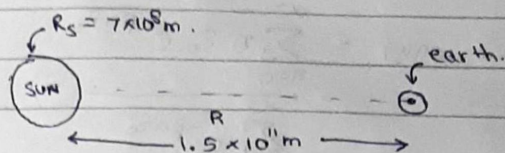


## EP 1108 ASSIGNMENT-2

1.



Let  $I$  be the intensity of radiation

Let  $P$  be the energy per unit second radiated by the sun.

$$\therefore \frac{P}{4\pi \times (1.5 \times 10^{11})^2 \text{ m}^2} = 1.4 \times 10^3 \text{ W/m}^2$$

$$P = 9\pi \times 10^{25} \times 1.4 \text{ W}$$

$$= \frac{63\pi \times 10^{25}}{5} \text{ W}$$

$\therefore$  By Stephan - Boltzmann Law,

$$\frac{63\pi \times 10^{25}}{5} = \sigma \times (4\pi) \times (7 \times 10^8)^2 T^4,$$

where  $\sigma$  = Stephan's constant =  $5.67 \times 10^{-8} \text{ W/m}^2 \text{ K}^4$

$T$  = temp. of sun

$$\begin{aligned} \therefore T^4 &= \frac{63\pi \times 10^{25}}{2\pi \times 4 \times 10^6 \times 5.67 \times 10^{-8} \times 10^{16}} \\ &= \frac{90,000}{2 \times 7 \times 5.67} \times 10^{12} \end{aligned}$$

$$\begin{aligned} T &= 5.802 \times 10^3 \text{ K} \\ &\approx 5802.74 \text{ K} \end{aligned}$$

## 2. Planks Spectral Distribution Law:

$$\bar{E}(\text{mean energy}) = \frac{\sum_{n=0}^{\infty} n \epsilon_0 e^{-\beta n \epsilon_0}}{\sum_{n=0}^{\infty} e^{-\beta n \epsilon_0}}$$

$$= -\frac{d}{d\beta} \left[ \ln \left( \sum_{n=0}^{\infty} e^{-\beta n \epsilon_0} \right) \right]$$

$$= \frac{\epsilon_0}{e^{\beta \epsilon_0} - 1}$$

We know

$$e(\lambda, T) = \frac{8\pi}{\lambda^4} \bar{E}$$

$$= \frac{8\pi}{\lambda^4} \times \frac{\epsilon_0}{e^{\beta \epsilon_0} - 1}$$

Now,

$$\epsilon_0 = \frac{hc}{\lambda} \quad \beta = \frac{1}{kT}$$

$$\therefore e(\lambda, T) = \frac{8\pi hc}{\lambda^5} \times \frac{1}{e^{\left(\frac{hc}{\lambda kT}\right)} - 1}$$

$$\boxed{\text{As } \lambda \rightarrow \infty, e^{\left(\frac{hc}{\lambda kT}\right)} - 1 \rightarrow \frac{hc}{\lambda kT}}$$

$$\therefore \lim_{\lambda \rightarrow \infty} e(\lambda, T) = \frac{8\pi hc}{\lambda^4} \times \frac{\lambda kT}{hc}$$

$$= \frac{8\pi}{\lambda^4} kT$$

This is Rayleigh-Jeans law which is obtained by simplification of Planks law at very long wavelengths.



3. Planck's Theory for  $e(\lambda, T)$ :

$$e(\lambda, T) = \frac{8\pi hc}{\lambda^5} \times \frac{1}{e^{\left(\frac{hc}{\lambda kT}\right)} - 1}$$

For  $e$  to peak [OR,  $R(\lambda, T)$  to peak],

$$\boxed{\frac{d e(\lambda, T)}{d \lambda} = 0}$$

$$8\pi hc \left[ \frac{-5}{\lambda^6} \times \frac{1}{e^{\left(\frac{hc}{\lambda kT}\right)} - 1} + \frac{1}{\lambda^5} \times \frac{-1}{\left[e^{\frac{hc}{\lambda kT}} - 1\right]^2} \times e^{\frac{hc}{\lambda kT}} \times \frac{-hc}{\lambda^2 kT} \right] = 0$$

or

$$\boxed{\frac{5}{\lambda^6} \times \frac{1}{e^{\left(\frac{hc}{\lambda kT}\right)} - 1} = \frac{1}{\lambda^5} \times \frac{1}{\left[e^{\frac{hc}{\lambda kT}} - 1\right]^2} \times e^{\frac{hc}{\lambda kT}} \times \frac{hc}{\lambda^2 kT}}$$

$$x = \frac{hc}{\lambda kT}$$

$$\frac{5}{(e^x - 1)} = \frac{1}{(e^x - 1)^2} \times e^x \times x$$

$$x e^x = 5(e^x - 1)$$

$$\boxed{x = 5(1 - e^{-x})}$$

$\therefore$

$$x = 4.965 \Rightarrow \frac{hc}{\lambda kT} = 4.965 \Rightarrow \lambda T = \frac{hc}{4.965 k} \text{ mk}$$

$$= 2.898 \times 10^{-3} \text{ mk}$$

$$= b$$

Wien's Displacement Law

4.

Emissive Power of

Bernard Star =  $\sigma T^4$ 

$$= 5.67 \times 10^{-8} \times 3^4 \times 10^{12} \text{ W/m}^2$$

$$= \boxed{4.593 \times 10^6 \text{ W/m}^2}$$

 $R(\lambda, T)$  peaks according to Wien's Displacement law

$$\therefore \lambda_{\text{peak}} \times T = b \rightarrow \text{Wien's Constant}$$

$$\lambda_{\text{peak}} \times 3 \times 10^3 = 2.898 \times 10^{-3}$$

$$\boxed{\begin{aligned} \lambda_{\text{peak}} &= 9.66 \times 10^{-7} \text{ m} \\ &= 966 \text{ nm.} \end{aligned}}$$