

$$\left. \begin{aligned} a_0 &= \frac{4\pi\epsilon_0\hbar^2}{me^2} \\ E_1 &= \frac{me^4}{32\pi^2\epsilon_0^2\hbar^2} \end{aligned} \right\} \Rightarrow E_1 = \frac{h}{2ma_0} \quad \text{where s.s. system is used } \frac{1}{4\pi\epsilon_0}$$

Helium Atom - Ground state energy

Variational Calculation / Theory

$\psi(\alpha)$  - Variational wavefunction

$$\frac{\langle \psi(\alpha) | H | \psi(\alpha) \rangle}{\langle \psi(\alpha) | \psi(\alpha) \rangle} = E(\alpha)$$

chosen  
Boundary condition  
Normalizable

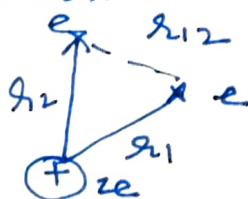
$$\frac{dE(\alpha)}{d\alpha} = 0 \Rightarrow \alpha \text{ is determined}$$

Put it back to  $\psi \Rightarrow$  Variational state

$$\rightarrow E = \underline{\text{Upper bound}}$$

Use Variational Method to find the energy of Ground state of He-atom.

Helium atom has nucleus with  $2e$  charge and 2 electrons



$$P.E \Rightarrow V = -\frac{Ze^2}{r_1} - \frac{Ze^2}{r_2} + \frac{e^2}{r_{12}}$$

$$H = KE + PE$$

$$H = -\frac{\hbar^2}{2m} (\nabla_1^2 + \nabla_2^2) - \frac{Ze^2}{r_1} - \frac{Ze^2}{r_2} + \frac{e^2}{r_{12}}$$

$$\Rightarrow H = -\frac{\hbar^2}{2m} (\nabla_1^2 + \nabla_2^2) - 2e^2 \left( \frac{1}{r_1} + \frac{1}{r_2} \right) + \frac{e^2}{r_{12}}$$

Interaction Energy

$\Downarrow$   
Neglect.

$$\psi(r_1, r_2) = u_{100}(r_1) u_{100}(r_2)$$

(2)

$$\psi_1^0 = \sqrt{\frac{Z^3}{\pi a_0^3}} e^{-\frac{Z r_1}{a_0}}$$

$$\psi_2^0 = \sqrt{\frac{Z^3}{\pi a_0^3}} e^{-\frac{Z r_2}{a_0}}$$

$$\psi(r_1, r_2) = \frac{Z^3}{4\pi a_0^3} e^{-\frac{Z}{a_0}(r_1 + r_2)}$$

$$Z = 2, \quad a_0 = \frac{\hbar^2}{m e^2}$$

$$E = \frac{Z^2 m e^4}{2 \hbar^2 n^2} = \frac{-Z^2 \omega_H}{n^2}$$

$$\text{where } \omega_H = \frac{m e^4}{2 \hbar^2} = \frac{e^2}{2 a_0} \quad \text{where } \boxed{a_0 = \frac{\hbar^2}{2 m e^2}}$$

$$\langle H \rangle = \langle KE \rangle + \langle PE \rangle$$

$$= T + V + \text{Interaction Energy}$$

$$\boxed{\langle H \rangle = \langle T \rangle + \langle V \rangle} + \left\langle \frac{e^2}{r_{12}} \right\rangle$$

For 2 Hydrogen

For Hydrogen atom

$$\langle T \rangle = \frac{Z^2 e^2}{2 a_0} \quad \text{and} \quad \langle V \rangle = -\frac{2 Z e^2}{a_0}$$

$$\langle T \rangle = \int \psi^* \left[ -\frac{\hbar^2}{2m} (\nabla_1^2 + \nabla_2^2) \right] \psi \, d\tau$$

$$\text{For Helium atom } \langle T \rangle = 2 \times \frac{Z^2 e^2}{2 a_0} = \frac{Z^2 e^2}{a_0}$$

$$\langle V \rangle = 2 \times -\frac{2 Z e^2}{a_0} = -\frac{4 Z e^2}{a_0}$$

$$\langle H \rangle = \frac{Z^2 e^2}{a_0} - \frac{4 Z e^2}{a_0} + \left\langle \frac{e^2}{r_{12}} \right\rangle$$

Now we need to find  $\left\langle \frac{e^2}{r_{12}} \right\rangle$

(3)

$$\begin{aligned}\left\langle \frac{e^2}{r_{12}} \right\rangle &= \iint \psi^*(r_1, r_2) \frac{e^2}{r_{12}} \psi(r_1, r_2) dr_1 dr_2 \\ &= \left( \frac{Z^2}{\pi a_0^3} \right)^2 e^2 \iint \frac{1}{r_{12}} e^{-\frac{2Z}{a_0}(r_1+r_2)} dr_1 dr_2\end{aligned}$$

Now  $\frac{2Z}{a_0} r_1 = \rho_1$  and  $\frac{2Z}{a_0} r_2 = \rho_2$

$$\left\langle \frac{e^2}{r_{12}} \right\rangle = \frac{Ze^2}{32\pi^2 a_0} \underbrace{\iint \frac{e^{-(\rho_1+\rho_2)}}{\rho_{12}} d\rho_1 d\rho_2}_{20\pi^2}$$

$$\left\langle \frac{e^2}{r_{12}} \right\rangle = \frac{Ze^2}{32\pi^2 a_0} \times 20\pi^2 = \frac{5Ze^2}{8a_0} = \text{Interaction Energy}$$

$$\langle H \rangle = \frac{Z^2 e^2}{a_0} - \frac{4Ze^2}{a_0} + \frac{5Ze^2}{8a_0}$$

$$= \frac{e^2}{a_0} \left( Z^2 - 4Z + \frac{5}{8}Z \right)$$

$$\langle H \rangle = \frac{e^2}{a_0} \left( Z^2 - \frac{27}{8}Z \right)$$

$$\frac{\partial \langle H \rangle}{\partial Z} = \frac{e^2}{a_0} \cdot \left( 2Z - \frac{27}{8} \right) = 0$$

For minimum

$$\Rightarrow Z = \frac{27}{16} = 1.69$$

$$\langle H \rangle = -2.85 \frac{e^2}{a_0}$$