

# An introduction to Molecular Dynamics

EMBO, June 2016

What is MD?

## Loading Tables

- Nodes and edges can have data associated with them

- Gene expression data

- Mass spectrometry data

- Protein structure information

- Gene Ontology terms, etc.

- Cytoscape supports multiple data types:

- Numbers, Text, Logical, Lists...

### Loading Networks

#### Loading Tables

- Nodes and edges can have data associated with them

- Gene expression data

- Mass spectrometry data

- Protein structure information

- Gene Ontology terms, etc.

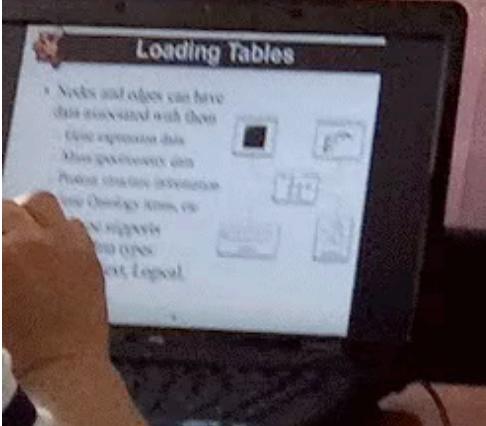
- Cytoscape supports multiple data types:

- Numbers, Text, Logical,

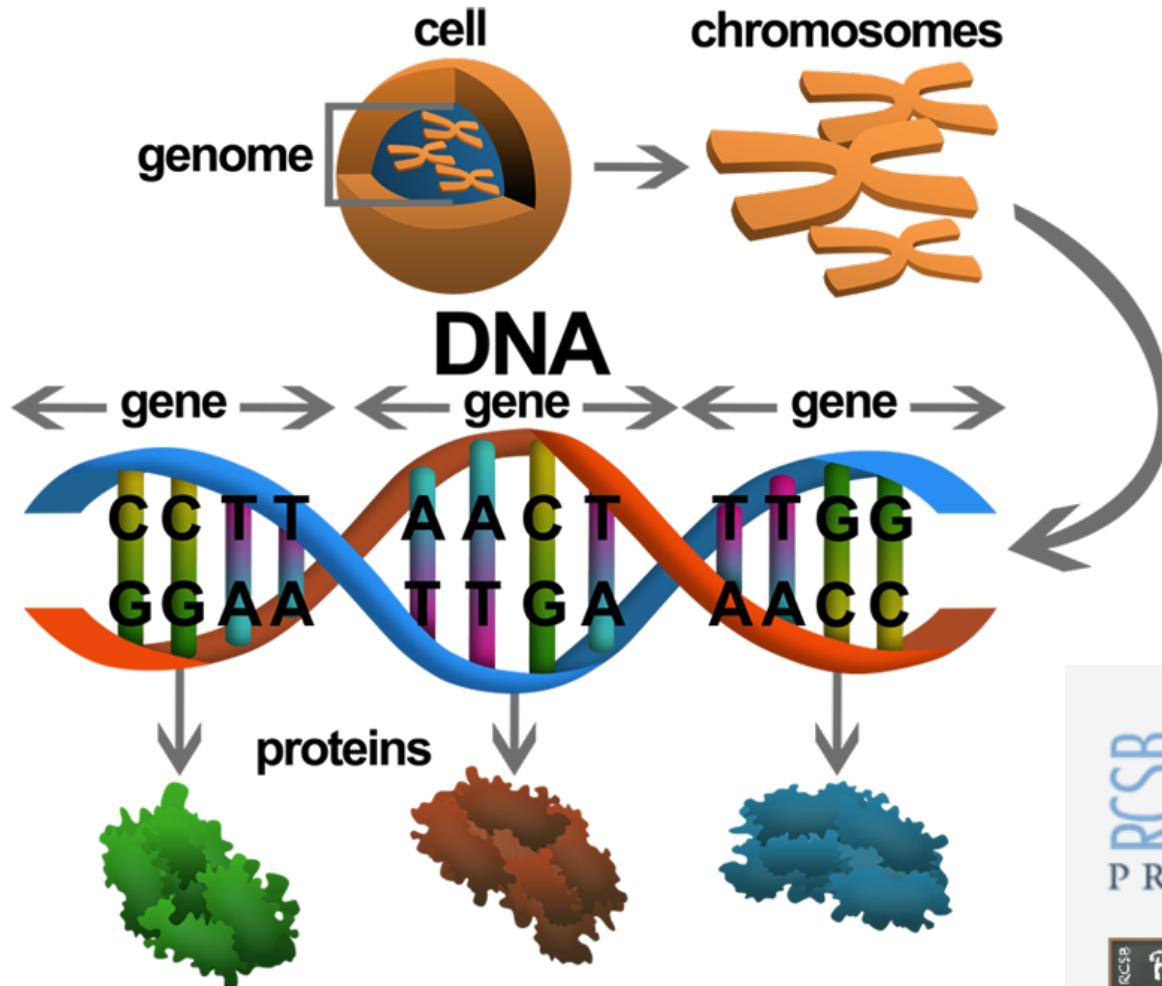
- Lists...

## Loading Tables

- Nodes and edges can have data associated with them
  - Gene expression data
  - Mass spectrometry data
  - Protein structure information
  - Ontology terms
  - Shape supports multiple data types



# Introduction to Molecular Dynamics



**RCSB PDB** An Information Portal to  
119137 Biological Macromolecular Structures

PDB-101

WORLDWIDE PROTEIN DATA BANK

EMDataBank Unified Data Resource for 3DEM

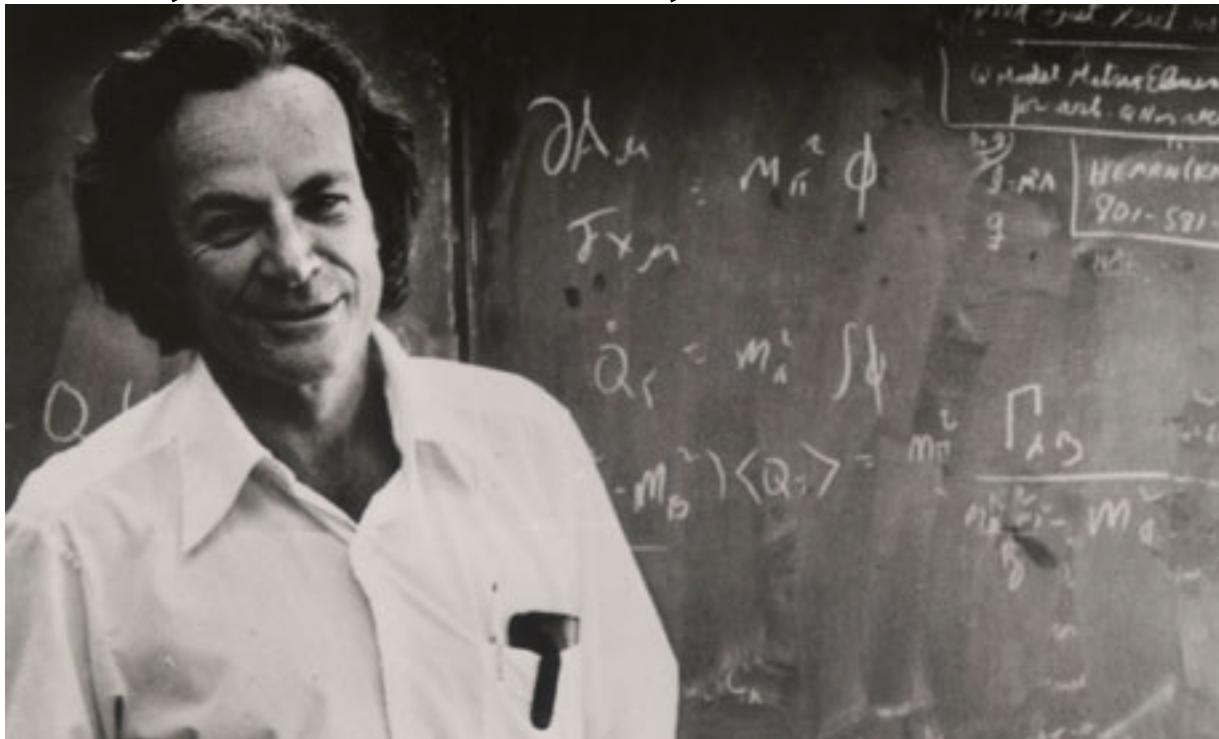
ndb NUCLEIC ACID DATABASE

Structural Biology Knowledgebase

# Introduction to Molecular Dynamics

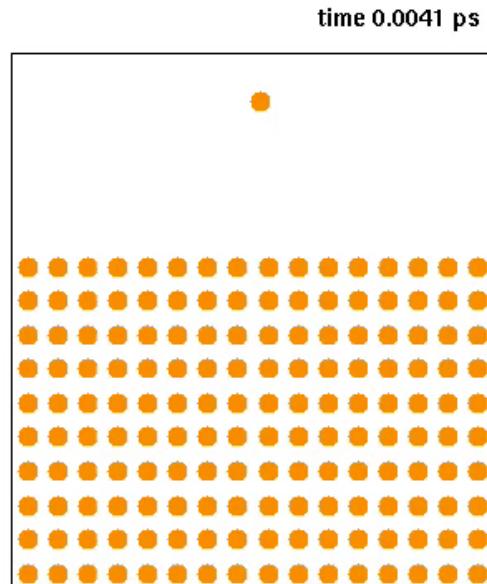
“everything that living things do can be understood in terms of the jiggling and wiggling of atoms.”

*The Feynman Lectures in Physics* vol. 1, 3-6 (1963)



# Introduction to Molecular Dynamics

- Molecular dynamics is a technique for computer simulation of complex systems, modelled at the atomic level.



- Example of a molecular dynamics simulation in a simple system: deposition of a single Cu atom on a Cu surface.

