

KEV: Constant Evaluator

<https://k-ev.org>

*Online optimizer of the equilibrium constants
of the chemical reactions*

Web Application User Guide

by Drs. G. Gamov and A. Meshkov

TABLE OF CONTENTS

I. ABOUT KEV	5
II. GETTING STARTED	6
2.1 Log in or Sign up	6
2.2 Select a Problem.....	6
2.3 Input Data.....	7
2.4 Check the Data and Evaluate	8
2.5 Download Data	8
III. EQUILIBRIUM COMPOSITION	9
3.1 Select the Problem	9
3.2 Data Description and Manual Input.....	9
3.2.1 Stoichiometric matrix (Stoichiometric coefficients)	11
3.2.2 Decimal logarithms of equilibrium constants.....	11
3.2.3 Total (or equilibrium) concentrations of reagents	12
3.2.4 Particle names	12
3.2.5 The particle in relation to the yields of products are calculated	13
3.3 Evaluate.....	13
3.4 Output (Calculated) Data	14
3.5 Option: Equilibrium Concentrations as Input.....	15
3.6 Option: Upload Data from CSV or Tab Delimited File(s)	15
3.6.1 Stoichiometric matrix	16
3.6.2 Total (or equilibrium) concentrations of reagents	16
3.6.3 Decimal logarithms of equilibrium constants.....	16
Bulk upload (all files at once).....	17
3.7 Option: Upload Data from XLSX Single File	21
3.7.1 Stoichiometric matrix	22
3.7.2 Total (or equilibrium) concentrations of reagents	22
3.7.3 Decimal logarithms of equilibrium constants.....	23
3.7.4 Particle to get fractions of.....	23
3.8 Download Results of Calculations.....	24
3.8.1 Single xlsx file	24
3.8.2 Zip archive	25

4.1 Select the Problem	26
4.2 Data Description and Manual Input	26
4.2.1 Stoichiometric matrix (Stoichiometric coefficients)	27
4.2.2 Decimal logarithms of equilibrium constants	28
4.2.3 Total (or equilibrium) concentrations of reagents	28
4.2.4 Particle names	29
4.2.5 Absorbance and deviations	29
4.2.6 Molar extinction coefficients	30
4.2.7 Constants to evaluate and wavelengths to use	31
4.2.8 Threshold	31
4.2.8 Search density	32
4.3 Evaluate	32
4.4 Output (Calculated) Data	32
4.5 Option: Upload Data from CSV or Tab Delimited File(s)	35
4.5.1 Stoichiometric matrix	35
4.5.2 Total (or equilibrium) concentrations of reagents	36
4.5.3 Decimal logarithms of equilibrium constants	36
4.5.4 Absorbance	36
4.5.5 Molar extinction coefficients	36
4.5.6 Target	37
4.6 Option: Upload Data from CSV or Tab Delimited File(s)	37
4.7 Option: Upload Data from XLSX Single File	37
4.7.1. absorbance	38
4.7.2. mol_ext_coefficients	38
4.7.3. target	40
4.8 Option: Molar Extinction Coefficients Calculation	41
4.8.1 Select the Problem	41
4.8.2 Data Description	42
4.8.3 Upload Data from CSV or Tab Delimited File(s)	42
4.8.4 Option: Upload Data from XLSX Single File	44
4.9. More playground	46
5.1 Select the Problem	48

5.2 Data Description and Manual Input	49
5.2.1 Stoichiometric matrix (Stoichiometric coefficients)	50
5.2.2 Decimal logarithms of equilibrium constants	51
5.2.3 Total (or equilibrium) concentrations of reagents	51
5.2.4 Particle names	52
5.2.5 EMF and deviations	52
5.2.6 E_0 and Slope	53
5.2.7 Constants to evaluate	53
5.3 Evaluate	53
5.4 Output (Calculated) Data	54
5.5 Option: Upload Data from CSV or Tab Delimited File(s)	55
5.5.1 Stoichiometric matrix	56
5.5.2 Total (or equilibrium) concentrations of reagents	56
5.5.3 Decimal logarithms of equilibrium constants	57
5.5.4 EMF	57
5.5.5 Target	57
5.6 Option: Upload Data from CSV or Tab Delimited File(s)	57
5.7 Option: Upload Data from XLSX Single File	57
5.7.1 emf	58
5.7.2 target	58
5.8. More playground	59

I. ABOUT KEV

KEV is a chemical software developed for solving two classes of problems of the chemical equilibria theory. It is free, open source and available both as an online service and scripts bundle.

1. Equilibrium concentrations. Calculation of the equilibrium composition of the compounds mixture using stoichiometric scheme of reactions, total or equilibrium concentrations of reagents and known equilibrium constants.
2. Equilibrium constants. Estimation of the equilibrium constants from the experimental data (UV-Vis spectroscopy or potentiometric measurements).

KEV implements the statistical approach in solving equilibrium problems thus being broadly applicable across different systems (any system could be processed; theoretically, the number of chemical reactions occurring in the calculated mixture is unlimited), more accurate and reliable comparing with more traditional graphical methods.

What is described in this guide is **KEV web application** (<https://k-ev.org/kev>). If you are familiar with R Programming Language you could use **R scripts directly**:

- Download KEV repository from <https://gitlab.com/a.meshkov/KEV/> You could do it with Git or (if you have no idea what Git is and how to deal with it) directly from the web page using *Download* button (cloud and arrow image).
- Open *kev.Rproj* in RStudio. If you do not use RStudio be sure to set the working directory to the root of project with *setwd()* function.
- Play with *run.r* file (do not source the whole file, try different datasets and options)

II. GETTING STARTED

2.1 Log in or Sign up

KEV web application requires the very simple authentication. Log in to <https://kev.org/kev> if you have already gained the login and password or register if you have not.

Please consider signing up is free and requires only sending a letter towards Dr. Aleksandr Meshkov or Dr. Georgiy Gamov via e-mails ameshkoff@protonmail.com or ggamov@isuct.ru. In reply, you will be provided with the login and the password.

2.2 Select a Problem

After logging in you should choose which type of problem should be solved. KEV interface is divided in tabs corresponding specific calculation problems (Fig. 1):

- **Equilibrium concentrations.** Calculate the equilibrium composition of the mixture
- **Spectrophotometry.** Evaluate unknown equilibrium constants from the experimental spectral data (UV-Vis spectroscopy)
- **Extinction Coefficients.** Calculate molar extinction coefficients from a series of the experimental spectra data
- **EMF.** Evaluate unknown equilibrium constants from the potentiometric experimental data

KEV: Chemistry Constant Evaluator

Calculate Equilibrium Concentrations

Column delimiter of your data files

☒ , ☐ ; ☐ tab

Particle to get fractions of

molecule1

Bulk upload / download (optional)

Upload all data

Choose CSV files or XLSX file with multiple sheets

Browse... No file selected

Download all data

zip xlsx

Upload or type input data

Stoichiometric coefficients				
	molecule1	molecule2	molecule3	molecule4
1	1	1	1	1
2	1	1	1	1

K: lg constants	
	k_constants_log10
1	1
2	1

Concentrations	
Input	Total
	molecule1

Figure 1. KEV interface. Different functionality tabs are marked with red

2.3 Input Data

KEV is very flexible in the ways you can provide the data to. Use one of the options described below or mix them together.

- Manual input
 - Type all the data in the tables provided
 - Add or remove rows and columns using context menu
 - Both “.” or “,” as decimal separator are available
 - Do not use “,” or “ ” as a mark between big interval decimals
 - Copy + paste from Excel spreadsheets or a similar app
 - Copy from the source app, select cells in the KEV table and paste using hotkeys (Ctrl+V in Windows)
- Upload file(s)
 - Upload all prepared data at once (*Bulk input*). Click *Upload all data* and upload one of the following options:
 - CSV or tab delimited files, one file for one table. Files should be properly named and properly formatted. Supported file formats:
 - CSV, both comma and semicolon separated
 - Tab separated file
 - Excel file with a few sheets (xlsx only). Every sheet should be properly named and properly formatted
 - Upload prepared data one table after another using *Choose CSV* under each table. So if you have 3 tables to fill prepare 3 files and then upload them using 3 different *Browse* buttons. Use this option if you want to change only part of the large dataset already uploaded. Supported file formats:
 - CSV, both comma and semicolon separated
 - Tab separated file

We recommend starting with *Copy + paste* option or *Upload all prepared data* at once.

Example input datasets could be obtained here:

<https://gitlab.com/a.meshkov/KEV/tree/master/input>

- *concentrations*: Equilibrium composition
- *spectrophotometry*: Unknown Equilibrium constants via UV-Vis spectroscopy
- *emf*: Unknown Equilibrium constants via EMF
- *molar.extinction.coefficients*: Molar extinction coefficients from a series of the experimental spectra data

IMPORTANT: It is not possible to add or remove rows if the data is uploaded from the disc and not typed manually or pasted from another application.

TIP: To switch between column separators (“,” “;” or *tab*) use *Column delimiter* in the left top of the window.

2.4 Check the Data and Evaluate

After the input data is provided check if all is OK. If not you will get a bunch of red and grey error messages. Most frequent error causes and fixes are:

- Check *Column delimiter* control. If you use csv file with semicolon “;” delimiter should be also “;”
- Check if particle (molecule) names are consistent between different tables i.e. coefficient, concentrations and molar extinction matrices
- Check the same for wavelenghts
- If you are uploading data table by table just try to load remaining data, the “error” could be a temporary thing disappearing after all the data is provided

Now when all is OK click *Evaluate* and wait till the unknown constants, concentrations and other stuff are calculated.

2.5 Download Data

Both the input and calculated data could be downloaded to your local computer.

- Download one table. Click *csv* or *xlsx* button under the table you prefer to download. If you choose the CSV option the downloaded CSV file use the column separator defined by *Column delimiter* control.
- Download all the data, input and calculated (if available) as well. Choose *zip* or *xlsx* option of the *Download all data* in the *Bulk upload / download* section.
 - *zip*. Zip archive containing CSV files. The downloaded CSV file use the column separator defined by *Column delimiter* control.
 - *xlsx*. Excel file (xlsx) with multiple sheets

III. EQUILIBRIUM COMPOSITION

3.1 Select the Problem

Select *Equilibrium concentrations* in the top menu to open the tab.

Let us consider the equilibria in the system containing phosphoric acid and Cu^{2+} ions at T 298.15 K. The example data could be obtained:

- Input Data
 - CSV comma separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/concentrations/ds.3p.long/csv.comma>
 - CSV semicolon separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/concentrations/ds.3p.long/csv.semicolon>
 - Tab separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/concentrations/ds.3p.long/txt.tab>
 - Excel (xlsx) file
https://gitlab.com/a.meshkov/KEV/tree/master/input/concentrations/ds.3p.long/example_eq_conc_calc.xlsx
- Output (calculated) data
 - Zip archive with CSV files
https://gitlab.com/a.meshkov/KEV/tree/master/output/concentrations/ds.3p.long/gui/eq_conc_calc_res.zip
 - Excel (xlsx) file
https://gitlab.com/a.meshkov/KEV/tree/master/output/concentrations/ds.3p.long/gui/eq_conc_calc_res.xlsx

3.2 Data Description and Manual Input

In order to calculate the equilibrium concentrations of reagents and reactions' product KEV requires the following data:

- Tables (matrices and vectors)
 - Stoichiometric matrix of reactions (a.k.a. Stoichiometric coefficients)
 - Decimal logarithms of equilibrium constants
 - Total (or equilibrium) concentrations of reagents
- Text input
 - The particle in relation to the yields of products are calculated

Column delimiter of your data files

Particle to get fractions of

Upload or type input data

Stoichiometric coefficients

	molecule1	molecule2	molecule3	molecule4
1	1	1	1	1
2	1	1	1	1
3	1	1	1	1
4	1	1	1	1

K: lg constants

	lg10(K)
1	1
2	1
3	1
4	1

Concentrations

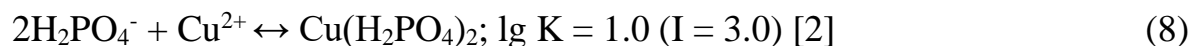
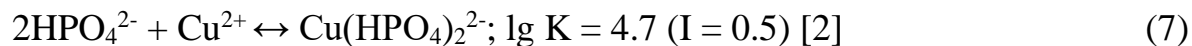
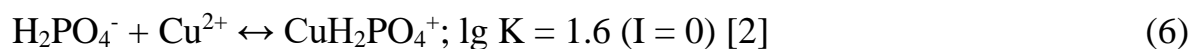
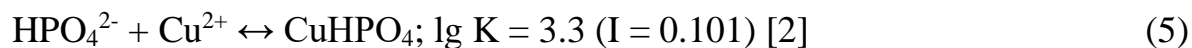
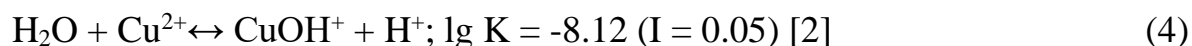
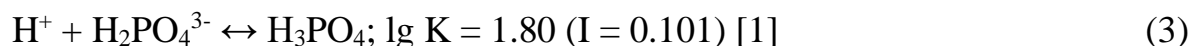
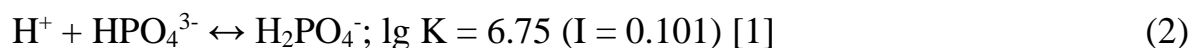
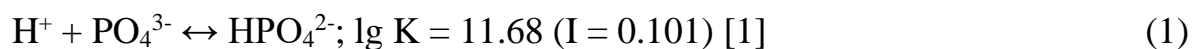
Input	Total	molecule1	molecule2	molecule3	molecule4
1	0.001	0.001	0.001	0.001	0.001
2	0.001	0.001	0.001	0.001	0.001
3	0.001	0.001	0.001	0.001	0.001
4	0.001	0.001	0.001	0.001	0.001
5	0.001	0.001	0.001	0.001	0.001
1	tot	tot	tot	tot	tot

Particle names, comma separated

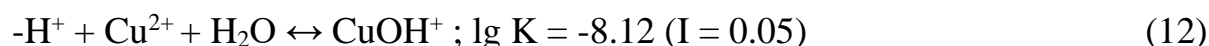
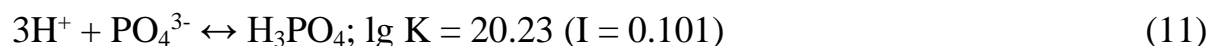
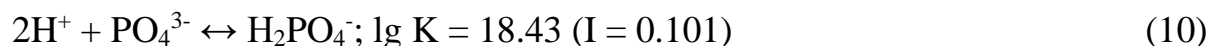
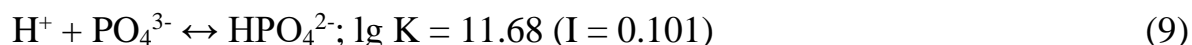
molecule1, molecule2, molecule3, molecule4

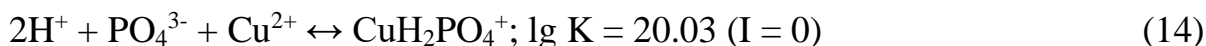
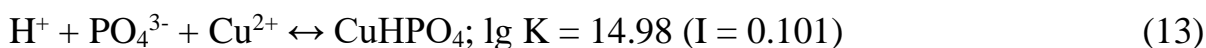
Figure 2. KEV interface. Table of stoichiometric coefficients is marked with red, table of decimal logarithms of equilibrium concentrations is marked with green, table of total (or equilibrium) concentrations of reagents is marked with blue, reagents names are marked with purple

First, one needs to scribe all the reactions known for system taken into consideration. For the example data it is:



It could be seen easily that all products could be obtained by combination of three reagents only. Namely those are H^+ ; PO_4^{3-} and Cu^{2+} . To make this more evident let us re-write the system of chemical reactions as follows:





H^+ , PO_4^{3-} and Cu^{2+} are so-called *basis particles* or *basis set* or just *basis*. The choice of basis should be justified by the following assumptions: a) all the products should be obtained by combining the basis particles; b) the number of basis particles should be minimal (but the first condition must be met). Now provide this data to KEV.

