207. Kupferblock in Styropor

$$\rho = 8930 \text{ kg/m}^3$$
; $c = 385 \text{ J/kg K}$

a)

209. Limonade mit Eis

$$m_1 = 0.24 \text{ kg};$$
 $T_1 = 306.15 \text{ K};$ $m_2 = 0.025 \text{ kg};$ $T_2 = 273.15 \text{ K}$

a)
$$Q = cm\Delta T$$
; $\Delta T = T_{Ende} - T_{Start}$

$$Q_{1} = cm_{1}(T - T_{1})$$

$$Q_{2} = cm_{2}(T - T_{2})$$

$$Q_{1} + 2Q_{2} = 0$$

$$\Rightarrow T = \frac{m_1 T_1 + 2m_2 T_2}{m_1 + 2m_2} = \underline{300.46 \text{ K}}$$
 (= 27.31 °C)

b)

$$Q_1 + 6Q_2 = 0$$

 $\Rightarrow T = \frac{m_1 T_1 + 6m_2 T_2}{m_1 + 6m_2} = \underline{293.46 \text{ K}}$ (= 20.31 °C)