# Introduction to Molecular Dynamics

- Dynamics
  - displacements from average structure - e.g., local sidechain motions that act as conformational gates in oxygen transport myoglobin, enzymes, ion channels
- Thermodynamics
  - equilibrium behaviour - e.g., energy of ligand binding

# Simulations: Modelling Strategies

Molecular  
Simulations

*Ab initio* QM  
Methods

Force Field  
Methods

# Quantum Mechanics

- postulates and theorems of quantum mechanics form the rigorous foundation for the prediction of observable chemical properties from first principles.
  - microscopic systems are described by wave functions that completely characterise all the physical properties of the system
  - operators applied to the wave function allow one to predict the probability of the system having a value or range of values.

# Quantum mechanics vs Force Field methods

- QM deals with electrons in system
  - Accurate
  - Can deal with reactions (bond breaking etc.)
  - Often used to parameterise force fields
  - Large number of particles means infeasibly time-consuming for molecules as large as proteins
  - Static models only (no time)
- FF methods
  - Molecular mechanics
  - Cannot answer questions that depend on electron distribution in a molecule
  - But fast and surprisingly useful

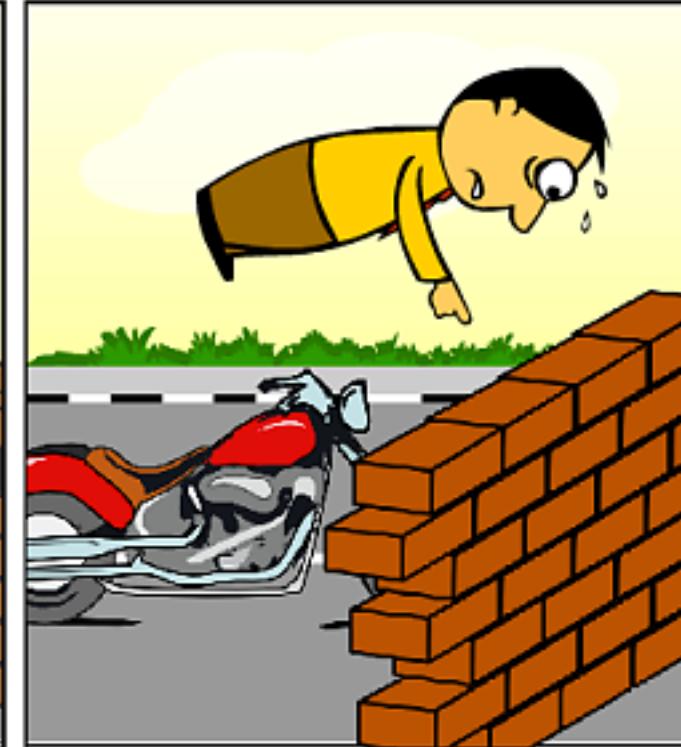
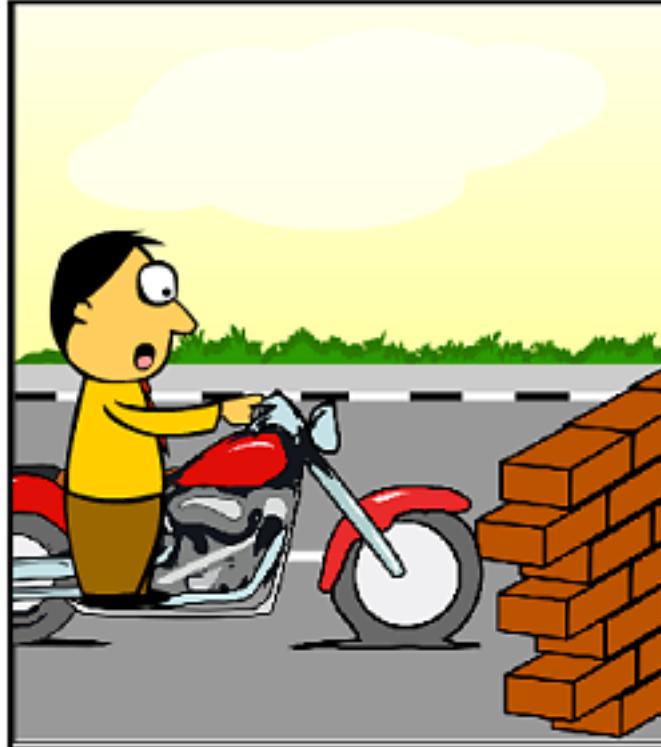
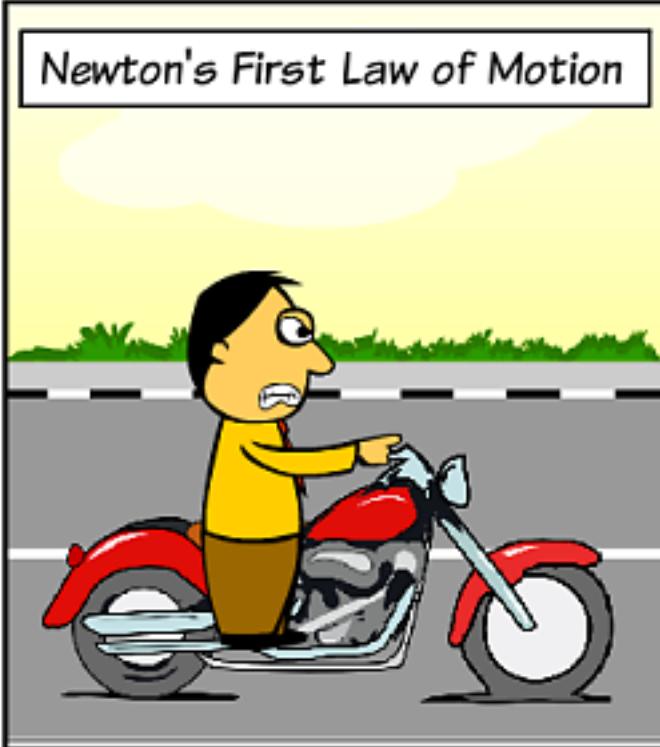
# Introduction to Molecular Dynamics

MD simulations boil down to numerically integrating Newton's equations of motion

*Newton's Laws of Motion*

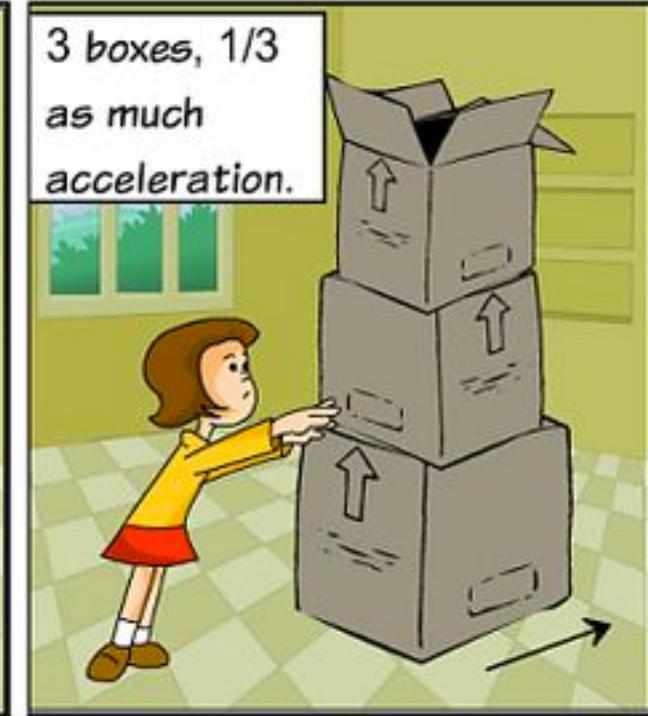
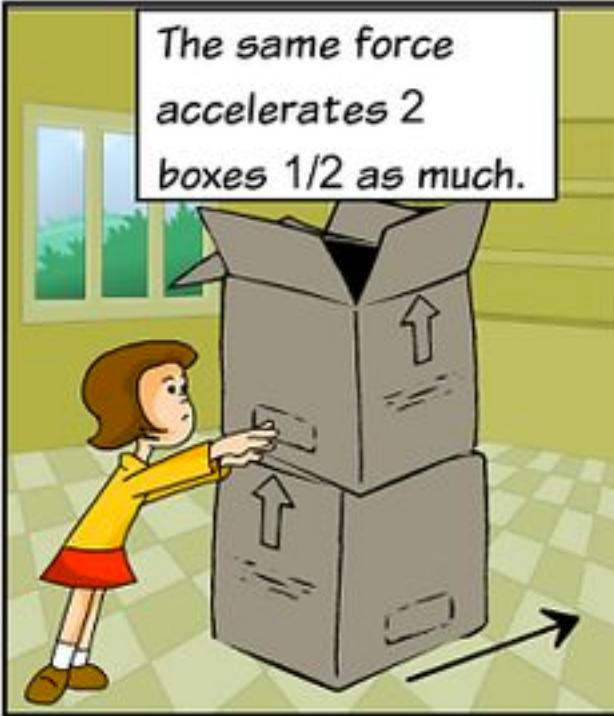
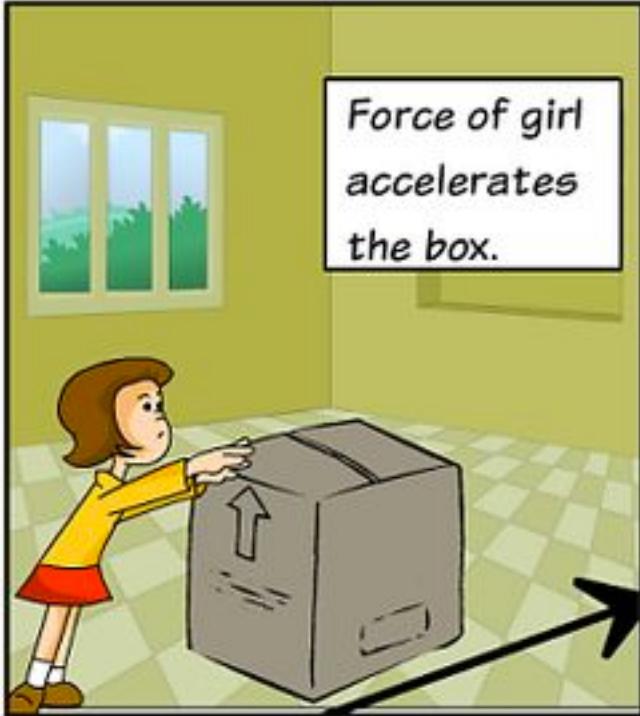
**NEWTON'S FIRST LAW** - BY AMAMAS

