

Exploring Hydrogen Bond Geometry in RNA With F-SAPT

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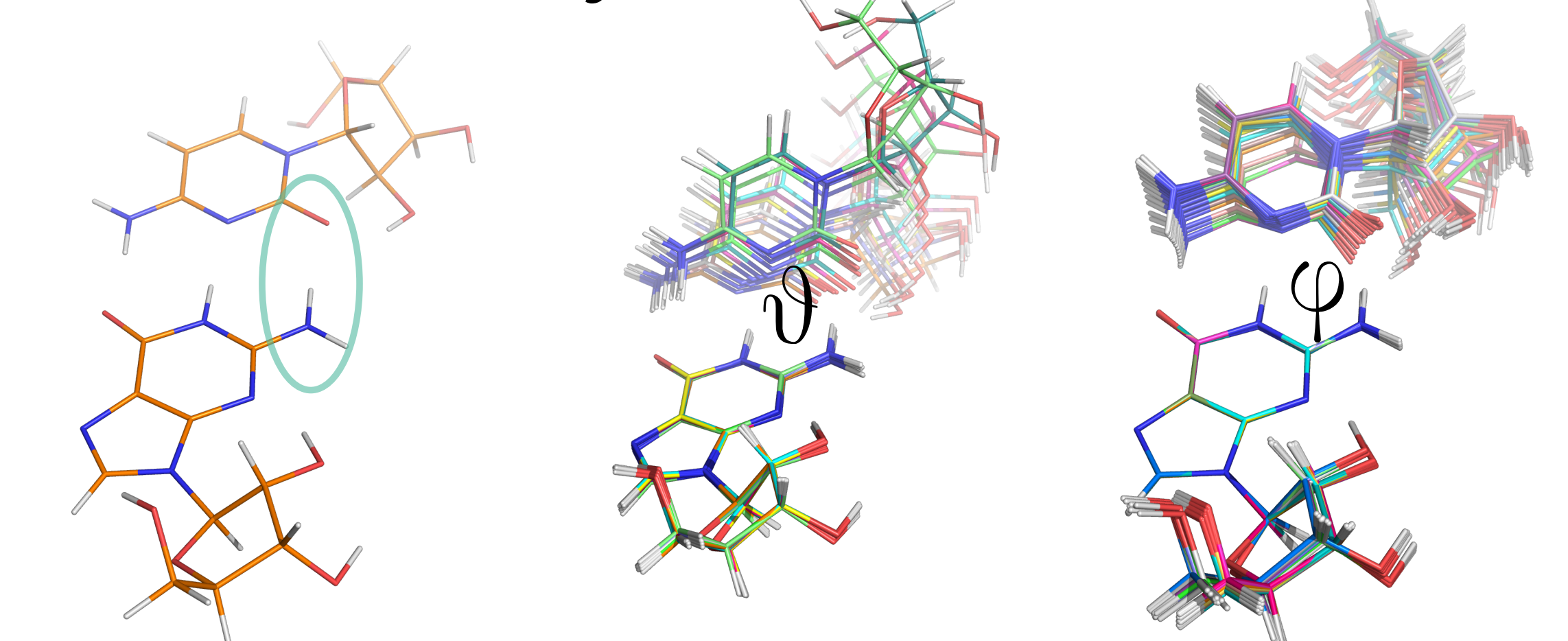
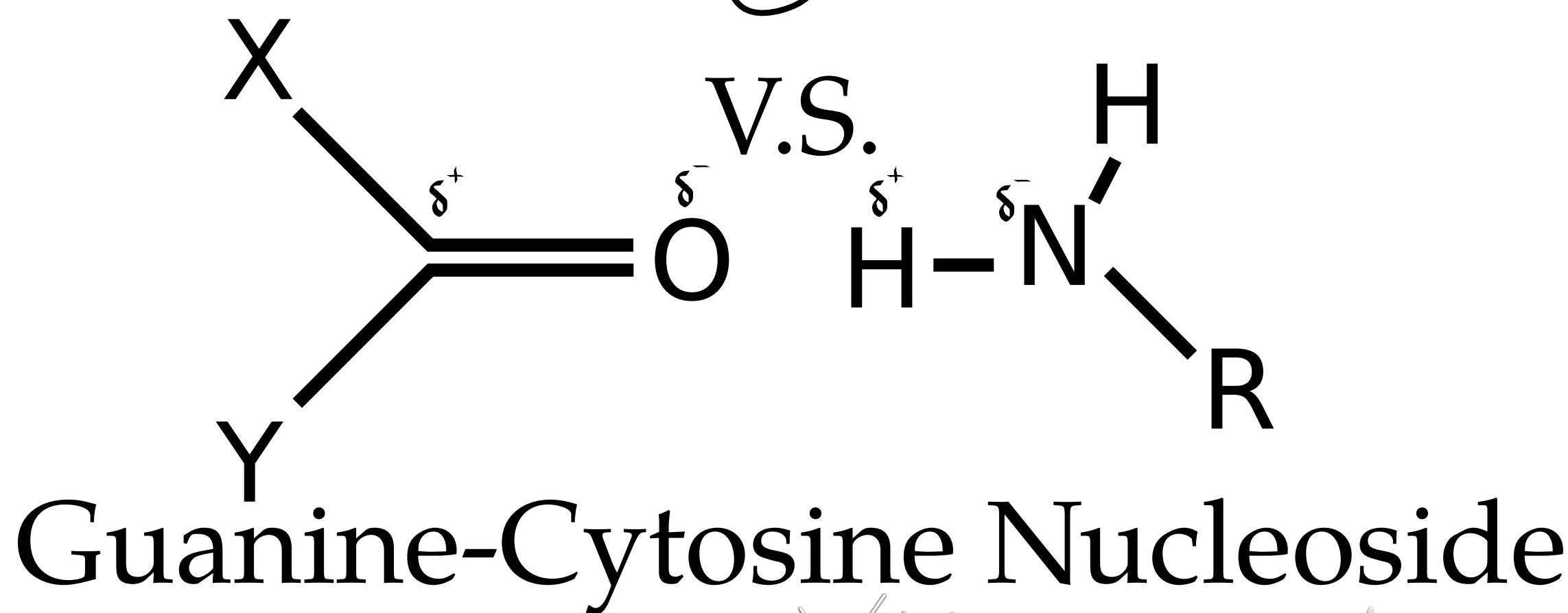
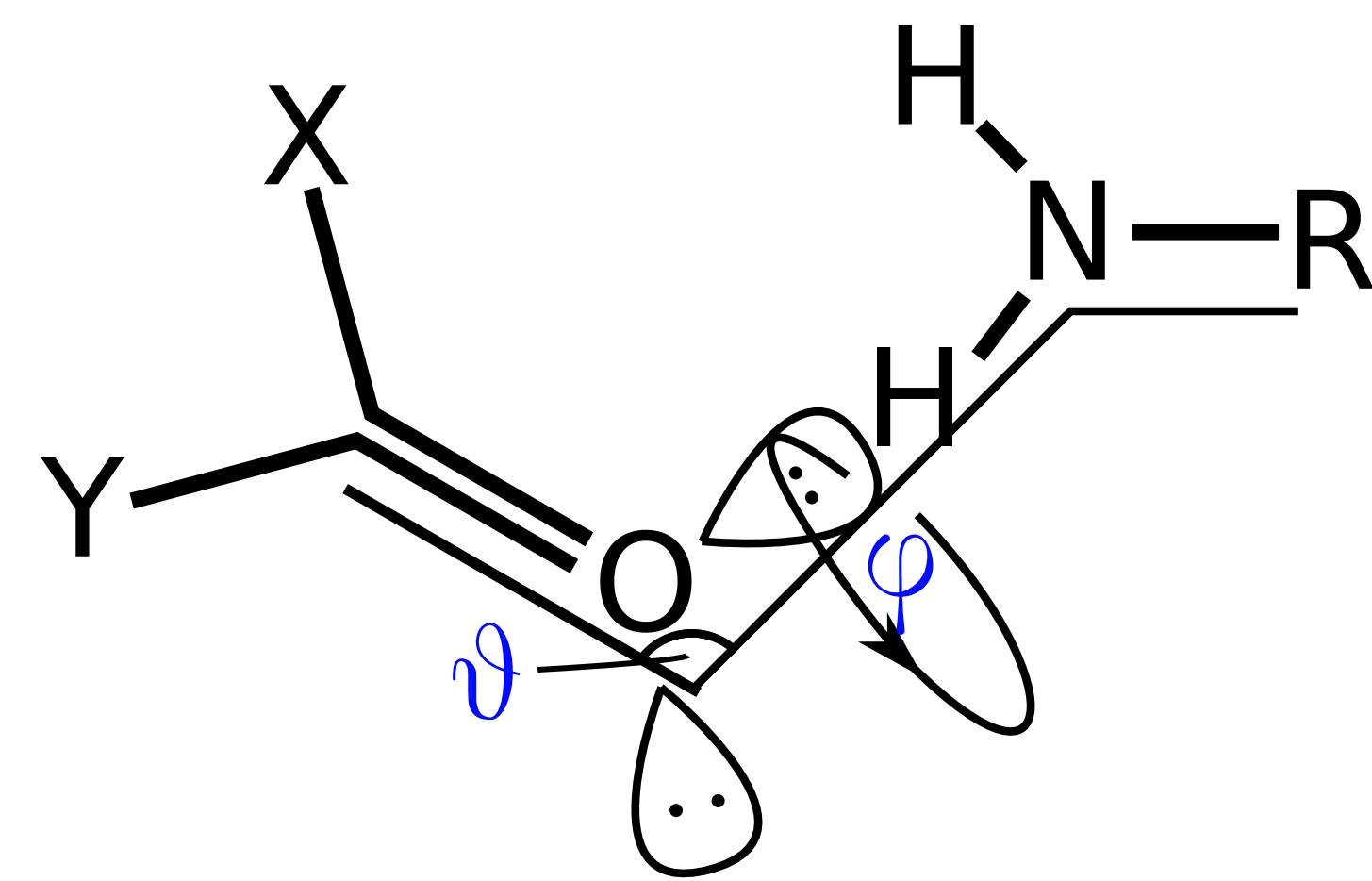
