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iv 1 Remembering General Chemistry: Electronic Structure and Bonding 1 Natural OrgaNic cOmpOuNds Versus syNthetic OrgaNic cOmpOuNds 2 1.1 The Structure of an Atom 3 1.2 How the Electrons in an Atom Are Distributed 4 1.3 Ionic and Covalent Bonds 6 1.4 How the Structure of a Compound Is Represented 12 prOBlem-sOlViNg strategy 14

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Nitrogen, oxygen, Nitrogen, oxygen and chlorine atoms can also be sp3 hybridized in organic and chlorine structures. Nitrogen has five valence electrons in its second shell. After hybridization, it will have three half-filled sp3 orbitals and can form three bonds. Oxygen has six valence electrons.

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