

Name:

ID:

QUIZ 8 – CH7

- 1) You built a cabin in the woods and for electricity you decide to build a small hydroelectric generator under a 75-m high waterfall. The flow rate of the waterfall is 10 m³/h, and you anticipate needing 750 kW h/wk to run your lights, air conditioner, and television. Calculate the maximum power theoretically available from the waterfall and see if it is sufficient to meet your needs.

($g = 9.82 \text{ m/s}^2$, $1 \text{ N} = 1 \text{ kg}\cdot\text{m/s}^2$, $1 \text{ J} = 1 \text{ N}\cdot\text{m}$, $1 \text{ J} = 2.778 \times 10^{-7} \text{ kW}\cdot\text{h}$)

- 2) Oxygen at 150 K and 41.64 atm has a tabulated specific volume of 4.684 cm³/g and a specific internal energy of 1706 J/mol. Calculate the specific enthalpy of O in this state.

(MW Oxygen = 32 g/mol, $1 \text{ L}\cdot\text{atm} = 101.32500 \text{ joules}$)

QUIZ SOLUTIONS:

1) 7.8

$$\Delta \dot{E}_p = \dot{m} g \Delta z = \frac{10^3 \text{ m}^3}{\text{h}} \left| \frac{10^3 \text{ L}}{1 \text{ m}^3} \right| \left| \frac{1 \text{ kg H}_2\text{O}}{1 \text{ L}} \right| \left| \frac{9.81 \text{ m}}{\text{s}^2} \right| \left| \frac{-75 \text{ m}}{1} \right| \left| \frac{1 \text{ N}}{1 \text{ kg}\cdot\text{m/s}^2} \right| \left| \frac{1 \text{ J}}{1 \text{ N}\cdot\text{m}} \right| \left| \frac{2.778 \times 10^{-7} \text{ kW}\cdot\text{h}}{1 \text{ J}} \right|$$

$$= -2.04 \times 10^4 \text{ kW}\cdot\text{h/h}$$

The maximum energy to be gained equals the potential energy lost by the water, or

$$\frac{2.04 \times 10^4 \text{ kW}\cdot\text{h}}{\text{h}} \left| \frac{24 \text{ h}}{1 \text{ day}} \right| \left| \frac{7 \text{ days}}{1 \text{ week}} \right| = \underline{\underline{3.43 \times 10^6 \text{ kW}\cdot\text{h/week}}} \text{ (more than sufficient)}$$

2)

$$7.12 \quad \hat{V} = \frac{32.00 \text{ g}}{\text{mol}} \left| \frac{4.684 \text{ cm}^3}{\text{g}} \right| \left| \frac{10^3 \text{ L}}{10^6 \text{ cm}^3} \right| = 0.1499 \text{ L/mol}$$

$$\hat{H} = \hat{U} + P\hat{V} = 1706 \text{ J/mol} + \frac{41.64 \text{ atm}}{1} \left| \frac{0.1499 \text{ L}}{\text{mol}} \right| \left| \frac{8.314 \text{ J / (mol}\cdot\text{K)}}{0.08206 \text{ L}\cdot\text{atm / (mol}\cdot\text{K)}} \right| = \underline{\underline{2338 \text{ J/mol}}}$$

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TEAM WORK 8 – CH 7

- 1) Suppose you pour 1 gallon of water on a yowling cat 10 ft below your bedroom window.
- How much potential energy (ft lb) does the water lose?
 - How fast is the water traveling (ft/s) just before impact?
 - True or false: The kinetic energy of the water before impact must equal the kinetic energy of the cat after impact.
- 2) Methane enters a 3-cm ID pipe at 30°C and 10 bar with an average velocity of 5.00 m/s and emerges at a point 200 m lower than inlet at 30°C and 9 bar. Calculate the ΔK and ΔP assuming the methane behaves as ideal gas. (Hint: You need STP)

TEAM WORK SOLUTIONS:**1)**

$$7.6 \quad (a) \quad \Delta E_p = mg\Delta z = \frac{1 \text{ gal}}{7.4805 \text{ gal}} \left| \frac{1 \text{ ft}^3}{1 \text{ ft}^3} \right| \frac{62.43 \text{ lb}_m}{1 \text{ ft}^3} \left| \frac{32.174 \text{ ft}}{\text{s}^2} \right| \frac{-10 \text{ ft}}{1} \left| \frac{1 \text{ lb}_f}{32.174 \text{ lb}_m \cdot \text{ft} / \text{s}^2} \right| = \underline{\underline{-83.4 \text{ ft} \cdot \text{lb}_f}}$$

$$(b) \quad E_k = -\Delta E_p \Rightarrow \frac{mu^2}{2} = mg(-\Delta z) \Rightarrow u = [2g(-\Delta z)]^{1/2} = \left[2 \left(32.174 \frac{\text{ft}}{\text{s}^2} \right) (10 \text{ ft}) \right]^{1/2} = \underline{\underline{25.4 \frac{\text{ft}}{\text{s}}}}$$

(c) False

2) 7.7

Mass flow must be the same at the inlet to attain steady-state condition.

$$\Delta K = m/2(v_2^2 - v_1^2) \text{ and } \Delta P = mg(h_2 - h_1)$$

Determine the mass flow:

$$\text{Volumetric flow at the inlet} = v_1 A$$

If methane behaves as an ideal gas:

$$V = v_1 A = \frac{mRT}{(MW)(P)}$$

Solving for mass flow:

$$m = \frac{(v_1 A)(MW)(P_1)}{RT_1} = 0.0225 \text{ kg/s}$$

$$\dot{m} = \frac{5 \text{ m}}{\text{s}} \left| \frac{\pi(1.5)^2 \text{ cm}^2}{10^4 \text{ cm}^2} \right| \left| \frac{1 \text{ m}^3}{303 \text{ K}} \right| \left| \frac{273 \text{ K}}{1.01325 \text{ bars}} \right| \left| \frac{1 \text{ kmol}}{22.4 \text{ m}^3 (\text{STP})} \right| \left| \frac{16.0 \text{ kg CH}_4}{1 \text{ kmol}} \right|$$

$$= 0.0225 \text{ kg/s}$$

Solving for ΔP :

$$\Delta P = mg(h_2 - h_1) = \left(0.0225 \frac{\text{kg}}{\text{s}}\right) \left(9.81 \frac{\text{m}}{\text{s}^2}\right) (-200 \text{ m})$$

$$\Delta P = -44.1 \frac{\text{J}}{\text{s}} = -44 \text{ W}$$

Determine v_2 :

$$P_1 V_1 = P_2 V_2$$

$$P_1(v_1 A) = P_2(v_2 A)$$

Solving for v_2 :

$$v_2 = v_1 \left(\frac{P_1}{P_2} \right) = 5.00 \frac{\text{m}}{\text{s}} \left(\frac{10 \text{ bar}}{9 \text{ bar}} \right) = 5.555 \text{ m/s}$$

Solving for ΔK :

$$\Delta K = \frac{m}{2} (v_2^2 - v_1^2) = \frac{1}{2} \left(0.0225 \frac{\text{kg}}{\text{s}} \right) \left(5.555^2 - 5.00^2 \frac{\text{m}^2}{\text{s}^2} \right)$$

$$\Delta K = 0.0659 \frac{\text{J}}{\text{s}} = 0.0659 \text{ W}$$