

F61: NMR–Spectroscopy

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Outline

Introduction and Theoretical Concepts

Part I: Relaxation Times

Part II. Chemical shift

- Theory

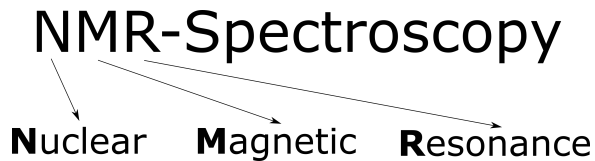
- Measurements

Imaging with NMR

- Theory

- Experiments

Introduction



Introduction

NMR-Spectroscopy

Nuclear

Magnetic

Resonance

Detection of substances

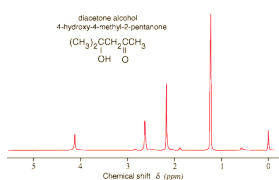


Figure: Carl Nave, Hyperphysics, hyperphysics.phy-astr.gsu.edu

Multidimensional imaging

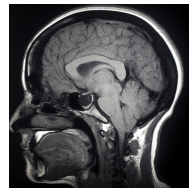


Figure: Sierra Vista Diagnostics, svdrads.com

Theoretical Concepts - Working Principle

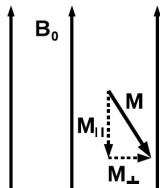


Figure: Magnetization

- ▶ Nuclei with spin I have a magnetic moment μ
- ▶ Ensemble of many nuclei: Measurable magnetization \vec{M}
- ▶ Minimal energy \rightarrow Dipole aligned parallel to B-field
- ▶ Ground state $\rightarrow M_{\perp} = 0$

Theoretical Concepts - Working Principle

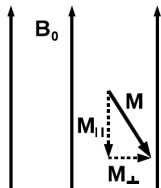


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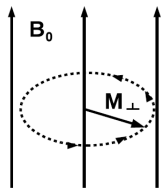


Figure: Larmor-Precession of M_{\perp}

- ▶ Excited states have a component $M_{\perp} \neq 0$
- ▶ M_{\perp} precesses around the field lines with the Larmor frequency

$$\omega_L = \gamma B_0 \quad (1)$$

- ▶ ω_L can be measured!

Theoretical Concepts - Working Principle

How can we create an excited state ?

- ▶ An oscillating B-Field \vec{B}_1 rotates the magnetization \vec{M} by an angle

$$\alpha = \gamma B_1 \Delta t \quad (2)$$

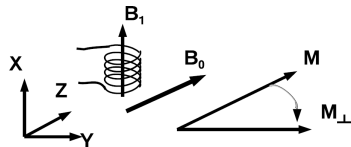


Figure: Rotation of M due to an HF-Pulse

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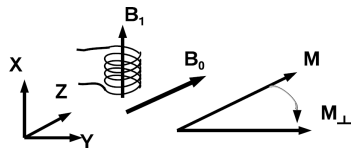


Figure: Rotation of M due to an HF-Pulse

- ▶ By choosing Δt , we can create:
 - ▶ A perpendicular magnetization (90°-Pulse)
 - ▶ An anti-parallel magnetization (180°-Pulse)

Setup and Measurement Principle

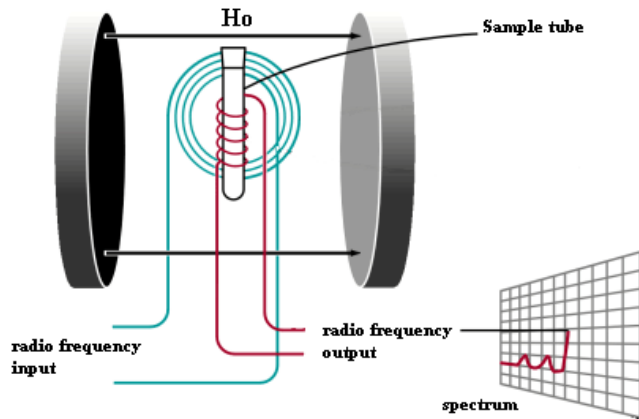


Figure: McGraw Hill Higher Education, mhhe.com

Theory of Relaxation

Excited states decay into the Ground State on a characteristic timescale.
The decay is of exponential nature and described in the *Bloch equations*:

$$\frac{dM_{\perp}(t)}{dt} = -\frac{M_{\perp}(t)}{T_2} \quad (3)$$

$$\frac{dM_{\parallel}(t)}{dt} = -\frac{M_{\parallel}(t) - M_0}{T_1} \quad (4)$$

- ▶ T_2 : Spin-Lattice Relaxation
- ▶ T_1 : Spin-Spin Relaxation

T_2 -Measurement: Spin Echo

Spin-Echo principle

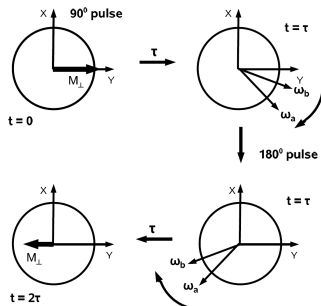


Figure: Principle of the spin-echo method

T_2 -Measurement: Spin Echo

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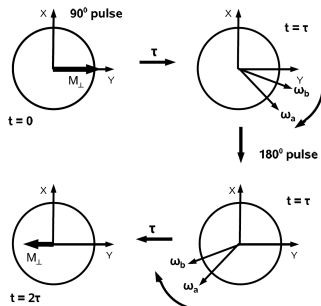


Figure: Principle of the spin-echo method

Pulse sequence

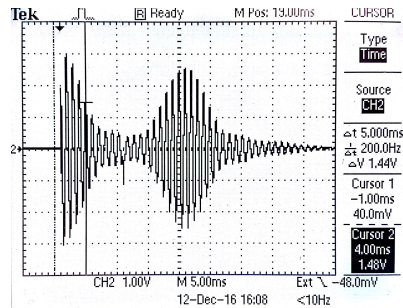


Figure: Spin-Echo measurement with $\tau = 10\text{ms}$

T_2 -Measurement: Spin Echo

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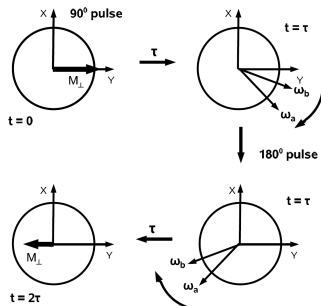


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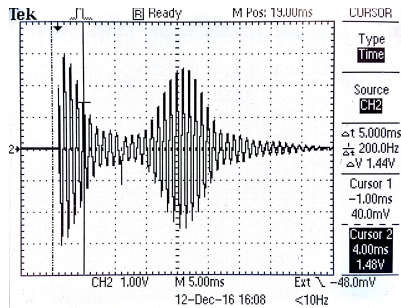


Figure: Spin-Echo measurement with $\tau = 10\text{ms}$

- **Disadvantage:** Dephasing for long measurement times!

T_2 -Measurement: Spin Echo

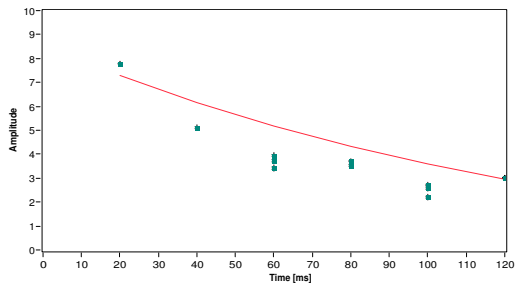


Figure: T2-Measurement Sample 1 with fit.

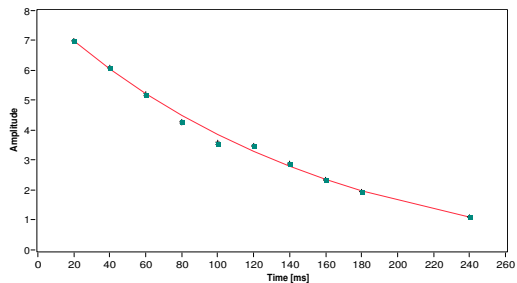


Figure: T2-Measurement Sample 3 with fit.

