## JEE EXPERT

## **JEE MAIN EXCLUSIVE TOPICS**

(SPECIAL TEST - 01)
PHYSICS, CHEMISTRY & MATHEMATICS
Batch: 12th Pass (A01 & A02) [19.12.2019]

### **Answer Key**

PHYSICS										
1	<b>(C)</b>	2	<b>(B)</b>	3	<b>(C)</b>	4	<b>(A)</b>	5	<b>(D</b> )	
6	<b>(A)</b>	7	<b>(C)</b>	8	<b>(C)</b>	9	<b>(C)</b>	10	<b>(C)</b>	
11	<b>(B)</b>	12	<b>(B)</b>	13	<b>(A)</b>	14	<b>(C)</b>	15	<b>(C)</b>	
16	<b>(B)</b>	17	<b>(B)</b>	18	<b>(D)</b>	19	<b>(A)</b>	20	<b>(A)</b>	
21	<b>(A)</b>	22	<b>(A)</b>	23	<b>(A)</b>	24	<b>(D</b> )	25	<b>(C)</b>	
26	<b>(A)</b>	27	<b>(A)</b>	28	<b>(A)</b>	29	<b>(C)</b>	30	<b>(A)</b>	
CHEMISTRY										
31	<b>(B)</b>	32	(A)	33	<b>(D)</b>	34	<b>(A)</b>	35	<b>(C)</b>	
36	<b>(C)</b>	37	(A)	38	<b>(C)</b>	39	<b>(C)</b>	40	<b>(B)</b>	
41	<b>(D)</b>	42	<b>(C)</b>	43	<b>(C)</b>	44	<b>(C)</b>	45	<b>(A)</b>	
46	<b>(B)</b>	47	<b>(D)</b>	48	<b>(A)</b>	49	<b>(B)</b>	50	<b>(C)</b>	
51	<b>(A)</b>	52	<b>(D)</b>	53	<b>(C)</b>	54	<b>(A)</b>	55	<b>(C)</b>	
<b>56</b>	(A)	57	<b>(A)</b>	58	<b>(C)</b>	59	<b>(A)</b>	60	<b>(A)</b>	
MATHEMATICS										
61	<b>(B)</b>	62	<b>(A)</b>	63	<b>(A)</b>	64	<b>(B)</b>	65	<b>(D)</b>	
66	(A)	67	<b>(B)</b>	68	<b>(D)</b>	69	<b>(C)</b>	70	<b>(C)</b>	
<b>71</b>	<b>(D)</b>	72	<b>(C)</b>	73	<b>(C)</b>	74	<b>(A)</b>	75	<b>(D)</b>	
<b>76</b>	<b>(D)</b>	77	<b>(B)</b>	78	<b>(C)</b>	79	<b>(C)</b>	80	<b>(C)</b>	
81	<b>(B)</b>	82	<b>(C)</b>	83	<b>(B)</b>	84	<b>(C)</b>	85	<b>(B)</b>	

**(A)** 

**89** 

**(C)** 

**90** 

**(A)** 

88

86

**(B)** 

**87** 

**(B)** 

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## JEE MAIN EXCLUSIVE TOPICS

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#### **Solutions**

### **PHYSICS**

- **3.** If the length .....
- **Sol.** B A =  $(b \pm \Delta b)$   $(a \pm \Delta a)$  =  $(b-a) \pm (\Delta b + \Delta a)$ =  $(4.19 - 3.25) \pm (0.01 + 0.01) = 0.94 \pm 0.02$ cm
- **4.** The vernier .....
- **Sol.** L.C = 1 MSD I VSD
- **5.** For a faulty ......
- Sol. Zero error = MSR + (n)(L.C) = -0.1 + (7)(0.01) = -0.03 cm Actual value = Measured value - Zero error = 8.65 - (-0.03) = 8.68 cm
- **6.** A block of .....
- $\textbf{Sol.} \quad A = \frac{F_0}{\sqrt{m^2 \left(\omega^2 \omega_d^2\right)^2 + \omega_d^2 b^2}}$

