



TO THE COURSE:

TECHNOLOGY OF TEXTILE PREPARATION AND FINISHING

(TXL-241)

About the course.....

3 credits (3-0-0); Pre-requisites: TXL110 OR TXL111 OR PHL110 OR MAL110 OR TXN101 OR TXL111L AND TXL111P

A. Preparatory Processing

Natural and added impurities in textiles, singeing, desizing, scouring, bleaching, mercerization and optical whitening of cotton. Combined preparatory processes, carbonization, scouring and bleaching of wool, degumming of silk. Heat setting. Machinery for preparation of textiles. Surfactants and their application.

B. Finishing

Introduction to chemical and mechanical finishes. Chemical finishes for hand modification. Bio-polishing, easy care, oil, water and soil repellent finishes. Fire retardancy, antimicrobial finishes. Finishes for wool. Mechanical finishes like shrink proofing and calendering; Raising, sueding and emerising. Low liquor application techniques and machinery; Stenters and dryers.

Instructor.....

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Introduction to Preparatory Processes

Reference Book

Books: Part A

- Chemical Technology in the pre-treatment process of textiles, S. R. Karmakar, 1999, Elsevier Science
 - Textile Scouring & Bleaching by E. R. Trotman, B.I. Publications, New Delhi
 - Handbook of Fibre Science and Technology- Volume I: Chemical properties of fibers and fabrics fundamentals and preparation Part-A and B. Ed. Mena Chem Lewin and Stephen B-Sello. Marcel Dekker Inc. New York
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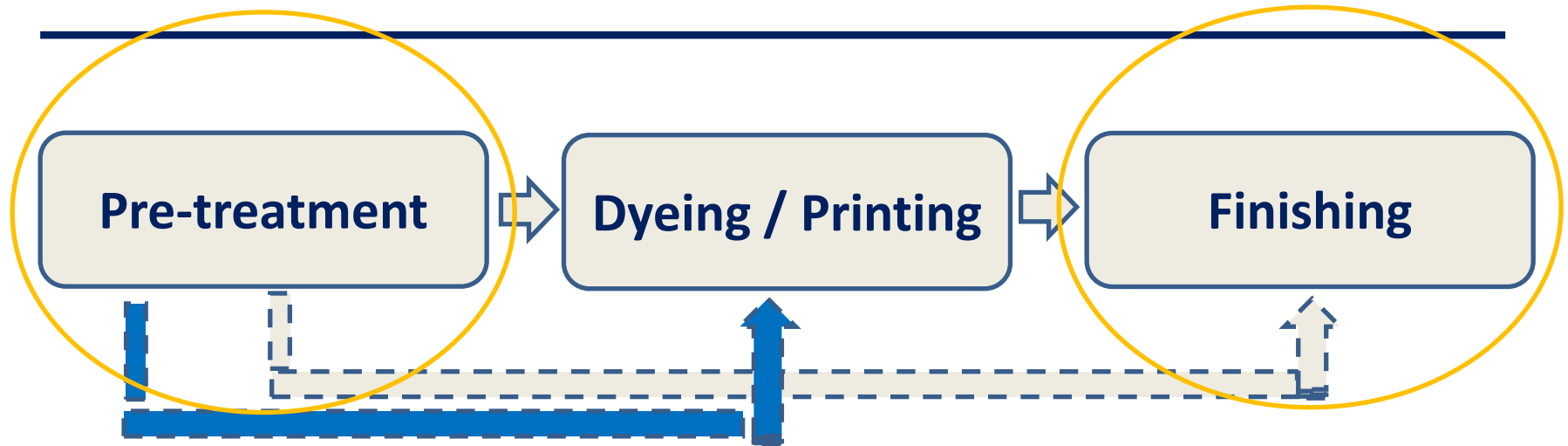
Major Textile Fibres

Natural:

- Cotton
- Wool
- Silk

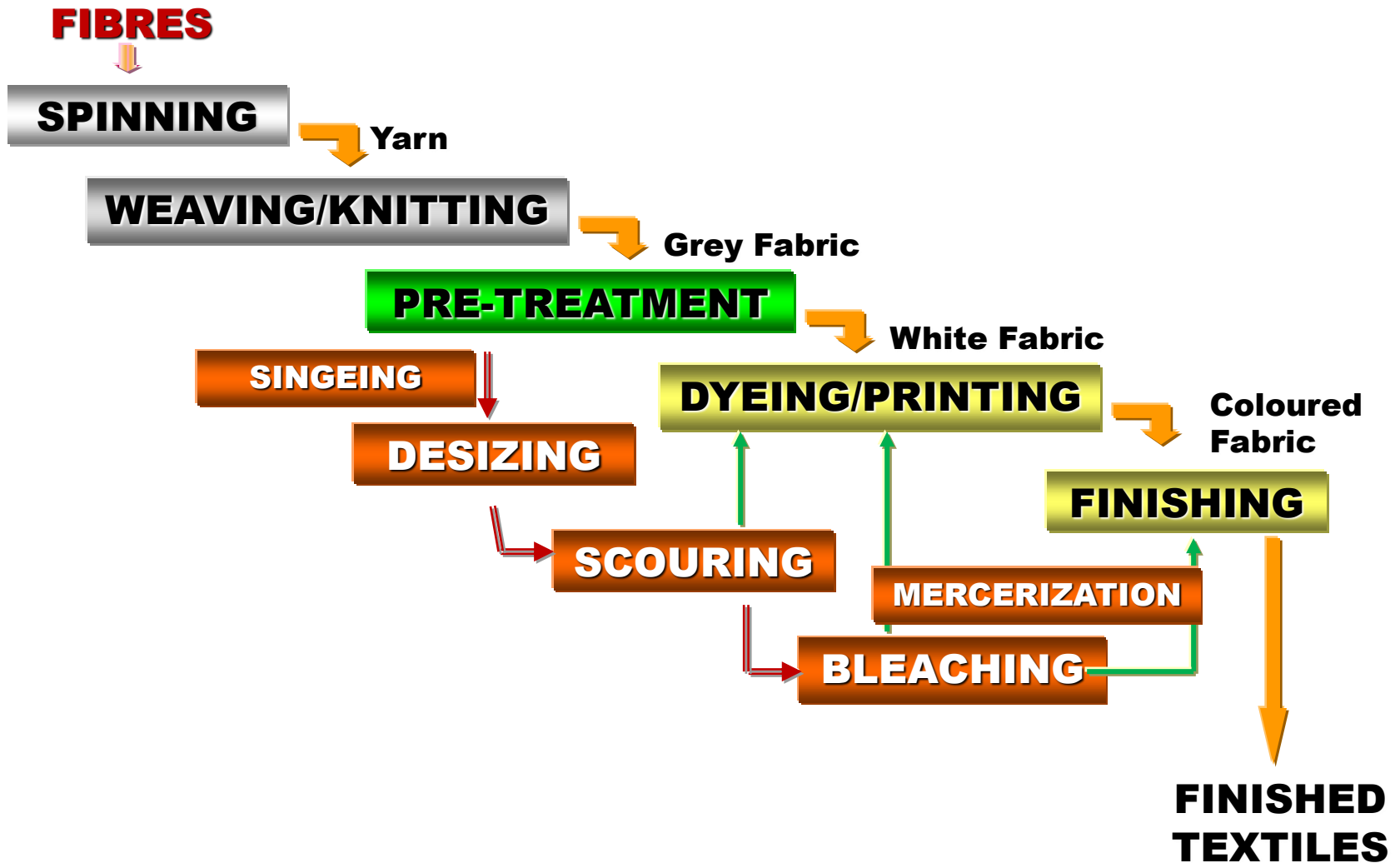
Synthetic / Man made:

