

N-PROPYL NITRATE

PNI

CAUTIONARY RESPONSE INFORMATION				4. FIRE HAZARDS	7. SHIPPING INFORMATION										
Common Synonyms	Liquid	Colorless to pale yellow	Ether-like odor	<p>4.1 Flash Point: 68°F C.C.</p> <p>4.2 Flammable Limits in Air: LEL: 2%; UEL: 100%</p> <p>4.3 Fire Extinguishing Agents: Alcohol foam, dry chemical, or carbon dioxide.</p> <p>4.4 Fire Extinguishing Agents Not to Be Used: Water.</p> <p>4.5 Special Hazards of Combustion Products: Toxic gases and vapors, such as oxides of nitrogen and carbon monoxide, may be released in a fire.</p> <p>4.6 Behavior in Fire: Currently not available</p> <p>4.7 Auto Ignition Temperature: 347°F.</p> <p>4.8 Electrical Hazards: I, B</p> <p>4.9 Burning Rate: Currently not available</p> <p>4.10 Adiabatic Flame Temperature: Currently not available</p> <p>4.11 Stoichiometric Air to Fuel Ratio: 20.2 (calc.)</p> <p>4.12 Flame Temperature: Currently not available</p> <p>4.13 Combustion Molar Ratio (Reactant to Product): 7.5 (calc.)</p> <p>4.14 Minimum Oxygen Concentration for Combustion (MOCC): Not listed</p>	<p>7.1 Grades of Purity: Technical.</p> <p>7.2 Storage Temperature: Ambient.</p> <p>7.3 Inert Atmosphere: No requirement.</p> <p>7.4 Venting: Not listed.</p> <p>7.5 IMO Pollution Category: Currently not available</p> <p>7.6 Ship Type: Currently not available</p> <p>7.7 Barge Hull Type: Currently not available</p>										
<p>Wear full chemical protective clothing, gloves, goggles and approved respirator.</p> <p>Evacuate</p> <p>Shut off ignition sources and call the fire department.</p> <p>Notify local health and pollution control agencies.</p> <p>Protect water intakes.</p>				<p>8. HAZARD CLASSIFICATIONS</p> <p>8.1 49 CFR Category: Flammable Liquid</p> <p>8.2 49 CFR Class: 3</p> <p>8.3 49 CFR Package Group: II</p> <p>8.4 Marine Pollutant: No</p> <p>8.5 NFPA Hazard Classification:</p> <table> <thead> <tr> <th>Category</th> <th>Classification</th> </tr> </thead> <tbody> <tr> <td>Health Hazard (Blue)</td> <td>2</td> </tr> <tr> <td>Flammability (Red)</td> <td>3</td> </tr> <tr> <td>Instability (Yellow)</td> <td>3</td> </tr> <tr> <td>Special (White)</td> <td>OX</td> </tr> </tbody> </table> <p>8.6 EPA Reportable Quantity: Not listed</p> <p>8.7 EPA Pollution Category: Not listed.</p> <p>8.8 RCRA Waste Number: Not listed</p> <p>8.9 EPA FWPCA List: Not listed</p>		Category	Classification	Health Hazard (Blue)	2	Flammability (Red)	3	Instability (Yellow)	3	Special (White)	OX
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<p>Fire</p> <p>Flammable.</p> <p>Wear full protective clothing with self-contained breathing apparatus.</p> <p>Extinguish fire with alcohol foam, dry chemical, or CO₂.</p> <p>Exposure</p> <p>CALL FOR MEDICAL AID.</p> <p>VAPOR</p> <p>Move victim to fresh air.</p> <p>If breathing has stopped, give artificial respiration.</p> <p>If breathing is difficult, give oxygen.</p> <p>LIQUID</p> <p>Remove contaminated clothing and shoes.</p> <p>Wash affected areas with soap and water.</p> <p>IF IN EYES, hold eyelids open and flush with plenty of water.</p> <p>IF SWALLOWED and victim is CONSCIOUS, give two glasses of water and induce vomiting.</p> <p>Water Pollution</p> <p>Effect of low concentrations on aquatic life is unknown.</p> <p>May be dangerous if it enters water intakes.</p> <p>Notify local health and wildlife officials.</p> <p>Notify operators of nearby water intakes.</p>				<p>5. CHEMICAL REACTIVITY</p> <p>5.1 Reactivity with Water: No reaction.</p> <p>5.2 Reactivity with Common Materials: Contact with either strong oxidizers or with combustibles may cause fires and explosions.</p> <p>5.3 Stability During Transport: Stable.</p> <p>5.4 Neutralizing Agents for Acids and Caustics: Not pertinent.</p> <p>5.5 Polymerization: Will not polymerize.</p> <p>5.6 Inhibitor of Polymerization: Not pertinent.</p> <p>6. WATER POLLUTION</p> <p>6.1 Aquatic Toxicity: Currently not available</p> <p>6.2 Waterfowl Toxicity: Currently not available</p> <p>6.3 Biological Oxygen Demand (BOD): Currently not available</p> <p>6.4 Food Chain Concentration Potential: Currently not available</p> <p>6.5 GESAMP Hazard Profile: Not listed</p>											
