

I n d e x

To determine the surface tension of water by ..
.. capillary rise

Aim : To experimentally determine surface-tension of water and to be able to explain the deviation of the experimental value from literature value.

Apparatus : 1. Capillary Tubes (of uniform bore and equal diameters)
2. Travelling Microscope
3. Thermometer
4. Glass stopp.
5. Needle
6. Clamping Stand
7. Glass beaker
8. Adjustable stand.

Working Formula :

$$T = \frac{1}{2} [(h + \gamma/r) \cdot \rho g]$$

γ is the radius of capillary tube

h is the height of water rise in the glass capillary.

ρ is the density of water used.

g is the value of acceleration due to gravity in lab.

Experimental Data:

Temperature of water at the beginning (T_b) = 26°C

Temperature " " " end (T_e) = 26°C

$$\text{Mean Temperature} (T_m) = (T_b + T_e)/2 = 26^\circ\text{C}$$

Density of water at $T_m^\circ\text{C}$ (26°C) = $\rho = 0.9965 \frac{\text{gm}}{\text{cc}}$

Angle of contact of water (θ) = 0° $0.9965162 \frac{\text{gm}}{\text{cc}}$

LC of Travelling Microscope : $0.01 \text{ mm } (1 - \frac{49}{50}) = 0.02$

A. Determination of capillary tube diameter... .

using Travelling Microscope :-

Tube No.	$a = \text{MSR}$	$b = \text{VSR} \times LC$	$A_1 = a + b$	$\frac{2000}{A_1 (\text{cm})}$	$a_1 = \text{MSR}$	$b_1 = \text{VSR} \times LC$	$B = a + b$
1(1)	3.12	0.1 X					
2(1)							
3(1)							
4(2)							
(2)							
(2)							
(3)							
(3)							
(3)							

Tube No. Zero
Corrected B_1
(cm) $B_1 - A_1 = AB$

Table for measuring horizontal diameter:-

Tube No.	$a = \text{MSR}$ (cm)	$b = \text{MSR}$ (VS _R X _L) (cm)	$A_1 = a+b$	$\frac{Z_{\text{MSR}}}{\text{Corrected}}$ A_1	$a_1 = \text{NSR}$ (cm)	$b_1 = \text{NSR}$ (VS _L X _R) (cm)	$B_1 = (a+b)$ Z_{NSR} Corrected $B_1 = AB$	
	2.00	0.035	2.035		2.25	0.030	2.280	0.245
1.	1.9	0.001	0.191		1.65	0.005	0.338	0.147
	8.75	0.01	0.165		9.0	0.030	0.285	0.120
2.	0.15	0.041 0.003	0.191		0.3	0.038	0.338	0.25
	0.2	0.030	0.230		0.05	0.020	0.070	0.150
	7	0.021	7.021		6.85	0.046	6.896	0.112
3.	0.15	0.015	0.165		0.25	0.035	0.285	0.270
	0.25 0.05	0.003 5.057			0.49	0.005 4.945		0.125
	5.1	0.010	5.110		4.95	0.040	4.990	0.120

由 Table for Measuring the vertical Diameter :-

Tube No.	$C = MSR$	$\delta = VSRXLC$	Tot. (C) Reading	$C_1 = MSR$	$\delta_1 = VSRXLC$	Tot. (D) Reading	$ C_1 - D_1 $
1	9.35	0.037	9.387	9.6	0.033	9.633	0.246
	9.8	0.040	9.840	9.05	0.042	9.092	0.252
	9.05	0.038	9.088	8.85	0.006	8.856	0.232
2.	9.4	0.01	9.41	9.55	0.045	9.595	0.185
	8.85	0.035	8.885	9.05	0.045	9.095	0.210
	8.85	0.025	8.875	9.05	0.006	9.056	0.181
3.	9.0	0.007	9.007	8.9	0.002	8.902	0.105
	9.0	0.036	9.036	8.85	0.045	8.895	0.141
	9.0	0.04	9.040	8.9	0.001	8.901	0.139

*all data are in (cm)

