{56}H_{73}N_5O_8S_2$  · MW: 1008.34



## Characteristics

Lambdachrome® number:	8800
CAS registry number:	54849-69-3
Appearance:	bronze, crystalline solid
Absorption maximum (in DMSO):	750 nm
Molar absorptivity:	$14.1 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	848
For research and development purposes only.	

## Lasing Performance

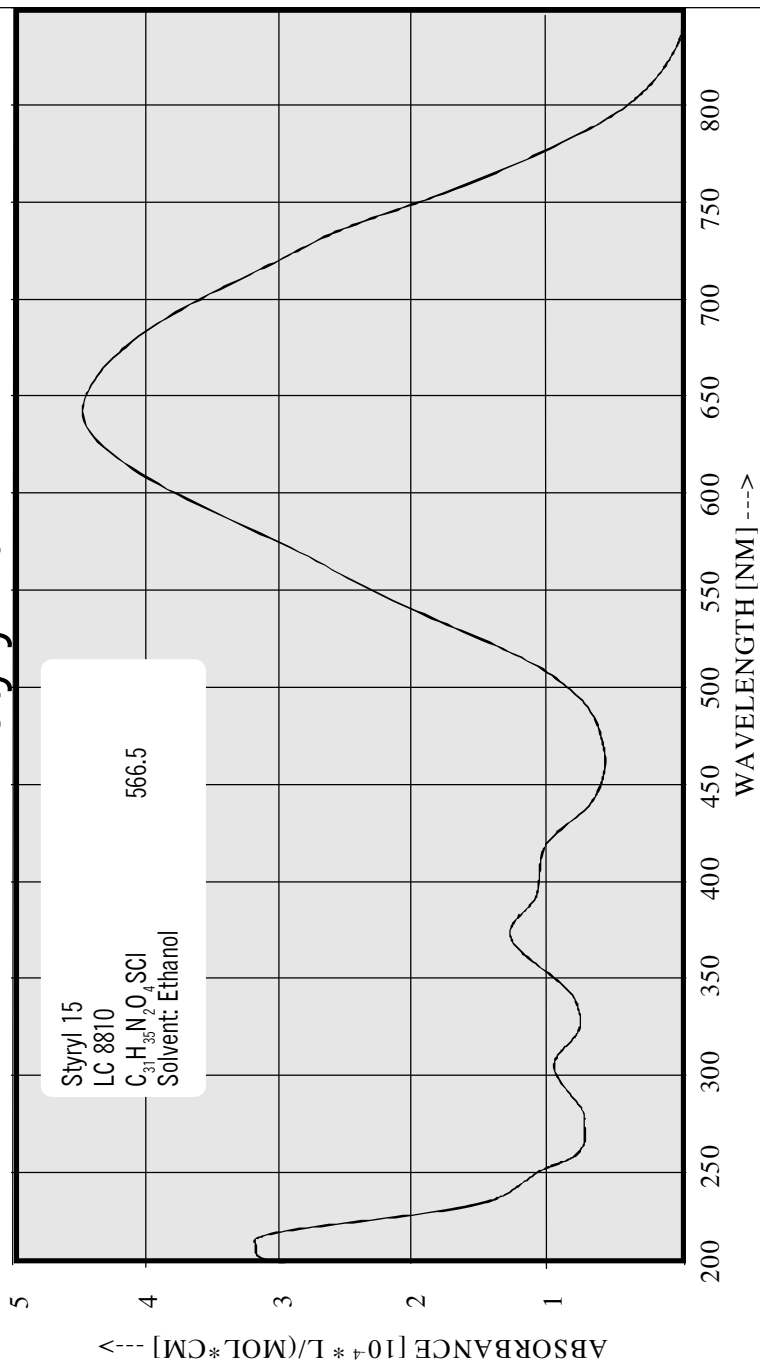
Laser dye for pulsed operation; tunable around 880 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	869	856 - 879	3	1.08	DMSO	1, 2
Nitrogen	337	874	862 - 892	rel.	1.61	DMSO	1, 3, 4
Nd:YAG, 2nd	532	867	-	6	0.30	DMSO	5, 6
Flashlamp	-	880	-	-	0.10	DMSO	7

## References

1. F. Bos, *Appl. Optics* **20**(20), 3553 (1981).
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B30**, 109 (1983).
3. Lambda Physik, *Data Sheet*.
4. B. M. Pierce, R. R. Birge, *IEEE J. Quantum Electr.* **QE-18**(7), 1164 (1982).
5. F. Bos, *Appl. Optics* **20**(10), 1886 (1981).
6. C. A. Moore, C. D. Decker, *J. Appl. Phys.* **49**(1), 47 (1978).
7. J. P. Webb et al., *IEEE J. Quantum Electr.* **QE-11**, 114 (1975).

## Styryl 15



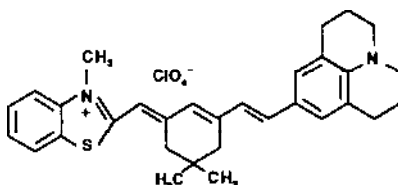


# Styryl 15 (LC 8810)

## Constitution

2-(6-(9-(2,3,6,7-Tetrahydro-1H,5H-benzo(i,j)-chinolizinium))-2,4-neopentylene-1,3,5-hexatrienyl)-3-methylbenzothiazolium Perchlorate

$C_{31}H_{35}N_2O_4SCl$  · MW: 566.5



## Characteristics

Lambdachrome® number:	8810
CAS Registry number:	-
Appearance:	dark green, crystalline solid
Absorption maximum (in ethanol):	648 nm
Molar absorptivity:	$4.42 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	-
For research and development purposes only.	

## Lasing Performance

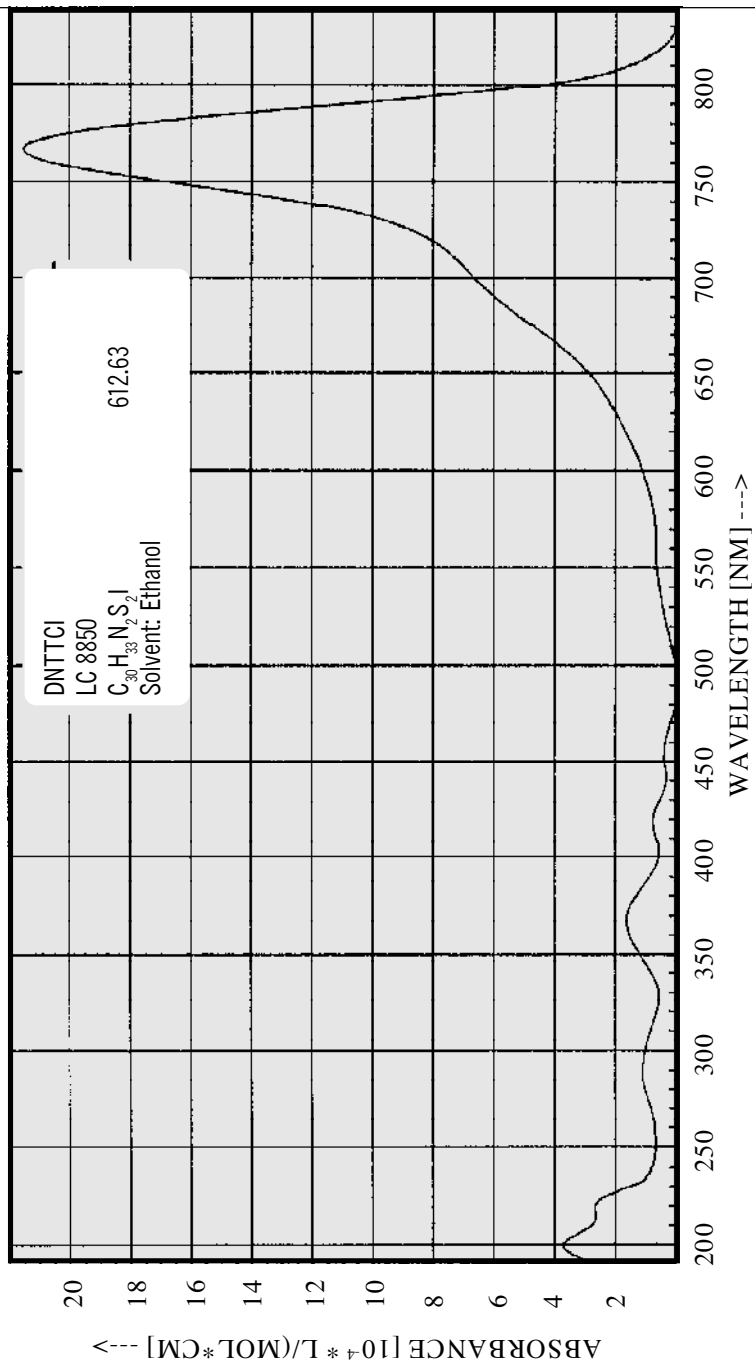
Efficient laser dye for pulsed and CW operation; tunable around 880 nm.

Source	Pump		Dye Laser Characteristics					Ref.
	Wavelength		Peak	Range	Effic.	Conc.	Solvent	
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		880	856 - 918	7	1.15	DMSO	1
Nd:YAG, 2nd	532		880	856 - 918	7	0.62	Pc.	1

## References

1. Lambda Physik, *Wall Chart* 1996.

# DNTTCI

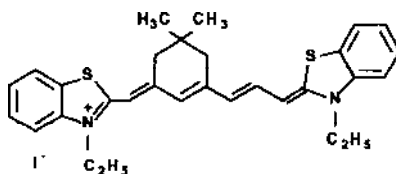


# DNTTCl (LC 8850)

## Constitution

3,3'-Diethyl-9,11-neopentylenethiatricarbocyanine Iodide

$C_{30}H_{33}N_2S_2I \cdot MW: 612.63$



## Characteristics

Lambdachrome® number:	8850
CAS registry number:	-
Appearance:	brass colored, crystalline solid
Absorption maximum (in ethanol):	765nm
Molar absorptivity:	$22.5 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum :	-
For research and development purposes only.	

