(a)

VIIIA	18	2 4.0026 He	1s ² Helium	18.998 10 20.180	Ne 18 ² 28 ² 2p ⁶	18 39.948	Ar	[Ne]3s ² 3p ⁶	5	36 83.80 K	[Ar]3d ¹⁰ 4s ² 4p ⁶	54 131.29	Xe	Kr 4d ¹⁰ 5s ² 5p ⁶ Xenon	86 (222)	Rn	Xe 4f ¹⁴ 5d ¹⁰ 6s ² 6p ⁶	TOTAL STATE		
VIIA	17				下	17 35.453	ರ	[Ne]38 ² 3p ⁵	Chlorine	35 79.904 Br	[Ar]3d ¹⁰ 4s ² 4p ⁵	53 126.90	I	[Kr]4d ¹⁰ 5s ² 5p ⁵ Iodine	(210)		[Xe]4r ¹⁴ 5d ¹⁰ 6s ² 6p ⁵	Asiamic		
VIA	16			8 15.999 9	O 18 ² 28 ² 2p ⁴	16 32.066	S	$[\mathrm{Ne}]3\mathrm{s}^23\mathrm{p}^4$	Sul	ň	[Ar]3d ¹⁰ 4s ² 4p ⁴	52 127.60	Te	[Kr]4d ¹⁰ 5s ² 5p ⁴ Tellurium	84 (210) 85	Po	[xe]4t ¹⁴ 5d ¹⁰ 6s ² 6p ⁴			
VA	15			7 14.007	N ^{18²28²2p³}	15 30.974	Ь	$[\mathrm{Ne}] \mathrm{3s}^2 \mathrm{3p}^3$	Phos	ri γ	[Ar]3d ¹⁰ 4s ² 4p ³	Arsenic 51 121.76	Sb	[Kr]4d ¹⁰ 5s ² 5p ³ Antimony	83 208.98	Bį	Xe 4f ¹⁴ 5d ¹⁰ 6s ² 6p ³	mpinera		
IVA	14			6 12.011 7	C 18 ² 28 ² 2p ²	14 28.086	Si	$[\mathrm{Ne}]3\mathrm{s}^23\mathrm{p}^2$	ij.	32 72.61 Ge	[Arj3d ¹⁰ 4s ² 4p ²	50 118.71	Sn	[Krj4d ¹⁰ 5s ² 5p ² Tin	82 207.2	Pb	[Xe]4f ^{]4} 5d ¹⁰ 6s ² 6p ²	read		
IIIA	13			5 10.811 6	$\displaystyle \mathop{B}_{^{1s^2ss^2zp^1}}$	13 26.982	Al	$[\mathrm{Ne}]3\mathrm{s}^23\mathrm{p}^1$	Alu	n	[Ar]3d ¹⁰ 4s ² 4p ¹	4	ln	Kr 4d ¹⁰ 5s ² 5p ¹ Indium	81 204.38	ŢŢ	Xe 4f ¹⁴ Sd ¹⁰ 6s ² 6p ¹			
IIB	12								6	Zn	[Ar]	48 112.41	Cd	[Kr]4d ¹⁰ 5s ² Cadmium	w	Hg	xe 4f ¹⁴ 5d ¹⁰ 6s ²	maram		
IB	11								8	Cu	Ϋ́	47 107.87	Ag	×	79	Au	Xe 4f 4 Sd ¹⁰ 6s ¹	200		
	10			ro	ids	1				$N_{\mathbf{i}}$	[Ar]3d ⁸ 4s ²	46 106.42	Pd	[Kr]4d ¹⁰ 5s ⁰ Palladium	1.	Pt	Xe 4f ¹⁴ 5d ⁹ 6s ¹	Tradition of		
VIII	6			metals	metalloids nonmetals					Co Co	[Ar]3d ⁷ 4s ²	45 102.91	Rh	[Kr]4d ⁸ 5s ¹ Rhodium	100	Ir	[Xe]4r ¹⁴ 5d ⁷ 6s ²			
	∞			weight	gases liquids solids					In Fe	[Ar]3d ⁶ 4s ²	44 101.07	Ru	[Kr]4d ⁷ 5s ¹ Ruthenium		SO	[Xe]4f ¹⁴ 5d ⁶ 6s ²			