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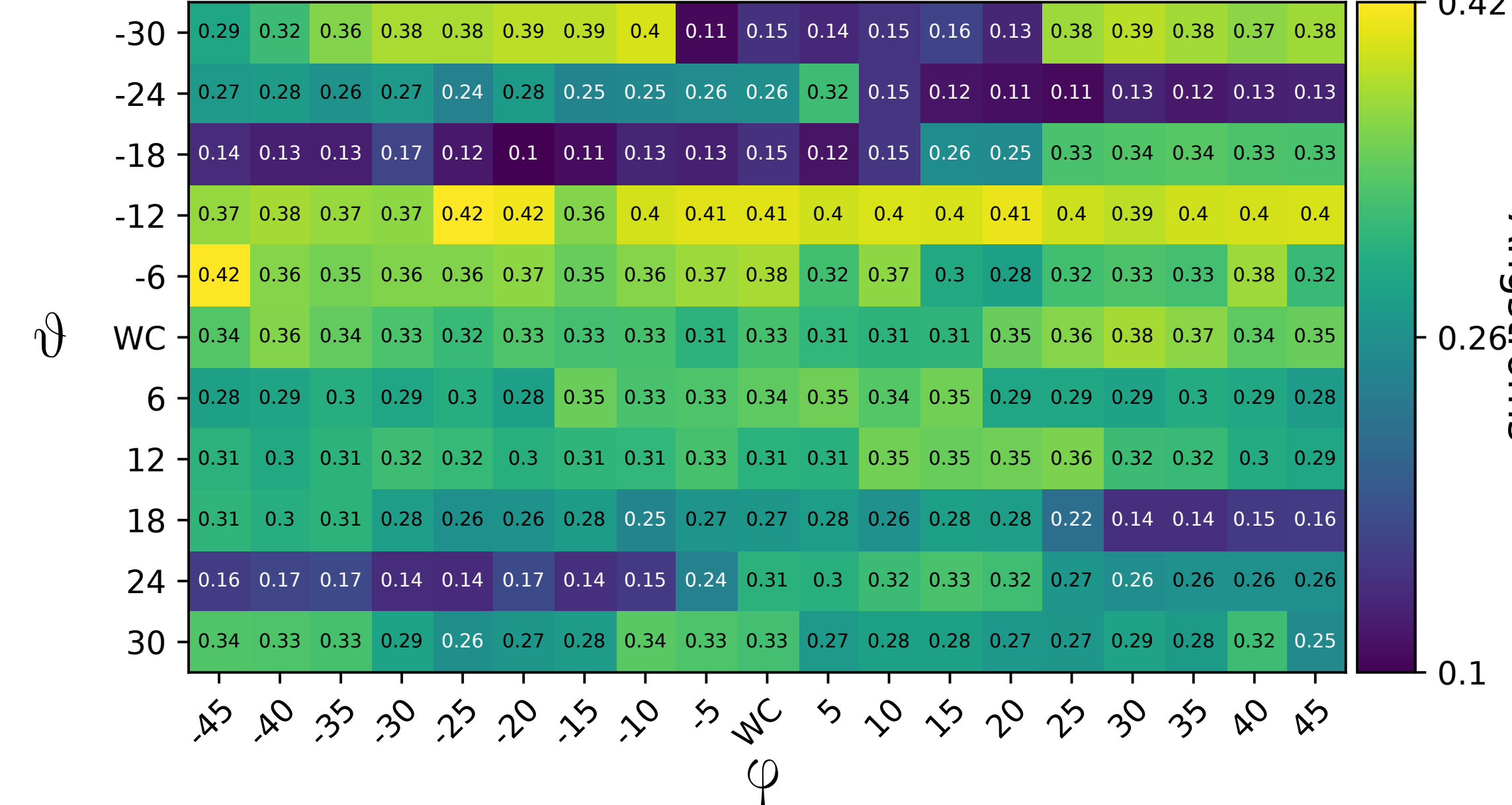
Introduction