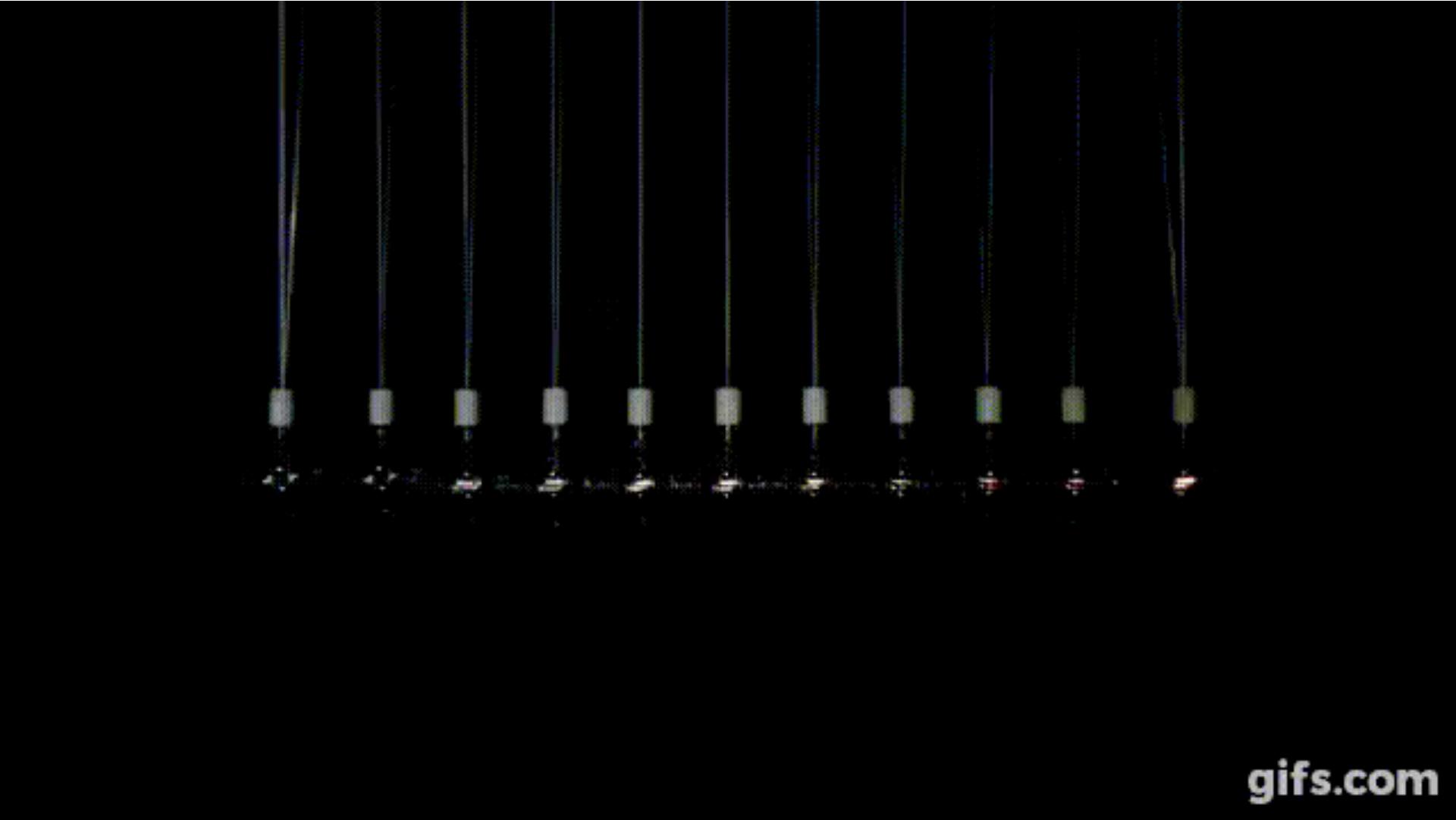
WWW.TOONDOO.COM



**NEWTON'S 2ND LAW OF MOT** - BY APEPEI

WWW.TOONDOO.COM





gifs.com

# Introduction to Molecular Dynamics

MD simulations boil down to numerically integrating Newton's equations of motion

## *Newton's Laws of Motion*

1. A body maintains its state of rest or of uniform motion in a straight line, unless acted upon by a force.
2. The applied force is equal to the rate of change of momentum.
3. For every action, there is an equal and opposite reaction.

# Introduction to Molecular Dynamics

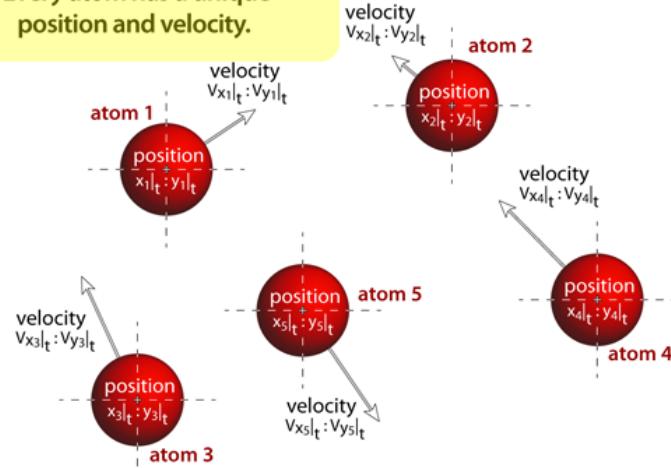
- Use Newtonian mechanics to calculate the net force and acceleration experienced by each atom
- Each atom  $i$  is treated as a point with mass  $m_i$  and fixed charge  $q_i$
- Determine the force  $F_i$  on each atom:

$$\vec{F}_i = m_i \vec{a}_i = -\frac{dV}{dr_i}$$

- Use positions and accelerations at time  $t$  to calculate new positions at time  $t_n + t_{n+1}$

# Introduction to Molecular Dynamics

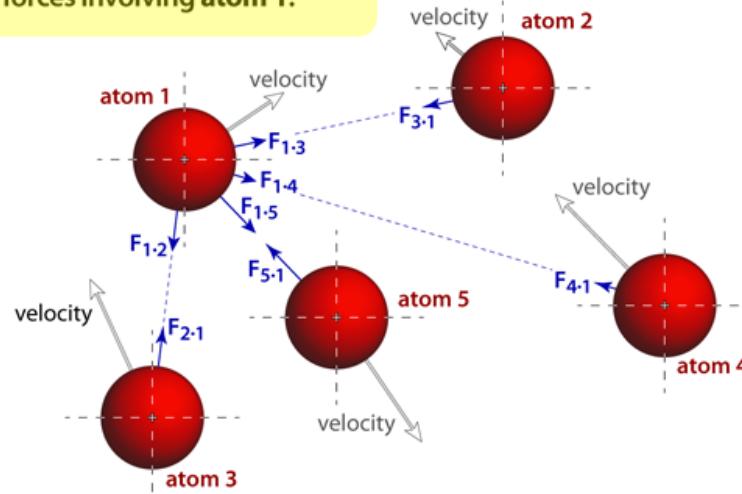
Every atom has a unique position and velocity.



$$F = ma$$

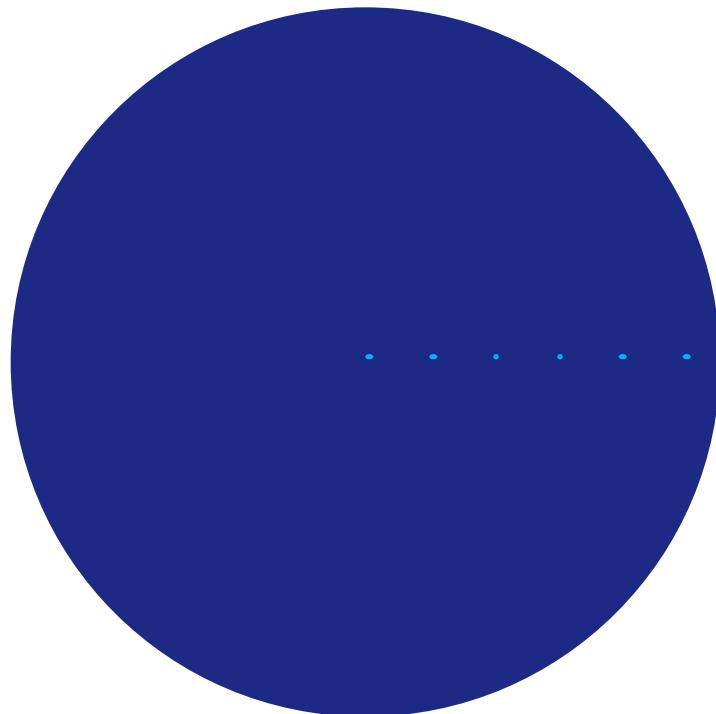
Forces are derived from interatomic potential functions (analytical approximations)

Calculate the interatomic forces involving atom 1.