Tube No.	For lower tip of the needle			For lower meniscus of water			height
	*MSR	*VSR	*TR	*MSR	*VSR	*TR	
	1.6	0.018	1.618	1.95	0.043	1.993	0.375
1.	1.5	0.04	1.504	2.00	0.027	2.027	0.487
	1.55	0.019	1.569	2.00	0.04	2.04	0.471
	1.6	0.018	1.618	2.2	0.04	2.240	0.622
2.	1.5	0.04	1.54	2.25	0.005	2.255	0.715
	1.55	0.019	1.569	2.25	0.017	2.267	0.698
	1.6	0.018	1.618	2.85	0.019	2.869	1.251
3.	1.5	0.04	1.540	2.95	0.025	2.975	1.435
	1.55	0.019	1.569	2.9	0.015	2.915	1.346

Calculation of Surface Tension

1. Tube-1 :

$$\text{Average Radius} = \frac{\text{Horizontal Diameter} + \text{Vertical Diameter}}{2 \times 2}$$

$$= 0.1245 \text{ cm}$$

Average height of lower meniscus of water (h)

$$= 0.4443 \text{ cm}$$

$$\rho \text{ of water} = 0.9965162 \text{ g/cm}^3$$

$$g = 980 \text{ cm/s}^2$$

$$\text{Surface Tension } T = \frac{1}{2} \left[\left(h + \frac{r}{3} \right) \rho g \right]$$

$$= \frac{1}{2} \left[0.4443 + \frac{0.1245}{3} \right] \times (0.1245) \times 980 \times (0.9965162)$$

$$= 29.53 \text{ dyne/cm}$$

$$= 2.953 \times 10^{-2} \text{ N/m}$$

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Error Analysis :-

i) Percentage Error :-
Actual value of Surface Tension = $7.197 \times 10^{-2} \text{ N/m}$
calculated value = $3.31192 \times 10^{-2} \text{ N/m}$
[Experimental]
% error = $\left(\frac{7.197 \times 10^{-2} - 3.31192 \times 10^{-2}}{7.197 \times 10^{-2}} \times 100 \right) \%$
 $= 53.981\%$.

2. Tube 2:

$$\text{Average Radius} = \frac{0.144 + 0.192}{4} = 0.084 \text{ cm.}$$

$$h(\text{capillary rise}) = 0.6783 \text{ cm.}$$

$$\rho = 0.9965162 \text{ g/cm.}$$

$$g = 980 \text{ cm/s}^2$$

$$\begin{aligned} \text{Surface Tension (T)} &= \frac{1}{2} [\sigma_3 + h] r \rho g \\ &= \frac{1}{2} \left[\frac{0.084}{3} + 0.6783 \right] \times \frac{0.084 \times 0.9965162}{980} \\ &= 28.97 \text{ dyne/cm} \\ &= 2.897 \times 10^{-2} \text{ N/m.} \end{aligned}$$

3. Tube 3: $r = 0.061325 \text{ cm}$, $h = 1.344 \text{ cm}$

$$\rho = 0.9965162 \text{ g/cm. } g = 980 \text{ cm/s}^2.$$

$$\begin{aligned} T &= \frac{1}{2} \left[\left(\frac{\sigma}{3} + h \right) r \rho g \right] = 40.8576 \text{ dyne/cm} \\ &= 4.08576 \times 10^{-2} \text{ N/m.} \end{aligned}$$

Average value of Surface Tension

$$= 3.31192 \times 10^{-2} \text{ N/m.}$$

② Discussion of Error :-

The sources of high error :-

- i) The levelling device was faulty
- ii) Difficulty in focusing
- * iii) The contact angle wasn't zero.
- iv) Impurities in Tap water changes density.
- v) Human Errors.

③ Conclusion :-

Experimental value of $T = 3.31192 \times 10^{-2} \text{ N/m}$

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Dynamical viscosity of a liquid & Stokes' law

Aim: To determine the co-efficient of viscosity of a liquid using a falling ball viscometer (using Stokes law), finding possible source of error.

Apparatus:-

- i) Falling fall viscometer
- ii) Balls of different radius.
- iii) Screw gauge
- iv) Scale.

Working Formula:-

$$\eta = \frac{2}{9} \frac{\pi^2 (\rho - \sigma) g}{v_t (1 + 2.4 \frac{r}{R})}$$

η = dynamic viscosity.

r = radius of the ball

ρ = density of ball

v_t = terminal velocity

R = radius of liq. cylinder.