T_2 -Measurement: Carr-Purcell Sequence

Improve dephasing problem of spin-echo method:

- ▶ Inject a 180° -Pulse on odd multiples of a time τ .
- ▶ The system is phase coherent on even multiples of a time τ .

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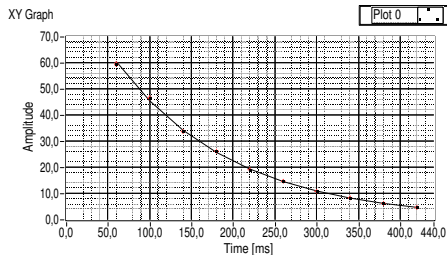


Figure: T2-Measurement using Carr-Purcell, Sample 1, with fit.

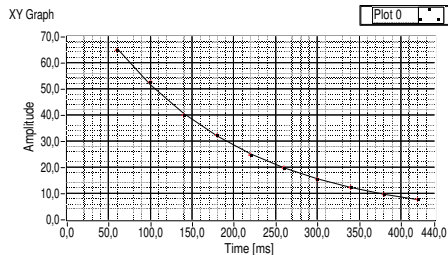


Figure: T2-Measurement using Carr-Purcell, Sample 3, with fit.

T_1 -Measurement

Start with a 180° -Pulse (Anti-parallel Magnetization) and probe the magnetization after time τ with a 90° -Pulse

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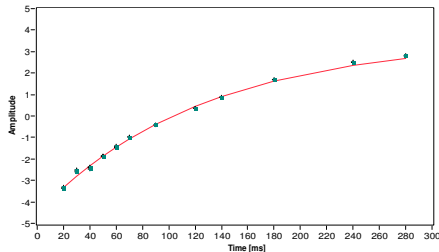


Figure: T1-Measurement Sample 1 with fit.

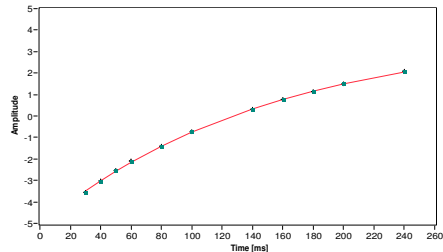


Figure: T1-Measurement Sample 3 with fit.

Relaxation Times: Evaluation

Table: Relaxation times – Measured values

Time Method	T_1 [ms] 180°-90°	T_2 [ms] Spin-Echo	T_2 [ms] Carr-Purcell
Sample 1 (Gd 1:500)	(125, 5 ± 0, 6)	(119, 5 ± 0, 5)	(140, 1 ± 0, 4)
Sample 3 (Gd 1:600)	(150, 5 ± 1, 2)	(139, 3 ± 0, 8)	(166, 9 ± 0, 4)

Chemical shift – theory

Aim: structure determination of chemical substances

Chemical shift – theory

Aim: structure determination of chemical substances

- ▶ electron orbitals contribute to B_0 :

$$\delta \vec{B} = \sigma \vec{B}_0$$

- ▶ modification of the Larmor frequency:

$$\omega_i = \omega_L (1 - \sigma_i)$$

Chemical shift – theory

Aim: structure determination of chemical substances

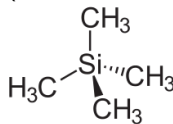
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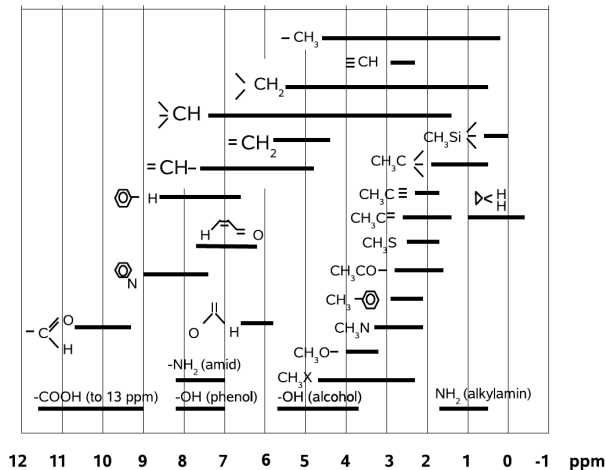
- ▶ reference substance: TMS
(Tetra-Methyl-Silan)



