

# METHYL FORMAL

MTF

CAUTIONARY RESPONSE INFORMATION				4. FIRE HAZARDS	7. SHIPPING INFORMATION								
Common Synonyms Dimethoxymethane Dimethyl formal Formaldehyde dimethylacetol Methylal Methylene dimethyl ether	Liquid	Colorless	Mild sweet odor	<p>4.1 Flash Point: 0°F O.C.</p> <p>4.2 Flammable Limits in Air: 1.6%-17.6%</p> <p>4.3 Fire Extinguishing Agents: Dry chemical, foam, carbon dioxide</p> <p>4.4 Fire Extinguishing Agents Not to Be Used: Water may be ineffective.</p> <p>4.5 Special Hazards of Combustion Products: Irritating formaldehyde gas may be present in smoke.</p> <p>4.6 Behavior in Fire: Not pertinent</p> <p>4.7 Auto Ignition Temperature: 459°F</p> <p>4.8 Electrical Hazards: Currently not available</p> <p>4.9 Burning Rate: 5.5 mm/min.</p> <p>4.10 Adiabatic Flame Temperature: Currently not available</p> <p>4.11 Stoichiometric Air to Fuel Ratio: 23.8 (calc.)</p> <p>4.12 Flame Temperature: Currently not available</p> <p>4.13 Combustion Molar Ratio (Reactant to Product): 7.0 (calc.)</p> <p>4.14 Minimum Oxygen Concentration for Combustion (MOCC): Not listed</p>	<p>7.1 Grades of Purity: 97+%</p> <p>7.2 Storage Temperature: Ambient</p> <p>7.3 Inert Atmosphere: No requirement</p> <p>7.4 Venting: Pressure-vacuum</p> <p>7.5 IMO Pollution Category: Currently not available</p> <p>7.6 Ship Type: Currently not available</p> <p>7.7 Barge Hull Type: Currently not available</p>								
Keep people away. Shut off ignition sources. Call fire department. Notify local health and pollution control agencies.				8. HAZARD CLASSIFICATIONS									
Fire	FLAMMABLE. Irritating gases may be produced when heated. Containers may explode in fire. Flashback along vapor trail may occur. Vapor may explode if ignited in an enclosed area. Extinguish with dry chemicals, foam or carbon dioxide. Water may be ineffective on fire. Cool exposed containers with water.				<p>8.1 49 CFR Category: Flammable liquid</p> <p>8.2 49 CFR Class: 3</p> <p>8.3 49 CFR Package Group: II</p> <p>8.4 Marine Pollutant: No</p> <p>8.5 NFPA Hazard Classification:</p> <table> <tr> <td>Category</td> <td>Classification</td> </tr> <tr> <td>Health Hazard (Blue).....</td> <td>2</td> </tr> <tr> <td>Flammability (Red).....</td> <td>3</td> </tr> <tr> <td>Instability (Yellow).....</td> <td>2</td> </tr> </table>	Category	Classification	Health Hazard (Blue).....	2	Flammability (Red).....	3	Instability (Yellow).....	2
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Exposure	Call for medical aid.  VAPOR Irritating to eyes, nose and throat. Harmful if inhaled. Move victim to fresh air. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen.  LIQUID Irritating to skin and eyes. Harmful if swallowed. Remove contaminated clothing and shoes. Flush affected areas with plenty of water. IF IN EYES, hold eyelids open and flush with plenty of water. IF SWALLOWED and victim is CONSCIOUS, having victim drink water or milk and have victim induce vomiting. IF SWALLOWED and victim is UNCONSCIOUS OR HAVING CONVULSIONS, do nothing except keep victim warm.				<p>8.6 EPA Reportable Quantity: Not listed.</p> <p>8.7 EPA Pollution Category: Not listed.</p> <p>8.8 RCRA Waste Number: Not listed</p> <p>8.9 EPA FWPCA List: Not listed</p>								
Water Pollution	Effect of low concentrations on aquatic life is unknown. May be dangerous if it enters water intakes. Notify local health and wildlife officials. Notify operators of nearby water intakes.				9. PHYSICAL & CHEMICAL PROPERTIES								
1. CORRECTIVE RESPONSE ACTIONS Dilute and disperse Stop discharge		2. CHEMICAL DESIGNATIONS 2.1 CG Compatibility Group: 41; Ether 2.2 Formula: CH <sub>2</sub> (OCH <sub>3</sub> ) <sub>2</sub> 2.3 IMO/UN Designation: 3.1/1234 2.4 DOT ID No.: 1234 2.5 CAS Registry No.: 109-87-5 2.6 NAERG Guide No.: 127 2.7 Standard Industrial Trade Classification: 51616											
3. HEALTH HAZARDS													
<p>3.1 Personal Protective Equipment: Self-contained breathing apparatus or all-purpose canister mask; rubber gloves; chemical safety goggles; impervious apron and boots.</p> <p>3.2 Symptoms Following Exposure: Inhalation causes irritation of respiratory system and depression of central nervous system. Liquid causes irritation of eyes and will irritate skin if allowed to remain. Ingestion causes depression of central nervous system.</p> <p>3.3 Treatment of Exposure: INHALATION: remove victim from contaminated area and administer artificial respiration and oxygen if necessary. EYES: flush with plenty of water; get medical attention. SKIN: flush with plenty of water. INGESTION: induce vomiting; administer gastric lavage and saline cathartics; subsequent treatment is symptomatic and supportive.</p> <p>3.4 TLV-TWA: 1,000 ppm</p> <p>3.5 TLV-STEL: Not listed.</p> <p>3.6 TLV-Ceiling: Not listed.</p> <p>3.7 Toxicity by Ingestion: Grade 1; LD<sub>50</sub> = 5 to 15 g/kg</p> <p>3.8 Toxicity by Inhalation: Currently not available.</p> <p>3.9 Chronic Toxicity: Liver and kidney injury may follow high exposures.</p> <p>3.10 Vapor (Gas) Irritancy Characteristics: Vapors cause a slight smarting of the eyes or respiratory system if present in high concentrations. The effect is temporary.</p> <p>3.11 Liquid or Solid Characteristics: Minimum hazard. If spilled on clothing and allowed to remain, may cause smarting and reddening of the skin.</p> <p>3.12 Odor Threshold: Currently not available</p> <p>3.13IDLH Value: 2,200 ppm</p> <p>3.14 OSHA PEL-TWA: 1,000 ppm</p> <p>3.15 OSHA PEL-STEL: Not listed</p> <p>3.16 OSHA PEL-Ceiling: Not listed.</p> <p>3.17 EPA AERGL: Not listed</p>					6. WATER POLLUTION								
<p>5.1 Reactivity with Water: No reaction</p> <p>5.2 Reactivity with Common Materials: No reaction</p> <p>5.3 Stability During Transport: Stable</p> <p>5.4 Neutralizing Agents for Acids and Caustics: Not pertinent</p> <p>5.5 Polymerization: Not pertinent</p> <p>5.6 Inhibitor of Polymerization: Not pertinent</p> <p>6.1 Aquatic Toxicity: Currently not available</p> <p>6.2 Waterfowl Toxicity: Currently not available</p> <p>6.3 Biological Oxygen Demand (BOD): Currently not available</p> <p>6.4 Food Chain Concentration Potential: None</p> <p>6.5 GESAMP Hazard Profile: Bioaccumulation: - Damage to living resources: - Human Oral hazard: 0 Human Contact hazard: I Reduction of amenities: X</p> <p>9.1 Physical State at 15°C and 1 atm: Liquid</p> <p>9.2 Molecular Weight: 76.1</p> <p>9.3 Boiling Point at 1 atm: 108°F = 42°C = 315°K</p> <p>9.4 Freezing Point: -157°F = -105°C = 168°K</p> <p>9.5 Critical Temperature: 419.0°F = 215°C = 488.2°K</p> <p>9.6 Critical Pressure: Not pertinent</p> <p>9.7 Specific Gravity: 0.861 at 20°C (liquid)</p> <p>9.8 Liquid Surface Tension: 21.1 dynes/cm = 0.0211 N/m at 20°C</p> <p>9.9 Liquid Water Interfacial Tension: Not pertinent</p> <p>9.10 Vapor (Gas) Specific Gravity: 2.6</p> <p>9.11 Ratio of Specific Heats of Vapor (Gas): 1.0888</p> <p>9.12 Latent Heat of Vaporization: 161.5 Btu/lb 89.8 cal/g = 3.76 X 10<sup>5</sup> J/kg</p> <p>9.13 Heat of Combustion: -10,970 Btu/lb = -6,100 cal/g = -255 X 10<sup>5</sup> J/kg</p> <p>9.14 Heat of Decomposition: Not pertinent</p> <p>9.15 Heat of Solution: Not pertinent</p> <p>9.16 Heat of Polymerization: Not pertinent</p> <p>9.17 Heat of Fusion: Currently not available</p> <p>9.18 Limiting Value: Currently not available</p> <p>9.19 Reid Vapor Pressure: Currently not available</p>					NOTES								

