

# Kinetic Monte Carlo Simulation of Electron Transfer

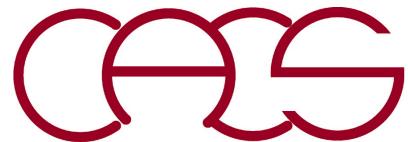
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Aiichiro Nakano

*Collaboratory for Advanced Computing & Simulations  
Dept. of Computer Science, Dept. of Physics & Astronomy,  
Dept. of Chemical Engineering & Materials Science  
Dept. of Biological Sciences  
University of Southern California*

Email: [anakano@usc.edu](mailto:anakano@usc.edu)



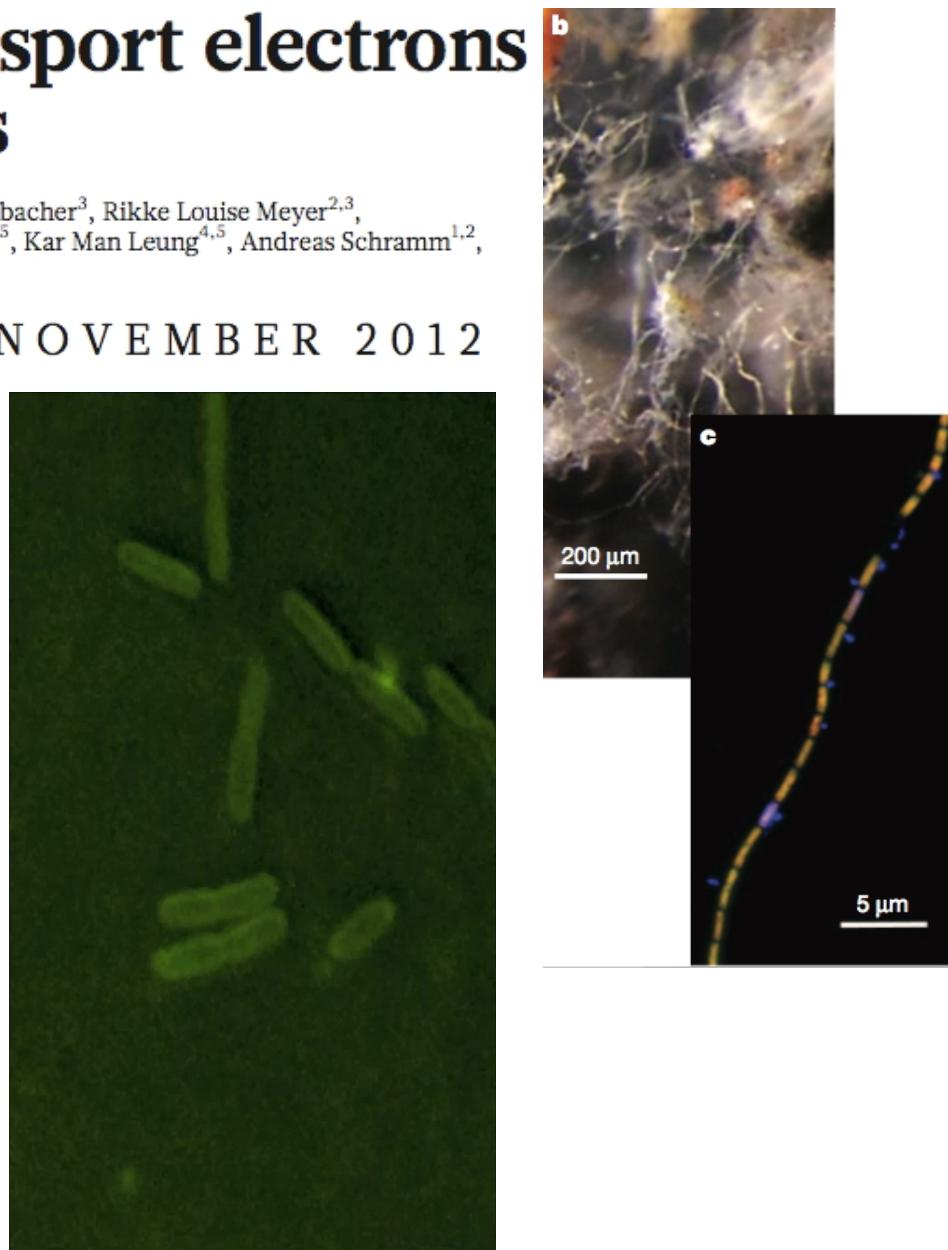
# Biological Electron Transfer

## Filamentous bacteria transport electrons over centimetre distances

Christian Pfeffer<sup>1</sup>, Steffen Larsen<sup>2</sup>, Jie Song<sup>3</sup>, Mingdong Dong<sup>3</sup>, Flemming Besenbacher<sup>3</sup>, Rikke Louise Meyer<sup>2,3</sup>, Kasper Urup Kjeldsen<sup>1</sup>, Lars Schreiber<sup>1</sup>, Yuri A. Gorby<sup>4</sup>, Mohamed Y. El-Naggar<sup>5</sup>, Kar Man Leung<sup>4,5</sup>, Andreas Schramm<sup>1,2</sup>, Nils Risgaard-Petersen<sup>1</sup> & Lars Peter Nielsen<sup>1,2</sup>

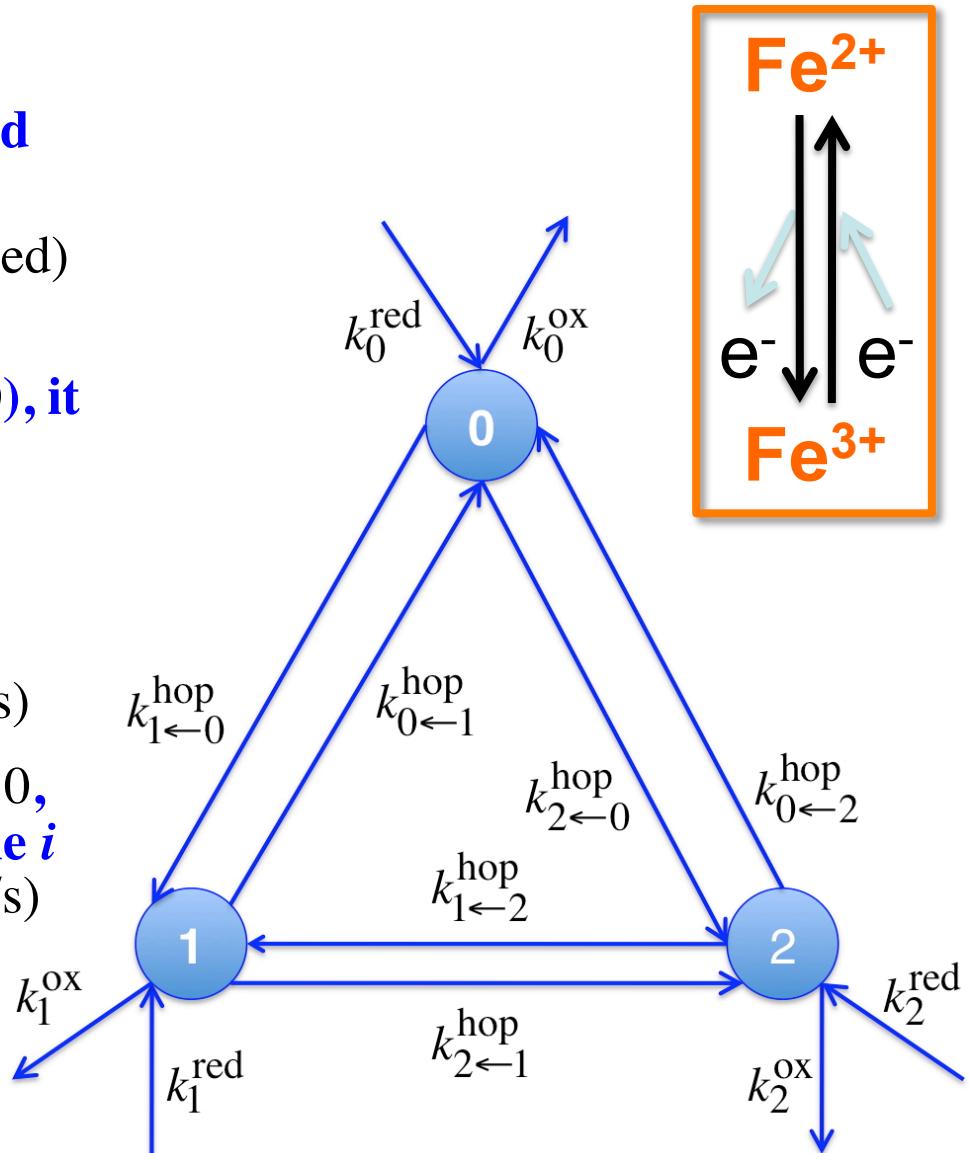
218 | NATURE | VOL 491 | 8 NOVEMBER 2012