- Modeling Nucleic Acids (NAs) using Molecular Mechanics contributes to biology, biochemistry, and medicine.
- Simulations lose native structure
- Some Hydrogen Bonds (HBs) bent
 - HBs essential for NA structure
 - Rosetta adopted dihedral HB term for peptide modeling
- Orientation preferences unknown
- Point charges shouldn't work for certain HBs
- Classic force fields (AMBER) only have radial nonbonded terms
- Lack of clean experimental data → computational chemistry
- Functional group (FG) Symmetry Adapted Perturbation Theory (F-SAPT) → interaction energy (IE) between two FGs¹
- FG-IEs sum to SAPT0 IEs with mean absolute error 1.0 kcal/mol
- Goal:** Patch classic NA force fields

Carbonyl-Amino HB



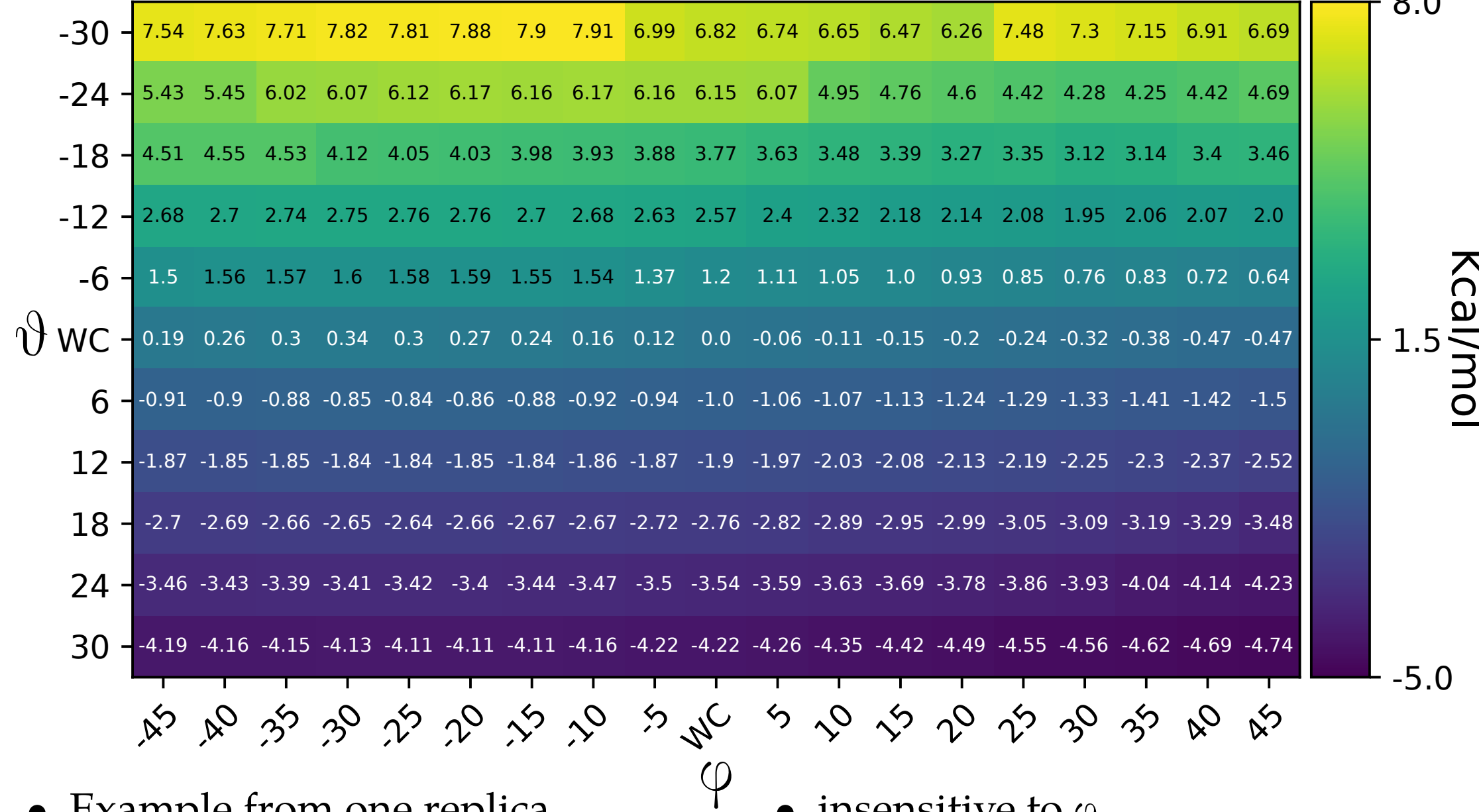
Mean heavy atom RMSD across six SA replicas



- Langevin thermostat is stochastic
 - Replicated process six times.
- Highest RMSD for WC atoms < 0.2 Å
 - differences from distant atoms