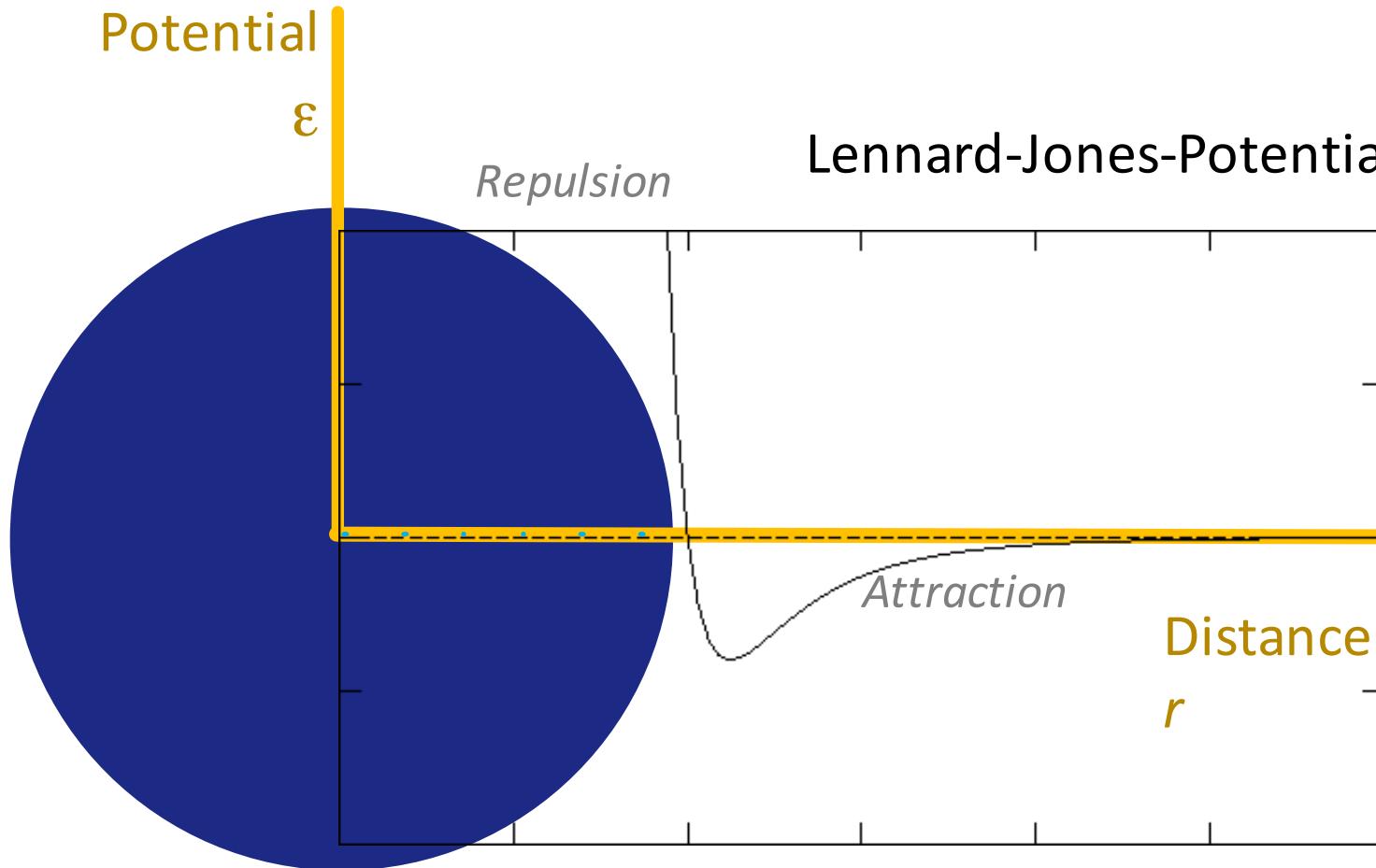




# Introduction to Molecular Dynamics



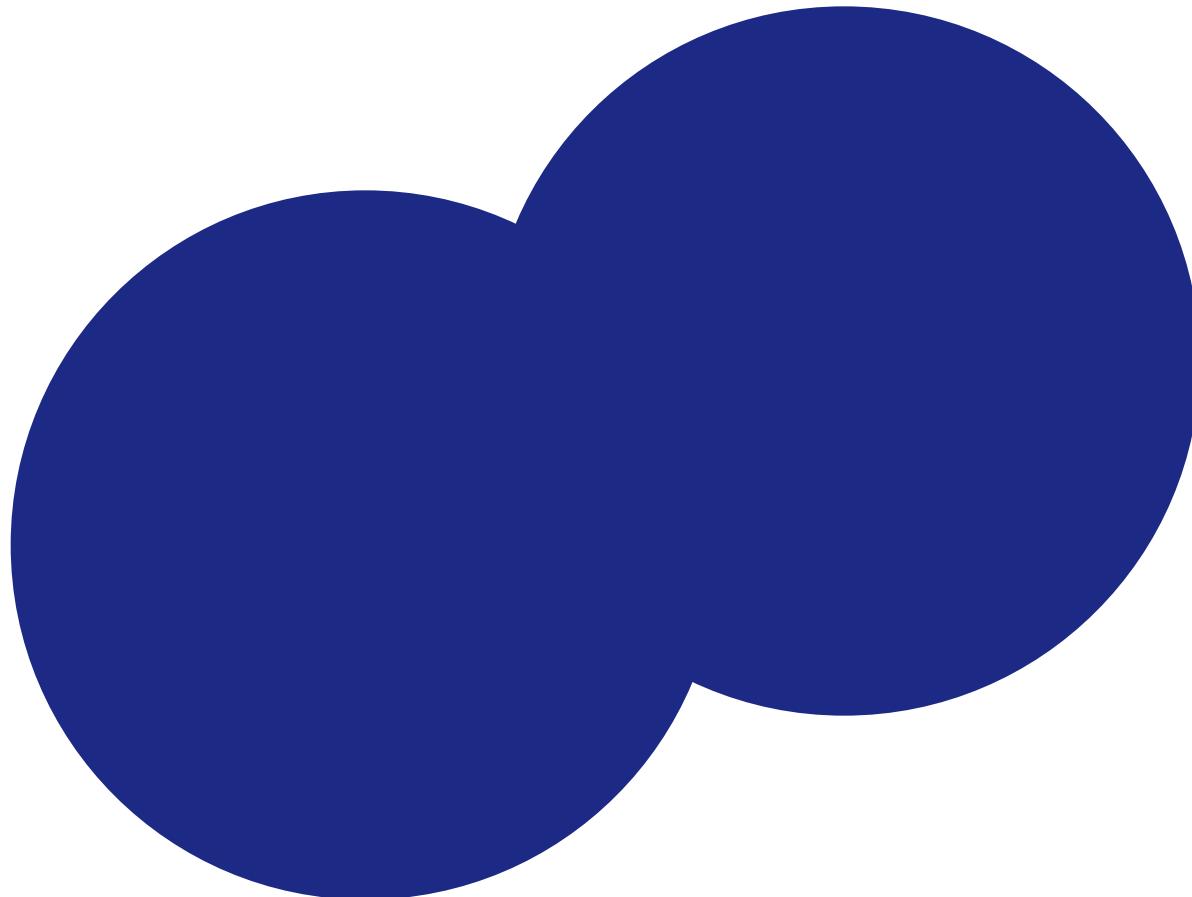
# Introduction to Molecular Dynamics



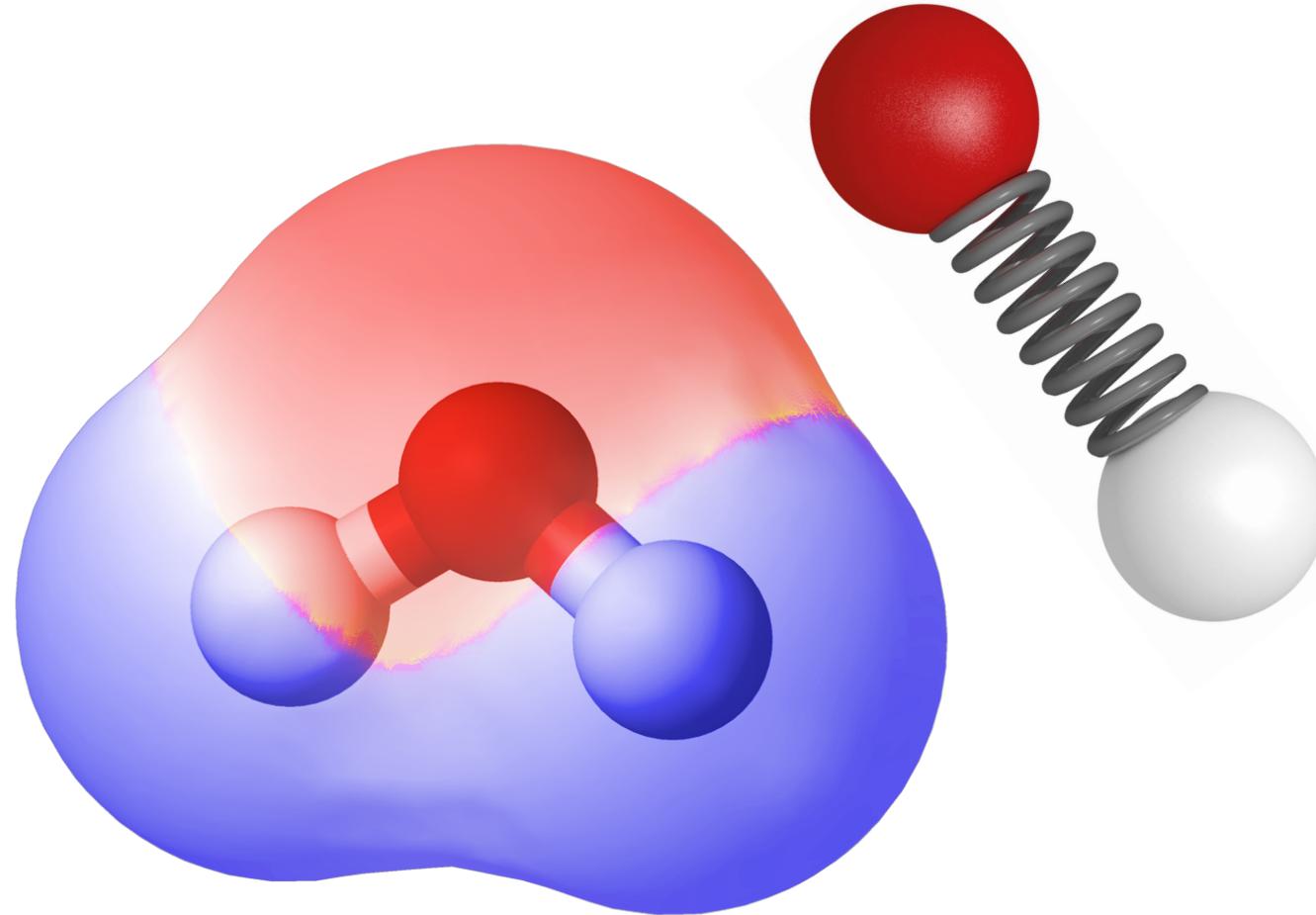
Lennard-Jones-Potential

*mathematically simple model  
that approximates the  
interaction between a pair of  
neutral atoms or molecules*

