

BARIUM NITRATE

BNT

CAUTIONARY RESPONSE INFORMATION		
Common Synonyms	Solid	White Odorless Sinks and mixes with water.
Restrict access. AVOID CONTACT WITH SOLID AND DUST. Wear goggles and self-contained breathing apparatus. Shut off ignition sources and call fire department. Notify local health and pollution control agencies. Protect water intakes.		
Fire	Not flammable. May cause fire on contact with combustibles. POISONOUS GASES MAY BE PRODUCED IN FIRE. Flood discharge area with water.	
Exposure	CALL FOR MEDICAL AID. DUST POISONOUS IF INHALED. Irritating to eyes, nose and throat. Move victim to fresh air. If in eyes, hold eyelids open and flush with plenty of water. If breathing is difficult, give oxygen. SOLID POISONOUS IF SWALLOWED. Irritating to skin and eyes. Remove contaminated clothing and shoes. Flush affected areas with plenty of water. IF IN EYES, hold eyelids open and flush with plenty of water. IF SWALLOWED and victim is CONSCIOUS, have victim drink water or milk and have victim induce vomiting. IF SWALLOWED and victim is UNCONSCIOUS OR HAVING CONVULSIONS, do nothing except keep victim warm.	
Water Pollution	Dangerous to aquatic life in high concentrations. May be dangerous if it enters water intakes. Notify local health and wildlife officials. Notify operators of nearby water intakes.	

