



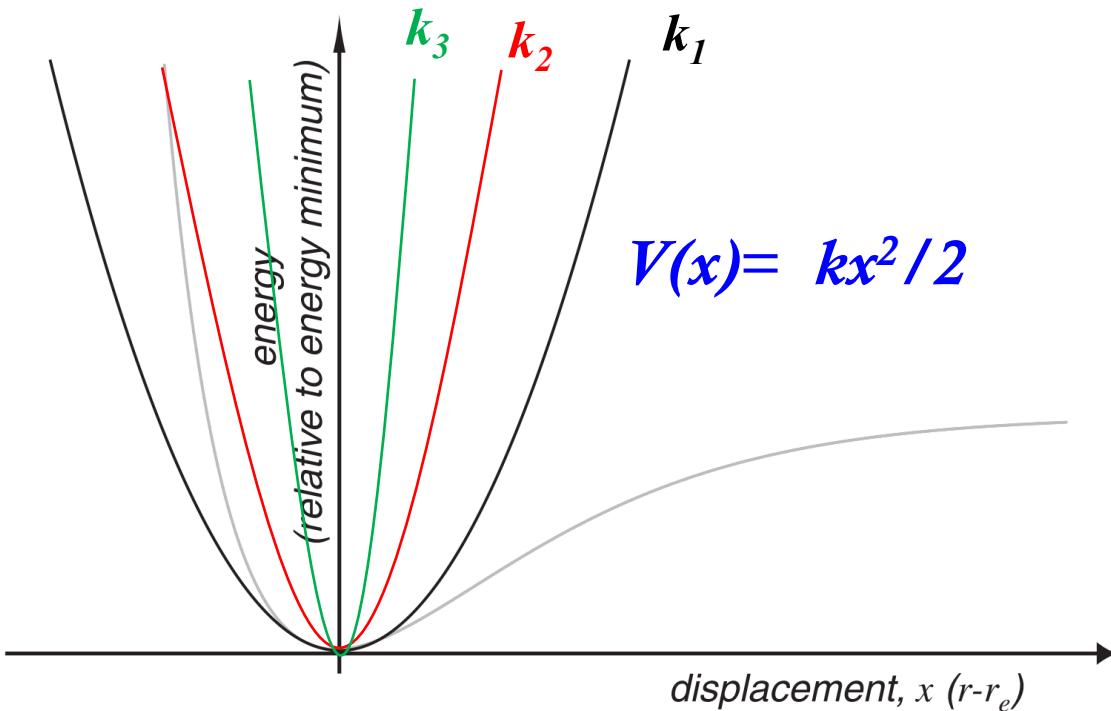
Molecular spectroscopy

Answers to the Questions 1-5



- Sketch on the same graph the potential energy functions for the harmonic oscillator with three different values of k . How do you interpret the meaning of k ?

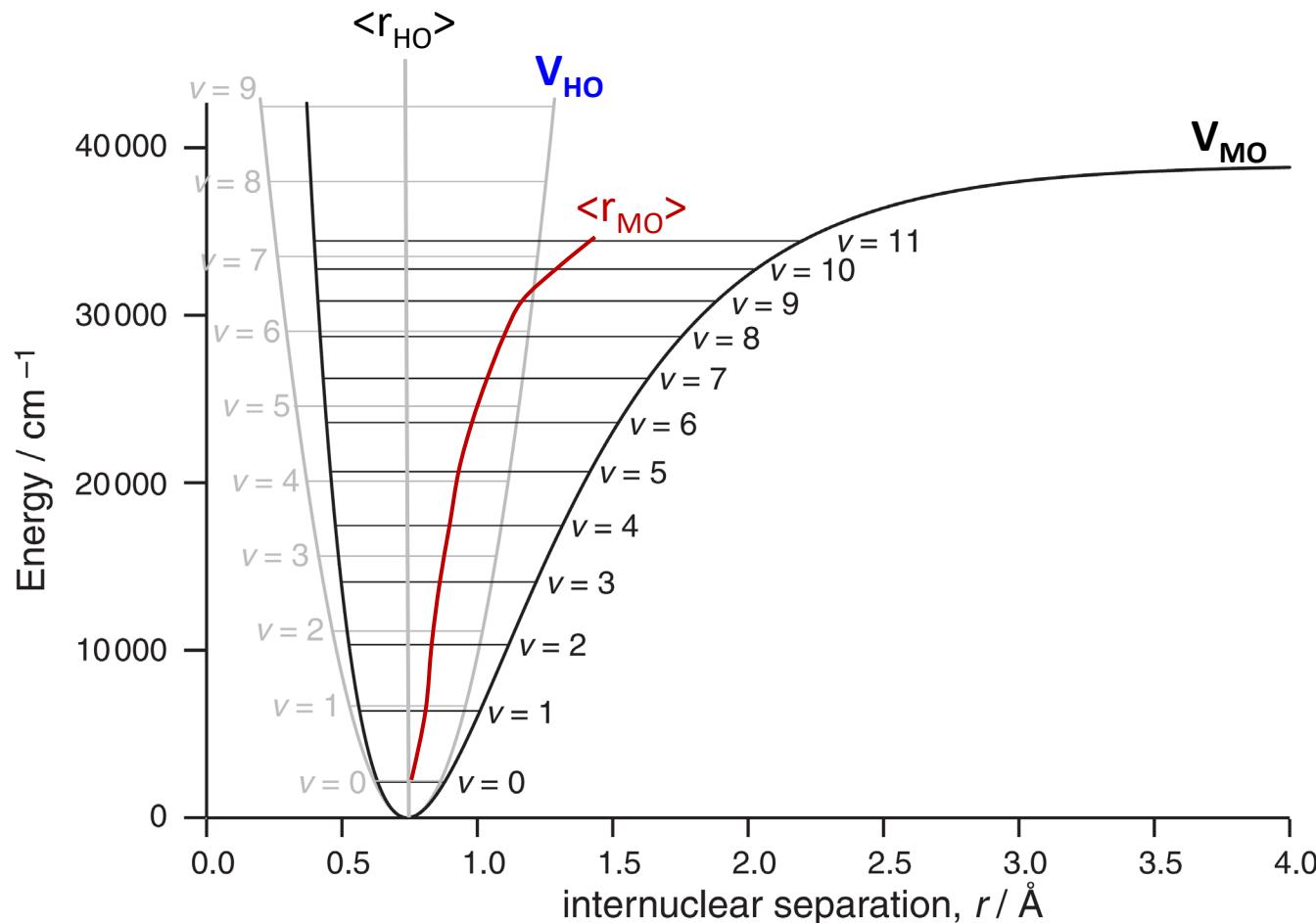
A: $k_1 < k_2 < k_3$; k 越大，越难以拉伸和压缩，势阱越狭窄！