- Polyester
 - Nylon
 - Acrylic
 - Viscose
-



Final step of fabric manufacturing processes

-
- Improved comfort
 - Improved downstream processing
 - Appearance (aesthetics)
 - Special functionalities (case basis)
 - Appearance (aesthetics)
 - Special functionalities
 - Feel (touch)
-



Define – Pretreatment

Textile pretreatment is a series of cleaning operations that removes impurities which might adversely affect downstream processes like dyeing, printing and finishing.

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- Natural impurities
 - Added impurities

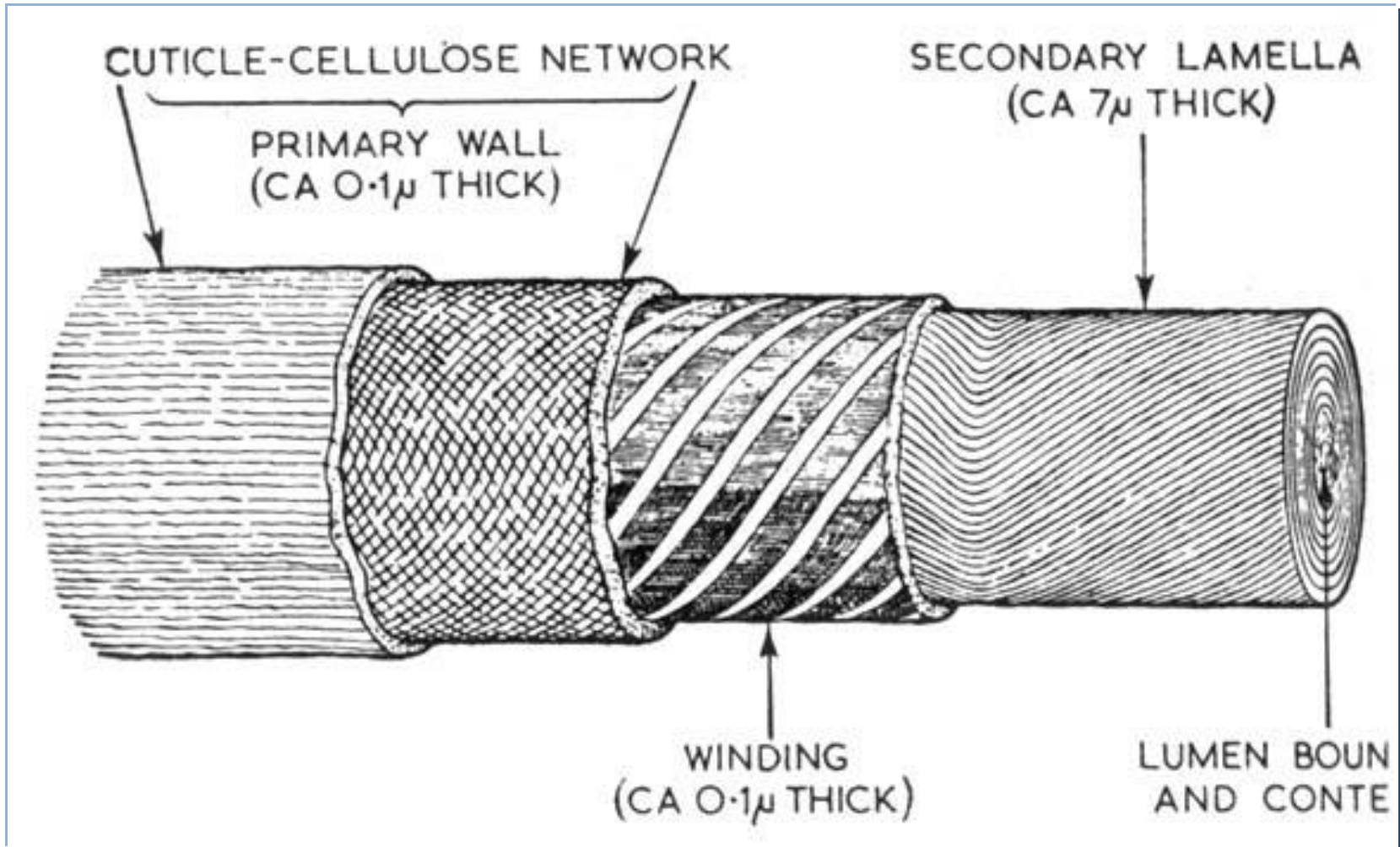
Pretreatment Objectives:

Improvement in absorbency

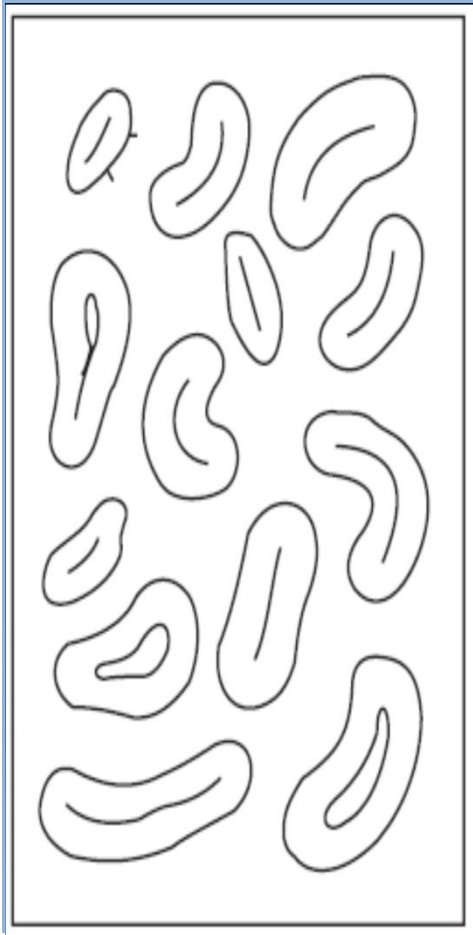
(Should be carried out with Minimum damage to the material)

- ☐ Improved comfort
 - ☐ Improved downstream processing (dyeing / finishing)
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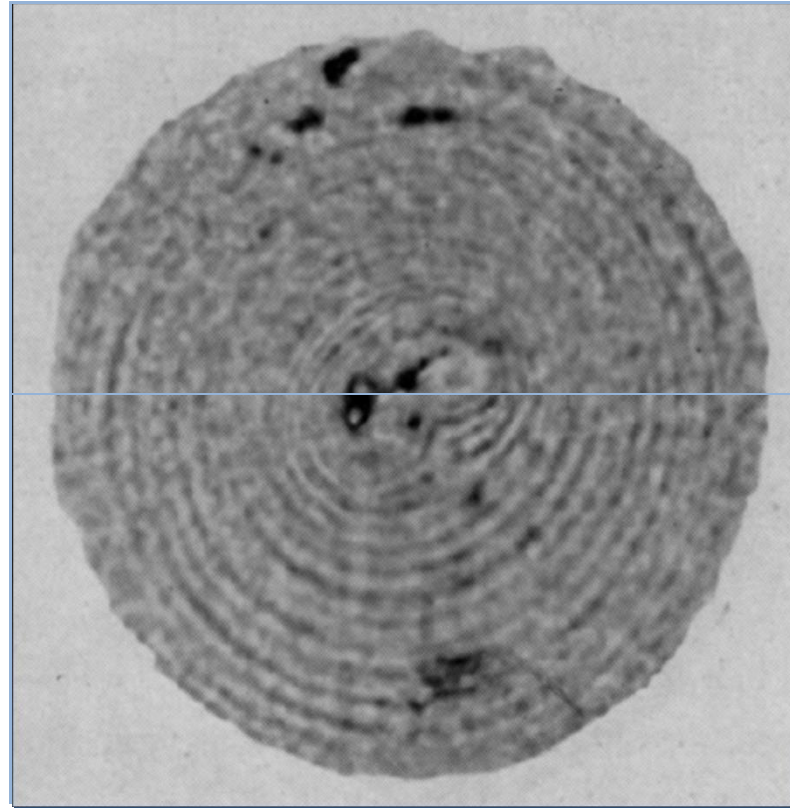
Layered components of a cotton fibre cell wall



Cotton fibres have a typical bean shaped x-section, convoluted, collapsed tube like structure