Calculations :-

I. Calculation for Least Count of Screw Gauge :-

for a full rotation ; MSR = 0.5 mm

i.e., for 50 marks = 0.5 mm

So, 1 mark of GS = 0.01 mm

So, L.C = 0.01 mm.

Zero Error of Screw Gauge :-

Negative Zero Error.

Reading = -0.2 mm.

Experimental Data :-

Table : Measure radius and density of steel balls.

Table No. _____

Ball NO.	MSR (mm)	CSR X LC Z (mm)	Total Reading (mm)	Zero corrected Reading (mm)
1(a)	3	0	3.00	3.20
1(b)	3	40 X 0.01	3.40	3.60
1(c)	3	38 X 0.01	3.38	3.58
1(d)	3	36 X 0.01	3.36	3.56
1(e)	3	43 X 0.01	3.43	3.63
1(f)	3	36 X 0.01	3.36	3.56
2(a)	3.5	21 X 0.01	3.71	3.91
2(b)	3.5	26 X 0.01	3.76	3.96
2(c)	3.5	18 X 0.01	3.68	3.88
2(d)	3.5	15 X 0.01	3.65	3.85
2(e)	3.5	22 X 0.01	3.72	3.92

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Ball No.	MSR(mm)	CSR XLC (mm)	Total Reading(mm)	Zero Corrected Reading (mm)
2(f)	3.5	27×0.01	3.77	3.97
3(a)	4.5	45×0.01	4.95	5.15
3(b)	4.5	48×0.01	4.98	5.18
3(c)	4.5	44×0.01	4.94	5.14
3(d)	4.5	43×0.01	4.93	5.13
3(e)	4.5	45×0.01	4.96	5.16
3(f)	4.5	46×0.01	4.96	5.16

Mass

Weight of Ball No.(1) [6 balls] = 0.77g

" " " (2) [6 "] = 1.54g

" " " " (3) [6 Balls] = 2.65g

So, 1 ball (Ball No.1) weight = 0.128g

(Ball No.2) " = 0.257g

(Ball No.3) " = 0.442g

Average radius of Ball no (1) = $\frac{(3.52)}{2}$ mm = 1.761 mm.

Ball " (2) = $\frac{(3.915)}{2}$ mm = 1.9575 mm

" " (3) = 2.577 mm

Ball No.	Radius (mm)	Volume (mm³)	Total vol.	Total Mass (g)	Density (g/mm³)
1(a)	1.6	17.164	(mm³)	(g)	
1(b)	1.8	24.438			
1(c)	1.79	24.033	137.958	0.728	0.00093
1(d)	1.78	24.633			0.00558
1(e)	1.815	25.054			
1(f)	1.78	23.633			

Ball No.	Radius (mm)	Vol (mm³)	Total Vol (mm³)	Total Weight(g)	Density (g/mm³)
2(a)	1.955	31.3114			
2(b)	1.98	32.5281			
2(c)	1.94	30.5962	188.655	1.54	0.00816
2(d)	1.925	29.8920			
2(e)	1.96	31.5523			
2(f)	1.985	32.7751			
3(a)	2.575	71.5476			
3(b)	2.59	72.8052			
3(c)	2.57	71.1316	430.132	2.65	0.00616
3(d)	2.565	70.7772			
3(e)	2.58	71.9651			
3(f)	2.58	71.9651			

Distance between P & Q = 70 cm

Ball No.	Average Time Taken(s)	Vf (cm/s)	Average Vf (cm/s)
1(a)	14.63	4.7846	
1(b)	14.47	4.8375	
1(c)	14.47	4.8375	4.83557
1(d)	14.40	4.8611	
1(e)	14.35	4.8780	
1(f)	14.54	4.8143	
2(a)	9.69	7.2239	
2(b)	9.75	7.1794	7.2217
2(c)	9.69	7.2239	
2(d)	9.66	7.2463	
2(e)	9.58	7.3068	
2(f)	9.79	7.1501	

Ball No.	Time taken(s)	v_t (cm/s)	Average v_t (cm/s)
3(a)	7.03	9.9573	
3(b)	7.03	9.9573	
3(c)	7.07	9.9009	9.96576
3(d)	6.90	10.1449	
3(e)	7.16	9.7765	
3(f)	6.96	10.057	

