

## **EUROPEAN COMMISSION**

JOINT RESEARCH CENTRE





# CERTIFIED REFERENCE MATERIAL BCR® – 032

# **CERTIFICATE OF ANALYSIS**

NATURAL MOROCCAN PHOSPHATE ROCK (Phosphorite)				
	Mass fraction based on dry mass		Number of	
	Certified value 1) [g/kg]	Uncertainty <sup>2)</sup> [g/kg]	accepted individual measurements	
Ca expressed as CaO	518	4	70	
Total P expressed as P <sub>2</sub> O <sub>5</sub>	329.8	1.7	85	
Carbonate Carbon expressed as CO <sub>2</sub>	51.0	0.8	60	
F	40.4	0.6	80	
Si expressed as SiO <sub>2</sub>	20.9	1.2	60	
Total S expressed as SO₃	18.4	0.8	75	
Al expressed as Al <sub>2</sub> O <sub>3</sub>	5.5	0.6	80	
Mg expressed as MgO	4.0	0.1	65	
Fe expressed as Fe <sub>2</sub> O <sub>3</sub>	2.3	0.1	65	

<sup>1)</sup> The certified value is the unweighted mean of individual measurements obtained by different laboratories. The certified value is traceable to SI.

This certificate is valid for one year after purchase.

Sales date:

The minimum amount of sample to be used is 1 g.

#### NOTE

This material has been certified by BCR (Community Bureau of Reference, the former reference materials programme of the European Commission). The certificate has been revised under the responsibility of IRMM.

Brussels, November 1979 Latest revision: March 2010

Signed:

Prof. Dr. Hendrik Emons
European Commission
Joint Research Centre
Institute for Reference Materials and Measurements
Retieseweg 111
B-2440 Geel, Belgium



The uncertainty is estimated standard deviation of reproducibility which. It accounts for the precision and bias of the participating laboratories as well as for any inhomogeneity of the material.

Indicative Values				
	Mass fraction based on dry mass		Number of	
	Certified value 1) [mg/kg]	Uncertainty <sup>2)</sup> [mg/kg]	accepted sets of data p	
As	9.5	0.5	7	
В	22.6	2.2	6	
Cd	20.8	0.7	12	
Cr	257	16	12	
Co	0.59	0.06	9	
Cu	33.7	1.4	14	
Hg	0.055	0.011	6	
Mn	18.8	1.3	13	
Ni	34.6	1.9	11	
Ti	171	10	10	
V	153	7	12	
Zn	253	6	9	

- 1) This value is the unweighted mean of p accepted sets of results. The certified value is traceable to SI.
- 2) The 95% confidence interval is a measure of the uncertainty and is applicable when the reference material is used for calibration purposes.

When the reference material is used to assess the performance of a method, the user should refer to the recommendations laid down in the last chapter (instructions for use) of the certification report. In particular he should use the values of the within-laboratory set standard deviation  $(S_W)$ , and of the between-laboratory set standard deviation  $(S_B)$  given there.

Additional Material Information			
	Mass fraction based on dry mass		
	Estimated value		
Na expressed as Na <sub>2</sub> O K expressed as K <sub>2</sub> O Organic C expressed as C Ag Mo Pb Sb Se Th U	8.6 g/kg 0.9 g/kg 1.6 g/kg 2 mg/kg 2-4 mg/kg 5.4 mg/kg 3 mg/kg 10 mg/kg 2 mg/kg 10 mg/kg		

# **DESCRIPTION OF THE SAMPLE**

The sample consists of approximately 100 g of thoroughly mixed finely ground material (particle size < 100  $\mu$ m) taken from a batch of a natural Moroccan phosphate rock usually employed for the production of phosphate fertilizers. The sample is homogenous at least to a 1 g level. The sample is available in brown glass bottles closed with a double plastic stopper.

# ANALYTICAL METHOD USED FOR CERTIFICATION

CaO : Volumetric method with KMnO<sub>4</sub>, titration with EDTA

P<sub>2</sub>O<sub>5</sub>: Quinoline phosphomolybdate gravimetry, spectrophotometric method, X-ray fluorescence

: Gravimetry of CO<sub>2</sub> evolved by acid attack, titration of CO<sub>2</sub> in non-aqueous medium,

conductimetric measurement of CO<sub>2</sub>

F : Spectrophotometric and volumetric methods after distillation of F, ion selective electrode

SiO<sub>2</sub> : Gravimetric methods, spectrophotometric method after alkaline fusion,

X- ray fluorescence, inductively coupled plasma

SO<sub>3</sub>: Gravimetry after acid dissolution, reduction to S<sup>2</sup> and titration with mercuric solution

Al<sub>2</sub>O<sub>3</sub>: Atomic absorption spectrometry, gravimetric method with 8-hydroxyquinoline,

spectrophotometric method, X-ray fluorescence, neutron activation analysis

MgO : Atomic absorption spectrometry, inductively coupled plasma

Fe<sub>2</sub>O<sub>3</sub> : Atomic absorption spectrometry, spectrophotometric methods, X-ray fluorescence,

inductively coupled plasma.

