

## Week 2 Report

### The Effect of Dipping Duration on Perovskite Formation:

PbI<sub>2</sub> was coated onto cleaned glass in air (no glove box), then dipping times in CH<sub>3</sub>NH<sub>3</sub>I 2-Propanol solution was varied according to the following table:

Sample Number	Temp of Glass (°C)	Dipping Duration (Sec)
F07	Room ~22	10
F08	Room ~22	20
F09	Room ~22	30
F10	65	10
F11	65	20
F12	65	30

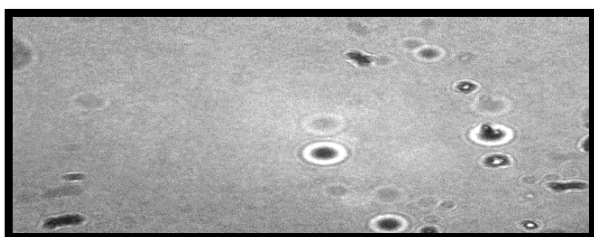
Glass temperature prior to dipping was also varied.

### Observations:

F10-F12 were more opaque than F07-F09, in general. Longer dipping times led to darker sample, more perovskite. But this may be due to the thick PbI<sub>2</sub> layer.

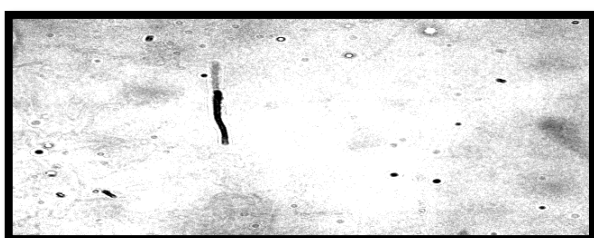
The samples were studied under a microscope and revealed the following:

F07: (10 Sec)



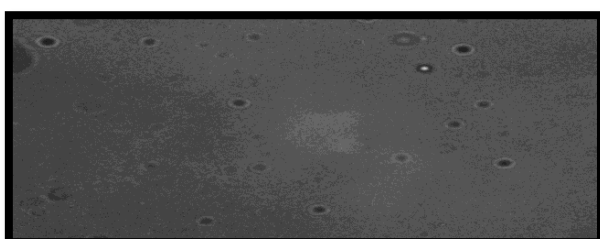
- Good, even coat
- Many black spots
- Red

F08: (20 Sec)



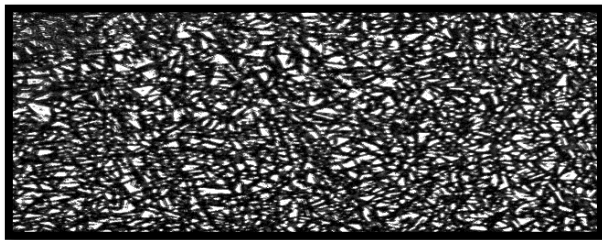
- Black spots
- Signs of cracking
- Very red

F09: (30 Sec)



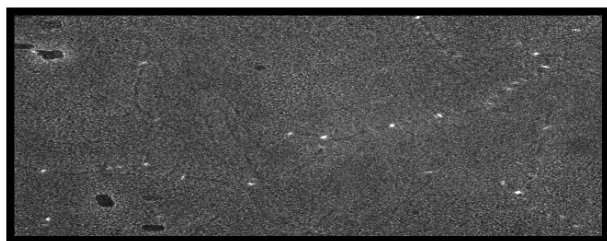
- Black spots
- No crack (maybe I was unable to see them)
- Very red

F10: (10 Sec)



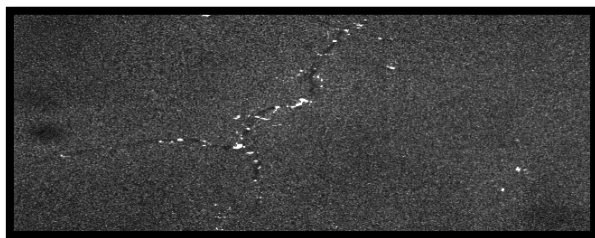
- $\frac{1}{2}$  Perovskite to hole ratio
- White (No red)

F11: (20 Sec)



- Few black spots
- Cracks

F12: (30 Sec)



- Few black spots
- Few holes
- Lots of cracks

#### Conclusion:

Longer dipping times yields darker samples, meaning higher perovskite density. The probability of cracks increases with time, and is proportional to the glass' temperature prior to dipping.