## Lasing Performance

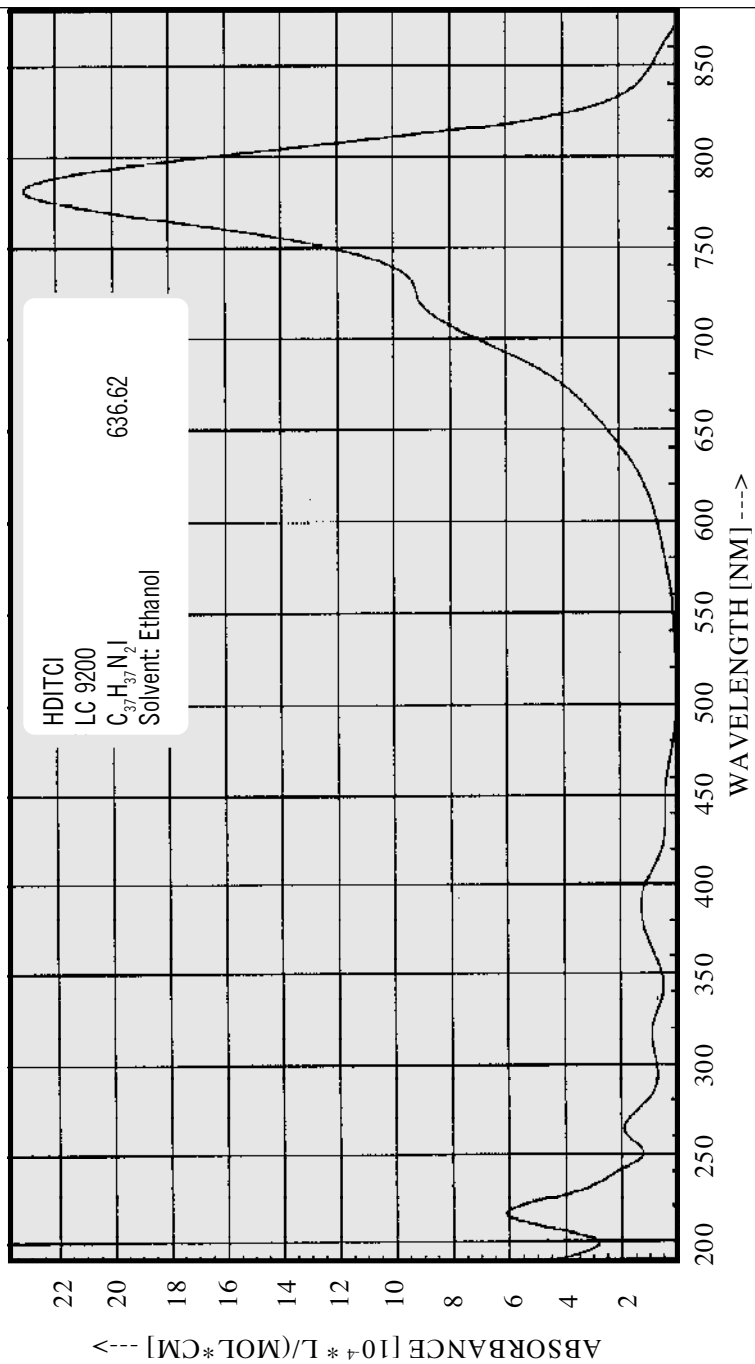
Laser dye for pulsed operation; tunable around 880 nm

Source	Pump		Dye Laser Characteristics				Solvent	Ref.
	Wavelength		Peak	Range	Effic.	Conc.		
	[nm]		[nm]	[nm]	[%]	[g/l]		
Flashlamp	-		880	-	-	0.12	DMSO	1

## References

1. A. Hirth, K. Vollrath, J. Faure, D. Loughnot, *Opt. Commun.* **7(4)**, 339 (1973).

# HDITCI

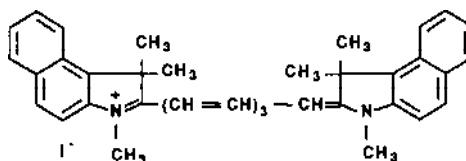


# HDITCI (LC 9200)

## Constitution

1,1',3,3,3',3'-Hexamethyl-4,4',5,5'-dibenzo-2,2'-indotricarbocyanine iodide  
Hexadibenzocyanin 3

$C_{37}H_{37}N_2I$  · MW: 636.62



## Characteristics

Lambdachrome® number: 9200  
CAS registry number: 23178-67-8  
Appearance: bronze colored, crystalline solid  
Absorption maximum (in ethanol): 780 nm  
Molar absorptivity:  $23.1 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum (in chloroform): 824  
For research and development purposes only.

## Lasing Performance

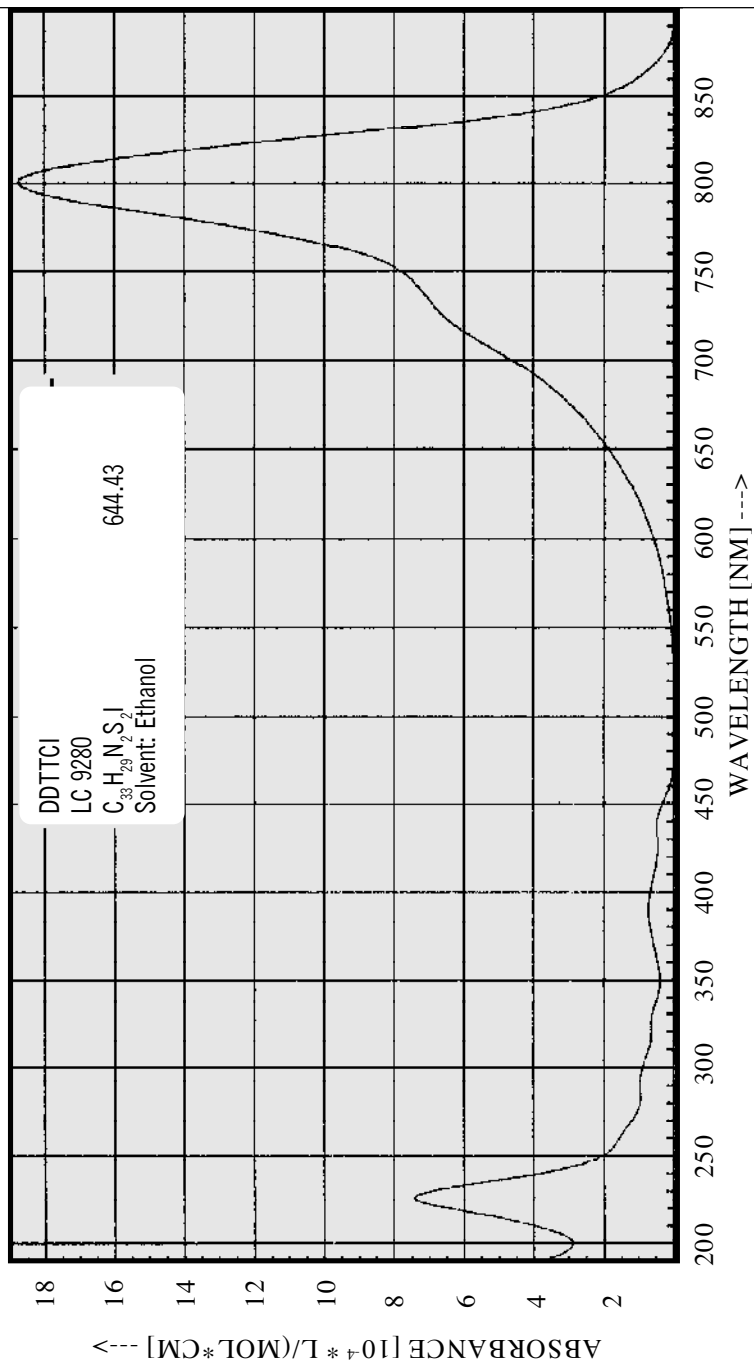
Laser dye for pulsed and CW operation; tunable around 920 nm. Saturable absorber for CW pumped Oxazine 170 dye lasers; applicable around 780 nm <sup>1,2</sup>.

Source	Pump	Dye Laser Characteristics					Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
XeCl-Excimer	308	932	899-975	4	0.90	DMSO	1, 2
CW, Kr <sup>+</sup>	red	920	880 - 960	2	0.95	DMSO/Eg.	3

## References

1. G. W. Fehrenbach et al., *Appl. Phys. Letters* **33**(2), 159 (1978).
2. V. S. Antonov, K. L. Hohla, *Appl. Phys.* **B30**, 109 (1983).
3. K. M. Romanek et al., *Opt. Commun.* **21**(1), 16 (1977).

# DDTTCI

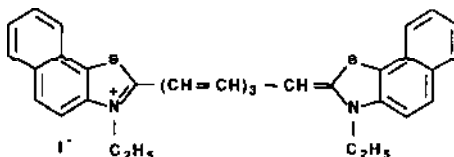


# DDTTCl (LC 9280)

## Constitution

3,3'-Diethyl-4,4',5,5'-dibenzothiatricbocyanine Iodide  
Hexadibenzocycaini 45

$C_{33}H_{29}N_2S_2I$  · MW: 644.43



## Characteristics

Lambdachrome® number:	9280
CAS registry number:	-
Appearance:	bronze colored, crystalline solid
Absorption maximum (in ethanol):	798 nm
Molar absorptivity:	$19.6 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum:	-
For research and development purposes only.	