3.2.1 Stoichiometric matrix (*Stoichiometric coefficients*)

Stoichiometric coefficients before H^+ , PO_4^{3-} and Cu^{2+} form the following matrix:

1 1 0

2 1 0

3 1 0

-1 0 1

1 1 1

2 1 1

2 2 1

4 2 1

Type them into KEV table *Stoichiometric coefficients*. Adding or removing columns/rows could be made with the context menu (mouse right click).

TIP: If you do not want to type numbers cell by cell (we don't), copy numbers into a blank Excel file, split data into columns and copy + paste it from the Excel file to the KEV table *Stoichiometric coefficients*

3.2.2 Decimal logarithms of equilibrium constants

Values of $\lg K_{\text{total}}$ (9-16) include stepwise protonation and metal complexation constants.

Total $\lg K$ values form $\lg K$ vector:

11.68

18.43

20.23

-8.12

14.98

20.03

28.06

37.86

Type it into KEV table *K: lg constants* preserving the order or copy + paste as for the stoichiometric matrix data.

IMPORTANT: Please note the decimal logarithms of equilibrium constants (lg K) should be input and not the K values

3.2.3 Total (or equilibrium) concentrations of reagents

Let us also give the total concentrations for three mixtures. For example, 1) $C(H^+) = 0.01$; $C(PO_4^{3-}) = 0.01$; $C(Cu^{2+}) = 0.01$; 2) $C(H^+) = 0.02$; $C(PO_4^{3-}) = 0.01$; $C(Cu^{2+}) = 0.01$; 3) $C(H^+) = 0.03$; $C(PO_4^{3-}) = 0.01$; $C(Cu^{2+}) = 0.01$.

It forms another matrix (particles as columns and mixtures as rows):

0.01 0.01 0.01

0.02 0.01 0.01

0.03 0.01 0.01

Type or copy + paste it to KEV table *Concentrations >> Input*

The footnote of the Table “Concentrations” contains a row of variables capable of taking two values: *tot* for total concentrations, and *eq* for equilibrium concentrations (if some input concentrations are equilibrium ones).

IMPORTANT: Check if the column number in *Stoichiometric coefficients* and *Concentrations* tables is equal. If one table has empty or excess columns remove them with the context menu (mouse right click)

3.2.4 Particle names

Columns in both *Stoichiometric coefficients* and *Concentrations* tables have the same names which are the names of the base particles. By default KEV uses particle names *molecule1*, *molecule2*, *molecule3* etc. If you want to change them (and you'd better do) type new base particle names into the *Particle names, comma separated* field.

H, PO4, Cu

IMPORTANT: Make sure the quantity of reagents in the string *Particle names, comma separated* corresponds to the quantity of columns of Table *Stoichiometric coefficients*. Delete or add columns if required

3.2.5 The particle in relation to the yields of products are calculated

Type one of the particle names into *Particle* to get fractions of field. Make sure it corresponds to one of the particle names in the *Particle names, comma separated* field.

Taking all required values into KEV looks as follows (Fig. 3).

The screenshot shows the KEV software interface with the following sections:

- Top Bar:** A search bar containing "PO4".
- Bulk upload / download (optional):**
 - Upload all data: Choose CSV files or XLSX file with multiple sheets. A "Browse..." button is present, with "No file selected" below it.
 - Download all data: Buttons for "zip" and "xlsx".
- Upload or type input data:**
 - Stoichiometric coefficients:** A table with columns H, PO4, and Cu.

	H	PO4	Cu
1	1	1	0
2	2	1	0
3	3	1	1
4	-1	0	1
5	1	1	1
6	2	1	1
7	2	2	1
8	4	2	1
 - K: lg constants:** A table with a single column log10(K).

	log10(K)
1	11.68
2	18.43
3	20.23
4	-8.12
5	14.98
6	20.03
7	28.06
8	37.66
 - Concentrations:** A table with columns Input and Total, and sub-columns H, PO4, and Cu.

	Input	Total	H	PO4	Cu
1	0.01			0.01	0.01
2	0.02			0.01	0.01
3	0.03			0.01	0.01
1	tot		tot		tot
- Choose CSV File:** Three "Browse..." buttons are present, one for each table, with "No file selected" below each.
- Particle names, comma separated:** A text field containing "H, PO4, Cu".

Figure 3. Data on equilibria in the system containing H^+ , PO_4^{3-} , Cu^{2+} input into KEV

3.3 Evaluate

After the input data is provided check if all is OK. If it is not you will get a bunch of red and grey error messages. Most frequent error causes and fixes are:

- Check *Column delimiter* control. If you use csv file with semicolon ";" delimiter should be also ";"
- Check if particle (molecule) names are consistent between different tables i.e. coefficient, concentrations and molar extinction matrices
- Check the same for wavelenghts
- If you are uploading data table by table just try to load remaining data, the "error" could be a temporary thing disappearing after all the data is provided

In order to calculate the equilibrium concentrations of the reagents and products click *Evaluate* button right under the input data tables.

CSV XLSX

Particle names, comma separated

H, PO4, Cu


Evaluate  Press me!

Figure 4. “Evaluate” button

3.4 Output (Calculated) Data

The results will appear below a few moments later (Fig. 5):

Evaluate

Equilibrium concentrations

	H_1	PO4_1	Cu_1	H + PO4_1	H + PO4_2	H + PO4 + Cu_1	- H + Cu_1	H + PO4 + Cu_2	H + PO4 + Cu_3	H + PO4 + Cu_4	H + PO4 + Cu
1	3.80944551713618e-08	9.6560308755452e-08	0.0020481910817524	0.00176059884422119	0.000377157005627523	1.85675420666495e-12	0.00040785735988877	0.00719500140731675	3.07533655047646e-05	0.00031819349408552	2.9134987020255
2	0.000147176408262386	1.29713295427535e-13	0.00757573885017166	9.13752534624635e-06	0.00756262726951504	5.3203694796299e-07	3.9046390526064e-07	0.00013811905232933	0.00226085507753635	3.17016731931698e-08	4.3328174127596
3	0.00996557011611662	2.8551797384337e-17	0.0076365677438667	1.36186984014782e-07	0.00763198927255231	3.66469730755774e-05	5.81294293571826e-09	2.0750750607411e-06	0.0023202629698885	7.09854619633506e-12	4.4480914884227

CSV XLSX

Fractions per PO4

	m	-lg(C(PO4))	PO4_1	H + PO4_1	H + PO4_2	H + PO4 + Cu_1	H + PO4 + Cu_2	H + PO4 + Cu_3	H + PO4 + Cu_4	H + PO4 + Cu_5
1	S_1	7.02	0.00	17.61	3.77	0.00	71.95	0.31	6.36	0.00
2	S_2	12.89	0.00	0.09	75.63	0.01	1.38	22.81	0.00	0.09
3	S_3	16.54	0.00	0.00	76.32	0.37	0.02	23.20	0.00	0.09

CSV XLSX

Residuals matrix

	H	PO4	Cu
1	4.91100216049034e-13	4.17200995972422e-13	3.75914577244174e-13
2	4.85722573273506e-15	5.03069808033274e-15	1.21430643181876e-15
3	1.2260886750326e-10	1.83545539877983e-10	5.21775331274732e-11

CSV XLSX

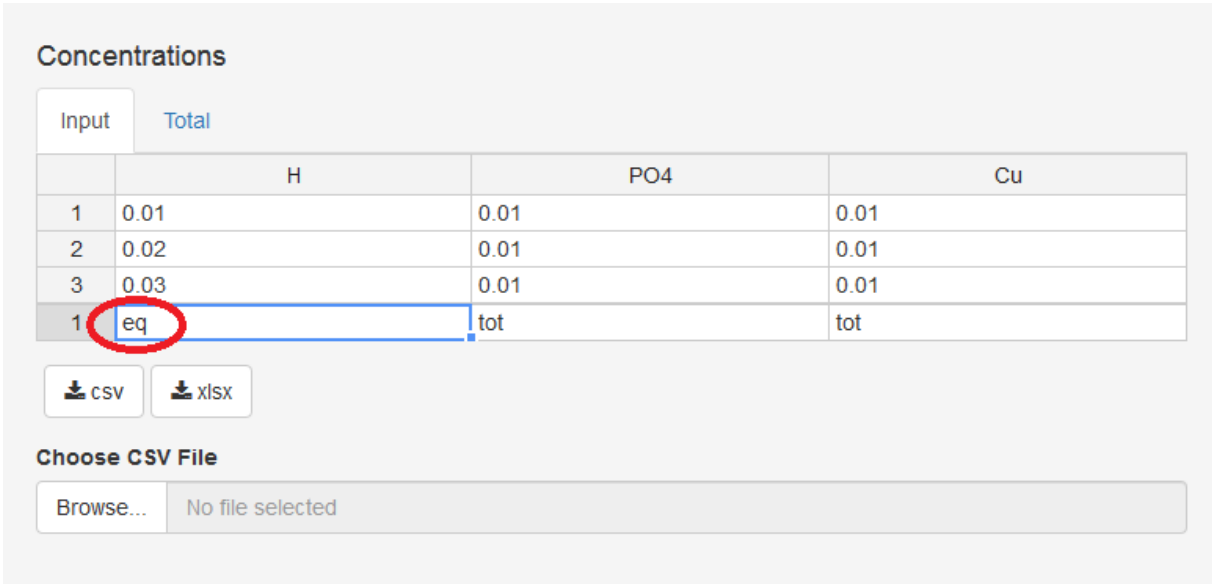
Figure 5. The results of equilibrium constants calculations for the system containing H^+ , PO_4^{3-} , Cu^{2+}

- *Equilibrium concentrations*: all the equilibrium concentrations for basis particles and products
- *Fractions per PO4*: yields of products in relation towards preliminarily chosen basis particle (namely, PO_4^{3-})
- *Residual matrix*: relative errors of the calculated total concentrations (differencies between the calculated and observed concentrations divided by observed calculations)

- *Correlations >> Total* (in the upper part of the window, tab near the observed concentrations): calculated total concentrations

3.5 Option: Equilibrium Concentrations as Input

Let us consider now the hydrogen ions concentration is equilibrium but not the total one. To make a correspondent change one needs to replace the *tot* variable by *eq* variable in the footnote of the Table “Concentrations” (Fig. 6).



The screenshot shows a web interface titled "Concentrations". It has two tabs: "Input" and "Total", with "Total" being the active tab. Below the tabs is a table with three columns: "H", "PO4", and "Cu". The table has four rows. The first three rows contain numerical values: 0.01, 0.02, and 0.03 for column "H"; 0.01, 0.01, and 0.01 for column "PO4"; and 0.01, 0.01, and 0.01 for column "Cu". The fourth row contains the text "eq" in the "H" column, and "tot" in the "PO4" and "Cu" columns. The "eq" cell is highlighted with a red circle. Below the table are two buttons: "CSV" and "xlsx". At the bottom, there is a section titled "Choose CSV File" with a "Browse..." button and a text box showing "No file selected".

	H	PO4	Cu
1	0.01	0.01	0.01
2	0.02	0.01	0.01
3	0.03	0.01	0.01
1	eq	tot	tot

Figure 6. How to introduce the equilibrium concentration of basis particle instead of total one

Then again press the button *Evaluate* and the equilibrium concentrations of basis particles and products will be calculated.

3.6 Option: Upload Data from CSV or Tab Delimited File(s)

All the data could be entered manually, however we recommend preparing the specific file (or files) containing all the required data.

TIP: Even if you input data manually you could save both input and output data in the file(s) so you could reproduce the research later or do minor modifications without retyping all the data from scratch.

CSV or tab delimited files are plain files with data delimited into columns with a specific delimiter (a.k.a. separator), mostly comma (,), semicolon (;) or tab (press *Tab* key in your preferred text editor to get the value).

These formats are very stable, platform independent and are used by a great range of applications and packages. There are at least four ways to format KEV input data as plain text files:

- Basic way: Type data using your preferred plain text editor such as Notepad on Windows
- Simplest way: Type or copy data to KEV, then download as CSV
- Dangerous way: Save from Excel using *Save as* option (could be some undesirable tricks)
- Advanced way: Use input from another app or package if format is the same as KEV requires

Please follow the recommendations below to format files in the right way.

TIP 1: Consider file names if you upload all data at once. If you upload file by file using *Choose CSV File* buttons there are no restrictions for file names

TIP 2: If you do not understand a word below use the Simplest way and consider only file names for the bulk input

3.6.1 Stoichiometric matrix

- File names allowed: *stoich_coefficients.csv*, *stoichiometric_coefficients.csv*, *stoich_coefficients*, *stoichiometric_coefficients*, *stoich_coefficients.txt*, *stoichiometric_coefficients.txt*. Prefix *input_* is also allowed. Remember: those names must be used if you upload all the files at once! If you upload them one by one, you are free to name the files whatever you want.
- First row: particle names
- Following rows: stoichiometric coefficients, one row for one reaction

3.6.2 Total (or equilibrium) concentrations of reagents

- File names allowed: *concentrations.csv*, *concentrations*, *concentrations.txt*. Prefix *input_* is also allowed
- First row: total (*tot*) or equal (*eq*) concentrations
- Second row: particle names
- Following rows: input concentrations, one row for one mixture

3.6.3 Decimal logarithms of equilibrium constants

- File names allowed: *k_constants_log10.csv*, *k_constants_log10*, *k_constants_log10.txt*. Prefix *input_* is also allowed
- First row: “k_constants_log10”
- Following rows: Decimal logarithms of equilibrium constants, one row for one reaction

Now when all files are prepared upload them to KEV.

Bulk upload (all files at once)

Select *Upload all data* in the section *Bulk upload / download (optional)* clicking on the *Browse...* data. Select all input data files for this project at once and click *Open*.

TIP: You can also load only some of the input data files e.g. *concentrations* and *k_constants_log10* and *not stoich_coefficients*. The missing data could be uploaded later (e.g. from another directory) or typed manually.

IMPORTANT: If input data requires some consistency between input files there could be some temporary errors if some data is already uploaded and some not. Don't panic! The errors to be gone when the full data is uploaded.

KEV: Chemistry Constant Evaluator

Calculate Equilibrium

Column delimiter of your data: ☐ , ☐ ; ☐ tab

Bulk upload / download (optional)

Upload all data

Choose CSV files or XLSX file with data

Browse...

Upload or type input data

Stoichiometric coefficients

	molecule1	molecule2	molecule3	molecule4
1	1	1	1	1
2	1	1	1	1
3	1	1	1	1
4	1	1	1	1

Choose CSV File

Browse... No file selected

K: lg constants

	k_constants_log10
1	1
2	1
3	1
4	1

Choose CSV File

Browse... No file selected

Concentrations

Input	Total
1	0.001
2	0.001
3	0.001
4	0.001
5	0.001
1	tot

Upload one file by another

Upload one file by another could be of interest if you need to correct only a part of the previously uploaded data or a small part of data is not yet uploaded at all.

