

# CHEM 153A Week 3

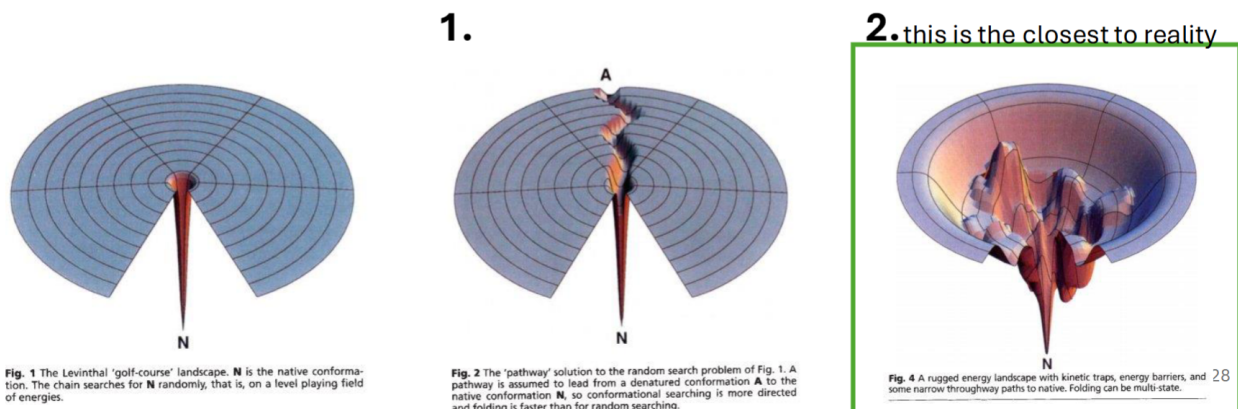
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## Protein Folding Continued

### More Questions - Levinthal's Paradox

- Other experiments like Anfinsen's raised more questions
  - Denatured proteins refold in 0.1-1000 seconds
  - Take a hypothetical protein with 100 amino acids
  - Due to allowed rotations, amino acids can have 3 conformations
  - That's roughly  $3^{100}$  possibilities ( $\approx 5 \times 10^{47}$ )
- If the protein can visit one conformation every picosecond ( $10^{-12}$  s), searching every possibility would take  $5 \times 10^{47} \times 10^{-12}$  seconds, or  $1.6 \times 10^{28}$  years.
- This is the left diagram



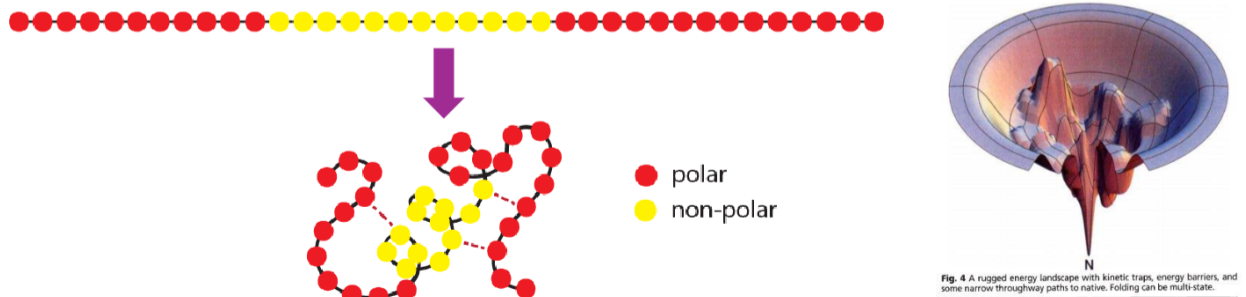
## The Thermodynamics of Protein Folding

- Free Energy Funnel:
  - Unfolded states have high degree of conformational entropy, thus there is high free energy.
  - The free energy funnel shows that the closer the protein is to its **native state**, the ideal, lowest energy, folded form, the lower energy it has.
    - \* The middle diagram suggests that there is a pathway guiding the protein folding to the lowest energy state. This is on the right track, but is not correct.
    - \* The right diagram states there are multiple stable intermediates leading to the final folded protein. This is the most accurate diagram.