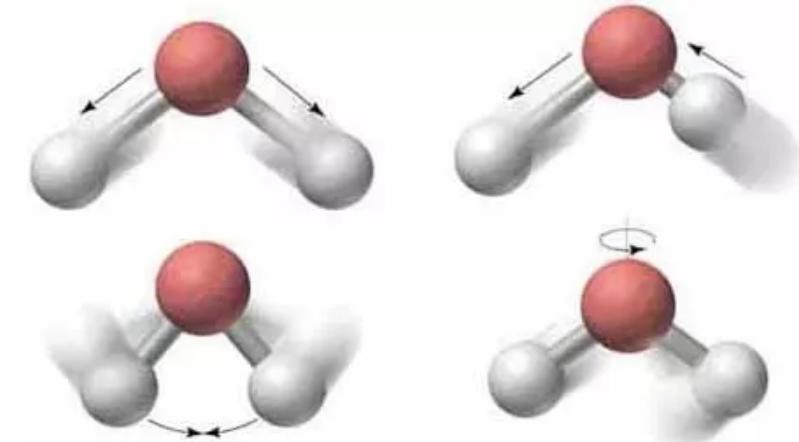
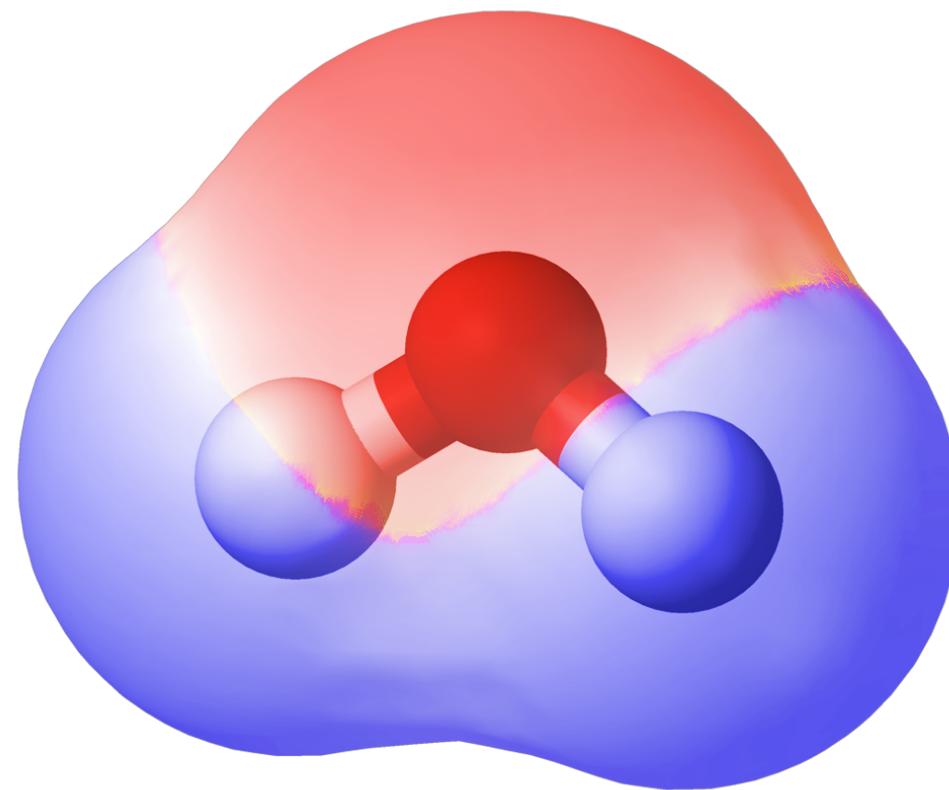
# Introduction to Molecular Dynamics



# Introduction to Molecular Dynamics



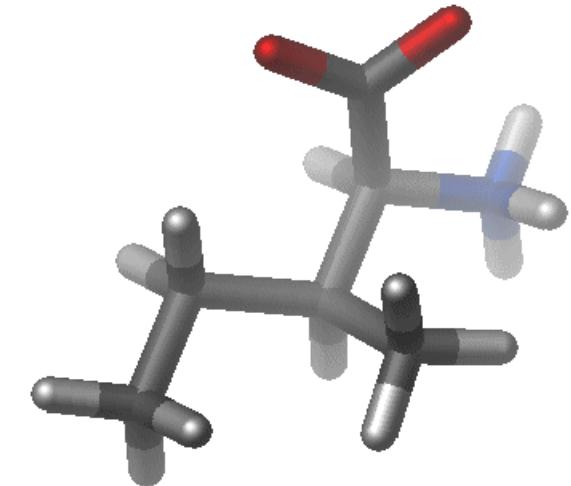
# Introduction to Molecular Dynamics



# Introduction to Molecular Dynamics

A force field is made up by the contributions of many terms that represent the different types of interactions between the atoms of the protein molecule (energy function)

- van der Waals energy
- Electrostatic energy
- Hydrogen bond
- Bond energy
- Bond angle energy
- Dihedral angel energy



# Introduction to Molecular Dynamics

A force field is made up by the contributions of many terms that represent the different types of interactions between the atoms of the protein molecule (energy function)

- van der Waals energy
- Electrostatic energy
- Hydrogen bond
- Bond energy
- Bond angle energy
- Dihedral angel energy

$$E_{\text{total}} = E_{\text{bonded}} + E_{\text{non-bonded}}$$

$$E_{\text{bonded}} = E_{\text{bond}} + E_{\text{angle}} + E_{\text{dihedral}}$$

$$E_{\text{electrostatic}} + E_{\text{vander Waals}} = E_{\text{non-bonded}}$$

# Example of a Single Step

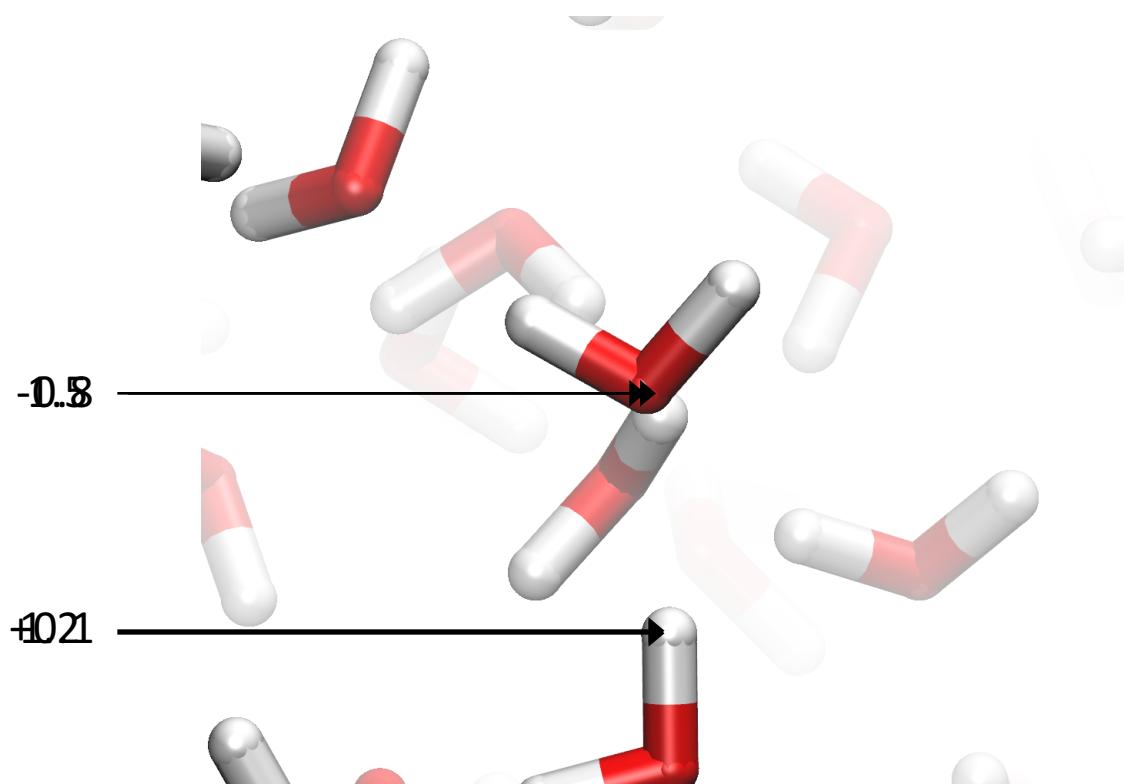
- Calculates forces acting on each atom
- Predicts new position after 2 fs ( $10^{-15}$ )
- Takes single step and recalculates forces
- Repeat...