Select *Choose CSV File* clicking on the *Browse...* button under the table you are going to fill. Let's begin e.g. from stoichiometric matrix. Remember, if you have chosen uploading one file by another there are no restrictions in file names – just

pick the properly formatted one. The strict names must be given to files if you upload them all simultaneously using *Bulk upload* option. See sections 3.6.1-3.6.3 for details.

Column delimiter of your data files

Particle to get fractions of

Bulk upload / download (optional)

Upload all data

Choose CSV files or XLSX file with multiple sheets

Browse... No file selected

Upload or type input data

Stoichiometric coefficients

	H	PO4
1	1	1
2	2	1
3	3	1
4	-1	0
5	1	1
6	2	1
7	2	2
8	4	2

Choose CSV File

Browse... StCoef.csv

Upload complete

Particle names, comma separated

H, PO4, Cu

Выборка файла

Имя файла: StCoef.csv

Открыть

Отмена

Choose CSV File

Browse... No file selected

Choose CSV File

Browse... No file selected

Then, let us upload the stoichiometric coefficients

Column delimiter of your data: ☐ tab

Bulk upload / download (optional)

Upload all data

Choose CSV files or XLSX file with multiple sheets

Browse... No file selected

Upload or type input data

Stoichiometric coefficients

	H	PO4	Cu
1	1	1	0
2	2	1	0
3	3	1	0
4	-1	0	1
5	1	1	1
6	2	1	1
7	2	2	1
8	4	2	1

Choose CSV File

Browse... StCoef.csv

Upload complete

Particle names, comma separated

H, PO4, Cu

Particle to get fractions of

molecule1

Download all data

zip xlsx

Concentrations

Input	Total
	H
1	0.001
2	0.001
3	0.001
4	0.001
5	0.001
1	tot
	tot

Choose CSV File

Browse... logK.csv

Upload complete

Choose CSV File

Browse... No file selected

Finally, here the concentrations go!

Column delimiter of your data files: ☐ tab

Bulk upload / download (optional)

Upload all data

Choose CSV files or XLSX file with multiple sheets

Browse... No file selected

Upload or type input data

Stoichiometric coefficients

	H	PO4	Cu
1	1	1	0
2	2	1	0
3	3	1	0
4	-1	0	1
5	1	1	1
6	2	1	1
7	2	2	1
8	4	2	1

Choose CSV File

Browse... StCoef.csv

Upload complete

Particle names, comma separated

H, PO4, Cu

Particle to get fractions of

molecule1

Concentrations

Input	Total
	H
1	1.00E-02
2	2.00E-02
3	3.00E-02
1	tot
	tot

Choose CSV File

Browse... Conc.csv

Upload complete

Create CSV files with KEV

- **Fill data in KEV and download.** If you are experiencing some problems with creating the input data files fill the data in KEV interface and download as a zip archive containing CSV files. Unzip archive and you can use them with the *Bulk upload* option next time.

Calculate Equilibrium Concentrations

Column delimiter of your data files
☐ , ☒ ; ☐ tab

Particle to get fractions of
 molecule1

Bulk upload / download (or)

Upload all data
 Choose CSV files or XLSX files
 Browse... 3 files

Download all data

Upload or type input data

Stoichiometric coefficients

	molecule1	molecule2	molecule3
1	1	1	0
2	0	1	1
3	0	2	1

K: lg constants

	k_constants_log10
1	3,14
2	1,45
3	2,79

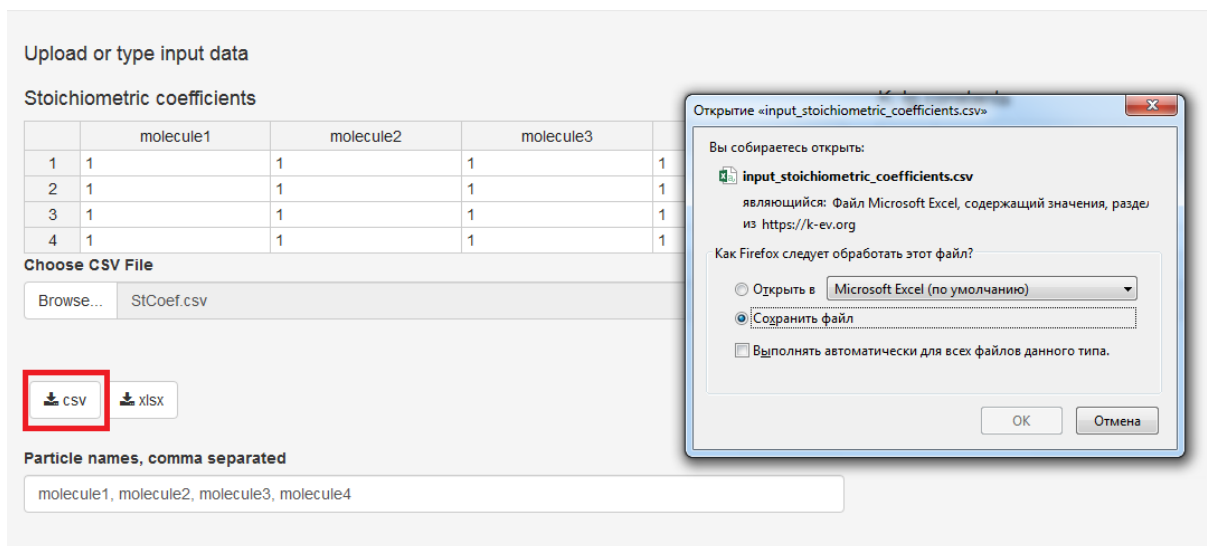
Concentrations

	Input	Total
		molecule1
1		1,00E-02
2		1,00E-03

- **Download, fill on your desktop and upload.** Or you can download example files (i.e. do not fill anything before downloading) and fill them with data on your desktop.

A1	:	X	✓	f _x	molecule1
	A	B	C	D	E
1	molecule1	molecule2	molecule3	molecule4	
2	1	1	1	1	
3	1	1	1	1	
4	1	1	1	1	
5	1	1	1	1	
6					
7					
8					
9					
10					

- **Download files one by one.** Or you can download, fill and upload files one by one if you prefer. Modify this blank file for stoichiometric coefficients; then save it in your preferable format, and all is done! We believe there is no need to repeat the same for other tables. The procedure is very similar: download, modify, save, upload.



3.7 Option: Upload Data from XLSX Single File

KEV supports uploading and downloading Excel files in *xlsx* format.

IMPORTANT: *xls*, *xlsm* and *xlsb* are not supported yet. Convert them to *xlsx*, it is easy with Excel 2007 or newer.

The properly formatted file of the example system considered above could be downloaded via https://gitlab.com/a.meshkov/KEV/tree/double/input/concentrations/ds.3p.long/example_eq_conc_calc.xlsx

The file consists of four sheets

- the sheets *stoich_coefficients*, *concentrations* and *k_constants_log10* cover KEV Tables
- the sheet *particle_names* contains the name of *particle to get fractions of* (Fig. 7).

IMPORTANT: Do not rename the sheets.

The input data should be written into file as it is shown for stoichiometric matrix (Fig. 7), concentrations, equilibrium constants logarithms and particle name (Fig. 8).

3.7.1 Stoichiometric matrix

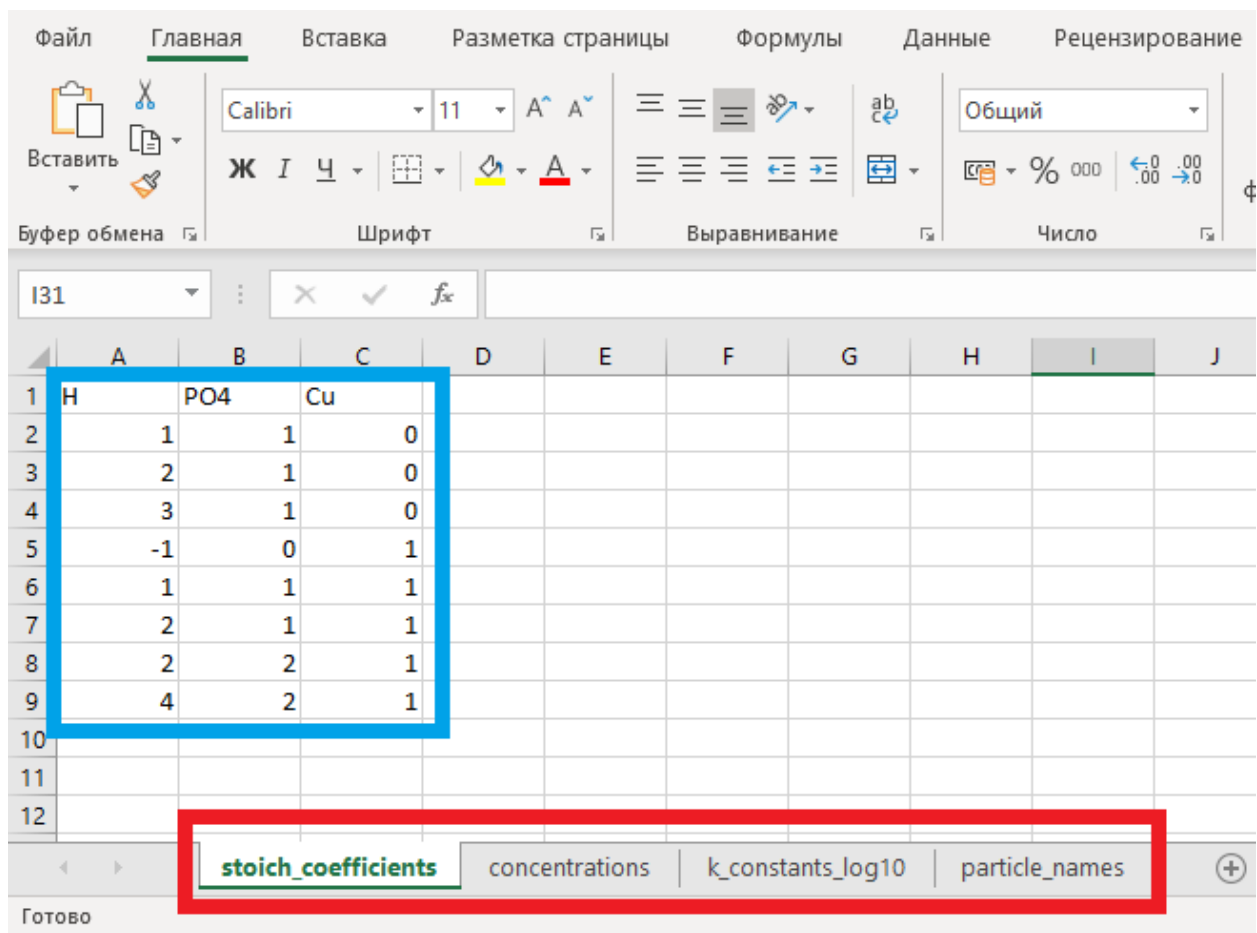


Figure 7. Input data file tabs (marked with red), and stoichiometric matrix (marked with blue)

3.7.2 Total (or equilibrium) concentrations of reagents

	A	B	C
1	tot	tot	tot
2	H	PO4	Cu
3	1.000E-02	1.000E-02	1.000E-02
4	2.000E-02	1.000E-02	1.000E-02
5	3.000E-02	1.000E-02	1.000E-02

Figure 8. Input data file – concentrations

3.7.3 Decimal logarithms of equilibrium constants

	A	B
1	k_constants_log10	
2	11.68	
3	18.43	
4	20.23	
5	-8.12	
6	14.98	
7	20.03	
8	28.06	
9	37.86	
10		

Figure 9. Input data file – equilibrium constants logarithms

3.7.4 Particle to get fractions of

	A	B	C
1	PO4		
2			
3			
4			
5			
6			

Figure 10. Input data file – particle name

When the xlsx data file is done upload it into KEV the same way as the bulk upload with CSV files i.e. select *Upload all data* in the *Bulk upload / download (optional)* section (Fig. 11).

Calculate Equilibrium Concentrations

The screenshot shows the KEV web interface for calculating equilibrium concentrations. A file upload dialog is open, showing the selection of 'example_eq_conc_calc.xlsx'. The interface includes sections for 'Bulk upload / download (optional)', 'Upload or type input data', and 'Concentrations'.

Bulk upload / download (optional)

Upload all data

Choose CSV files or XLSX file with multiple sheets

Browse... No file selected

Upload or type input data

Stoichiometric coefficients

	molecule1	molecule2	molecule3	molecule4
1	1	1	1	1
2	1	1	1	1
3	1	1	1	1
4	1	1	1	1

Choose CSV File

Browse... StCoef.csv

K: lg constants

	k_constants_log10
1	1
2	1
3	1
4	1

Choose CSV File

Browse... logK.csv

Concentrations

Input	Total	molecule1	molecule2	molecule3	molecule4
1	0.001	0.001	0.001	0.001	0.001
2	0.001	0.001	0.001	0.001	0.001
3	0.001	0.001	0.001	0.001	0.001
4	0.001	0.001	0.001	0.001	0.001
5	0.001	0.001	0.001	0.001	0.001
1	tot	tot	tot	tot	tot

Choose CSV File

CSV XLSX

Figure 11. *How to upload the single .xlsx file containing all the data*

3.8 Download Results of Calculations

Just click *Download all data* in the *Bulk upload / download (optional)* section and choose your preferred format (Fig. 12):

- zip: zip archive containing CSV files
- xlsx: single Excel file with multiple sheets

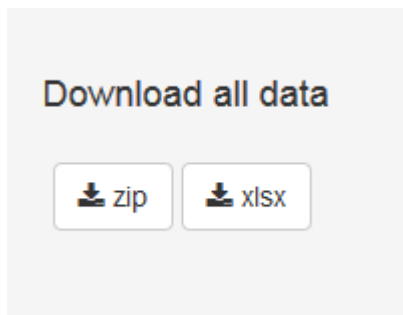
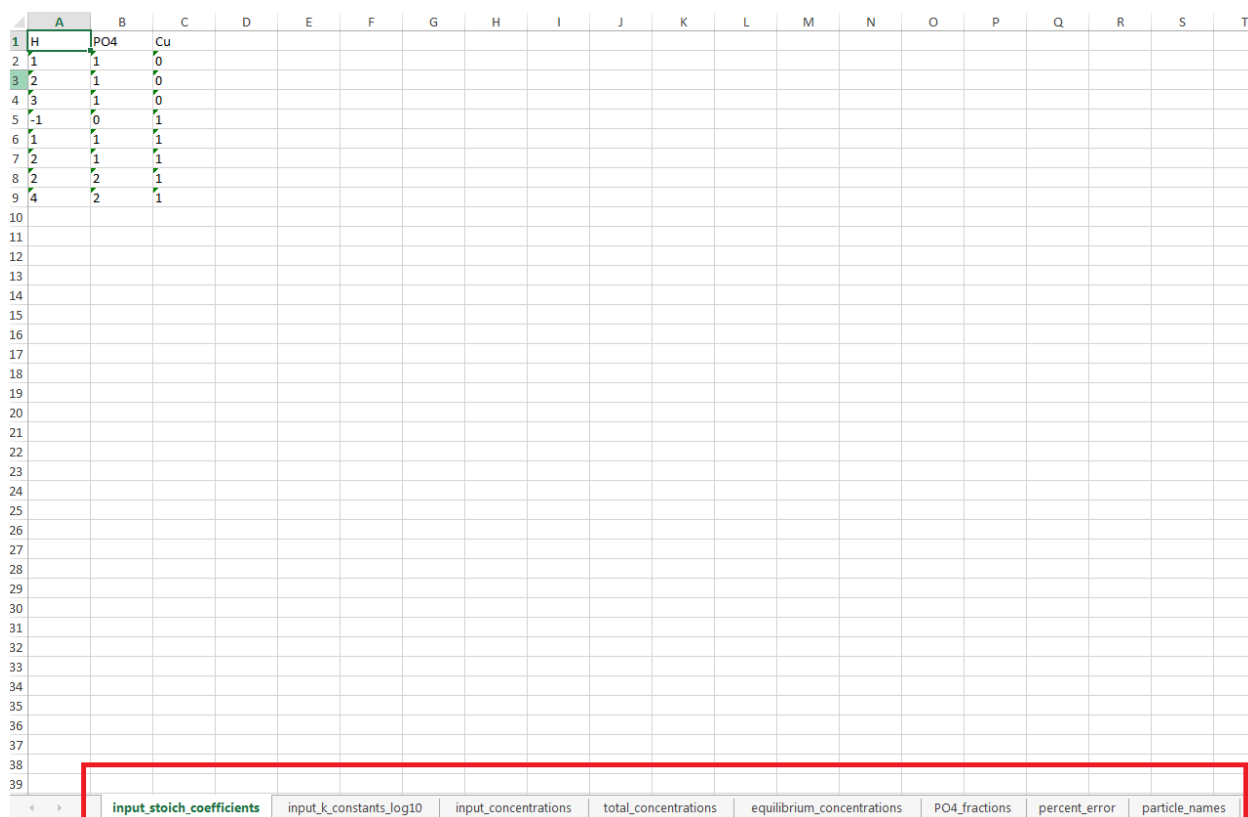


Figure 12. *Download the results together with input data*

3.8.1 Single xlsx file

The resulting .xlsx file contains all the input data and results in separate sheets (Fig. 13).