VIIB	7			- atomic weight	symbol red: orange: black:				i	g 2		43 (98)	Tc	[Kr]4d ⁵ 5s ¹ [Kr]4d ⁵ 5s ² Molybdenium Technetium	75 186.21	Re	Xe 4f ¹⁴ 5d ⁵ 6s ²			
VIB	9			-1 1.0079	二	ingement.			3	Cr		42 95.94	Mo	[Kr]4d ⁵ 5s ¹ Molybdenium	73 180.95 74 183.85 75 186.21	M	Xej4r ¹⁴ Sd ⁴ 6s ²	magaini		
VB	ß			atomic number —	configuration —				0	Z3 50.942 V		41 92.906	qN	[Kr]4d ⁴ 5s ¹ Njobjum		Ta	[Xe]4f ¹⁴ 5d ³ 6s ²	Tantana		
IVB	4			atomic	electron configuration					7. Ti		40 91.224	Zr	[Kr]4d ² 5s ² Zirconium	72 178.49	Hf	Xe 4f ⁴ 5d ² 6s ²			
IIIB	က									Sc 44.956	[Ar]3d ¹ 4s ²	87.62 39 88.906	Υ	[Kr]4d ¹ 5s ² Yttrium	57	La-		89- 103	Ac-Lr	
IIA	2			4 9.0122	$\mathop{\rm Be}_{{}^{1s^22s^2}}$	11 22.990 12 24.305	Mg	[Ne]3s ²	Magnesium	K Ca	[Ar]48 ²	ă I	Sr	[Kr]5s ² Strontium	- m	Ba	[Xe]6s ²	(223) 88 (226)	Ra	[Rn]7s ² Radium
IA	1	1 1.0079	1s ¹ Hydrogen	3 6.941	$\Gamma_{1s^2 2s^1}$	11 22.990	Na	[Ne]3s1	Sodium	K K	[Ar]4s ¹	37 85.468 38	Rb	[Kr]5s ¹ Rubidium	55 132.91	Cs	[Xe]6s ¹	87 (223)	Fr	[Rn]7s ¹ Francium
		П			2		ď)		4	-		r.)		9	>		7	

	57 138.91	58 140.12	59 140.91	60 144.24	61 (147)	62 150.36	63 151.96	64 157.25	65 158.93	66 162.50	57 13891 58 140.12 59 140.91 60 144.24 61 (147) 62 150.36 63 151.96 64 157.25 65 158.93 66 162.50 67 164.93 68 167.26 69 168.93 70 173.04 71 174.97	68 167.26	69 168.93	70 173.04	71 174.97
Lanthanides La Ce Pr Nd Pm Sm Eu Gd Tb Dy Ho Er Tm Yb Lu	La	Ce	Pr	pN	Pm	Sm	Eu	Вd	Tb	Dy	Но	Er	Tm	Yb	Γ'n
	$[\mathrm{Xe}]4f^0\mathrm{Sd}^16\mathrm{s}^2$	[xe]4f ⁰ 5d ¹ 6s ² [xe]4f ² 5d ⁰ 6s ²	[Xe]4f ³ 5d ⁰ 6s ²	[Xe]4f ⁴ 5d ⁰ 6s ²	Xe 4f ⁵ 5d ⁰ 6s ²	[Xe]4f ⁶ 5d ⁰ 6s ²	[Xe]4f ⁷ 5d ⁰ 6s ²	$[{\rm Ke}]_4 f^4 {\rm Sd}^0 6 s^2 = [{\rm Ke}]_4 f^5 {\rm Sd}^0 6 s^2 = [{\rm Ke}]_4 f^5 {\rm Sd}^0 6 s^2 = [{\rm Ke}]_4 f^7 {\rm Sd}^1 6 s^$	$[Xe]4f^95d^06s^2$	$[\mathrm{Xe}]4\mathrm{f}^{10}\mathrm{5d}^{0}\mathrm{6s}^{2}$	[Xe]4f ¹⁰ 5d ⁰ 6s ² [Xe]4f ¹¹ 5d ⁰ 6s ² [$[\mathrm{Xe}]4\mathrm{f}^{12}\mathrm{5d}^{0}6\mathrm{s}^{2}$	[Xe]4f ¹³ 5d ⁰ 6s ²	Xe 4r ¹³ 5d ⁰ 6s ² Xe 4r ¹⁴ 5d ⁰ 6s ²	[Xe]4f ¹⁴ 5d ¹ 6s ²