Oxygen consumption in marine sediments is often coupled to the oxidation of sulphide generated by degradation of organic matter in deeper, oxygen-free layers. Geochemical observations have shown that this coupling can be mediated by electric currents carried by unidentified electron transporters across centimetre-wide zones. Here we present evidence that the native conductors are long, filamentous bacteria. They abounded in sediment zones with electric currents and along their length they contained strings with distinct properties in accordance with a function as electron transporters. Living, electrical cables add a new dimension to the understanding of interactions in nature and may find use in technology development.



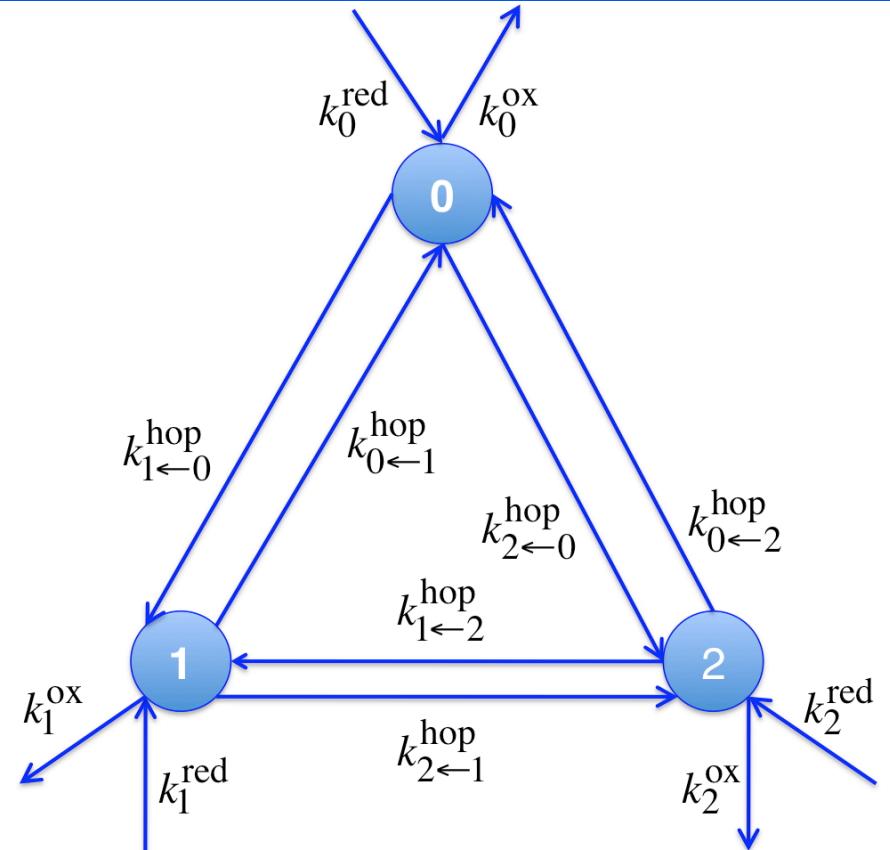
# Model Definition

- A set of  $N$  redox molecules
- Each molecule  $i \in [0, N-1]$  can hold up to one electron:  
 $occ(i) = 1$  (occupied) or 0 (unoccupied)
- **Reduction:** If molecule  $i$  is not occupied by an electron ( $occ(i) = 0$ ), it can be reduced (or an electron is injected to it) at a rate of  $k_i^{\text{red}}$  (1/s)
- **Oxidation:** If  $occ(i) = 1$ , molecule  $i$  can be oxidized (or the electron is ejected from it) at a rate of  $k_i^{\text{ox}}$  (1/s)
- **Hopping:** If  $occ(i) = 1$  and  $occ(j) = 0$ , the electron can hop from molecule  $i$  to molecule  $j$  at a rate of  $k_{j \leftarrow i}^{\text{hop}}$  (1/s)



# Molecular Network Topology

- Directed graph consisting of  $N$  nodes (= molecules) and directed edges (= possible hopping from a node to a neighbor node)
- $ngb(i)$ : Number of outgoing edges for molecule  $i$  ( $\text{Max\_ngb}$  = maximum number of outgoing edges per node)
- $lsngb[N][\text{Max\_ngb}+1]$   
 $lsngb[i][0] = ngb(i)$   
 $lsngb[i][k] = \text{molecular ID of the } k\text{-th outgoing neighbor of molecule } i,$   
 $\text{where } k \in [1, ngb(i)]$



Example:  $\text{Max\_ngb} = 2$

$lsngb[i][k]$	$k = 0$	1	2
$i = 0$	2	1	2
1	2	0	2
2	2	0	1

# Data Structures

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- Occupation

$$occ[N]: occ[i] = \begin{cases} 1 & \text{molecule } i \text{ is occupied by an electron} \\ 0 & \text{unoccupied} \end{cases}$$

- Rates (constants)

$$\begin{aligned} rate[N][Max\_ngb + 2] \\ \left\{ \begin{array}{l} rate[i][0] = k_i^{\text{red}} \\ rate[i][1] = k_i^{\text{ox}} \\ rate[i][k + 1] = k_{lsngb[i][k] \leftarrow i}^{\text{hop}} \end{array} \right. \end{aligned}$$

- Occupation-modified rates (dynamic variables)

$$\begin{aligned} rate\_occ[N][Max\_ngb + 2] \\ \left\{ \begin{array}{l} rate\_occ[i][0] = k_i^{\text{red}} \times (1 - occ[i]) \\ rate\_occ[i][1] = k_i^{\text{ox}} \times occ[i] \\ rate\_occ[i][k + 1] = k_{lsngb[i][k] \leftarrow i}^{\text{hop}} \times (1 - occ[lsngb[i][k]]) \times occ[i] \end{array} \right. \end{aligned}$$

