

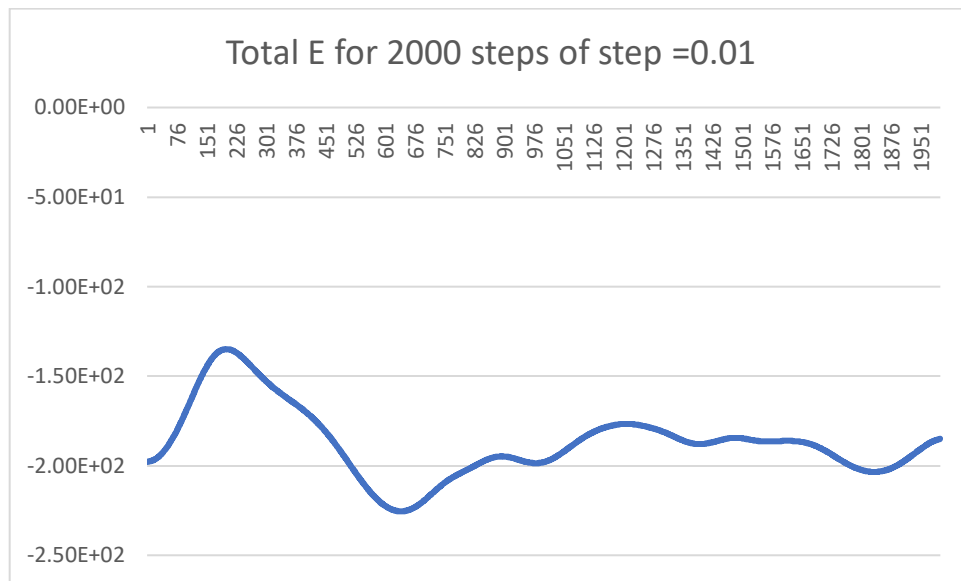
Albert Wiryawan  
[Avw2@illinois.edu](mailto:Avw2@illinois.edu)  
673431511  
PHYS 466: Atomic scale simulation