Calculation for η :

1. For ball 1 :- $\rho = 0.00558 \text{ g/mm}^3$
 $r = 1.461 \text{ mm}$ $R = 46.07 \text{ mm}$.
 $\sigma = 0.961 \text{ g/cm}^3$
 $v_t = 4.83557 \text{ cm/s.}$

$$\eta = \frac{2 \left(\frac{r^2}{980} \right) (1000 \times \rho - \sigma) g}{v_t \left(1 + 2.4 \frac{r}{R} \right)}$$

$$= 5.90973 \text{ cgs-unit}$$

$$= 0.590973 \text{ Pa.s.}$$

2. For ball 2 :- $\rho = 0.00816$
 $r = 1.9575 \text{ mm}$
 $\sigma = 0.961 \text{ g/cm}^3$ $R = 46.07 \text{ mm}$
 $v_t = 7.2217 \text{ cm/s.}$

$$\eta = \frac{2}{980} \left(\frac{(1.9575)^2}{100} \right) (1000 \times 0.00816 - 0.961) \times 9.80 = 7.551875 \text{ cgs.}$$

$$= 0.7551875 \text{ Pa.s.}$$

For Day 3 :-

$$\rho = 0.00616 \text{ g/cm}^3$$

$$r = 2.577 \text{ mm}$$

$$\sigma = 0.961 \text{ g/cm}^3 \quad R = 46.07 \text{ mm}$$

$$V_f = 9.96576 \text{ cm/s.}$$

$$\eta = \frac{2}{9} \left(\frac{2.577^2}{100} \right) (1000 \times 0.00616 - 0.961) \times 980$$

unit in
each step

$$9.96576 \times (1 + 2.4 \times \frac{2.577}{46.07})$$

$$6.65139$$

$$= 9.965762 \text{ cgs-unit.}$$

$$= 0.0996576 \text{ Pa.s. } 0.665139 \text{ Pa.s.}$$

Average Dynamic Viscosity =

$$\eta_{\text{experimental}} = 0.67043374 \text{ Pa.s}$$

Error Analysis :-

$$\eta_{\text{theoretical}} = 0.650 \text{ Pa.s.}$$

$$\eta_{\text{experimental}} = 0.67043374 \text{ Pa.s}$$

$$\text{Relative Error} = \frac{(0.67043374 - 0.650)}{0.650} \times 100\%$$

$$\therefore = 3.14\%$$

5) Standard deviation of $n = 0.0822$
(Experimental)

Standard deviation in ~~and~~ measurement of
diameter of balls.

Ball 1 : 0.159

Ball 2 : 0.0459

Ball 3 : 0.0175

Standard deviation in measurement of time
taken to travel 70 cm (distance b/w P & Q) :-

Ball 1 : 0.09053

Ball 2 : 0.07284

Ball 3 : 0.89610

Error Propagation :-

$$\frac{\partial n}{\partial v_f} = \left(-\frac{1}{v_{\text{mean}}} \right)^2 \left(\frac{2}{9} r^2 \left(\frac{m}{4131123} - T \right) g \right) \frac{1}{\left(1 + 2.4 \frac{r}{R} \right)}$$

$$= 1.22137$$

$$\sigma_{v_f} = 0.331845$$

$$\frac{\partial n}{\partial r} = 41734528$$

$$\text{So, } \sigma_n = \sqrt{\left(\frac{\partial n}{\partial r} \sigma_r \right)^2 + \left(\frac{\partial n}{\partial v_f} \sigma_{v_f} \right)^2} = 1.703885$$

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no. of Measurements = 6.

$$\text{So, standard error} = \frac{\sqrt{17.41935308}}{\sqrt{6}} = 1.70388 \\ = 0.170388 (\text{SI unit})$$

$$\text{So, value of } n_{\text{exp}} = 0.67043374 \pm 0.170388$$

$$n_1 = 0.590973 \pm 0.170388$$

Source of Error:-

- 1) Standard deviation is very high in measurement of radius for different faults.
- 2) Error in measurement of time due to reaction threshold for human reflex.
- 3) ~~At~~ @ Upper air-gap in the fluid tube (air resistance).
- 4) Mechanical, Instrument faults. (if any).