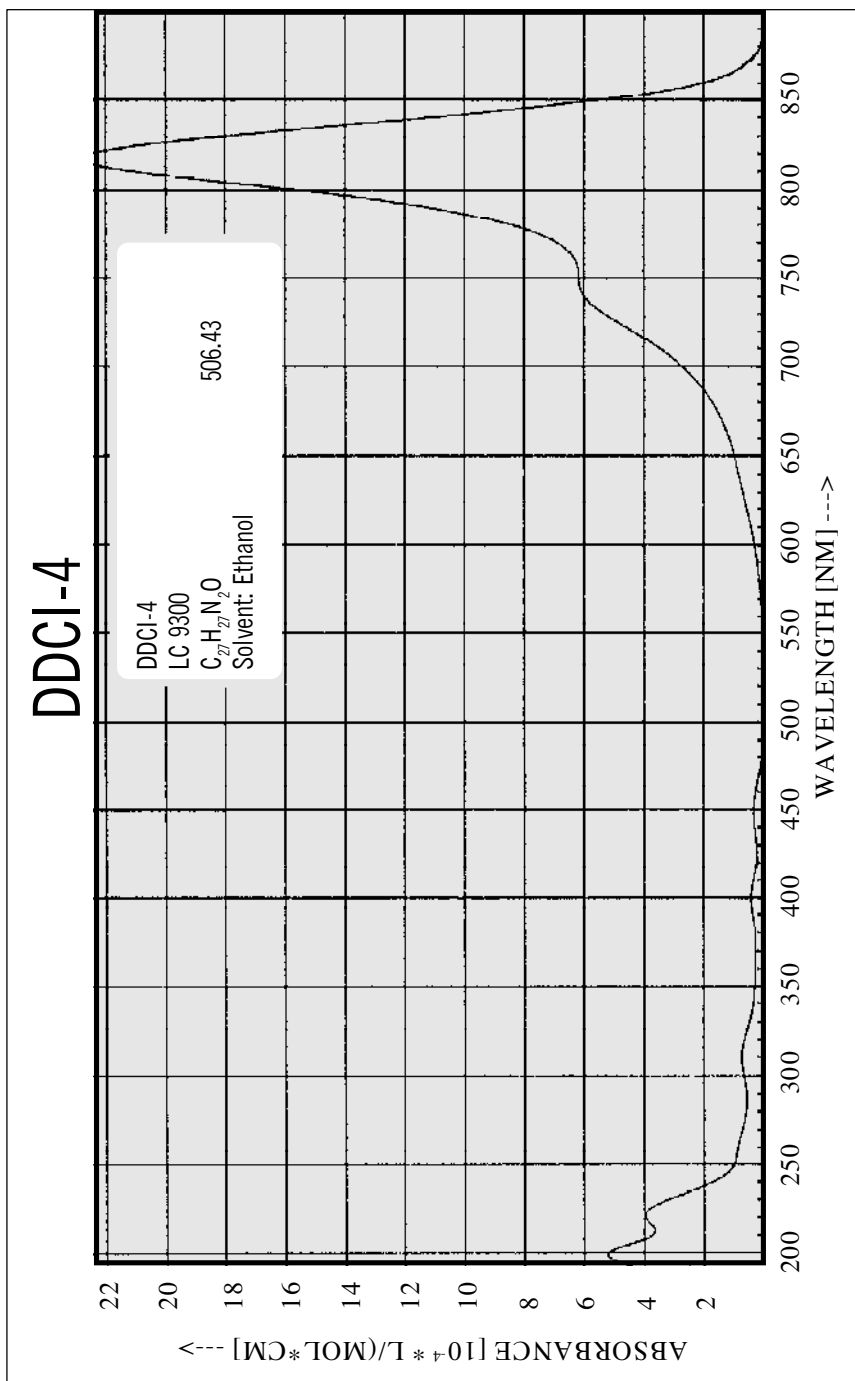
## Lasing Performance

Laser dye for pulsed operation; tunable around 930 nm

<i>Pump</i>		<i>Dye Laser Characteristics</i>				<i>Solvent</i>	<i>Ref.</i>
<i>Source</i>	<i>Wavelength</i> <i>[nm]</i>	<i>Peak</i> <i>[nm]</i>	<i>Range</i> <i>[nm]</i>	<i>Effic.</i> <i>[%]</i>	<i>Conc.</i> <i>[g/l]</i>		
<i>XeCl-Excimer</i>	<i>308</i>	<i>932</i>	<i>899 - 975</i>	<i>5</i>	<i>0.92</i>	<i>DMSO</i>	<i>1, 2</i>
<i>Nitrogen</i>	<i>337</i>	<i>870</i>	<i>-</i>	<i>rel.</i>	<i>0.32</i>	<i>Acetone</i>	<i>3</i>
<i>Flashlamp</i>	<i>-</i>	<i>946</i>	<i>-</i>	<i>-</i>	<i>-</i>	<i>Pc./DMSO</i>	<i>4</i>

## References

1. Lambda Physik.
2. H. Telle, W. Hüffer, D. Basting, *Opt. Commun.* **38**(5,6), 402 (1981).
3. Chinlon Lin, *IEEE J. Quantum Electr.* **QE-11**, 61 (1975).
4. A. Hirth, J. Faure, D. Lougnot, *Opt. Commun.* **8**(4), 318 (1973).



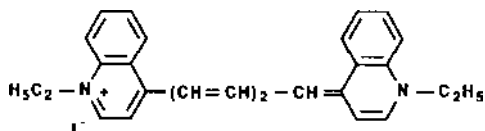


# DDCI-4 (LC 9300)

## Constitution

1,2'-Diethyl-4,4'-dicarbocyanine Iodide  
NK 1144

$C_{27}H_{27}N_2O$  · MW: 506.43



## Characteristics

Lambdachrome® number:	9300
CAS registry number:	-
Appearance:	green, crystalline solid
Absorption maximum (in ethanol):	815 nm
Molar absorptivity:	$23.6 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	850 nm
For research and development purposes only.	

## Lasing Performance

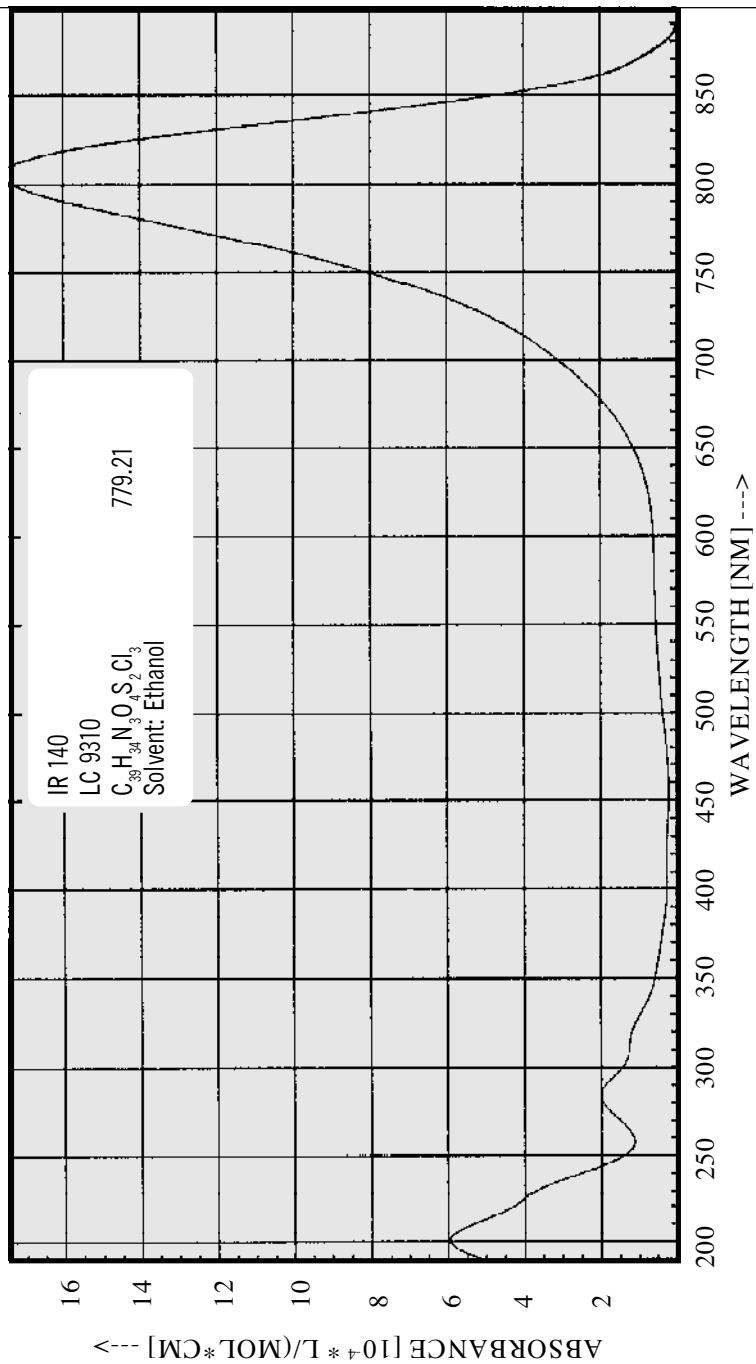
Laser dye for pulsed operation; tunable around 930 nm

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. Conc. [%] [g/l]			
Nitrogen	337	930	-	rel.	0.50	Acetone	1

## References

1. Chinlon Lin, *IEEE J. Quantum Electr.* **QE-11**, 61 (1975).

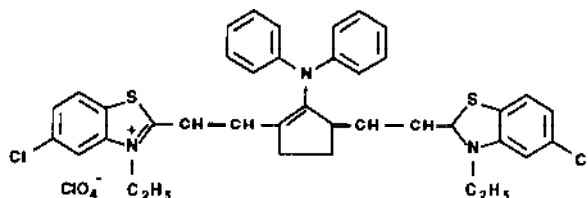
## IR 140



# IR 140 (LC 9310)

## Constitution

$C_{39}H_{34}N_3O_4S_2Cl_3$  · MW: 779.21



## Characteristics

Lambdachrome® number:	9310
CAS registry number:	53655-17-7
Appearance:	brown, crystalline solid
Absorption maximum (in ethanol):	810 nm
Molar absorptivity:	$15.0 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	860 nm
For research and development purposes only.	

## Lasing Performance

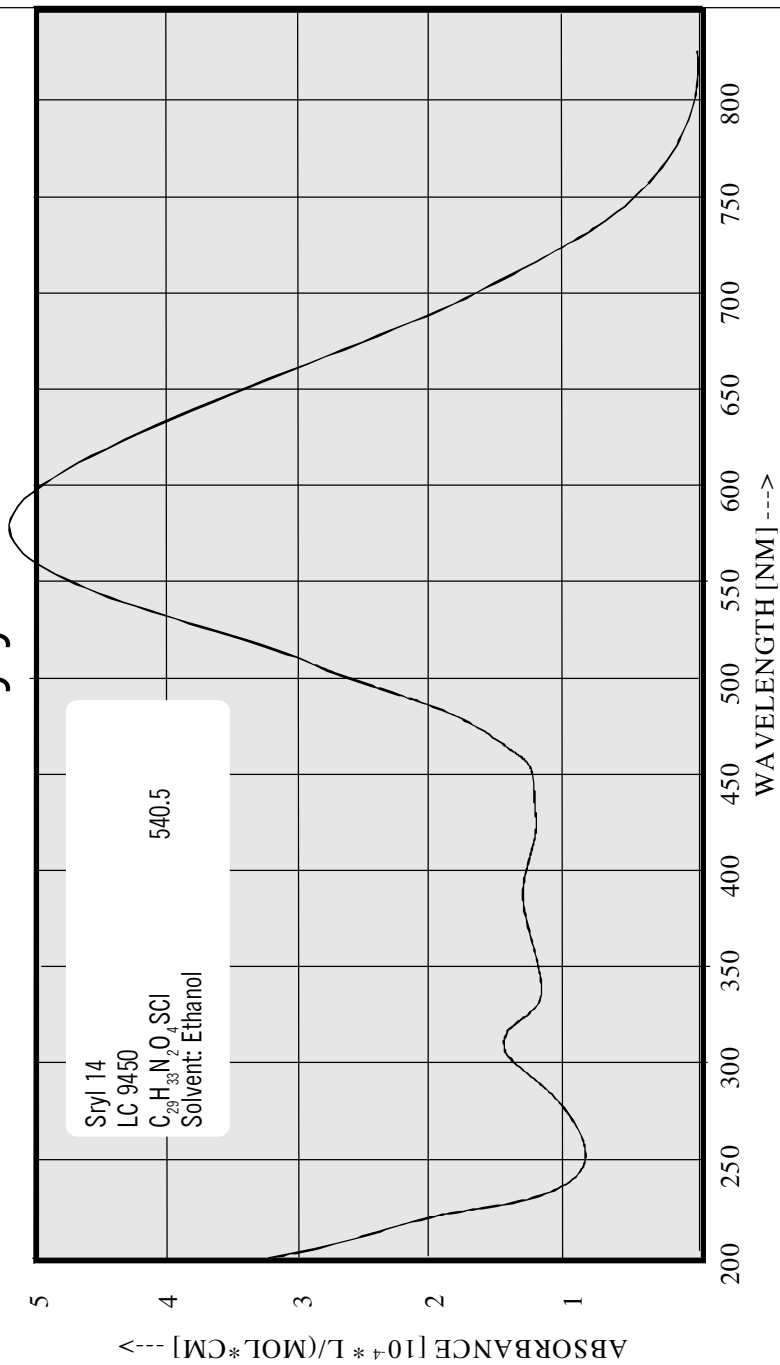
Laser dye for pulsed operation; tunable around 950 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	950	882 - 985	3	1.10	DMSO	1, 2, 3
Nitrogen	337	910	900 - 936	rel.	0.78	DMSO	3, 4
Nd:YAG, 2nd	532	890	-	5	0.31	DMSO	5, 6, 7
Flashlamp	-	950	-	-	0.08	DMSO	8
CW, Kr <sup>+</sup>	VIS	970	880 - 1010	14	0.71	DMSO/Eg.	9, 10

## References

See page 236.

