Jharkhand University of Technology, Ranchi

Diploma 1st Semester Examination, 2024 (NEP-2024)

Subject : Engineering Chemistry

Subject Code: BSC103

Time Allowed: 3 Hours

Full Marks: 70

Pass Marks: 21

Answer in your own words.

Answer five questions in which Question No. 1 is compulsory.

The figures in the margin indicate full marks.

All questions carry equal marks.

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1.		e the correct option:		2×7=14	
	000000	The electronic configuration of Cu ²⁺ ion is		2 10 1	
	ليا	(a) [Ar] 4s ¹ 3d ⁸	(b)	$[Ar] 4s^2 3d^{10} 4p^1$	
		(c) $[Ar] 4s^1 3d^{10}$	(d)	[Ar] $4s^2 3d^{10} 4p^1$ [Ar] $3d^9$	
(ii) Bohr proposed that while revolving in discrete orbits, the electrons				ts, the electrons	
	((a) gain energy	(b)	lose energy	
	((c) do not radiate energy	(d)	first lose energy and then gain energy	
	(iii) V	ii) What is the most difficult atom to ionize?			
	((a) Hydrogen	(b)	Helium	
	(c) Beryllium	(d)	Neon	
(iv) Which of the following is a synthetic polymer				monly used in making plastic?	
	(;	a) Cellulose	(b)	Starch	
	(c) Protein	(d)	Polyethylene	
	(v) W	Which of the following is not a type of electrochemical cell?			
	(a) Voltaic cell	(b)	Photovoltaic cell	
	(c) Electrolytic cell	(d)	Fuel Cell	
(vi) Which industry causes all three: air, water and land pollution?			pollution?		
	(a) Fertilizer and pesticides	(b)	Oil refineries and iron, steel	
	. (c) Oil refineries and caustic soda	(d)	Iron, steel and caustic soda	

(c) Gun Metal

(d) Thermal Insulators

(e) BOD and COD