Coulomb's Law:

$$F = \frac{q_i q_j}{4\pi\epsilon_0 r_{ij}}$$

Lennard-Jones:

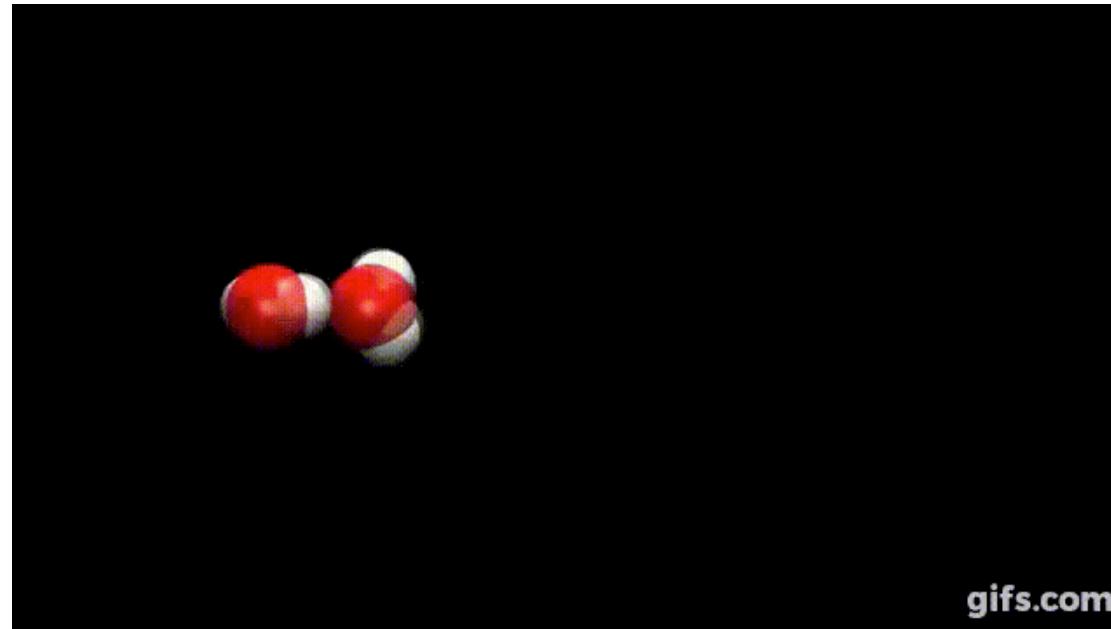
$$\left(\frac{r_{0ij}}{r_{ij}}\right)^{12} - 2\left(\frac{r_{0ij}}{r_{ij}}\right)^6$$



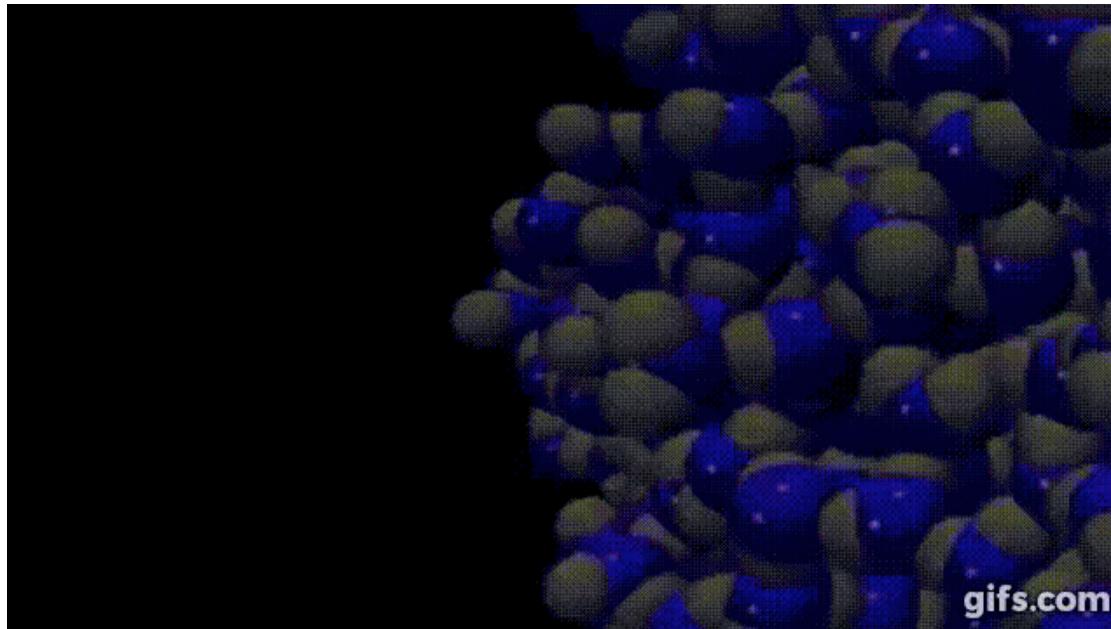
# What is Molecular Dynamics

- Use force fields to describe molecular properties
  - These force fields are tuned to reproduce experimental observables
  - Specialised for particular molecules (e.g. Glycam)
- $\mu\text{s}$  ( $10^{-6}$ ) simulations of large (100k atoms) are now feasible
  - Exponential growth in computing speed

# Introduction to Molecular Dynamics

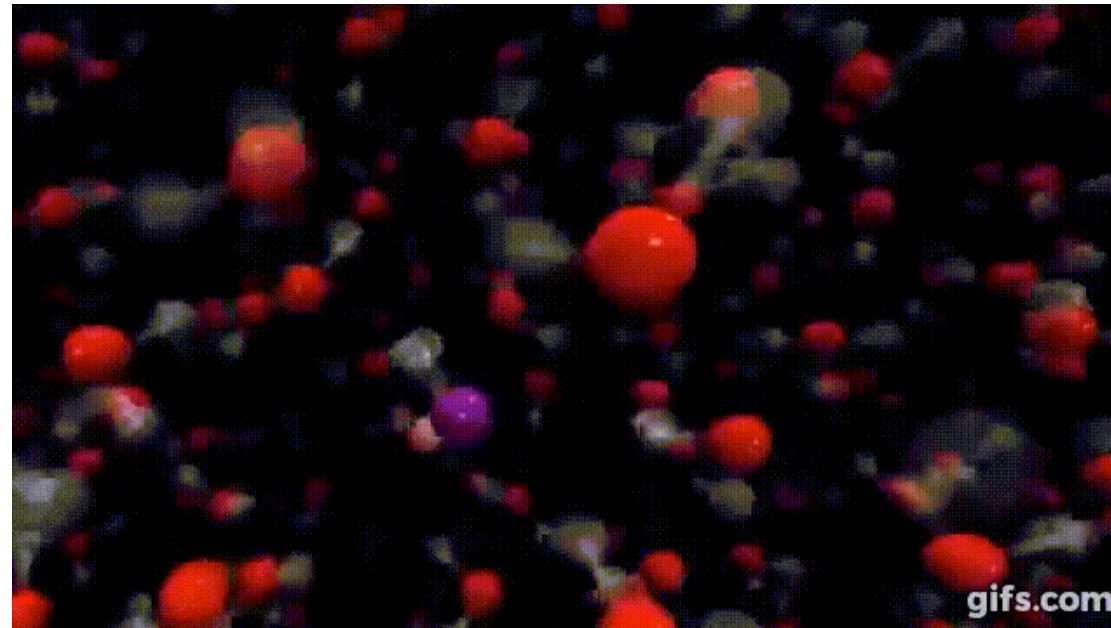


# Introduction to Molecular Dynamics



# Introduction to Molecular Dynamics

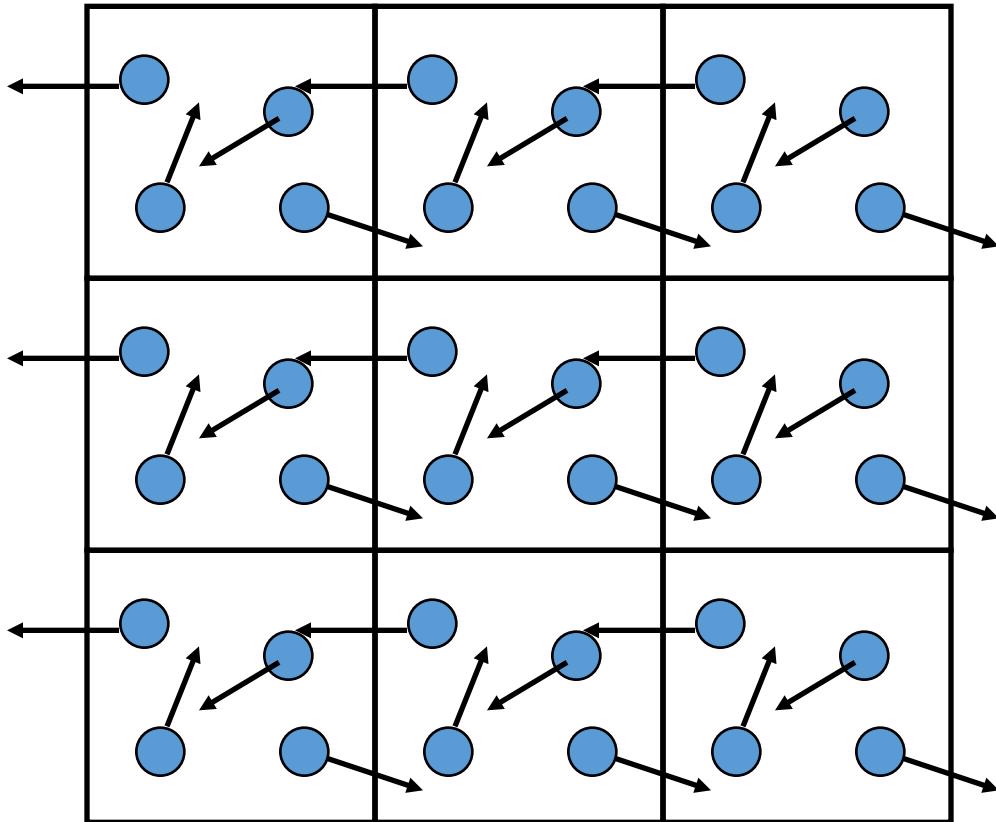
Radial distribution function



*If we plot the average probability of finding an Oxygen in the space around a specific water molecule.*

\*Structure

# Introduction to Molecular Dynamics



- *Periodic Boundary Conditions*
  - your structure doesn't know that it's in a box (non-infinite space).