In this case  $\omega_d = \omega = \sqrt{\frac{k}{m}} = 20 \text{ rad/s}$ 

so, 
$$A = \frac{F_0}{\omega_d b} \Rightarrow b = \frac{F_0}{\omega_d A} = \frac{0.1}{20 \times 5} = 10^{-3}$$

- 7. A carnot's .....
- **Sol.** Efficiency =  $\eta = 1 \frac{T_L}{T_H} = 1 \frac{300}{900} = \frac{2}{3}$

:. If heat rejected is E

 $\Rightarrow$  heat supplied is 3E and work done is 2E

So, 
$$2E = \frac{1}{2} \left( \frac{mR^2}{2} \right) \omega^2 \Rightarrow \omega = \sqrt{\frac{8E}{mR^2}}$$

- **8.** A monochromatic .....
- Sol.  $\frac{1}{2} = \frac{1}{2} \varepsilon_0 E_{rms}^2 \Rightarrow 1 = \frac{1}{2} \varepsilon_0 E_0^2$  $\frac{P}{4\pi r^2} = \frac{1}{2} \varepsilon_0 E_0^2 C \Rightarrow E_0 = \frac{1}{r} \sqrt{\frac{P}{2\pi \varepsilon_0 C}}$ and  $C = \frac{1}{\sqrt{\mu_0 \varepsilon_0}} \Rightarrow \frac{1}{\varepsilon_0} = \mu_0 C^2$

so, 
$$E_0 = \frac{1}{r} \sqrt{\frac{P}{2\pi C} \times \mu_0 C^2} = \frac{1}{r} \sqrt{\frac{\mu_0 PC}{2\pi}}$$

- **9.** In a Searle's .....
- **Sol.** Young's modulus of elasticity is given by

$$Y = \frac{stress}{strain} = \frac{F/A}{\ell/L} = \frac{FL}{\ell A} = \frac{FL}{\ell \left(\frac{\pi d^2}{4}\right)}$$

Substituting the values, we get

$$y = \frac{50 \times 11 \times 4}{\left(1.25 \times 10^{-3}\right) \times \pi \left(5.0 \times 10^{-4}\right)^{2}} = 2.24 \times 10^{11}$$

 $N/m^2$ 

Now, 
$$\frac{\Delta Y}{Y} = \frac{\Delta L}{L} + \frac{\Delta \ell}{\ell} + 2\frac{\Delta d}{d}$$
  
=  $\left(\frac{0.1}{110}\right) + \left(\frac{0.001}{0.125}\right) + 2\left(\frac{0.001}{0.05}\right) = 0.0489$ 

 $\Delta Y = (0.0489)Y = (0.0489) \times (2.24 \times 10^{11}) \text{ N/m}^2$ = 1.09 × 10<sup>10</sup> N/m<sup>2</sup>

- **10.** A paramagnetic .....
- **Sol.** If the temperature is decreased, the thermal vibration will be reduced. So there would not be negative effect on magnetisation.
- **11.** The following ......
- **Sol.** Out of upper ABD gate =  $A\overline{B}$ Output of lower AND gate =  $B\overline{A}$

Output  $Y = A\overline{B} + B\overline{A}$ 

This is Boolean expression for XOR gate.

- **13.** A Ge specimen .....
- **Sol.** When Ge specimen is doped with A  $\ell$ , then concentration of acceptor atoms is also called concentration of holes.