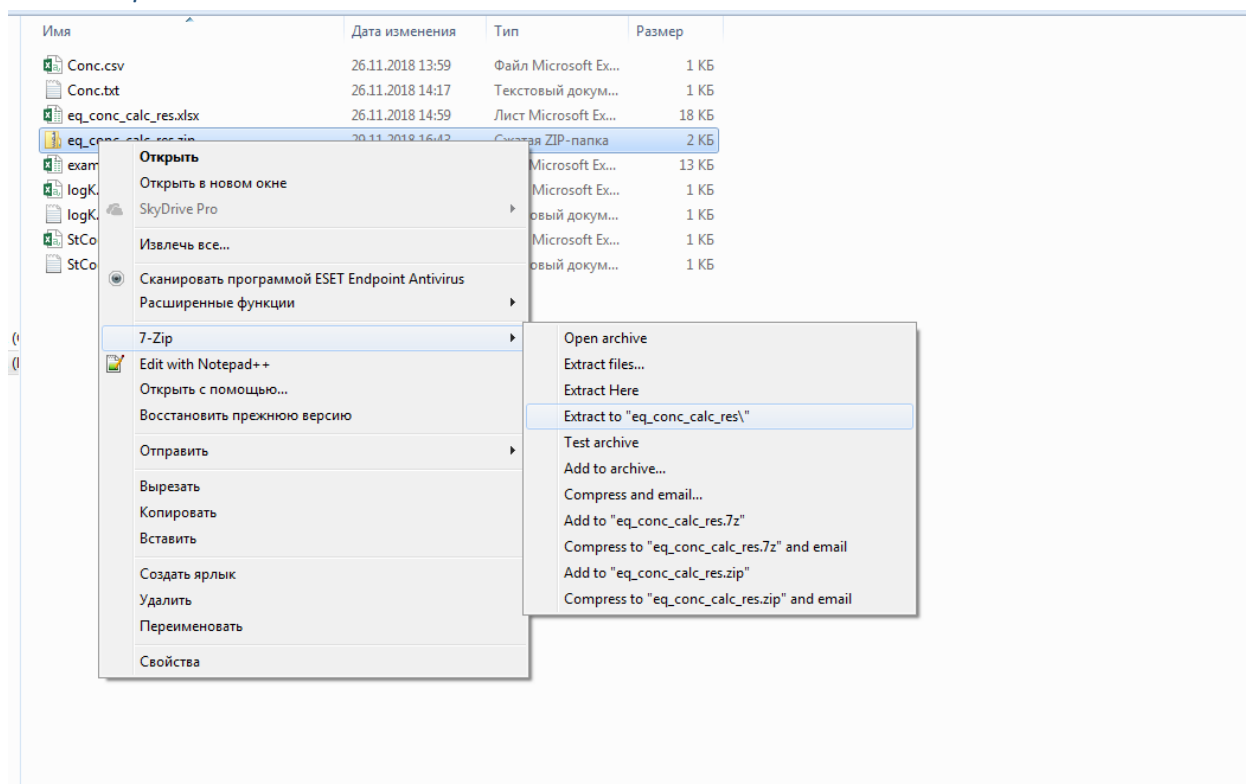


	A	B	C	D	E	F	G	H	I	J	K	L	M	N	O	P	Q	R	S	T
1	H	PO4	Cu																	
2	1	1	0																	
3	2	1	0																	
4	3	1	0																	
5	-1	0	1																	
6	1	1	1																	
7	2	1	1																	
8	2	2	1																	
9	4	2	1																	
10																				
11																				
12																				
13																				
14																				
15																				
16																				
17																				
18																				
19																				
20																				
21																				
22																				
23																				
24																				
25																				
26																				
27																				
28																				
29																				
30																				
31																				
32																				
33																				
34																				
35																				
36																				
37																				
38																				
39																				

At the bottom of the spreadsheet, there is a red box highlighting the sheet names: **input_stoich_coefficients**, input_k_constants_log10, input_concentrations, total_concentrations, equilibrium_concentrations, PO4_fractions, percent_error, and particle_names.

Figure 13. *Resulting .xlsx file*

3.8.2 Zip archive



Archive unpacked contains following files:

Имя	Дата изменения	Тип	Размер
equilibrium_concentrations.csv	29.11.2018 13:43	Файл Microsoft Ex...	1 КБ
input_concentrations.csv	29.11.2018 13:43	Файл Microsoft Ex...	1 КБ
input_k_constants_log10.csv	29.11.2018 13:43	Файл Microsoft Ex...	1 КБ
input_stoichiometric_coefficients.csv	29.11.2018 13:43	Файл Microsoft Ex...	1 КБ
particle_names.csv	29.11.2018 13:43	Файл Microsoft Ex...	1 КБ
percent_error.csv	29.11.2018 13:43	Файл Microsoft Ex...	1 КБ
PO4_fractions.csv	29.11.2018 13:43	Файл Microsoft Ex...	1 КБ
total_concentrations.csv	29.11.2018 13:43	Файл Microsoft Ex...	1 КБ

References to Section III

1. K. J. Powell et. al. *Pure&Applied Chemistry* **2005**, 77(4), 739-800, doi: 10.1351/pac200577040739
2. K. J. Powell et. al. *Pure&Applied Chemistry* **2007**, 79(5), 895-950, doi: 10.1351/pac200779050895

IV. EQUILIBRIUM CONSTANTS EVALUATION FROM UV-Vis DATA

4.1 Select the Problem

Select *Spectrophotometry* in the top menu to open the tab.

Let us consider the reaction of hydrazone formation between pyridoxal 5'-phosphate (PLP) and 2-methyl-3-furoylcarbohydrazide (2MF3H) at pH 7.0 [1]:



The example data could be obtained:

- Input Data
 - CSV comma separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/spectrophotometry/dsl.9/csv.comma>
 - CSV semicolon separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/spectrophotometry/dsl.9/csv.semicolon>
 - Tab separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/spectrophotometry/dsl.9/txt.tab>
 - Excel (xlsx) file
https://gitlab.com/a.meshkov/KEV/tree/double/input/spectrophotometry/dsl.9/UV-Vis_input_example.xlsx
- Output (calculated) data
 - Zip archive with CSV files
https://gitlab.com/a.meshkov/KEV/tree/double/output/spectrophotometry/dsl.9/gui/UV_Vis_results.zip
 - Excel (xlsx) file
https://gitlab.com/a.meshkov/KEV/tree/double/output/spectrophotometry/dsl.9/gui/UV_Vis_results.xlsx

4.2 Data Description and Manual Input

In order to evaluate the equilibrium constant(s) from UV-Vis experimental results KEV requires the following data:

- Tables (matrices and vectors)
 - Stoichiometric matrix of reactions (a.k.a. Stioichiometric coefficients)
 - Decimal logarithms of equilibrium constants
 - Total (or equilibrium) concentrations of reagents

- Absorbance values at certain wavelengths with their experimental deviations
- Molar extinction coefficients at the *very same* wavelengths as absorbance (optional)
- Text input (Fig. 16)
 - The name of constant(s) to be evaluated
 - Threshold (search algorithm precision)
 - Search density (default to 1)

The initial three tables are the same as in the case of calculation of the equilibrium composition. See Sections 3.2.1-3.2.3; 3.6-3.7 for details.

However, the additional information input is also in order. It includes absorbance values with the errors of their measuring and molar extinction coefficients (Fig. 15).

	data	wavelength	S1	S2	S3	S4	S5
1	observation	110	100	100	100	100	100
2	observation	120	100	100	100	100	100
3	observation	130	100	100	100	100	100
4	deviation	110	0.001	0.001	0.001	0.001	0.001
5	deviation	120	0.001	0.001	0.001	0.001	0.001
6	deviation	130	0.001	0.001	0.001	0.001	0.001

	wavelength	molecule1	molecule2
1	110	0	0
2	120	0	0
3	130	0	0

Figure 15. Screenshot of KEV interface. Experimental absorbances table is marked with red, molar extinction coefficients table is marked with blue

Constants to evaluate
Particle names, comma separated
molecule1

Threshold
Search algorithm precision $0 < \zeta < 1$
1e-7

Search density
Do not change unless you fully understand what you are doing
1

Figure 16. Additional information required for the calculations

4.2.1 Stoichiometric matrix (Stoichiometric coefficients)

For the example given in Section 4.1 (reaction of hydrazone formation between pyridoxal 5'-phosphate (PLP) and 2-methyl-3-furoylcarbohydrazide (2MF3H) at pH 7.0) there are two basis particle, 2MF3H and PLP, and the only reaction. Therefore, the stoichiometric matrix is following:

1 1

Is that all? Not yet: there is also an additional column *name*. It serves for naming the products of reactions. Let us name the product SB (Schiff Base). Then, the matrix looks like that:

1 1 SB

Type them into KEV table *Stoichiometric coefficients*. Adding or removing columns/rows could be made with the context menu (mouse right click).

TIP: If you do not want to type numbers cell by cell (we don't), copy numbers into a blank Excel file, split data into columns and copy + paste it from the Excel file to the KEV table *Stoichiometric coefficients*

4.2.2 Decimal logarithms of equilibrium constants

There is the only reaction, so, the vector of $\lg K_{\text{total}}$ contains the only number:

5

Type it into KEV table *K: lg constants* preserving the order or copy + paste as for the stoichiometric matrix data.

IMPORTANT: Please note the decimal logarithms of equilibrium constants ($\lg K$) should be input and not the K values

4.2.3 Total (or equilibrium) concentrations of reagents

Let us also give the total concentrations for three mixtures.

They form another matrix (particles as columns, the first one is 2MF3H, the second one is PLP, and mixtures as rows):

0.0000144 0.0001486

0.0000288 0.0001486

0.0000432 0.0001486

0.0000576 0.0001486

0.0000720 0.0001486

0.0000864 0.0001486

0.0001008 0.0001486

0.0001152 0.0001486

0.0001296 0.0001486

0.0001440 0.0001486

Type or copy + paste it to KEV table *Concentrations >> Input*

The footnote of the Table "Concentrations" contains a row of variables capable of taking two values: *tot* for total concentrations, and *eq* for equilibrium concentrations (if some input concentrations are equilibrium ones). However, in the case of constants calculations *eq* value would hardly appear anywhere.

IMPORTANT: Check if the column number in *Stoichiometric coefficients* and *Concentrations* tables is equal. If one table has empty or excess columns remove them with the context menu (mouse right click)

After input of these data, it should look like that (Fig. 17):

The screenshot shows a web interface for data input. It has three main sections for data entry:

- Stoichiometric coefficients:** A table with columns for particle names (2MF3H, PLP) and their coefficients (1, 1). Below it is a 'Choose CSV File' button and a 'Browse...' button for 'stoichiometric_coefficient.csv'.
- K: lg constants:** A table with columns for particle names (2MF3H, PLP) and their log constants (1, 5). Below it is a 'Choose CSV File' button and a 'Browse...' button for 'log10.csv'.
- Concentrations:** A table with columns for particle names (2MF3H, PLP) and their concentrations (1, 1.44e-05, 2.88e-05, etc.). Below it is a 'Choose CSV File' button and a 'Browse...' button for 'Concentrations.csv'.

At the bottom, there is a field for 'Particle names, comma separated' containing the text '2MF3H, PLP'.

Figure 17. *Stoichiometric coefficients, log K and concentrations input data for example (17)*

4.2.4 Particle names

Columns in both *Stoichiometric coefficients* and *Concentrations* tables have the same names which are the names of the base particles. By default KEV uses particle names *molecule1*, *molecule2*, *molecule3* etc. If you want to change them (and you'd better do) type new base particle names into the *Particle names, comma separated* field.

2MF3H, PLP

IMPORTANT: Make sure the quantity of reagents in the string *Particle names, comma separated* corresponds to the quantity of columns of Table *Stoichiometric coefficients*. Delete or add columns if required

4.2.5 Absorbance and deviations

The *Absorbance and deviations* table consists of following columns:

- *data*. The values in column *data* can take *observation* or *deviation* value. *observation* is your experimental value; the *deviation* is its experimental error of determination.
- *wavelength*. We recommend typing there the real wavelengths, for example, 297, 337, and 388 nm. However, you can mark them by some symbols you desire – just not get entangled later!
- Columns with experimental values and experimental deviations for every mixture (solution). They could be named *S1*, *S2*, *S3*... *SK* where *K* is the number of solutions used in calculations. In case of our example there are 9 solutions.

IMPORTANT: The *wavelength* cells for *deviations* must contain the same designations of wavelengths you have given to experimental values. If you marked the latter as 297, 337, 388, the deviations should be given also for 297, 337, 388 wavelengths.

The following table should be applied for the example (17):

observation 297	0.383	0.664	0.948	1.240	1.500	1.729	1.989	2.213	2.361
observation 337	0.432	0.54	0.64	0.744	0.831	0.913	1.009	1.089	1.145
observation 388	0.714	0.679	0.624	0.574	0.514	0.48	0.439	0.403	0.374
deviation 297	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
deviation 337	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
deviation 388	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003

The upper three rows are experimental values of absorbance at 297, 337, 388 wavelengths. The lower three rows are the deviations at the same 297, 337, 388 wavelengths.

