

$R_1/R_1' \Rightarrow O_3/m/H_2O$  or  $O_3/PPh_3$ . Page - 2

$R_2/R_2' \Rightarrow \text{Hg}^{2+}/\text{dil. H}_2\text{SO}_4$  or  $\text{HgSO}_4/\text{H}_3\text{O}^+$ .

$R_3' / R_3'' \Rightarrow$  PCC;  $\text{CrO}_3$  in acetone

$R_4 \neq R_4' \Rightarrow Ca(OH)_2 / \Delta$  or  $Ca(OH)_2$  dry distillation.

$R_5 \Rightarrow 1t_2 / \text{Pd} / \text{BaSO}_4$  Rosenmund reduction.

or DIBALH.

$R_6' \Rightarrow (CtB_3)_2 Cd$  or  $(CtB_3)_2 Cu di.$

R<sub>1</sub>  $\Rightarrow$  DIBALH or SnCl<sub>2</sub> + HgCl<sub>2</sub>(aq)

$\alpha_6 \Rightarrow \text{DIS}$

$R_1 \rightarrow Cl_3 MSX$  followed by  $1h_3 O^+$ .

$R_6 \Rightarrow C_6 H_5 Cl$   
 $R_7 \Rightarrow C_6 H_5 Br$

$R_7 \Rightarrow DIBAH$

$$R_s' \Rightarrow \text{Eig}-\text{CH}_2-\text{CH}_3. (2^{\circ} \text{ Alcohol})$$

$R_S \Rightarrow C_{13}CH_2OH$  (rAldoh)

$$R_9' \Rightarrow Cl_3 Cl_2 - Cl_3.$$

$$Rg \Rightarrow Cb_3 - Ch_3$$

$$\text{R} \rightarrow \text{CH} - \text{CH} = \text{CH} - \text{CH} = 0 \quad (\text{Cootonanoldenone})$$

$$R_{10} \Rightarrow C_3H_7 - CH = CH - CH_2 - CH_3 = 0 \quad (\text{Aldol product})$$

at OH (50)       $C_3H_7 - CH - CH_2 - CH_3$   
- 100% isomers

$$R_{10}' \Rightarrow Cl_3-C\overset{CH}{\underset{Cl_3}{|}}-C=CH_2 + Cl_3-C\overset{CH}{\underset{OH}{|}}-C=CH_2$$

at 0.14% S.C.