# Algorithm

```

 $occ[i] \leftarrow 0 \text{ } (i = 0 \text{ to } N-1)$ 
 $n_{\text{red}} \leftarrow 0 \text{ } // \text{ number of injected electrons}$ 
 $n_{\text{ox}} \leftarrow 0 \text{ } // \text{ number of ejected electrons}$ 
 $t \leftarrow 0$ 
for  $step \leftarrow 1$  to  $Max\_step$   $// Max\_step = \text{total number of KMC steps}$ 
     $r \leftarrow 0$ 
    for  $i \leftarrow 0$  to  $N-1$ 
         $r += (rate\_occ[i][0] = rate[i][0]*(1-occ[i]))$ 
         $r += (rate\_occ[i][1] = rate[i][1]*occ[i])$ 
        for  $k \leftarrow 1$  to  $lsngb[i][0]$ 
             $r += (rate\_occ[i][k+1] = rate[i][k+1]*(1-occ[lsngb[i][k]])*occ[i])$ 
         $t = -\ln(\text{rand()}/RAND\_MAX)/r$ 
         $r_{\text{th}} \leftarrow r*\text{rand()}/RAND\_MAX$ 
         $r_{\text{acc}} \leftarrow 0$ 
        for  $i \leftarrow 0$  to  $N-1$ 
            if ( $r_{\text{th}} < (r_{\text{acc}} += rate\_occ[i][0])$ )  $// \text{reduction occurs}$ 
                 $occ[i] \leftarrow 1$ ; print  $t, ++n_{\text{red}}, n_{\text{ox}}$ ; break
            else if ( $r_{\text{th}} < (r_{\text{acc}} += rate\_occ[i][1])$ )  $// \text{oxidation occurs}$ 
                 $occ[i] \leftarrow 0$ ; print  $t, n_{\text{red}}, ++n_{\text{ox}}$ ; break
            else
                for  $k \leftarrow 1$  to  $lsngb[i][0]$ 
                    if ( $r_{\text{th}} < (r_{\text{acc}} += rate\_occ[i][k+1])$ )  $// \text{hopping from } i \text{ to } k\text{-th neighbor occurs}$ 
                         $occ[lsngb[i][k]] \leftarrow 1$ ;  $occ[i] \leftarrow 0$ ; break
            if ( $r_{\text{th}} < r_{\text{acc}}$ ) break

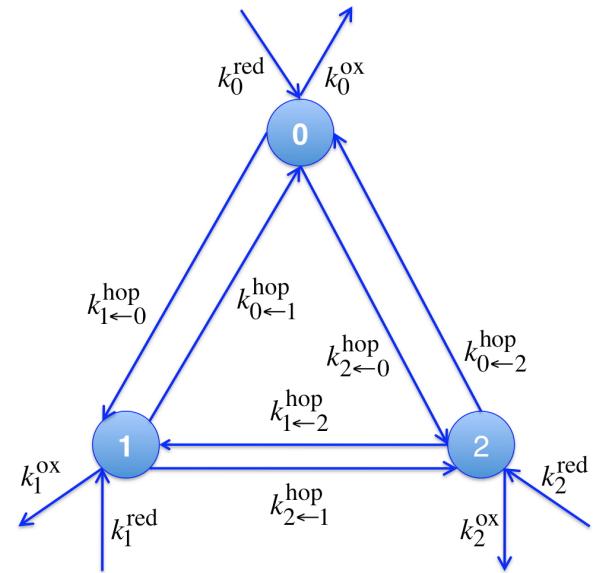
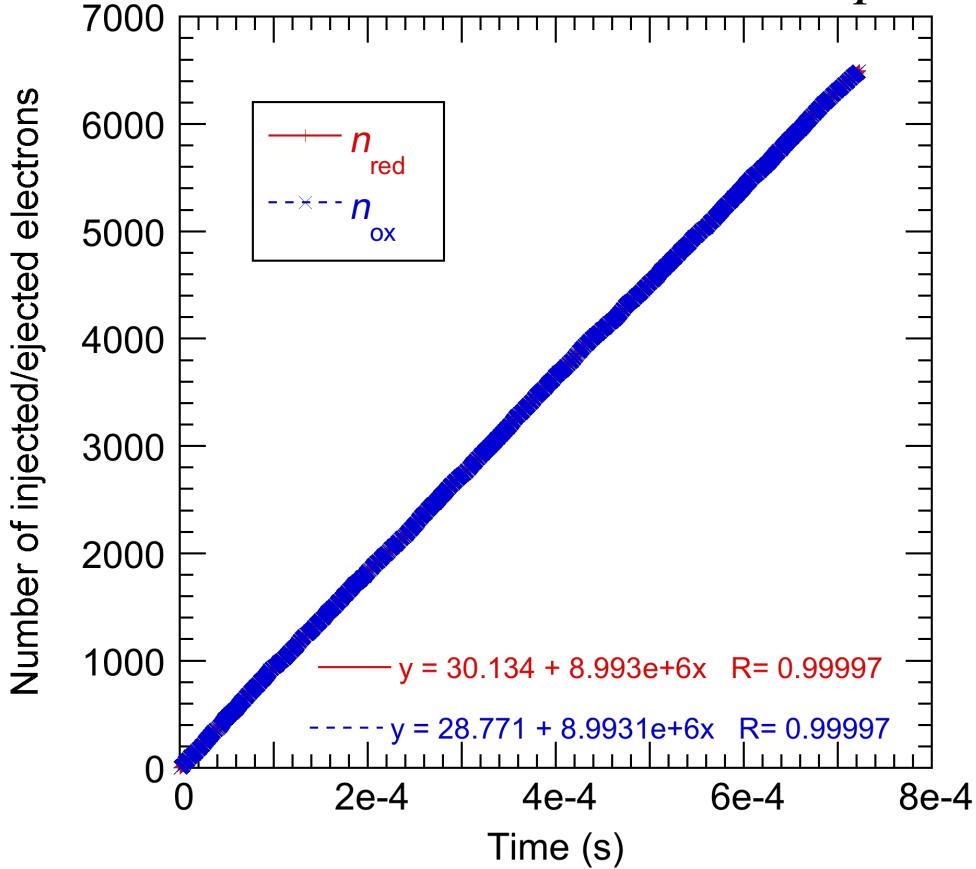
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$r = \sum_{\text{event}} r_{\text{event}}$   
