

Atomvibrationen in einem Metall

$$\text{a) } \rho = \frac{3}{4\pi R^3}; \quad V = \frac{4}{3}\pi r^3; \quad q = e\rho V; \quad k = \frac{1}{4\pi\epsilon_0}$$

$$E = \frac{kq}{r^2} = \frac{1}{4\pi\epsilon_0} \frac{er}{R^3}$$

$$F = qE = kr$$

$$\omega = \sqrt{\frac{k}{mM}} = \sqrt{\frac{e^2}{4\pi\epsilon_0 MR^3}}$$

b)

$$4R^2 = 3a^2 \quad \Rightarrow \quad R = \frac{\sqrt{3}a}{2}$$

$$\omega = \sqrt{\frac{e^2}{4\pi\epsilon_0 MR^3}} = \underline{\underline{5.11 \times 10^{10} \text{ Hz}}}$$

$$\text{c) } N = 9; \quad V = a^3$$

$$v = \frac{\Omega_D}{\sqrt[3]{\frac{6\pi^2 N}{V}}} = \underline{\underline{1715 \text{ m/s}}}$$

Zwei-atomige Kette

$$\text{a) } \omega^2 = G \left(\frac{1}{m} + \frac{1}{M} \right) \pm G \sqrt{\left(\frac{1}{m} + \frac{1}{M} \right)^2 - \frac{4 \sin^2(ka)}{Mm}}; \quad ka \ll 1; \quad \mu = \frac{1}{m} + \frac{1}{M}$$

$$\begin{aligned} \omega^2 &= G \left(\frac{1}{m} + \frac{1}{M} \right) \pm G \sqrt{\left(\frac{1}{m} + \frac{1}{M} \right)^2 - \frac{4 \sin^2(ka)}{Mm}} \approx \frac{G}{\mu} \pm G \sqrt{\frac{1}{\mu^2} - \frac{4(ka)^2}{Mm}} = \\ &= \frac{G}{\mu} \pm \frac{G}{\mu} \sqrt{1 - \frac{4(ka)^2 \mu^2}{Mm}} \approx \frac{G}{\mu} \pm \frac{G}{\mu} \left(1 - \frac{2(ka)^2 \mu^2}{Mm} \right) \end{aligned}$$

$$\omega_- = \sqrt{\frac{2G(ka)^2 \mu}{Mm}} = ka \sqrt{\frac{2G\mu}{Mm}}$$

$$v = \frac{\omega_-}{k} = \underline{\underline{a \sqrt{\frac{2G\mu}{Mm}}}}$$

b)

c)

d)

Eigenschaften eines Natriumkristalls

$$m = 22.9897 \text{ u}; \quad \rho = 0.968 \text{ g/cm}^3$$

$$\text{a) } V\rho = M$$

$$a = \sqrt[3]{\frac{9m}{\rho}} = \underline{\underline{7.08 \times 10^{-10} \text{ m}}}$$

$$\text{b) } I \propto |F_{\text{hkl}}|^2$$

$$|F_{111}|^2 = f^2 \left(\left| e^0 + e^{3i\pi} \right|^2 \right) = 0$$

$$|F_{110}|^2 = f^2 \left(\left| e^0 + e^{2i\pi} \right|^2 \right) = 2f^2$$

The Intensity for a (111) Lattice is lower than for a (110) lattice.

$$\text{c) } T_D = 150 \text{ K}$$

$$\Omega_D = \frac{T_D k_B}{\hbar} = \underline{\underline{1.96 \times 10^{13} \text{ Hz}}}$$