(Mesityl oxide) (Alcohol)  $\Rightarrow$

$R_{10}' \Rightarrow (CH_2CO_2H + CO_2 + H_2O)$

$$R_{11} \Rightarrow \text{CH}_3\text{CO}_2\text{H}$$

$R_{12} \text{ Cl}_3-\text{C}_2\text{H}_5-\text{O}-\text{CH}_3$

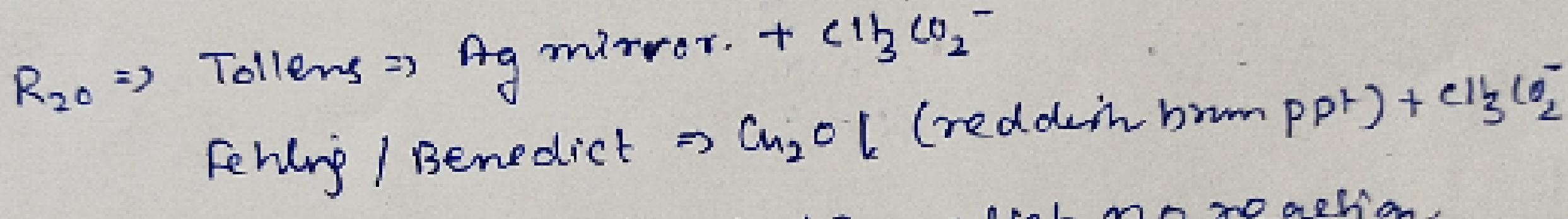
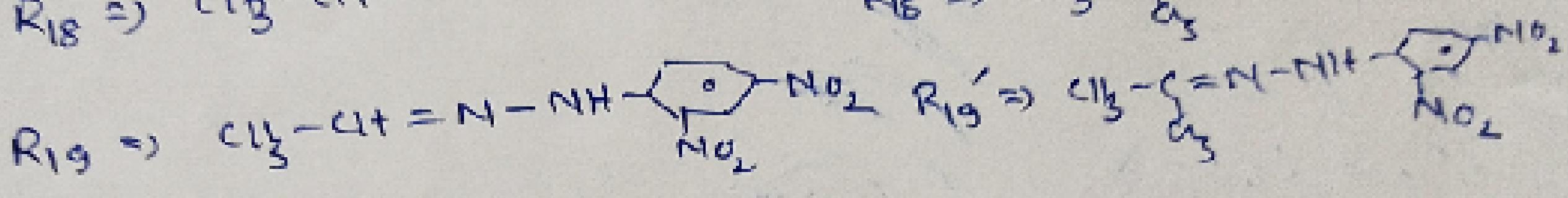
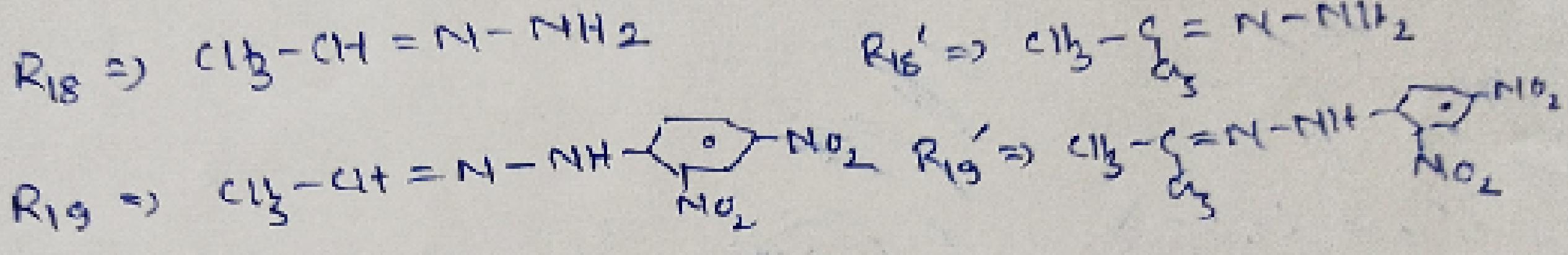
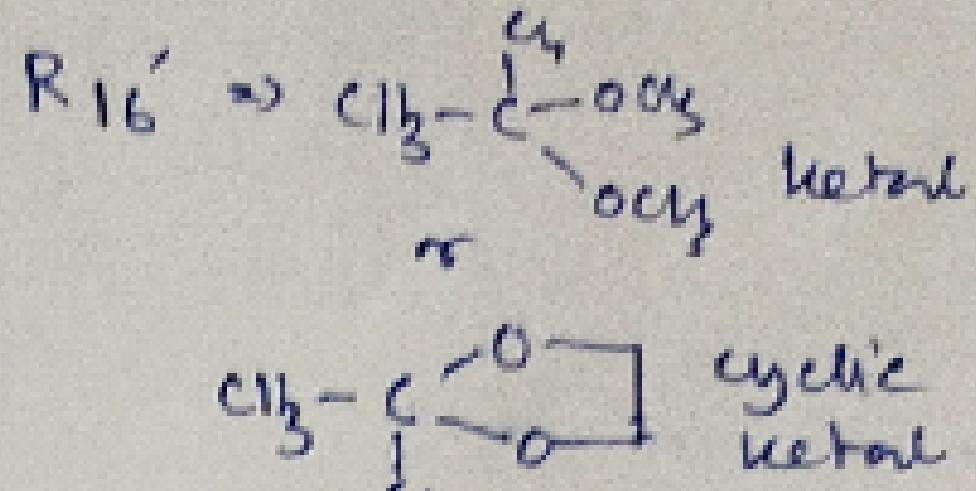
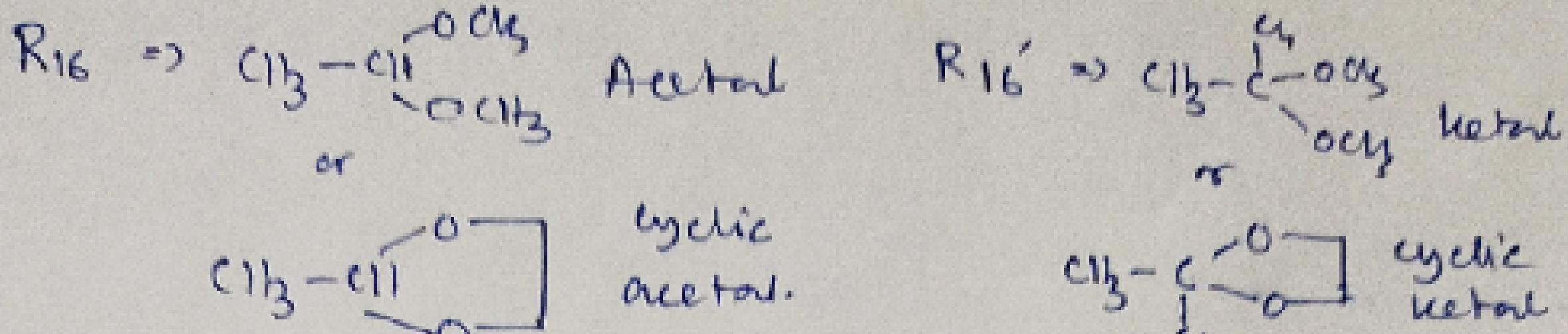
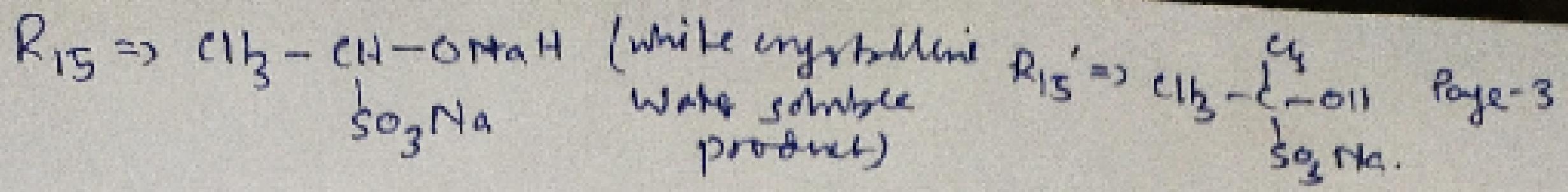
$R_{12} \Rightarrow Cl_3CO_2H$  (Bayer-vilsmeyer oxidation)