Conclusion:-

The value of $n_{\text{exp}} = 0.67043374 \text{ Pa.s.}$

Discussion - What you learnt?

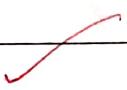
Heat Capacity of Solids :-

Aim: a) To determine specific heat capacity of different metals.
 b) To estimate molar heat capacity of those metals and hence to verify Dulong-Petit law.



Apparatus:

- a) Calorimeter
- b) Digital Thermometer
- c) Electronic Balance
- d) Heater
- e) Waterboiling container.
- f) Stand.
- g) 3 blocks of same material (3 set)



Working Formula :

$$C = \frac{(C + C_{W0}m_W)(T_e - T_w)}{m_1 \cdot (T_1 - T_e)}$$

m_1 = known mass (blocks)

C_W = Specific heat capacity of water $\approx 4.19 \text{ J/g}^\circ\text{C}$

C = heat capacity of calorimeter $\approx 80 \text{ J}^\circ\text{C}$

m_W = Mass of water in calorimeter.

T_e = Equilibrium Temperature.

T_w = known Temp. of calorimeter at first.

T_1 = known Temp. of the metal blocks at first.

Experimental Data:-

Table 1:- Mass measurement of the blocks with nylon thread

1. Iron	0.182 Kg
2. Brass	0.180 Kg
3. Aluminium	0.179 Kg

Table 2: Measurement of Temperatures*

	T_1 (°C)	T_{10} (°C)	T_e (°C)	$T_e - T_{10}$ (°C)	$T_1 - T_e$ (°C)	m_w (g)	
Iron	100.1	30.6	34.5	3.9	69.5	182g	↑
	100.6	34.5	38.3	3.8	66.1	182g	↓
	100.6	34.8	38.9	4.1	65.8	182g	↑
Brass	100.3	31.8	34.4	2.6	68.5	180g	↑
	100.1	28.6	31.4	2.8	71.5	180g	↓
	100.3	33.6	36.5	2.9	66.7	180g	↓
Aluminium	100.6	33.7	41.8	8.1	66.9	179g	↑
	100.1	32.2	41.2	9	68	179g	↑
	100.1	32.1	39.8	7.7	62.9	179g	↓

* T_1 = Temperature of the solid after heating.

T_{10} = Temperature of calorimeter + water (initial)

T_e = Temperature of the equilibrium state (thermal)

$$\text{Formula: } \frac{(C + c_{\text{water}})(T_e - T_w)}{m_1 \cdot (T_i - T_e)} = C$$

calculations :-

Sl.no.

3. $C = \frac{524.3 \text{ J}}{(182 \text{ g}) \times (69.5^\circ \text{C})} = 0.29087 \text{ J/g}^\circ \text{C}$
2. $C = \frac{1337 \text{ J/g}^\circ \text{C} \times 8.1^\circ \text{C}}{179 \text{ g} \times 66.9^\circ \text{C}} = 0.90435 \text{ J/g}^\circ \text{C}$
1. $C = \frac{1337 \text{ J/g}^\circ \text{C} \times 2.8^\circ \text{C}}{180 \text{ g} \times 71.5^\circ \text{C}} = 0.29087 \text{ J/g}^\circ \text{C}$
4. $C = \frac{1337 \text{ J/g}^\circ \text{C} \times 38^\circ \text{C}}{182 \text{ g} \times 66.1^\circ \text{C}} = 0.42232 \text{ J/g}^\circ \text{C}$
5. $C = \frac{1337 \text{ J/g}^\circ \text{C} \times 2.6^\circ \text{C}}{180 \text{ g} \times 68.5^\circ \text{C}} = 0.28193 \text{ J/g}^\circ \text{C}$
6. $C = \frac{1337 \text{ J/g}^\circ \text{C} \times 77^\circ \text{C}}{179 \text{ g} \times 68^\circ \text{C}} = 0.84579 \text{ J/g}^\circ \text{C}$
7. $C = \frac{1337 \text{ J/g}^\circ \text{C} \times 4.1^\circ \text{C}}{182 \text{ g} \times 65.8^\circ \text{C}} = 0.45773 \text{ J/g}^\circ \text{C}$
8. $C = \frac{1337 \text{ J/g}^\circ \text{C} \times 2.9^\circ \text{C}}{180 \text{ g} \times 66.7^\circ \text{C}} = 0.32294 \text{ J/g}^\circ \text{C}$
9. $C = \frac{1337 \text{ J/g}^\circ \text{C} \times 9^\circ \text{C}}{179 \text{ g} \times 67.9^\circ \text{C}} = 0.99003 \text{ J/g}^\circ \text{C}$



