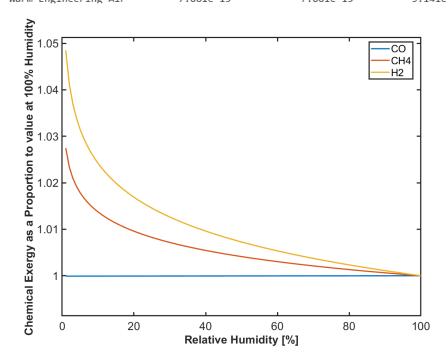
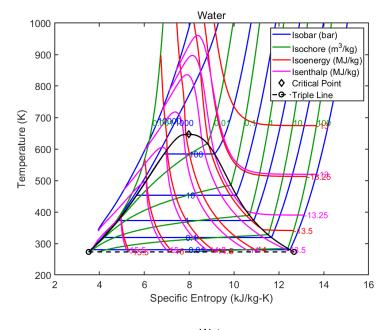
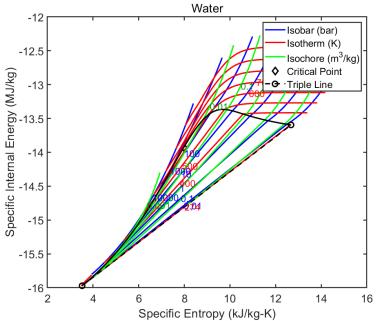
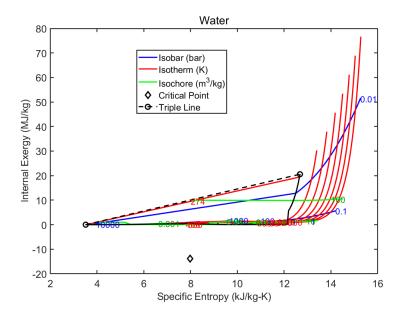
## ME370B Project 2

| Gas                        | LHV [J/kg] | HHV [J/kg] | Exergy [J/kg] | Flow Exergy [J/kg] |
|----------------------------|------------|------------|---------------|--------------------|
| Hydrogen                   | 1.200e+08  | 1.418e+08  | 1.167e+08     | 1.167e+08          |
| Carbon Monoxide            | 1.010e+07  | 1.010e+07  | 9.806e+06     | 9.806e+06          |
| Methane                    | 5.003e+07  | 5.551e+07  | 5.172e+07     | 5.172e+07          |
| Propane                    | 4.635e+07  | 5.034e+07  | 4.869e+07     | 4.869e+07          |
| Nitrogen                   | 0.000e+00  | 0.000e+00  | 2.463e+04     | 2.463e+04          |
| 0xygen                     | 0.000e+00  | 0.000e+00  | 1.240e+05     | 1.240e+05          |
| Carbon Dioxide             | 0.000e+00  | 0.000e+00  | 4.416e+05     | 4.416e+05          |
| Natural Gas                | 4.613e+07  | 5.109e+07  | 4.771e+07     | 4.771e+07          |
| Simplified Syngas          | 2.081e+07  | 2.293e+07  | 2.036e+07     | 2.054e+07          |
| Engineering Air            | -7.881e-15 | -7.881e-15 | 3.611e+03     | 3.611e+03          |
| Compressed Engineering Air | -7.881e-15 | -7.881e-15 | 1.241e+05     | 2.015e+05          |
| Cold Engineering Air       | -7.881e-15 | -7.881e-15 | 4.731e+03     | 4.731e+03          |
| Warm Engineering Air       | -7.881e-15 | -7.881e-15 | 3.141e+05     | 3.141e+05          |
|                            |            |            |               |                    |









The two options for water's dead state is as a vapor or as a liquid with the difference being the latent heat of water at STP. I chose that the dead state of water should be as a liquid. I believed that this was more appropriate because at standard temperature and pressure water is a liquid; it is only a vapor at room temperatures when the vapor has a low partial pressure. Thus I assert that liquid water is closest to a true environmental dead state.

## Total time: ~15 hrs

I thought that while the exergy and LHV/HHV function creation was useful in understanding the concepts, much of my time wasnt related to the thermodynamics but with confusion interfacing with GRI30. I know that some recommendations are given, but I could have saved a lot of frustration is more was given on the GRI30 functions, specifically which functions are bulk (applying to all 53 species), specific to a species, or specific to an element (much of my time was sorting this out for 10-15 functions). Questions 3 and 4 seem less useful conceptually although I guess plotting practice is a skill.