R<sub>13</sub>: CH-CH-OH (Iodofer)  
 R<sub>13</sub>: CH-CH<sub>2</sub> (Step column)

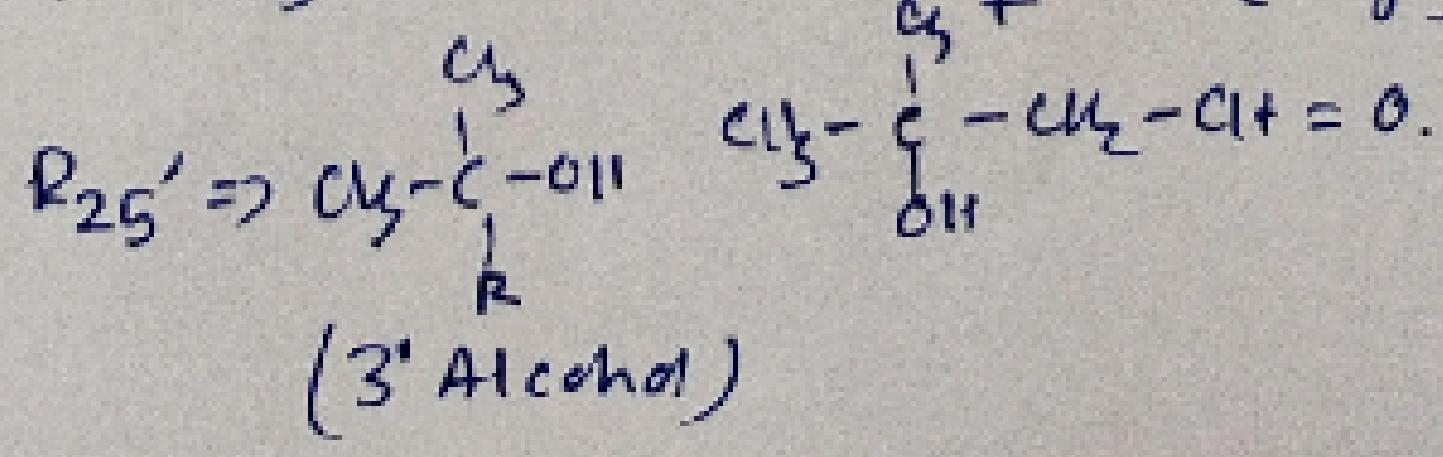
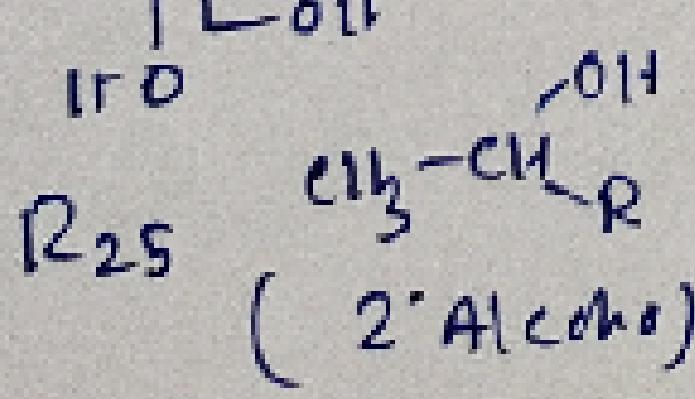
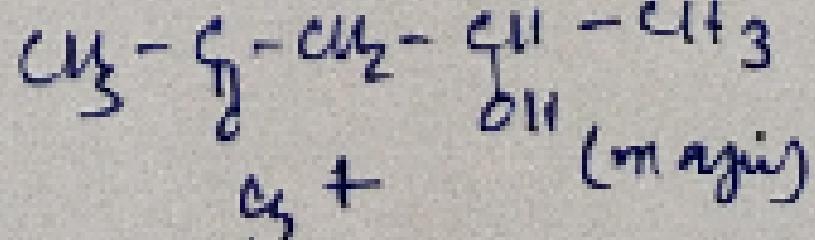