Using formula  $n_i^2 = n_e n_h$ 

 $n_i = \text{concentration of electron hole pair} = 10^{19} \text{/m}^3$ 

 $n_e$  = concentration of electrons

 $n_h = concentration of holes = 10^{21} atoms/m^3$ 

$$(10^{19})^2 = 10^{21} \times n_e, n_e = 10^{17} / m^3$$

- **14.** In a given .....
- Sol. Given at saturation

$$V_{CE} = 0V, V_{BE} = 0.8V$$

$$V_{CE} = V_{CC} - l_C R_C$$

$$l_C = V_{CC}/R_C = 5.0V/1.0k\Omega = 5.0mA$$

Therefore  $l_B = /\beta = 5.0 \text{ mA}/250 = 20 \,\mu\text{A}$ 

The input voltage at which the transistor will go into saturation is given by

$$\begin{split} V_{lH} &= \, V_{BB} \, = \, l_B R_B \, + \, V_E \quad = & 20 \, \mu A \times 100 k \, \Omega \, + \\ 0.8 V &= 2.8 V \end{split}$$

- **15.** The current .....
- **Sol.** Given that,

$$V_1 = 0.01 \text{ volt}$$

$$R_1 = 1k\Omega = 10^3 \Omega$$

$$\therefore l_b = \frac{V_I}{R_I} = \frac{0.01}{1 \times 10^3} = 0.01 \times 10^{-3} \Omega = 0.01 \text{ mA}$$

Further,  $l_C = \beta l_b = 50 \times 0.01 \text{mA}$ 

$$= 0.5 \text{ mA} = 500 \,\mu\text{A}$$

- **16.** The current gain .....
- Sol.  $\alpha = \frac{I_C}{I_E} = \frac{I_E I_B}{I_E} \Rightarrow 0.95 = \frac{10 I_B}{10}$  $\Rightarrow I_B = 0.5 \text{ mA}$
- **17.** In an npn .....
- Sol. Collector current,

$$I_c = 10 \text{ mA} \; ; \; \alpha = \frac{90}{100} = 0.9$$

$$\alpha = \frac{I_{C}}{I_{E}} \Longrightarrow Emitter \ current, \ I_{E} = \frac{I_{C}}{\alpha}$$

$$I_E = \frac{10}{0.9} = 11 \text{mA}$$

Base current,  $I_B = I_E - I_C$ 

$$I_{E} = \frac{10}{0.9} 11 \text{mA}$$

Base current,  $I_B = I_E - I_C$ 

$$I_B = 11 - 10 = 1 \text{ mA}$$

- **18.** Following diagram .....
- **Sol.** We know that combination of AND gate and NOT gate is known as NAND gate since the given symbol contains AND gate and NOT, therefore it is a NAND gate.
- **19.** If p is the ......
- **Sol.** Stress = weight/area =  $A\ell \rho g$  /A =  $\ell \rho g$   $\Rightarrow \sigma = \ell \rho g$
- **20.** A lift is tied ......
- **Sol.** The tension T in the rope of the lift when it goes upward is given by

T = (m) (g + a) = 1000 (9.8 + 1.2) = 11000 NLet r be the radius of the wire, then the

maximum stress will be  $\frac{T}{\pi r^2}$ 

Hence, 
$$\frac{T}{\pi r^2} = 1.4 \times 10^8$$

or

$$r^{2} = \frac{T}{\pi \times 1.4 \times 10^{8}} = \frac{11000}{3.14 \times \left(1.4 \times 10^{8}\right)} = \frac{1}{4 \times 10^{4}}$$

Now 
$$r = \frac{1}{2 \times 10^2} = \frac{1}{200} = 0.005 m$$

Therefore, diameter of the wire = 2r = 0.01 m.

**21.** What is the .....

**Sol.** 
$$m\frac{g}{2} = mg - F_B - F_V$$

$$F_V + F_B = \frac{1}{2} \, mg$$

$$6\pi\eta rv + \sigma \frac{4}{3}\pi r^3g = \frac{1}{2}\rho \frac{4}{3}\pi r^3g$$

$$v = \frac{r^2 g}{9\eta} (\rho - 2\sigma)$$

- **22.** The excess .....
- Sol.  $\frac{P_1}{P_2} = 4$  or  $P_1 = 4P_2$   $\frac{4s}{r^1} = \frac{4 \times 4s}{r_2} \Rightarrow \frac{r_1}{r_2} = \frac{1}{4}$  $\frac{V_1}{V_2} = \frac{r_1^3}{r_2^3} = \left(\frac{1}{4}\right)^3 = \frac{1}{64}$
- **23.** A mercury drop .....
- **Sol.** Energy expended will be the work done against the increase in surface area i.e.