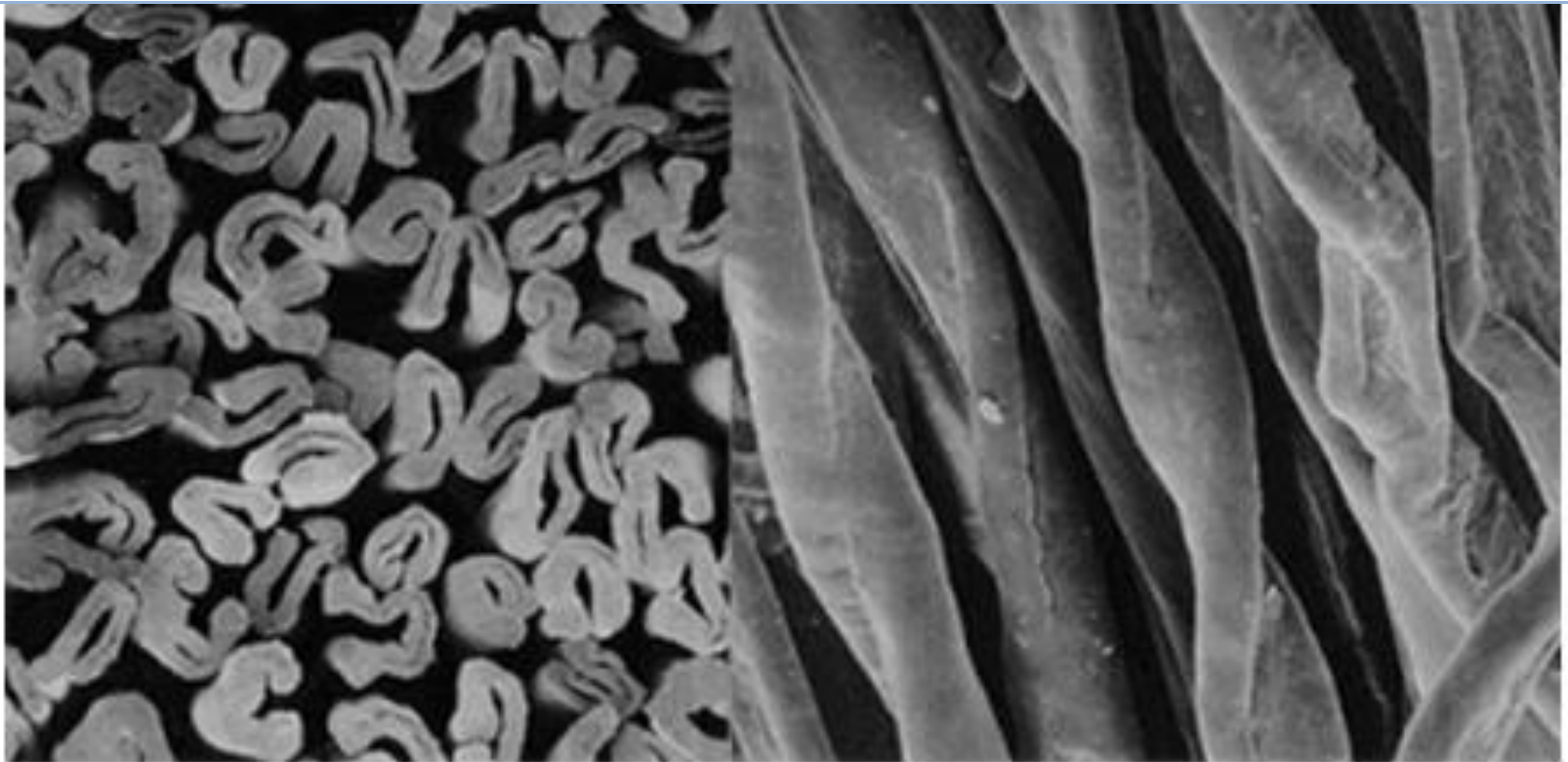


**Typical cross-sections
of matured cotton fibres**

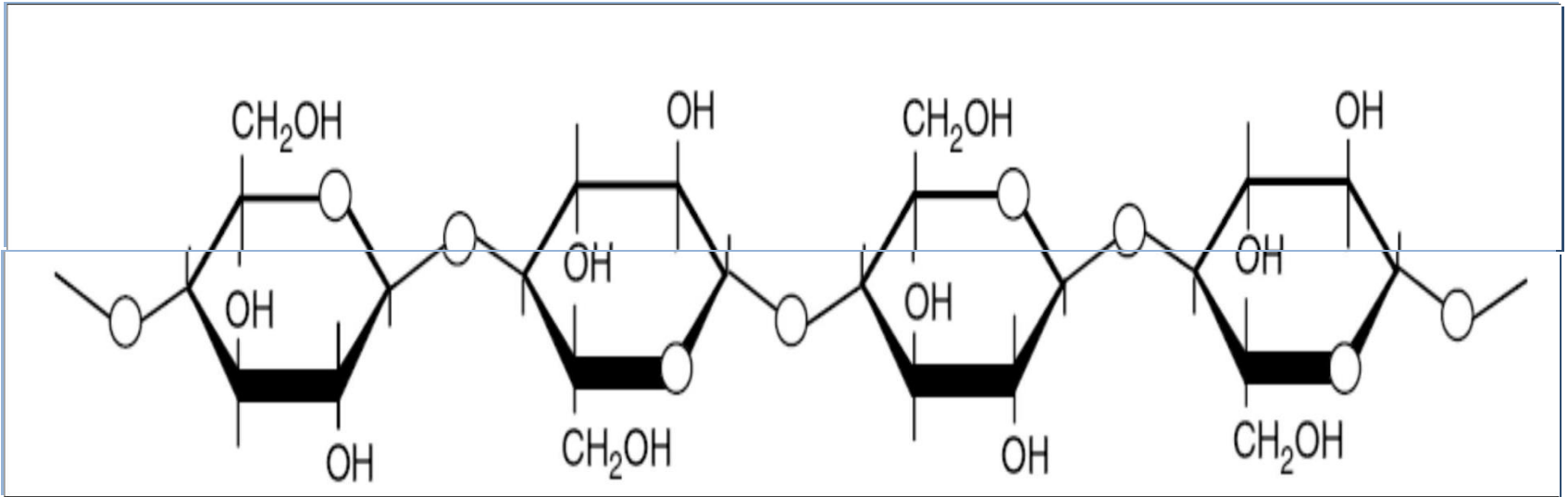


**Cross-section of swollen cotton
fibre showing daily growth rings**

Microscopic view of cotton

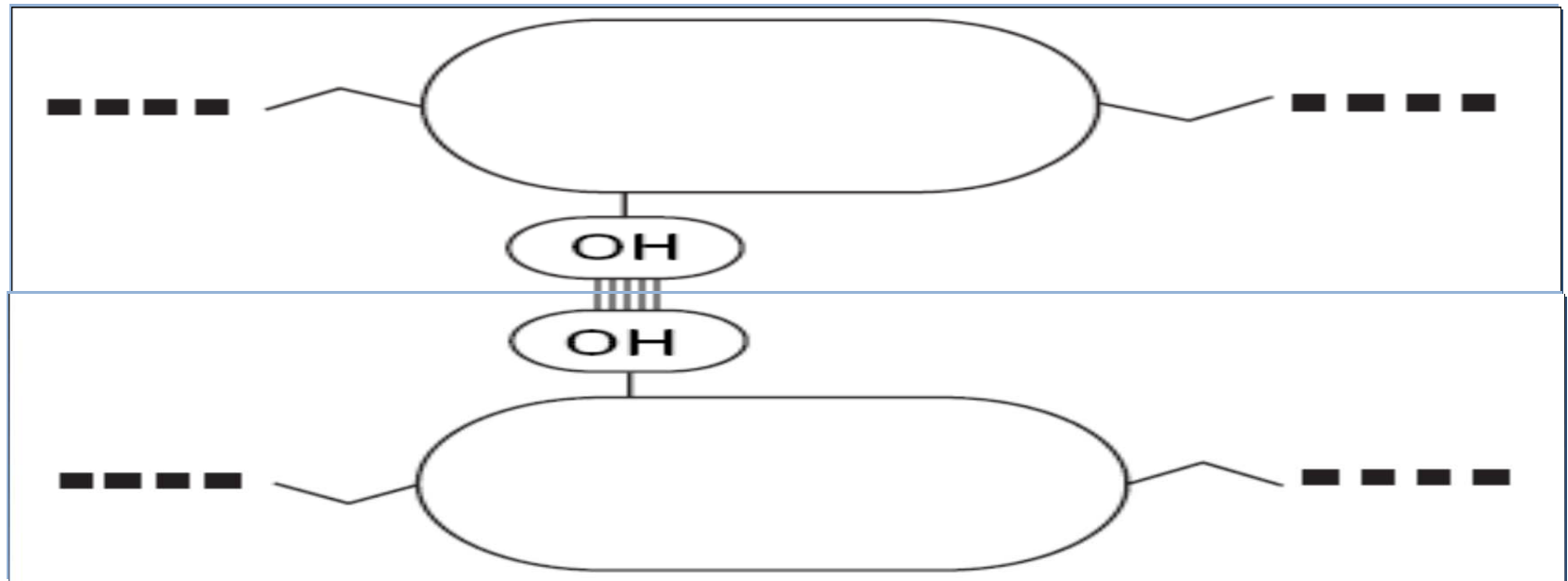
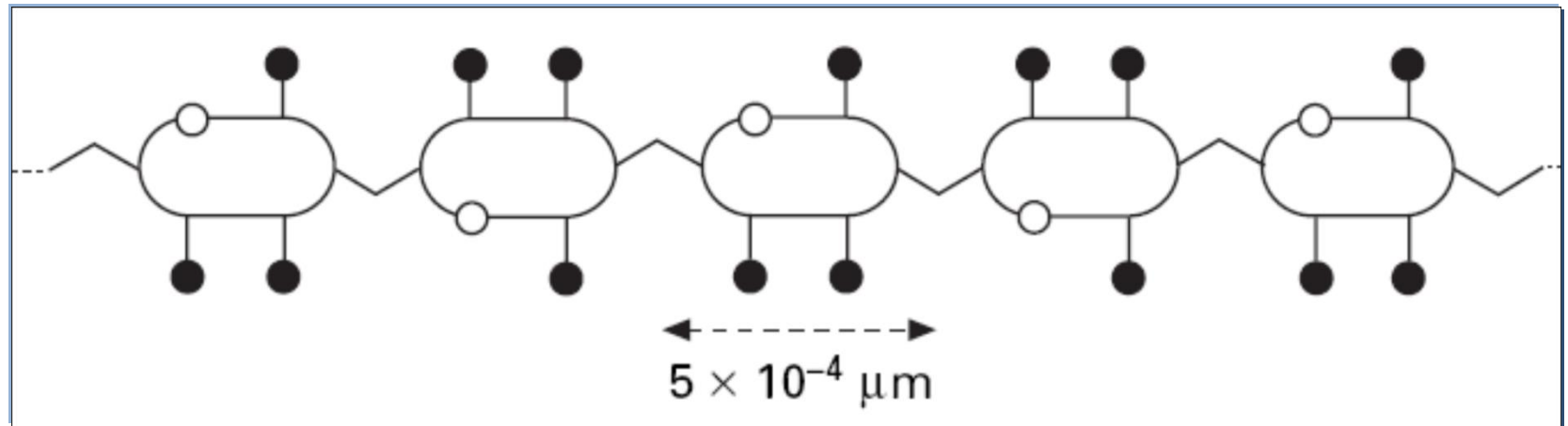


Chemical drawing of Cellulose



Polymer made up of a long chain of glucose molecules linked by C-1 to C-4 oxygen bridges with elimination of water (glycoside bonds). The anhydroglucose units are linked together as **beta-cellobiose**; therefore, anhydro-beta-cellobiose is the repeating unit of the polymer chain

Features of the cellulose molecule chain



Cotton Fibre

Constituents	Composition of a fibre		
	Typical (%)	Low (%)	High (%)
Cellulose	94	88	96