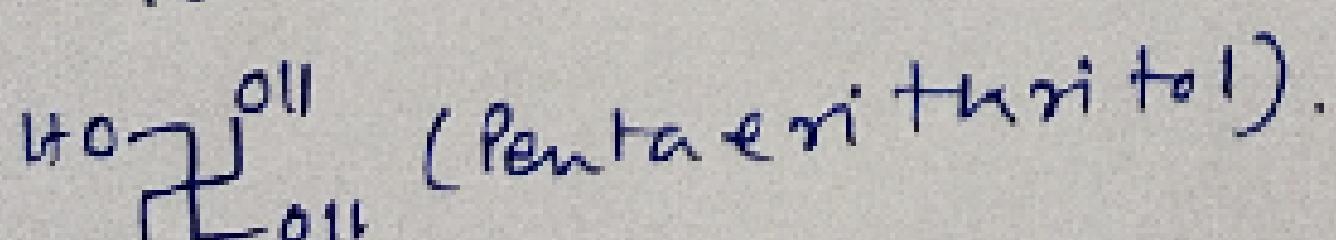
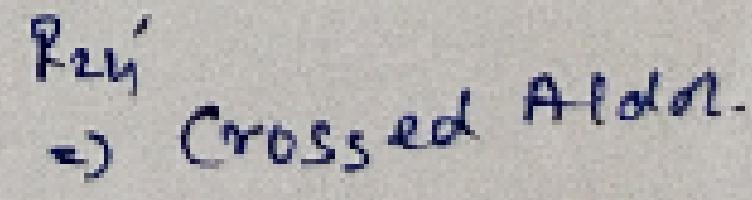
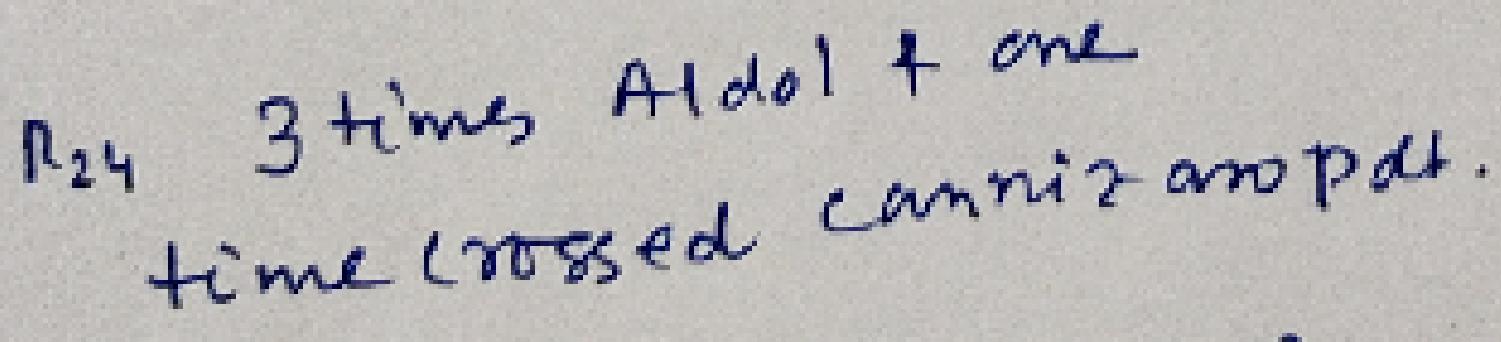
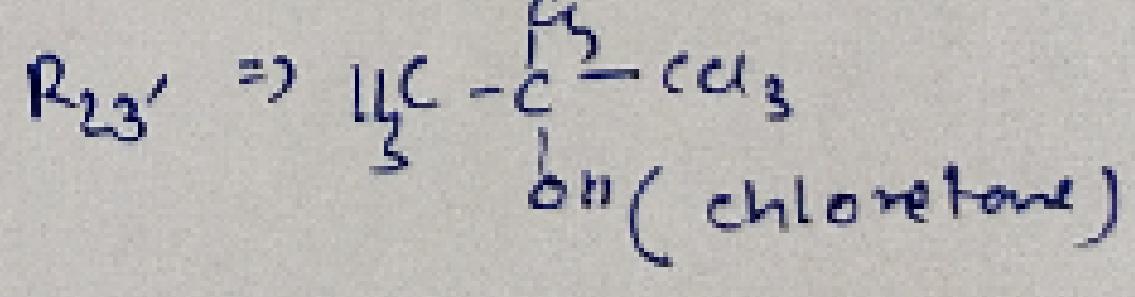
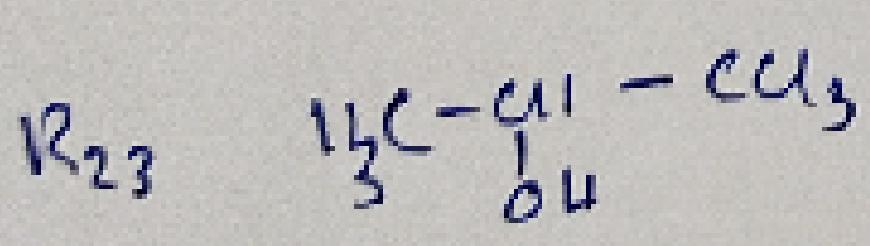