1. CORRECTIVE RESPONSE ACTIONS	2. CHEMICAL DESIGNATIONS	3. HEALTH HAZARDS	4. FIRE HAZARDS	5. CHEMICAL REACTIVITY	6. WATER POLLUTION	7. SHIPPING INFORMATION	8. HAZARD CLASSIFICATIONS	9. PHYSICAL & CHEMICAL PROPERTIES										
Dilute and disperse Stop discharge Collection Systems: Dredge	2.1 CG Compatibility Group: Not listed. 2.2 Formula: Ba(NO ₃) ₂ 2.3 IMO/UN Designation: 5.1/1446 2.4 DOT ID No.: 1446 2.5 CAS Registry No.: 10022-31-8 2.6 NAERG Guide No.: 141 2.7 Standard Industrial Trade Classification: 52359	3.1 Personal Protective Equipment: Goggles or face shield; dust respirator; rubber gloves and shoes; suitable coveralls. 3.2 Symptoms Following Exposure: Inhalation or contact with eyes or skin causes irritation. Ingestion causes excessive salivation, vomiting, colic, diarrhea, convulsive tremors, slow, hard pulse, elevated blood pressure. Hemorrhages may occur in the stomach, intestines, and kidneys. Muscular paralysis may follow. 3.3 Treatment of Exposure: Get medical attention. Alert doctor to possibility of barium poisoning, particularly if compound was swallowed. INHALATION: remove to fresh air. EYES: flush with water for at least 15 min. SKIN: flush with water. INGESTION: oral administration of an aqueous 10% solution of magnesium or sodium sulfate; in severe intoxication, calcium or a magnesium salt may have to be given I.V. with caution; treatment otherwise is supportive and symptomatic. 3.4 TLV-TWA: 0.5 mg/m ³ as barium. 3.5 TLV-STEL: Not listed. 3.6 TLV-Ceiling: Not listed. 3.7 Toxicity by Ingestion: Grade 3; oral rat LD ₅₀ = 355 mg/kg 3.8 Toxicity by Inhalation: Currently not available. 3.9 Chronic Toxicity: Barium poisoning 3.10 Vapor (Gas) Irritant Characteristics: Currently not available 3.11 Liquid or Solid Characteristics: Currently not available 3.12 Odor Threshold: Odorless 3.13IDLH Value: 50 mg/m ³ 3.14 OSHA PEL-TWA: 0.5 mg/m ³ as barium. 3.15 OSHA PEL-STEL: Not listed. 3.16 OSHA PEL-Ceiling: Not listed. 3.17 EPA AEGL: Not listed	4.1 Flash Point: Not flammable (but see 7.2) 4.2 Flammable Limits in Air: Not flammable 4.3 Fire Extinguishing Agents: Not pertinent 4.4 Fire Extinguishing Agents Not to Be Used: Not pertinent 4.5 Special Hazards of Combustion Products: Yields toxic gaseous oxides of nitrogen when involved in fire. 4.6 Behavior in Fire: Mixtures with combustible materials are readily ignited and may burn fiercely. Containers may explode. 4.7 Auto Ignition Temperature: Not pertinent 4.8 Electrical Hazards: Not pertinent 4.9 Burning Rate: Not pertinent 4.10 Adiabatic Flame Temperature: Currently not available 4.11 Stoichiometric Air to Fuel Ratio: Not pertinent 4.12 Flame Temperature: Currently not available 4.13 Combustion Molar Ratio (Reactant to Product): Not pertinent 4.14 Minimum Oxygen Concentration for Combustion (MOCC): Not listed	5.1 Reactivity with Water: No reaction 5.2 Reactivity with Common Materials: Contact with combustible material may cause fire. 5.3 Stability During Transport: Stable 5.4 Neutralizing Agents for Acids and Caustics: Not pertinent 5.5 Polymerization: Not pertinent 5.6 Inhibitor of Polymerization: Not pertinent	6.1 Aquatic Toxicity: 500 ppm/1658 hr/stickle back/average survival/fresh water 6.2 Waterfowl Toxicity: Currently not available 6.3 Biological Oxygen Demand (BOD): None 6.4 Food Chain Concentration Potential: None 6.5 GESAMP Hazard Profile: Bioaccumulation: 0 Damage to living resources: 0 Human Oral hazard: 2 Human Contact hazard: I Reduction of amenities: XX	7.1 Grades of Purity: Technical; Reagent 7.2 Storage Temperature: Ambient 7.3 Inert Atmosphere: No requirement 7.4 Venting: Open 7.5 IMO Pollution Category: Currently not available 7.6 Ship Type: Currently not available 7.7 Barge Hull Type: Currently not available	8.1 49 CFR Category: Oxidizer 8.2 49 CFR Class: 5.1 8.3 49 CFR Package Group: II 8.4 Marine Pollutant: Yes 8.5 NFPA Hazard Classification: <table border="1"> <thead> <tr> <th>Category</th> <th>Classification</th> </tr> </thead> <tbody> <tr> <td>Health Hazard (Blue)</td> <td>0 1</td> </tr> <tr> <td>Flammability (Red)</td> <td>0 0</td> </tr> <tr> <td>Instability (Yellow)</td> <td>0 0</td> </tr> <tr> <td>Special (White)</td> <td>OX OX</td> </tr> </tbody> </table> <p>* First column refers to non-fire situation.</p>	Category	Classification	Health Hazard (Blue)	0 1	Flammability (Red)	0 0	Instability (Yellow)	0 0	Special (White)	OX OX	9.1 Physical State at 15° C and 1 atm: Solid 9.2 Molecular Weight: 261.35 9.3 Boiling Point at 1 atm: Decomposes 9.4 Freezing Point: 1,098°F = 592°C = 865°K 9.5 Critical Temperature: Not pertinent 9.6 Critical Pressure: Not pertinent 9.7 Specific Gravity: 3.24 at 23°C (solid) 9.8 Liquid Surface Tension: Not pertinent 9.9 Liquid Water Interfacial Tension: Not pertinent 9.10 Vapor (Gas) Specific Gravity: Not pertinent 9.11 Ratio of Specific Heats of Vapor (Gas): Not pertinent 9.12 Latent Heat of Vaporization: Not pertinent 9.13 Heat of Combustion: Not pertinent 9.14 Heat of Decomposition: Not pertinent 9.15 Heat of Solution: 36 Btu/lb = 20 cal/g = 0.84 X 10 ⁵ J/kg 9.16 Heat of Polymerization: Not pertinent 9.17 Heat of Fusion: 22.6 cal/g (est) 9.18 Limiting Value: Currently not available 9.19 Reid Vapor Pressure: Currently not available
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NOTES																		

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9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
	NOT PERTINENT		NOT PERTINENT		NOT PERTINENT		NOT PERTINENT

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
34	5.322		NOT		NOT		NOT
36	5.644						
38	5.966						
40	6.288						
42	6.610						
44	6.933						
46	7.255						
48	7.577						
50	7.899						
52	8.222						
54	8.544						
56	8.866						
58	9.188						
60	9.510						
62	9.833						
64	10.150						
66	10.480						
68	10.800						
70	11.120						
72	11.440						
74	11.770						
76	12.090						
78	12.410						
80	12.730						
82	13.050						
84	13.380						