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SHAMBHUNATH INSTITUTE OF ENGINEERING AND TECHNOLOGY, PRAYAGRAJ

Subject Code : BAS202

Subject : Engg. Chemistry

Course: B.Tech.

SEMESTER: II

THIRD SESSIONAL EXAMINATION, EVEN SEMESTER, (2024-2025)

Branch : All

Time -1hr 30 min

Maximum Marks - 30

1. Attempt any FIVE questions.

Q N	QUESTION	Marks	CO	BL
a.	100 ml of water sample has a hardness equivalent to 12.5 ml of 0.08 N MgSO_4 . What is its hardness in degree French.	2	CO4	L5
b.	Why does $\text{Mg}(\text{HCO}_3)_2$ require double amount of lime for softening?	2	CO4	L1
c.	Why do we express hardness of water in terms of CaCO_3 equivalent?	2	CO4	L3
d.	Calculate the weight and volume of air required for combustion of 3Kg of carbon.	2	CO4	L1
e.	Give the composition of biogas with the help of diagram.	2	CO4	L1
f.	Give the condition when $\text{GCV}=\text{NCV}$.	2	CO4	L2

2. Attempt any ONE of the following.

Q N	QUESTION	Marks	CO	BL
a.	The ultimate analysis of a coal gave the following result: C=84%, S=1.5%, N=0.6%, H=5.5%, and O=8%. Calculate the GCV and NCV of the coal using Dulong's formula.	5	CO4	L1
b.	Why ion exchange method is considered more efficient than zeolite process in some applications?	5	CO4	L2
c.	Calculate the amount of lime (90%) and soda (98%) for the treatment of 1000000 Liters of water containing: $\text{Ca}(\text{HCO}_3)_2 = 8.1\text{ppm}$, $\text{CaCl}_2 = 33.3\text{ppm}$, $\text{HCO}_3^- = 91.5\text{ppm}$, $\text{MgCl}_2 = 38\text{ppm}$, and $\text{Mg}(\text{HCO}_3)_2 = 14.6\text{ppm}$. The coagulant $\text{Al}_2(\text{SO}_4)_3$ was added at the rate of 17.1 mg/L of water.	5	CO4	L2

3. Attempt any FIVE questions.

Q N	QUESTION	Marks	CO	BL
a.	What is meant by functionality of monomers?	2	CO5	L1
b.	What is the limitation of raw rubber? How can it improve?	2	CO5	L1
c.	Why do all simple organic molecules not produce polymers?	2	CO5	L2
d.	Write the structure of Ziegler-Natta catalyst.	2	CO5	L1
e.	Give a brief account of the applications of conducting polymers.	2	CO5	L1
f.	How will you synthesise Nylon-6 from cyclohexanone oxime.	2	CO5	L2

4. Attempt any ONE of the following.

Q N	QUESTION	Marks	CO	BL
a.	Write preparation, properties and uses of: (i) PET (ii) PMMA .	5	CO5	L1
b.	How will you synthesise 2-hydroxypropane and t-butyl alcohol from CH_3MgBr .	5	CO5	L2
c.	Differentiate between thermoplastic and thermosetting polymers with examples.	5	CO5	L1

Bloom's Taxonomy Level (BL) :-Remember (L1), Understanding (L2), Apply (L3), Analyze (L4), Evaluating(L5), Creating(L6)