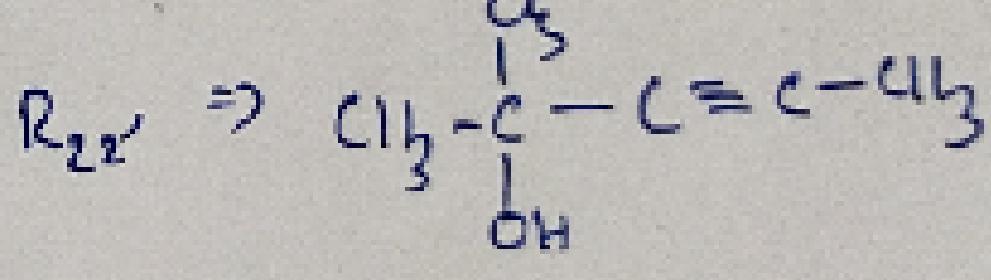
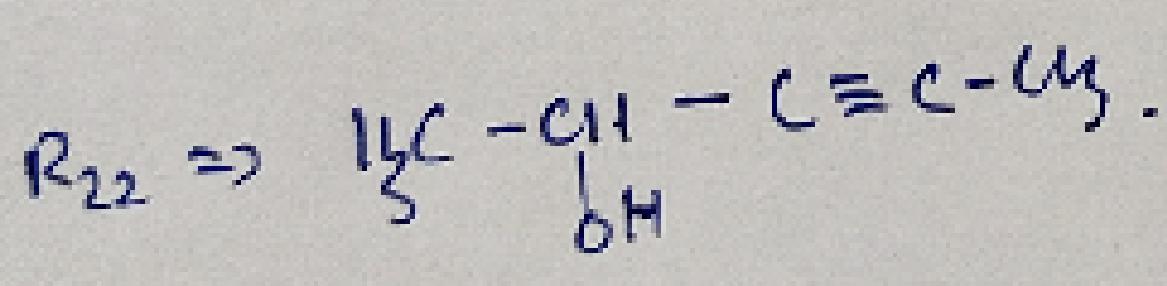
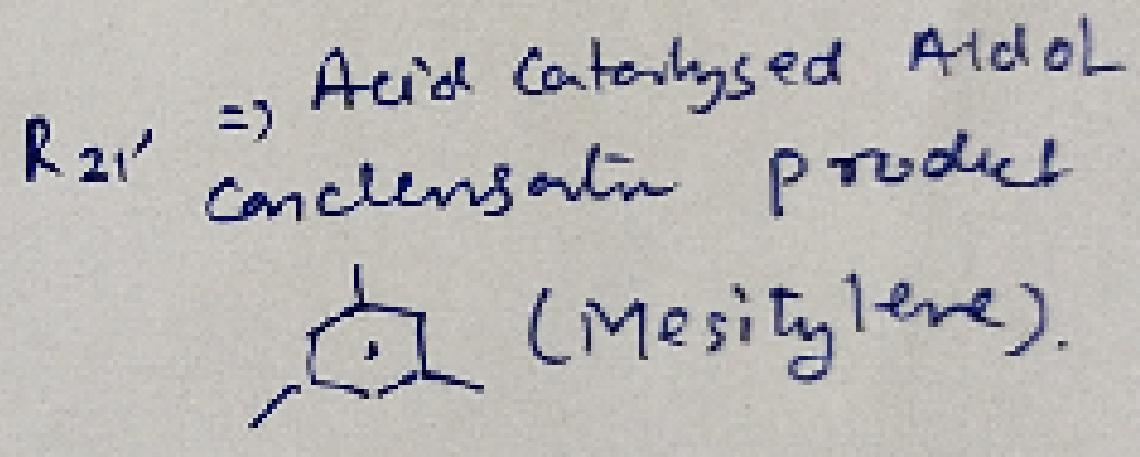
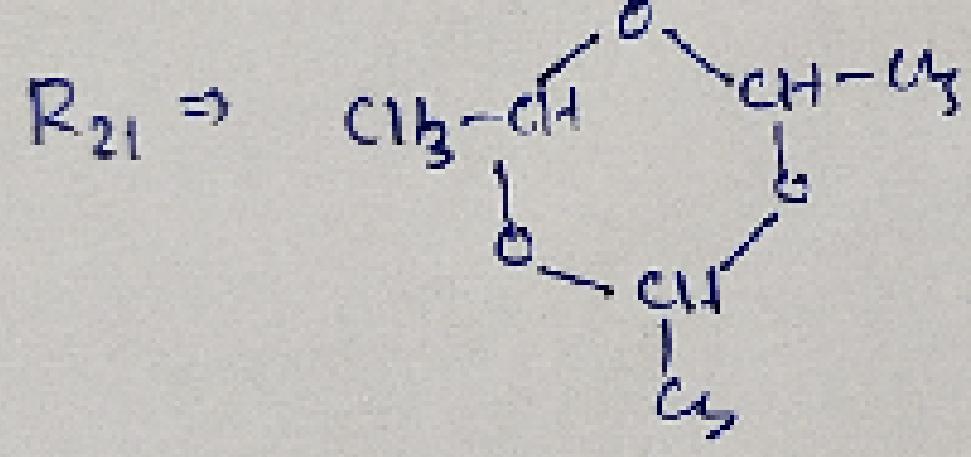
R<sub>13</sub> 2) HCO<sub>2</sub>H (Iodoform)  
(step down)