## Homework #2

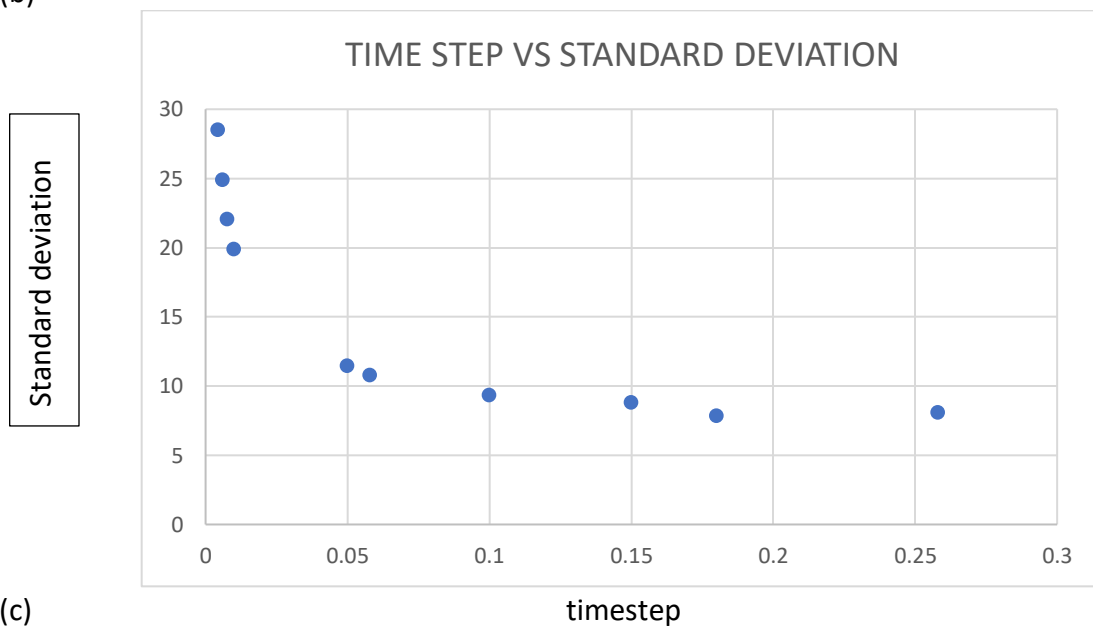
teijrhgogrojeiojgr

### Problem #1

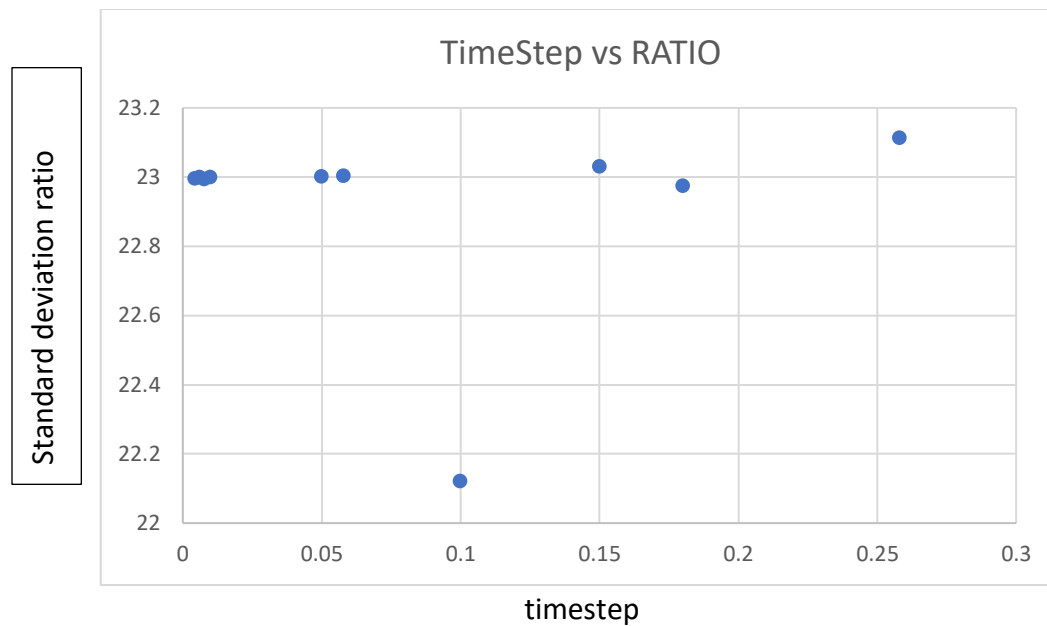
(a)



(b)



(c)



(d)

If you pick a very large time step the program has an error in the potentially energy function that is caused by a division of zero

## Problem #2

(a)

$$r(t+h) = r(t) + v(t) h + 0.5a(t) h^2$$

$$r(t-h) = r(t) - v(t)h + 0.5a(t) h^2$$

When  $r(t+h)$  and  $r(t-h)$  are added together, the total comes out to:  $2r(t) - r(t-h) + a(t)h^2$ .

This is equal to the equation below.

$$r(t+h) = 2r(t) - r(t-h) + a(t)h^2$$

This equivalence proves time reversal invariance.

(b)

Since the float precision is precise to the 7<sup>th</sup> decimal point. The  $h^2$  term in the verlet equation which is said to have a time step  $h = 10^{-4}$ . The decimals beyond that are essentially truncated and will not affect the estimation.

(c)

The equation for (9) is given by  $r(t+h) = 2r(t) - r(t-h) + a(t)h^2$ . This equation can be converted to the form of:  $v\left(t - \frac{h}{2}\right) + \frac{a(t)h}{2}$ . Taking this equation and applying the derivative law, the equation turns into  $v(t) = \frac{r(t+h)-r(t-h)}{2h}$ . Equation 10 on the other hand is  $r(t+h) = r(t) + v(t)h + \frac{a(t)h^2}{2}$  can be rearranged into  $v\left(t + \frac{h}{2}\right) = v(t) + \frac{a(t)h}{2}$ . Thus, this shows that (9) is equivalent to velocity Verlet.

