

MERCURIC CHLORIDE

MRC

CAUTIONARY RESPONSE INFORMATION		
Common Synonyms	Solid	White
KEEP PEOPLE AWAY. AVOID CONTACT WITH SOLID AND DUST. Wear a dust respirator and rubber overclothing (including gloves). Notify local health and pollution control agencies.		
Fire	Not flammable. POISONOUS GASES MAY BE PRODUCED WHEN HEATED.	
Exposure	CALL FOR MEDICAL AID. DUST POISONOUS IF INHALED OR IF SKIN IS EXPOSED. If inhaled will cause coughing or difficult breathing. If in eyes, hold eyelids open and flush with plenty of water. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen. SOLID POISONOUS IF SWALLOWED OR IF SKIN IS EXPOSED. Irritating to skin and eyes. If swallowed will cause nausea and vomiting. Remove contaminated clothing and shoes. Flush affected areas with plenty of water. IF IN EYES, hold eyelids open and flush with plenty of water. IF SWALLOWED and victim is CONSCIOUS, have victim drink water or milk and have victim induce vomiting. IF SWALLOWED and victim is UNCONSCIOUS OR HAVING CONVULSIONS, do nothing except keep victim warm.	
Water Pollution	HARMFUL TO AQUATIC LIFE IN VERY LOW CONCENTRATIONS. May be dangerous if it enters water intakes. Notify local health and wildlife officials. Notify operators of nearby water intakes.	

1. CORRECTIVE RESPONSE ACTIONS	2. CHEMICAL DESIGNATIONS	3. HEALTH HAZARDS
Stop discharge	<p>2.1 CG Compatibility Group: Not listed.</p> <p>2.2 Formula: <chem>HgCl2</chem></p> <p>2.3 IMO/UN Designation: 6.1/1624</p> <p>2.4 DOT ID No.: 1624</p> <p>2.5 CAS Registry No.: 7447-94-7</p> <p>2.6 NAERG Guide No.: 154</p> <p>2.7 Standard Industrial Trade Classification: 52329</p>	<p>3.1 Personal Protective Equipment: Bu. Mines approved airline respirator; impervious suit; appropriate eye protection</p> <p>3.2 Symptoms Following Exposure: All forms of exposure to this compound are hazardous; acute systemic mercurialism may be fatal within a few minutes, but death by uremic poisoning is usually delayed 5-12 days. Acute poisoning has resulted from inhaling dust concentrations of 1.2-8.5 mg/m³ of air; symptoms include tightness and pain in chest, coughing, and difficulty in breathing. Ingestion causes necrosis, pain, vomiting, and severe purging; as little as 0.5 gm can be fatal. Contact with eyes causes ulceration of conjunctiva and cornea. Contact with skin causes irritation and possible dermatitis; systemic poisoning can occur by absorption through skin.</p> <p>3.3 Treatment of Exposure: Act promptly! Alimentary absorption is very rapid, and the first 10-15 minutes determine the prognosis. INHALATION: remove victim to fresh air; get medical attention. INGESTION: give egg whites, milk, or activated charcoal; induce vomiting; consult physician. EYES or SKIN: wash with water for 15 min.</p> <p>3.4 TLV-TWA: 0.025 mg/m³ (as mercury)</p> <p>3.5 TLV-STEL: Not listed.</p> <p>3.6 TLV-Ceiling: Not listed.</p> <p>3.7 Toxicity by Ingestion: Grade 4; oral LD₅₀ = 1 mg/kg (rat)</p> <p>3.8 Toxicity by Inhalation: Currently not available.</p> <p>3.9 Chronic Toxicity: Currently not available</p> <p>3.10 Vapor (Gas) Irritant Characteristics: Currently not available</p> <p>3.11 Liquid or Solid Characteristics: Currently not available</p> <p>3.12 Odor Threshold: Currently not available</p> <p>3.13 IDLH Value: Not listed.</p> <p>3.14 OSHA PEL-TWA: Not listed.</p> <p>3.15 OSHA PEL-STEL: Not listed.</p> <p>3.16 OSHA PEL-Ceiling: 0.1 mg/m³ (as mercury)</p> <p>3.17 EPA AEGL: Not listed</p>