- ▶ relative chemical shift in ppm:

$$\delta_i = \sigma_i - \sigma_{TMS} = \frac{\omega_{TMS} - \omega_i}{\omega_L}$$

Chemical shifts δ_i of compounds relative to TMS



Measurements

- ▶ five chemical substances, with and without TMS
- ▶ inhomogeneities and diffusion processes reduce resolution
⇒ thin glass tube, put into rotation with pressure air
- ▶ result:
 - ▶ without rotation:
FWHM = 200 Hz, $I = 0,25$
 - ▶ with rotation:
FWHM = 20 Hz, $I = 1,9$
- ▶ energy resolution:
$$\Delta E_{NMR} = h \cdot \Delta \nu = 8,28 \cdot 10^{-14} \text{ eV}$$

Identification of the Probes

Probe C: acetic acid $\text{CH}_3 - \text{COOH}$

Peaks of C+ [ppm]	Peaks of C [ppm]	Chem. shift.	
$p_1 = 16,7$	$p_1 = 17,0$	$\delta_i = 11,6$	COOH-group
$p_2 = 26,2$	$p_2 = 26,6$	$\delta_i = 2,1$	Methyl group CH_3
$p_3 = 28,3$	–	–	TMS

Identification of the Probes



Peaks of B+ [ppm]	Peaks of B [ppm]	Chem. shift	
$p_1 = 22,7$	$p_1 = 22,7$	$\delta_i = 7,0$	Benzene group
$p_2 = 27,5$	$p_2 = 27,5$	$\delta_i = 2,2$	Methyl group, Peak twice as high as p_1
$p_3 = 29,7$	—	—	TMS

Identification of the Probes

Probe E: toluol CH3-c1ccccc1

Peaks of E+ [ppm]	Peaks of E [ppm]	Chem. shift	
$p_1 = 19,5$	$p_1 = 23,1$	$\delta_i = 7,3$	Benzene group
$p_2 = 24,4$	$p_2 = 23,1$	$\delta_i = 2,4$	Methyl group, peaks have same height
$p_3 = 26,8$	—	—	TMS

Identification of the Probes

Probe A: fluoroacetone $\text{FCH}_2 - \text{CO} - \text{CH}_3$

Peaks of A+ [ppm]	Peaks of A [ppm]	Chem. shift	
$p_1 = 22,2$	$p_1 = 23,8$	$\delta_i = 6,3$	FCH ₂ -group
$p_2 = 24,6$	$p_2 = 21,4$	$\delta_i = 3,9$	
$p_3 = 26,4$	$p_3 = 19,6$	$\delta_i = 2,1$	Methyl group CH ₃
$p_4 = 28,5$	—	—	TMS

Identification of the Probes

Probe D: fluoroacetonitril $\text{FCH}_2\text{—CN}$

Peaks of D+ [ppm]	Peaks of D [ppm]	Chem. shift	
$p_1 = 30,8$	—	—	TMS
$p_2 = 34,8$	$p_2 = 26,6$	$\delta_i = 6,4$	FCH ₂ group
$p_3 = 37,2$	$p_3 = 24,2$	$\text{delta}_i = 4,0$	

one dimensional imaging – theory

- ▶ **position dependent magnet fields**
- ▶ Superposition of the static field \vec{B}_0 with gradient fields \vec{B}^x , \vec{B}^y , \vec{B}^z
- ▶ two techniques:

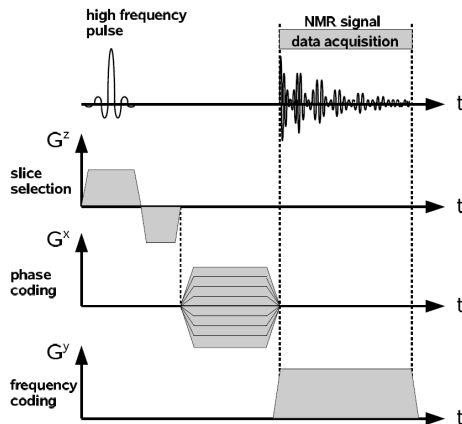
frequency coding

- ▶ Larmor frequency
$$\omega_L = \gamma(B_0 + G^z \cdot z) = \omega_L^0 + \omega_z$$
- ▶ measured NMR signal $S(t)$ is Fourier transform of $M_{\perp}^{rot}(z)$

phase coding

- ▶ apply gradient field, increase strength
- ▶ phase rotates: $\phi(z) = (\gamma G^z T_{Ph})z = k_z z$

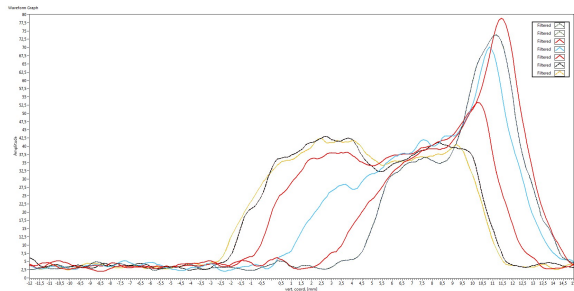
two dimensional imaging – theory



One dimensional imaging measurements

- ▶ Bruker[®] NMR analyzer mq7.5
- ▶ Glass tube filled with 15 mm of oil
- ▶ Glass tube filled with 50 mm of water
- ▶ glass tube with teflon structure
- ▶ examination of an infiltration process:

Fick's second law: $\frac{\partial c}{\partial t} = D \frac{\partial^2 c}{\partial x^2}$



two dimensional imaging measurements

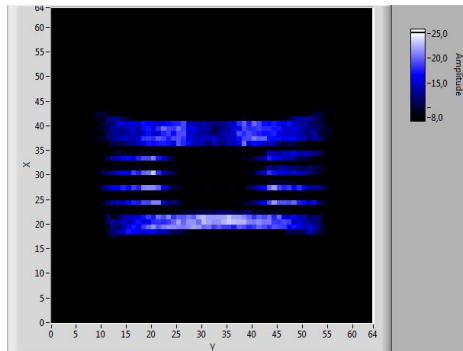


Figure: teflon structure

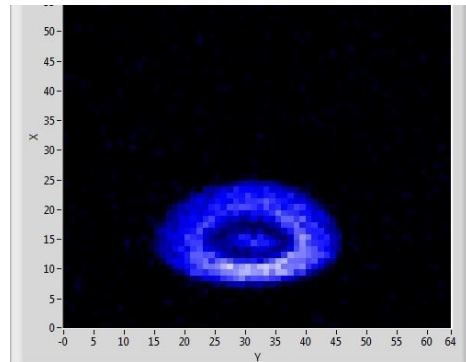


Figure: olive

two dimensional imaging measurements

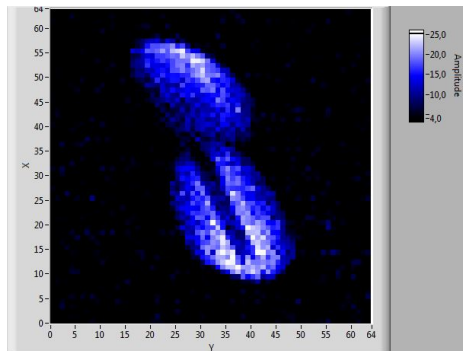


Figure: peanut shell

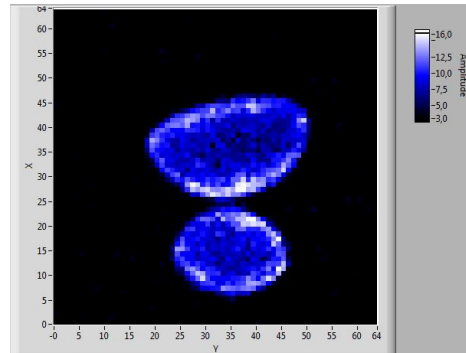


Figure: aloe vera