$$\Delta S = n(4\pi r^2) - 4\pi R^2$$

$$\Delta E = W = T\Delta S = T.4\pi(nr^2 - R^2)$$

But the total volume remains constant.

i.e. 
$$\frac{4}{3}\pi R^3 = n\frac{4}{3}\pi r^3$$

or 
$$r = \frac{R}{(n)^{1/3}}$$
 and hence,  $E = 4 \pi R^2 T(n^{1/3} - 1)$ 

- **24.** Water rises in .....
- **Sol.**  $L = \frac{h'}{\cos \phi} = \frac{\left(\frac{2T\cos\theta}{\rho gr'}\right)}{\cos \phi} = \frac{3h}{\cos \phi}$
- **25.** An air bubble .....
- **Sol.** Excess pressure inside an air bubble in soap solution

$$\frac{2T}{r} = \frac{2 \times 2.50 \times 10^{-2}}{5.00 \times 10^{-3}} = 10 \text{ pa}$$

Total pressure inside the air bubble 2T

$$=P_0+h\rho g+\frac{2T}{r}$$

$$= 1.01 \times 10^5 + 0.4 \times 1.2 \times 10^3 \times 9.8 + 10$$

$$=1.06\times10^{5} \, \text{Pa}$$

- **26.** In Amplitude .....
- **Sol.** Modulation index =  $\frac{Am}{Ac}$
- **27.** Consider an .....
- Sol. Optical source frequency

$$f = \frac{c}{\lambda} = 3 \times 10^8 / (800 \times 10^{-9})$$
$$= 3.8 \times 10^{14} \text{ Hz}$$

Bandwidth of channel (1% of above) =  $3.8 \times 10^{12}$  Hz

Number of channel =(Total bandwidth of channel)/ (Bandwidth needed per channel)

Number of channels for audio signal

$$=(3.8\times10^{12})/(8\times10^3)=4.8\times10^8$$

- **28.** The diameter .....
- **Sol.**  $d\theta = \frac{1.22\lambda}{a}$
- **29.** A simple microscope ......
- **Sol.** R.P. =  $\frac{2\mu \sin 30^{\circ}}{\lambda}$
- **30.** Linearly polarised .....
- **Sol.** In incident light,  $\frac{E_z}{E_v} = \sqrt{3} \Rightarrow \vec{E}$

is at angle 60° from y-axis

whereas transmission axis (y = z) is at  $45^{\circ}$  from y-axis

So, 
$$E_{transmitted} = E_{incident} \cos(60^{\circ} - 45^{\circ})$$
  
=  $2E_0 \cos 15^{\circ}$ 

#### **CHEMISTRY**

31. The phenomenon of scattering of light from the surface of collidal particles is called tyndall effect. The minimum conc. required for micelle formation is called critical micelle concentration.

For soap solution above CMC tyndall effect is observed.

- 33. Colloidal solution shows brownian movement. The Brownian movement does not allow the collidal particles to settle down to gravity & thus responsible for their stability.
- 34. Easily liquefiable gases like  $CO_2$  are adsorbed to a greater extent than gases like  $O_2$ ,  $N_2$  and  $H_2$
- **35.** Higher the gold number, lesser will be the protective power of colloid.

36. Mass adsorbed = 
$$\frac{100(0.3 - 0.125)60}{1000}$$
$$= (0.3 - 0.125)6$$

Mass adsorbed per gram of charcoal 
$$(0.3-0.125)6$$

Mass adsorbed per gram of charcoal
$$= \frac{(0.3 - 0.125)6}{0.8} = 1.31g$$

- **37.** (1)  $\Delta G = \Delta H - T\Delta S < O$  as  $\Delta S < O$  so  $\Delta H$  has to be negative.
  - (2) micelles formation will take place above T<sub>k</sub> and above CMC.
  - (4) Fe<sup>3+</sup> ions will have greater flocculatibility power so smaller flocculating value.

