# BIRLA INSTITUTE OF TECHNOLOGY AND SCIENCE, PILANI HYDERABAD CAMPUS

## **INSTRUCTION DIVISION, FIRST SEMESTER 2016-2017**

# Course Handout (Part II)

Date: 12/5/2016

In addition to part-I (General Handout for all courses appended to the time table) this portion gives further specific details regarding the course.

Course No. : CHE F311

Course Title : Kinetics and Reactor Design

Instructor-in-charge : Balaji Krishnamurthy

## **Course Description:**

Kinetics of homogeneous, heterogeneous reactions; ideal reactors, non-ideal flow; selectivity; analysis and design of chemical reactors.

## **Scope and Objective of the course:**

This course is an introduction to the chemical reaction kinetics, design and performance of various types of chemical reactors for chemically reacting systems which yield industrially important products. The emphasis in this course will be to understand the fundamentals of kinetics of homogeneous reactions, design and analysis of ideal reactors; and non-ideal flow.

#### **Text Book:**

- 1. Scott Fogler, H. Scott "Elements of Chemical Reaction Engineering", Pearson Edu, 4th Ed, 2006.
- 2. Octave Levenspiel, Chemical Reaction Engineering. 3<sup>rd</sup> Ed

### **Reference Books:**

1. Schmidt Lanny D., "The Engineering of Chemical Reactions", Oxford University Press, 2nd Ed., 2005.

### **Course Plan:**

Lec.	<b>Learning Objectives</b>	Topics to be covered	
L1-L3	Introduction	Scope and objectives of the course,	TB
		methodology, concept of mole balances	-1&2
L4-L7	Kinetics of reactions	Different types of reactors, mole balances	TB
			-1&2
L8-L12	Batch reactors	Conversion and reactor sizing	TB
			-1&2
L13-L18	CSTR and PFR reactors	Equations governing conversions	TB
			-1&2
L-19-L23	Multiple reactor systems	Reactors in series and parallel	TB
			-1&2
L-24-L28	Multiple reactions	Reactions of various orders	TB
			-1&2
L29-L35	Solid catalyzed	Pore diffusion factors fitting first and second	TB
	reactions	order Catalyst deactivation	-1&2
L36-L42	Basics of non-ideal	Non ideal behavior	TB
	reactor		-1&2

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#### **Evaluation Schedule:**

Component	Duration (Minutes)	Weightage	Date & Time	Remarks
Test-1	1hr	25%		CB
Test-2	1hr	25%		OB
Surprise/quiz tests*	-	10%		CB
Comprehensive Exam.	3hr	40%		CB/OB

**Closed Book Test:** No reference material of any kind will be permitted inside the exam hall.

**Open Book Exam:** Use of any printed / written reference material (books and notebooks) will be permitted inside the exam hall. Loose sheets of paper will not be permitted. Computers/mobile of any kind will not be allowed inside the exam hall. Use of calculators will be allowed in all exams. No exchange of any material will be allowed.

Number and Sequence of lecture may be changed depending on the situation/requirements. Quiz/ surprise test will be conduct for 2 times and average of the marks will be taken as final.

**Chamber Consultation Hour:** To be announced in the class.

**Notices:** All notices concerning this course will be displayed on the Notice Board of Chemical Engineering or CMS

**Make-up Policy:** Make-up for the test (test-1 and test-2) may be granted only with prior permission and valid justification from the Instructor-in-charge. No makeup for the quiz/surprise tests will be granted.

Instructor-in-charge Kinetics and Reactor Design