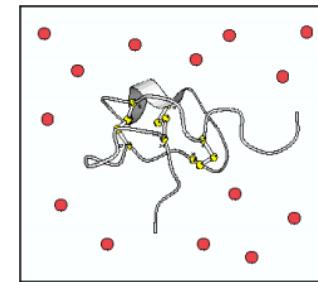
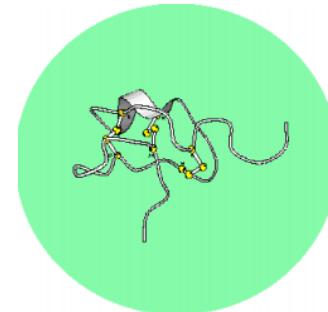
# Introduction to Molecular Dynamics



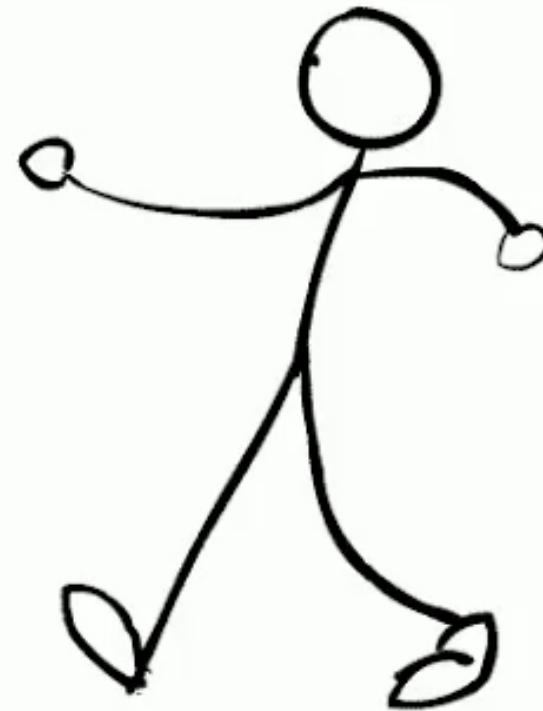
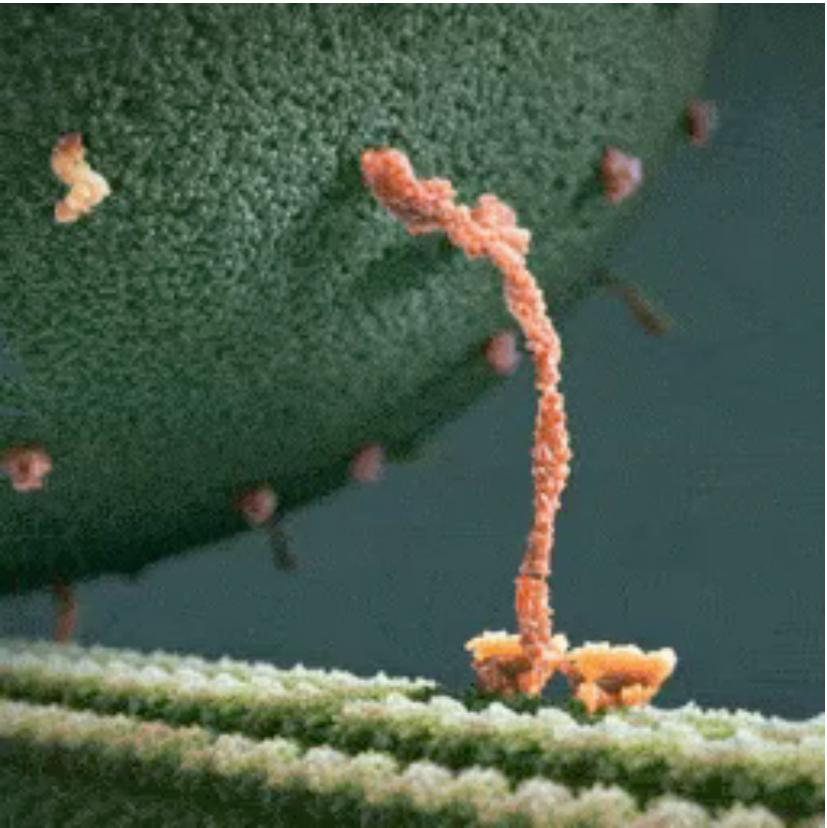
- *Periodic Boundary Conditions*
  - *your structure doesn't know that it's in a box (non-infinite space).*

# Introduction to Molecular Dynamics

- Protocol for an MD simulation
- Initial Coordinates
  - X-ray diffraction or NMR coordinates from the Protein Data Bank
  - Coordinates constructed by modeling (homology)
- Treatment of non-bonded interactions
- Treatment of solvent
  - implicit
  - explicit
- If using explicit treatment of solvent
  - Periodic boundary conditions (PBC)
  - Solvation sphere
  - Active site dynamics

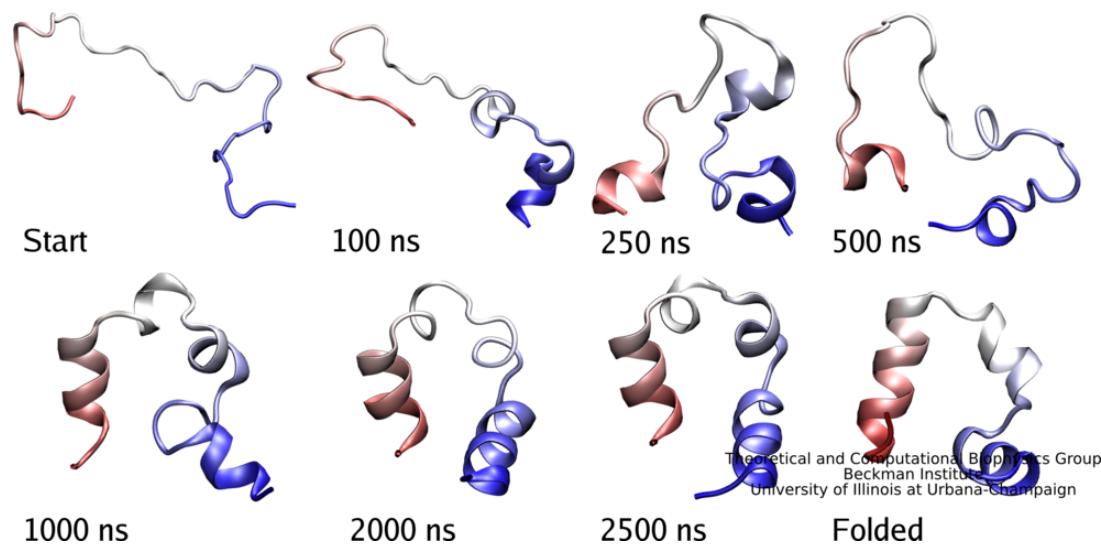


# Introduction to Molecular Dynamics

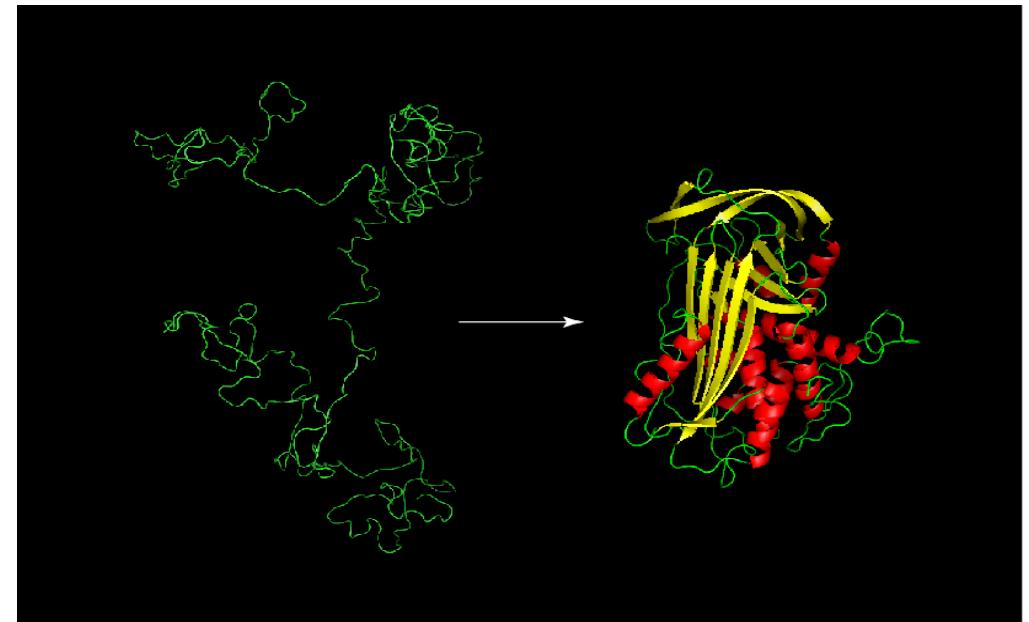


# Introduction to Molecular Dynamics

## Folding pathways



Theoretical and Computational Biophysics Group  
Beckman Institute  
University of Illinois at Urbana-Champaign



