

SECTION-I

Select the best alternatives: (60 × 1 = 60)

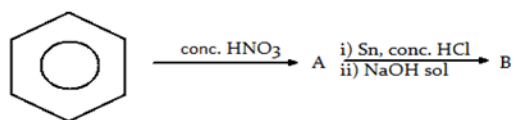
- Which of the following is an open statement
 - x is a natural numbers**
 - Give me a glass of water
 - Wish you best of luck
 - Good morning to all
- If A, B and C are any three sets, then $A - (B \cup C)$ is equal to
 - $(A-B) \cup (A-C)$
 - $(A-B) \cap (A-C)$**
 - $(A-B) \cap C$
 - $(A-B) \cup C$
- If $A = \begin{bmatrix} 2 & 1 & 3 \\ -2 & 1 & 4 \\ 9 & -12 & 2 \end{bmatrix}$ then ratio of minor to co factor of the element a_{23} is
 - 4
 - 4
 - 1
 - 1**
- $(1+i)^6 + (1-i)^6 =$
 - 2^6
 - 2^7
 - 0**
 - -2^7
- The roots of the equation $a(x^2+1) - (a^2+1)x=0$ are
 - $a, \frac{1}{a}$**
 - $a, 2a$
 - $a, \frac{1}{2a}$
 - $\frac{1}{a}, \frac{1}{2a}$
- $1 - \frac{1}{2} + \frac{1}{3} - \frac{1}{4} + \frac{1}{5} \pm \dots =$
 - $\ln 2$**
 - $\log 2$
 - $-\ln 2$
 - $-\log 2$
- The coefficient of x^3y^4 in $(2x+3y^2)^5$ is
 - 360
 - 720**
 - 240
 - 1080
- If $\tan m\theta = \tan n\theta$, then the general value of θ will be in
 - AP**
 - GP
 - HP
 - AGP
- A, B, C are the angles of a triangle, then $\sin^2 A + \sin^2 B + \sin^2 C - 2\cos A \cos B \cos C =$
 - 1
 - 2**
 - 3
 - 4
- The equation of the line perpendicular to the line $\frac{x}{a} - \frac{y}{b} = 1$ and passing through the point at which it cuts x-axis, is
 - $\frac{x}{a} + \frac{y}{b} + \frac{a}{b} = 0$
 - $\frac{x}{a} + \frac{y}{b} - \frac{a}{b} = 0$**
 - $\frac{x}{a} + \frac{y}{b} = 0$
 - $\frac{x}{b} + \frac{y}{a} - \frac{a}{b} = 0$
- The equation of the lines passing through the origin and having slopes 3 and $-1/3$ is
 - $3y^2 + 8xy - 3x^2 = 0$
 - $3x^2 + 8xy - 3y^2 = 0$**
 - $3y^2 - 8xy + 3x^2 = 0$
 - $3x^2 + 8xy + 3y^2 = 0$
- If the line $x + 2by + 7 = 0$ is a diameter of the circle $x^2 + y^2 - 6x + 2y = 0$, then $b =$
 - 3
 - 5
 - 1
 - 5**
- If α, β, γ are the direction angles of a line then the value of $\cos 2\alpha + \cos 2\beta + \cos 2\gamma =$
 - 1
 - 1**
 - 2
 - 0
- The equation of the plane which is parallel to y-axis and cuts off intercepts of length 2 and 3 from x-axis and z-axis is
 - $3x + 2z = 1$
 - $3x + 2z = 6$**
 - $2x + 3z = 6$
 - $3x + 2z = 0$
- The integral $\int \sin(\log x) dx$ is equal to
 - $\frac{1}{2} x [\cos(\log x) - \sin(\log x)]$
 - $\cos(\log x) - x$
 - $\frac{1}{2} x [\sin(\log x) - \cos(\log x)]$**
 - $-\cos(\log x)$
- $x\sqrt{1+y} + y\sqrt{1+x} = 0$, then $\frac{dy}{dx} =$
 - $1+x$
 - $(1+x)^{-2}$
 - $-(1+x)^{-1}$
 - $-(1+x)^{-2}$**
- Suppose $f(x)$ is differentiable at $x=1$ and $\lim_{h \rightarrow 0} \frac{f(1+h) - f(1)}{h} = 5$ then $f'(1)$ equals
 - 3
 - 4
 - 5**
 - 6
- The order of the differential equation of family of curves represented by an equation containing four arbitrary constants, will be
 - 2
 - 4**
 - 5
 - 6
- A bag contains 4 white, 5 black and 6 red balls. If a ball is drawn at random, then what is the probability that the drawn ball is either white or red
 - $1/2$
 - $2/3$**
 - $2/5$
 - $4/15$
- If mean = (3 median - mode) k, then the value of k is
 - 1
 - 2
 - $1/2$**
 - $3/2$
- Select the pair whose dimensions are same
 - Pressure and Stress**
 - Stress and Strain
 - Pressure and Force