For the future we will stick with room temperature glass prior to dipping.

#### **Air Drying After 2-Propanol Rinse in Glove Box:**

Dipping was attempted with the same methods previously used, but in a glove box. Unlike in air, letting the sample dry in the glove box did not result in any perovskite dissolving. The use of an air jet is no longer needed when dipping is done in a humidity free environment.

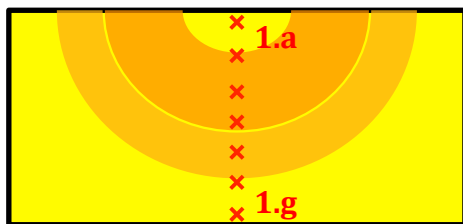
#### **O<sub>2</sub> vs. O<sub>3</sub> Priming, & Heated vs. Unheated Glass Prior to Spin Coating:**

Now we take a closer look at spin coating and its preceding steps to arrive at a thinner and more uniform initial coat of PbI<sub>2</sub>. The sample were treated according to the following table:

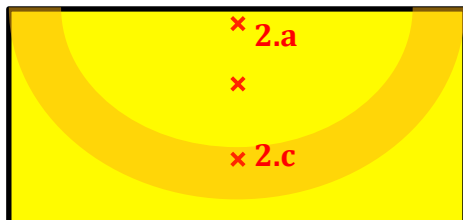
Sample	Primer	Glass Temp. (°C)
F13	O <sub>2</sub>	70
F14	O <sub>2</sub>	70
F15	O <sub>2</sub>	70
F16	O <sub>2</sub>	Room ~20
F17	O <sub>2</sub>	Room ~20
F18	O <sub>2</sub>	Room ~20
F19	O <sub>3</sub>	70
F20	O <sub>3</sub>	70
F21	O <sub>3</sub>	70
F22	O <sub>3</sub>	Room ~20
F23	O <sub>3</sub>	Room ~20
F24	O <sub>3</sub>	Room ~20

#### Observation 1 (Heated vs. Unheated Glass):

Heated samples exhibited a multi color ring pattern, while unheated sample were uniformly colored. F19 & F23 were compared using a SEM, by taking pictures across the ‘rings.’ The following figures show where these pictures were taken:



F19: Images were taken starting from point 1.a, with 2mm spacing, and ending at point 1.g.



F23: Images were taken starting from point 2.a, with 5mm spacing, and ending at point 2.c.

The images revealed that the surface becomes more jagged as we move away from the center. Point 1.a is more even than 1.g, but exhibits more holes (Fig 1.a,f). Points 2.a, 2.b, 2.c were similar to each other, more jagged than 1.a, but more even than 1.g (Fig 2.a,c).