# Styryl 14

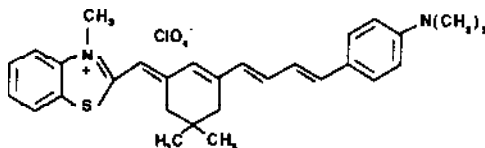


# Styryl 14 (LC 9450)

## Constitution

2-(8-(4-p-Dimethylaminophenyl)-2,4-neopentylene-1,3,5,7-octatetraenyl)-3-methylbenzothiazolium Perchlorate

$C_{29}H_{33}N_2O_4SCl$  · MW: 540.5



## Characteristics

Lambdachrome® number:	9450
CAS Registry number:	-
Appearance:	dark green, crystalline solid
Absorption maximum (in ethanol):	588 nm
Molar absorptivity:	$5.17 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in ethanol):	-
For research and development purposes only.	

## Lasing Performance

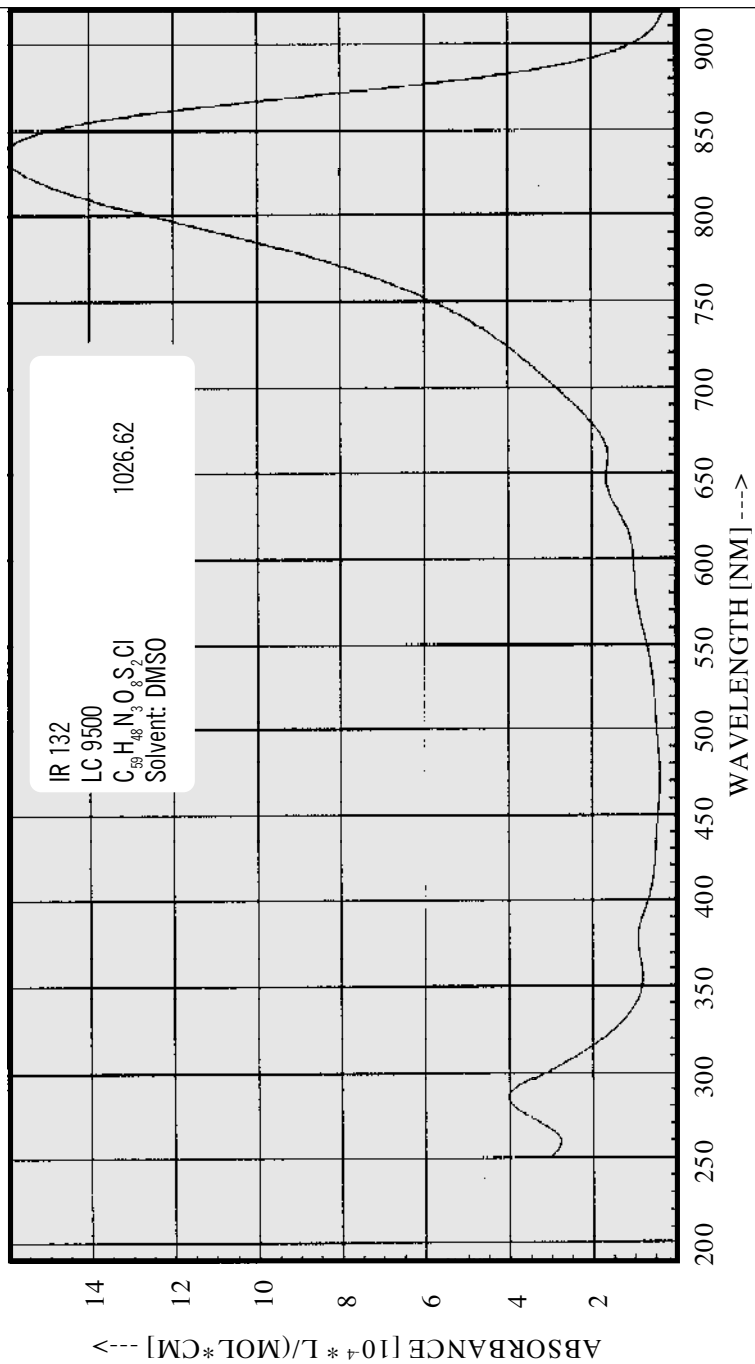
Efficient laser dye for pulsed and CW operation; tunable around 950 nm.

Source	Pump Wavelength [nm]	Dye Laser Characteristics				Solvent	Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
XeCl-Excimer	308	945	904 - 992	9	1.10	DMSO	1
Nd:YAG, 2nd	532	945	904 - 990	9	0.27	Pc.	1

## References

1. Lambda Physik, *Wall Chart* 1996.

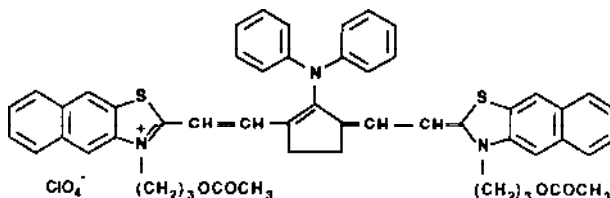
## IR 132



# IR 132 (LC 9500)

## Constitution

$C_{59}H_{48}N_3O_8S_2Cl \cdot MW: 1026.62$



## Characteristics

Lambdachrome® number:	9500
CAS registry number:	62669-62-9
Appearance:	red, crystalline solid
Absorption maximum (in DMSO):	830 nm
Molar absorptivity:	$15.9 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in chloroform):	861 nm
For research and development purposes only.	

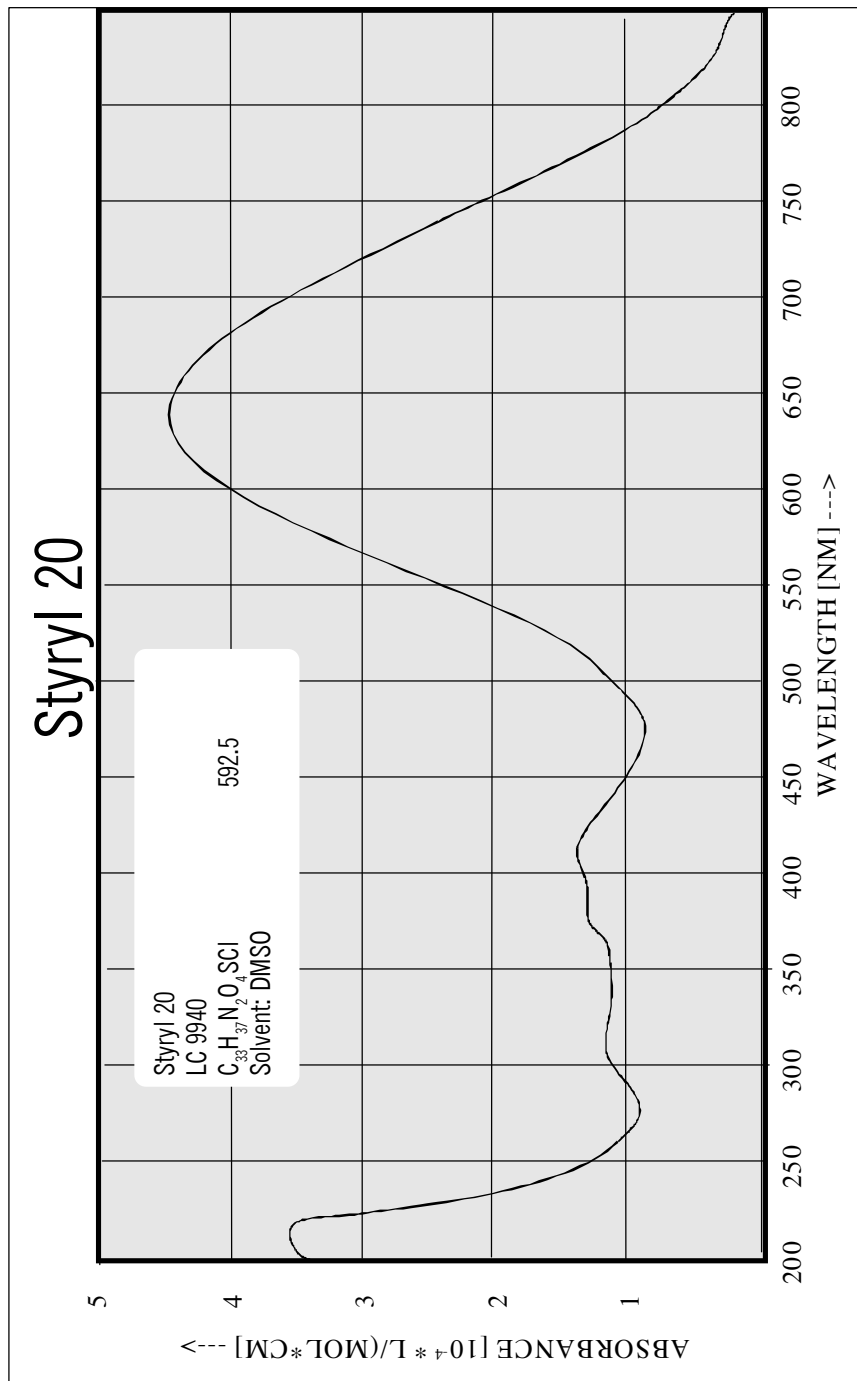
## Lasing Performance

Laser dye for pulsed operation; tunable around 950 nm

Source	Pump Wavelength [nm]	Dye Laser Characteristics					Ref.
		Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]	Solvent	
Nd:YAG, 2nd	532	909	-	1	0.51	DMSO	1
Flashlamp	-	972	-	-	0.10	DMSO	2
ML, Kr <sup>+</sup>	VIS	-	863 - 1048	-	0.80	DMSO	3

## References

1. C. D. Decker, *Appl. Phys. Letters* **27**(11), 607 (1975).
2. J. P. Webb et al., *IEEE J. Quantum Electr.* **QE-11**, 114(1975).
3. M. Leduc, *Opt. Commun.* **31**(1), 66(1979).