 $\Delta t = -\ln(u_{\text{random}})/r$

# Example Run

$rate[i][k]$ (1/s)	$k = 0$	1	2	3
$i = 0$	1e7	0	1e9	2e9
1	0	0	3e9	1e7
2	0	2e7	2e7	4e9

$Max\_step = 10^6$

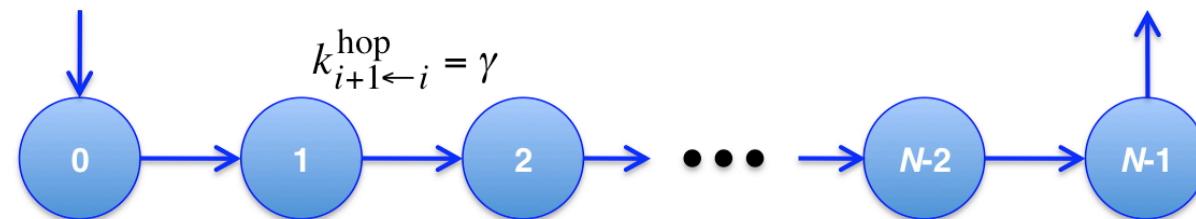


## Electric current

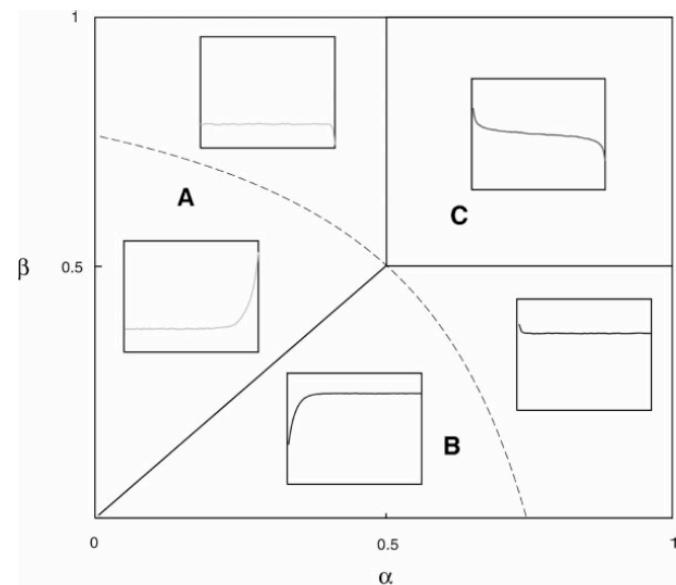
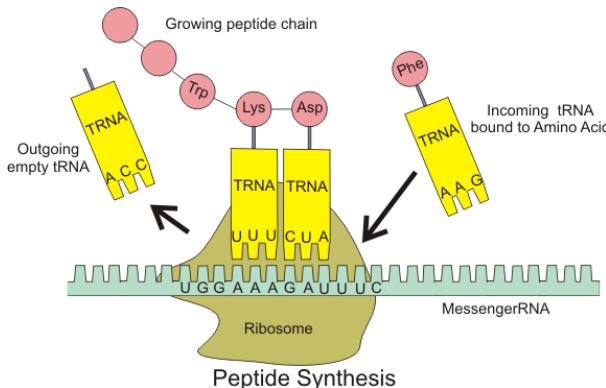
$$8.99 \times 10^6 \text{ (1/s)} \times 1.602 \times 10^{-19} \text{ (C)} \\ = 1.44 \times 10^{-12} \text{ (C/s)} \\ = 1.44 \text{ (pA)}$$

# Asymmetric Simple Exclusion Process

- Linear chain of  $N$  molecules
- *Reduction* can occur only at molecule 0 with  $k^{\text{red}} = \alpha$  (1/s)
- *Oxidation* can occur only at molecule  $N-1$  with  $k^{\text{ox}} = \beta$  (1/s)
- *Hopping* can occur from molecule  $i$  to  $i+1$  ( $i \in [0, N-2]$ ) with  $k_{i+1 \leftarrow i}^{\text{hop}} = \gamma$  (1/s)



- Nonequilibrium phase transition from low-density (LD) to high-density (HD) phase with increasing  $\alpha/\beta$
- Applications in traffic flow and ribosome motion on mRNA during translation

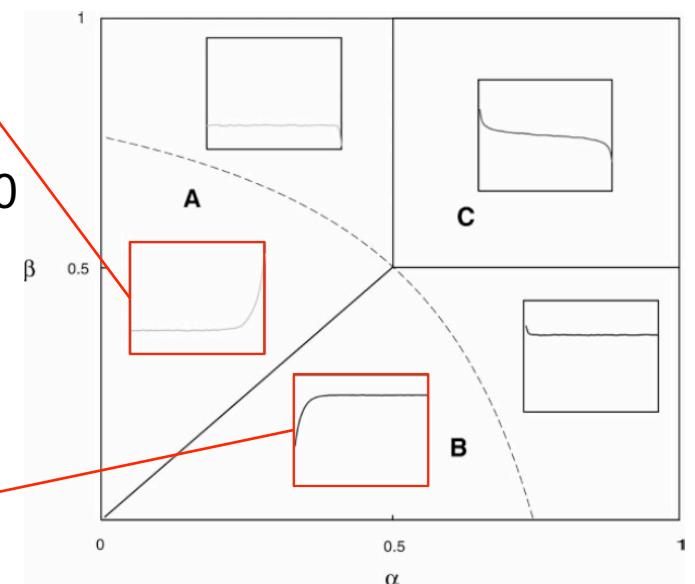
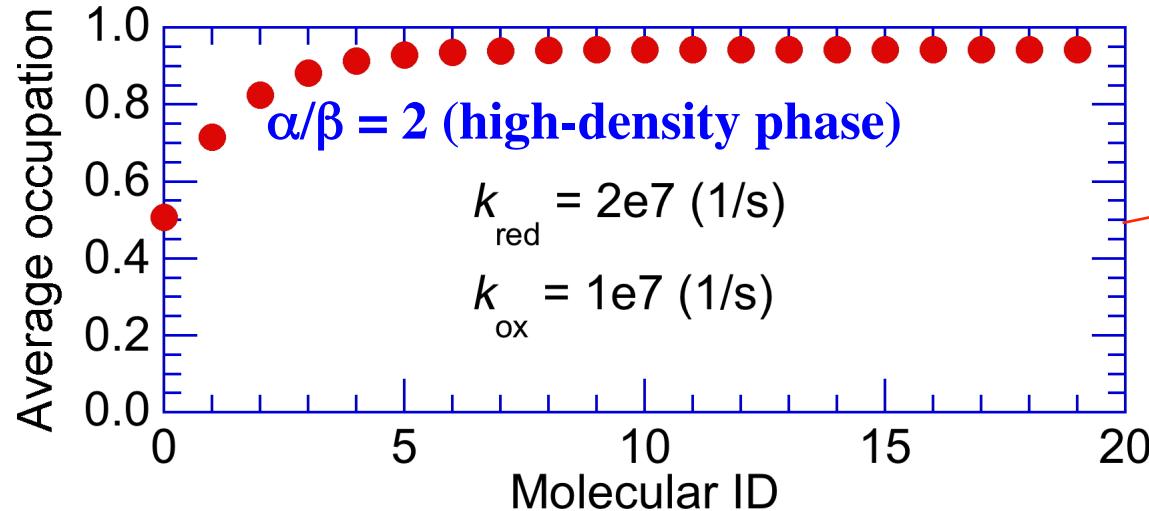
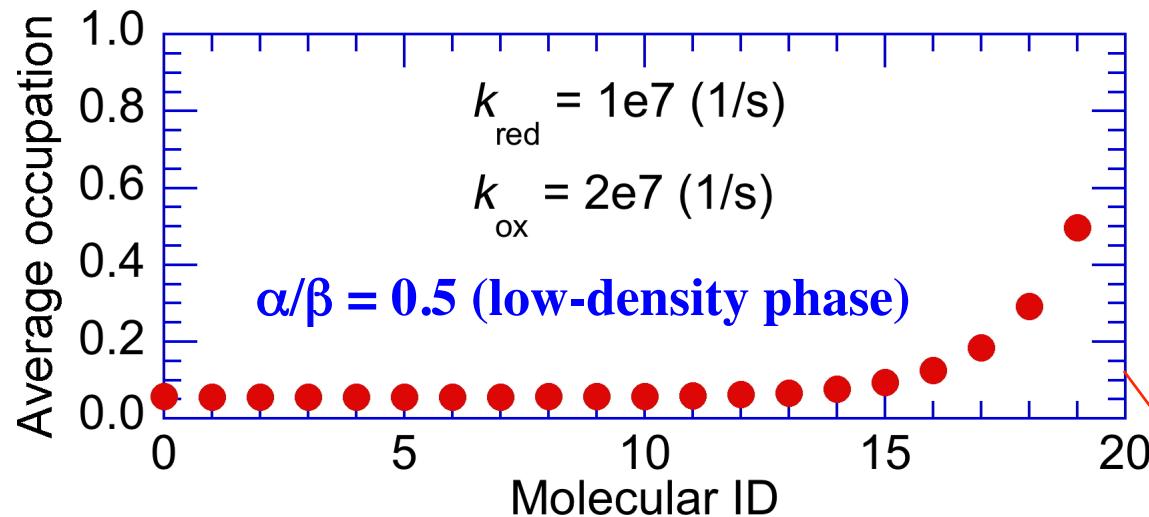


A. Shadschneider, *Physica A* **285**, 101 ('00)

# ASEP Occupation Profile

## KMC simulation results

$N = 20$ ;  $k_{\text{hop}} = 1\text{e}9$  (1/s);  $\text{Max\_step} = 10^6$



# Algorithm for KMC Simulation of ASEP

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 $k^{\text{red}} = \alpha (k^{\text{red}}, k^{\text{ox}}) = (1\text{e}7 \text{ s}^{-1}, 2\text{e}7 \text{ s}^{-1}) \text{ and } (2\text{e}7 \text{ s}^{-1}, 1\text{e}7 \text{ s}^{-1}) \quad k^{\text{ox}} = \beta$ 
 $n_{\text{red}} \leftarrow 0 \quad // \text{number of injected electrons}$ 
 $n_{\text{ox}} \leftarrow 0 \quad // \text{number of ejected electrons}$ 
 $t \leftarrow 0 \quad \text{Max\_step} = 10^6$ 