2. Describe qualitatively how the average bond length changes with increasing vibrational energy for a vibrating diatomic using (a) the harmonic oscillator model and (b) the Morse model.

A: 谐振子每个量子态的平均键长均处于 r_e ；摩尔斯振子的平均键长随着能级的升高而增大





3. The far infrared spectrum of $^{39}\text{K}^{35}\text{Cl}$ has an intense line at 278.0 cm^{-1} . Calculate the force constant for KCl.

A: 分子的约化质量 $\mu = m_{\text{K}}m_{\text{Cl}}(m_{\text{K}}+m_{\text{Cl}})^{-1}$
 $= 39 \times 35 / (39+35) \text{ au} = 18.446 \text{ au} = 3.063 \times 10^{-26} \text{ kg}$

谐振子模型下， $v=0 \rightarrow v=1$ 的跃迁最强，吸收光频率即为分子的特征波长，
即有： $\Delta\varepsilon_{1-0} = \tilde{\omega} = (2\pi\tilde{c})^{-1}\sqrt{k/\mu}$

$$k = \mu(2\pi\tilde{c}\Delta\varepsilon_{0-1})^2 = 83.99 \text{ N m}^{-1}$$



4. The force constant of $^{79}\text{Br}_2$ is 240 N m^{-1} . Calculate the fundamental vibrational frequency and the zero-point energy of $^{79}\text{Br}_2$.

A: 分子的约化质量 $\mu = m_{\text{Br}}/2 = 39.5 \text{ au} = 39.5(\tilde{N}_0)^{-1} \times 10^{-3} \text{ kg}$

$$\tilde{\omega} = (2\pi\tilde{c})^{-1}\sqrt{k/\mu} = 321 \text{ cm}^{-1}$$

由谐振子近似由: $ZPE = \tilde{\omega}/2 = 160.5 \text{ cm}^{-1}$



5. From the data in the spectrum of carbon monoxide shown on page 12, determine ω and x_e and hence k_M , D_e , D_0 and β , and explain what each of these terms means.

A: 已知 $\Delta\varepsilon_{1-0} = 2143.26 \text{ cm}^{-1} = \tilde{\omega} - 2\tilde{\omega}x_e$ $\Delta\varepsilon_{2-0} = 4260.04 \text{ cm}^{-1} = 2\tilde{\omega} - 6\tilde{\omega}x_e$

$$\tilde{\omega} = 3\Delta\varepsilon_{1-0} - \Delta\varepsilon_{2-0} = 2169.74 \text{ cm}^{-1}$$

$$2\tilde{\omega}x_e = 2\varepsilon_{1-0} - \varepsilon_{2-0} = 26.48 \text{ cm}^{-1} \quad x_e = 2\tilde{\omega}x_e/(2\tilde{\omega}) = 0.006102$$

$$\tilde{\omega} = (2\pi\tilde{c})^{-1}\sqrt{k_M/\mu} \quad k_M = \mu(2\pi\tilde{c}\tilde{\omega})^2 = 1902.0 \text{ N/m}$$

$$\tilde{D}_e (\text{in cm}^{-1}) = \tilde{\omega}(4x_e)^{-1} = 88924 \text{ cm}^{-1}$$

$$\tilde{D}_0 (\text{in cm}^{-1}) = \tilde{D}_e - \varepsilon_0 = \tilde{D}_e - (\tilde{\omega}/2 - \tilde{\omega}x_e/4) = 87842 \text{ cm}^{-1}$$

$$\beta = \sqrt{2\mu x_e \omega / \hbar} = \sqrt{2\tilde{c}\mu x_e \tilde{\omega} / \hbar} = 2.3207 \times 10^{-10} \text{ m}^{-1} = 2.32 \text{ \AA}^{-1}$$