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9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
34	54.860	10	0.484	51	1.048	35	0.396
36	54.790	15	0.486	52	1.048	40	0.388
38	54.720	20	0.488	53	1.048	45	0.380
40	54.650	25	0.490	54	1.048	50	0.372
42	54.580	30	0.491	55	1.048	55	0.365
44	54.520	35	0.493	56	1.048	60	0.358
46	54.450	40	0.495	57	1.048	65	0.351
48	54.380	45	0.497	58	1.048	70	0.344
50	54.310	50	0.498	59	1.048	75	0.338
52	54.240	55	0.500	60	1.048	80	0.332
54	54.170	60	0.502	61	1.048	85	0.326
56	54.100	65	0.503	62	1.048	90	0.320
58	54.030	70	0.505	63	1.048	95	0.315
60	53.960	75	0.507	64	1.048	100	0.309
62	53.890	80	0.509	65	1.048		
64	53.820			66	1.048		
66	53.750			67	1.048		
68	53.680			68	1.048		
70	53.610			69	1.048		
72	53.540			70	1.048		
74	53.470			71	1.048		
76	53.410			72	1.048		
				73	1.048		
				74	1.048		
				75	1.048		
				76	1.048		

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
60	32.000	35	2.738	35	0.03924	0	0.294
		40	3.122	40	0.04430	20	0.301
		45	3.551	45	0.04989	40	0.309
		50	4.029	50	0.05604	60	0.317
		55	4.560	55	0.06281	80	0.324
		60	5.149	60	0.07024	100	0.332
		65	5.800	65	0.07836	120	0.339
		70	6.518	70	0.08724	140	0.347
		75	7.310	75	0.09693	160	0.354
		80	8.181	80	0.10750	180	0.362
		85	9.136	85	0.11890	200	0.369
		90	10.180	90	0.13130	220	0.376
		95	11.330	95	0.14480	240	0.383
		100	12.580	100	0.15930	260	0.390
		105	13.940	105	0.17500	280	0.397
		110	15.420	110	0.19180	300	0.404
						320	0.411
						340	0.417
						360	0.424
						380	0.431
						400	0.437
						420	0.444
						440	0.450