- d. Power and Force
22. A particle moves in a plane with constant acceleration in a direction different from the initial velocity. The path of the particle will be
a. st. line b. arc of circle
c. **parabola** d. ellipse
23. It is not possible to write directly on blotting paper or newspaper with ink pen
a. viscosity b. inertia
c. friction d. **capilarity**
24. w_w is weight of 5 liter of benzene in winter and w_s is that in summer. Then,
a. **$w_w > w_s$** b. $w_w < w_s$
c. $w_w = w_s$ d. None
25. The amount of radiation emitted by a perfectly black body is proportional to
a. T b. T^2
c. T^3 d. **T^4**
26. When a plane mirror is rotated through an angle θ then the reflected ray turns through the angle 2θ then the size of the image
a. is doubled b. is halved
c. **remains same** d. becomes ∞
27. A thin convex lens of focal length 10 cm is placed in contact with a concave lens of same material and of same focal length. The focal length of combination will be
a. 0 b. 10 cm
c. 20 cm d. **∞**
28. By Huygen's wave theory of light, we cannot explain the phenomenon of
a. Interference b. Diffraction
c. **Photoelectric effect** d. Polarisation
29. The velocity of light emitted by a source S observed by an observer O, who is at rest with respect to S is c. If the observer moves towards S with velocity v, the velocity of light as observed will be
a. $c+v$ b. $c-v$
c. **c** d. **$\sqrt{c^2 - v^2}$**
30. A charge q is placed at the centre of the line joining two equal charges Q. The system of the three charges will be in equilibrium, if q is equal to
a. $-Q/2$ b. **$-Q/4$**
c. $Q/2$ d. $Q/4$
31. The example for non-ohmic resistance is
a. copper wire b. **diode**
c. carbon resistance d. tungsten wire
32. In electromagnetic induction, the induced charge in a coil is independent of
a. change in flux b. **time**
c. resistance d. none
33. The rest mass of the photon is
a. **0** b. ∞
c. between 0 and ∞ d. 10^7 m/s
34. A photon, an electron and a uranium nucleus all have the same wavelength. The one with the most energy
a. **photon** b. electron
c. uranium nucleus d. equal energy
35. In the depletion region of an unbiased P-N junction diode there are
a. only electrons b. only holes
c. both electrons and holes d. **only fixed ion**
36. What is the modulation index of an over modulated wave
a. 0 b. 1
c. <1 d. **>1**
37. In a reaction the rate of reaction is proportional to its active mass, this statement is known as
a. **Law of mass action**
b. Le-Chatlier's principle
c. Faraday law of electrolysis
d. Law of constant proportion
38. Rutherford's experiment on scattering of particles showed for the first time that the atom has
a. electrons b. protons
c. **nucleus** d. neutrons
39. The oxidation number of Cl in HOCl
a. **+1** b. -1
c. 0 d. +2
40. Ionization potential is lowest for
a. Halogens b. Inert Gases
c. Alkaline earth metals d. **Alkali metal**
41. The minimum energy a molecule should possess in order to enter into a fruitful collision is known as
a. Reaction energy
b. collision energy
c. activation energy
d. **threshold energy**
42. The process, in which no heat enters or leaves the system, is termed as
a. isochoric b. isobaric
c. isothermal d. **adiabatic**
43. Among the following, identify the compound which cannot act as both oxidising and reducing agents
a. H_2O_2 b. SO_2
c. **H_2** d. Cl_2
44. Nitrogen reacts with metal to give