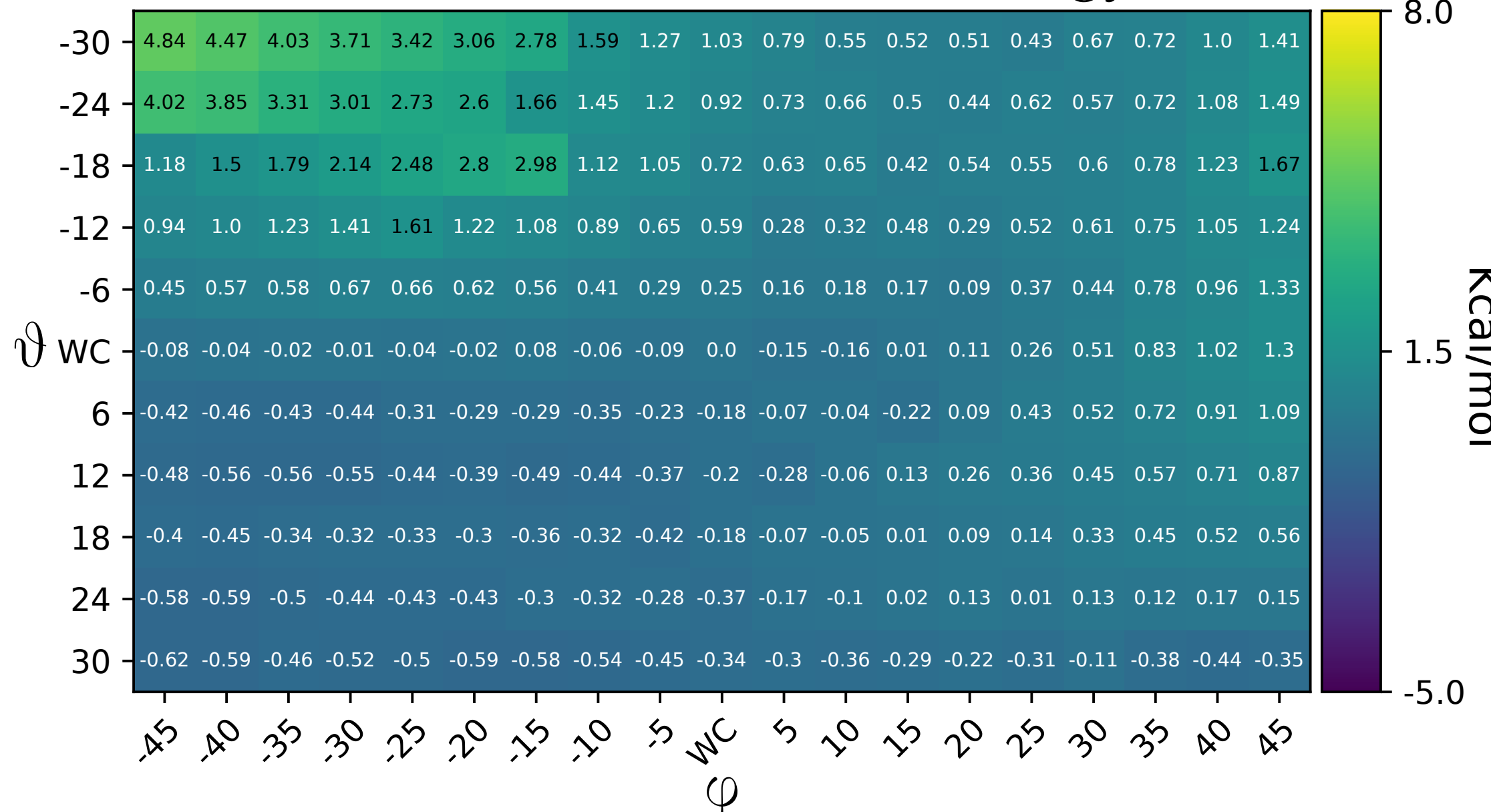
Force Field is Indifferent to Phi

AMBER Interaction Energy



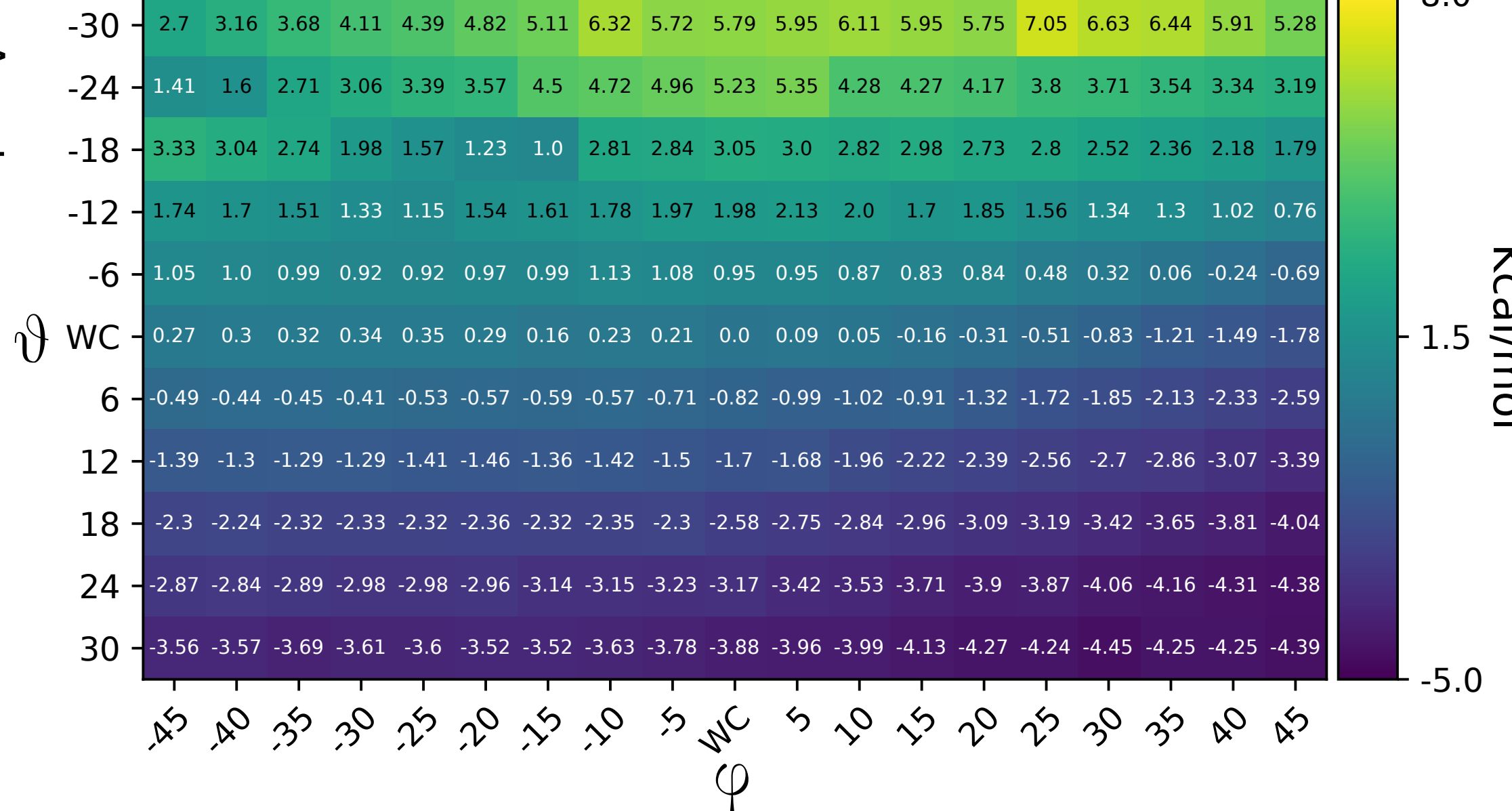
- Example from one replica
- Zeroed at Watson-Crick (WC) (φ, θ)
- insensitive to φ