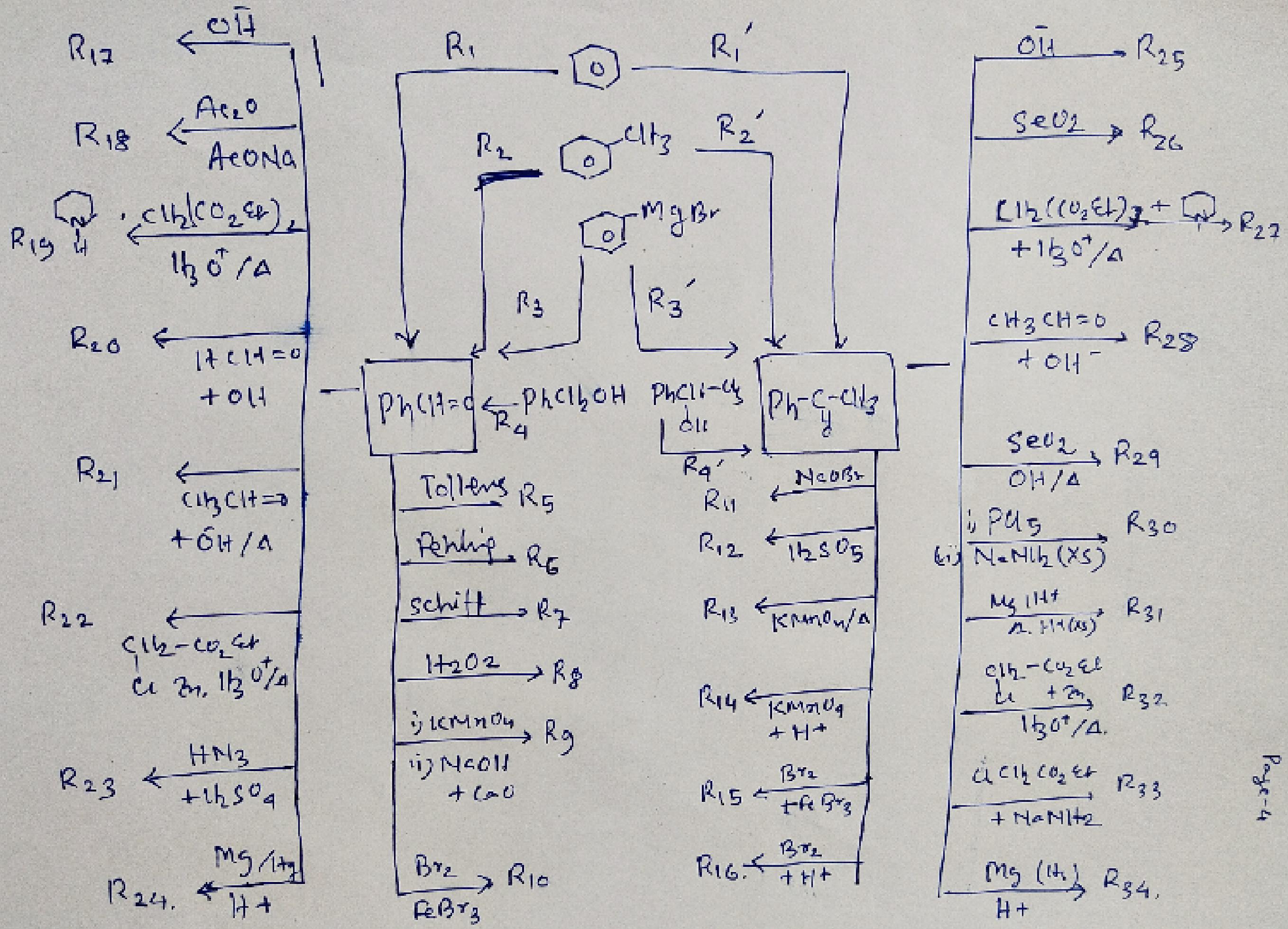
$$R_{14} \Rightarrow Cl_3 - \frac{C}{Cl_{11}} - CN$$