- b. He advised her not to read so fast.
 c. He requested her not to read so fast.
 d. He ordered her not to read so fast.
65. She is not only a receptionist but also an accountant. It is ____ sentence:
 a. simple **b. compound**
 c. complex d. compound-complex
66. The horse and carriage _____ at the door.
 a. are **b. is**
 c. have d. has
67. If $\log_a m$, $\log_b m$, $\log_c m$ are in HP, then a,b,c are in
 a. AP **b. GP**
 c. HP d. AGP
68. Eight points are chosen from a line and five from a line parallel to it. How many triangles can be formed by using these points?
a. 220 b. 80
 c. 140 d. 40
69. The angles A,B,C of triangle ABC are in AP and $\frac{b}{c} = \sqrt{\frac{3}{2}}$ then A =
 a. 60° b. 45°
c. 75° d. 90°
70. $\left| \frac{(1+i)(2+i)}{(3+i)} \right| =$
 a. $\frac{1}{\sqrt{5}}$ b. $\frac{2}{\sqrt{5}}$
 c. $\frac{1}{2}$ **d. 1**
71. If $2^x + 2^y = 2^{x+y}$ then $\frac{dy}{dx}$ at $x=y=1$ is
 a. 1 **b. -1**
 c. 2 d. -2
72. The value of $\int_1^e \log x \, dx =$
 a. e **b. 1**
 c. e-1 d. 1-e
73. $\lim_{x \rightarrow \infty} \sqrt{x^2+8x+3} - \sqrt{x^2+4x+3}$
 a. 0 b. 1
c. 2 d. doesn't exist
74. The area bounded by curve $x^2 = 4ay$ and st. line $y = 2a$ in square units is
a. $\frac{16\sqrt{2}}{3} a^2$ b. $\frac{16}{3} a^2$
 c. $\frac{8}{\sqrt{3}} a^2$ d. $\frac{8\sqrt{2}}{3} a^2$
75. If the roots of $px^2 + qx + r = 0$ be in the ratio of 4:5 then

- a. $12pr = 25q^2$ b. $25pr = 12q^2$
c. $81q^2 = 20pr$ d. $25pr = 3q^2$
76. $\begin{vmatrix} 1 & 0 & -k \\ 2 & 1 & 3 \\ k & 0 & 1 \end{vmatrix} = 0$ for k =
 a. -1 b. 0
 c. 1 **d. for all k in R**
77. The equation of tangent to the parabola $y^2 = 5x$ perpendicular to the line $x+2y-7=0$ is:
a. $16x-8y+5=0$ b. $4x-2y+8=0$
 c. $4x-4y+9=0$ d. $2x-y+11=0$
78. If $(\tan^{-1} x)^2 + (\cot^{-1} x)^2 = \frac{5\pi^2}{8}$ then x =
a. -1 b. 0
 c. 1 d. 2
79. A variable plane is at a constant distance p from the origin and meets the axes at A, B, C. The locus of the centroid of the triangle ABC is
 a. $x^2+y^2+z^2 = p^2$
 b. $x^2+y^2+z^2 = 9p^2$
 c. $x^2+y^2+z^2 = p^{-2}$
d. $x^2+y^2+z^2 = 9p^{-2}$
80. In what ratio the XY-plane divides the line segment joining points (-1,3,4) & (2,-5,6) is
 a. 2:3 internally
b. 2:3 externally
 c. 3:2 internally
 d. 3:2 externally
81. If the position vectors A and B are $3\mathbf{i}-2\mathbf{j}+\mathbf{k}$ and $2\mathbf{i}+4\mathbf{j}-3\mathbf{k}$ the |AB|
 a. $\sqrt{14}$ b. $\sqrt{29}$
 c. $\sqrt{43}$ **d. $\sqrt{53}$**
82. A body of mass 2kg has an initial velocity of 5m/s along OA is subjected a force 4N perpendicular to OA. The displacement of the body after 4 seconds will be:
 a. 12 m b. 20 m
c. 26 m d. 36 m
83. The distance between two rails is 1 m on a circular track of radius 400 m. The outer rail must be raised by how much distance so that a train moves with a speed of 20 m/s without any friction.
 a. 5 cm **b. 10 cm**
 c. 15 cm d. 20 cm
84. A soap bubble is blown to a diameter of 7cm. if 36960 erg work is done to blow it further, then the new radius when the surface tension is 40dyne/cm is:
a. 7 cm b. 15 cm