- Hydride atomic absorption spectrometry
- Neutron Activation analysis
- Spectrophotometry
- Voltammetry

 $CO_2$ 

- Photoneutron activation analysis
- Photon activation analysis
- Charged particle activation analysis
- Inductively coupled plasma emission spectrometry
- Neutron capture activation analysis
- Atomic absorption spectrometry
- Isotope dilution mass spectrometry
- Potentiometric stripping analysis
- Graphite furnace atomic absorption spectrometry
- Solid sample atomic absorption spectrometry
- Micro wave plasma emission spectrometry
- Cold vapour atomic absorption spectrometry
- Cold vapour atomic fluorescence spectrometry

#### **PARTICIPANTS**

- ANIC, Milano (IT)
- APC, Toulouse (FR)
- ARBED SA, Esch-sur-Alzette (LU)
- Bundesanstalt für Materialprüfung, Berlin (DE)
- Centre National de la Recherche Scientifique (CNRS), Centre de Recherches Petrografiques et Geochimiques, Vandœuvre-le-Nancy (FR)
- Centro Italiano Studi Esperienze (CISE), Milano (IT)
- Centro Nazionale Ricerche; Centro di Radiochimica e Analisi per Attivazione, Pavia (IT)
- CNR, Centro Radiochimica, Pavia (IT)
- ECN Netherlands Energy Research Foundation; Research Centre, Petten (NL)
- European Commission, Joint Research Centre, Chemistry Division and CETIS, Ispra (IT)
- Fabbrica Perfosfati, Cerea (IT)
- Général des Engrais SA, Rouen (FR)
- Gesellschaft für Strahlen- und Umweltforschung, Neuherberg (DE)
- Joint Research Centre Commission of the European Communities (EEC), Ispra (IT)
- Laboratoria van het SCK/CEN, Mol (BE)
- Laboratory of the Governemental Chemist, London (GB)
- Landwirtschaftskammer Rheinland, Landwirtschaftliche Untersuchungs- und Forschungsantstalt Bonn, Bonn (DE)
- Landwirtschaftskammer Schleswig-Holstein, Landwirtschaftliche Untersuchungs- und Forschungsantstalt Kiel, Kiel (DE)
- Nitriging Eireann Teoranta, Arklow (IE)
- Produits Chimiques Ugine Kuhlmann, Levallois (FR)
- Rijkslandbouwproefstation, Maastricht (NL)
- Rijksuniversiteit Gent; Instituut voor Nucleaire Wetenschappen, Gent (BE)
- Services Techniques de l'Agriculture, Ettelbruck (LU)

- Station Agronomique de l'Aisne, Laon (FR)
- Technical and Analytical Services, Stockton-on-Tees (GB)
- Ugine-Kuhlmann, Levallois-Perret (FR)
- UKF-SBB, Geleen (NL)
- Università di Bologna, Istituto Chimica Agraria, Bologna (IT)
- Universtiät Hohenheim, Landesanstalt für Landwirtschaftliche Chemie, Stuttgart (DE)
- Windmill Holland BV, Vlaardingen (NL)

#### SAFETY INFORMATION

The usual laboratory safety precautions apply.

#### INSTRUCTIONS FOR USE

Once the bottle has been opened, the material is susceptible to contamination (e.g. by laboratory dust or vapour) or losses. Precautions with regards to storage container and temperature should be taken. The portion for analysis shall be taken as it is. The moisture content should be determined by drying a portion of the sample at 105 °C during 2 hours.

#### **STORAGE**

The material can be stored at room temperature.

However, the European Commission cannot be held responsible for changes that happen during storage of the material at the customer's premises, especially of opened samples.

## **LEGAL NOTICE**

Neither IRMM, its subsidiaries, its contractors nor any person acting on their behalf.

(a) make any warranty or representation, express or implied that the use of any information, material, apparatus, method or process disclosed in this document does not infringe any privately owned intellectual property rights;

or

(b) assume any liability with respect to, or for damages resulting from, the use of any information, material, apparatus, method or process disclosed in this document save for loss or damage arising solely and directly from the negligence of IRMM or any of its subsidiaries.

#### NOTE

A technical report on the production of BCR-032 is available on the internet (<a href="http://www.irmm.jrc.be">http://www.irmm.jrc.be</a>). A paper copy can be obtained from IRMM on request.