```

**for**  $step \leftarrow 1$  to  $\text{Max\_step}$   $// \text{Max\_step} = \text{total number of KMC steps}$

$r \leftarrow 0$

**for**  $i \leftarrow 0$  to  $N-1$

$r += (\text{rate\_occ}[i][0] = \text{rate}[i][0] * (1 - \text{occ}[i]))$

$r += (\text{rate\_occ}[i][1] = \text{rate}[i][1] * \text{occ}[i])$

**for**  $k \leftarrow 1$  to  $\text{lsngb}[i][0]$

$r += (\text{rate\_occ}[i][k+1] = \text{rate}[i][k+1] * (1 - \text{occ}[\text{lsngb}[i][k]]) * \text{occ}[i])$

$t = \ln(\text{rand}()) / \text{RAND\_MAX} / r$

$r_{\text{th}} \leftarrow r * \text{rand}() / \text{RAND\_MAX}$

$r_{\text{acc}} \leftarrow 0$

**for**  $i \leftarrow 0$  to  $N-1$

**if** ( $r_{\text{th}} < (r_{\text{acc}} += \text{rate\_occ}[i][0])$ )  $// \text{reduction occurs}$

$\text{occ}[i] \leftarrow 1$ ; **print**  $t, ++n_{\text{red}}, n_{\text{ox}}$ ; **break**

**else if** ( $r_{\text{th}} < (r_{\text{acc}} += \text{rate\_occ}[i][1])$ )  $// \text{oxidation occurs}$

$\text{occ}[i] \leftarrow 0$ ; **print**  $t, n_{\text{red}}, ++n_{\text{ox}}$ ; **break**

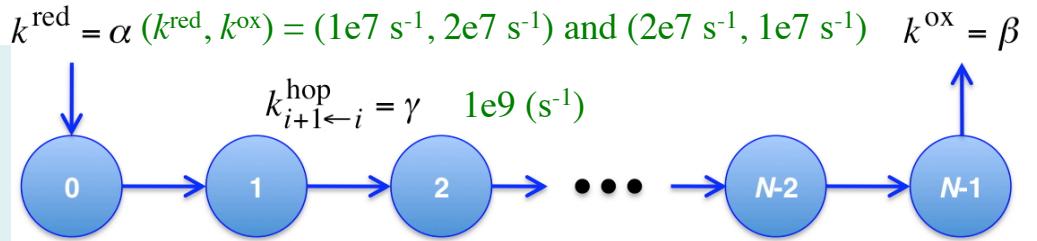
**else**

**for**  $k \leftarrow 1$  to  $\text{lsngb}[i][0]$

**if** ( $r_{\text{th}} < (r_{\text{acc}} += \text{rate\_occ}[i][k+1])$ )  $// \text{hopping from } i \text{ to } k\text{-th neighbor occurs}$

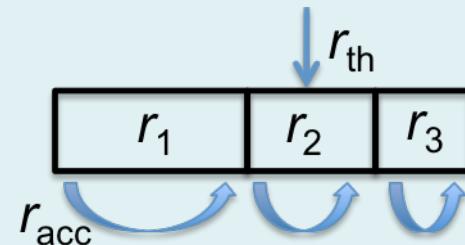
$\text{occ}[\text{lsngb}[i][k]] \leftarrow 1$ ;  $\text{occ}[i] \leftarrow 0$ ; **break**

**if** ( $r_{\text{th}} < r_{\text{acc}}$ ) **break**



$$r = \sum_{\text{event}} r_{\text{event}}$$

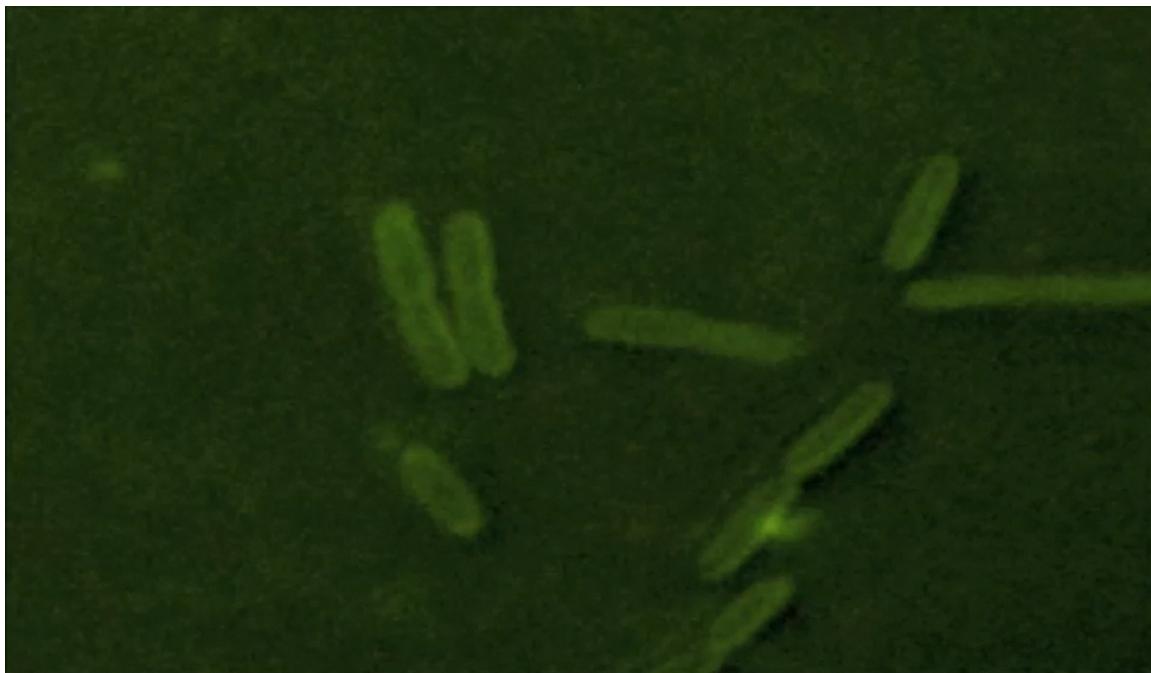
$$\Delta t = -\ln(u_{\text{random}}) / r$$



# Motivation: Bacterial Nanowires

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- Electron transfer (ET) governs all known energy-conversion (redox) processes in biology
- A remarkable example is the recent discovery of rapid ET along bacterial nanowires produced by *Shewanella oneidensis* MR-1



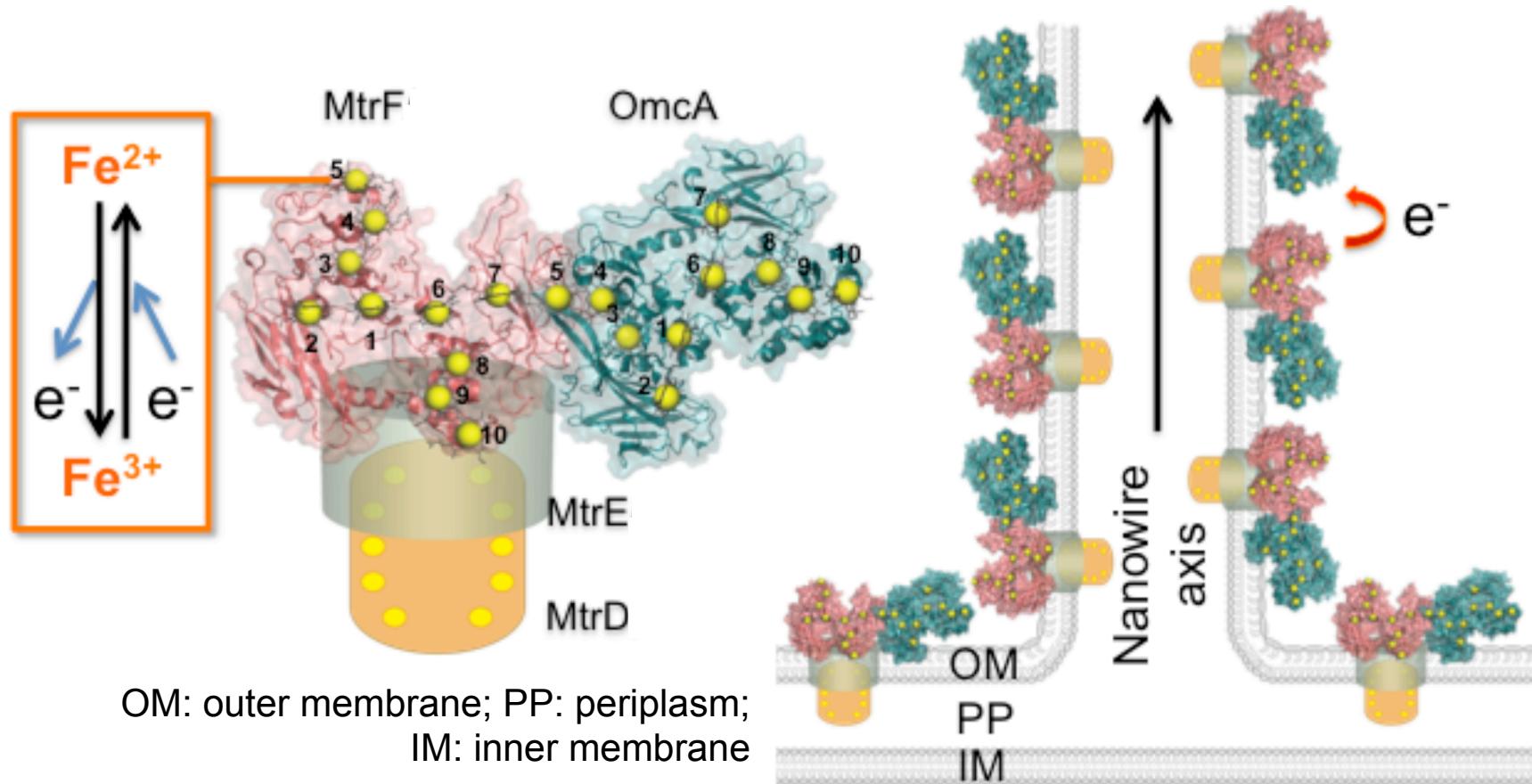
*Shewanella* emitting electrically conducting nanowires