Protein	1.3	1.1	1.9
Pectic matter	0.9	0.7	1.2
Wax	0.6	0.4	1.0
Mineral matter	1.2	0.7	1.6
Malic, citric and other organic acids	0.8	0.5	1.0
Total sugars	0.3	-	-

Proteins

Nitrogenous compounds:

- Present in primary cell wall and lumen
- Yellow colour of cotton is due to presence of proteins and some colouring matter

Some amino acids:

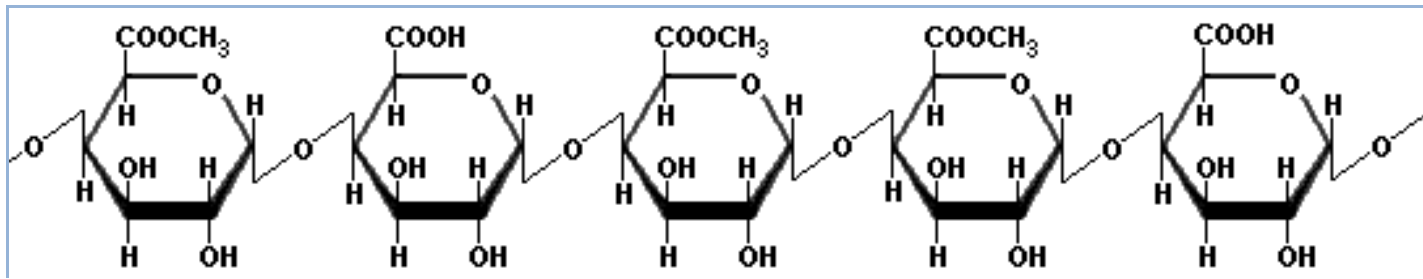
- Leucine
 - Valine
 - Proline
 - Alanine
-

Pectins

- Derivatives of pectic acid
- Found in the cover of citrus fruits
- Polymer of high molecular weight



- Pectin is a polysaccharide that acts as a cementing material in the cell walls of all plant tissues. It is a polymer of α -Galacturonic acid with a variable number of methyl ester groups.
- Some COOH groups are present as Ca and Mg salts.