4.2.6 Molar extinction coefficients

It is strongly recommended to determine molar extinction coefficients for, at least, some of the reagents or products prior to the main experiment. Those pre-determined values help greatly in making the constant calculation results more reliable and valid.

The *Molar Extinction coefficients* table contains following columns:

- *wavelength*. We recommend typing there the real wavelengths, for example, 297, 337, and 388 nm. The values should correspond to *Absorbance and deviations* wavelengths.
- Columns with molar extinctions coefficients for particles with known molar extinction coefficients (both reagents and products). Name them with particle names, e.g. *2MF3H*, *PLP*. There is no need in specifying somewhere, which particles are with unknown molar extinction coefficients – just do not add/remove the corresponding column from this table, and KEV does the rest.

IMPORTANT: *wavelength* and particle names should correspond to *Absorbance and deviations* wavelengths and *Stoichiometric coefficients* reagent and product names respectively. I.e. if you have 297, 337 and 388 nm wavelengths in Absorbance and deviations table do not include 403 nm wavelength in Molar Extinction coefficients table.

IMPORTANT: All the known molar extinction coefficients values should be specified in this table (even if all of them are zeros). KEV treats the reagents or products unspecified in this table as ones with unknown molar extinction coefficients, which should be determined.

Making preliminary calibration plots experiments, we determined the molar extinction coefficients for 2MF3H and PLP at 297, 337, 388 nm:

0 888

0 2252

0 5078

Type or copy + paste it to KEV table

The absorbance and molar extinction coefficients tables look now as follows (Fig. 18).

Absorbance and deviations										
	data	wavelength	1	2	3	4	5	6	7	8
1	observation	297	0.383	0.664	0.948	1.24	1.5	1.729	1.989	2.213
2	observation	337	0.432	0.54	0.64	0.744	0.831	0.913	1.009	1.089
3	observation	388	0.714	0.679	0.624	0.574	0.514	0.48	0.439	0.403
4	deviation	297	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
5	deviation	337	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
6	deviation	388	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003

Molar extinction coefficients			
Peaks (up to 10)			
297, 337, 388			
	wavelength	2MF3H	PLP
1	297	0	888
2	337	0	2252
3	388	0	5078

Figure 18. The data on absorbance and molar extinction coefficients for the reaction (17) at wavelengths values of 297, 337, 388 nm

4.2.7 Constants to evaluate and wavelengths to use

Type one or more product names into *Constants to evaluate* field (Fig. 16). Make sure they correspond to the ones specified in the *Stoichiometric coefficients* table.

The wavelengths KEV uses for calculations should be specified in the *Molar extinction coefficients* field right above molar extinction coefficients table in the *Peaks (up to 10)* field. (Fig. 18). If dozens of molar extinction coefficients are known it is a good idea to restrict number of wavelengths used in the evaluation process to peaks only. If none of peaks are specified the first 10 wavelengths to be used.

4.2.8 Threshold

The threshold parameter (a.k.a. ζ , spells *zeta* or *dzeta*) defines the search algorithm precision, default to $1e-7$. The threshold should be $1 > \zeta > 0$.

- Increase it to speed up the algorithm (e.g. to $1e-4$)
- Decrease it to increase accuracy (e.g. to $1e-8$)
- However, in most cases the default value is the best choice

4.2.8 Search density

The search density parameter defines how many steps the algorithm performs in every direction on each step. The default value is 1 as in the classic direct search algorithm.

Increasing the search density could help in complex systems with many local minimums. E.g. try increase it to 2. Now on every step the algorithm checks not 1 possible value in each direction but 2: one equal to *learning rate * initial step* and the second one equal to *learning rate * initial step * 2*. I.e. it performs as you have started with two different learning rates simultaneously (or from different initial values).

The search density should be integer > 0. Do not use values > 3 for it slows down the algorithm too much and increases the risk of the computation error.

4.3 Evaluate

After the input data is provided check if all is OK. If it is not you will get a bunch of red and grey error messages. Most frequent error causes and fixes are:

- Check *Column delimiter* control. If you use csv file with semicolon ";" delimiter should be also ";"
- Check if particle (molecule) names are consistent between different tables i.e. coefficient, concentrations and molar extinction matrices
- Check the same for wavelenghts
- If you are uploading data table by table just try to load remaining data, the "error" could be a temporary thing disappearing after all the data is provided

In order to calculate the equilibrium concentrations of the reagents and products click *Evaluate* button right under the input data tables.

4.4 Output (Calculated) Data

The results will appear below a few moments later (Fig. 19):

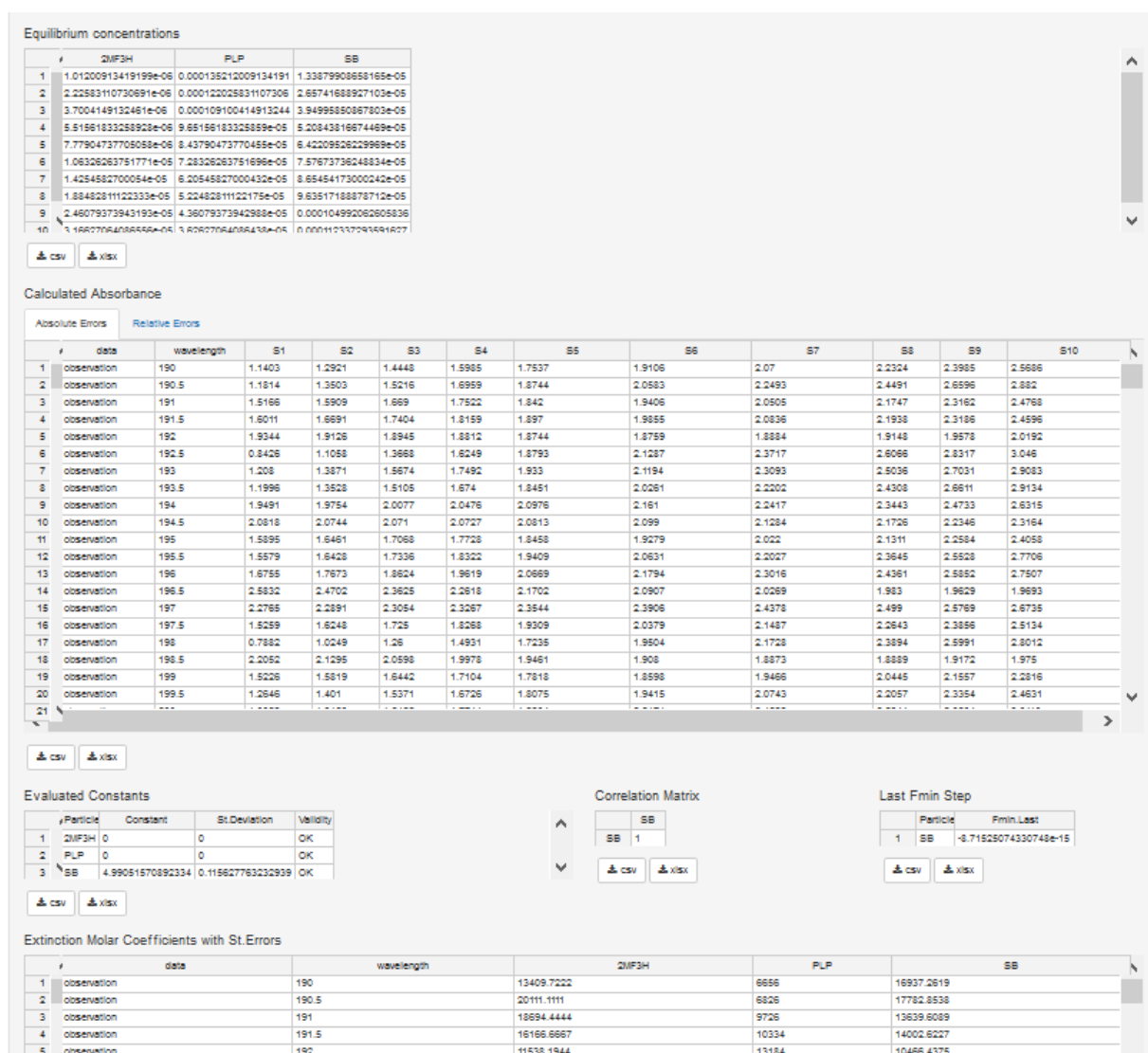


Figure 19. Results of constant evaluation for example (17)

- *Equilibrium concentrations*: all the equilibrium concentrations for basis particles and products
- *Calculated Absorbance*: the sum of superposition of input (calculated) molar extinction coefficients on the equilibrium concentrations of light-absorbing compounds.
- *Evaluated Constants*: the input and evaluated equilibrium constants. The standard deviations of evaluated constants are also specified there. *Validity* column takes one of the following values
 - *OK*: everything seems right
 - *Non-Sensitive*: varying the evaluated log K value does not alter the minimizing function
 - *-Inf* or *Inf*: no global minima of minimizing functions were found

- *Insignificant*: the evaluated value of log K has a standard deviation exceeding 10% of log K value
- *Correlation matrix*: the matrix, which diagonal elements are equal to 1 while non-diagonal ones are the Pearson correlation coefficients between each pair of optimized parameters. Since the only constant was evaluated in this example, the correlation matrix consists of the only element equal to 1.
- *Last F_{min} Step*: the last alteration of minimizing function under variation of evaluated log K. A useful information, if something goes wrong.
- *Extinction Molar Coefficients with St. Errors*: the input or calculated molar extinction coefficients. For the latter, the deviation of their determination by the least squares method is specified.

TIP: Pay attention to the *Validity* comments. In case of any not-OK signal appearance, try doing the following:

1. Check typos and errors in the input data. the table *Calculated Absorbance* with errors could give a clue, where the blunder had been made
2. Change the initial approximated value(s) of log K.
3. Increase the value of threshold (Fig. 16)
4. Increase the search density. The value should be a positive integer
5. Remember, that Insignificant comment is rather subjective. For example, the constant value of 0.8 ± 0.1 might be accurate enough
6. If none of these helped, the experimental data might be insufficient or experimental design is unsuccessful

For example, let us imagine that the error has been made in the 7th solution at wavelength of 337, and 0.109 was input instead of correct 1.009. The high standard deviation indicates something went wrong, and in the appropriate cell of table of calculated absorbance we can see high absolute/relative error (Figure 20).

Calculated Absorbance

Absolute Errors

Relative Errors

	data	wavelength	S1	S2	S3	S4	S5	S6	S7	S8	S9	S10
1	observation	297	0.4128	0.6898	0.9619	1.2274	1.4842	1.7292	1.9586	2.1677	2.3516	2.5071
2	observation	337	0.4181	0.5004	0.5813	0.6602	0.7365	0.8093	0.8775	0.9396	0.9943	1.0405
3	observation	388	0.7067	0.6594	0.613	0.5677	0.5239	0.4821	0.443	0.4074	0.376	0.3494
4	error	297	0.0298	0.0258	0.0139	-0.0126	-0.0158	0.0002	-0.0304	-0.0453	-0.0094	0.0711
5	error	337	-0.0139	-0.0396	-0.0587	-0.0838	-0.0945	-0.1037	0.7685	-0.1494	-0.1507	-0.1285
6	error	388	-0.0073	-0.0196	-0.011	-0.0063	0.0099	0.0021	0.004	0.0044	0.002	-0.0036

csv

xlsx

Evaluated Constants

Particle	Constant	St.Deviation	Validity
1	2MF3H 0	0	OK
2	PLP 0	0	OK
3	SB 5.0470507144928	0.8690006620703	Insignificant

csv

xlsx

Correlation Matrix

SB
SB 1

csv

xlsx

Last Fmin Step

Particle	Fmin Last
1 SB	-1.4432899320127e-15

csv

xlsx

Extinction Molar Coefficients with St.Errors

	data	wavelength	2MF3H	PLP	SB
1	observation	297	0	888	21685.5754
2	observation	337	0	2252	8432.5391
3	observation	388	0	5078	1530.4663
4	error	297	0	0	141.4187
5	error	337	0	0	1153.4754
6	error	388	0	0	38.2143

Figure 20. If something goes wrong, look for input errors and typos

4.5 Option: Upload Data from CSV or Tab Delimited File(s)

All the data could be entered manually, however we recommend preparing the specific file (or files) containing all the required data.

TIP: Even if you input data manually you could save both input and output data in the file(s) so you could reproduce the research later or do minor modifications without retyping all the data from scratch.

CSV or tab delimited files are plain files with data delimited into columns with a specific delimiter (a.k.a. separator), mostly comma (,), semicolon (;) or tab (press *Tab* key in your preferred text editor to get the value).

These formats are very stable, platform independent and are used by a great range of applications and packages. There are at least four ways to format KEV input data as plain text files:

- Basic way: Type data using your preferred plain text editor such as Notepad on Windows
- Simplest way: Type or copy data to KEV, then download as CSV
- Dangerous way: Save from Excel using *Save as* option (could be some undesirable tricks)
- Advanced way: Use input from another app or package if format is the same as KEV requires

Please follow the recommendations below to format files in the right way.

TIP 1: Consider file names if you upload all data at once. If you upload file by file using *Choose CSV File* buttons there are no restrictions for file names

TIP 2: If you do not understand a word below use the Simplest way and consider only file names for the bulk input

4.5.1 Stoichiometric matrix

- File names allowed: *stoich_coefficients.csv*, *stoichiometric_coefficients.csv*, *stoich_coefficients*, *stoichiometric_coefficients*, *stoich_coefficients.txt*, *stoichiometric_coefficients.txt*. Prefix *input_* is also allowed. Remember: those names must be used if you upload all the files at once! If you upload them one by one, you are free to name the files whatever you want.
- First row: particle names, “name”
- Following rows: stoichiometric coefficients, one row for one reaction, and the name of reaction product.

4.5.2 Total (or equilibrium) concentrations of reagents

- File names allowed: *concentrations.csv*, *concentrations*, *concentrations.txt*. Prefix *input_* is also allowed
- First row: total (*tot*) or equal (*eq*) concentrations
- Second row: particle names
- Following rows: input concentrations, one row for one mixture

4.5.3 Decimal logarithms of equilibrium constants

- File names allowed: *k_constants_log10.csv*, *k_constants_log10*, *k_constants_log10.txt*. Prefix *input_* is also allowed
- First row: “k_constants_log10”
- Following rows: Decimal logarithms of equilibrium constants, one row for one reaction

4.5.4 Absorbance

- File names allowed: *absorbance.csv*, *absorbance*, *absorbance.txt*. Prefix *input_* is also allowed. Remember: those names must be used if you upload all the files at once! If you upload them one by one, you are free to name the files whatever you want.
- First row: “data”, “wavelength”, solution numbers, e.g. “1”, “2”, “3”, etc.
- Following rows
 - Column *data*: either “observation” or “deviation”
 - Column *wavelength*: the wavelength identifier (e.g., “322”)
 - Following columns: experimental absorbance values or their deviations depending on what was specified in the first cell. The table must contain as many “observation” lines as “deviation” rows so every “observation” line corresponds to a “deviation” line with the same wavelength.