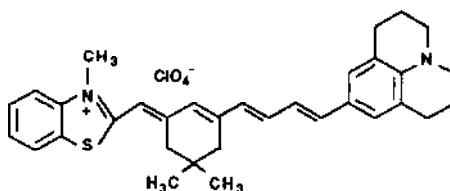


# Styryl 20 (LC 9940)

## Constitution

2-(8-(9-(2,3,6,7-Tetrahydro-1H,5H-benzo(i,j)chinolizinium))-2,4-neopentylene-1,3,5,7-octatetraenyl)-3-methylbenzothiazolium Perchlorate

C<sub>33</sub>H<sub>37</sub>N<sub>2</sub>O<sub>4</sub>SCl · MW: 592.5



## Characteristics

Lambdachrome® number:	9940
CAS registry number:	-
Appearance:	dark green, crystalline solid
Absorption maximum (in ethanol):	645 nm
Molar absorptivity:	4.70 x 10 <sup>4</sup> L mol <sup>-1</sup> cm <sup>-1</sup>
Fluorescence maximum:	-
For research and development purposes only.	

## Lasing Performance

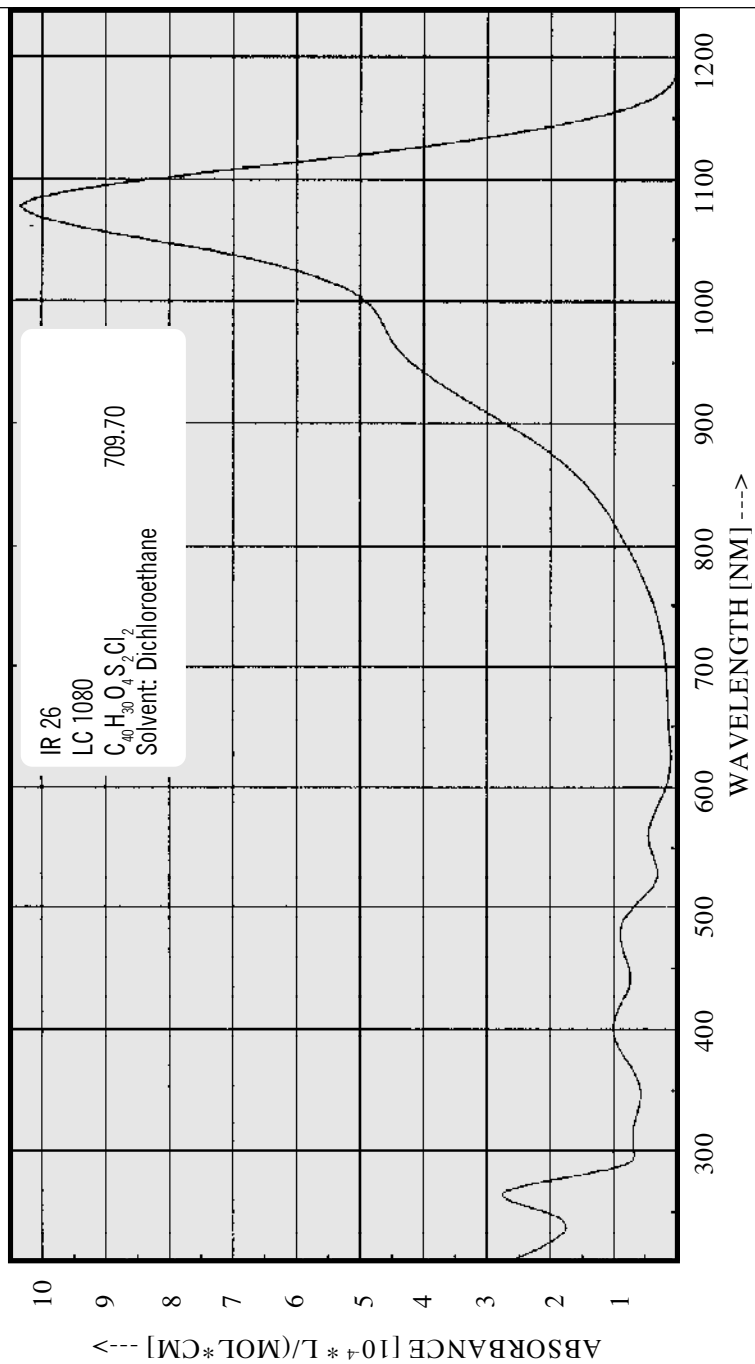
Efficient IR laser dye for pulsed and CW operation; tunable around 990 nm.

Source	Pump		Dye Laser Characteristics					Ref.
	Wavelength		Peak	Range	Effic.	Conc.	Solvent	
	[nm]		[nm]	[nm]	[%]	[g/l]		
XeCl-Excimer	308		994	970 - 1036	4	1.10	DMSO	1
Nd:YAG, 2nd	532		994	970 - 1036	4	0.68	Pc.	1

## References

1. Lambda Physik, *Wall Chart* 1996.

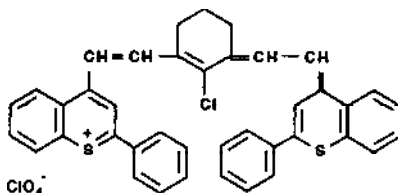
## IR 26



# IR 26 (LC 1080)

## Constitution

$C_{40}H_{30}O_4S_2Cl_2 \cdot MW: 709.70$



## Characteristics

Lambdachrome® number:	1080
CAS registry number:	76871-75-5
Appearance:	dark green, crystalline solid
Absorption maximum (in dichloroethane):	1080 nm
Molar absorptivity:	$10.3 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$
Fluorescence maximum (in benzyl alcohol):	1180 nm
For research and development purposes only.	

## Lasing Performance

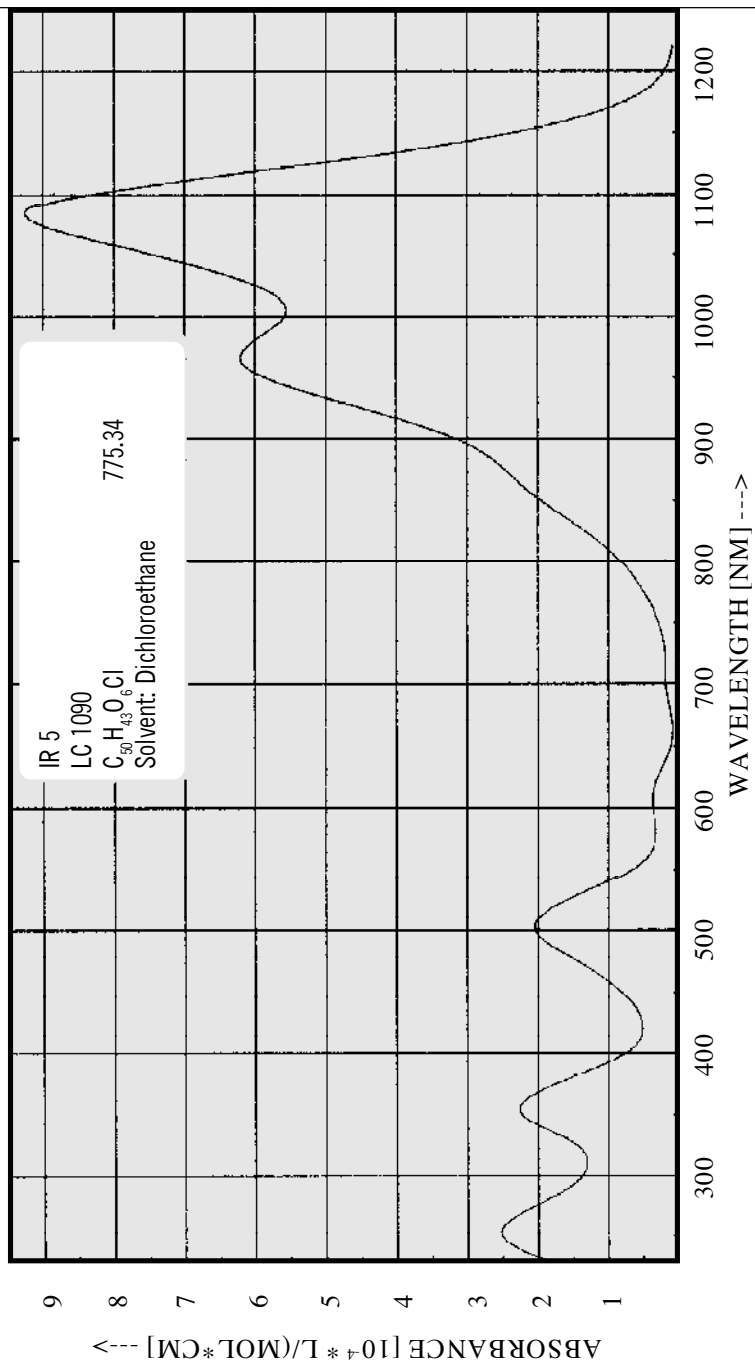
Efficient laser dye for synchronously pumped (Nd:YAG) dye lasers; tunable around 1030 nm. Saturable absorber for Nd:YAG lasers.

Source	Pump	Dye Laser Characteristics					Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]			
Nd:YAG	1064	1290	1200 - 1320	4	0.71	Bz.	1, 2, 3, 4	

## References

1. W. Kranitzky et al., *Opt. Commun.* **36**(2), 149 (1981).
2. A. Seilmeier et al., *Optics. Letters* **8**(4), 205 (1983).
3. A. Seilmeier, *Opt. Quantum Electr.* **16**, 89 (1984).
4. K. Kato, *IEEE J. Quantum Electr.* **QE-20**(7), 698 (1984).

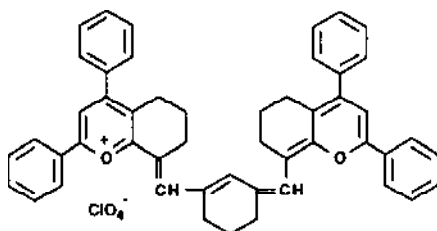
## IR 5



# IR 5 (LC 1090)

## Constitution

$C_{50}H_{43}O_6Cl$  · MW: 775.34



## Characteristics

Lambdachrome® number: 1090  
CAS registry number: 61010-01-3  
Appearance: dark green, crystalline solid  
Absorption maximum (in dichloroethane): 1090 nm  
Molar absorptivity:  $9.30 \times 10^4 \text{ L mol}^{-1} \text{ cm}^{-1}$   
Fluorescence maximum: -  
For research and development purposes only.

## Lasing Performance

Laser dye for synchronously pumped (Nd:YAG) dye lasers; tunable around 1300 nm. Saturable absorber for Nd:glass lasers<sup>2)</sup>.

Source	Pump	Dye Laser Characteristics				Solvent	Ref.
	Wavelength [nm]	Peak [nm]	Range [nm]	Effic. [%]	Conc. [g/l]		
Nd:YAG	1064	1320	1180 - 1400	10	0.64	DCE	1, 2,

## References

1. T. Elsaesser et al., *IEEE J. Quantum Electr.* QE-20(3), 191 (1984).
2. R. R. Alfano et al., *IEEE J. Quantum Electr.* QE-17(3), 290 (1981).

