

## REPORT ON COMPUTATIONAL ASSIGNMENT

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Aim- Plotting temporal variation of methane pyrolysis by initial decomposition pathways by MATLAB. And also using degree of rate control method finding the rate determining step (RDC).

Theory- Micro-Kinetic Model is a method that helps to develop a feasible reaction scheme. By this method we get a kinetic parameter of all steps.

Degree of rate control (DRC):

$$X_i = (k_i/r) * (\partial r / \partial k)_{k_j \neq i}$$

$$x_i^+ = \frac{k_i}{r} (\partial r / \partial k_i)_{k_j \neq i, k_{-j}}$$

$$x_i^- = k_{-i}/r \left( \frac{\partial r}{\partial k_{-i}} \right)_{k_j \neq i, k_j}$$

$$X_i = X_i^+ + X_i^-$$

Where,

$k_i$  = forward reaction rate constant

$k_{-i}$  = backward reaction rate constant

$r$  = overall rate of reaction (mols/m<sup>3</sup> sec)

$x$  = degree of rate control

Result:

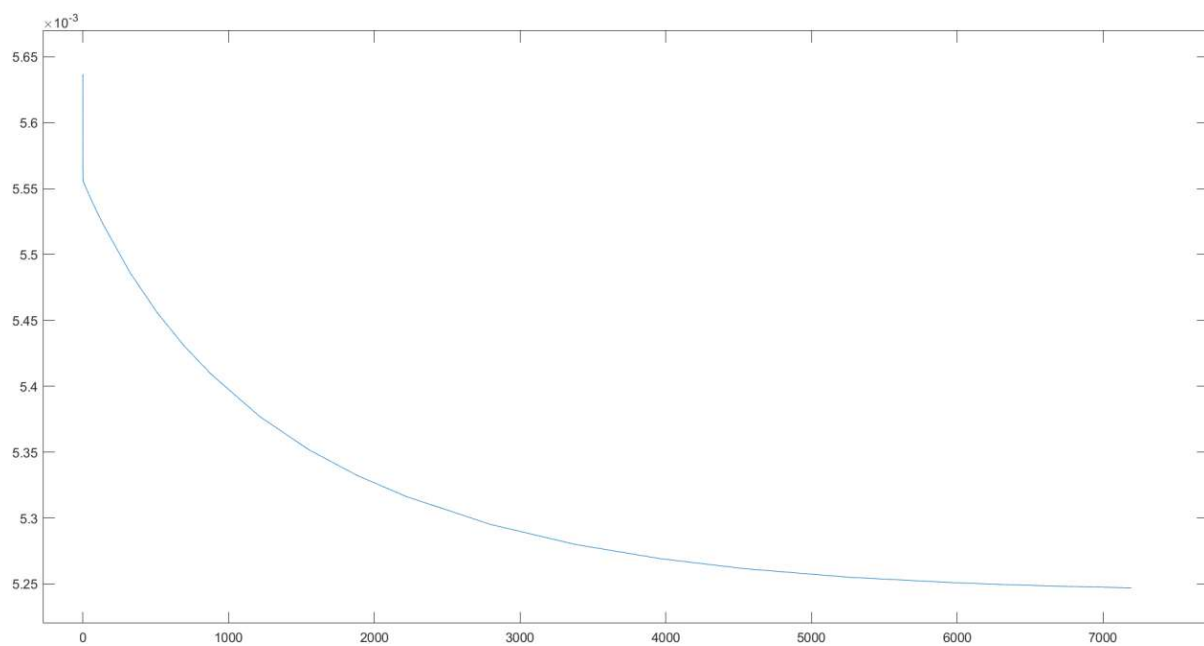
Initially at higher value of concentration there is negligible variation in time. But as time increases the moles decreases asymptotically and then remains constant.

Sensitivity analysis:

Sensitivity analysis is done to simplify the reaction network using degree of rate control

$X_i = (k_i/r) * (\partial r / \partial k_i) |_{k_j \neq i}$  (Where  $X_i$  is the degree of rate control);

If  $X_i = 0$  then the reaction is insensitive and overall rate will not depend on that  $i^{\text{th}}$  reaction.



TIME (IN SEC.)

### RDS BY SENSITIVITY ANALYSIS