F-SAPT Interaction Energy



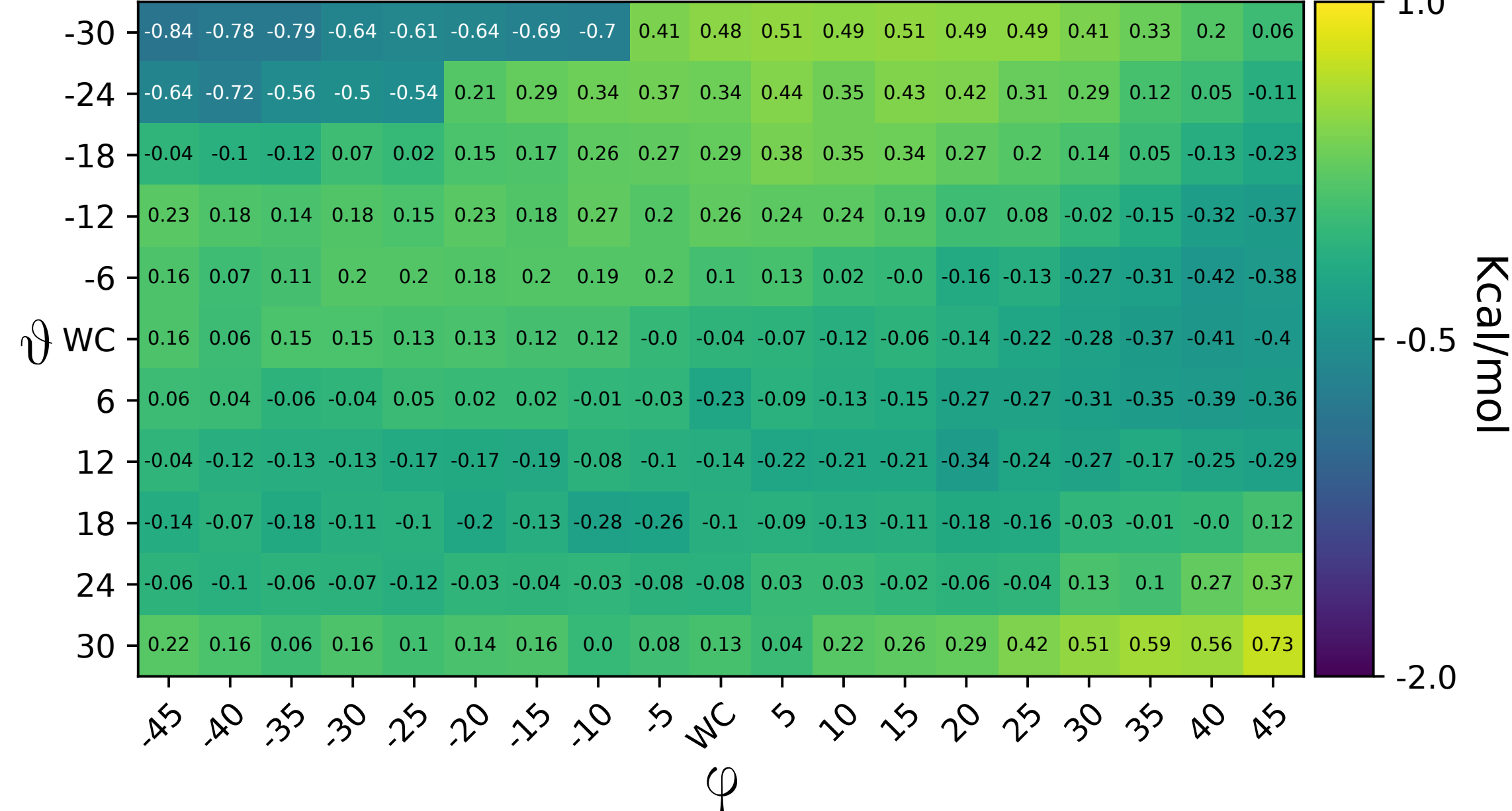
- Used as fitting target
- FG-IE for same conformations
- Subtracted from AMBER IEs below
- Δ FG-IE < Δ AMBER IE