	Lanthanum	Cerium	Praseodymium	Neodynium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Certum Proscotymism Neodynium Promethium Bronethium Buropium Gadolinium Gedolinium Terbium Dysprosium Holmium Brbium Thulium Thulium	Erbium	Thulium	Ytterbium	Lutetium
	89 (227)	90 232.04	91 231.04	92 238.03	93 (237)	94 (242)	95 (243)	96 (247)	97 (247)	98 (249)	89 (227) 90 232.04 91 231.04 92 238.08 93 (237) 94 (242) 95 (243) 96 (247) 97 (247) 97 (247) 98 (249) 99 (254) 100 (253) 101 (256) 102 (254) 103	100 (253)	101 (256)	(254)	(257)
Actinides	Ac	Th	Pa	n	dN	Pu	Am	Cm	Bk	Cf	Ac Th Pa U Np Pu Am Cm Bk Cf Es Fm Md No Lr	Fm	Md	No	Ľ
	$ \mathrm{Rn} 5f^06d^47s^2$	[Rn]5f ⁰ 6d ² 7s ²	$\rm [Rn]5f^26d^17s^2$	[Rn]5f ³ 6d ¹ 7s ² [Rr	[Rn]5f ⁴ 6d ¹ 7s ²	$[\rm Rn]5r^66d^07s^2$	[Rn]5f ⁷ 6d ⁰ 7s ²	$[{ m Rn}]{ m Sf}^4{ m 6d}^4{ m 7s}^2$ $[{ m Rn}]{ m Sf}^6{ m 6d}^0{ m 7s}^2$ $[{ m Rn}]{ m Sf}^7{ m 6d}^4{ m 7s}^2$ $[{ m Rn}]{ m Sf}^7{ m 6d}^4{ m 7s}^2$	[Rn]5f ⁹ 6d ⁰ 7s ²	$[\mathrm{Rn}]\mathrm{Sf}^{10}\mathrm{6d}^{07\mathrm{s}^2}$	$ \text{Rn} \text{St}^9\text{od}^{0}{\tau_8}^2 \text{Rn} \text{St}^{10}\text{od}^{0}{\tau_8}^2 \text{Rn} \text{St}^{11}\text{od}^{0}{\tau_8}^2 \text{Rn} \text{St}^{12}\text{ed}^{0}{\tau_8}^2 \text{Rn} \text{St}^{13}\text{od}^{0}{\tau_8}^2 \text{Rn} \text{St}^{13}\text{od}^{0}{\tau_8}$	$ \mathrm{Rn} 5 f^{12} 6 d^{0} 7 s^2$	[Rn]5f ¹³ 6d ⁰ 7s ²	[Rn]5f ¹⁴ 6d ⁰ 7s ²	Rn 5f ¹⁴ 6d ^{17s²}
	Actinium	Thorium	Proactinium	Uranium	Neptunium	Plutonium	Americium	Curium	Berkelium	Californium	Thorium Proactinium Uranium Neptunium Putonium Putonium Americium Curium Berkelium Californium Berkelium Guinium Berkelium Nobelium Nobeli	Fermium	Mendelevium	Nobelium	Lawrencium

Fig. 2.9 (continued)