HW5

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1 Thermal Physics - Homework 5

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```
[3]: import numpy as np
  import seaborn as sns
  import matplotlib.pyplot as plt
  import scipy.constants as con
  from scipy.special import comb
  from matplotlib_inline.backend_inline import set_matplotlib_formats

set_matplotlib_formats('pdf')
  plt.rcParams |= {
     'text.usetex': True,
     'figure.figsize': (10, 4)
  }
  sns.set_theme()
  set_matplotlib_formats('svg', 'pdf')
```

1.1 Question 1

Consider two-state paramagnet of N=150 identical dipoles in a magentic field of strength B, each of which has a magnetic dipole moment of μ

1.1.1 Numeriacal Solution

$$\frac{U}{\mu B} = N - 2N_\uparrow \qquad \frac{S}{k} = \Omega(N_\uparrow) = \binom{N}{N_\uparrow}$$

1.1.2 Analytical Solution

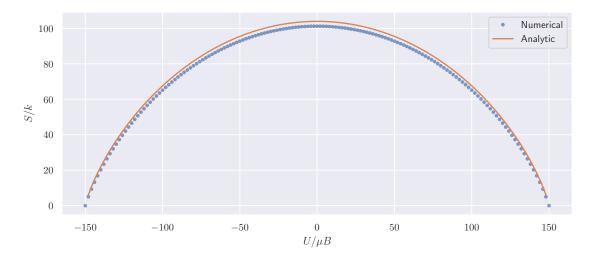
We take (3.28)

$$\frac{S}{k} = N \ln N - N_{\uparrow} \ln N_{\uparrow} - (N - N_{\uparrow}) \ln (N - N_{\uparrow})$$

and for substituting (3.30) in (3.31) we have

$$\frac{U}{\mu B} = -N \tanh \frac{1}{2} \ln 2 \frac{N_{\uparrow}}{N - N_{\uparrow}}$$

```
[39]: N = 150
                         N_{up} = np.arange(N + 1)
                         # unitless energy: U/mu B
                         energy = N - 2 * N_up
                         # unitless entropy: S/k
                         entropy = np.log(comb(N, N_up))
                         # remove first and last val, invlid
                         N_{up} = N_{up}[1:-1]
                         # analytic unitless energy: U/mu B
                         energy_anal = -N * np.tanh(0.5 * np.log((N - energy[1:-1])) / (N + energy[1:-1])
                            -1])))
                         # analytic unitless entropy: S/k
                         entropy_anal = N * np.log(N) - N_up_ * np.log(N_up_) - (N - N_up_) * np.log(N - vertex + ve
                             →N_up_)
                         plt.plot(energy, entropy, '.', label='Numerical', alpha=0.7)
                         plt.plot(energy_anal, entropy_anal, label='Analytic')
                         plt.xlabel(r'$U/\mu B$')
                         plt.ylabel(r'$S/k$')
                         plt.legend()
                         plt.show()
```



1.2 Question 1

1.2.1 Numerical Solution

We use the following realtion between entropy, energy and temperatur

$$T = \frac{\partial U}{\partial S} \qquad \frac{kT}{\mu B} = -2 \frac{\Delta N_{\uparrow}}{\Delta \Omega(N_{\uparrow})}$$

Note is always $\Delta N_{\uparrow}=1$

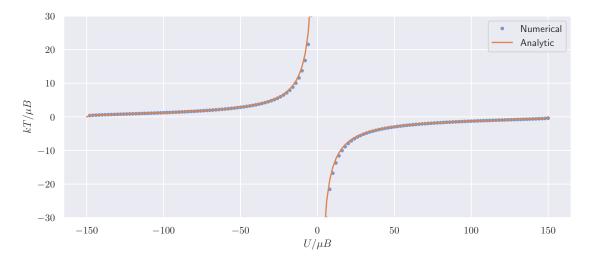
1.2.2 Analytic Solution

Using

$$\frac{\mu B}{kT} = \frac{1}{2} \ln \left(\frac{N - U/\mu B}{N + U/\mu B} \right)$$

```
[58]: # Numerical unitless temprature
temp = -2 / np.diff(entropy)
# Analytical unitless temprature
with np.errstate(divide='ignore'):
    # ignore warning of dividing by zero
    temp_anal = 1 / (0.5* np.log((N - energy) / (N + energy)))

plt.plot(energy[:-1], temp, '.', label='Numerical', alpha=0.7)
plt.plot(energy, temp_anal, label='Analytic')
plt.xlabel(r'$U/\mu B$')
plt.ylabel(r'$kT/\mu B$')
plt.ylim([-30, 30])
plt.legend()
plt.show()
```



1.3 Question 3

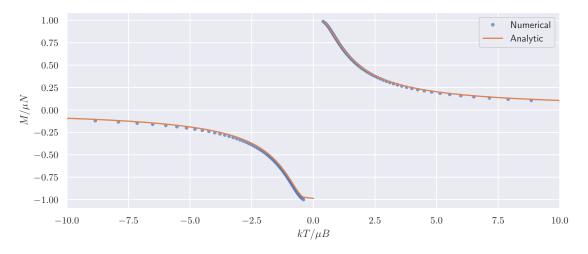
$$\frac{M}{N\mu} = -\frac{U}{B}$$

From
$$(3.32)$$

$$\frac{M}{N\mu} = \tanh\frac{\mu B}{kT}$$

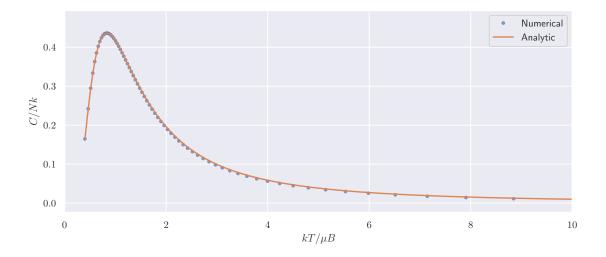
```
[78]: # Numerical Magetiztion
x = 1 / temp
# Analytic Magetiztion
Mag = - energy / N

plt.plot(temp, Mag[:-1], '.', label='Numerical', alpha=0.7)
plt.plot(temp_anal[:-1], np.tanh(x), label='Analytic')
plt.xlabel(r'$kT/\mu B$')
plt.ylabel(r'$M/\mu N$')
plt.xlim([-10, 10])
plt.legend()
plt.show()
```



1.4 Question 4

We use (3.33) For analytical solution.



1.5 Question 4

Thermodynamic identity is

$$dU = T dS - P dV$$

The enthaply is given by

$$H = U + PV$$

At constant volume the identity becomes $dU=T\,dS$, and the definition of capacity at constant volume is $C_V=\frac{\partial U}{\partial T}$

Then we substituse the reduced indentity into the later, we get

$$C_V = T \left(\frac{\partial S}{\partial T} \right)_V$$

The differential of H is

$$dH = dU + V dP + P dV$$

Or from the identity

$$dH = T dS + V dP$$

But at constant pressure we get, $H=T\,dS$ By dimension analysis, the capacity at constant pressure is

$$C_P = \left(\frac{\partial H}{\partial T}\right)_P = T \left(\frac{\partial S}{\partial T}\right)_P$$