4.5.5 Molar extinction coefficients

- File names allowed: *molar_extinction_coefficients.csv*, *mol_ext_coefficients.csv*, *molar_extinction_coefficients*, *mol_ext_coefficients*, *molar_extinction_coefficients.txt*, *mol_ext_coefficients.txt*. Prefix *input_* is also allowed. Remember: those names must be used if you upload all the files at once! If you upload them one by one, you are free to name the files whatever you want.
- First row: “wavelength”, particle names
- Following rows
 - Column *wavelength*: the wavelength identifier (e.g. “322”)
 - Following columns: molar extinction coefficients

4.5.6 Target

- File names allowed: *constants_names.csv*, *target.csv*, *constants_names*, *target*, *constants_names.txt*, *target.txt*. Remember: those names must be used if you upload all the files at once! If you upload them one by one, you are free to name the files whatever you want.
- First row: “constant”, product name (it must correspond to one of the names given in stoichiometric matrix)
- Second row: “wavelength”, wavelength name(s) (e.g. “322”). Optional. If *wavelength* row is missing the first 10 provided wavelengths to be used in calculation.

4.6 Option: Upload Data from CSV or Tab Delimited File(s)

Now when all files are prepared upload them to KEV.

See Section 3.6 for details. The uploading (all files at one time, one file by another) is pretty the same as in the case of equilibrium concentrations calculations.

4.7 Option: Upload Data from XLSX Single File

The first three sheets of this single .xlsx file are the same as for equilibrium composition problem (Section 3.7):

- *stoich_coefficients*
- *concentrations*
- *k_constants_log10*

New sheets are:

- *absorbance*
- *mol_ext_coefficients*
- *target*

IMPORTANT: Do not rename the sheets

	A	B	C	D	E	F	G	H	I	J	K	L
1	2MF3H	PLP	name									
2	1		1 SB									
3												
4												
5												
6												
7												
8												
9												
10												
11												
12												

Sheet tabs: **stoich_coefficients** | k_constants_log10 | concentrations | absorbance | mol_ext_coefficients | target

Figure 21. Sheets in the input .xlsx file

4.7.1. absorbance

absorbance sheet contains absorbance data. Type peaks data only or paste the whole spectra. Have a look how should they be formatted (Fig. 22). Do not forget the table of deviations should follow the observation (Fig. 23)

IMPORTANT: Do not forget specifying the deviations after experimental values in the same tab

	A	B	C	D	E	F	G	H	I	J	K	L
1	data	wavelength	1	2	3	4	5	6	7	8	9	10
2	observation	190	1.462	0.672	3.494	2.881	1.759	1.209	3.709	1.287	1.613	2.047
3	observation	190.5	1.812	2.643	2.249	2.103	1.709	1.116	2.089	3.009	2.646	2.038
4	observation	191	1.899	1.394	1.112	2.331	3.148	2.597	0.942	2.123	1.527	2.852
5	observation	191.5	1.678	2.039	2.283	1.779	3.1	Number of sample	1.299	1.987	1.781	2.879
6	observation	192	1.333	wavelengths		1.573	2.1	2.093	2.3	Absorbance values		1.195
7	observation	192.5	1.153	1.354	2.568	2.025	2.105	2.688	2.626	2.1		2.18
8	observation	193	1.336	1.507	1.423	1.213	2.594	2.775	1.789	2.2		.556
9	observation	193.5	2.326	1.74	1.993	1.907	2.626	1.797	1.718	2.038	3.227	2.313

Figure 22. Formatting the absorbance tab with full spectra of mixtures 2MF3H + PLP

612	observation	495	0.002	0.004	-0.001	0	0	0	0	0	0	0
613	observation	495.5	0.002	0.004	-0.001	0	0	0	0	0	0	0
614	observation	496	0.002	0.004	-0.001	0	0	0	0	0	0	0
615	observation	496.5	0.002	0.003	-0.001	0	0	0	0	0	0	0
616	observation	497	0.002	0.003	-0.001	-0.001	0	-0.001	-0.001	-0.001	0	0
617	observation	497.5	0.001	0.003	-0.001	-0.001	0	-0.001	-0.001	0	0	0
618	observation	498	0.001	0.003	-0.001	-0.001	0	-0.001	-0.001	-0.001	0	-0.001
619	observation	498.5	Deviations must follow the experimental values!							-0.001	-0.001	-0.001
620	observation	499	0.001	0.003	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001
621	observation	499.5	0.001	0.003	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001
622	observation	500	0.001	0.003	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001	-0.001
623	deviation	190	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
624	deviation	190.5	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
625	deviation	191	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
626	deviation	191.5	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
627	deviation	192	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
628	deviation	192.5	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
629	deviation	193	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
630	deviation	193.5	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
631	deviation	194	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
632	deviation	194.5	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003
633	deviation	195	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003	0.003

Figure 23. Yep, those are important too

4.7.2. mol_ext_coefficients

Type known molar extinction coefficients for particles and wavelengths or calculate them using KEV's *Extinction Coefficient* tab (see the next Section 4.8).

	A	B	C	D	E	F	G	H	I	J	K	L
1	wavelength	2MF3H	2MF3H_ac	PLP	PLP_adj.r.squared							
2	190	13409,72	-0,17431	6656	0,419347							
3	190.5	20111,11	0,695758	6826	0,233609							
4	191	18694,44	0,948224	9726	0,39325							
5	191.5	16166,67	0,959953	10334	0,941263							
6	192	11538,19	0,1432	13184	0,750931							
7	192.5	10611,11	-0,11839	3888	0,31573							
8	193	16541,67	0,549601	6930	0,767526							
9	193.5	26166,67	0,896377	7064	0,865942							
10	194	22506,94	0,566991	12970	0,829524							
11	194.5	13472,22	0,603712	14080	0,513074							
12	195	18159,72	0,549295	10336	0,756934							
<div> <div>stoich_coefficients</div> <div>k_constants_log10</div> <div>concentrations</div> <div>absorbance</div> <div style="border: 1px solid green;">mol_ext_coefficients</div> <div>target</div> </div>												

Figure 24. Molar extinction coefficients tab

TIP: If you calculated the molar extinction coefficients with *Extinction Coefficients* tba just before evaluating the constant(s), you may upload them into the appropriate KEV table on the tab *Spectrophotometry* simply pressing the button *From memory* (Fig. 25).

Molar extinction coefficients

Peaks (up to 10)

110, 120, 130

	wavelength	2MF3H	2MF3H_adj.r.squared	PLP	PLP_adj.r.squared
3	191	18694.4444444444	0.948223875725258	9725.99999999999	0.393250077337198
4	191.5	16166.6666666667	0.959953001349617	10334	0.941262764776496
5	192	11538.1944444444	0.143200127214927	13184	0.750930708148839
6	192.5	10611.1111111111	-0.118390641778676	3888	0.315730065446006
7	193	16541.6666666667	0.549601260870289	6930	0.767526342960483
8	193.5	26166.6666666667	0.896376797577931	7064	0.865942350591561
9	194	22506.9444444444	0.566991278449665	12970	0.829523752418321
10	194.5	13472.2222222222	0.603712277024861	14080	0.513073872695311
11	195	18159.7222222222	0.549295147898487	10336	0.756934100459761
12	195.5	26729.1666666667	0.513819171647628	9944	0.647613300007799
13	196	17920.1388888889	0.905128043261782	10674	0.911674076311049
14	196.5	10875	0.23442522797718	18172	0.80000390526827
15	197	14034.7222222222	0.946963735342395	15254	0.85332439768307
16	197.5	11392.3611111111	0.697734365699167	9610	0.941844782446417
17	198	11010.4166666667	0.776718195496359	3704	0.260474892540191
18	198.5	15680.5555555556	0.918131044400003	15380	0.837882877529265
19	199	14559.0277777778	0.763329635909546	9862	0.994286196568424
20	199.5	8111.11111111111	0.998132035067811	7590	0.785508289023749
21	200	8399.30555555556	0.997580453215015	5708	0.863144588690495
22	200.5	9496.52777777778	0.9968243465431	5526	0.85192451065985

Choose CSV File

Browse...
No file selected

From memory

Figure 25. If there are some preliminary calculations on the tab “Extinction coefficients”, this button makes them into the table

4.7.3. target

Finally, the sheet *target* defines constant(s) should be evaluated and wavelengths KEV should use for that aim (Fig. 26).

IMPORTANT: The name of constant(s) to be determined should correspond to those listed in the tab *stoich_coefficients*. Choose one or more from the reactions you have written.

	A	B	C	D
1	constant	SB		
2	wavelength	297	337	388
3				
4				

Figure 26. Target tab

You might add some wavelengths either in KEV window on the appropriate field (Figure 27) or in the *target* tab of the .xlsx file you have uploaded. Remember that maximal number of wavelengths involved into calculations is limited to 10. The most characteristics values should be chosen (as a rule, they correspond to the absorbance maxima of reagents or products).

TIP: Do not use the values from non-informative ranges with strong solvent interference

Molar extinction coefficients			
Peaks (up to 10)			
	297, 337, 388		
	wavelength	2MF3H	2MF3H_adj.r.squared
3	191	18694.4444444444	0.948223875725258
4	191.5	16166.6666666667	0.959953001349617
5	192	11538.1944444444	0.143200127214927

Figure 27. Wavelengths could be added if required, but their total number should not exceed 10

When the xlsx data file is done upload it into KEV the same way as the bulk upload with CSV files i.e. select *Upload all data* in the *Bulk upload / download (optional)* section (Fig. 11).

TIP: Since the whole spectra were taken for calculations, the molar extinction coefficients values at all wavelengths are calculated for the unknown particles. It allows plotting their spectrum in any software with this functional (MS Excel, Origin

etc.). Wonderful, isn't it? It also provides an additional possibility to check the validity of constant calculation – if the calculated spectrum differs drastically from the experimental one or contains non-sense values (negative or too high ones), the results are incorrect.

The results could be downloaded by hitting the appropriate button (Fig. 28) as single .xlsx file whose tabs contain all the input and calculated tables or as an archived bunch of .csv files.

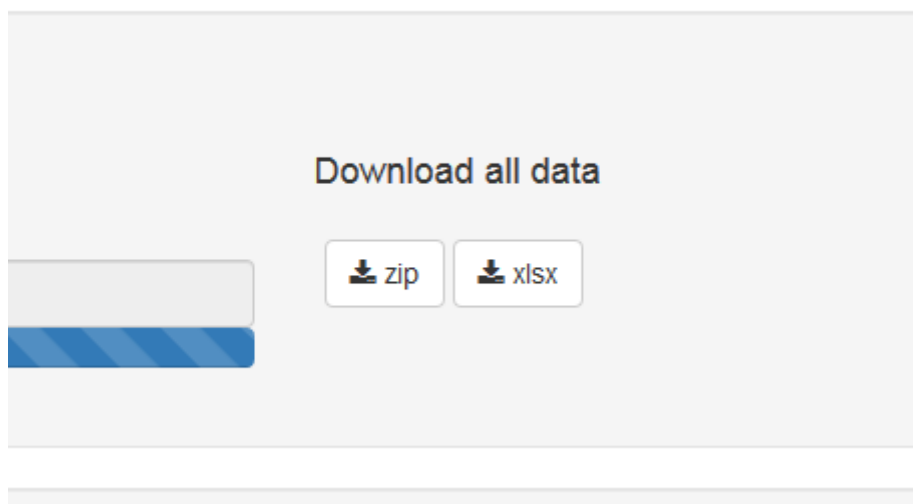


Figure 28. The results might be downloaded

See Section 3.8 if you need details of results downloading.

4.8 Option: Molar Extinction Coefficients Calculation

A quick, convenient and reliable way to calculate the molar extinction coefficients is the calibration plot method. Finding them for some reagents, at least, makes the constant evaluation more reliable. KEV allows calculating molar extinction coefficients for every wavelength, for entire spectra of compounds. They could be easily taken for the constant evaluations and unknown particles spectra calculations.

4.8.1 Select the Problem

Select *Extinction Coefficients* in the top menu to open the tab (Fig. 29).

KEV
Equilibrium concentrations
Spectrophotometry
Extinction Coefficients
EMF
Home

KEV: Chemistry Constant Evaluator

Calculate Molar Extinction Coefficients

Column delimiter

☐ ,
☐ ;
☒ tab

Bulk upload

Upload all data

Choose CSV, TXT files or XLSX file with multiple sheets

Browse...
No file selected

Figure 29. Extinction Coefficients tab

4.8.2 Data Description

In order to calculate the extinction coefficients UV-Vis experimental results KEV requires the following data:

- Tables (matrices and vectors)
 - Absorbance values at certain wavelengths with particle name and its total concentrations provided

There is no possibility to input data manually, because we assumed that the large amounts of values are to be processed here.

4.8.3 Upload Data from CSV or Tab Delimited File(s)

Please follow the recommendations below to format files in the right way.

- File names allowed: Any
- First row: particle name
- Second row: “wavelength”, total concentrations
- Following rows
 - Column *wavelength*: wavelength identifier, e.g. 190, 190.5, 191, 191.5, 192 etc.

- Following columns: experimental absorbance value for the solution with corresponding total concentration

IMPORTANT: All files should be formatted with the same delimiter (e.g. comma, semicolon or tab).

Now when all files are prepared upload them all at once selecting *Upload all data* clicking *Browse...* in the *Bulk upload section*.

KEY Equilibrium concentrations Spectrophotometry Extinction Coefficients EMF Home

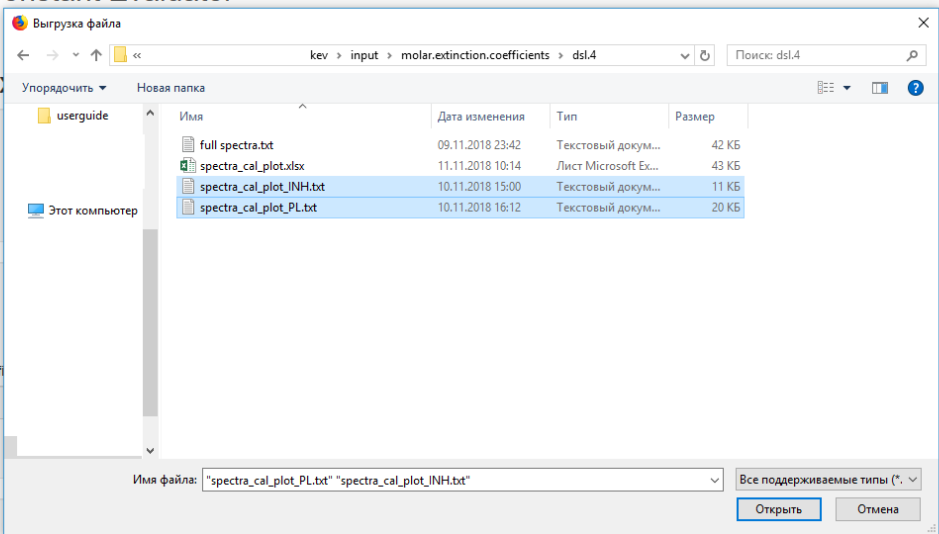
KEY: Chemistry Constant Evaluator

Calculate Molar Extinction Coefficients

Column delimiter
☐ , ☐ ; ☒ tab

Bulk upload
 Upload all data
 Choose CSV, TXT files or XLSX file
 Browse... No file selected

Molar extinction coefficients



The calculations start immediately after uploading the file. The results should look as below:

Upload all data
 Choose CSV, TXT files or XLSX file with multiple sheets
 Browse... PLP.txt
 Upload complete