#### **MATHEMATICS**

- The number of .....
- **Sol.**  $R_1 = \{(a, a), (b, b), (c, c)\}$  $R_2 = (a, b), (b, a), (a, a), (b, b), (c, c)$  $R_3 = (a, c), (c, a), (a, a), (c, c), (b, b)$

 $R_4 = (b, c), (c, b), (b, b), (c, c), (a, a)$ 

 $R_5 = \{(a, a), (b, b), (c, c), (a, b), (b, a), (a, c), \}$ (c, a)

(b, c), (c, b)

- **62.** Let A and B be .....
- **Sol.**  ${}^{6}C_{3} + {}^{6}C_{4} + {}^{6}C_{5} + {}^{6}C_{6} = 2^{6} {}^{6}C_{0} {}^{6}C_{1} {}^{6}C_{2}$ = 64 - 1 - 6 - 15 = 64 - 22 = 42
- Let A and B be .....
- **Sol.** B:  $\frac{(x-4)^2}{9} + \frac{(y-3)^2}{4} \le 1$ A: 2 < x < 4 and 2 < y < 4
- **64.** Consider the following .....
- **Sol.** The given relation is only symmetric
- **65.** Let  $A = \{(x, y) : \dots$
- **Sol.** APA  $\Rightarrow$  reflexive area (A - B) = 0 and area (B - A) = $4\pi - \pi = 3\pi$  $\Rightarrow$  (A, B)  $\notin$  P  $\Rightarrow$  symmetric and transitive
- Let R and S be .....
- **Sol.** Let  $A = \{1, 2, 3\}$  $R = \{(1, 1), (1, 2)\}$

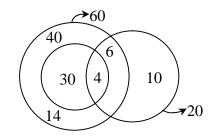
- $S = \{(2, 2), (2, 3)\}$  be transitive relations on A. Then  $R \cup S = \{(1, 1), (1, 2), (2, 2), (2, 3)\}$ Which shows that  $R \cup S$  is not transitive Since  $(1, 2) \in R \cup S$  and  $(2, 3) \in R \cup S$  but  $(1,3) \notin \mathbb{R} \cup \mathbb{S}$
- Let  $A = \{z : |z|^2 .....$ **67.**
- The given equation are the equations of two circles touching, each other extremally.
- **68**. Consider relations .....
- **Sol.**  $R_1$  is reflexive, not symmetric and transitive R<sub>2</sub> is not reflexive, symmetric and not transitive
- The intersection of .....
- Sol. Obvious
- **70.** If 'R' is an equivalence .....  $\forall a \in A, (a,a) \in R \Rightarrow (a,a) \in R^{-1}$  $[(a,b) \in R \Rightarrow (b,a) \in R]$  $\Rightarrow$  { $(a,b) \in R^{-1}$ }  $\land$  { $(b,a) \in R^{-1}$ }  $[(a,b) \in R \land (b,c) \in R] \rightarrow (a,c) \in R$  $[(c,b) \in R^{-1} \land (b,a) \in R^{-1}] \rightarrow (c,a) \in R^{-1}$ : It is an equivalence relation
- 71. Let  $P = \left\{\theta : \sin^2 \theta + \cos^4 \theta = \frac{1 \sin \theta}{\cos \theta}\right\}$ .....
- **Sol.**  $\sin^2 \theta + \cos^4 \theta = \cos^2 \theta + \sin^4 \theta$ and  $\frac{1-\sin\theta}{\cos\theta} = \cot\left(\frac{\pi}{4} + \frac{\theta}{2}\right)$
- **72.** In a survey of ......
- **Sol.**  $n(M' \cap C' \cap P') = n(\bigcup) n(M \cap P \cap C)$  $\Rightarrow$  n(M $\cup$ P $\cup$ C) = 450  $n(M \cap P \cap C) = 20$ who liked exactly one = 95 + 40 + 190 =325
- Number of elements .....
- **Sol.**  $P = \frac{2n^5 + 3n^4 + 4n^3 + 4n^2 + 6}{n}$  is integer  $\Rightarrow$  n = 1, 2, 3, 6