AMBER - F-SAPT

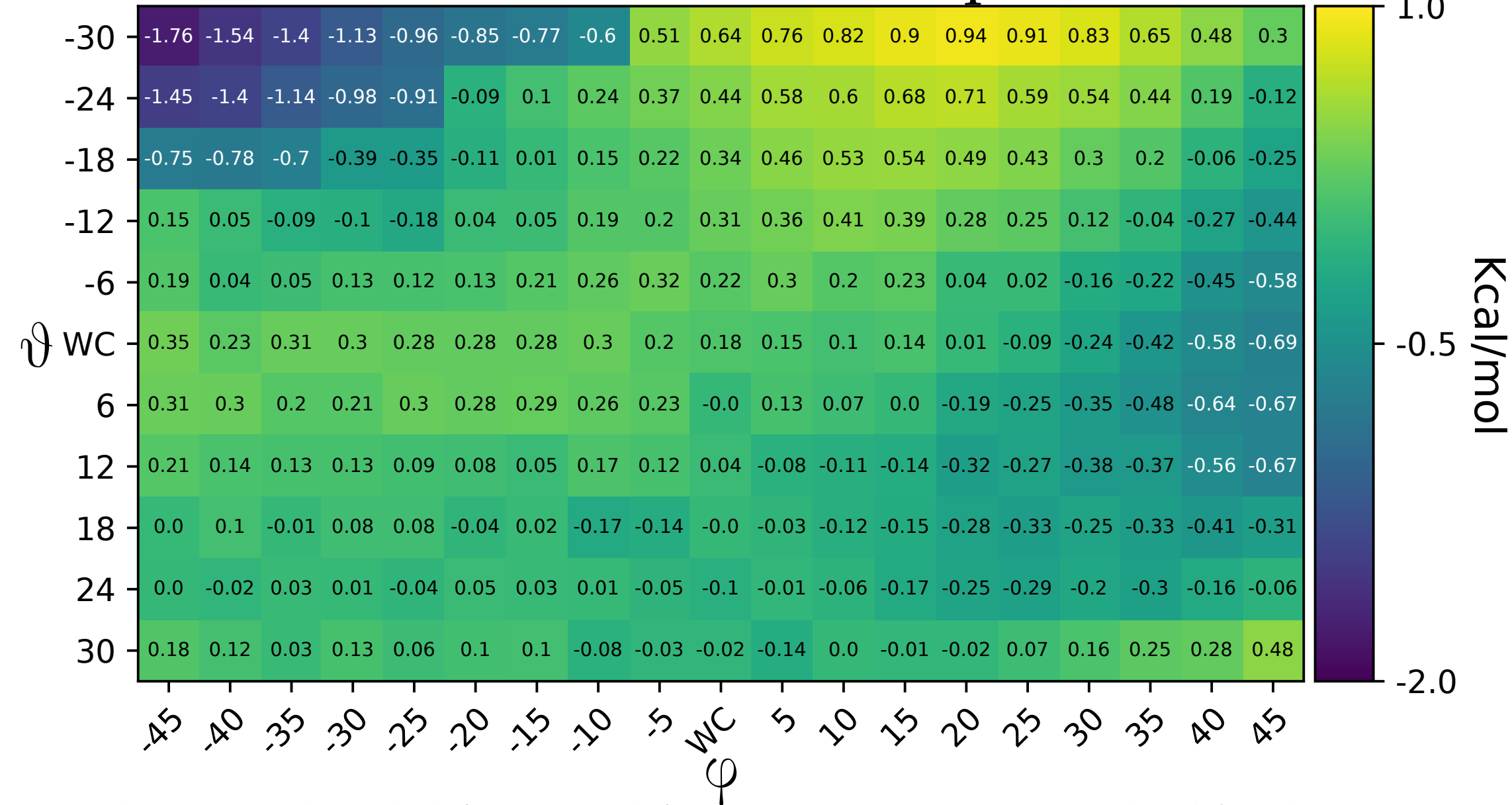


Fit to F-SAPT

Residuals, initial parameters from AMBER



Residuals, random initial parameters



- classic nonbonded functional form
- optimize parameters using basin-hopping (global search)
- Cell values = modeled - target
- IE surfaces (targets) plotted to left
- AMBER target: positive control
- AMBER guess: AMBER parameters
- random guess: Random parameters
- Guess: RMSE (kcal/mol)
 - AMBER: 0.26075
 - random: 0.26069
 - AMBER target: 0.00488
- Constraints
 - H parameters must be equal
 - H charge > 0
 - 1 < partial charges < 1
 - all VDW parameters > 0

Conclusions

- Fitting without the restrictions of Atom type or broader context accurately predicted.
- best-possible case for a parameter set or model. Unzeroed interaction energies in those regions are similar.
- regions very close to WC are more **Classic force field functional form cannot match QM.**

Future Directions

- Need greater diversity of systems
- Cross-validate above fits
- Compare parameters to AMBERs
 - scan other HBs in GC pair
 - scan AU pair
 - structures from full nonbond fit
- fit all replicas simultaneously
- Scan distance with this system
- Competitive fit of trial patch terms
- Fit patch term in the context of refitting other parameters

⇒ See our other poster, **B109**, on Tuesday in RNA Structure and Dynamics!

