

Michelson-Interferometer

a)

b)

Bestimmung des Wirkungsquerschnitts

a) A cathode releases electrons, which pass through a hole in a membrane into a chamber full of nitrogen. At the end the electrons hit a detector and from the current I_0 the electrons produce one can determine the cross-section of electrons and nitrogen.

b) Instead of only varying the distance between Nitrogen molecules in one plane only, changing the pressure will yield a similar result while being much easier to achieve. When varying the distance in one plane you have to keep the molecules from escaping in other directions, which is pretty hard using electrically neutral molecules.

$$I(p) = I_0 e^{-\beta p}$$

c) $x = 2.5 \text{ m}; \quad T = 300 \text{ K}$
 $p_1 = 2 * 10^{-2} \text{ Pa}; \quad p_2 = 10^{-2} \text{ Pa}; \quad p_3 = 10^5 \text{ Pa}; \quad p_4 = 7 * 10^4 \text{ Pa}$

Geladene Teilchen in \vec{E} - und \vec{B} - Feldern

$$d = 0.105 \text{ m}; \quad B = 1 \text{ mT}; \quad U_b = 220 \text{ V}$$

a) $F_L = qv_{\perp} B = \frac{mv_{\perp}}{r} \Rightarrow v_{\perp} = \frac{dqB}{m}; \quad E = Uq = \frac{mv_{\perp}^2}{2} \Rightarrow v_{\perp}^2 = \frac{2Uq}{m}$

$$\left(\frac{dqB}{m} \right)^2 = \frac{2Uq}{m}$$

$$\frac{q}{m} = \frac{2U}{d^2 B^2} = \underline{\underline{3.99 * 10^{10} \text{ C/kg}}}$$

$$\text{b) } v = \sqrt{\frac{2Uq}{m}}$$

$$F_L = q(E + vB) = 0$$

$$E = -vB = \sqrt{\frac{2Uq}{m}} B$$

$$\frac{q}{m} = \frac{E^2}{\underline{\underline{UB^2}}}$$