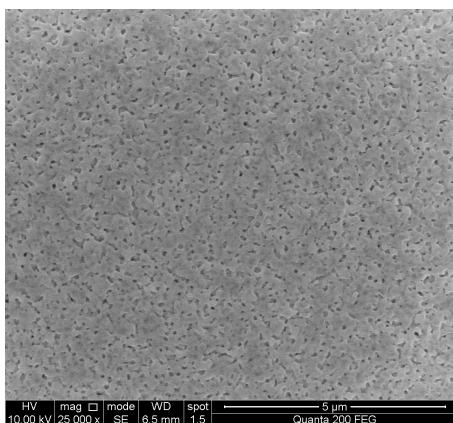


Fig. 1.a

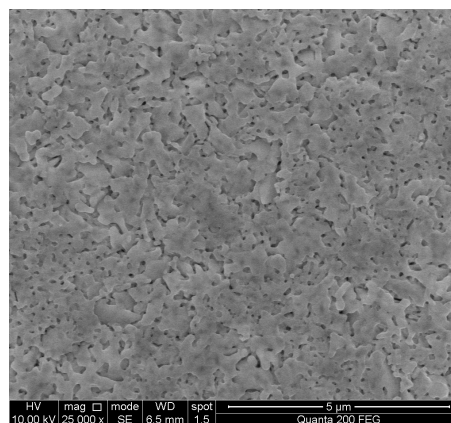


Fig. 1.f

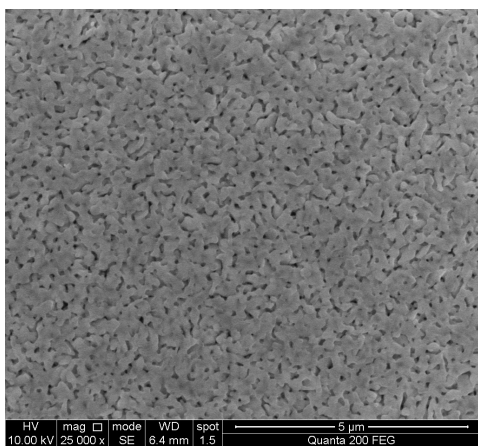


Fig. 2.a

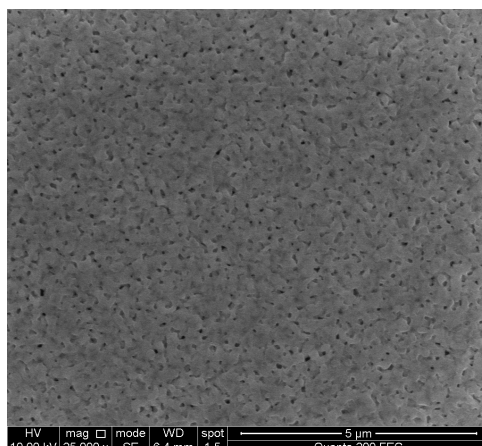


Fig. 2.c

### Conclusion 1:

It seems that the best results are attained at the center of the heated glass, but become much worse than unheated glass at the edges. We must either use unheated glass or find a variation to the procedure remedies the jagged edges in heated glass. Makhsud proposed applying DMF to the heated glass prior to adding the  $\text{PbI}_2$  solution during spin coating. Trying this resulted in no apparent rings, but no SEM images were taken yet. This variation to the procedure will be studied further.

### Observation 2 ( $\text{CH}_3\text{NH}_3\text{I}$ Layer After Rinsing in 2-Propanol):

The rest of the samples aside from F19 & F23 were dipped in  $\text{CH}_3\text{NH}_3\text{I}$  2-Propanol solution. While drying a white layer seemed to appear on the last surfaces to dry. It is speculated that this layer is  $\text{CH}_3\text{NH}_3\text{I}$ . This is troublesome since it will act as an insulator and dramatically decrease the efficiency of the solar cell.

### Conclusion 2:

A method to remove access  $\text{CH}_3\text{NH}_3\text{I}$  that appears after dipping must be found. This will be studied further, and several methods such as double dipping, decreased solution concentration, and others will be tested.

### Observation 3 ( $\text{O}_2$ vs. $\text{O}_3$ as a Primer):

Room temperature glass prior to spin coating was revealed to be more prone to micro-cracks than heated glass, when studied under a microscope.  $\text{O}_3$  treated glass also seemed to be more prone to micro-cracks than  $\text{O}_3$  glass (Fig.3.a,b).

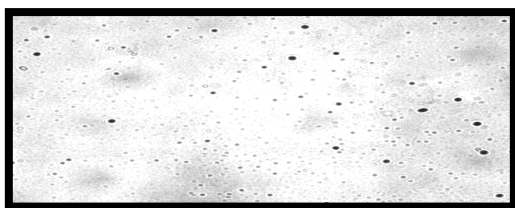


Fig. 3.a Sample F16

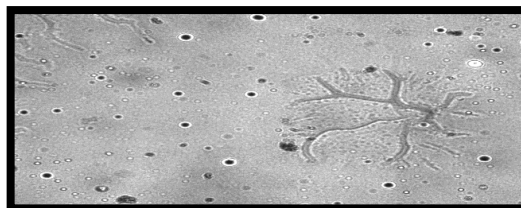


Fig. 3.a Sample F24

Conclusion 3:

O<sub>2</sub> should be used, and glass should be heated prior to spin coating to avoid cracks.

**Next Week:**

- Longer dipping times and variations in solution temperature will be more carefully tested with properly spin coated samples (spin coat in glove box).
- Applying DMF prior to the PbI<sub>2</sub> solution during spin coating, on heated glass, will be tested further.
- A method must be formulated and tested to remove the insulating CH<sub>3</sub>NH<sub>3</sub>I layer that results after dipping.