$R_{14} \Rightarrow \text{CH}_3-\text{CH}-\text{CN}$   
 $\text{BH} \subset \text{cyanohydrin} \Rightarrow$



for ketone, with Tollens / Fehling / Benedict no reaction.





$R_1 \Rightarrow CO + HCl + \text{anhydrous } AlCl_3$  (Gatterman Koch synthesis) Page 5  
or  
 $HCONH + HCl + \text{anhydrous } AlCl_3$  Gatterman Aldehyde Synthesis.

$R_1' \Rightarrow Cl_3-C\equiv C + AlCl_3$  Friedel-Craft Acylation  
or  
 $Ac_2O + AlCl_3$  (Acylation in ring)

$R_2 \Rightarrow CrO_2Cl_2Cl_2$  or  $Cr$ -acetate +  $H_2O$  (Etard reaction)

$R_2' \Rightarrow KMnO_4$ ;  $SOCl_2$ ;  $(CH_3)_2Cd$ .

$R_3 \Rightarrow HCN$  followed by  $H_3O^+$  addition.

$R_3' \Rightarrow Cl_3-C\equiv N$  followed by  $H_3O^+$  addition.

$R_4 \Rightarrow MnO_2$  (oxidises Benzylic  $1^\circ$  Alcohol to aldehyde)

$R_4' \Rightarrow MnO_2$  (oxidises benzylic  $2^\circ$  Alcohol to ketone)

$R_4$  or  $R_4'$  can be  $Cu/300^\circ C$ .

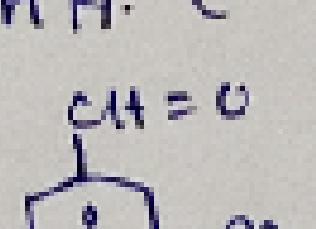
With Tollens  $R_5 \Rightarrow PhCO_2^- + Ag$  mirror.