Components of Fats and Waxes

- The wax present in the primary cell wall of cotton protects the fibre from environmental agencies
 - Responsible for the smooth handle and is a source of hydrophobicity
 - In the presence of wax, cotton has poor wettability
 - The wax consists of long chains of fatty alcohols, fatty acids, their esters, cholesterin & hydrocarbons
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Fats and Waxes

Fats and waxes:

- ☐ Fatty acids
 - ☐ Stearic acid
 - ☐ Palmitic acid
 - ☐ Oleic acid
 - ☐ Fatty alcohols
 - ☐ Cetyl alcohol ($\text{C}_{26}\text{H}_{53}\text{OH}$)
 - ☐ Montanyl alcohol ($\text{C}_{28}\text{H}_{57}\text{OH}$)
 - ☐ Gossipyl alcohol ($\text{C}_{30}\text{H}_{61}\text{OH}$)
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Mineral Matters

- Depends on soil composition
- Determined by ash analysis

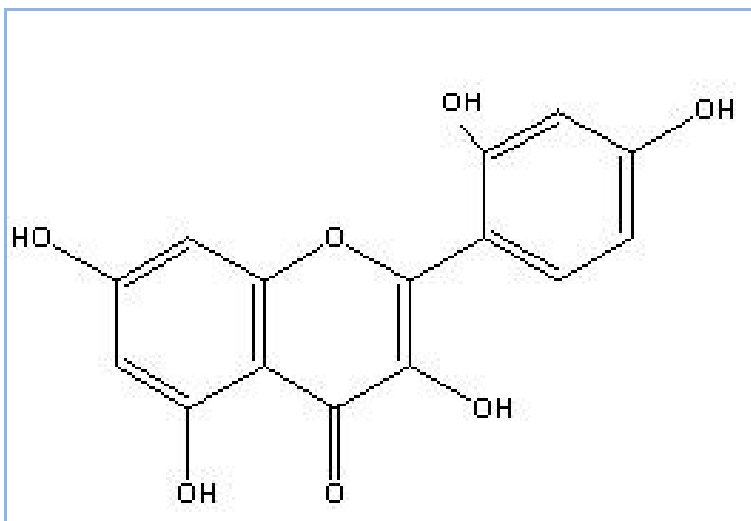
Potassium carbonate	44.8
Calcium carbonate	10.3
Potassium chloride	9.9
Potassium sulphate	9.3
Calcium sulphate	9.0
Magnesium sulphate	8.4
Aluminum oxide	5.0
Ferric oxide	3.0

Colouring Matter

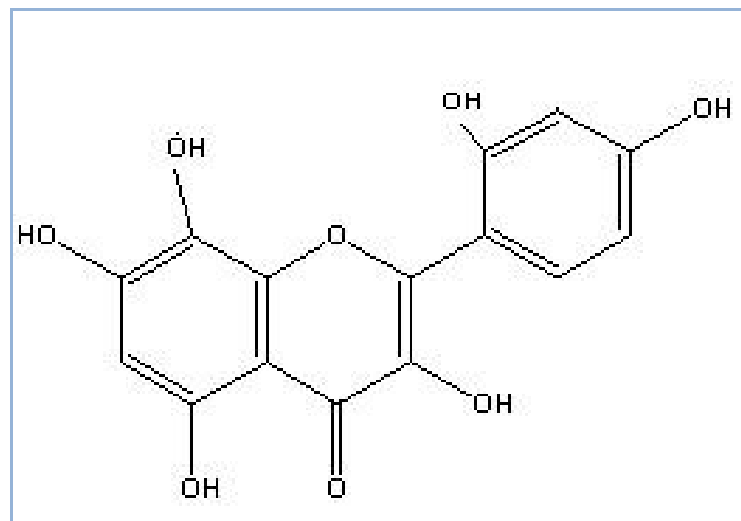
Colored pigments:

Flavones (flavus—Latin for yellow)

3,5,7,2',4' penta hydroxy flavone (Morrin)



3,5,8,2',4' Hexa hydroxy flavone (gossypetin)



Added / acquired impurities

- Mainly sizing matter (protective coating for warp yarns)
 - Machine oils, lubricants, grease, etc.
 - In knitting-coning oil
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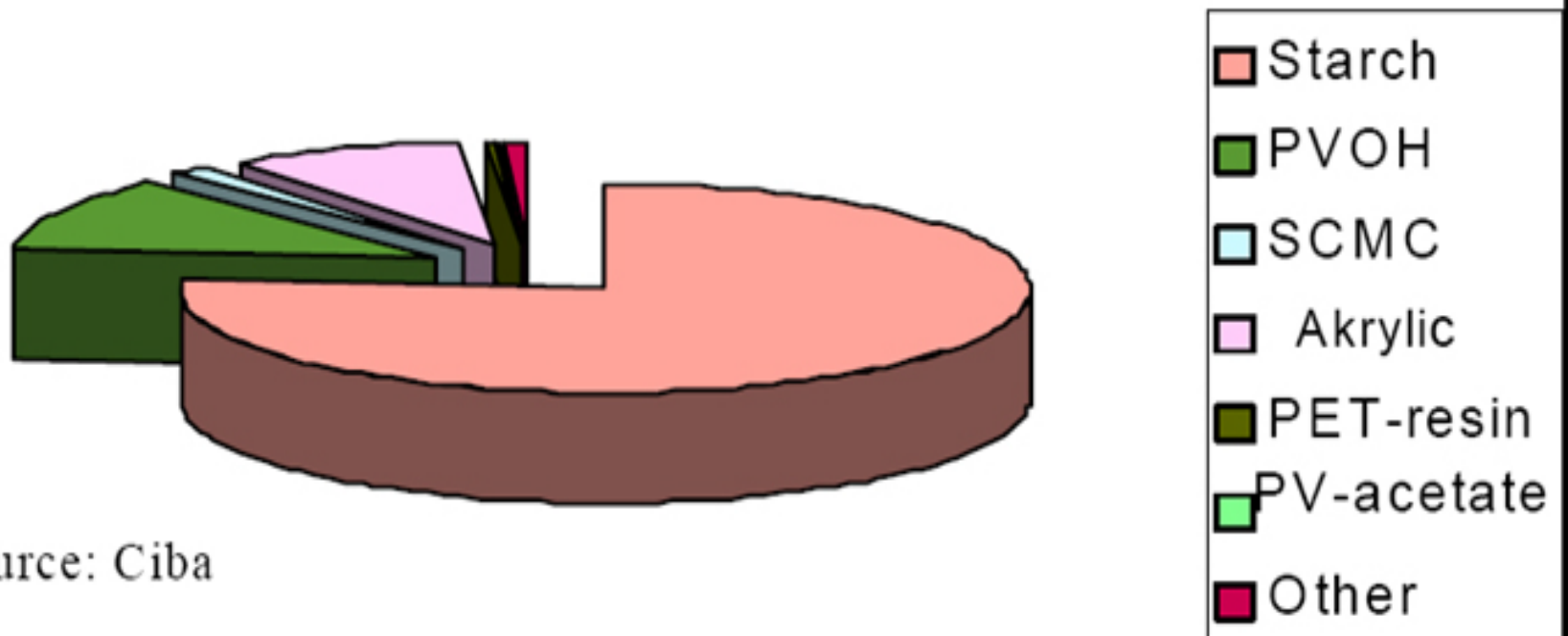
Chemistry of Size