Table 3:-

Table for calculating value of specific heat capacity (C)
units: $C + C_{\text{mm}}(J/\text{g}^\circ\text{C})$; $C(J/\text{g}^\circ\text{C})$

Sl.no. Set-I.:

	Iron	value of $C + C_{\text{mm}}$	value of $(C + C_{\text{mm}})(T_e - T_n)$	value of C	standard deviation
1.	Brass	1337	5214.3746.6	0.29087	
2.	Aluminium	1337	10829.7	0.90435	
3.	Iron	1337	5214.3	0.41223	

Set II.:

		val. of $C + C_{\text{mm}}$	value of $C(J/\text{g}^\circ\text{C})$
4.	Iron	1337	0.42232
5.	Brass	1337	0.28193
6.	Aluminium	1337	0.84579

Set III.:

	val. of $C + C_{\text{mm}}$	val. of $(C + C_{\text{mm}})(T_e - T_n)$	value of C
7.	Iron	1337	5481.7
8.	Brass	1337	3877.3
9.	Aluminium	1337	12033

Standard deviation in value of $C(J/\text{g}^\circ\text{C})$

$$\text{Iron} = 0.023900$$

$$\text{Brass} = 0.215608$$

$$\text{Aluminium} = 0.0725490$$

Metal	Specific heat capacity (J/g°C) Measured	Specific heat capacity (J/g°C) Literature	Molar Mass (g/mol)	Molar Heat capacity (J/mol°C)
Iron	0.430763	0.452	55.85	24.05795
Brass	0.298585	0.385	64.28	19.19304
Aluminium	0.9138910	0.896	26.98	24.64829

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Error Analysis :-

Maximum possible error (relative) in c

$$\frac{\delta c}{c} = \frac{\delta m_w}{m_w} + \frac{\delta m_i}{m_i} + \frac{\delta T_e + \delta T_w}{T_e - T_w} + \frac{\delta T_i + \delta T_e}{T_i - T_e}$$

$\delta m_w, \delta m_i, \delta T_i, \delta T_w, \delta T_e$ are maximum possible error in m_w, m_i, T_i, T_w & T_e .

$\delta T_e = \delta T_w = \delta T_i$ = least count for thermometer we've used.

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$$= 0.1^\circ\text{C}$$

least count for weighing machine = 0.001 kg.
= 1 g.

(underlined)

1. For Iron -

Avg of $(T_e - T_w) = 3.933^\circ\text{C}$ Avg val. of $c = 0.430763 \text{ J/g}^\circ\text{C}$

Avg of $(T_i - T_e) = 67.133^\circ\text{C}$

$$\text{So, } \frac{\delta c}{0.430763 \text{ J/g}^\circ\text{C}} = \left(\frac{1}{300} + \frac{1}{182} + \frac{0.2}{3.933} + \frac{0.2}{67.133} \right)$$

$$\delta c = 0.026990 \text{ J/g}^\circ\text{C}$$

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2. For Brass -

Avg of $(T_e - T_w) = 2.7667^\circ\text{C}$ Avg of $c = 0.29859 \text{ J/g}^\circ\text{C}$

Avg of $(T_i - T_e) = 68.9^\circ\text{C}$

$$\delta c = 0.29859 \left(\frac{1}{300} + \frac{1}{180} + \frac{0.2}{2.7667} + \frac{0.2}{68.9} \right) = 0.0251 \text{ J/g}^\circ\text{C}$$