S. Pirbadian *et al.*, PNAS 111, 12883 ('14)

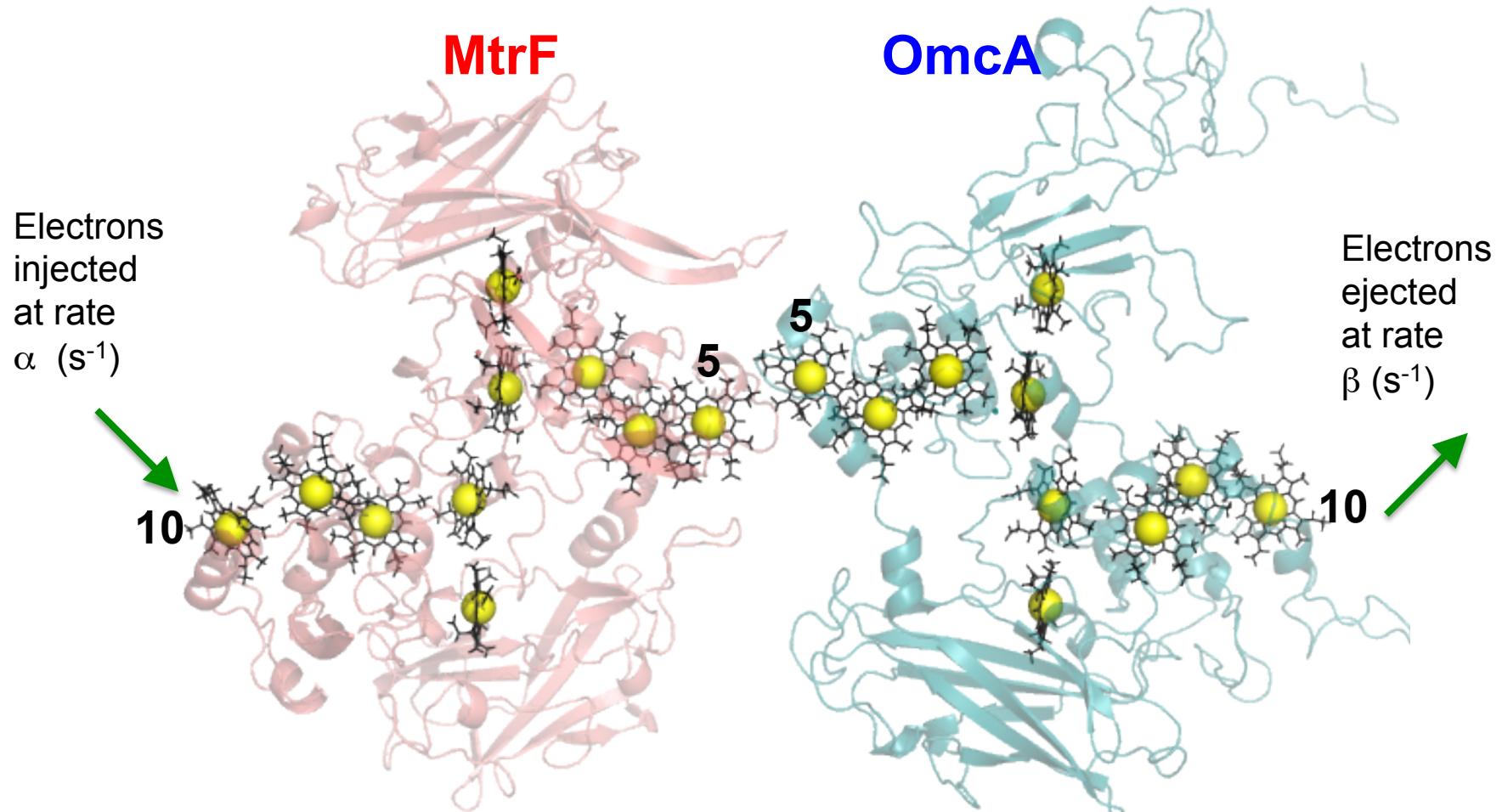
# Scientific Problem & Research Goal

- Outer-membrane cytochromes, MtrF & OmcA, are hypothesized media for ET, but how they are assembled into a conducting complex remains a mystery
- **Goal:** Determine the structure of MtrF-OmcA complex & visualize ET dynamics in it to understand electric conduction mechanisms



