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| Course Name: <b>Communication Skills Lab</b>   |                           |
| Course Code: <b>HS-102</b>   |                           |
| Contact Hours/Week: <b>2P</b>  | Course Credits: <b>01</b> |
| <b>Course Objectives</b> <ul style="list-style-type: none"> <li>• To provide skills for listening with understanding and speaking</li> <li>• To provide skills 'correct' pronunciation of English language</li> <li>• To enable the students to make oral and technically aided presentations</li> </ul>   |                           |
| <b>List of Experiments</b><br>Activities based on language software Sky Pronunciation/others: <ol style="list-style-type: none"> <li>1. Sky Pronunciation: Introduction to the Speech Sounds of English</li> <li>2. Sky Pronunciation: Syllable and Organs of Speech</li> <li>3. Sky Pronunciation: Vowel and Consonant Sounds</li> <li>4. Sky Pronunciation: Similar sounds and test</li> <li>5. Word Stress and Intonation using available software</li> <li>6. Listening and Comprehension using available software</li> <li>7. Listening to Native speakers of English language</li> <li>8. Watching short talks for learning effective presentation skills</li> <li>9. Presentation skills using technology enabled slides</li> <li>10. Just a Minute (JAM) Sessions</li> <li>11. Describing Objects/Situations/People</li> <li>12. Interview skills using available software/interview videos</li> </ol> |                           |
| <b>Course Outcomes</b><br>Upon successful completion of the course, the students will be able to<br>CO1: Speak coherently<br>CO2: Make effective Presentations<br>CO3: Listen and comprehend English language  |                           |

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| Course Name: <b>Engineering Chemistry Lab</b>   |                           |
| Course Code: <b>CY-102</b>  |                           |
| Contact Hours/Week: <b>2P</b>   | Course Credits: <b>01</b> |
| <b>Course Objectives</b> <ul style="list-style-type: none"> <li>• To analyse water samples for different parameters like amount of chloride ions, residual chlorine, alkalinity and hardness</li> <li>• To measure physical properties of liquids</li> <li>• To estimate the percentage of a particular metal in its ore or alloy</li> <li>• To familiarize students about the characterization method like absorption spectroscopy</li> </ul>  |                           |
| <b>List of Experiments</b> <ol style="list-style-type: none"> <li>1. Estimation of residual Chlorine in a given sample of water</li> <li>2. Estimation of chloride content in a given sample of water by Mohr's method (Argentometrically)</li> <li>3. Estimation of concentration of hydroxyl, carbonate, bicarbonate and total alkalinity in a given sample of water</li> <li>4. Estimation of Hardness (Temporary and Permanent) in a given sample of water</li> <li>5. Determination of quantity of Ferrous ions in a sample of water by <math>\text{KMnO}_4</math> titration</li> <li>6. Estimation of concentration of iron in an iron ore by dichrometry.</li> <li>7. Estimation of Cu in a given sample of brass</li> <li>8. Determination of Viscosity of unknown liquid by Ostwald's viscometer</li> <li>9. Determination of surface tension of unknown liquid by drop number method.</li> <li>10. Estimation of calcium in Limestone or Dolomite</li> <li>11. Verification of the absorption laws by using Colorimetric method</li> <li>12. Determination of the concentration of nickel using Absorption technique</li> </ol> <p><b>Note:</b> The concerned Course Coordinator will prepare the actual list of experiments/problems at the start of semester based on above generic list.</p> |                           |
| <b>Course Outcomes</b> <p>Upon successful completion of the course, the students will be able to</p> <p>CO1: Quantify different pollutants in water samples</p> <p>CO2: Identify the unknown liquid from their surface tension and viscosity measurement.</p> <p>CO3: Analytically measure the composition of alloy and ores</p>  |                           |

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| Course Name: <b>Electrical Engineering Lab</b>   |                           |
| Course Code: <b>EE-102</b>   |                           |
| Contact Hours/Week: <b>2P</b>  | Course Credits: <b>01</b> |
| <b>Course Objectives</b> <ul style="list-style-type: none"> <li>• To impart basic knowledge of electrical quantities such as current, voltage, power, energy etc.</li> <li>• To familiarize students with basic circuit components and their connections.</li> <li>• To explain working principle of electrical measuring instruments such as ammeter, voltmeter, wattmeter, energy meter, etc.</li> </ul>   |                           |
| <b>List of Experiments</b> <ol style="list-style-type: none"> <li>1. To verify Ohm's law for BPLL (Bilateral Passive Linear Lumped) element.</li> <li>2. To find for a filament lamp:             <ol style="list-style-type: none"> <li>i. Variation of resistance with voltage.</li> <li>ii. Variation of power with voltage.</li> </ol> </li> <li>3. To find minimum fusing current and fuse constant of a given fuse wire.</li> <li>4. To calibrate a given voltmeter with the help of standard ammeter and resistance.</li> <li>5. To calibrate a given ammeter with the help of standard voltmeter and resistance.</li> <li>6. To find voltage current relationship in R-L series circuit and to determine power factor of the circuit.</li> <li>7. To calibrate given wattmeter by direct loading.</li> <li>8. To calibrate single phase energy meter by direct loading.</li> <li>9. Verification of Kirchhoff's Laws:             <ol style="list-style-type: none"> <li>i. KVL (Kirchhoff's Voltage Law)</li> <li>ii. KCL (Kirchhoff's Current Law)</li> </ol> </li> <li>10. Determination of inductance of a coil using voltmeter, ammeter methods.</li> <li>11. To verify total resistance <math>R</math> of the series connected resistances <math>R = R_1 + R_2 + R_3</math></li> </ol> |                           |
| <b>Course Outcomes</b> <p>Upon successful completion of the course, the students will be able to</p> <p>CO1: Verify fundamental laws like Ohm's Law, KCL, KVL, etc.</p> <p>CO2: Use different meters and instruments for the measurement of common electrical quantities</p> <p>CO3: Understand the importance of fuse as a safety device and study the parameters related with the selection of fuse wire</p>   |                           |