For Aluminium :-

$$\text{Avg } (T_e - T_w) = 8.266 \text{ } ^\circ\text{C}$$

$$\text{Avg } (T_i - T_e) = 67.6 \text{ } ^\circ\text{C}$$

$$\text{Avg val. of } c = 0.91339 \text{ J/g}^\circ\text{C}$$

$$\text{So, } \frac{\delta c}{0.91339 \text{ J/g}^\circ\text{C}} = \left(\frac{1}{300} + \frac{1}{179} + \frac{0.2}{8.266} + \frac{0.2}{67.6} \right)$$

$$\delta c = 0.03295 \text{ J/g}^\circ\text{C}$$



- Maximum possible error in c ---

$$\text{Iron} = 0.026990 \text{ J/g}^\circ\text{C}$$

$$\text{Brass} = 0.0251 \text{ J/g}^\circ\text{C}$$

$$\text{Aluminium} = 0.03295 \text{ J/g}^\circ\text{C}$$



⇒ Deviation from literature value :-

	Literature value of c	Measured value of c	Relative Error
Iron	0.43076 J/g°C	0.452 J/g°C	4.699%
Brass	0.29858 J/g°C	0.385 J/g°C	22.447%
Aluminium	0.91339 J/g°C	0.896 J/g°C	2.166%



■ Discussion of Error:-

- 1) Two potential sources of error - bias error (zero being shifted by some degrees) and linearity error. These are also called as calibration error.
- 2) Poor thermal contact can cause a reading less than

its actual value.

3) The composition of brass metal we used can be different from our literature one giving more relative error.

4) If small fraction of heat can be lost by radiation while transferring the blocks from heating pot to calorimeter.

Conclusion:-

The so we learnt the process of calculating the heat capacity by the experimental procedure.

The molar heat capacity of a solid (in temperature range of Room Temp \rightarrow Melting Pt) is $3R = 24.9433 \text{ J/mol}^\circ\text{C}$ according to Dulong Petit law and our experimental data (ie. $C_{Iron} = 24.05795 \frac{\text{J}}{\text{J/g}^\circ\text{C}}$, $C_{Brass} = 19.19304 \frac{\text{J}}{\text{J/g}^\circ\text{C}}$, $C_{Al} = 24.64329 \frac{\text{J}}{\text{J/g}^\circ\text{C}}$)

Supports the fact.

Study of resonance phenomena in Series LCR AC circuit :

Aim: To be able to draw resonance curve and to obtain quality factor from resonance curve for two different values of R

Apparatus:

- I) LCR test set
- II) Digital Oscilloscope.
- III) Crocodile to BNC cable.
- IV) Wires.

Working Formulae:

$$i = \frac{VR}{R}$$

i = current passing through the system
 VR = voltage across resistor
 R = value of R

$$f_0 = \frac{1}{(2\pi\sqrt{LC})}$$

f_0 = resonance frequency
 L = inductance.
 C = capacitance.

$$Q = \frac{f_0}{\Delta f}$$

f_0 = resonance freq.
 Δf = full width of resonance curve.

$$= \frac{\sqrt{L}}{RC}$$

L = Inductance
 R = Resistance
 C = Capacitance.

■ Experimental Data :- $R = 220\Omega$

$L = 40\text{mH}$; Capacitance = (0.1 microF)

1. Table 1 : Data of Freq. (kHz) with corresponding voltage across resistance and calculated current (I) (mA)

SNO.	Frequency (kHz)	Voltage across resistance (V)	Current (I) (mA)
1	0.1152	0.144	0.6545
2	0.53	0.624	12.8363
3	0.893	0.203	0.02272
4	1.25	1.72	0.7.8181
5	1.626	2.4	10.9091
6	2	3	13.6364
7	2.415	3.32	15.09091
8	2.611	3.3	15
9	2.809	3.24	14.7273
10	3.026	3.12	14.18182
11	3.226	2.96	13.4546
12	3.636	2.68	12.1819
13	4.024	2.48	11.2728
14	4.405	2.32	10.5455
15	4.831	2.12	9.6364
16	5.208	2.04	9.2728
17	5.65	1.92	8.7273
18	6.02	1.8	7.6364
19	6.41	1.68	7.4546
20	6.826	1.64	7.0901
21	7.246	1.56	6.5455
22	7.874	1.44	6.3637
23	8.264	1.44	6.1819
24	8.621	1.4	6.1819
25	9.091	1.36	5.81819
26	9.434	1.36	5.81819
27	n. nn1	1.28	5.81819

$$\text{Lo} \quad R = 470\Omega \quad \text{G}$$

Table 2 : Data for $F_{eq}(KHz)$ vs Voltage across resistance(V) and calculated current(mA).