- c. 3.5 cm d. 14 cm
85. A block of metal of specific heat capacity $0.10 \text{ cal/g}^\circ\text{C}$ fall from a height of 100 m. If 60% of energy of fall is converted into heat then rise in temperature is:
 a. **1.43°C** b. 1.53°C
 c. 1.62°C d. 0.95°C
86. A carnot engine whose source is at 400K , takes 200 calories of heat and rejects 150 calories of heat to sink. The temperature of sink is:
 a. 267 K b. 200 K
 c. **300 K** d. 250 K
87. To make an achromatic combination a convex lens of focal length 42 cm having dispersive power 0.14 is placed in contact with a concave lens of dispersive power 0.21. The length of the convex lens should be
 a. 14 cm b. 21 cm
 c. **63 cm** d. 42 cm
88. A capacitor has capacitance C in air. The either half part of the plate is filled with a dielectric of constant K . The new capacitance will be:
 a. KC b. $(K+1)C$
 c. **$\frac{K+1}{2}C$** d. $(K-1)C$
89. A current in the windings on a toroid is 2A. There are 400 turns and the mean circumferential length is 40cm. If the magnetic field inside is 1T, the relative permeability is :
 a. 200 b. 250
 c. **400** d. 150
90. An ammeter reads up to 1 A. Its internal resistance is $0.81\ \Omega$. To increase the range to 10 A, the value of shunt required is:
 a. $0.9\ \Omega$ b. **$0.09\ \Omega$**
 c. $0.03\ \Omega$ d. $0.3\ \Omega$
91. A cord attached to a vibrating tuning fork is divided into six string segments under a tension of 36 N. it will be divided into 4 segments if the tension is
 a. 16 N b. 24 N
 c. 48 N d. **81 N**
92. If the electron in hydrogen atom jumps from third orbit to second orbit, the wavelength of the emitted radiation is given by:
 a. **$36/5R$** b. $5R/36$
 c. $5/R$ d. $R/6$
93. The count rate of a radioactive source at $t=0$ was 1600 count/sec and at $t=8\text{ sec}$ it was 100 count/sec. The count rate at $t=6\text{ sec}$ will be
 a. 150 b. **200**
- c. 300 d. 400
94. A current 2 A when passed for 5 hours through a molten metal salt, deposits 22.2 g of metal of atomic weight 177. The oxidation state of the metal in the metal salt is
 a. +1 b. +2
 c. **+3** d. +4
95. 6.3gm of hydrated oxalic acid ($\text{H}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$) is present in 100 ml normal solution of the acid. The mole of water of crystallization is
 a. **2** b. 3
 c. 4 d. 5
96. A container of volume V contains 0.28 g of N_2 gas, if some amount of an unknown gas under similar condition of temperature and pressure weight 0.44 g, the molecular mass of the gas is
 a. 22 b. **44**
 c. 66 d. 88
97. Identify the incorrect statement with respect to ozone
 a. Ozone is formed in the upper atmosphere by a photochemical reaction involving dioxygen
 b. Ozone is more reactive than oxygen
 c. Ozone is diamagnetic whereas dioxygen is paramagnetic
 d. **Ozone protects the earth's inhabitants by absorbing γ radiations**
98. The charge/size ratio of a cation determines its polarizing power. Which one of the following sequences represents the increasing order of the polarizing power of the cationic species, K^+ , Ca^{2+} , Mg^{2+} , Be^{2+} ?
 a. $\text{Ca}^{2+} < \text{Mg}^{2+} < \text{Be}^{2+} < \text{K}^+$
 b. $\text{Mg}^{2+} < \text{Be}^{2+} < \text{K}^+ < \text{Ca}^{2+}$
 c. $\text{Be}^{2+} < \text{K}^+ < \text{Ca}^{2+} < \text{Mg}^{2+}$
 d. **$\text{K}^+ < \text{Ca}^{2+} < \text{Mg}^{2+} < \text{Be}^{2+}$**
99. An organic compound A reacts with sodium metal and forms B. On heating with conc. H_2SO_4 , A gives diethyl ether. What are A and B?
 a. **$\text{C}_4\text{H}_9\text{OH}$ and $\text{C}_4\text{H}_9\text{ONa}$**
 b. CH_3OH and CH_3ONa
 c. $\text{C}_2\text{H}_5\text{OH}$ and $\text{C}_2\text{H}_5\text{ONa}$
 d. $\text{C}_3\text{H}_7\text{OH}$ and CH_3ONa
100. What are A and B in the given sequence, respectively?



- a. Aldehyde, nitro compound
- b. Nitro compound, phenyl amine**
- c. Phenyl amine, nitro compound
- d. Phenthalene, phenyl amine

