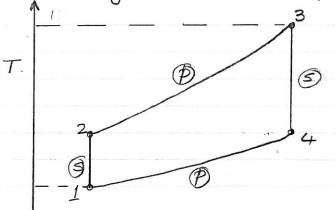
Brayton Cycle

· Single-phase

· gas turbine plants

· gas-cooled reactor plants



process

- D constant pressure
- 5) constant enloopy

Simple Brayton Cycle.

Pressure or compression ratio of the cycle $\gamma_p = \frac{\rho_2}{\rho} = \frac{\rho_3}{\rho_4}$

For a perfect gas with isentopic processo. $TV^{-1} = Constant$: PV = compt.

 $7 = c_p/c_v$ enthalpy $\Delta h = c_p \Delta T$.

Cp - constant.

Turbine and Compressor Work

Turbine: WIT = mcp (T3-T4) = mcp T3 (1- T4)

for isentapic process W= mg T3 [1- (1)

Compressor:
$$|\lambda|_{cp} = mc_p (T_2 - T_i) = mc_p T_i \left[(Y_p)^T - 1 \right]$$
Heat input from reactor: $Q_R = mc_p (T_3 - T_2) = mc_p T_i \left[\frac{T_3}{T_i} - (Y_p)^T \right]$
Heat rejected by heat exchanger: $Q_{Hx} = mc_p (T_4 - T_i) = mc_p T_3 \left[\frac{1}{T_i} - \frac{T_i}{T_i} \right]$

Maximum useful work

$$W_{u, max} = Q_{R} = m c_{p} T_{1} \left[\frac{T_{3}}{T_{1}} - (\delta_{p})^{r} \right]$$

Broyton nuclear plant thermodynamic efficiency:
$$\frac{1}{\sqrt{1-1/4}} \left[\frac{1}{\sqrt{1-1/4}} - \frac{1}{\sqrt{1-1/4}} \right] \left[1 - \frac{1}{\sqrt{1-1/4}} \right] \left[\frac{1}{\sqrt{1-1/4}} - \frac{1}{\sqrt{1-1/4}} \right] \right]$$

$$\xi = 1 - \frac{1}{(\gamma_p)^{\gamma-1/\gamma_1}}$$

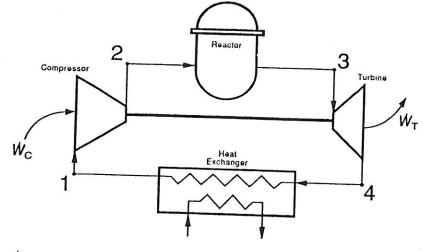
Optimum pressure ratio for meximum net work:

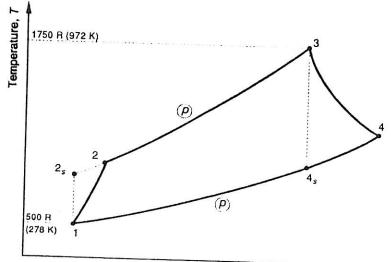
$$\frac{d[\dot{w}_{T} - \dot{w}_{CP}]}{dr_{p}} = 0 : (r_{p})_{aphinum} = (\overline{T_{1}})$$

Brayton Cycle with Real Components

- 1. Helium working fluid (assume perfect gas)

 Cp = 5230 Jlkg°k, 7 = 1.658,
- 2 $T_p = 4$. 3 $T_3 = 972^{\circ} K$, $T_1 = 278^{\circ} K$.
- 4. Isentapic efficiencies of turbine & compressor 90%





Entropy, s

$$\eta_{T} = \frac{\text{Actual work out of turbine}}{\text{ideal turbine work}} = \frac{\dot{w}_{T}}{\dot{w}_{Tc}} = \frac{\dot{m} c_{p}(T_{3} - T_{4})}{\dot{r}_{0}c_{p}(T_{3} - T_{4})}$$

$$\frac{1}{100} = \eta_{T} \dot{w}_{T_{1}} = \eta_{T} \dot{m} c_{p} T_{3} \left(1 - \frac{T_{4}s}{T_{3}}\right) = \eta_{T} \dot{m} c_{p} T_{3} \left(1 - \frac{1}{(7 - 1/4)}\right) = \frac{1.935 \, \dot{m}}{(7 - 1/4)} = \frac{1.935 \, \dot{m}}{(7 - 1/4)} = \frac{1.935 \, \dot{m}}{(7 - 1/4)}$$