- **74.** Find cardinal number .....
- **Sol.** Let A be the set of positive integers not exceeding 1000 that are divisible by 7 and B→divisible by 11

$$n(A \cup B) = n(A) + n(B) - n(A \cap B)$$

$$= \left\lceil \frac{1000}{7} \right\rceil + \left\lceil \frac{1000}{11} \right\rceil - \left\lceil \frac{1000}{77} \right\rceil = 142 + 90 - 12 = 220$$

- **75.** The relation R .....
- Sol. Obvious
- **76.** Which of the following ......
- Sol. Obvious
- **77.** Consider relation ......
- **Sol.** (a, b) is not related to (a, b) as LCM(a, a)  $\neq$  LCM(b, b)
- **78.** If A, B, C are 3 .....
- Sol.



- **79.** The relation T .....
- **Sol.**  $a \ge a^{\frac{2}{3}} \Rightarrow a^3 \ge a^2$  not true for  $\overline{Z}$  so not reflexive

 $27 \ge 8^{\frac{2}{3}}$  but  $8 \not\ge 27^{\frac{2}{3}}$  so not symmetric

$$a \ge b^{\frac{2}{3}}$$
 and  $b^3 C^{\frac{2}{3}}$ 

 $(:: a \ge 0, b \ge 0)$ 

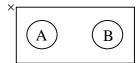
$$\Rightarrow a \ge (C^{\frac{2}{3}})^2 \Rightarrow a^3 C^{\frac{2}{3}}$$

 $\left(\because (C^{\frac{2}{3}})^2 \ge C^{\frac{2}{3}}\right)$ 

- ⇒ transitive
- **80.** Which of the following .....
- **Sol.** Obvious
- **81.** Consider relation .....
- Sol. Obvious
- **82.** Let  $A = \{2, 3, 4, ..., 17, 18\}$
- **Sol.** (21 to 22)

$$(3,2) \simeq (c,d) \Rightarrow 3d = 2c \Rightarrow c = \frac{3d}{2}$$
  
  $\Rightarrow d = 2, 4, 6, 8, 10, 12$ 

- **83.** Let A and B be .....
- **Sol.**  $(A \cap B = \emptyset)$  $\Rightarrow A \subset (X - B)$



- **84.** Let A consist of .....
- **Sol.** The subsets of A which contain atteast one of  $a_1$ ,  $a_2$ , .......  $a_r$  and all of  $a_{s+1}$ ,  $a_{s+2}$  .....  $a_n$  are obtained, once each, as the unions of  $\{a_{s+1}, a_{s+2}, \dots, a_n\}$  with the subsets of  $\{a_1, a_2, \dots, a_s\}$  which contains at east one of  $a_1, a_2, \dots, a_s$  which contains at least

$$=2^{s-r}(2^r-1)$$

- **85.** Relation  $\otimes$  is defined ......
- Sol. Obvious
- **86.** The number of .....
- **Sol.** Total no. of relations =  $2^{4\times4} = 2^{16}$ In cartesian product we have 16 ordered pairs in which 4 are compulsory of reflexive relation, out of 12 ordered pairs. We can take

$$^{12}$$
C<sub>0</sub> +  $^{12}$ C<sub>1</sub> +  $^{12}$ C<sub>2</sub> + ..... +  $^{12}$ C<sub>12</sub> =  $2^{12}$ 

- **87.** Consider relation .....
- **Sol.**  $(a, b) \in T$  and  $(b, a) \in T$  but  $(a, a) \notin T$  so not transitive
- **88.** If an examination .....
- **Sol.** Only failed in physics = 37% 19% = 18%= alone pass in maths. let total = xx = 1700
- **89.** Relation C is defined .....
- Sol. Obvious
- **90.** Consider the two .....
- Sol. Obvious