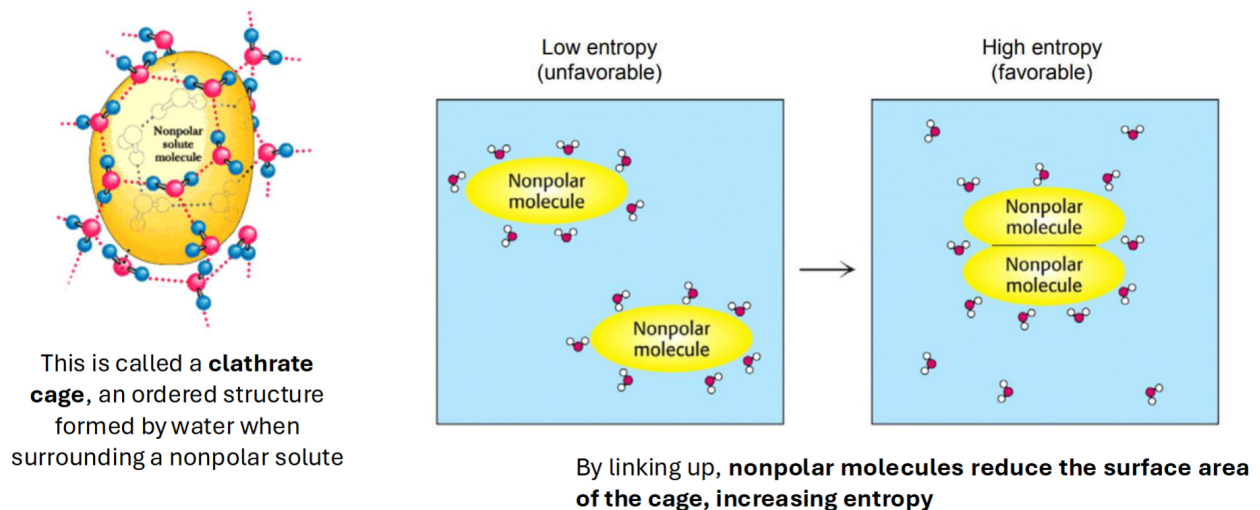
## Hydrophobic Collapse → Molten Globule

**Hydrophobic collapse** is the rapid burying of hydrophobic residues in the center of the protein - they want to escape their watery environment.

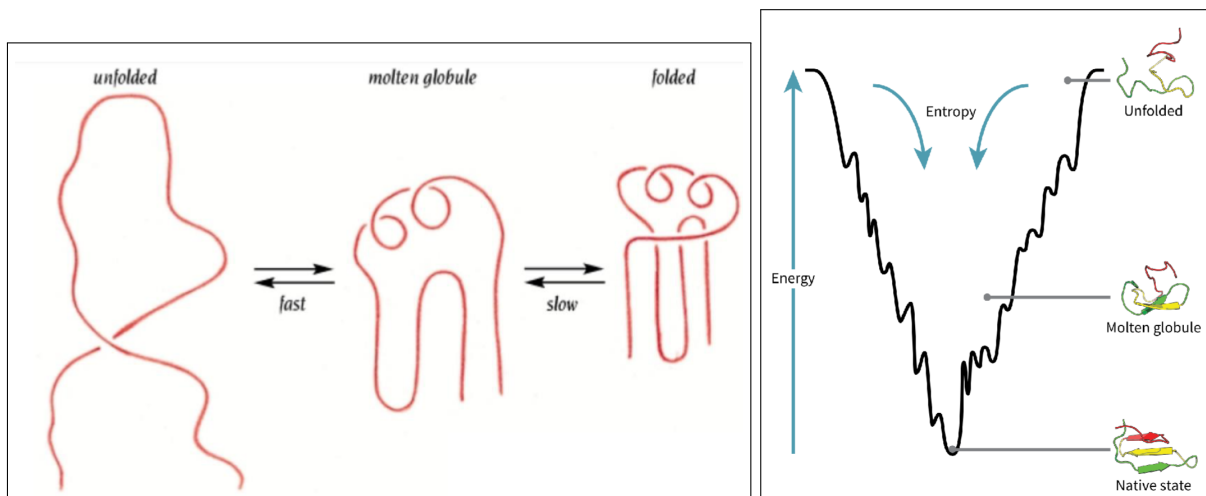
- e.g., Val, Leu, Ile, etc.