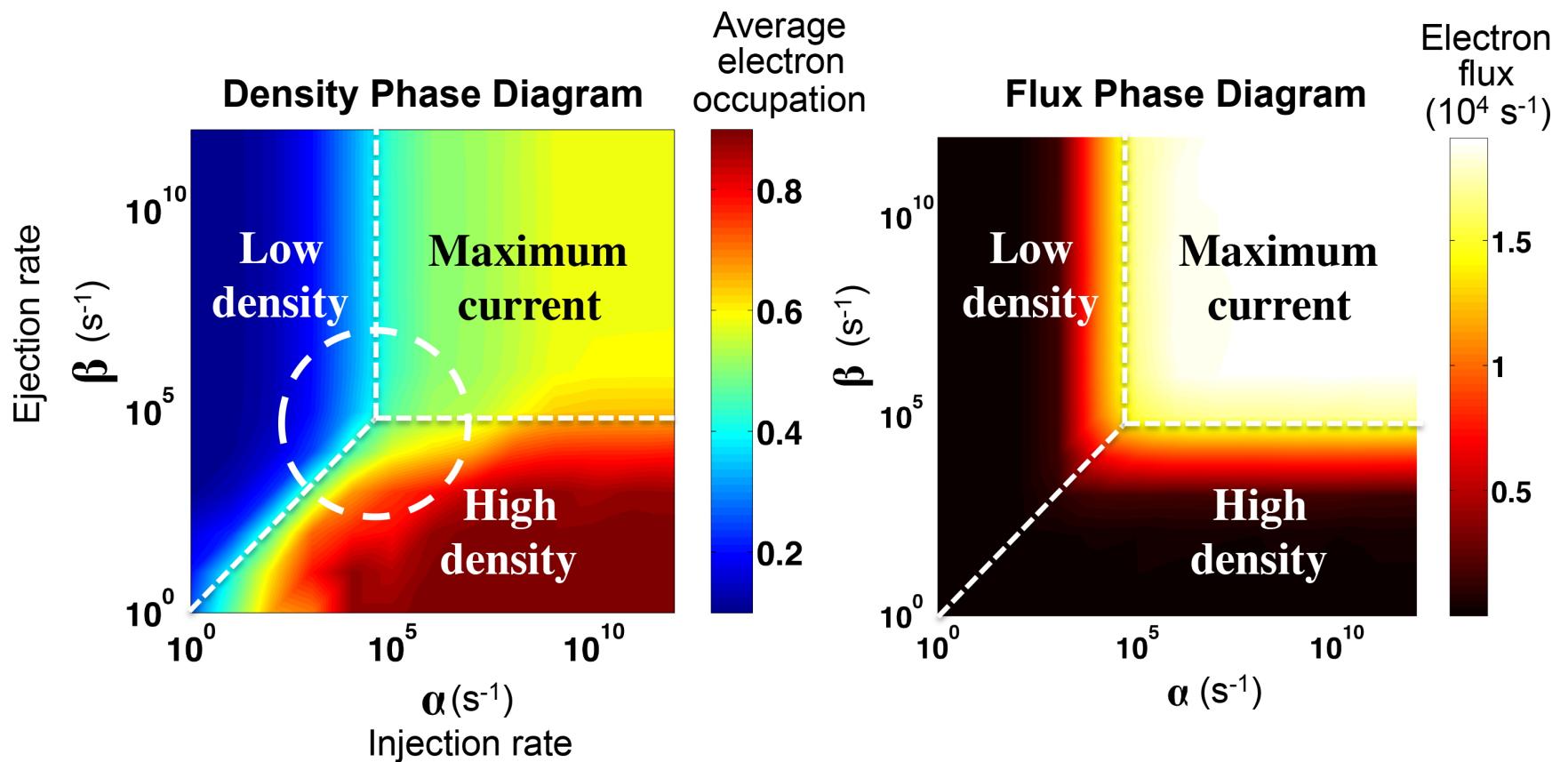
# Result: Complex Structure

## Top-ranked MtrF-OmcA complex structure



- H. S. Byun *et al.*, *ChemElectroChem* **1**, 1932 ('14)  
C. M. Nakano *et al.*, *Comput. Phys. Commun.* **193**, 1 ('15)  
C. M. Nakano *et al.*, *J. Mol. Graph. Model.* **65**, 94 ('16)  
T. Wei *et al.*, *J. Phys. Chem. Lett.* **7**, 929 ('16)

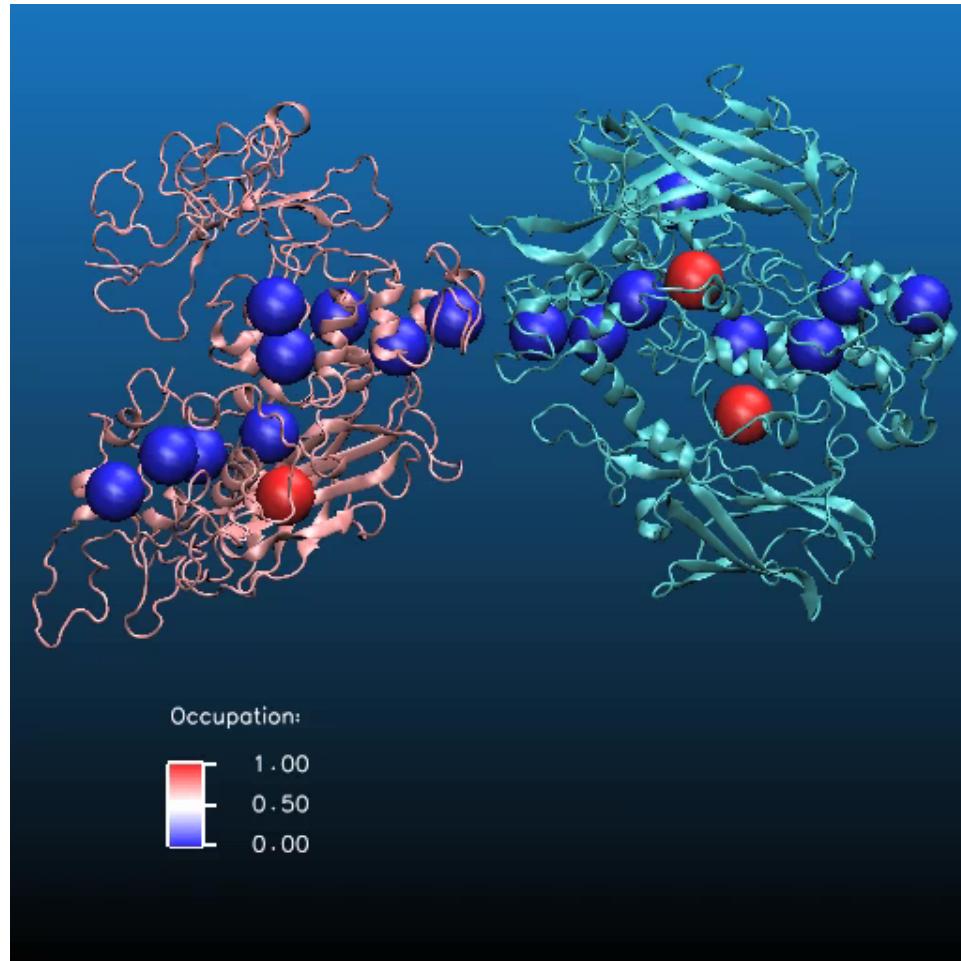
# Result: ET Phase Diagram



- KMC results exhibit a nonequilibrium phase transition from low-density to high-density phases for increased ratio of electron injection rate ( $\alpha$ ) to ejection rate ( $\beta$ )
- When both  $\alpha$  &  $\beta$  exceed the smallest  $k_{ij}$ , another transition to the maximum-current phase was found
- Observed respiration rates ( $10^3$ - $10^4$  s<sup>-1</sup>) indicate that “life operates around the triple phase junction,” thus a small change in the electrochemical environment effectively triggers a large electric response

# Animation: Low-Density Phase

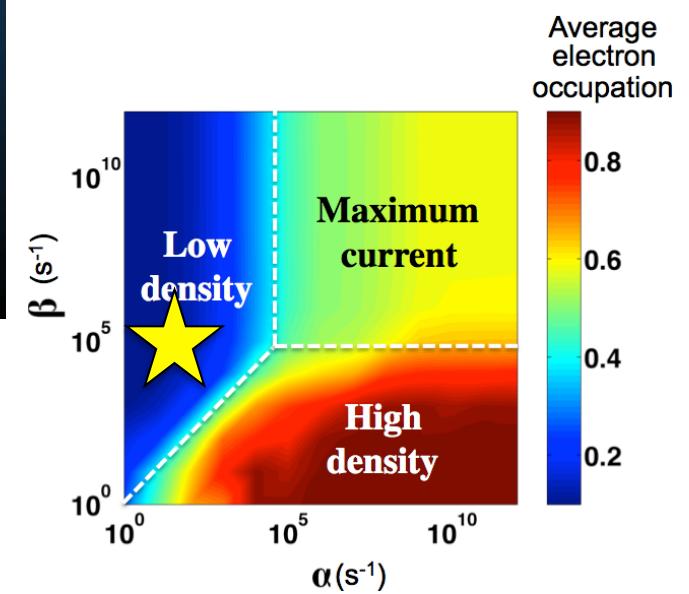
Electron injection rate:  
 $\alpha = 10^2 \text{ s}^{-1}$