References

- Parrish, R. M.; Parker, T. M.; Sherrill, C. D. *J. Chem. Theory Comput.* 2014, 10 (10), 4417–4431.
- Bergonzo, C.; Henriksen, N. M.; Roe, D. R.; Cheatham, T. E. *RNA* 2015, 21 (9), 1578–1590.

What is the orientation dependence for RNA base-base hydrogen bonds? Can standard force fields mimic this behavior?

Strategy

- Scan of functional group interaction energy over HB orientation
- Guanine-Cytosine (GC) Nucleoside
 - most stable pair
 - largest dipole moment
 - no costly phosphate in F-SAPT
- Explore GC → reaction coordinate
- Distort Watson-Crick (WC) HB
- Comparison of functional form, not parameters → model fitting
- Capability of model → quality of fit
- Favor classic functional form by
 - Overdetermined fitting set
 - No transferability
 - Just fit atoms in HB
 - Just constrain chemically identical atoms (here, H)
 - 1 ≤ atomic charges ≤ 1
 - 0 ≤ hydrogen charges ≤ 1
- Design patch terms
- Fit patch terms in context of global refit of nonbonded terms

Methods

Scan

- Nucleoside dimer built in tLeap
- Manipulated using restrained Simulated Annealing (SA) in openMM 7.x
- φ: dihedral between carbonyl C, O (C), amino N and C (G)
- θ: angle between carbonyl C, carbonyl O (C), and amino N (G)
- WC (φ, θ) = (27.9°, 116.4°)
- Restraints
 - Bumper restraint

$$B(r_{ij}) = \begin{cases} \frac{k}{2}(r_{ij} - r_0)^2 & r_{ij} < r_0, \\ 0 & r_{ij} \geq r_0 \end{cases}$$

- * O2'(C) to C1'(G), $r_0 = 10 \text{ Å}$
- * O3'(C) to C1'(G), $r_0 = 10 \text{ Å}$

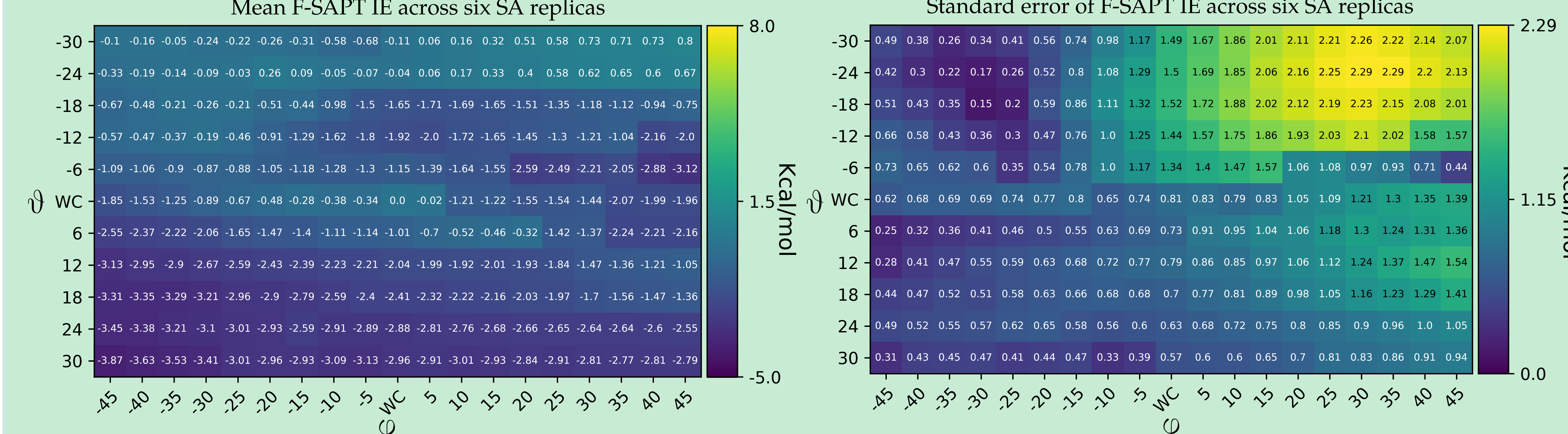
Interaction Energies from F-SAPT

- Used Psi4 v1.2 to perform F-SAPT
- Selected recommended parameters
 - Basis: jun-cc-DVPZ
 - Freeze-Core: True
 - Guess: SAD
- Computed IE for each conformation
- FG partition in postprocessing
- Chose **Functional Groups**
 - C: carbonyl C and O
 - G: amino N, both Hs, and C
- Doesn't cut bonds with DOCC > 0.1

Coordinate Comparison & Fits

- Structures iteratively aligned to average with LOOS 3.0 (See QR)
- mean RMSD from average per (φ, θ) plotted as heatmap
- fits performed using basin-hopping routine from scipy.optimize(1.2.1)
 - 1000 Metropolis-Hastings Monte-Carlo steps per fit.
 - Local optimization using L-BFGS-B, then accept.

How sensitive is F-SAPT FG-IE to distant atoms?



- Calculated 209×6 = 1254 QM IEs
- Colorbar from AMBER IE scan
- IEs lower, smoother than AMBER
- Retains trends in φ
- Computed SEM using 6 – 1 samples
- No SEM for AMBER > 0.1 kcal/mol
- Variability is from small changes in distant atoms