# Saturable Absorbers

a: Dye/LC #

b: Application

c: Literature

a: DASPI/LC 4660

b: s.a. for FL-pumped C1-, C102-, C466-, C6H-dye laser  
applicable in the 480 - 500 nm range

c: Sibbett et al., *Opt. Commun.* 46(1), 32 (1983)

a: DASBTI/LC 5280

b: s.a. for FL-pumped C6-, C522-, C153-, Rh110-dye laser  
applicable in the 500 - 540 nm range

c: W. Sibbett et al., *Opt. Commun.* 44(2), 121 (1982)

W. Sibbett et al., *Appl. Phys.* B29, 191 (1982)

W. Sibbett et al., *IEEE J. Quantum Electr.* QE-19(4), 558 (1983)

a: DOCI/LC 5410

b: s.a. for FL-pumped C102-, C500-, XeCl-pumped C102-dye laser  
applicable in the 480 - 500 nm range

c: J.C. Mialocq et al., *Appl. Phys. Lett.* 33(9), 819 (1978)

R. Wyatt, *Opt. Commun.* 38(1), 64 (1981)

Th. Varghese, *Opt. Commun.* 44(5), 353 (1983)

M. Watanabe et al., *Appl. Phys. Lett.* 45(9), 929 (1984)

a: DMETCI/LC 5460

b: s.a. for FL-pumped C153-dye laser  
applicable in the 530 - 550 nm range

c: W. Sibbett et al., *Opt. Commun.* 43(1), 50 (1982)

a: DQOCI/LC 5920

b: s.a. for FL-pumped Fluorol 7GA-, Rh6G-dye laser  
applicable in the 550 - 590 nm range

c: E. Lill et al., *Opt. Commun.* 20(2), 223 (1977)

R.S. Adrain et al., *Opt. Commun.* 12(2), 140 (1974)

a: DCI-2/LC 5950

b: s.a. for FL-pumped Rh6G-dye laser  
applicable in the 560 - 600 nm range

c: M. Maeda et al., *Japan J. Appl. Phys.* 13(1), 193 (1974)

- a: Malachit Green/LC 6220
- b: additive for CW-pumped Rh6G-dye laser  
applicable in combination with DODCI as s.a.
- c: M. Young, *Appl. Optics* 18(19), 3212 (1979)  
A. Watanabe et al., *IEEE J. Quant. Electr.* QE-19(4), 533 (1983)
  
- a: DTCI/LC 6250
- b: s.a. for FL-pumped Rh6G-dye laser  
applicable in the 580 nm range
- c: M. Maeda et al., *J. Appl. Phys.* 13(1), 193 (1974)
  
- a: DQTCI/LC 6290
- b: s.a. for FL-pumped RhB-dye laser  
applicable in the 600 - 620 nm range
- c: E. G. Arthurs et al., *Appl. Phys. Lett.* 20(3), 125 (1972)  
E. Lill et al., *Opt. Commun.* 22(1), 107 (1977)
  
- a: DODCI/LC 6550
- b: "state of the art"-s.a. for CW- and FL-pumped Rh6G-dye laser  
applicable in the 570 - 600 nm range
- c: elsewhere
  
- a: DTDCI/LC 7260
- b: s.a. for FL-pumped Rh101-, CV/Rh6G-dye laser  
applicable in the 630 - 650 nm range
- c: J. Negran et al., *Appl. Optics* 17(17), 2812 (1978)  
E.G. Arthurs et al., *Appl. Phys. Lett.* 20(3), 125 (1972)
  
- a: Cryptocyanine/LC 7450
- b: s.a. for Ruby-lasers
- c: I.K. Krasnyuk et al., *JETP Letters* 7(4), 89 (1968)  
H.W. Mocker et al., *Appl Phys. Lett.* 7(10), 270 (1965)  
V.I. Malyshev et al., *JETP Letters* 6(2), 34 (1967)
  
- a: DDI/LC 7700
- b: s.a. for FL-pumped CV/Rh6G-, Rh700-dye laser, Ruby-laser  
applicable in the 680 - 710 nm range
- c: E.G. Arthurs et al., *Appl. Phys. Lett.* 20(3), 125 (1972)  
M.E. Mack, *IEEE J. Quant. Electr.* QE-4, 1015 (1968)  
W. Sibbett et al., *IEEE J. Quant. Electr.* QE-20(2), 108 (1984)
  
- a: DOTCI/LC 7880
- b: s.a. for FL-pumped CV-dye laser  
applicable in the 680 - 700 nm range
- c: E.G. Arthurs et al., *Appl. Phys. Lett.* 20(3), 125 (1972)

- a: HITCI/LC 8500
- b: s.a. for FL-pumped DOTCI-dye laser  
applicable in the 750 - 800 nm range
- c: A. Hirth et al., *Opt. Commun.* 7(4), 339 (1973)
  
- a: HDITCI/LC 9200
- b: s.a. for CW-pumped Oxazine 750-dye laser  
applicable in the 750 - 830 nm range
- c: G.W. Fehrenbach et al., *Appl. Phys. Lett.* 33(2), 159 (1978)
  
- a: IR 140/LC 9310
- b: s.a. for FL- and CW-pumped Styryl 9-dye laser  
applicable in the 840 nm range
- c: K. Smith et al., *Opt. Commun.* 49(5), 359 (1984)
  
- a: IR 26/LC 1080
- b: s.a. for Nd:YAG-laser
- c: B. Kopainsky et al., *Appl. Phys.* B-29, 15 (1982)
  
- a: IR 5/LC 1090
- b: s.a. for Nd:glass-laser
- c: R.R. Alfano et al., *IEEE J. Quant. Electr.* QE-17(3), 290 (1981)

Abbreviations used:

- C Coumarin
- CV Cresyl Violet
- CW continuous wave
- FL flashlamp
- LC Lambdachrome®
- Rh Rhodamine
- s.a. saturable absorber



# Reference List

LC No.	Lambdachrome® Dye	corresponds to
3300	BM-Terphenyl	DMT
3400	PTP	p-Terphenyl
3500	TMQ	TMQ
3570	BMQ	-
3590	DMQ	-
3600	Butyl-PBD	BPBD-365
3640	PBD	PBD
3650	TMI	-
3690	QUI	-
3700	PPO	PPO
3720	PPF	-
3740	PQP	p-Quaterphenyl
3780	BBD	-
3800	Polyphenyl 1	-
3810	Polyphenyl 2	-
3860	BiBuQ	BBQ
3900	Quinolon 390	LD 390
3950	a-NPO	a-NPO
3990	Furan 2	-
4000	PBBO	PBBO
4090	DPS	DPS
4100	Stilbene 1	-
4150	BBO	BBO
4200	Stilbene 3	Stilbene 420
4220	Carbostyryl 7	Carbostyryl 124
4230	POPOP	POPOP
4240	Coumarin 4	Umbelliferon 47
4250	Bis-MSB	Bis-MSB
4300	BBOT	-
4350	Carbostyryl 3	Carbostyryl 165
4400	Coumarin 120	Coumarin 440
4500	Coumarin 2	Coumarin 450
4650	DASPI	-
4660	Coumarin 466	LD 466, C1H

<b>LC No.</b>	<b>Lambdachrome® Dye</b>	<b>corresponds to</b>
4700	Coumarin 47	Coumarin 460, Coumarin 1
4800	Coumarin 102	Coumarin 480
4810	Coumarin 152A	Coumarin 481, Coumarin 35
4850	Coumarin 152	Coumarin 485
4900	Coumarin 151	Coumarin
4910	Coumarin 6H	LD 490
5000	Coumarin 307	Coumarin 503
5010	Coumarin 500	Coumarin 500
5040	Coumarin 314	Coumarin 504
5100	Coumarin 510	Coumarin 510
5150	Coumarin 30	Coumarin 515
5210	Coumarin 334	Coumarin 521
5220	Coumarin 522	Coumarin 522
5280	DASBTI	-
5350	Coumarin 7	Coumarin 535
5360	Brillant Sulfaflavine	Brillant Sulfaflavin
5370	Coumarin 6	Coumarin 540
5400	Coumarin 153	Coumarin 540A
5410	DOCI	DOC, NK 85
5460	DMETCI	-
5520	Uranin	Disodium Fluorescein
5530	Fluorescein 27	Fluorescein 548
5700	Rhodamine 110	Rhodamine 560
5750	Rhodamine 19	Rhodamine 575
5900	Rhodamine 6G	Rhodamine 590
5901	Rhodamine 6G (Perchl.)	Rhodamine 590 (Perchl.)
5920	DQOCI	-
5950	DCI-2	-
6100	Rhodamine B	Rhodamine 610
6101	Rhodamine B (Perchl.)	Rhodamine 610 (Perchl.)
6200	Sulforhodamine B	Kiton Red 620
6220	Malachit Green	-
6250	DTCI	NK 76
6290	DQTCI	-
6400	Rhodamine 101	Rhodamine 640
6500	DCM	DCM

<b>LC No.</b>	<b>Lambdachrome® Dye</b>	<b>corresponds to</b>
6501	DCM-spec.	-
6550	DODCI	DODC-Iodide
6600	Sulforhodamine 101	Sulforhodamine 640
6700	Cresyl Violet	Cresyl Violet 670
6900	Nile Blue	Nile Blue 690
6950	Oxazine 4	LD 690
7000	Rhodamine 700	LD 700
7100	Pyridin 1	LDS 698
7210	Oxazine 170	Oxazine 720
7250	Oxazine 1	Oxazine 725
7260	DTDCI	DTDC-Iodide, NK 136
7270	Oxazine 750	Oxazine 750
7300	Pyridin 2	LDS 722
7400	HIDCI	Hexacyanine 2, NK 529
7500	Styryl 6	LDS 730
7550	Styryl 8	LDS 751
7700	DDI	NK 1456
7710	Pyridin 4	-
7800	Methyl-DOTCI	DMOTC-Iodide, NK 199
7880	DOTCI	DEOTC-Iodide
7950	Styryl 11	LDS 798
8000	Rhodamine 800	-
8410	Styryl 9 (M)	LDS 820 (821)
8500	HITCI	Hexacyanine 3, NK 125
8630	IR 125	IR 125
8760	DTTCI	DTTC-Iodide, NK 126
8800	IR 144	IR 144
8810	Styryl 15	-
8850	DNTTCI	-
9300	DDCI-4	NK 1144
9310	IR 140	IR 140
9500	IR 132	IR 132
9940	Styryl 20	-
10600	IR 25	-
10800	IR 26	IR 26
10810	IR 26 (HFB)	-
10900	IR 5	IR 5

# Tuning Curves

Tuning curves of excimer-, nitrogen-, Nd:YAG-, and CW-laser pumped dye lasers are shown on the following pages. Parameters given are defined as follows:

## **Peak**

Output maximum of the tuning curve, in nanometers.