$F_{eq}(KHz)$	Voltage across resistance (V)	Current (I) (mA)
0.148	0.38	0.80851
0.513	0.41 1.3	0.7659
0.982	2.52	5.36170
1.421	3.6	7.65057
1.805	4.32	9.191489
2.208	4.8	10.21276
2.625	4.8	10.21276
3.017	4.48	9.531914
3.436	4.32	8.68085
3.802	4.08	8.510638
4.228	4	7.82078
4.619	3.68	7.65057
5.013	3.6	7.310148
5.415	3.44	6.978723
5.814	3.28	6.63820
6.211	3.12	6.468085
6.608	3.04	6.29787
7.018	2.96	6.127659
7.407	2.88	5.78723
7.812	2.72	5.65057
8.22	2.66	5.31914
8.613	2.6	5.404255
9.05	2.54	5.23404
9.432	2.46	5.10638
9.911	2.4	

III Calculation:-

$$f_0 = \frac{1}{2\pi\sqrt{LC}}$$

$$Q = \frac{\sqrt{L}}{(R\sqrt{C})}$$

1) $L = 40 \mu H$ & Capacitance (C) = $0.1 \mu F$

$$f_{0I} = \frac{1}{2\pi\sqrt{40 \times 10^{-6} \times 0.1 \times 10^{-6}}} = 2.51646 \text{ kHz}$$

$$Q_I = 2.8748$$

II) $L =$ & $C =$

$$f_{0II} =$$

$$Q_{II} =$$

IV Data from graph-fitting \Rightarrow

Experimental

$$f_{0I} = 2.7894 \quad \text{Standard Error} = 0.09809$$

$$f_{0II} = 2.72465 \quad \text{Standard Error} = 0.06099$$

$$Q_I = 0.87752 \quad \text{Standard Error} = 0.07245$$

$$Q_{II} = 0.5501 \quad " \quad " = 0.0724 \\ 0.02452$$

Error Calculation :-For $R = 220\Omega$;

$$Q_I \text{ (Theoretical)} = 2.8748$$

$$Q_I \text{ (Exp)} = 0.87752$$

$$\begin{aligned} \text{Relative Error} &= \frac{(2.8748 - 0.87752)}{(2.8748)} \times 100\% \\ &= 69.47544\% . \end{aligned}$$

$$f_{0I} \text{ (Theoretical)} = 2.51646 \text{ kHz}$$

$$f_{0I} \text{ (Experimental)} = 2.7894 \text{ kHz}$$

$$\begin{aligned} \text{Relative Error} &= \frac{(2.7894 - 2.51646)}{2.51646} \times 100\% \\ &= 10.846\% . \end{aligned}$$

For $R = 470\Omega$

$$Q_{II} \text{ (Theoretical)} = 1.34565$$

$$Q_{II} \text{ (Experimental)} = 0.550$$

$$\begin{aligned} \text{Relative Error} &= \frac{(1.34565 - 0.550)}{1.34565} \times 100\% \\ &= 7.61\% \text{ } 59.12\% . \end{aligned}$$

$$f_{0II} \text{ (Theoretical)} = 2.72465 \text{ kHz}$$

$$f_{0II} \text{ (Experimental)} = 2.51646 \text{ kHz}$$

$$\text{Relative Error} = 7.64\%$$

Observation:-

for $f_1 > f_2$, Q_1 for $R = 220\Omega$

and $f_2 > Q_2$ for $R = 470\Omega$

Comparing the experimental values we found -

$$f_1 > f_2$$

and

$$Q_1 > Q_2$$

 Discussion of Error:-

i) The intrinsic properties of the circuit were not considered, for example internal resistance wasn't considered.

ii) The grounding wasn't done properly

iii) The oscilloscope was fluctuating if it did not give a fixed value. So we took the mean.

iv) Due to these conditions the experimental error was very high.

Conclusion

Conclusion:

Our findings in the experiment can be summarized as follows

A series circuit composed of Resistance 220Ω, capacitance 0.1μF and 40mH inductance has a Quality factor 0.87752 (59% error).

An increase in the resistance of the circuit decreases the Q factor to 0.5501 (59% error) and as Q factor infers the sharpness of resonance.

We conclude decreasing resistance increases the sharpness of resonance, as the Quality factor increases.