Molar extinction coefficients

	wavelength	PLP	PLP_adj r squared
1	190.0	6656.00000	0.41930
2	190.5	6826.00000	0.23360
3	191.0	9726.00000	0.39330
4	191.5	10334.00000	0.94130
5	192.0	13184.00000	0.75090
6	192.5	3888.00000	0.31570
7	193.0	6938.00000	0.78750
8	193.5	7064.00000	0.86590
9	194.0	12970.00000	0.82950
10	194.5	14080.00000	0.51310
11	195.0	10336.00000	0.75690
12	195.5	9944.00000	0.64760
13	196.0	10674.00000	0.91170
14	196.5	18172.00000	0.80000
15	197.0	15254.00000	0.85330
16	197.5	9610.00000	0.94180
17	198.0	3704.00000	0.26050
18	198.5	15380.00000	0.83790
19	199.0	9862.00000	0.99430
20	199.5	7590.00000	0.78550
21	200.0	5708.00000	0.86310
22	200.5	5526.00000	0.85190
23	201.0	4588.00000	0.87740

The calculated values of molar extinction coefficients are accompanied by adjusted R^2 values. Some of them are marked with red (Fig. 23). It means, that the R^2_{adj} is too low (<0.95). Several reasons may lead to that result:

- Using the non-informative wavelengths range, and this is the case in the example. Make sure you are processing the data obtained in the spectral region, where the solvent makes no disturbance
- Another reason might be an error during the solution preparation, spectrophotometric measurements, or typo in the input file

TIP: Do not use the wavelengths from the troublesome wavelength range for constant evaluations later

The results could be downloaded from KEV in .xlsx or .csv format for further using in preparing the data for constant evaluation (see Fig. 26):

Bulk upload

Upload all data

Choose CSV, TXT files or XLSX file with multiple sheets

Browse... PLP.txt Upload complete

Molar extinction coefficients

	wavelength	PLP	PLP_adj r.squared
1	190.0	6656.00000	0.41930
2	190.5	6826.00000	0.23360
3	191.0	9726.00000	0.39330
4	191.5	10334.00000	0.94130
5	192.0	13184.00000	0.75090
6	192.5	3888.00000	0.31570
7	193.0	6930.00000	0.76750
8	193.5	7064.00000	0.86590
9	194.0	12970.00000	0.82950
10	194.5	14080.00000	0.51310
11	195.0	10336.00000	0.75690
12	195.5	9944.00000	0.64760
13	196.0	10674.00000	0.91170
14	196.5	16172.00000	0.80000
15	197.0	15254.00000	0.85330
16	197.5	9610.00000	0.94180
17	198.0	3704.00000	0.26050
18	198.5	15380.00000	0.83790
19	199.0	9862.00000	0.99430
20	199.5	7590.00000	0.78550
21	200.0	5708.00000	0.86310
22	200.5	5526.00000	0.85190
23	201.0	4588.00000	0.87740

CSV XLSX

4.8.4 Option: Upload Data from XLSX Single File

The convenient way to calculate extinction coefficients is to prepare .xlsx file, containing the following information

- name of the compound
- wavelengths
- total concentrations of compound
- absorbance values at every wavelength

Several compounds could be input simultaneously on different sheets (See e.g. Fig. 30).

	A	B	C	D	E
1	PLP				
2	wavelength	2,0E-4	5,0E-5	1,0E-4	1,5E-4
3	190,00	1,420	0,629	0,914	1,869
4	190,50	1,796	1,108	0,988	2,337
5	191,00	1,728	0,642	0,711	2,316
6	191,50	1,946	0,488	0,711	1,504
7	192,00	2,549			17
8	192,50	1,563	0,802	1,331	0,992
9	193,00	1,627	0,605	1,246	1,645
10	193,50	1,742	0,582	1,032	1,084
11	194,00	2,706	0,925	0,878	2,020
12	194,50	2,170	0,445	1,040	2,905
13	195,00	2,249	0,762	0,664	1,371
14	195,50	1,863	0,570	1,097	2,190
15	196,00	2,212	0,715	0,807	1,819
16	196,50	3,561	0,666	0,757	1,666
17	197,00	2,963	0,562	1,049	1,473
18	197,50	2,008	0,488	1,251	1,496
19	198,00	1,586	0,921	1,652	1,509
20	198,50	2,975	0,537	1,046	1,422
21	199,00	2,039	0,582	1,077	1,637
22	199,50	1,818	0,692	0,701	1,118
23	200,00	1,434			1,222
24	200,50	1,473			1,011
25	201,00	1,316	0,629	0,703	0,936
26	201,50	1,334	0,663	0,669	0,947
27	202,00	1,300	0,666	0,662	0,989
28	202,50	1,315	0,666	0,670	0,996
29	203,00	1,324	0,673	0,673	0,998
30	203,50	1,354	0,675	0,676	1,003
31	204,00	1,367	0,677	0,683	1,021
32	204,50	1,391	0,680	0,691	1,037
33	205,00	1,410	0,684	0,700	1,053
34	205,50	1,433		0,709	1,068
35	206,00	1,458		0,710	1,086
36	206,50	1,481		0,711	1,104
37	207,00	1,506	0,688	0,714	1,125
38	207,50	1,537	0,618	0,758	1,146
39	208,00	1,566	0,701	0,773	1,169

Figure 30. How to prepare .xlsx file for molar extinction coefficients calculation

Upload the prepared xlsx file should be uploaded to KEV (Fig. 31) and the computation begins immediately.

Bulk upload

Upload all data

Choose CSV, TXT files or XLSX file with multiple sheets

Browse... No file selected

Figure 31. Uploading the data for molar extinction coefficients calculation

After a short time, the results will appear (Fig. 32).

	wavelength	2MF3H	2MF3H_adj.r.squared	PLP	PLP_adj.r.squared
1	190.0	13409.72220	-0.17430	6656.00000	0.41930
2	190.5	20111.11110	0.69580	6826.00000	0.23360
3	191.0	18694.44440	0.94820	9726.00000	0.39330
4	191.5	16166.66670	0.96000	10334.00000	0.94130
5	192.0	11538.19440	0.14320	13184.00000	0.75090
6	192.5	10611.11110	-0.11840	3888.00000	0.31570
7	193.0	16541.66670	0.54960	6930.00000	0.76750
8	193.5	26166.66670	0.89640	7064.00000	0.86590
9	194.0	22506.94440	0.56700	12970.00000	0.82950
10	194.5	13472.22220	0.60370	14080.00000	0.51310
11	195.0	18159.72220	0.54930	10336.00000	0.75690
12	195.5	26729.16670	0.51380	9944.00000	0.64760
13	196.0	17920.15990	0.90510	10674.00000	0.91170
14	196.5	10875.00000	0.23440	18172.00000	0.80000
15	197.0	14034.72220	0.94700	15254.00000	0.85330
16	197.5	11392.36110	0.69770	9610.00000	0.94180
17	198.0	11010.41670	0.77670	3704.00000	0.26050
18	198.5	15680.55560	0.91810	15380.00000	0.83790
19	199.0	14559.02780	0.76330	9862.00000	0.99430
20	199.5	8111.11110	0.99810	7590.00000	0.78550
21	200.0	8399.30560	0.99760	5708.00000	0.86310
22	200.5	9496.52780	0.99680	5526.00000	0.85190
23	201.0	9111.11110	0.99910	4588.00000	0.87740

Figure 32. The results of calculation of molar extinction coefficients

72	225.5	5951.38890	0.99940	11106.00000	0.99870
73	226.0	5888.88890	0.99950	11092.00000	0.99880
74	226.5	5833.33330	0.99960	11076.00000	0.99890
75	227.0	5788.19440	0.99950	11036.00000	0.99900
76	227.5	5763.88890	0.99960	11012.00000	0.99910
77	228.0	5729.16670	0.99950	10972.00000	0.99930
78	228.5	5718.75000	0.99960	10952.00000	0.99930
79	229.0	5704.86110	0.99960	10922.00000	0.99930
80	229.5	5704.86110	0.99960	10888.00000	0.99940
81	230.0	5715.27780	0.99960	10842.00000	0.99950
82	230.5	5725.69440	0.99960	10792.00000	0.99950
83	231.0	5750.00000	0.99970	10750.00000	0.99960
84	231.5	5760.41670	0.99960	10702.00000	0.99960
85	232.0	5795.13890	0.99960	10628.00000	0.99960

CSV XLSX

Figure 33. Downloading the results of calculation of molar extinction coefficients

The results could be downloaded from KEV in .xlsx or .csv format (Fig. 33).

4.9. More playground

If you like to see the examples of constant evaluation for more complicated system including those with several parallel reactions or simultaneous determination of two constants, please, see [2] and [3], respectively.

References to the Section IV

1. G. A. Gamov et al. *Russ. J. Phys. Chem. A* **2019**, 93(2), 240-245, doi: 10.1134/S0036024419020110
2. <https://gitlab.com/a.meshkov/KEV/tree/master/input/spectrophotometry/dsl.5>

3. <https://gitlab.com/a.meshkov/KEV/tree/master/input/spectrophotometry/dsl.3>

V. EQUILIBRIUM CONSTANTS

EVALUATION FROM

POTENTIOMETRIC DATA

5.1 Select the Problem

Select *EMF* in the top menu to open the tab.

Let us consider the reaction of pyridoxine protonation in aqueous solution [1]. The experiment was designed as follows: pyridoxine hydrochloride (0.01017 M) was in the potentiometric cell (19.77 ml), and standardized NaOH (0.1347 M) served as a titrant. Therefore, there will be three basis particles: H, L, and OH. The following reactions should be considered:



The example data could be obtained:

- Input Data
 - CSV comma separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/emf/dsp.1/csv.comma>
 - CSV semicolon separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/emf/dsp.1/csv.semicolon>
 - Tab separated files
<https://gitlab.com/a.meshkov/KEV/tree/master/input/emf/dsp.1/txt.tab>
 - Excel (xlsx) file
<https://gitlab.com/a.meshkov/KEV/tree/double/input/emf/dsp.1/data.input.xlsx>
- Output (calculated) data
 - Zip archive with CSV files
<https://gitlab.com/a.meshkov/KEV/tree/double/output/emf/dsp.1/gui/kev.emf.constants.data.zip>
 - Excel (xlsx) file
<https://gitlab.com/a.meshkov/KEV/tree/double/output/emf/dsp.1/gui/kev.emf.constants.data.xlsx>

5.2 Data Description and Manual Input

In order to evaluate the equilibrium constant(s) from UV-Vis experimental results KEV requires the following data:

- Tables (matrices and vectors)
 - Stoichiometric matrix of reactions (a.k.a. Stoichiometric coefficients)
 - Decimal logarithms of equilibrium constants
 - Total (or equilibrium) concentrations of reagents
 - EMF values for certain indicator particle with their experimental deviations
 - Standard potential and Nernst slope for the *very same* indicator particle as EMF
- Text input
 - The name of constant(s), which should be evaluated
 - Threshold (search algorithm precision)
 - Search density (default to 1)

The initial three tables are the same as in the case of calculation of the equilibrium composition. See Sections 3.2.1-3.2.3; 3.6-3.7 for details.

However, the additional information input is also in order. It includes the values of potential difference between indicator and reference electrodes (or EMF) with the errors of their measuring as well as Nernst equation constants, namely, standard potential and slope (Fig. 33).

Particle names, comma separated
molecule1, molecule2, molecule3, molecule4

EMF and deviations

	data	particle	S1	S2	S3	S4	S5
1	observation	molecule1	100	100	100	100	100
2	deviation	molecule1	0.01	0.01	0.01	0.01	0.01

Choose CSV File
Browse... No file selected

E₀ and Slope

	Parameter	Value
1	standard potential	388
2	slope	59

Choose CSV File
Browse... No file selected

Figure 34. Screenshot of KEV interface. Experimental EMF table is marked with red, standard potential and Nernst slope are marked with blue.

The name of indicator particle should be specified in the “particle” column of the “EMD and deviations” table!

Potentiometric titration is a standard and often-used method of the constant determination. Therefore, you can be puzzled right now failing to find the fields where the volumes of titrant and titrand should be input. Let’s make it clear: there are none. The table of total concentrations serves instead of them. Every line of this table is a single point of titration.

How to convert the titration data into total concentrations? Let us consider the following values to be known: initial volume of titrand V_0 , volumes of titrant added V_t at every i^{th} point (up to V_T , the total volume of titrant added), total concentration of titrand C_{cell} and titrant C_{titr} . In this case, the total concentration of compounds at every i^{th} titration point could be calculated using following equations:

$$C_i(\text{titrand}) = \frac{C_{\text{cell}} \cdot V_0}{V_0 + V_t} \quad (21),$$

$$C_i(\text{titrant}) = \frac{C_{\text{titr}} \cdot V_t}{V_0 + V_t} \quad (22).$$

If titrand presents initially in the titrant solution with concentration $C_{\text{titr}}(\text{titrand})$ and titrant presents initially in the cell with concentration $C_{\text{cell}}(\text{titrant})$ (but what do they do there?) the Eqs. (18), (19) should be replaced by more general expressions:

$$C_i(\text{titrand}) = \frac{C_{\text{cell}}(\text{titrand}) \cdot V_0 + C_{\text{titr}}(\text{titrand}) \cdot V_t}{V_0 + V_t} \quad (23),$$

$$C_i(\text{titrant}) = \frac{C_{\text{cell}}(\text{titrant}) \cdot V_0 + C_{\text{titr}}(\text{titrant}) \cdot V_t}{V_0 + V_t} \quad (24).$$

The calculations of concentration, thus, could be made easily using, e.g., MS Excel (more reliable) or couple of PhD students with calculators (more fun to the professor). Then just paste calculated concentrations into KEV table.

As an example of processing the results of potentiometric titration using KEV, let us consider the determination of pyridoxine protonation constant in aqueous solution [1]. The experiment was designed as follows: pyridoxine hydrochloride (0.01017 M) was in the potentiometric cell (19.77 ml), and standardized NaOH (0.1347 M) served as a titrant. Therefore, there will be three basis particles: H, L, and OH. The following reactions (18-20) should be considered. The process (18) constant is to be evaluated.

5.2.1 Stoichiometric matrix (Stoichiometric coefficients)

The following stoichiometric matrix corresponds to the processes (22)-(24):

$$\begin{array}{ccc} 1 & 1 & 0 \\ -1 & 1 & 0 \\ 1 & 0 & 1 \end{array}$$

Is that all? Not yet: there is also an additional column, “name”. It serves for naming the products of reactions. Let us name the products “HL” (protonated ligand), “L-“ (dissociated ligand), “HOH” (water). Then, the table looks like that:

$$\begin{array}{cccc} 1 & 1 & 0 & \text{HL} \end{array}$$

-1	1	0	L-
1	0	1	HOH

Type it into KEV table *Stoichiometric coefficients*. Adding or removing columns/rows could be made with the context menu (mouse right click).

TIP: If you do not want to type numbers cell by cell (we don't), copy numbers into a blank Excel file, split data into columns and copy + paste it from the Excel file to the KEV table *Stoichiometric coefficients*

5.2.2 Decimal logarithms of equilibrium constants

The vector of $\lg K_{\text{total}}$ contains the following values:

4.5

-8.35

13.88

Type it into KEV table *K: lg constants* preserving the order or copy + paste as for the stoichiometric matrix data.

IMPORTANT: Please note the decimal logarithms of equilibrium constants ($\lg K$) should be input and not the K values

5.2.3 Total (or equilibrium) concentrations of reagents

So, the pyridoxine hydrochloride (0.01017 M) was in the potentiometric cell (19.77 ml), and standardized NaOH (0.1347 M) served as a titrant.