Electron ejection rate:  
 $\beta = 10^5 \text{ s}^{-1}$

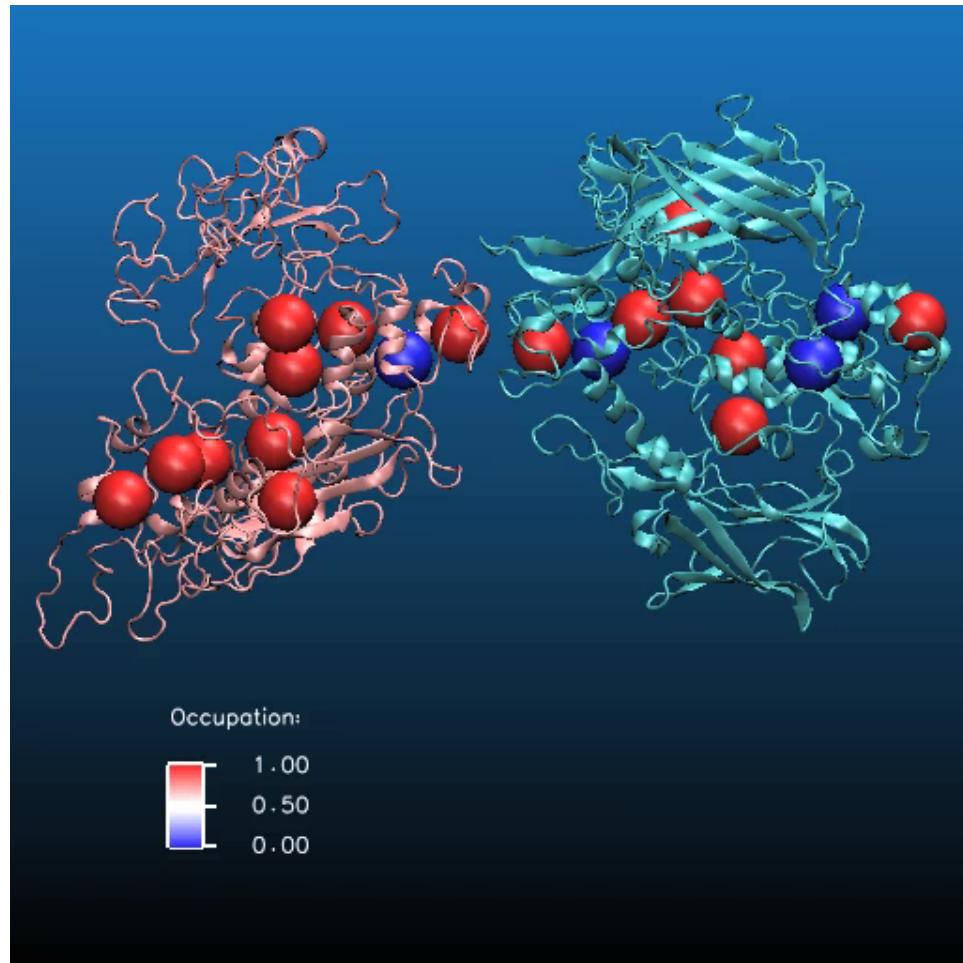


- Electron flow is limited by the small number of electrons



# Animation: High-Density Phase

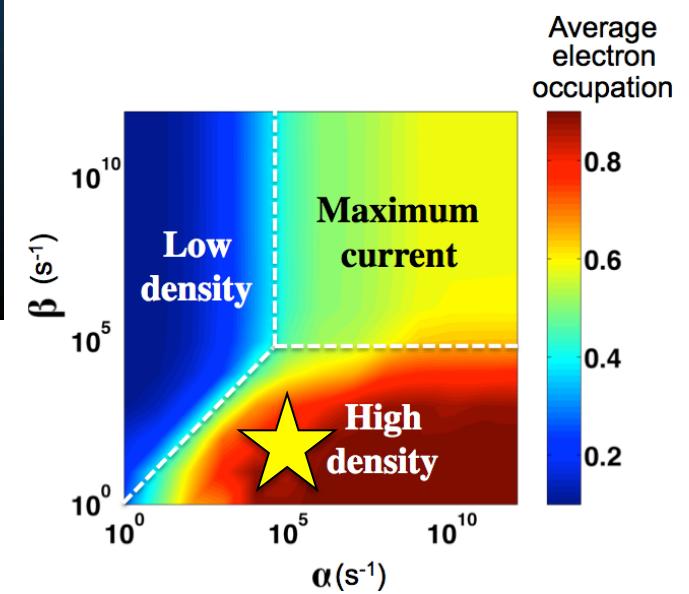
Electron injection rate:  
 $\alpha = 10^5 \text{ s}^{-1}$



Electron ejection rate:  
 $\beta = 10^2 \text{ s}^{-1}$

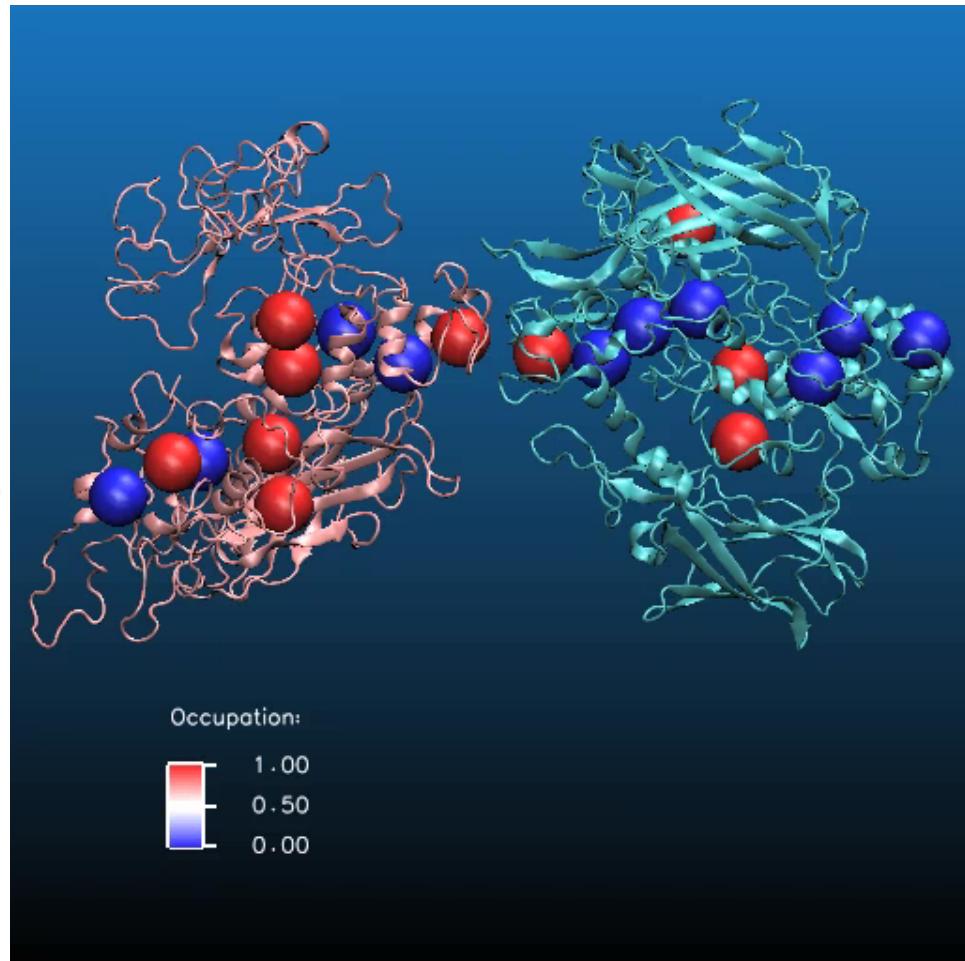


- Electron flow is limited by the congestion of electrons



# Animation: Maximum-Current Phase

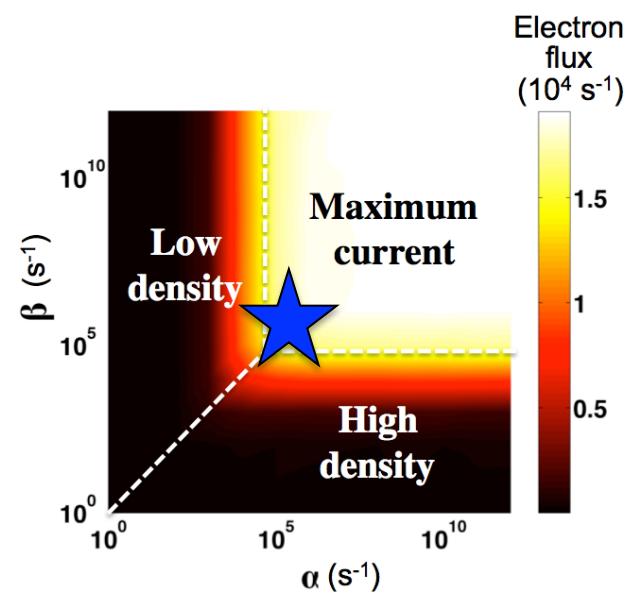
Electron injection rate:  
 $\alpha = 10^5 \text{ s}^{-1}$



Electron ejection rate:  
 $\beta = 10^5 \text{ s}^{-1}$



- Large electric current is facilitated by balanced electron injections & ejections
- Life operates around the triple junction



# Animation: Maximum-Current Phase