# Introduction to Molecular Dynamics

## Protein Interactions

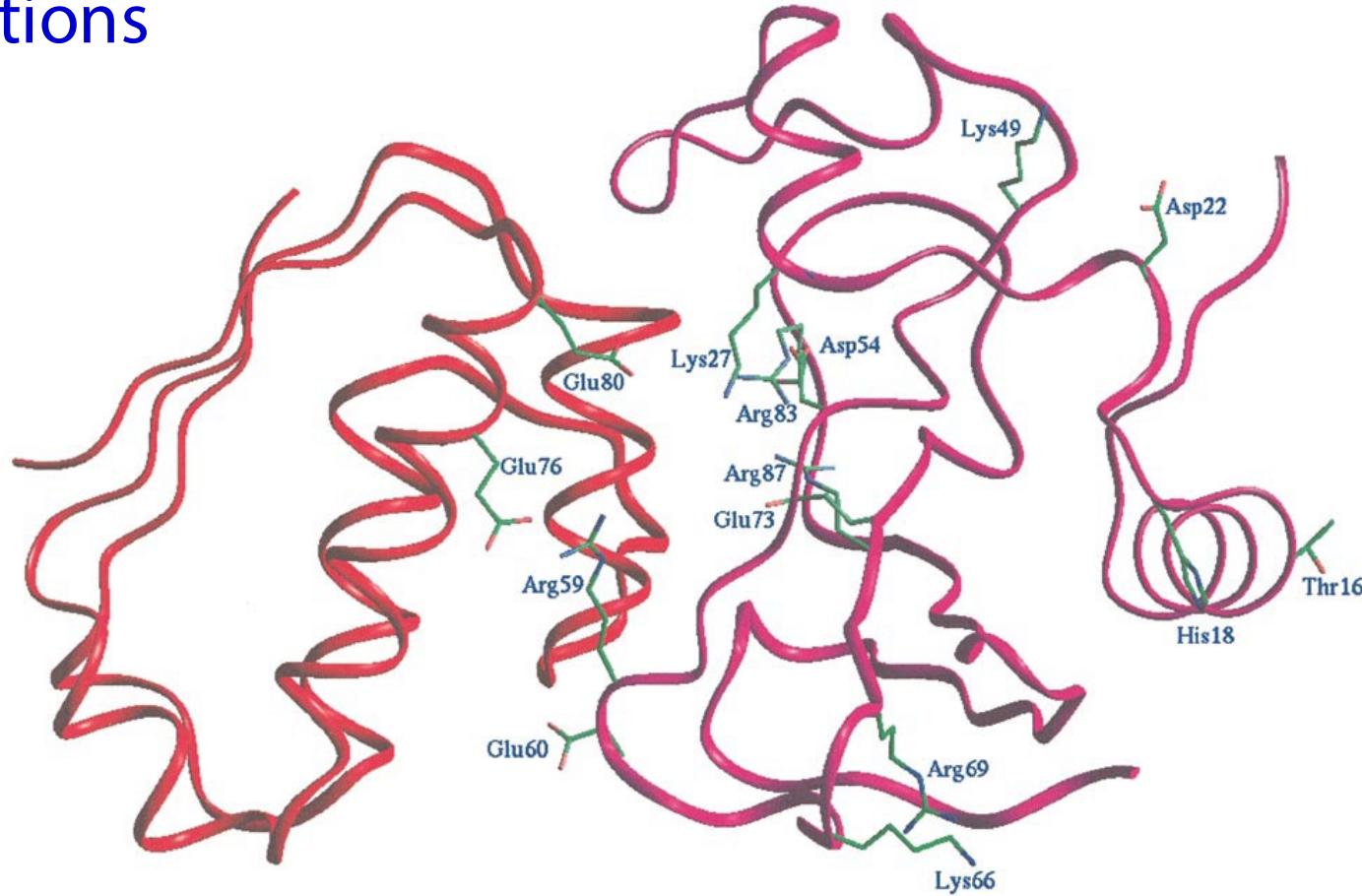


Image: Vijayakumar, et al., *J. Mol. Biol.* 278, 1015 (1998)

# Introduction to Molecular Dynamics

## Binding of Drugs to their Molecular Targets

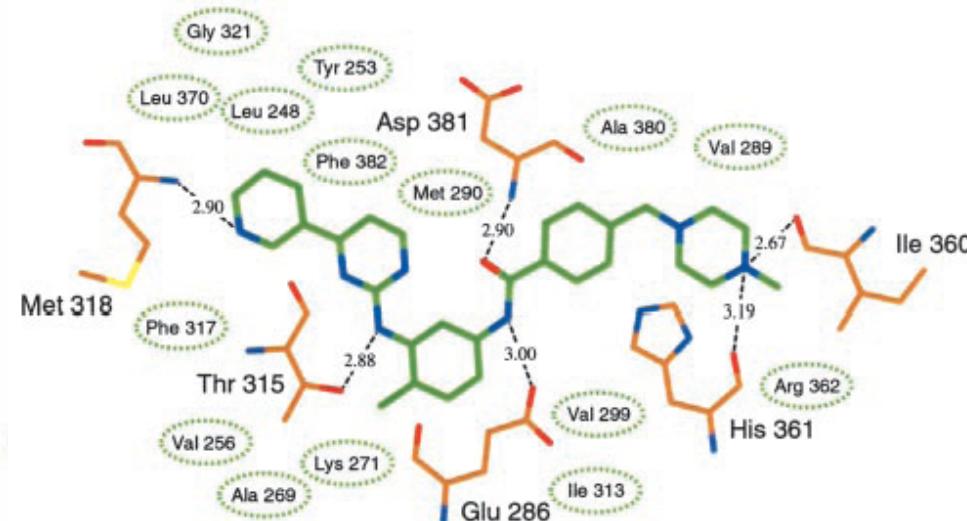
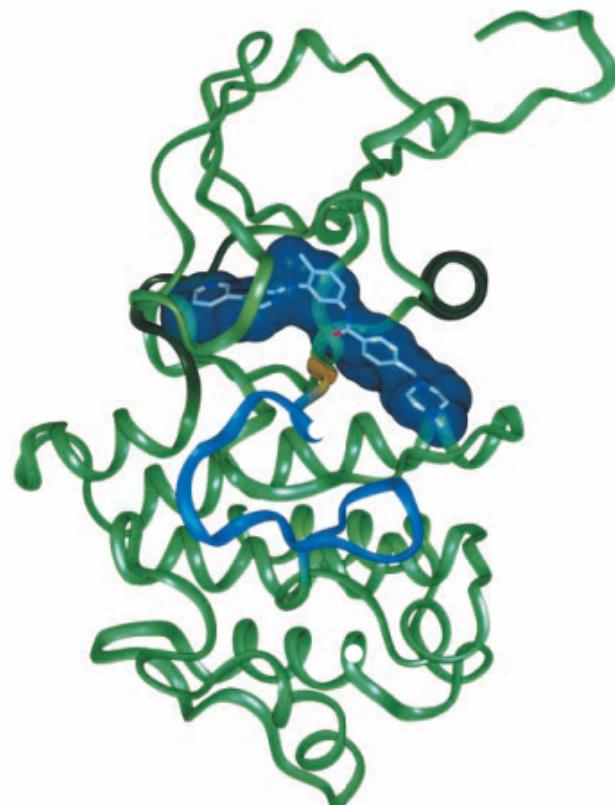
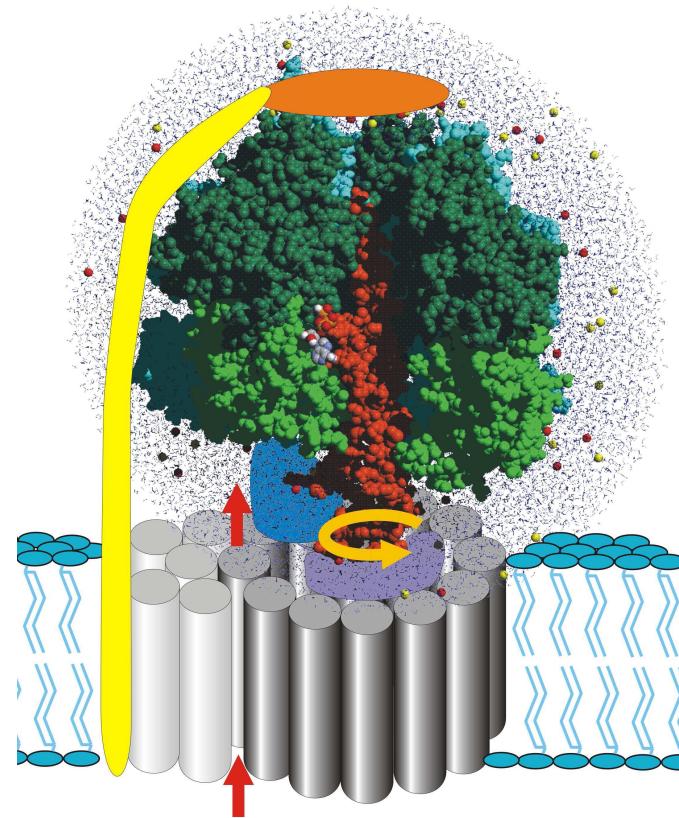
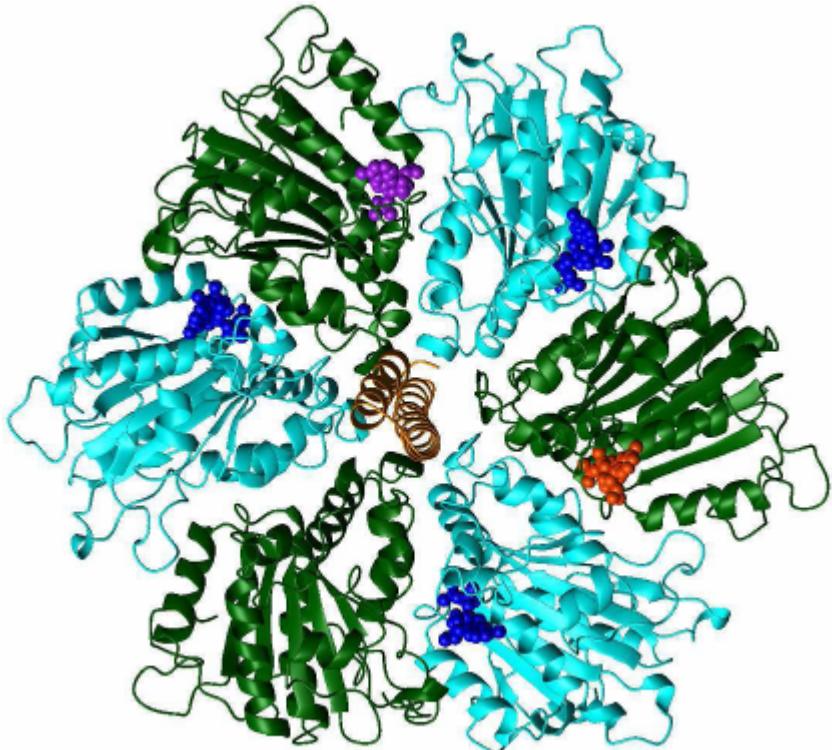


Image: Nagar, et al., *Cancer Res.* 62, 4236 (2002)

# Introduction to Molecular Dynamics

## Mechanisms of Intracellular machines



*Image: H. Grubmüller, in Attig, et al. (eds.),  
Computational Soft Matter (2004)*



“I... a universe of atoms, an atom in the universe.”  
*Feynman*