<p>1. CORRECTIVE RESPONSE ACTIONS</p> <p>Stop discharge</p> <p>Collection Systems: Pump</p> <p>Do not burn</p> <p>2. CHEMICAL DESIGNATIONS</p> <p>2.1 CG Compatibility Group: Not listed.</p> <p>2.2 Formula: CH₃CH₂NO₂</p> <p>2.3 IMO/UN Designation: Currently not available</p> <p>2.4 DOT ID No.: 1865</p> <p>2.5 CAS Registry No.: 627-13-4</p> <p>2.6 NAERG Guide No.: 131</p> <p>2.7 Standard Industrial Trade Classification: 51140</p> <p>3. HEALTH HAZARDS</p> <p>3.1 Personal Protective Equipment: Full, impervious chemical protective clothing and gloves, goggles, and approved respirator.</p> <p>3.2 Symptoms Following Exposure: Exposure can cause anoxia and cyanosis. Other effects are weakness, dizziness, and severe headaches.</p> <p>3.3 Treatment of Exposure: Get medical attention. INHALATION: Remove to fresh air. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen. EYES: Flush with water for at least 15 min., lifting lids occasionally. Contact lenses should not be worn when working with this chemical. SKIN: Remove contaminated clothing and shoes. Wash with soap and water.</p> <p>INGESTION: Give two glasses of water and induce vomiting.</p> <p>3.4 TLV-TWA: 25 ppm</p> <p>3.5 TLV-STEL: Not listed.</p> <p>3.6 TLV-Ceiling: 40 ppm</p> <p>3.7 Toxicity by Ingestion: Currently not available</p> <p>3.8 Toxicity by Inhalation: Currently not available.</p> <p>3.9 Chronic Toxicity: Currently not available</p> <p>3.10 Vapor (Gas) Irritant Characteristics: Vapors cause a slight smarting of the eyes or respiratory system if present in high concentrations. The effect is temporary.</p> <p>3.11 Liquid or Solid Characteristics: Minimum hazard. If spilled on clothing and allowed to remain, may cause smarting and reddening of the skin.</p> <p>3.12 Odor Threshold: 50 ppm.</p> <p>3.13 IDLH Value: 500 ppm</p> <p>3.14 OSHA PEL-TWA: 25 ppm</p> <p>3.15 OSHA PEL-STEL: Not listed.</p> <p>3.16 OSHA PEL-Ceiling: Not listed.</p> <p>3.17 EPA AEGL: Not listed</p>				<p>9. PHYSICAL & CHEMICAL PROPERTIES</p> <p>9.1 Physical State at 15°C and 1 atm: Liquid.</p> <p>9.2 Molecular Weight: 105.1</p> <p>9.3 Boiling Point at 1 atm: 231°F = 111°C = 384°K</p> <p>9.4 Freezing Point: <-150°F = <-101°C = <172°K</p> <p>9.5 Critical Temperature: Currently not available</p> <p>9.6 Critical Pressure: Currently not available</p> <p>9.7 Specific Gravity: 1.06</p> <p>9.8 Liquid Surface Tension: Currently not available</p> <p>9.9 Liquid Water Interfacial Tension: Currently not available</p> <p>9.10 Vapor (Gas) Specific Gravity: 3.6</p> <p>9.11 Ratio of Specific Heats of Vapor (Gas): Currently not available</p> <p>9.12 Latent Heat of Vaporization: Currently not available</p> <p>9.13 Heat of Combustion: Currently not available</p> <p>9.14 Heat of Decomposition: Currently not available</p> <p>9.15 Heat of Solution: Currently not available</p> <p>9.16 Heat of Polymerization: Not pertinent.</p> <p>9.17 Heat of Fusion: Currently not available</p> <p>9.18 Limiting Value: Currently not available</p> <p>9.19 Reid Vapor Pressure: Currently not available</p>											
NOTES															

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9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
C U R R E N T L Y N O T A V A I L A B L E			C U R R E N T L Y N O T A V A I L A B L E		C U R R E N T L Y N O T A V A I L A B L E		C U R R E N T L Y N O T A V A I L A B L E

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
I N S O L U B L E		68	0.348	68	0.00646		C U R R E N T L Y N O T A V A I L A B L E