

## Thermal Physics II

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### Assignment II

**Q.1)** Consider a metal (say Copper) at  $300K$  with the following values,  $V = 7.06 \text{ cm}^3/\text{mol}$ ,  $K_T = 7.78 \times 10^{-12} \text{ N/m}^2$ ,  $\beta = 50.4 \times 10^{-6} \text{ K}^{-1}$ ,  $C_p = 24.5 \text{ J/molK}$ . Determine  $C_v$ .

**Q.2)** Prove that the ratio of adiabatic  $\left[\alpha_S = \frac{1}{V}(\frac{\partial V}{\partial T})_S\right]$  to isobaric  $\left[\alpha_P = \frac{1}{V}(\frac{\partial V}{\partial T})_P\right]$  coefficient of expansion is  $\frac{1}{1-\gamma}$ . Also, prove that the ratio of adiabatic  $\left[E_S = -V(\frac{\partial P}{\partial V})_S\right]$  to isothermal  $\left[E_T = -V(\frac{\partial P}{\partial V})_T\right]$  elasticities is equal to the ratio of specific heats.

**Q.3)** Prove that the ratio of adiabatic  $\left[\beta_S = \frac{1}{P}(\frac{\partial P}{\partial T})_S\right]$  to isochoric  $\left[\beta_V = \frac{1}{P}(\frac{\partial P}{\partial T})_V\right]$  pressure coefficient of expansion is  $\frac{\gamma}{\gamma-1}$ .

**Q.4)** (a) If equation of state of certain material satisfies  $P = \frac{RT}{V}(1 + \frac{B''}{V})$  where  $B'' = B''(T)$ , show that

$$C_V = -\frac{RT}{V} \frac{d^2}{dT^2}(B''T) + C_V^\infty,$$

where  $C_V^\infty$  represents the value of  $C_V$  when  $V$  is very large. (b) In case  $P = \frac{RT}{V}(1 + B'P)$  where  $B' = B'(T)$ , show that

$$C_P = RTP \frac{d^2}{dT^2}(B'T) + C_P^0,$$

where  $C_P^0$  represents the value of  $C_P$  when pressure tends to zero.

**Q.5)** Using Berthelot's equation of state  $P = \frac{RT}{V-b} - \frac{a}{TV^2}$ , show that the critical constants are

$$P_c = \frac{1}{12b} \sqrt{\frac{2aR}{3b}}, \quad V_c = 3b, \quad T_c = \sqrt{\frac{8a}{27bR}}; \quad \frac{RT_c}{P_c V_c} = \frac{8}{3}.$$

**Q.6)** The boiling point of a liquid at pressure  $P_0$  is  $T_0$ . Its molar latent heat of vaporisation is  $L$  and molar volume of the liquid phase is negligible as compared to vapour phase. The vapour phase obeys the ideal gas equation. Show that the boiling point  $T$  at pressure  $P$  is given by,

$$\ln\left(\frac{P}{P_0}\right) = \frac{L}{RT_0} \left(1 - \frac{T_0}{T}\right).$$