4. FIRE HAZARDS	7. SHIPPING INFORMATION
4.1 Flash Point: Not flammable	7.1 Grades of Purity: Reagent; Analytical
4.2 Flammable Limits in Air: Not flammable	7.2 Storage Temperature: Ambient
4.3 Fire Extinguishing Agents: Not pertinent	7.3 Inert Atmosphere: No requirement
4.4 Fire Extinguishing Agents Not to Be Used: Not pertinent	7.4 Venting: Open
4.5 Special Hazards of Combustion Products: Heat of fire may cause material to form fumes of mercuric chloride, which are toxic.	7.5 IMO Pollution Category: Currently not available
4.6 Behavior in Fire: Currently not available	7.6 Ship Type: Currently not available
4.7 Auto Ignition Temperature: Not pertinent	7.7 Barge Hull Type: Currently not available
4.8 Electrical Hazards: Not pertinent	
4.9 Burning Rate: Not pertinent	
4.10 Adiabatic Flame Temperature: Currently not available	
4.11 Stoichiometric Air to Fuel Ratio: Not pertinent	
4.12 Flame Temperature: Currently not available	
4.13 Combustion Molar Ratio (Reactant to Product): Not pertinent	
4.14 Minimum Oxygen Concentration for Combustion (MOCC): Not listed	
5. CHEMICAL REACTIVITY	9. PHYSICAL & CHEMICAL PROPERTIES
5.1 Reactivity with Water: No reaction	9.1 Physical State at 15° C and 1 atm: Solid
5.2 Reactivity with Common Materials: Currently not available	9.2 Molecular Weight: 271.50
5.3 Stability During Transport: Stable	9.3 Boiling Point at 1 atm: 576°F = 302°C = 575°K
5.4 Neutralizing Agents for Acids and Caustics: Not pertinent	9.4 Freezing Point: 531°F = 277°C = 550°K
5.5 Polymerization: Not pertinent	9.5 Critical Temperature: Not pertinent
5.6 Inhibitor of Polymerization: Not pertinent	9.6 Critical Pressure: Not pertinent
6. WATER POLLUTION	9.7 Specific Gravity: 5.4 at 20°C (solid)
6.1 Aquatic Toxicity: 0.82 ppm/7 days/goldfish/TL ₅₀ /fresh water 0.075 ppm/48 hr/pink shrimp/TL ₅₀ /salt water 4.2 ppm/48 hr/oyster/TL ₅₀ /sea water	9.8 Liquid Surface Tension: Not pertinent
6.2 Waterfowl Toxicity: Currently not available	9.9 Liquid Water Interfacial Tension: Not pertinent
6.3 Biological Oxygen Demand (BOD): None	9.10 Vapor (Gas) Specific Gravity: Not pertinent
6.4 Food Chain Concentration Potential: Many organisms are capable of accumulating mercury from water. Bioconcentrative up to 10,000-fold.	9.11 Ratio of Specific Heats of Vapor (Gas): Not pertinent
6.5 GESAMP Hazard Profile: Bioaccumulation: + Damage to living resources: B4 Human Oral hazard: 4 Human Contact hazard: II Reduction of amenities: XXX	9.12 Latent Heat of Vaporization: Not pertinent
	9.13 Heat of Combustion: Not pertinent
	9.14 Heat of Decomposition: Not pertinent
	9.15 Heat of Solution: Not pertinent
	9.16 Heat of Polymerization: Not pertinent
	9.17 Heat of Fusion: 15.3 cal/g
	9.18 Limiting Value: Currently not available
	9.19 Reid Vapor Pressure: Currently not available

NOTES

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9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
	NOT PERTINENT		NOT PERTINENT		NOT PERTINENT		NOT PERTINENT

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
34	3.666						
36	3.833						
38	4.000						
40	4.166						
42	4.333						
44	4.500						
46	4.666						
48	4.833						
50	5.000						
52	5.166						
54	5.333						
56	5.500						
58	5.666						
60	5.833						
62	6.000						
64	6.166						
66	6.333						
68	6.500						
70	6.666						
72	6.833						
74	7.000						
76	7.166						
78	7.333						
80	7.500						
82	7.666						
84	7.833						