(d)

If you pick a very time step there is a division by zero error.

### Problem #3

(a)

Take the derivative of  $u_{LJ} = 2\epsilon \left[ A_{12} \left( \frac{\sigma}{r_n} \right)^{12} - A_6 \left( \frac{\sigma}{r_n} \right)^6 \right]$  with respects to lattice parameter since  $r_n = \frac{a}{\sqrt{2}}$  and set this equal to zero.

This derivative turns out to be  $u_{LJ}' = 2\epsilon \left[ -12A_{12} \left( \frac{\sigma}{r_n} \right)^{13} + 6A_6 \left( \frac{\sigma}{r_n} \right)^7 \right]$

This allows us to solve for the two parameters, thus, the  $\sigma = 0.1137 \text{ \AA}^{-1}$  and  $\epsilon = 2004.84 \text{ eV per atom/}$

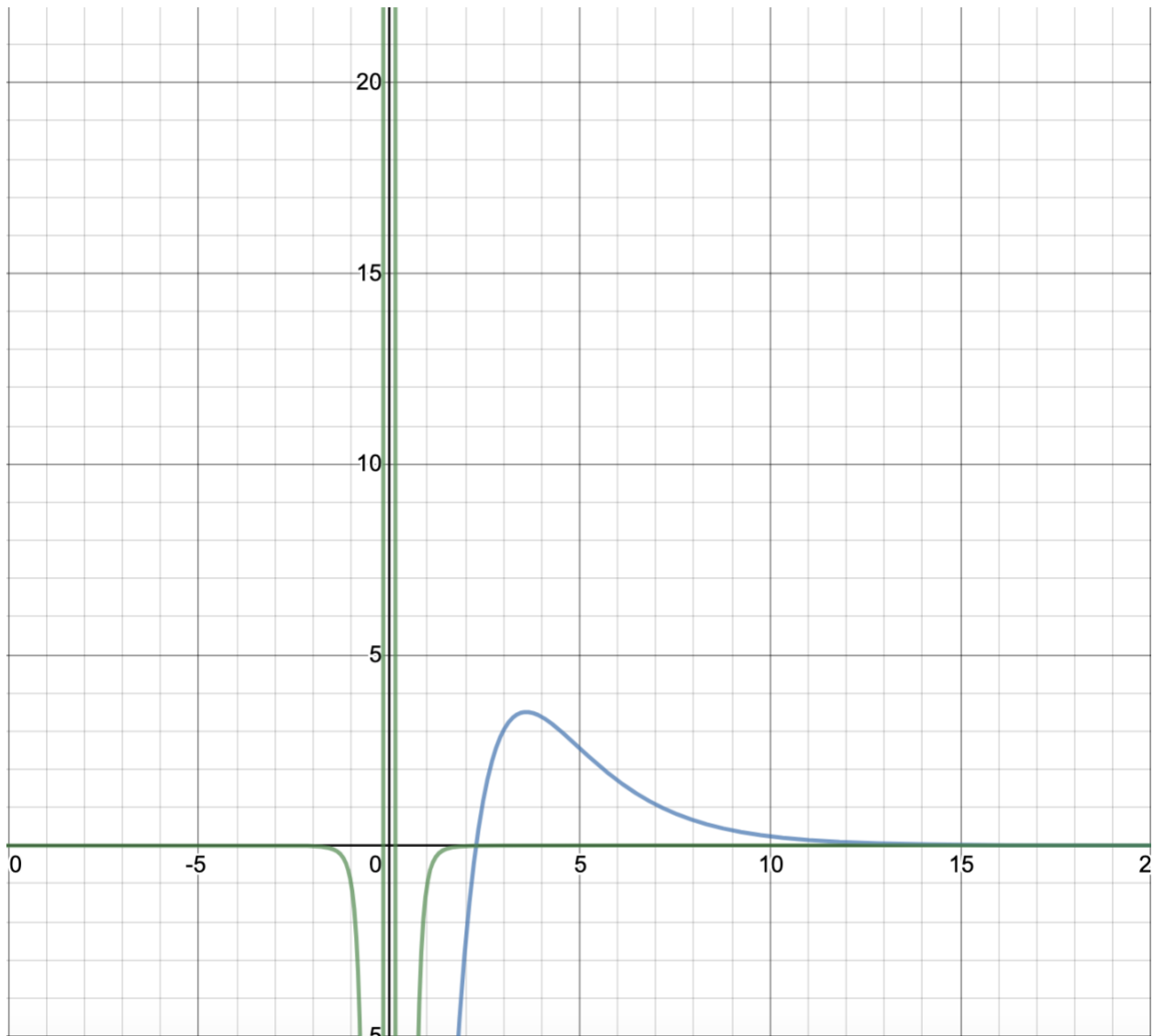
(b)