The following results of potentiometric titration were obtained:

V_{titr} , ml	EMF, mV	δEMF , mV
0.001	163.700	0.010
0.081	142.000	0.010
0.175	123.300	0.010
0.281	110.700	0.010
0.360	102.400	0.010
0.444	94.200	0.010
0.684	72.200	0.010
0.781	63.500	0.010
0.857	56.000	0.010

0.942 44.200 0.010

We are free to calculate the total concentrations of the reagents using Eqs. (21), (22). Here are the results!

0.010169486	0.010169486	6.81301E-06
0.010128502	0.010128502	0.00054963
0.010080767	0.010080767	0.001181875
0.010027475	0.010027475	0.001887721
0.009988122	0.009988122	0.002408942
0.009946616	0.009946616	0.002958682
0.009829906	0.009829906	0.004504488
0.009783509	0.009783509	0.005119006
0.009747462	0.009747462	0.005596446
0.009707459	0.009707459	0.006126275

Remember the columns correspond to the particle names, so name the first column *H*, the second one *L*, and the last one *OH*.

5.2.4 Particle names

Columns in both *Stoichiometric coefficients* and *Concentrations* tables have the same names which are the names of the base particles. By default KEV uses particle names *molecule1*, *molecule2*, *molecule3* etc. If you want to change them (and you'd better do) type new base particle names into the *Particle names, comma separated* field.

H, L, OH

IMPORTANT: Make sure the quantity of reagents in the string *Particle names, comma separated* corresponds to the quantity of columns of Table *Stoichiometric coefficients*. Delete or add columns if required

5.2.5 EMF and deviations

The *EMF and deviations* table consists of following columns:

- *data*. The values in column *data* can take *observation* or *deviation* value. *observation* is your experimental value; the *deviation* is its experimental error of determination.
- *particle*. The name of indicated particle. It is *H* in our example system.

- Columns with experimental values and experimental deviations for every mixture (solution). They could be named $S1$, $S2$, $S3$... SK where K is the number of solutions used in calculations. In case of our example there are 9 solutions.

IMPORTANT: The *particle* cells for *deviations* must contain the same designation of indicator particle as you have given to experimental values

The following matrix should be applied for the example (18-20):

observation H	163.7	142.0	123.3	110.7	102.4	94.2	72.2	63.5	56.0	44.2
deviation H	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01

The upper row contains the experimental values of EMF. The lower one is for the deviations.

5.2.6 E_0 and Slope

Standard potential and Nernst slope must be determined prior the calculations using, e.g., calibrating plot graph, i.e. the dependence of EMF on the logarithm of indicator particle concentration.

IMPORTANT: Pay attention that during the calibration plot determination of E_0 and slope, the logarithms of indicator particle value should be used, but not the pH (pC) values

5.2.7 Constants to evaluate

Type one or more product names into *Constants to evaluate* field (Fig. 16). Make sure they correspond to the ones specified in the *Stoichiometric coefficients* table.

5.3 Evaluate

After the input data is provided check if all is OK. If it is not you will get a bunch of red and grey error messages. Most frequent error causes and fixes are:

- Check *Column delimiter* control. If you use csv file with semicolon ";" delimiter should be also ";"
- Check if particle (molecule) names are consistent between different tables i.e. coefficient, concentrations and molar extinction matrices
- Check the same for wavelengths
- If you are uploading data table by table just try to load remaining data, the "error" could be a temporary thing disappearing after all the data is provided

In order to calculate the equilibrium concentrations of the reagents and products click *Evaluate* button right under the input data tables.

5.4 Output (Calculated) Data

The input data are ready for calculations (Fig. 35). Do not forget to specify the constant name to be evaluated, the standard potential and the slope determined from independent calibration plot experiment, where the EMF values were measured as function of known pH.

Upload or type input data

Stoichiometric coefficients

	H	L	OH	name
1	1	1	0	HL
2	-1	1	0	L-
3	1	0	1	HOH

Choose CSV File

Browse... No file selected

CSV XLSX

Particle names, comma separated

H, L, OH

K: lg constants

	k_constants_log10
1	4.5
2	-4.35
3	13.88

Choose CSV File

Browse... No file selected

CSV XLSX

Concentrations

	H	L	OH
1	0.010169486	0.010169486	6.81301e-06
2	0.010128502	0.010128502	0.00054963
3	0.010080767	0.010080767	0.001181875
4	0.010027475	0.010027475	0.001887721
5	0.009988122	0.009988122	0.002408942
6	0.009946616	0.009946616	0.002958682
7	0.009829906	0.009829906	0.004504488
8	0.009783509	0.009783509	0.005119006
9	0.009747462	0.009747462	0.005596446
10	0.009707459	0.009707459	0.006126275
1	tot	tot	tot

Choose CSV File

Browse... No file selected

CSV XLSX

EMF and deviations

	data	particle	S1	S2	S3	S4	S5	V1	V2	V3	V4	V5
1	observation	H	163.7	142	123.3	110.7	102.4	94.2	72.2	63.5	56	44.2
2	deviation	H	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01

Choose CSV File

Browse... No file selected

CSV XLSX

E₀ and Slope

	Parameter	Value
1	standard potential	388.42
2	slope	64.56

Choose CSV File

Browse... No file selected

CSV XLSX

Figure 35. Input data of potentiometric titration for pyridoxine protonation constant determination in aqueous solution

Press “Evaluate”. The results will appear below a few moments later (Fig. 36):

Equilibrium concentrations

	H	L	OH	HL	L-	HOH
1	0.000320488071412937	0.000327291958081053	2.80238092191515e-16	0.00964218948025314	4.56166580157672e-09	6.81301017290852e-06
2	0.000144796026128232	0.000694383183800933	5.00395962529773e-14	0.00943409739506043	2.14211386295047e-08	0.000549630013226915
3	7.48413702482869e-05	0.00125656637610473	2.08175660324912e-13	0.00882412562692757	7.49969676921337e-08	0.00118187502668392
4	4.46563154441929e-05	0.00193190081251318	5.57256225958165e-13	0.00809529093629993	1.93251186881847e-07	0.00188772103959653
5	3.29455612773452e-05	0.00244122558621326	9.63894365179021e-13	0.0075456542673665	3.3098705009783e-07	0.0024089420477887
6	2.48633454129693e-05	0.00298249456227456	1.56743496172418e-12	0.00696358604693992	5.35390785497869e-07	0.00295868205518827
7	1.2547914697495e-05	0.0045138222804947	4.73231759582094e-12	0.00531447693099269	1.60684095785135e-06	0.00450448806951236
8	9.68651672094208e-06	0.00512396693097367	6.9648215169506e-12	0.00465717922963593	2.36283939039499e-06	0.00511900607241739
9	7.89372404412004e-06	0.0055960042091368	9.34609913776375e-12	0.00414629003806259	3.16775278061955e-06	0.00559644607341704
10	6.22936319190555e-06	0.00612372218056846	1.29644124990737e-11	0.00357934573460202	4.39108482951387e-06	0.00612627507284992

CSV XLSX

Calculated EMF

Absolute Errors

Relative Errors

	data	particle	S1	S2	S3	S4	S5	S6	S7	S8	S9	S10
1	observation	H	162.8352	140.5584	122.0546	107.5764	99.049	91.1799	71.9838	64.7273	58.9855	52.3493
2	error	H	-0.8648	-1.4416	-1.2454	-3.1236	-3.351	-3.0201	-0.2162	1.2273	2.9885	8.1493

CSV XLSX

Evaluated Constants

Particle	Constant	St Deviation	Validity	
1	H	0	OK	
2	L	0	OK	
3	OH	0	OK	
4	HL	4.97234451770782	0.0185462562141481	OK
5	L-	-4.35	0	OK
6	HOH	13.88	0	OK

CSV XLSX

Correlation Matrix

	HL
HL	1

CSV XLSX

Last Fmin Step

Particle	Fmin Last
1	HL
2	-3.85995235774317e-09

CSV XLSX

Figure 35. Results of calculations for potentiometric experiment

- *Equilibrium concentrations*: all the equilibrium concentrations for basis particles and products

- *Calculated EMF*: the EMF values calculated using calculated equilibrium concentrations of the indicator particle and input standard potential and Nernst slope.
- *Evaluated Constants*: the input and evaluated equilibrium constants. The standard deviations of evaluated constants are also specified there. *Validity* column takes one of the following values
 - *OK*: everything seems right
 - *Non-Sensitive*: varying the evaluated log K value does not alter the minimizing function
 - *-Inf* or *Inf*: no global minima of minimizing functions were found
 - *Insignificant*: the evaluated value of log K has a standard deviation exceeding 10% of log K value
- *Correlation matrix*: the matrix, which diagonal elements are equal to 1 while non-diagonal ones are the Pearson correlation coefficients between each pair of optimized parameters. Since the only constant was evaluated in this example, the correlation matrix consists of the only element.
- *Last F_{min} Step*: the last alteration of minimizing function under variation of evaluated log K. A useful information, if something goes wrong.

TIP: Pay attention to the *Validity* comments. In case of any not-OK signal appearance, try doing the following:

1. Check typos and errors in the input data. the table *Calculated Absorbance* with errors could give a clue, where the blunder had been made
2. Change the initial approximated value(s) of log K.
3. Increase the value of threshold (Fig. 16)
4. Increase the search density. The value should be a positive integer
5. Remember, that Insignificant comment is rather subjective. For example, the constant value of 0.8 ± 0.1 might be accurate enough
6. If none of these helped, the experimental data might be insufficient or experimental design is unsuccessful

Since we have evaluated the only value of equilibrium constant, the table of correlation matrix consists of the only cell. Note that the *Evaluated constants* table contains the column *St. Deviation* reflecting the quality of model describing the experimental spectral data.

In order to avoid the errors caused by the incorrect manual input, we strongly recommend preparing the files (xlsx, CSV or TXT) containing all the data required.

5.5 Option: Upload Data from CSV or Tab Delimited File(s)

All the data could be entered manually however we recommend preparing the specific file (or files) containing all the required data.

TIP: Even if you input data manually you could save both input and output data in the file(s) so you could reproduce the research later or do minor modifications without retyping all the data from scratch.

CSV or tab delimited files are plain files with data delimited into columns with a specific delimiter (a.k.a. separator), mostly comma (,), semicolon (;) or tab (press *Tab* key in your preferred text editor to get the value).

These formats are very stable, platform independent and are used by a great range of applications and packages. There are at least four ways to format KEV input data as plain text files:

- Basic way: Type data using your preferred plain text editor such as Notepad on Windows
- Simplest way: Type or copy data to KEV, then download as CSV
- Dangerous way: Save from Excel using *Save as* option (could be some undesirable tricks)
- Advanced way: Use input from another app or package if format is the same as KEV requires

Please follow the recommendations below to format files in the right way.

TIP 1: Consider file names if you upload all data at once. If you upload file by file using *Choose CSV File* buttons there are no restrictions for file names

TIP 2: If you do not understand a word below use the Simplest way and consider only file names for the bulk input

5.5.1 Stoichiometric matrix

- File names allowed: *stoich_coefficients.csv*, *stoichiometric_coefficients.csv*, *stoich_coefficients*, *stoichiometric_coefficients*, *stoich_coefficients.txt*, *stoichiometric_coefficients.txt*. Prefix *input_* is also allowed. Remember: those names must be used if you upload all the files at once! If you upload them one by one, you are free to name the files whatever you want.
- First row: particle names, “name”
- Following rows: stoichiometric coefficients, one row for one reaction, and the name of reaction product.

5.5.2 Total (or equilibrium) concentrations of reagents

- File names allowed: *concentrations.csv*, *concentrations*, *concentrations.txt*. Prefix *input_* is also allowed
- First row: total (*tot*) or equal (*eq*) concentrations
- Second row: particle names
- Following rows: input concentrations, one row for one mixture

5.5.3 Decimal logarithms of equilibrium constants

- File names allowed: *k_constants_log10.csv*, *k_constants_log10*, *k_constants_log10.txt*. Prefix *input_* is also allowed
- First row: “k_constants_log10”
- Following rows: Decimal logarithms of equilibrium constants, one row for one reaction

5.5.4 EMF

- File names allowed: *emf.csv*, *emf*, *emf.txt*. Prefix *input_* is also allowed. Remember: those names must be used if you upload all the files at once! If you upload them one by one, you are free to name the files whatever you want.
- First row: “data”, “particle”, the different solutions names, e.g. “1”, “2”, “3”, etc.
- Following rows
 - Column *data*: either “observation” or “deviation”
 - Column *particle*: indicator particle symbol (e.g., “H”)
 - Following columns: experimental EMF values or their inaccuracies depending on what was specified in the first cell.

This table must contain one “observation” row and one “deviation” row.

5.5.5 Target

- File names allowed: *constants_names.csv*, *target.csv*, *constants_names*, *target*, *constants_names.txt*, *target.txt*. Remember: those names must be used if you upload all the files at once! If you upload them one by one, you are free to name the files whatever you want.
- First row: “standard.potential”, E_0 value
- Second row: “slope”, Nernst slope value

5.6 Option: Upload Data from CSV or Tab Delimited File(s)

Now when all files are prepared upload them to KEV.

See Section 3.6 for details. The uploading (all files at one time, one file by another) is pretty the same as in the case of equilibrium concentrations calculations.

5.7 Option: Upload Data from XLSX Single File

The first three sheets of this single .xlsx file are the same as for equilibrium composition problem (Section 3.7):

- *stoich_coefficients*
- *concentrations*
- *k_constants_log10*

New sheets are:

- *emf*
- *target*

IMPORTANT: Do not rename the sheets


	A	B	C	D	E	F	G	H	I
1	H	L	OH	name					
2	1.000	1.000	0.000	HL					
3	-1	1.000	0.000	L-					
4	1.000	0.000	1.000	H2O					
5									
6									
7									
8									
9									
10									
11									
12									

Figure 36. Sheets in the input .xlsx file


5.7.1 *emf*

The formatted tab containing experimental EMF values looks as Fig. 37. Specify also the experimental inaccuracies.

	A	B	C	D	E	F	G	H	I	J	K	L
data	particle	1	2	3	4	5	6	7	8	9	10	
observati	H	163.700	142.000	123.300	110.700	102.400	94.200	72.200	63.500	56.000	44.200	
deviaton	H	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	



Specify the name of indicator particle!



Input EMF values here

Figure 37. Tab containing experimental values of EMF

5.7.2 *target*

Finally, the sheet *target* defines constant(s) should be evaluated, E_0 and Nernst slope (Fig. 26).

IMPORTANT: The name of constant(s) to be determined should correspond to those listed in the tab *stoich_coefficients*. Choose one or more from the reactions you have written.

	A	B	
1	constant	HL	
2	standard.	388.420	
3	slope	64.560	
4			
5			
6			

Figure 38. “targets” sheet

Upload the .xlsx file and press “Evaluate”.

The results could be downloaded by hitting the appropriate button (Fig. 31) as single xlsx file whose tabs contain all the input and calculated tables or as an archived bunch of CSV files.

5.8. More playground

If you like to see the examples of constant evaluation for more complicated system including those with determination of two constants, please, visit [2].

References to the Section V

1. G. A. Gamov et al. *J. Chem. Thermodyn.* **2016**, 97, 322-330, doi: 10.1016/j.jct.2016.02.011
2. <https://gitlab.com/a.meshkov/KEV/tree/master/input/emf/dsp.2>