## **Tuning Range**

The tuning range is defined as the range, in nanometers, giving an efficiency larger than 10 percent of the maximum. All tuning ranges are restricted to broadband operation.

## **Efficiency**

Dye laser output at the maximum of the tuning range relative to pump laser input, in percent. Efficiency may change in other configurations or pump power levels.

## **Pump Wavelength**

Pump wavelength used in nanometers.

## **Solvent**

BZ=Benzyl Alcohol, EG=Ethylene Glycol, CH=Cyclohexane, DI=Dioxane, ME=Methanol, DMSO=Dimethylsulfoxide, PC=Propylene Carbonate.

## **Concentration**

Amount of dye, in grams, for 1 liter stock solution. There is an optimum concentration for a given dye, wavelength and input power. In case of continuously pumped dye lasers this optimum concentration generally lies between 60 percent and 80 percent absorption of the pump power, and in the case of transversally pumped pulsed dye lasers, at 99 percent absorption of the pump energy within 1 millimeters of the dye solution. Higher concentration causes the tuning curves to be shifted slightly to the red, while lower concentrations will result in blue shift. Optimization of the dye concentration is accomplished by adding either pure solvent or a solution of higher concentration than that recommended to the solution in the dye circulation system, until optimum power is at maximum.

### Stability

The accumulated pump energy  $t$ , in Wh, causing a decrease in dye laser output to 50 percent of the initial value for 1 liter dye solution (resp. 1 liter amplifier solution in the case of excimer pumped dye laser), measured at 10 Hz. The following classification has been used in the case of excimer laser pumped dye lasers: -:  $\tau$  10-30 Wh, +:  $\tau$  30-50 Wh, ++:  $\tau$  > 50 Wh. In the case of Nd:YAG-pumped lasers: -:  $\tau$  < 50Wh, +:  $\tau$  50-100 Wh, ++:  $\tau$  100-300 Wh, +++:  $\tau$  > 300 Wh. The stability of dyes being continuously pumped is classified as: -:  $\tau$  < 100 Wh, +:  $\tau$  100-500Wh, ++:  $\tau$  > 500 Wh.

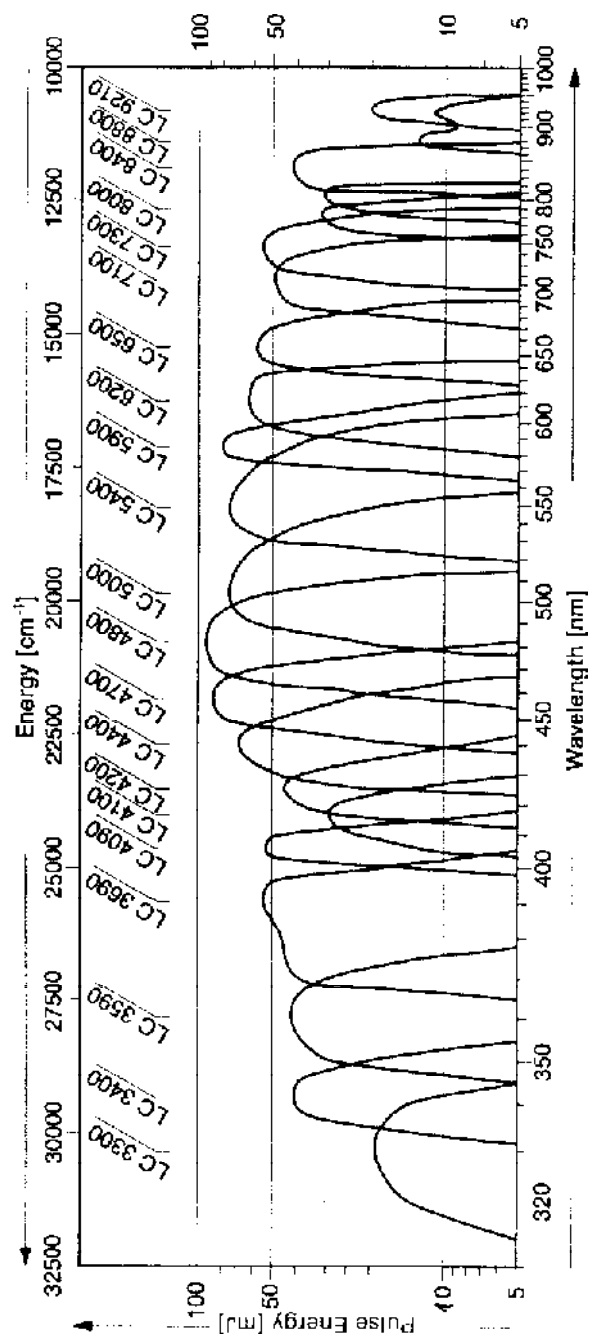
### General Remarks

The output power of dye lasers is strongly dependent on the quality of the dye used. To overcome reduced quantum efficiency and instability due to impurities, Lambdachrome® laser dyes are examined by experienced chemist for their chemical and spectral properties and purified by specially developed techniques.

Lambda Physik reserves the right to modify any information given herein. Every effort is made to ensure utmost accuracy; no liability, however, is assumed for errors occurring. Nothing here is to be constructed as recommending any practice or any product in violation of any patent.

Cautious handling of dyes and dye solutions is advised, since the exact toxicity in most cases is not well known. The responsibility for the safe use of our laser dyes must rest in all cases with the user.

# Dyes for EXCIMER LASER Pumped Dye Lasers



Lambdachrome Dye	Peak (nm)	Tuning Range (nm)	Effic. (%)	Pump Source Wavelength (nm)	Char. Energy (mJ)	Solvent	Conc. (g/l)	Stability*
LC 3300	334	312-343	4	248	150	CH	0.50	-
LC 3400	343	332-360	8	308	400	DI	0.24	+
LC 3590	360	346-377	9	308	400	DI	0.23	++
LC 3600	363	356-385	5	308	400	DI	0.30	++
LC 3690	390	368-402	11	308	400	DI	0.20	++
LC 3800	381	363-408	12	308	400	EG	0.20	++
LC 3860	388	367-405	11	308	400	DI	0.25	++
LC 3810	397	386-418	10	308	400	EG	0.25	+
LC 3990	399	388-426	8	308	400	ME	0.50	+
LC 4000	396	386-420	7	308	400	DI	0.40	++
LC 4090	406	399-415	11	308	400	DI	0.25	-
LC 4200	425	412-443	9	308	400	ME	0.65	-
LC 4400	441	423-462	15	308	400	ME	0.82	+
LC 4500	448	432-475	15	308	400	ME	1.50	+
LC 4700	456	440-484	18	308	400	ME	1.59	+
LC 4800	480	460-510	18	308	400	ME	2.30	+
LC 5000	500	479-553	16	308	400	ME	3.40	-
LC 5400	540	522-600	15	308	400	ME	4.20	++
LC 5900	581	569-608	16	308	400	ME	1.20	+
LC 6100	600	588-644	12	308	400	ME	0.91	+
LC 6400	623	614-672	12	308	400	ME	0.75	-
LC 6500	658	632-690	12	308	400	DMSO	0.71	-

\*Values are as follows: -:  $\tau$  10-30 Wh, +:  $\tau$  30-50 Wh, ++:  $\tau$  >50 Wh

## Dyes for EXCIMER LASER Pumped Dye Lasers (cont.)

Lambdachrome Dye	Peak (nm)	Tuning Range (nm)	Effic. (%)	Pump Source Wavelength (nm)	Char. Energy (mJ)	Solvent	Conc. (g/l)	Stability*
LC 7000	Rhodamine 700	723	11	308	400	ME	0.85	-
LC 7100	Pyridine 1	710	10	308	400	DMSO	0.84	++
LC 7300	Pyridine 2	740	11	308	400	DMSO	0.72	+
LC 7710	Pyridine 4	771	7	308	400	DMSO	0.75	++
LC 8000	Rhodamine 800	810	6	308	400	DMSO	1.00	+
LC 8400	Styryl 9	840	9	308	400	DMSO	1.10	-
LC 8500	HiTCI	868	4	308	400	DMSO	1.20	-
LC 8810	Styryl 15	880	7	308	400	DMSO	1.15	-
LC 9210	IR125	920	4	308	400	DMSO	2.00	-
LC 9450	Styryl 14	945	9	308	400	DMSO	1.10	-
LC 9940	Styryl 20	994	4	308	400	DMSO	1.10	-

\*Values are as follows: -:  $\tau$  10-30 Wh, +:  $\tau$  30-50 Wh, ++:  $\tau$  >50 Wh



# Dyes for Nd:YAG LASER Pumped Dye Lasers

Lambdachrome Dye	Peak (nm)	Tuning Range (nm)	Effic. (%)	Pump geometry	Pump Wavelength (nm)	Energy (mJ)	Solvent	Conc. (g/l)	Stability**
LC 3900 Quinolon 390	390	384-394	4	transv.	355	200	ME	0.25	-
LC 3990 Furan 2	402	392-422	15		355	200	ME	0.50	+
LC 4260 Furan 1	421	410-435	10		355	200	ME	0.26	+
LC 4200 Stilbene 3	428	415-439	15	transv	355	200	ME	0.25	-
LC 4400 Coumarin 120	440	420-470	16	transv.	355	200	ME	0.25	+
LC 4700 Coumarin 47	460	444-476	15	transv	355	200	ME	0.30	+
LC 4800 Coumarin 102	480	462-497	15	transv	355	200	ME	0.40	+
LC 5000 Coumarin 307	508	485-546	15	transv	355	200	ME	0.70	-
LC 5100 Coumarin 500	518	498-546	10		355	200	ME	0.70	-
LC 5400 Coumarin 153	540	516-575	18		355	200	ME	2.36	+
LC 5530 Fluorescein 27	550	540-575	28	long.	532	200	ME	0.64	++
LC 5750 Rhodamine 19	567	556-586	31	long.	532	200	ME	0.22	+++
LC 5900 Rhodamine 6G	566	555-585	32	long.	532	200	ME	0.10	+++
LC 6100 Rhodamine B	594	584-619	29	long.	532	200	ME	0.22	+++
LC 6200 Sulforhodamine B	588	579-600	29	long.	532	200	ME	0.27	++
LC 6400 Rhodamine 101	621	611-662	26	long.	532	200	ME	0.50	++
LC 6600 Sulforhodamine 101	628	619-673	15	long.	532	200	PC	0.30	++
LC 6500 DCM	639	615-666	27	long.	532	200	PC	0.50	++
LC 7100 Pyridine I	697	667-736	32	transv	532	200	PC	0.36	+++
LC 7300 Pyridine 2	750	725-776	21		532	200	PC	0.22	+++