The morse potential model is given by the equation  $u_M = D_e(e^{-2\alpha(r_n-r_e)} - 2e^{-\alpha(r_n-r_e)})$ .  $r_e$  can then be found by taking the derivative of this equation with respect to the lattice parameter (a), since  $r_n$  is equal to  $\frac{a}{\sqrt{2}}$ . This results to  $u_M' = D_e(-2\alpha e^{-2\alpha(r_n-r_e)} - 2\alpha e^{-\alpha(r_n-r_e)})$ . Thus,  $r_e = r_n = \frac{a}{\sqrt{2}} = 2.546 \text{ \AA}$  as found when plugging  $r_e$  back to the original equation. From here,  $D_e$  is found to be  $-u = 3.5 \text{ eV}$ . Using the bulk modulus,  $B$  in the form of  $B = v \frac{\partial}{\partial v} \frac{\partial u}{\partial v} = v \frac{\partial}{\left(\frac{\partial v}{\partial r}\right)^2} \frac{\partial^2 u}{\partial r^2} = \frac{r^3}{\sqrt{2}} \frac{\partial}{\frac{9r^4}{2}} \frac{\partial^2 u}{\partial r^2} = \frac{\sqrt{2}}{9r} D_e(4\alpha^2 e^{-2\alpha(r_n-r_e)} + 2\alpha^2 e^{-\alpha(r_n-r_e)})$ , the constant  $B=134 \text{ GPA}$  can be used to find  $\alpha$  which is  $0.7437 \text{ \AA}^{-1}$ .

(c)

In the Lennard-Jones model, the bulk's modulus equation is the same except for u. Thus,  $B = v \frac{\partial}{\partial v} \frac{\partial u}{\partial v} = v \frac{\partial}{\left(\frac{\partial v}{\partial r}\right)^2} \frac{\partial^2 u}{\partial r^2} = \frac{r^3}{\sqrt{2}} \frac{\partial}{\frac{9r^4}{2}} \frac{\partial^2 u}{\partial r^2} = \frac{2\sqrt{2}\epsilon}{9r} \left[ -156A_{12} \left( \frac{\sigma}{r_n} \right)^{14} + 42A_6 \left( \frac{\sigma}{r_n} \right)^8 \right]$ . Using the parameters found from part a the Bulk's modulus was determined to be  $148.52 \text{ GPa}$ .

(d)



$$u_M = D_e \left( e^{-2\alpha(r_n - r_e)} - 2e^{-\alpha(r_n - r_e)} \right),$$

The green plot is given by

$$u_{\text{LJ}} = 2\epsilon \left[ A_{12} \left( \frac{\sigma}{r_n} \right)^{12} - A_6 \left( \frac{\sigma}{r_n} \right)^6 \right]$$

While the blue plot is given by