Sizes are mainly formulations, with high molecular weight film forming polymers being the main components.

The size material applied on the warp yarns for facilitating weaving process.

- Natural / their derivatives
 - Synthetic
 - Blend of all these
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Various sizing materials



Source: Ciba

Sizes based on Natural polymers/derivatives

Starch and its derivatives (75%)

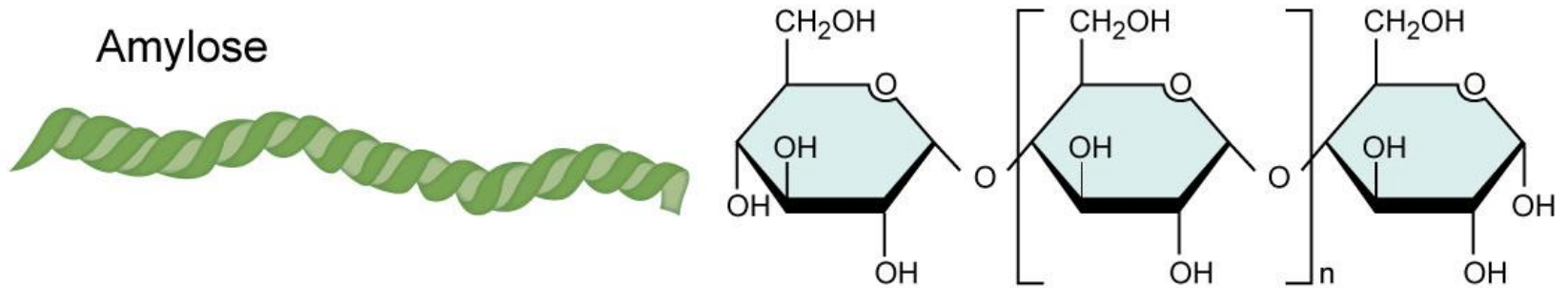
Starch has two components:

- ❑ Amylose: Relatively lower mol. wt. & water soluble (20%)
- ❑ Amylopectin: Higher mol. wt. and difficult to remove (80%)

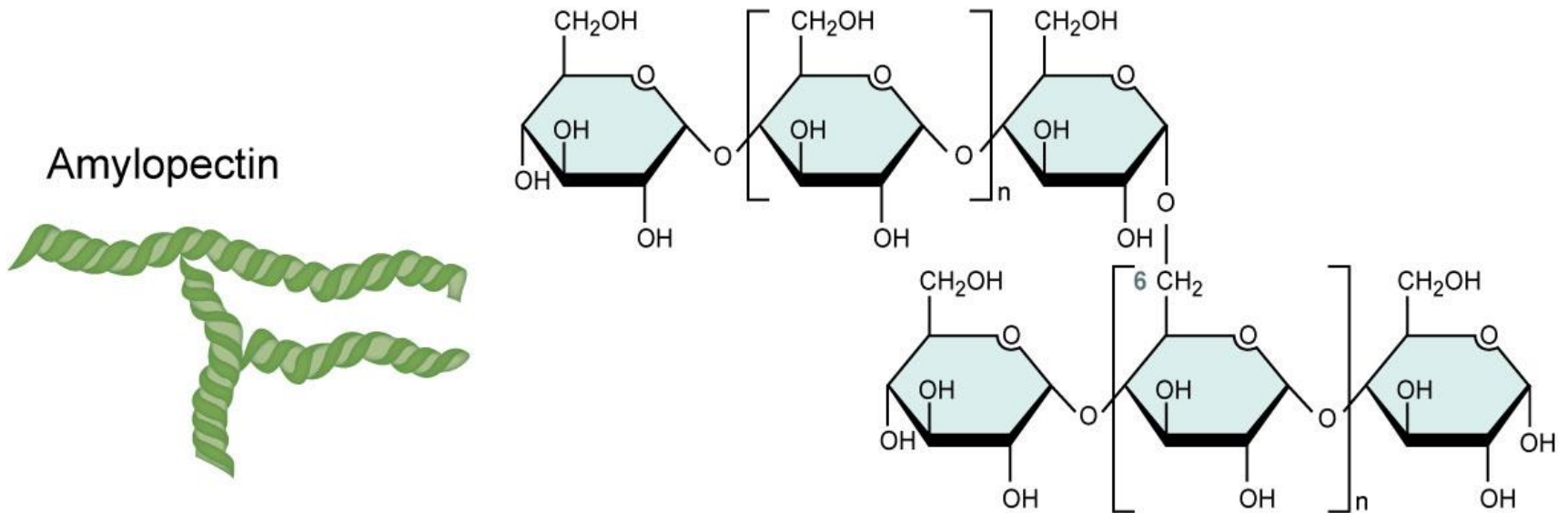
In case of sizes based on natural polymers, add-on is of the order of ~ 15% and they generally require removal by chemical degradation. This results in high pollution of discharge.