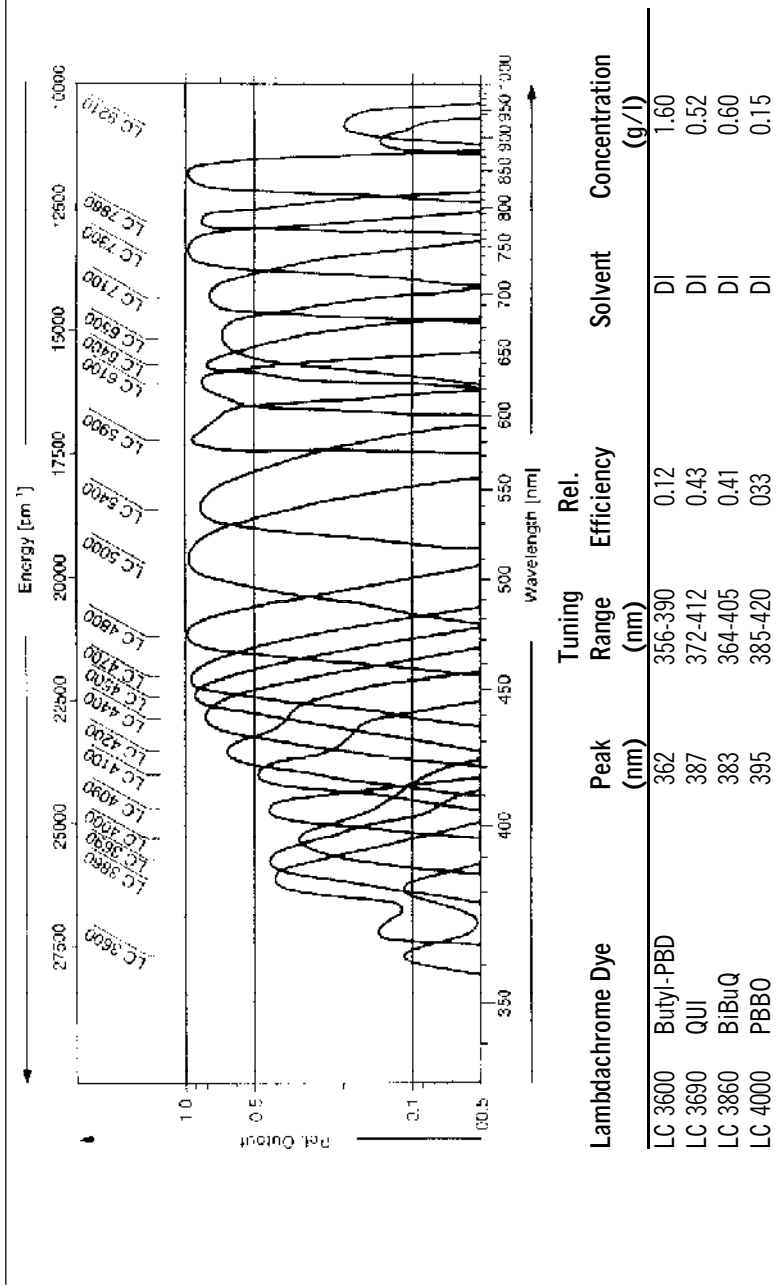
\*\*Values are as follows: -:  $\tau < 50$ , =:  $\tau 50$ -100 Wh, ++:  $\tau 100$ -300 Wh, +++:  $\tau > 300$ Wh

## Dyes for Nd:YAG LASER Pumped Dye Lasers (cont.)

Lamdbdachrome Dye	Peak (nm)	Tuning Range (nm)	Effic. (%)	Pump geometry	Pump Source Char. Wavelength (nm)	Energy (mJ)	Solvent	Conc. (g/l)	Stability**
LC 7500 Styryl 6	721	708-735	16		532	200	PC	0.28	++
LC 7600 Styryl 7	720	701-749	16	transv	532	200	ME	0.12	+++
LC 7550 Styryl 8	750	717-780	13	transv.	532	200	ME	0.15	+++
LC 8400 Styryl 9M	824	797-851	15	transv.	532	200	PC	0.26	++
LC 8810 Styryl 15	880	856-918	7		532	200	PC	0.62	++
LC 9450 Styryl 14	945	904-990	9		532	200	PC	0.27	++
LC 9940 Styryl 20	994	970-1036	4		532	200	PC	0.68	++

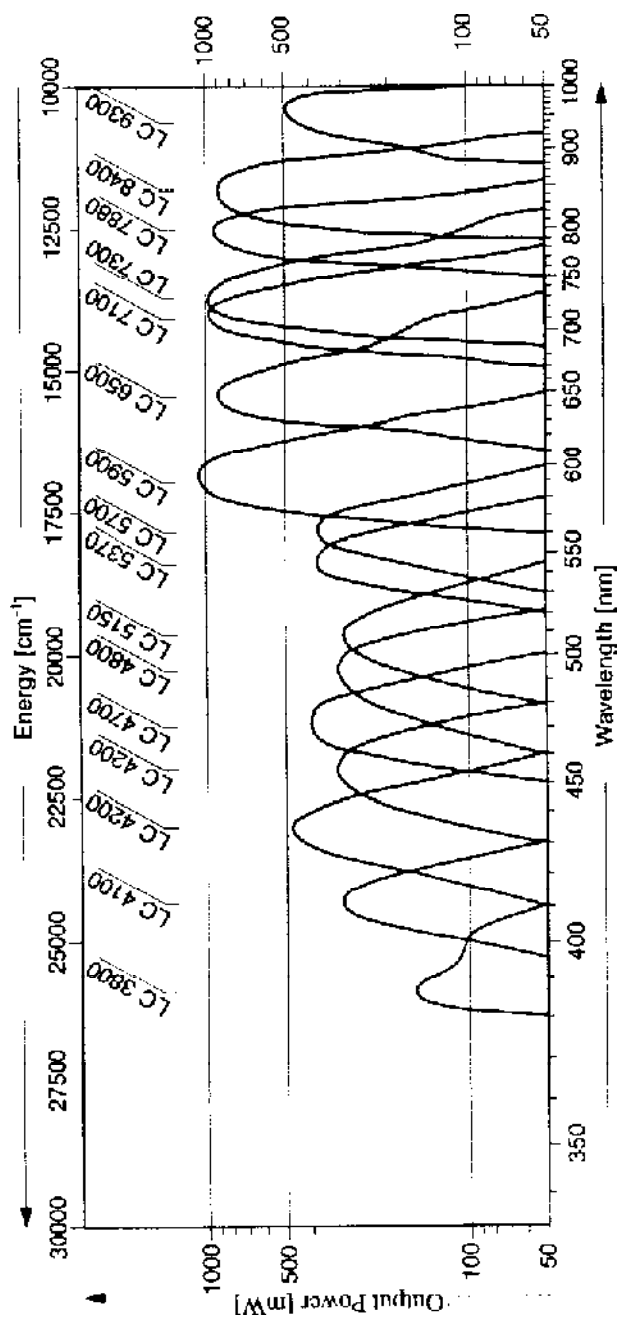
\*\*Values are as follows: -:  $\tau < 50$ , =:  $\tau$  50-100 Wh, ++:  $\tau$  100-300 Wh, +++:  $\tau > 300$ Wh

# Dyes for NITROGEN LASER Pumped Dye Lasers



Lambdachrome Dye	Peak (nm)	Tuning Range (nm)	Rel. Efficiency	Solvent	Concentration (g/l)
LC 4090 DPS	404	394-416	0.43	DI	0.12
LC 4100 Stilbene 1	417	405-446	0.49	EG	0.20
LC 4200 Stilbene 3	424	408-457	0.66	ME	0.22
LC 4250 Bis-MSB	421	412-435	0.59	DI	0.14
LC 4400 Coumarin 120	438	418-465	0.83	ME	0.25
LC 4500 Coumarin 2	444	426-475	0.94	ME	0.40
LC 4700 Coumarin 47	453	436-486	0.95	ME	0.66
LC 4800 Coumarin 102	470	454-506	1.00	ME	1.44
LC 5000 Coumarin 307	504	478-547	1.00	ME	1.60
LC 5400 Coumarin 153	537	517-590	0.87	ME	3.10
LC 5900 Rhodamine 6G	581	573-618	0.93	ME	1.63
LC 6100 Rhodamine B	622	600-646	0.91	ME	2.85
LC 6200 Sulforhodamine B	622	600-646	0.91	ME	2.85
LC 6400 Rhodamine 101	648	623-676	0.82	ME	2.36
LC 6500 DCM	659	626-703	0.69	DMSO	0.50
LC 7100 Pyridine 1	703	675-750	0.78	DMSO	0.88
LC 7210 Oxazine 170	705	672-727	0.35	ME	0.79
LC 7300 Pyridine 2	743	710-790	1.00	DMSO	0.85
LC 7800 Methyl-DOTC	780	768-820	0.86	DMSO	0.51
LC 7880 DOTC/HITC	823	794-867	0.74	DMSO	1.23/0.03
LC 8400 Styryl 9	840	803-875	1.00	DMSO	1.03
LC 8760 DTTC/IR 144	871	859-886	0.18	DMSO	0.65/2.52
LC 8800 IR 144/IR 125	887	872-935	0.14	DMSO	2.52/1.94
LC 9210 IR 125	918	893-958	0.21	DMSO	1.94
LC 9301 IR 140	910	900-963	0.11	DMSO	0.78

# Dyes for ION LASER Pumped Dye Lasers



LambdaChrom Dye	Peak (nm)	Tuning Range (nm)	Pump source char. Wavelength (nm)	Power (W)	SolCent	Conc. Stability*** (g/l)
LC 3810 Polyphenyl 2	384	370-406	Ar+, UV, 300-336nm	2.0	EG	2.0 +
LC 4100 Stilbene 1	415	403-428	Ar+, all lines UV	3.0	EG	0.75 +
LC 4200 Stilbene 3	435	410-485	Ar+, all lines UV	5.0	EG	1.0 +
LC 4800 Coumarin 102	482	463-515	Ar+, UV, 350-386nm	3.0	BZ/EG	2.0
LC 5370 Coumarin. 6	535	510-550	Ar+, 488 nm	6.0	BZ	2.0 +
LC 5700 Rhodamine 110	550	530-580	Ar+, 514.5 nm	6.0	EG	0.75 +
LC 5900 Rhodamine 6G	575	560-625	Ar+, 514.5 nm	6.0	EG	0.75 ++
LC 6200 Sulforhodamine B	625	598-650	Ar+, 514.5 nm	6.0	EG	2.5 ++
LC 6500 DCM Special	645	610-695	Ar+, 514.5 nm	6.0	BZ/EG	2.0 +
LC 7000 Rhodamine 700	740	690-785	Kr+, all lines red	4.6	EG	1.0 ++
LC 7300 Pyridine 2	720	675-783	Ar+, 514.5 nm	7.5	PC/EG	1.5 ++
LC 8400 Styryl 9	830	785-900	Ar+, 514.5 nm	6.0	PC/EG	2.0 +

\*\*\*Valeus are as follows: -:  $\tau < 100$  Wh, +:  $\tau$  100-500 Wh, ++:  $\tau > 500$  Wh