$$\dot{W}_{CP} = \frac{W_{CPi}}{\eta_{CP}} = \frac{\dot{\eta}}{\eta_{CP}} C_{P} T_{I} \left(\frac{T_{2S}}{T_{I}} - I \right) = \frac{\dot{\eta}_{CP}}{\eta_{CP}} T_{I} \left(\frac{\eta_{CP}}{\eta_{CP}} - I \right)$$

$$= 1.184 \dot{\eta}_{M} T_{I}/S.$$

$$\dot{W}_{NET} = \dot{W}_{T} - \dot{W}_{CP} = \dot{m} (1.935 - 1.184) = 0.752 \dot{m} M W$$

$$\dot{Q}_{R} = \dot{m}_{CP} (T_{3} - T_{2})$$

$$\dot{W}_{CPi} = \dot{m}_{CP} [T_{25} - T_{1}] = \dot{m}_{CP} T_{1} [\gamma_{p}^{Y} - 1] = 1.066 \dot{m} M J/s.$$

$$T_{2} - T_{1} = \frac{\dot{W}_{CP}}{\dot{m}_{CP}} = \frac{\dot{W}_{QP}}{\dot{m}_{CP}} = 226.5 \dot{m} K$$

$$\dot{T}_{2} = 226.5 + 278 = 504.5 \dot{m} K$$

$$\dot{Q}_{R} = \dot{m}_{CP} (T_{3} - T_{2}) = 2.45 \dot{m} M W.$$

$$\dot{\eta}_{H} = \dot{M}_{NET} / \dot{Q}_{R} = \frac{0.752 \dot{m}}{2.45 \dot{m}} = 0.307$$

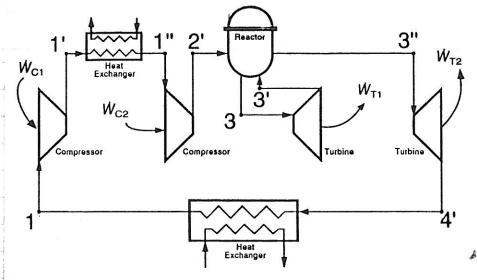
$$= 30.7 / 6$$

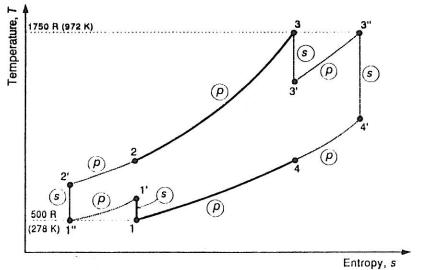
Brayton Cycle with Reheat and Intercooling

1. Fluid - He perfect gas,
$$C_p = 5230 \text{ J/kg°k}$$
, $\gamma = 1.658$
2. pressure rapio = 4, Intercooling: $P_1/P_1 = P_2/P_1 = 3p'$; $T_1'' = T_1$
Reheat: $\frac{P_3'}{P_4} = \frac{P_3'}{P_3'} = \frac{rp'}{P_3'}$ $T_3'' = T_3$.

3.
$$T_3 = 972^{\circ}K$$
, $T_1 = 278^{\circ}K$

4. Isombropic efficiencial of trustine de compresser 90%
 $W_{Cp} = \sin C_p (T_1 - T_1) + m c_p (T_2 - T_1'')$
 $= m c_p T_1 [(r_p)^{\frac{1}{2}} - 1] + m c_p T_1'' [(\tilde{r}_p') - 1]$
 $= 0.920. \text{ m MW}.$
 $W_1 = m c_p (T_3 - T_3') + m c_p (T_3 - T_4') = m c_p T_3 [1 - \frac{1}{124}]^{-1}$
 $+ m c_p T_3'' [1 - \frac{1}{124}]^{-1}$





$$\frac{T_{3}''}{T_{3}} = \frac{1}{3^{1/4}} \quad T_{3}' = 738.3^{\circ} K$$

$$T_{2}' = T_{1}'' (r_{p}) \quad and \quad T_{1}'' = T_{1} = 278^{\circ} K$$

$$T_{2}' = 365.8^{\circ} K$$

$$\frac{1}{2} = 365.8^{\circ} K$$