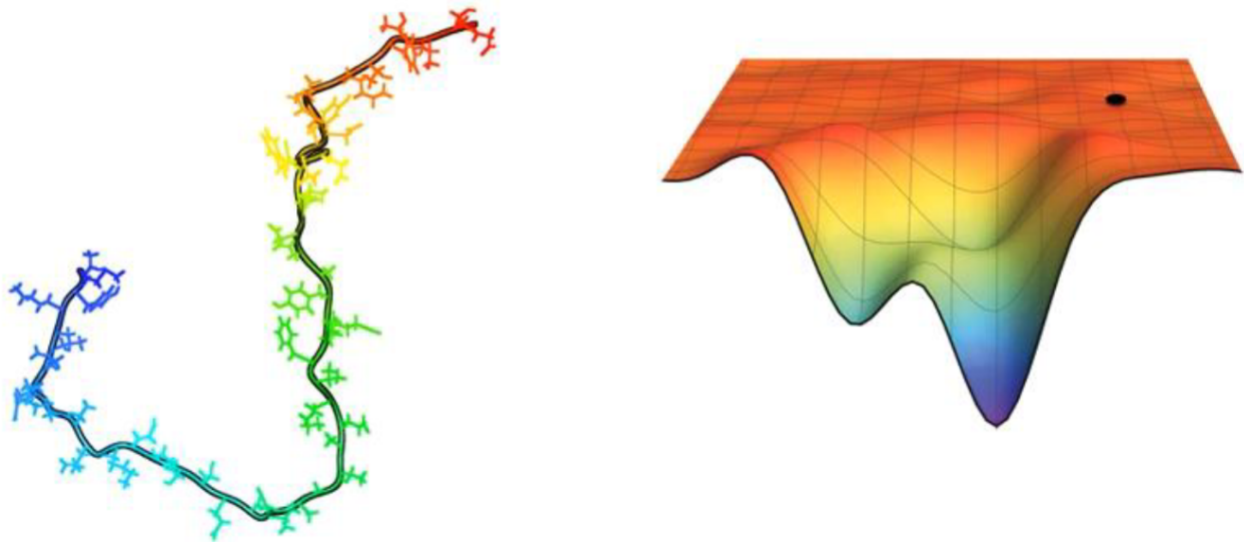
- Hydrophobic collapse is entropy-driven: water molecules become more ordered around hydrophobic residues, and the collapse releases that ordered water, increasing entropy.



- This collapse forms the **molten globule**, an intermediate transitioning to the final form of the protein



## Protein Folding and Energy Landscape



## Other Significant Factors in Protein Folding

- **Chain conformational entropy** is the entropy *decrease* due to the formation of an ordered polypeptide
- **Hydrogen bonding** serves an important role, mediating interactions with the surrounding water as well as connecting the outer surface of the protein with the hydrophobic core
  - They also stabilize interactions between peptide chains (secondary structure)
  - Enthalpy decrease
- **London dispersion forces** hold together the hydrophobic core (enthalpy decrease)

## Protein Structures

- Protein segments can adopt regular secondary structures such as the  $\alpha$  helix and the  $\beta$  conformation.
- These structures are defined by particular values of  $\phi$  and  $\psi$  and their formation is impacted by the amino acid composition on their segment.
- All of the  $\phi$  and  $\psi$  values for a given protein structure can be visualized using a Ramachandran plot.

## Secondary Structure

- **secondary structure** = describes the spatial arrangement of the main-chain atoms in a segment of a polypeptide chain
  - *regular* secondary structure =  $\phi$  and  $\psi$  remain the same throughout the segment
  - common types =  $\alpha$  helix,  $\beta$  conformation,  $\beta$  turn, random coils

### **The $\alpha$ Helix is a Common Protein Secondary Structure**

- $\alpha$  helix = simplest arrangement, maximum number of hydrogen bonds
  - backbone wound around an imaginary longitudinal axis
  - R groups protrude out from the backbone
  - Each helical turn = 3.5 residues,  $\sim 5.4$  Å