With Fehling & Schiff, benzaldehyde no reaction.

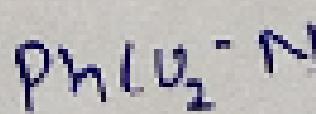
With Fehling & Schiff, benzaldehyde no reaction.

$R_6 \Rightarrow PhCO_2H$  (with  $H_2O_2$ , Baeyer-Villiger oxidation)

$R_7 \Rightarrow PhH$ . (Oxidation followed by decarbonylation).

$R_8 \Rightarrow$   [EAS; bromination].

$R_9 \Rightarrow$   [EAS; bromination].

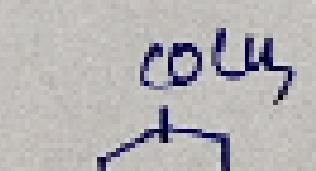
$R_{10} \Rightarrow$   [EAS; bromination].

$R_{11} \Rightarrow PhCO_2^+ Na^+$  (Bromoform reaction; step down).

$R_{12} \Rightarrow CF_3-C\equiv O-Ph$  [ $H_2SO_5$  also can participate in

Baeyer-Villiger Oxidation]

$R_{13}/R_{14} \Rightarrow PhCO_2K / PhCO_2H$ .

$R_{15} \Rightarrow$   [EAS; bromination in ring].

$R_{16} \Rightarrow Ph-C\equiv C-Br$  [Acid catalysed halogenation].

$R_{17} \Rightarrow PhCO_2^+ + PhCO_2OH$  [Cannizzaro reaction]

[disproportionation reaction]

$R_{18} \Rightarrow PhCH=CH-CO_2H$  (Cinnamic Acid) ; Perkin reaction.

$R_{19} \Rightarrow PhCH=CH-CO_2H$  Knoevenagel condensation. Paarls.

$R_{20} \Rightarrow HCO_2^- + PhCH_2OH$  crossed Cannizzaro reaction.

$R_{21} \Rightarrow PhCH=CH-CH=O$ . Claisen Schmidt reaction.

$R_{22} \Rightarrow PhCH=CH-CO_2H$  Reformatsky reaction

$R_{23} \Rightarrow Ph(\equiv N) + PhNHCHO$  Schmidt reaction

$R_{24} \Rightarrow \begin{matrix} PhCH & - & CO_2Ph \\ \text{on} & & \text{on} \end{matrix}$  [Pinacol formation by free radical mechanism].

$R_{25} \Rightarrow \begin{matrix} Ph-\overset{\text{C}}{\underset{\text{H}}{\overset{\text{O}}{\parallel}}} & = & C-Ph \\ \text{by} & & \text{by} \end{matrix}$  Aldeol condensation product (0°C) very  
 $\begin{matrix} Ph-\overset{\text{C}}{\underset{\text{H}}{\overset{\text{O}}{\parallel}}} & = & C-Ph \\ \text{by} & & \text{by} \end{matrix}$  Aldeol product (0°C) low temp).

$R_{26} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}}{\overset{\text{O}}{\parallel}}} = O$  [  $SeO_2 \Rightarrow$  Monocarbonyl  $\rightarrow$  Dicarbonyl ].

$R_{27} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH-CH_3$  Knoevenagel condensation.

$R_{28} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH-CH_3$  (crossed Aldeol condensation).

$R_{29} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH_2O \rightarrow \begin{matrix} Ph & - & CO_2^- \\ \text{on} & & \text{on} \end{matrix}$  [ Intra crossed Cannizzaro reaction ]

$R_{30} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH_2O \xrightarrow[\text{(x)}]{NaBH_4} Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} Na^+$ .

$R_{31} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH-CO_2H$  Reformatsky reaction.

$R_{32} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH-CO_2H \xrightarrow{\text{P}-\text{C}_6H_5} Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH-CO_2Et$ .

$R_{33} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH-CO_2Cu \rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH-CO_2Ph$ .

$R_{34} \Rightarrow Ph-\overset{\text{O}}{\underset{\text{H}_3}{\overset{\text{C}}{\parallel}}} = \overset{\text{O}}{\underset{\text{H}_3}{\overset{\text{C}}{\parallel}}} - CH_3$  (Pinacol)

$R_{31} \Rightarrow Ph-\overset{\text{C}}{\underset{\text{H}_3}{\overset{\text{O}}{\parallel}}} = CH-CO_2H$

(Pinacolone).  
Pinacol  $\rightarrow$  Pinacolone rearrangement.