

1. If that temperature is T , then from the ~~red~~ relation

$$C/5 = \frac{F-32}{9}, \text{ we get } \frac{T}{5} = \frac{T-32}{9}$$

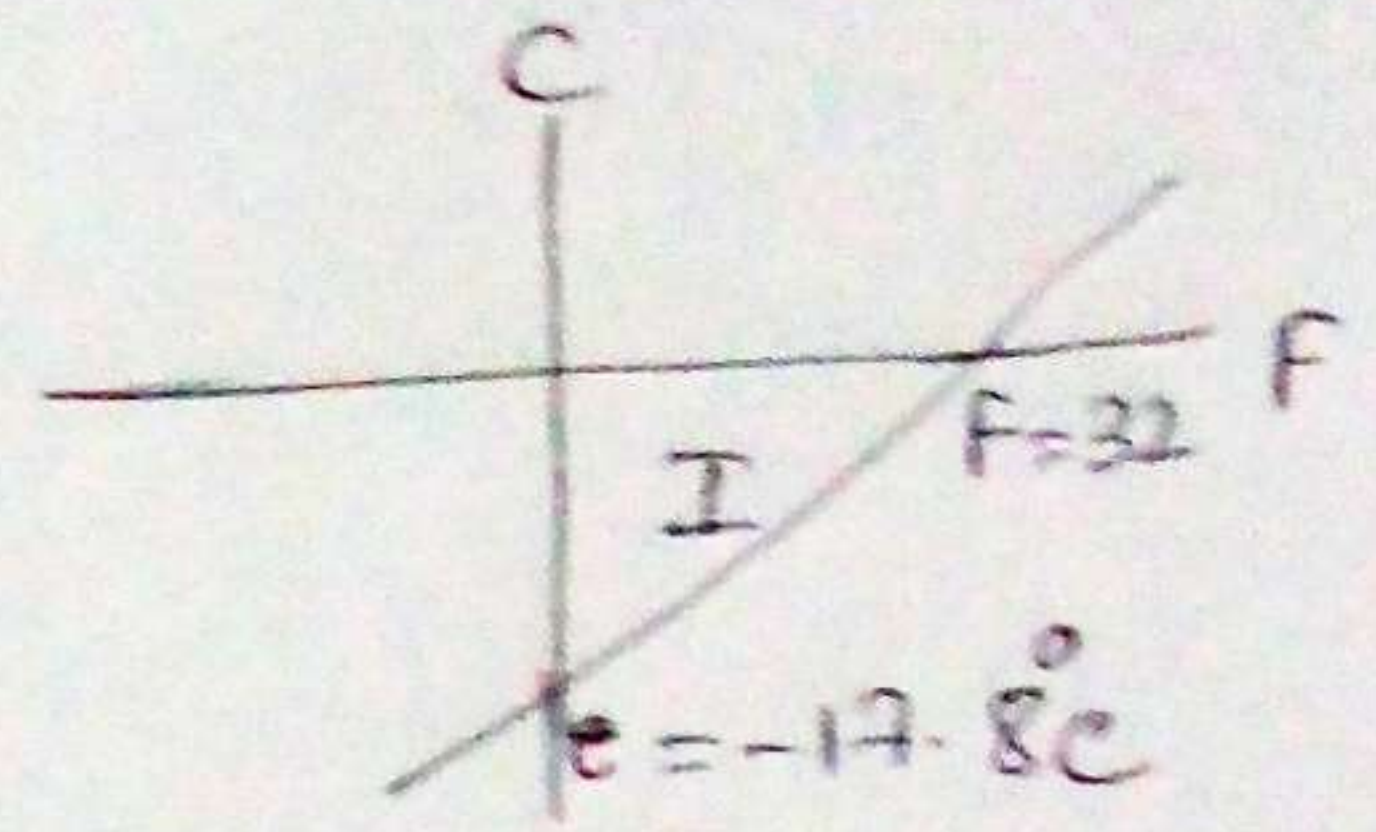
$$\Rightarrow 9T - 5T = -160$$

$$\text{or } T = -\frac{160}{4} = -40^\circ \text{ (unit can be both)}$$

2. If $F = 32$, then $C = 0^\circ\text{C}$

$$\text{if } F = 0, \text{ then } C = -\frac{5 \times 32}{9} = -17.8^\circ\text{C}$$

So "I" was the correct line.



$$\begin{aligned} 3. \quad T &= 100 \frac{P_T - P_0}{P_{100} - P_0} = 100 \frac{100 - 80}{109.3 - 80} \\ &= \frac{20 \times 100}{29.3} = 68.26^\circ\text{C} \end{aligned}$$

$$\begin{aligned} 4. \quad T_1 V_1^{\gamma-1} &= T_2 V_2^{\gamma-1} \\ \therefore T_2 &= T_1 \left(\frac{V_1}{V_2} \right)^{\gamma-1} = (273 + 20) \left(\frac{1}{1.4} \right)^{1.4-1} \\ &= 256.1^\circ\text{K} = (256.1 - 273)^\circ\text{C} = -16.9^\circ\text{C} \end{aligned}$$

$$5. \quad P_1 V_1 = P_2 V_2 \quad \text{"isothermal expansion"}$$

$$\therefore P_2 = P_1 \frac{V_1}{V_2} = \frac{1}{2} \times 10^6$$

$$P_2 V_2^\gamma = P_3 V_3^\gamma \quad \text{"adiabatic expansion"}$$

$$\therefore P_3 = P_2 \left(\frac{V_2}{V_3} \right)^\gamma = \frac{1}{2} \times 10^6 \times \left(\frac{1}{2} \right)^{1.4}$$

$$= 1.895 \times 10^5 \text{ dynes/cm}^2$$