

## HOMEWORK ASSIGNMENT - 1

## Indraprastha Insitute of Information Technology, Delhi

COMPUTER SCIENCE AND APPLIED MATHEMATICS

# **Introduction to Quantitative Biology**

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#### 1 Code

```
1 compare()
g function input_binary = convert_to_bin(input_sequence)
5 input_binary = '';
7 for i = 1 : length(input_sequence)
      if input_sequence(i) == 'A' || input_sequence(i) == 'T'
          input_binary = strcat(input_binary, '0');
10
11
      elseif input_sequence(i) == 'G' || input_sequence(i) == 'C'
12
          input_binary = strcat(input_binary, '1');
13
14
15
16
17 end
18
19 end
20
21
function bond_energy = get_bond_energy(input_binary)
23
bond_energy = 0;
25
for i = 1 : length(input_binary)
      if input_binary(i) == '0'
27
          bond_energy = bond_energy + 2;
28
29
          bond_energy = bond_energy + 3;
30
31
      end
32 end
зз end
34
35 function compare()
36
  fprintf ("Please note that the binary encoding is as follows: \n 1. 0 stands for A and T \n 2. 1 stands for G and C\n\n
seq_a = input('Enter base sequence A: ', 's');
seq_b = input('Enter base sequence B: ', 's');
42 bin_a = convert_to_bin(seq_a);
 bin_b = convert_to_bin(seq_b);
45 fprintf("\nThe binary encoding of base sequence A is: %s\n", bin_a);
  fprintf("The binary encoding of base sequence B is: %s \n\n", bin_b);
be_a = get_bond_energy(bin_a);
 be_b = get_bond_energy(bin_b);
fprintf("The bond energy of sequence A is: \%d\epsilon(k_b)T\n", be_a);
fprintf ("The bond energy of sequence B is: \%d\epsilon(k_b)T\n\n", be_b);
if be_a > be_b
     fprintf ("Molecule A will have a higher melting point n");
selseif be_a < be_b
      fprintf ("Molecule B will have a higher melting point n");
57
58 else
      fprintf ("Both the molecules have the same melting point n");
59
60
61 end
62 end
```

### 2 Input and Output

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Below is the MATLAB console output of the above code run for 3 different cases.

MATLAB Command Window

Figure 1: Input and Output of the code shown in 3 cases

1 of 1

## >> be\_cal Please note that the binary encoding is as follows: 1. 0 stands for A and T 2. 1 stands for G and C Enter base sequence A: ACGT Enter base sequence B: GCATATGC The binary encoding of base sequence A is: 0110 The binary encoding of base sequence B is: 11000011 The bond energy of sequence A is: $10\varepsilon(k_b)T$ The bond energy of sequence B is: $20\varepsilon(k_b)T$ Molecule B will have a higher melting point >> be cal Please note that the binary encoding is as follows: 1. 0 stands for A and T 2. 1 stands for G and C Enter base sequence A: AGCTTAGC Enter base sequence B: TAGC The binary encoding of base sequence A is: 01100011 The binary encoding of base sequence B is: 0011 The bond energy of sequence A is: $20\varepsilon(k_b)T$ The bond energy of sequence B is: $10\varepsilon(k_b)T$ Molecule A will have a higher melting point >> be\_cal Please note that the binary encoding is as follows: 1. 0 stands for A and T 2. 1 stands for G and C Enter base sequence A: ATCG Enter base sequence B: CTAG The binary encoding of base sequence A is: 0011 The binary encoding of base sequence B is: 1001 The bond energy of sequence A is: $10\varepsilon(k_b)T$ The bond energy of sequence B is: $10\varepsilon(k_b)T$ Both the molecules have the same melting point >>

## 3 Using this code to find molecule with higher melting point

The above code can be used to compare the binding energy of two molecules and decide which of the molecules will have a higher melting point.

The **compare()** function in the code takes two strings as input and prints out the string with a higher melting point. This is done by iterating over the binary encoding of the string.

Whenever a 0 is encountered, it means it is an Adenine or Thymine molecule, which contains 2 H-bonds. Similarly whenever a 1 is encountered, it indicates a Cytosine or Guanine which contains 3 H-bonds. Hence, corresponding numbers are added to the total number of bonds and the bond energy is calculated. This is done in the **get\_bond\_energy**() function.

The **compare**() function then compares the bond energy and gives the one with a higher melting point.