Amylose and Amylopectin

Amylose



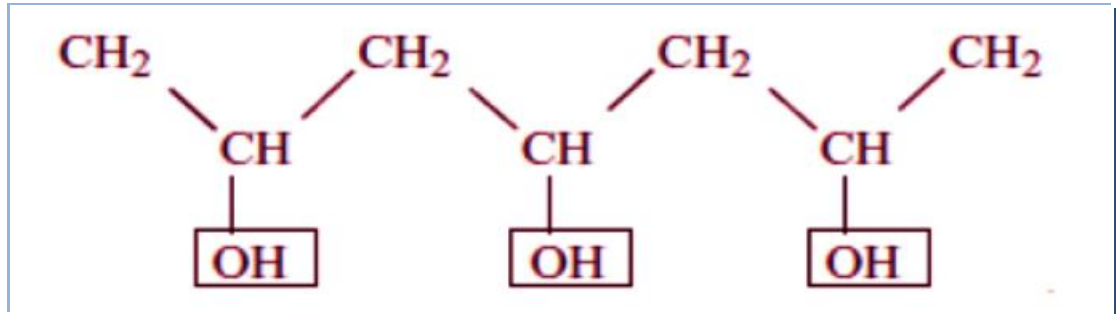
Amylopectin



Sizes based on Synthetic Polymers

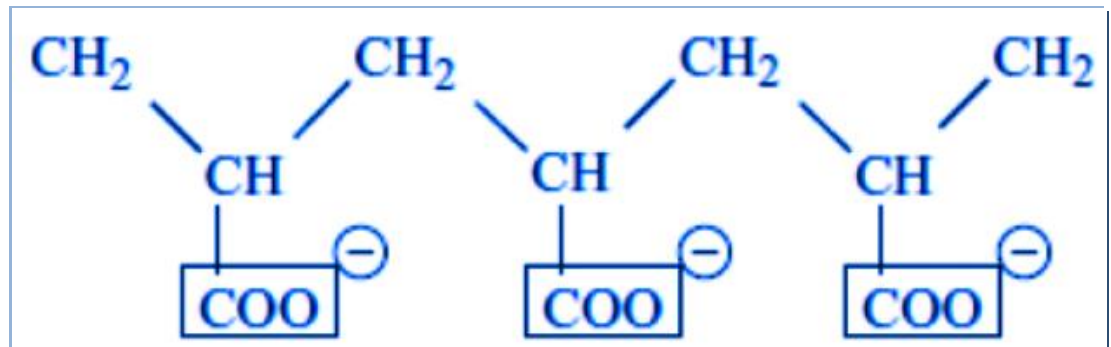
Polyvinyls:

(PVA)

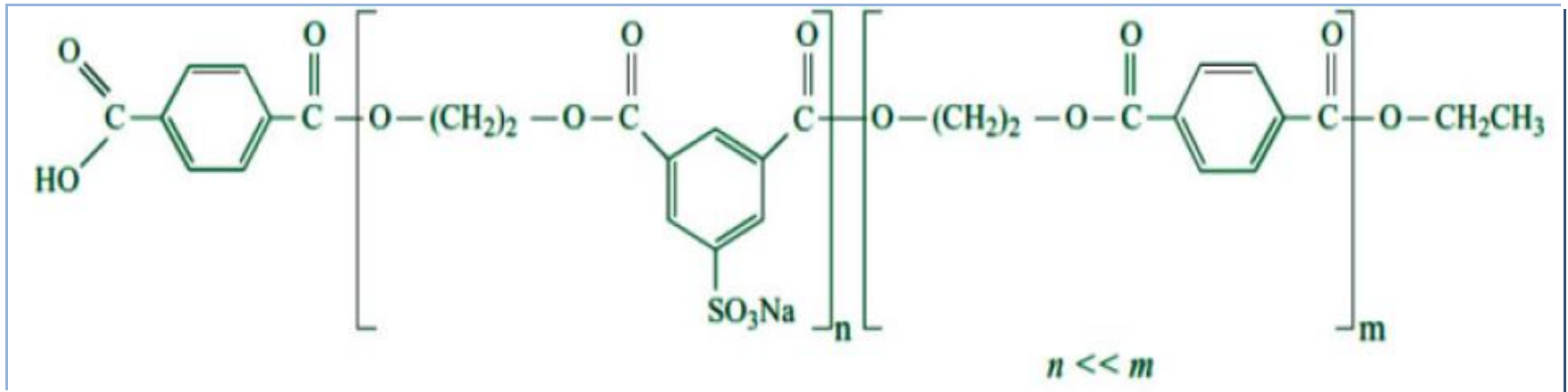


Polyacrylates:

(Polyacrylate)



Water dispersible Polyester



Mixed sizes:

These sizes are generally either water soluble or water dispersible. This results in lower pollution of discharge waters (recovery and recyclability)

Removal of impurities

- ☐ Solubilization
- ☐ Emulsification
- ☐ Chemical breakdown by
 - Hydrolysis
 - Oxidation

Important parameters:

- ☐ Temperature
 - ☐ pH
 - ☐ Time
 - ☐ Circulation
 - ☐ M:L ratio
-

Major Preparatory Processes

Singeing



Desizing



Scouring



Bleaching (*optional*)



OBA treatment (*optional*)




Mercerization (*optional*)

IMPURITIES REMOVED DURING PRE-TREATMENT

Short Fibres  **Singeing**

Applied Impurities (Size Material)  **Desizing**

Natural Impurities (Oil, Wax, Pectins, Proteins)  **Scouring**

Artificial Impurities (Oil, Stains, Dust, Dirt)  **Scouring**

Colour Pigments (Naturally present in cotton)  **Bleaching**
